

Electronic Supplementary Information

Self-assembled ultrasmall mixed Co-W phosphide nanoparticles on pristine graphene with remarkable synergistic effects as highly efficient electrocatalysts for hydrogen evolution

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Contents

1. FTIR spectra and high-resolution W 4f XPS spectra of G@HPW and G@HPW (without

- HR) and TEM images of G@HPW (without HR) and G@HPW (Fig. S1).
- EDX mapping of G@HPW after the hydrothermal reaction (Fig. S2).
 - High-resolution Co 2p_{3/2} and P 2p XPS spectra of the sample before phosphidation (Fig. S3).
 - Size distribution of Co-W-P NPs in G@Co-W-P (Fig. S4).
 - TEM and HRTEM images of Co-W-P/G and CoP/G (Fig. S5).
 - XRD patterns of one-step hydrothermal reaction and the sample with no HPW added in the second-step hydrothermal reaction (Fig. S6).
 - TEM image and high-resolution W 4f (B) and P 2p (C) XPS spectra of WP₂/G (Fig. S7).
 - Detailed comparison of the performance of G@W-Co-P in 0.5 M H₂SO₄ with those of representative non-noble-metal HER catalysts (Table S1).
 - Enlarged LSV curve of WP₂/G in 0.5 M H₂SO₄ (Fig. S8).
 - Detailed comparison of the performance of G@W-Co-P in 1 M KOH with those of representative non-noble-metal HER catalysts (Table S2).
 - XRD pattern of G@Co-W-P after the durability tests (Fig. S9).

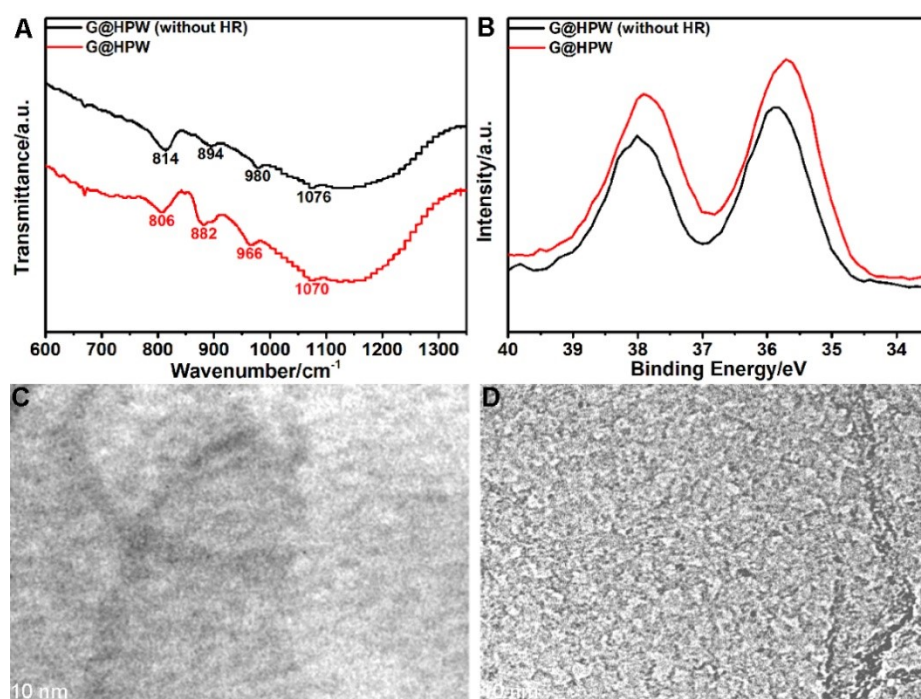


Fig. S1 FTIR spectra (A) and high-resolution W 4f XPS spectra (B) of G@HPW and G@HPW (without HR) and TEM images of G@HPW (without HR) (C) and G@HPW (D).

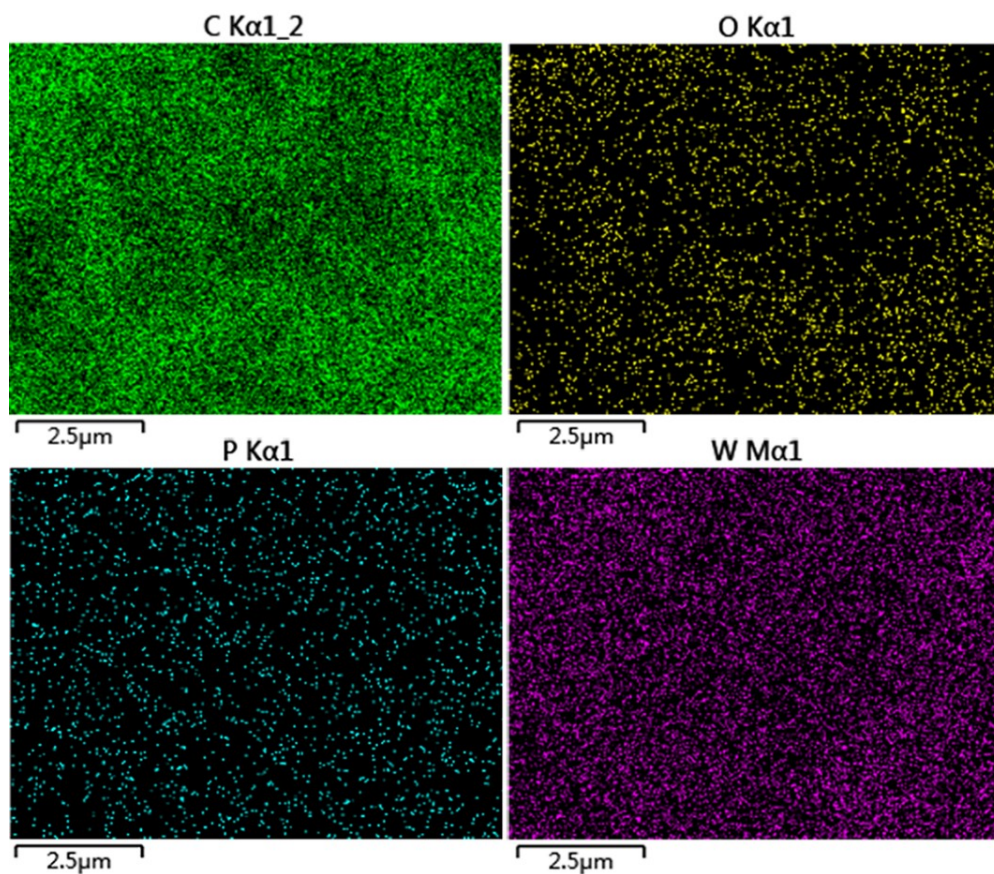


Fig. S2 EDX mapping of G@HPW after the hydrothermal reaction.

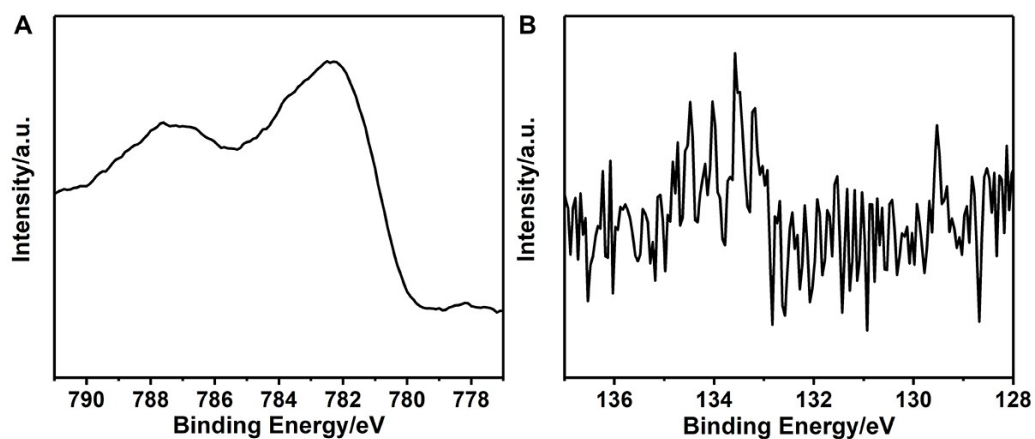


Fig. S3 High-resolution Co 2p_{3/2} (A) and P 2p (B) XPS spectra of the sample before phosphidation.

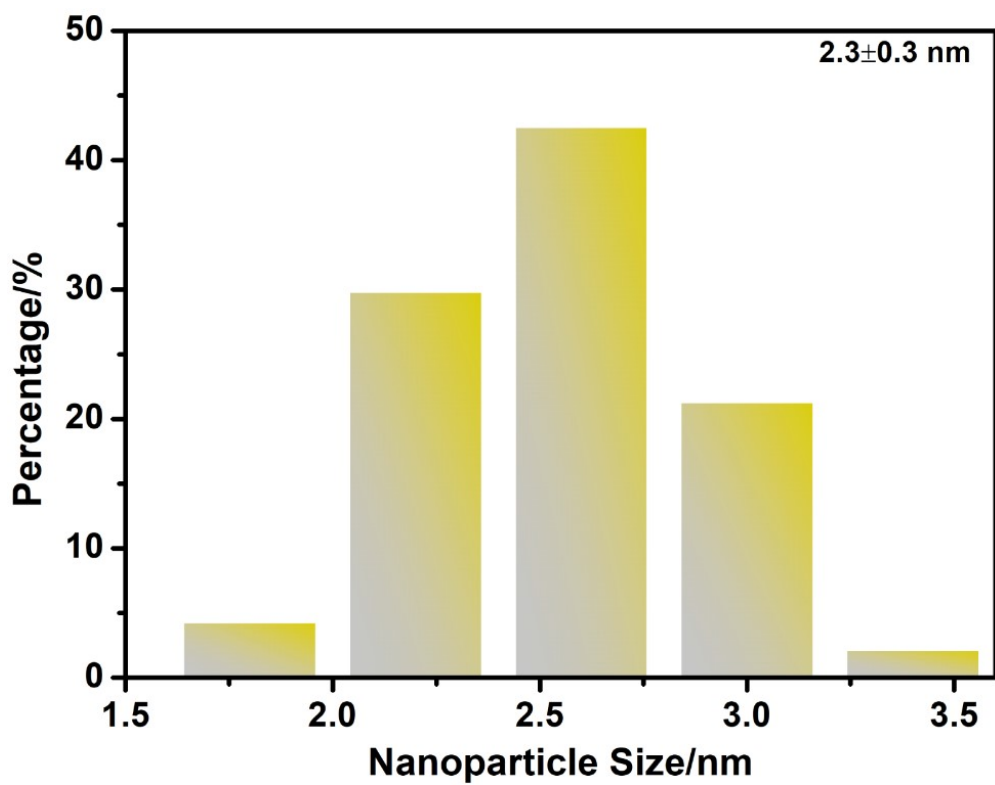


Fig. S4 Size distribution of Co-W-P NPs in G@Co-W-P.

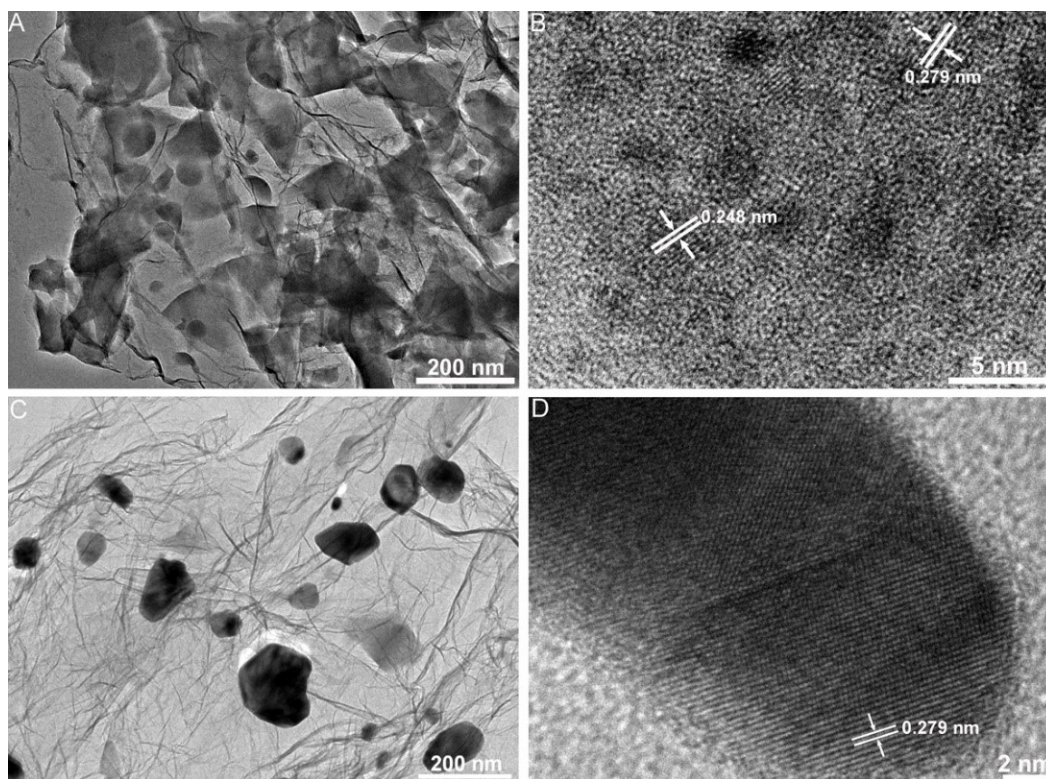


Fig. S5 TEM (A and C) and HRTEM (B and D) images of Co-W-P/G (A and B) and CoP/G (C and D).

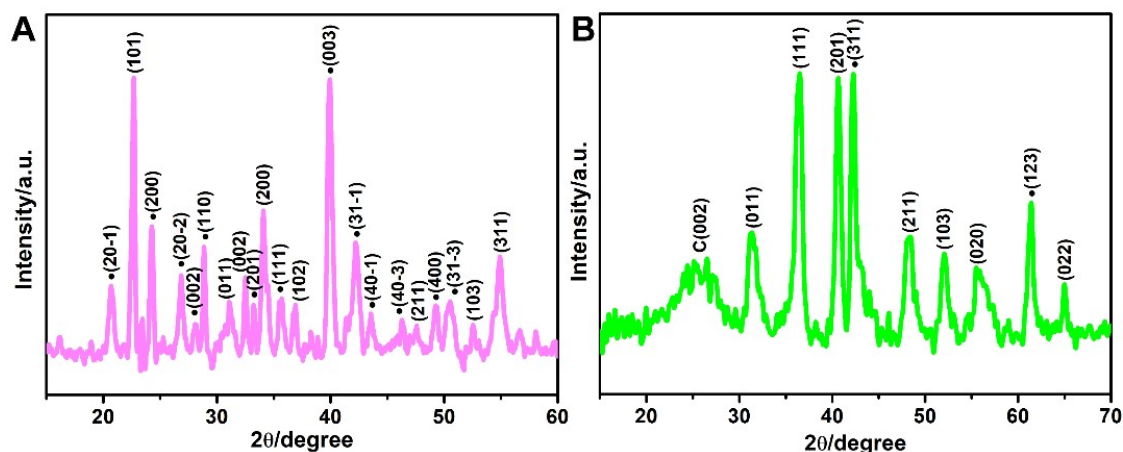


Fig. S6 XRD patterns of one-step hydrothermal reaction (A) and the sample with no HPW added in the second-step hydrothermal reaction (B). • above the XRD pattern in (A) represents characteristic peaks of WP_2 (JCPDS card no. 35-1467) and the other peaks are from CoP (JCPDS card no. 29-0497) (except the peak C (002), which is from graphene). There are some peaks from $Co_3(PO_4)_2$ (JCPDS card no. 77-0225) in (B) (labeled with •), which is formed due to the surface oxidation under high temperature.

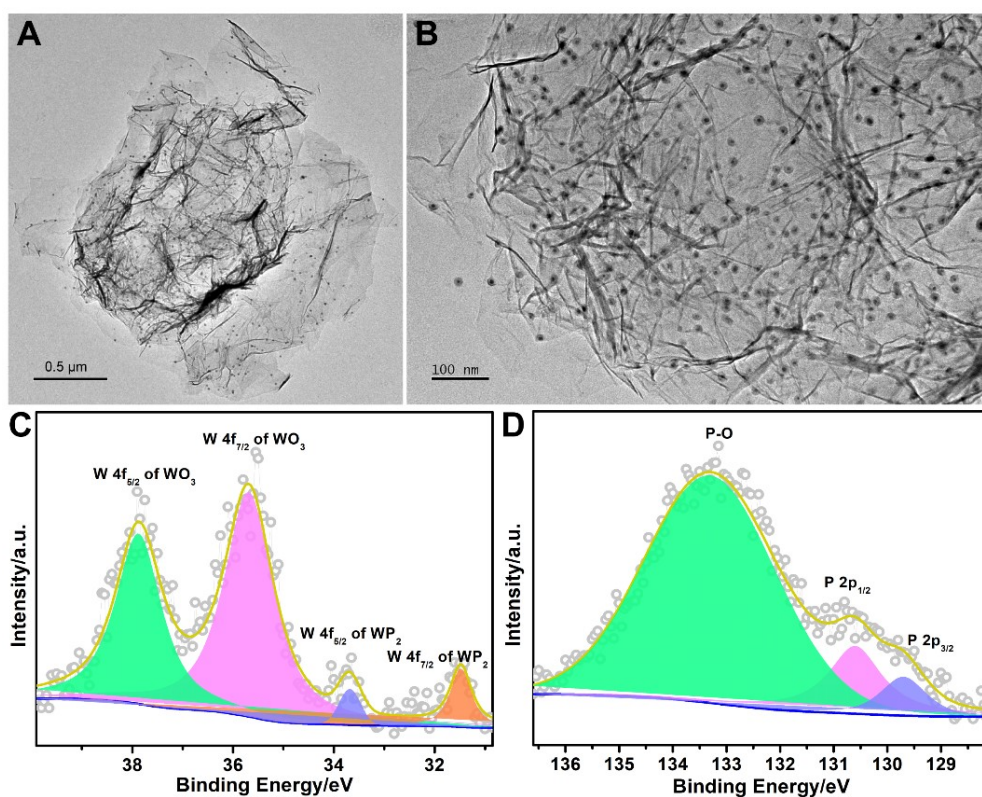


Fig. S7 TEM image and high-resolution W 4f (B) and P 2p (C) XPS spectra of WP_2/G .

Table S1 Detailed comparison of the performance of G@W-Co-P in 0.5 M H₂SO₄ with those of representative non-noble-metal HER catalysts.

Catalyst	Loading (mg cm ⁻²)	j_0 (mA/cm ²)	Onset overpotential (mV)	η_{10} (mV)	Tafel slope (mV dec ⁻¹)	Ref.
FeP ₂ /C NPs	0.425	0.00175		220	66	S1
CoP NFs	0.265			122	54.8	S2
W-W ₂ C/CNT-6	0.28			155	56	S3
Fe doped NiS ₂				198	42	S4
MoPS/NC	0.56		23	92	56	S5
CoP/CNT	0.285	0.13	40	122	54	S6
N ₂ P-MoxC NF	0.265			107	65.1	S7
CoP/NCNHP				140	53	S8
CoP@NG	0.283			158	63.8	S9
Mo ₂ C-N-CNFs	0.255	0.0473		167	70	S10
Ni ₂ P/CNT	0.2	0.0537		124	53	S11
WP ₂ nanorods	0.285	0.013	101	148	52	S12
FeP GS	0.28	0.12	30	123	50	S13
Ni ₂ P@NPCNFs	0.56			63.2	56.7	S14
MoP@PC	0.41			153	66	S15
CoP/Co ₂ P@NC-2				126	79	S16
FeP NPs@NPC	1.4	0.126		130	67	S17
CoP-OMC	0.285	0.161	77.74	112.18	56.67	S18
α -MoC _{1-x} /NC				142	74	S19
CoP-N-C	0.425	0.16	31	91	42	S20
Ni ₂ P hollow NPs	1	0.033		117	46	S21
G@W-Co-P	0.5	0.128	13	91.5	40.7	This work

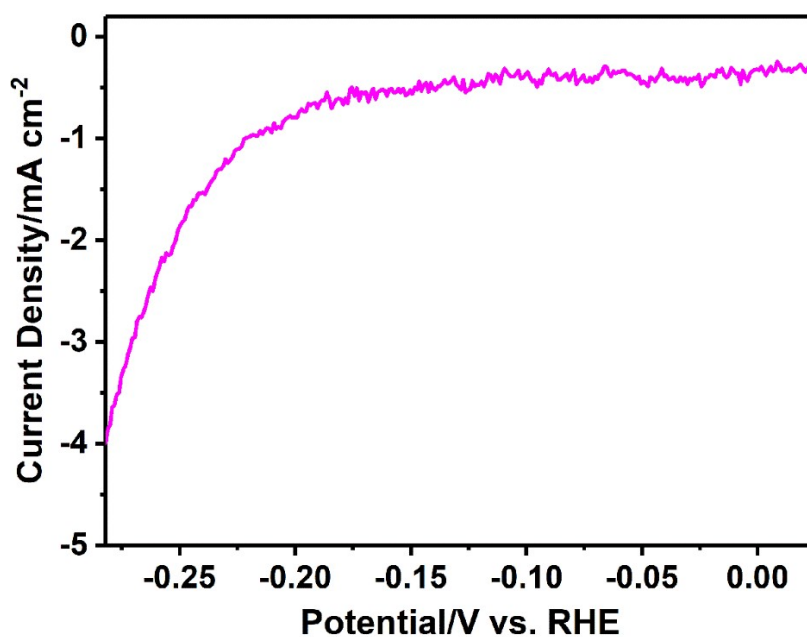


Fig. S8 Enlarged LSV curve of WP₂/G in 0.5 M H₂SO₄.

Table S2 Detailed comparison of the performance of G@W-Co-P in 1 M KOH with those of representative non-noble-metal HER catalysts.

Catalyst	Loading (mg cm ⁻²)	Onset overpotential (mV)	η_{10} (mV)	Tafel slope (mV dec ⁻¹)	Ref.
CoP NFs	0.265		136	56.2	S1
CoTeNR/NF	1.3		202	115	S22
Co ₃ O ₄ /MoS ₂	2		205	98	S23
CoP/CC	0.92	45	106	93	S24
NiCoP/NPC HFSs	0.2	~80	128	70	S25
CoP/rGO-400	0.285		150	38	S26
Cu _{0.3} Co _{0.27} P/NC	0.4		220	112	S27
CoP/NCNHP			115	66	S8
CoP@NG	0.283		182	59.6	S9
NiCoP/CNF900		100	130	83	S28

Co-PNCNFs	0.56		249	92	S29
Ni _{0.69} Co _{0.35} P	3.5		167	47	S30
Ni ₁₂ P ₅ /NF	3		170	106	S31
FeP NAs/CC	1.5	86	218	146	S32
W-W ₂ C/CNT-6	0.28		147	51	S3
Co ₄ Ni ₁ P NTs	0.19		129	52	S33
CoFe/NF			110	35	S34
NiCoP@NF			155	115	S35
WP ₂ nanorods		149	225	84	S12
EG/H-Co _{0.85} Se ₁ P	2.1		150	83	S36
NiCoP/rGO	0.19		209	124.1	S37
G@W-Co-P	0.5	16.1	102.3	61.2	This work

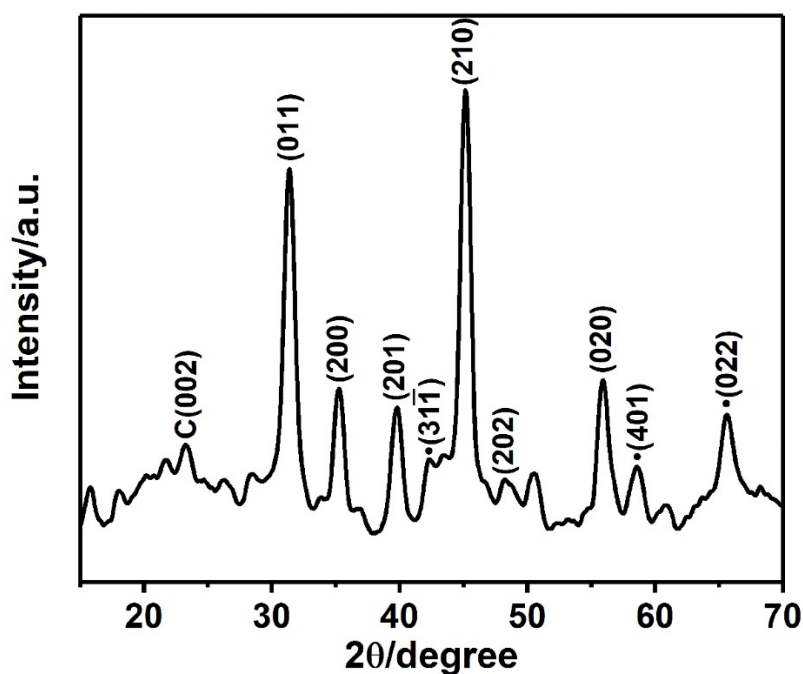


Fig. S9 XRD pattern of G@Co-W-P after the durability tests.

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