

Supporting Information

Designed Synthesis of MOR Zeolites using Gemini-type Bis(methylpyrrolidinium)

Dications as Structure Directing Agents and Their DME Carbonylation

Performance

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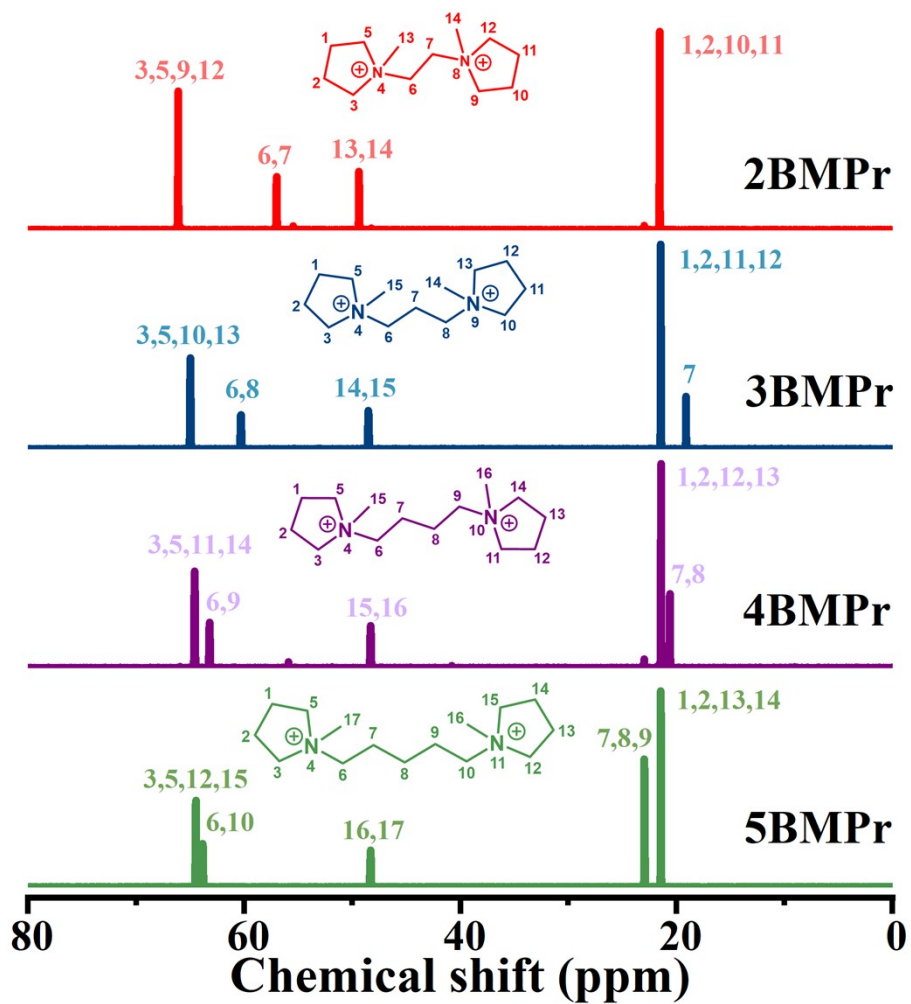


Fig. S1 The ^{13}C NMR spectra of the four organic dications solvated in D_2O .

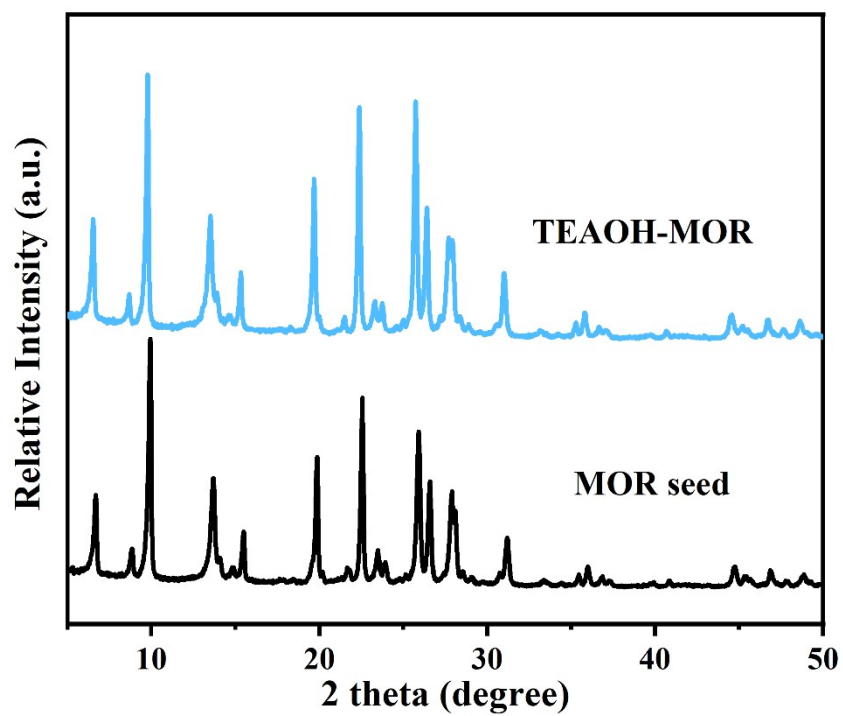


Fig. S2 XRD patterns of the seed and reference samples.

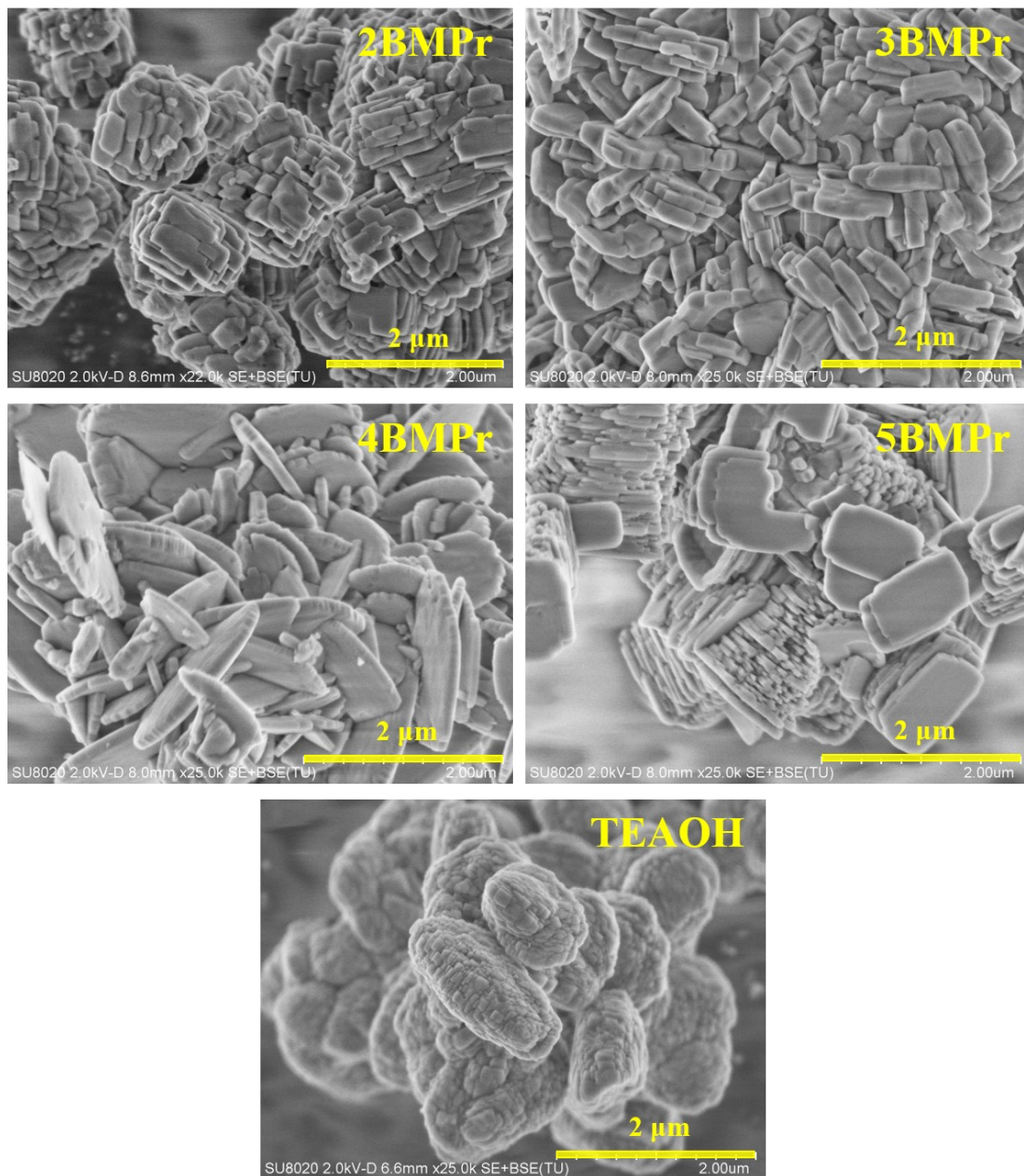


Fig. S3 SEM images of the as-synthesized MOR samples.

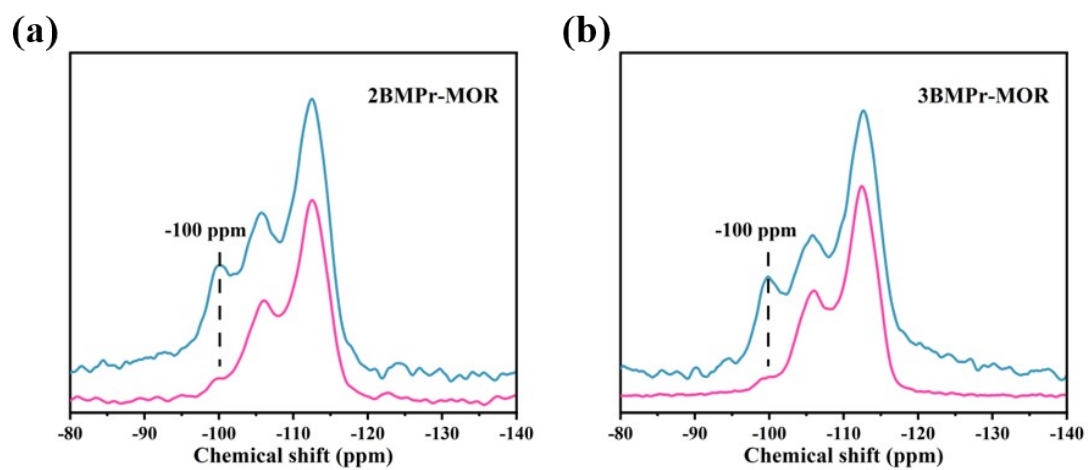


Fig. S4 The ^{29}Si MAS NMR (pink) and ^1H - ^{29}Si CP MAS NMR spectra (blue) of the as-synthesized samples: (a) 2BMPPr-MOR, (b) 3BMPPr-MOR.

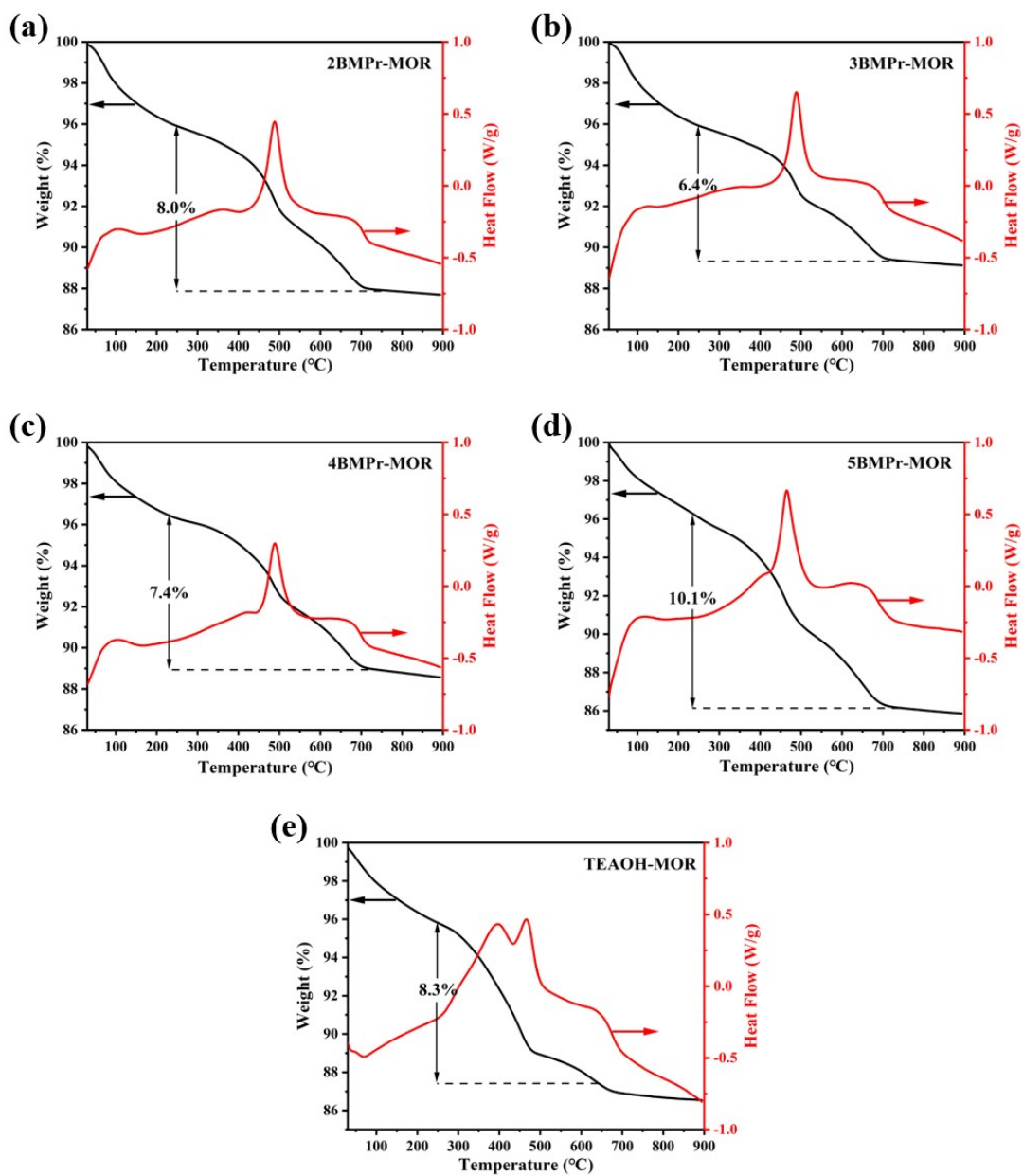


Fig. S5 Thermal analysis of the as-synthesized MOR samples: (a) 2BMPPr-MOR, (b) 3BMPPr-MOR, (c) 4BMPPr-MOR, (d) 5BMPPr-MOR, (e) TEAOH-MOR.

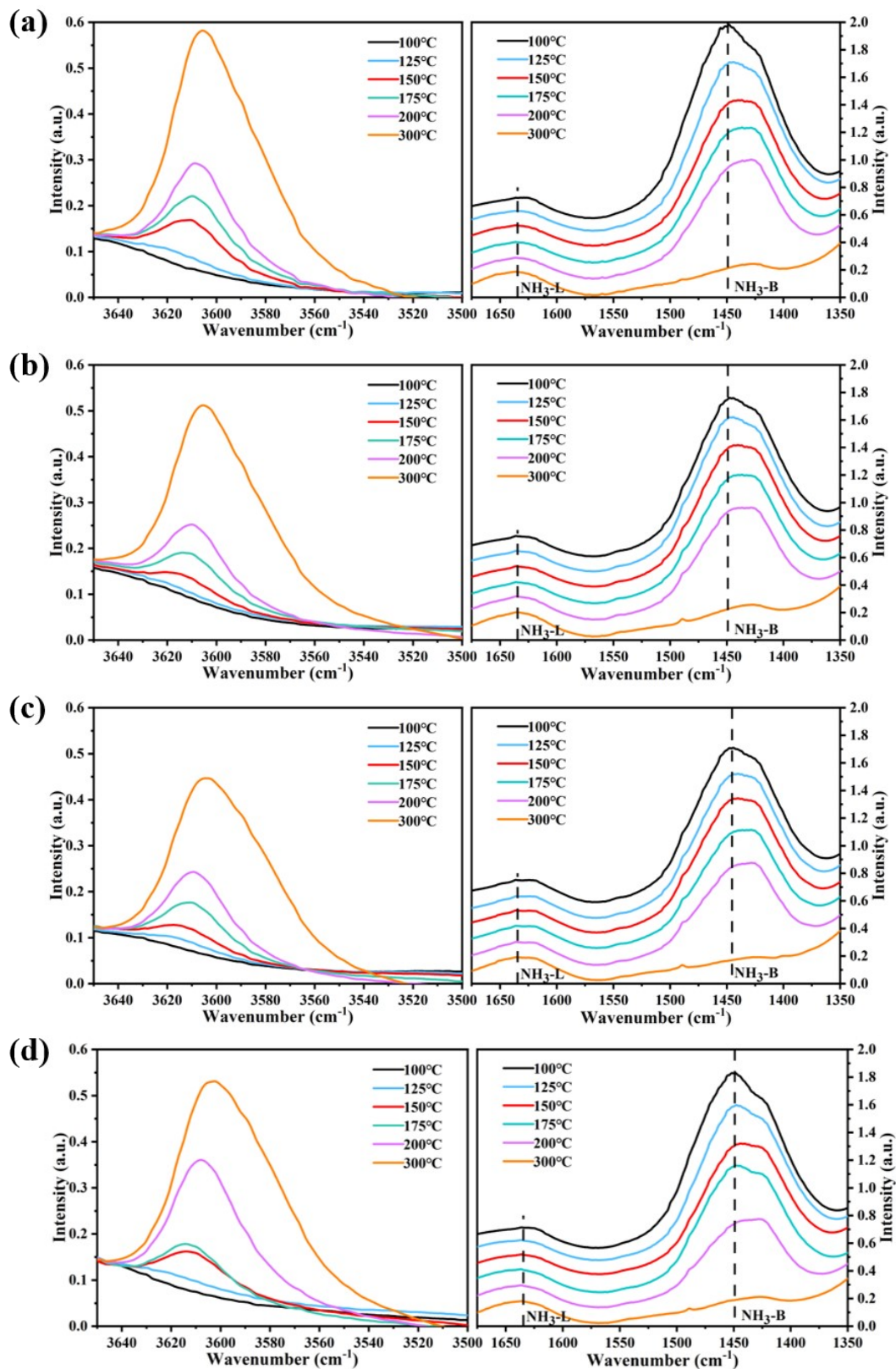


Fig. S6 NH_3 -adsorbed FTIR spectra of H-MOR samples after NH_3 desorption at different temperatures: (a) 2BMP_r-MOR, (b) 3BMP_r-MOR, (c) 4BMP_r-MOR, (d) TEAOH-MOR.

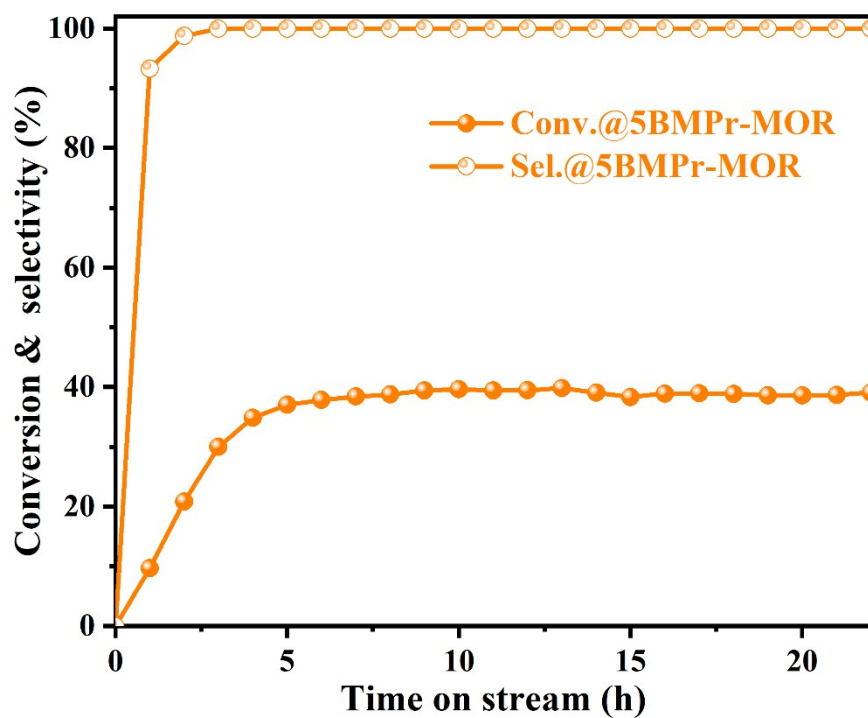


Fig. S7 The catalytic performance of DME carbonylation over 5BMPPr-MOR. Reaction conditions: 200 °C, 2 MPa, DME/CO/N₂ = 5/35/60, GHSV = 7200 mL/g/h.

Table S1. Synthesis results for MOR zeolites using nBMPPr [n=2, 3, 4, 5].

SDA	Gel molar composition ^a		Times(days)	product		Solid yield ^c (%)	
	SiO ₂ /Al ₂ O ₃	Na ₂ O/SiO ₂		Phase	SAR ^b		
2BMPPr	22	0.18	2	MOR	10.2	92	
		0.2	2	MOR	9.5	84	
	26	0.18	2	MOR	12.0(12.4)	86	
		0.2	2	MOR	10.5	76	
	30	0.2	2	MOR (amorphous)			
			2.5	MOR	12.9	85	
			3	MOR	12.4	83	
	34	0.2	2	Amorphous			
			2.5	MOR+Quartz			
	3BMPPr	22	0.2	2	MOR	9.4	82
0.18			2	MOR	11.9(12.0)	84	
26		0.2	2	MOR	10.9	85	
			3	MOR+ANA			
			4	MOR+Quartz			
			28	0.2	2	MOR	12.1
30		0.2	2.5	MOR	11.9	78	
			2	Amorphous			
			2.5	MOR	12.4	82	
			3	MOR+ZSM-5			
34	0.2	2	Amorphous				
		2.5	MOR+Nu-6				
		3	MOR+Quartz				
4BMPPr	26	0.18	5	Amorphous			
		0.24	2	Amorphous			
	30	0.18	5	Amorphous			
		0.2	4	MOR+ZSM-5			
		0.22	3	Amorphous			
	32	0.18	0.24	4	MOR	9.7	65
			4.5	Amorphous			
			5	MOR	12.1(12.0)	73	
5BMPPr	26	0.16	5.5	MOR+ZSM-5			
			4	Amorphous			
			5	MOR+ZSM-5			
	30	0.18	3	Amorphous			
			4	MOR+ZSM-5+SSZ-24			
			3	MOR+ZSM-5			
30	0.22	0.2	2	MOR	9.6(9.4)	64	
		0.24	2	MOR	8.6	55	
		0.18	5	MOR+ZSM-5+ZSM-57			
	0.24	2	MOR+ZSM-5+SUZ-4				

32	0.18	5	MOR+ZSM-5+SUZ-4
34	0.24	2	Amorphous

^a Gel molar composition: H₂O/SiO₂=15, OSDA/SiO₂=0.12, seed addition: 6 wt% relative to SiO₂ resource; crystallization temperature: 180 °C. The SAR (Si/Al ratio) of MOR seed are 12.9 derived from XRF.

^b The SAR was derived from XRF and the value in the bracket was calculated from ²⁹Si MAS NMR.

^c The yield was calculated based on the mass of silica and alumina.

Table S2. OSDA stabilization energy for the four dications used in this work.

OSDA	OSDA stabilization energy (kJ/mol Si) ^a
2BMP _r	-5.01
3BMP _r	-4.85
4BMP _r	-4.80
5BMP _r	-4.60

^a OSDA stabilization energy (nBMP_r) = E(total) - E(zeolite) - E(nBMP_r).

Table S3. Summary of the product STY in the MOR-catalyzed DME carbonylation reaction.

Literature	Reaction conditions				STY of MA (mmol/h/g)	Pyridine modification
	DME : CO (%)	T (°C)	P (MPa)	GHSV (mL/g/h)		
[1]	2:93	165	1	12024	1.9	
[2]	5:50	200	1	1250	0.8	Yes
[3]	5:50	200	1	1250	1.7	
[4]	3:95.5	210	1.5	5280	4.0	
[5]	5:35	200	2	1500	1.3	
[6]	1:47	200	1.5	4500	1.8	
[7]	5:35	200	3	1500	2.8	
[8]	2.4:50	210	2	2100	3.2	
[9]	1:49	200	1.5	6000	1.6	
[10]	5:50	200	1	1250	1.3	
[11]	5:35	200	2	1500	3.0	Yes
[12]	1:47	200	1.5	3000	1.1	Yes
[13]	5:76	200	1	2500	2.5	
[14]	1:49	200	1.5	6000	1.6	
[15]	2:98	190	2	2000	1.8	
[16]	3:95.5	190	1.5	2640	2.8	
[17]	5:35	200	2	3600	7.2	Yes
[18]	1:49	200	1.5	6000	6.5	
[19]	1:49	200	1.5	6000	7.2	
[20]	1:49	200	1.5	6000	6.4	
[21]	5:35	200	2	2250	4.5	
[22]	5:35	200	2	3600	6.8	Yes
[23]	5:35	200	2	3600	6.4	Yes
Our work	5:35	200	2	7200	12.5	Yes

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