## NiFe double hydroxide coated on sulfur-modified NiMoO<sub>4</sub> nanorods as core-shell structured catalysts for oxygen evolution reaction

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2 Key Laboratory of New Inorganic Nonmetallic Composite of Handan, College of Materials Science and Engineering, Hebei University of Engineering, Handan, 056038, China Table S1 The overpotentials of NiFe LDH@S-NiMoO<sub>4</sub> are compared with related catalytic electrodes in 1 M KOH electrolyte solution.

Catalysts	Current density	Overpotential	Ref.
CoP <sub>3</sub> /NiMoO <sub>4</sub> heterostructures on Ni foam	10 mA cm <sup>-2</sup>	347 mV	(1)
Fe-doped NiMoO <sub>4</sub>	10 mA cm <sup>-2</sup>	299 mV	(2)
FeOOH modified NiMoO <sub>4</sub> nanowires	100 mA cm <sup>-2</sup>	253 mV	(3)
Fe-S-NiMoO <sub>4</sub> / MoO <sub>3</sub> on Ni foam	100 mA cm <sup>-2</sup>	249 mV	(4)
N-NiMoO <sub>4</sub> /NiS <sub>2</sub> Nanowires/Nanosheets	10 mA cm <sup>-2</sup>	283 mV	(5)
Ni <sub>9</sub> S <sub>8</sub> /MoS <sub>2</sub> Nanosheets Decorated NiMoO <sub>4</sub> Nanorods	10 mA cm <sup>-2</sup>	360 mV	(6)
NiFe <sub>2</sub> O <sub>4</sub> /NiMoO <sub>4</sub> Nanorods on Ni foam	50 mA cm <sup>-2</sup>	270 mV	(7)
NiMoO <sub>4</sub> @Co <sub>3</sub> O <sub>4</sub> Core–Shell Nanorods	100 mA cm <sup>-2</sup>	282 mV	(8)
Core/shell -structured NiMoO <sub>4</sub> @ MoSe <sub>2</sub> /NixSey Nanorod on Ni Foam	10 mA cm <sup>-2</sup>	290 mV	(9)
P/NiFe doped NiMoO <sub>4</sub> micro-pillars arrays	100 mA cm <sup>-2</sup>	360 mV	(10)
P-doped Ni(OH) <sub>2</sub> /NiMoO <sub>4</sub> hierarchical nanosheet arrays grown on Ni foam	100 mA cm <sup>-2</sup>	380 mV	(11)
Oxygen vacancy enriched NiMoO <sub>4</sub> nanorods	50 mA cm <sup>-2</sup>	255 mV	(12)
Ce-Doped Ni–S nanosheets on Ni foam supported NiMoO <sub>4</sub> micropillars	100 mA cm <sup>-2</sup>	244 mV	(13)
NiMoO <sub>4</sub> /NiFe LDH/rGO multicomponent nanosheets	10mA cm <sup>-2</sup>	270 mV	(14)
NiFe <sub>2</sub> O <sub>4</sub> -x/NiMoO <sub>4</sub> nanowire arrays on Ni Foam	600 mA cm <sup>-2</sup>	326 mV	(15)
Crystalline-Amorphous Ni <sub>2</sub> P <sub>4</sub> O <sub>12</sub> /NiMoO <sub>x</sub> Nanoarrays	20 mA cm <sup>-2</sup>	250 mV	(16)
This work	100 mA cm <sup>-2</sup>	277mV	



Fig. S1 TEM images (a-d) of NiFe LDH<sub>30min</sub>@S-NiMoO<sub>4</sub>, NiFe LDH@S-NiMoO<sub>4</sub>, NiFe LDH<sub>50min</sub>@S-NiMoO<sub>4</sub> and NiFe LDH<sub>60min</sub>@S-NiMoO<sub>4</sub>.



Fig. S2 EDS mapping images (a-c) of NiFe LDH<sub>30min</sub>@S-NiMoO<sub>4</sub>, NiFe LDH<sub>50min</sub>@S-NiMoO<sub>4</sub> and NiFe LDH<sub>60min</sub>@S-NiMoO<sub>4</sub>.



Fig. S3 Atomic specific gravity of Fe element at different times of electrodeposition.



Fig. S4 XRD patterns of NiFe LDH<sub>30min</sub>@S-NiMoO<sub>4</sub>, NiFe LDH<sub>50min</sub>@S-NiMoO<sub>4</sub> and NiFe LDH<sub>60min</sub>@S-NiMoO<sub>4</sub>.



Fig. S5. Low-magnification (a) and High-magnification (b) SEM image of NiFe  $LDH_{30min}@S-NiMoO_4$ .

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