

1 **Supporting Information for**

2 **Na⁺/Vacancy Disordered Manganese-Based Oxide Cathode with**
3 **Ultralow Strain Enabled by Tuning Charge Distribution**

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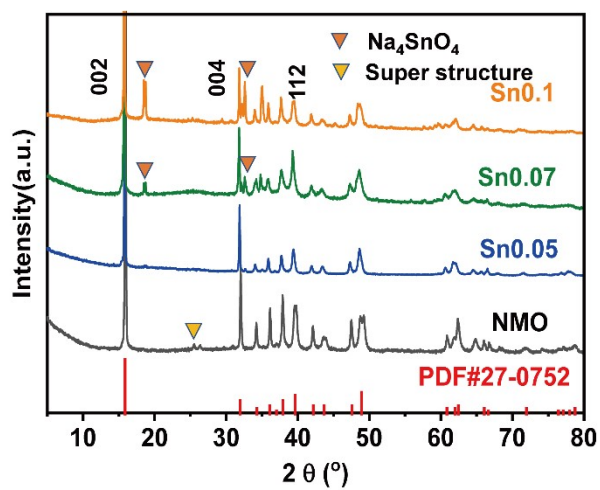
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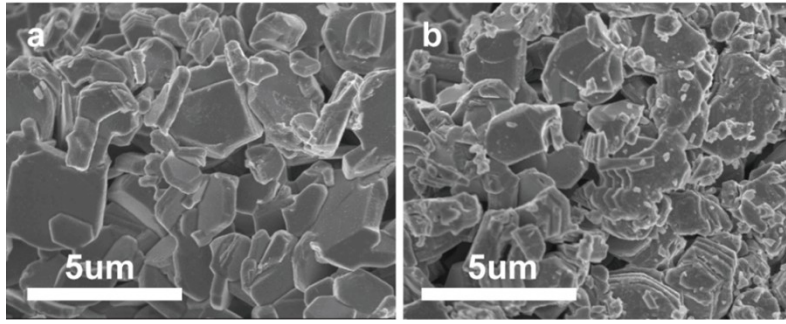
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2 **Fig. S1.** XRD patterns of $\text{Na}_{0.67}\text{Sn}_x\text{Mn}_{1-x}\text{O}_2$ ($x=0, 0.05, 0.07, 0.10$).

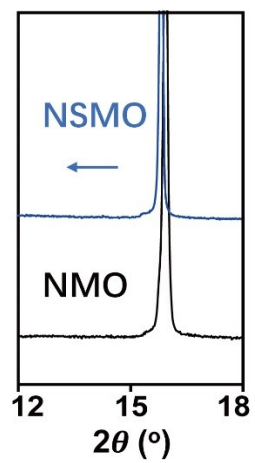
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2 **Fig. S2.** SEM images of (a) NMO and (b) NSMO.

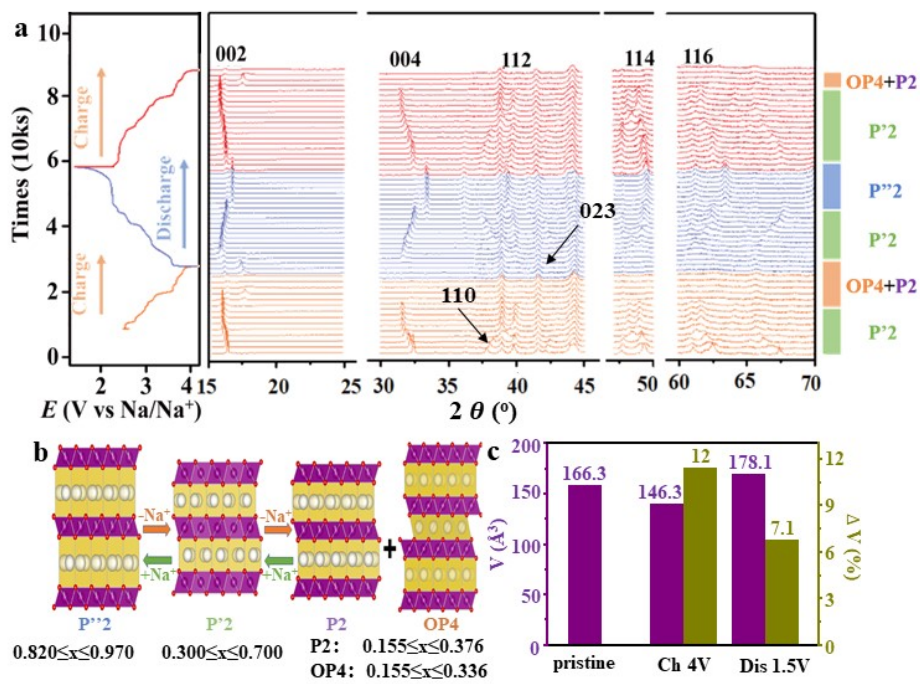
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2 **Fig. S3.** Enlarged XRD patterns of NMO and NSMO from 12° to 18°.

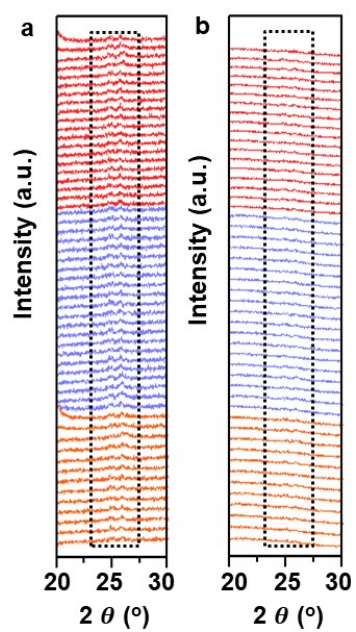
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2 **Fig. S4.** Structural evolution of NMO electrode. (a) *In-situ* XRD patterns. (b) Illustration of phase transition
 3 during cycles. (c) Volume variations upon Na⁺ extraction and insertion.

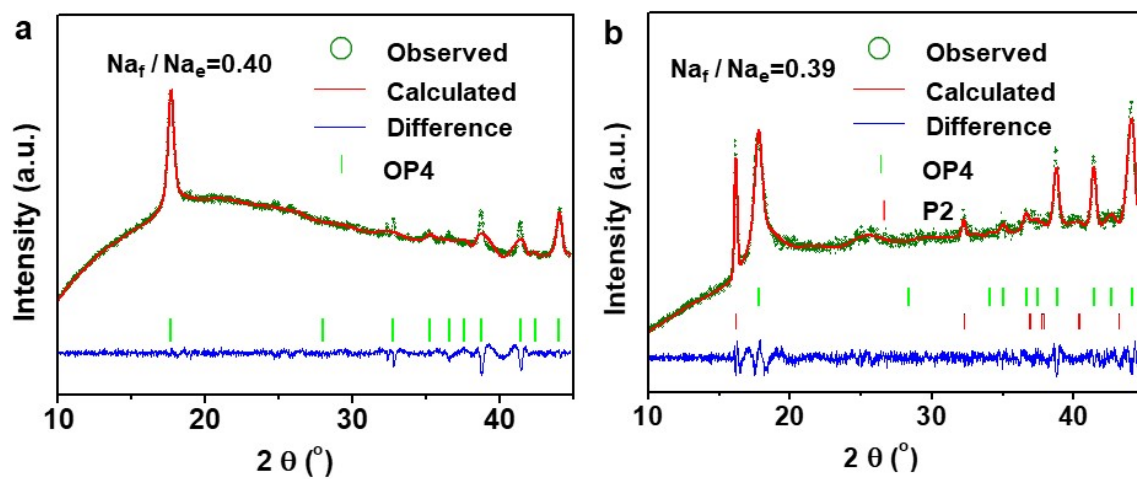
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2 **Fig. S5.** Enlarged *in-situ* XRD patterns of (a) NMO and (b) NSMO between 20° and 30°.

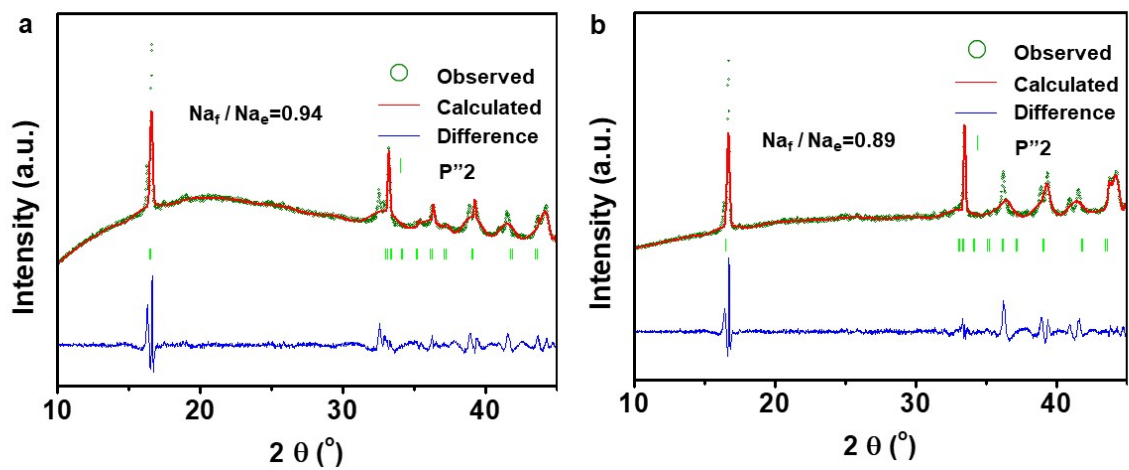
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2 **Fig. S6.** Rietveld refinement profiles of the powder XRD data for the charged (a) NSMO and (b) NMO
 3 electrodes.

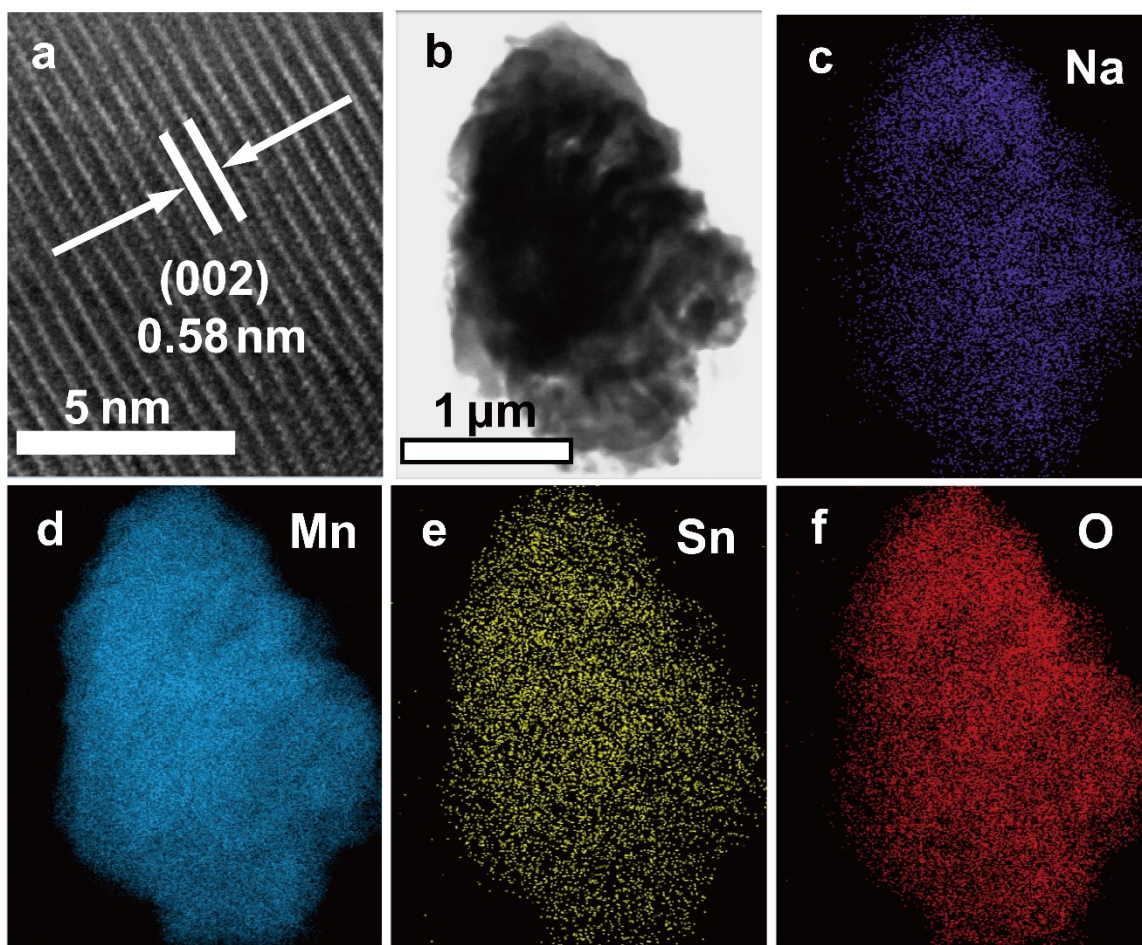
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2 Fig. S7. Rietveld refinement of powder XRD data for (a) NSMO and (b) NMO electrodes cycled at 1.5 V.

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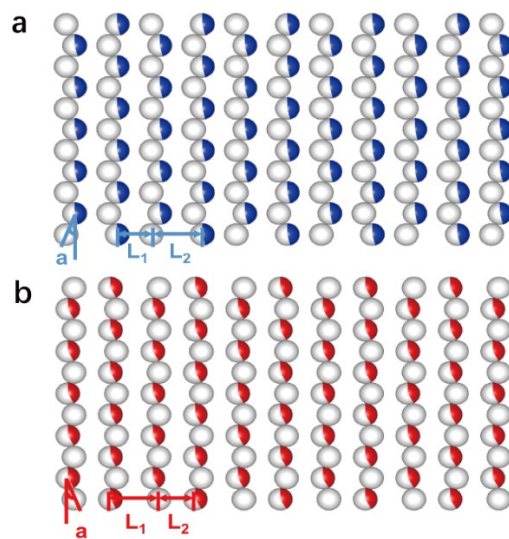


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2 **Fig. S8.** (a) HRTEM image of NSMO after 10 cycles and (b-f) the corresponding EDS elemental mappings.

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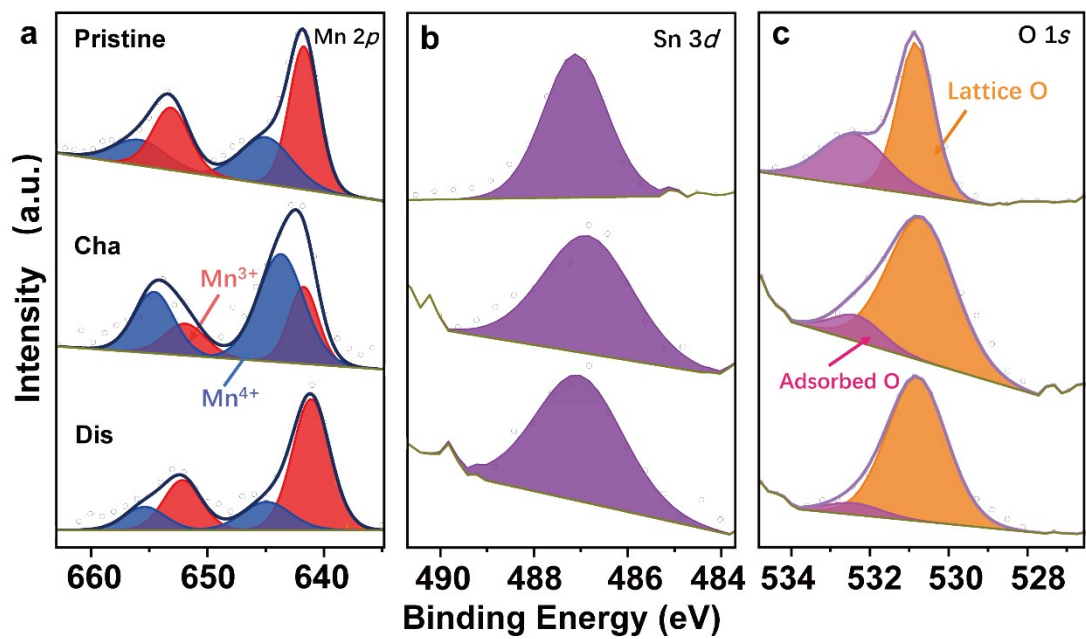
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2 **Fig. S9.** Top view of the Na layers for (a) NMO and (b) NSMO based on XRD refined data.

3

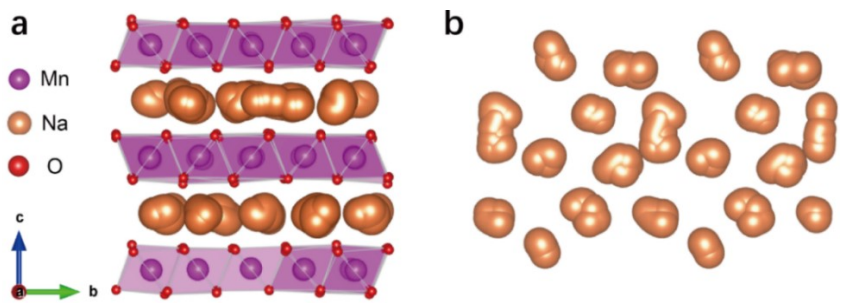


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2 **Fig. S10.** Charge compensation mechanism of NSMO characterized by ex-situ XPS. (a) Mn 2p spectra. (b)

3 Sn 3d spectra. (c) O 1s spectra.

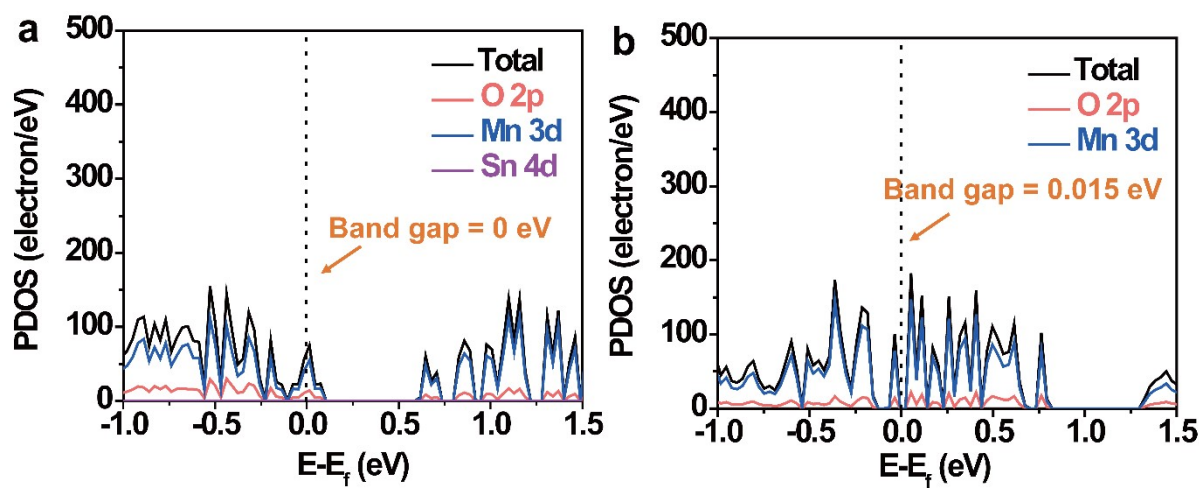
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2 **Fig. S11.** (a) Trajectories of Na⁺ in NMO simulated at 900 K. (b) Top view of Na layer.

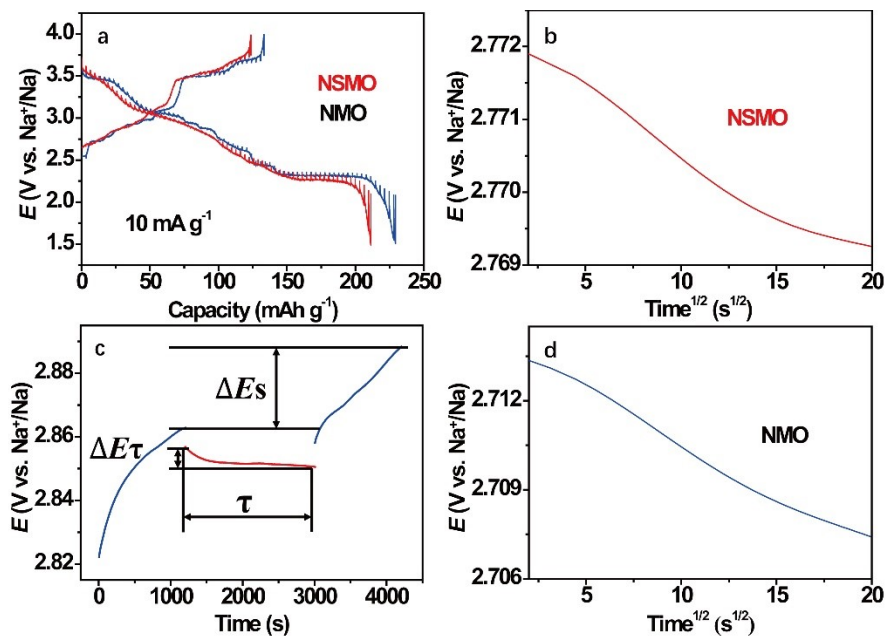
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2 Fig. S12. The calculated DOS for NSMO (a) and NMO (b).

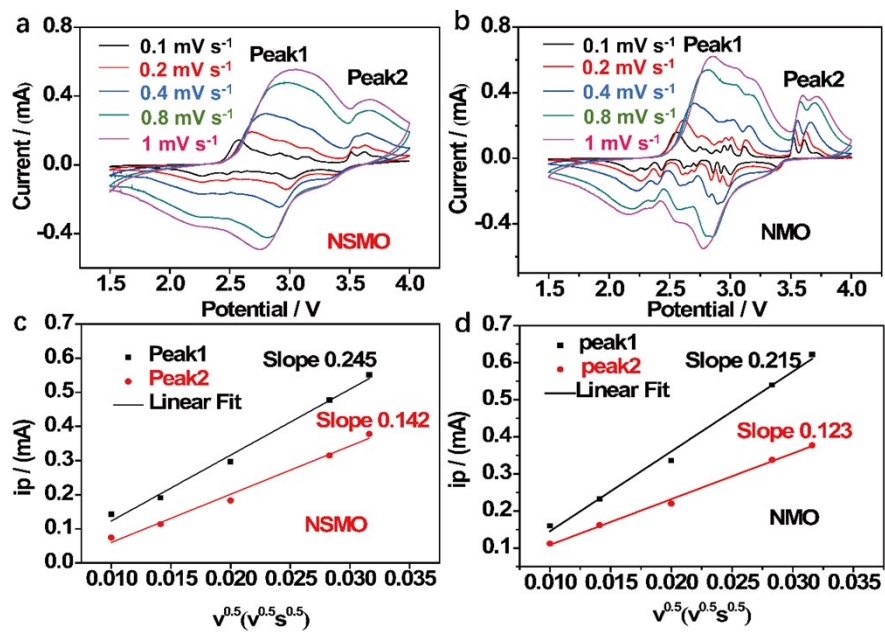
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2 **Fig. S13.** (a) GITT curves of NMO and NSMO. The linear relationship between voltage (E) and square root
 3 of resting time ($\text{Time}^{1/2}$) for (b) NSMO and (d) NMO. (c) Current step diagram of the first desodiation of
 4 NMO at 2.535 V.

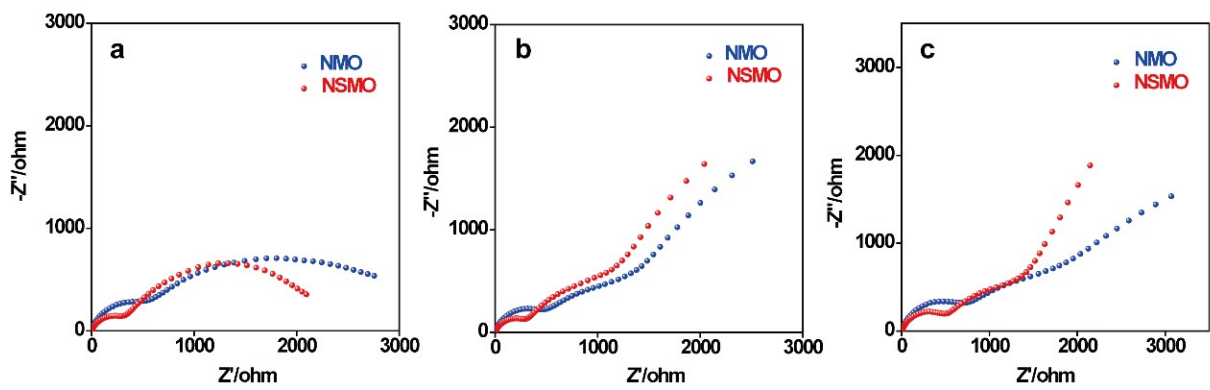
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2 **Fig. S14.** CV curves of (a) NSMO and (b) NMO electrodes at different scan rates. Fitted plots of peak current
 3 (i_p) versus scan rate ($v^{0.5}$) square root for (c) NSMO and (d) NMO electrodes.

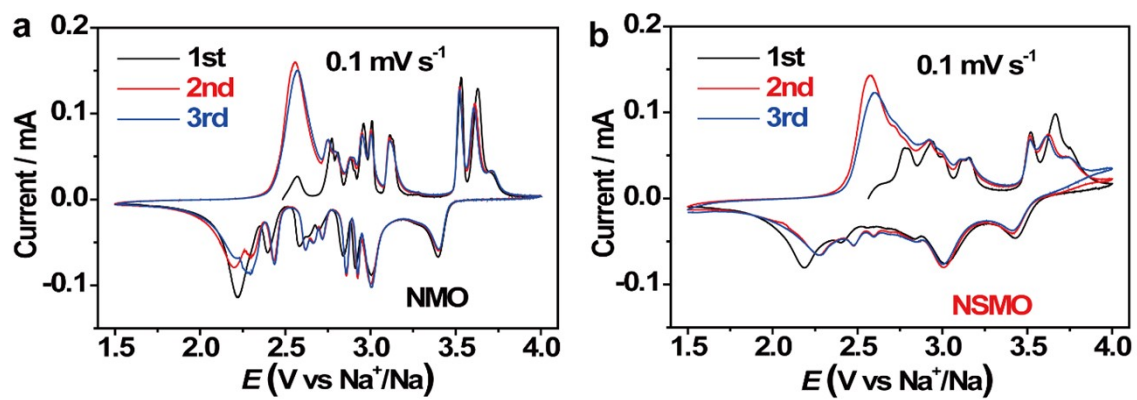
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2 **Fig. S15.** EIS plots of NSMO and NMO (a) at open-circuit voltage and (b) after 1 cycle and (c) ten cycles.

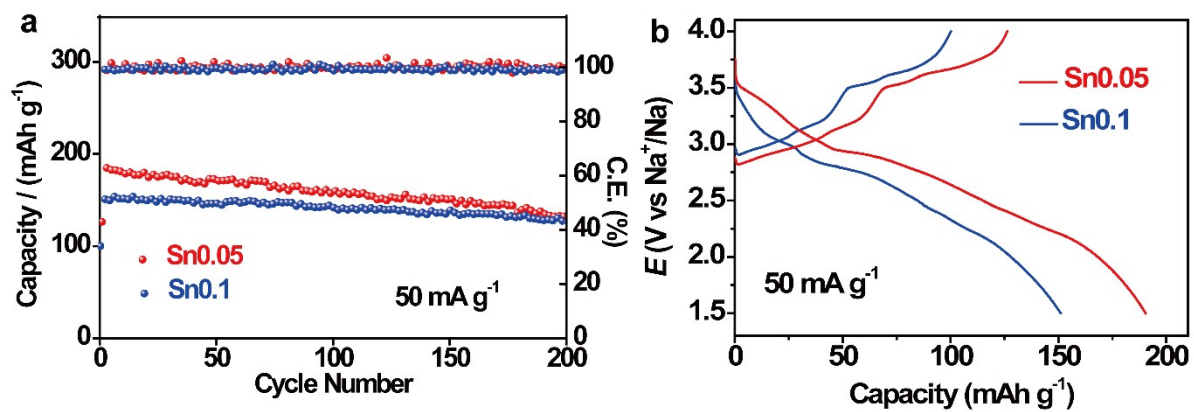
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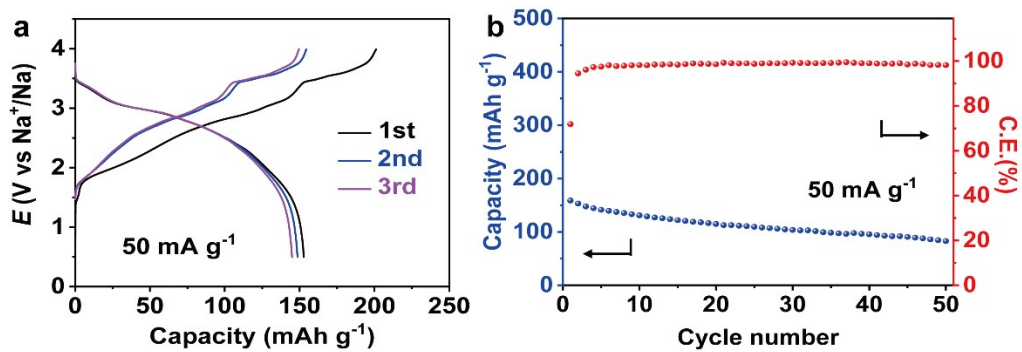
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2 Fig. S16. CV curves of (a) NMO and (b) NSMO electrodes measured at 0.1 mV s^{-1} within 1.5-4.0 V.

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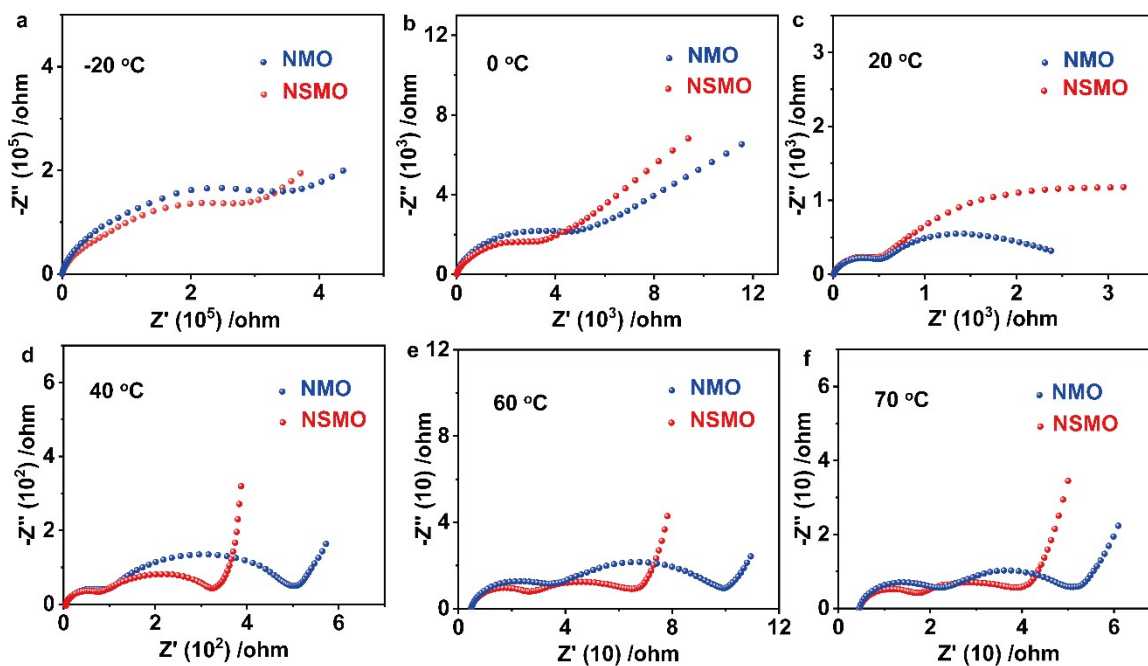
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 2 Fig. S17. (a) Cycle performance of NSMO and Na_{0.67}Sn_{0.10}Mn_{0.90}O₂ and (b) their selected charge/discharge
 3 curves.
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2 **Fig. S18.** (a) Charge/discharge profiles and (b) cycle performance of NSMO//HC full cells.

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2 **Fig. S19.** EIS plots of NSMO and NMO at (a) -20 °C, (b) 0 °C, (c) 20 °C, (d) 40 °C, (e) 60 °C and (f) 70 °C
 3 under open-circuit voltage.

4

1 **Table S1.** Lattice parameters of NSMO based on Rietveld refinement.

Atom	Site	Occupancy	x	y	z
Na_f	4c	0.25(9)	0	0.113	0.250
Na_e	4c	0.41(1)	0	0.673	0.250
Mn	4a	0.95(5)	0	0	0
Sn	4a	0.45(5)	0	0	0
O	8f	1	0	0.675	0.890

Space group: *Cmcm*, a = 2.830 Å, b = 5.231 Å, c = 11.420 Å, $\alpha = \beta = \gamma = 90^\circ$, V = 169.059 Å³,
Rwp = 4.3%

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1 **Table S2.** Lattice parameters of NMO based on Rietveld refinement.

Atom	Site	Occupancy	x	y	z
Na_f	4c	0.21(9)	0	0.236	0.250
Na_e	4c	0.45(1)	0	0.696	0.250
Mn	4a	1	0	0	0
O	8f	1	0	0.669	0.899

Space group: *Cmcm*, a = 2.831 Å, b = 5.240 Å, c = 11.209 Å, $\alpha = \beta = \gamma = 90^\circ$, V = 166.279 Å³, Rwp = 4.3%

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1 **Table S3.** Mn-O bond length, TMO₂ layer thicknesses, Na interlayer spacings, and distances between
 2 adjacent Mn columns for NSMO and NMO.

	$d_{(\text{Mn-O})}$	$d_{(\text{MnO}_2)}$	$d_{(\text{O-Na-O})}$	$d_{(\text{interslab})}$
NSMO (Å)	2.01/1.95	2.20	3.42	5.71
NMO (Å)	2.07/2.01	2.26	3.34	5.60
Change ratio	2.99%/3.08%	2.72%	2.34%	1.92%

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1 **Table S4.** Refined structural parameters of OP4 phase in the charged NSMO electrode.

Atom	Site	Occupancy	x	y	z
Na_r	2d	0.04(1)	0.667	0.333	0.250
Na_e	2c	0.10(2)	0.333	0.667	0.250
Na₁	2a	0.05(1)	0	0	0.5
Mn	4f	0.95	0.667	0.333	0.360
Sn	4f	0.05	0.667	0.333	0.360
O1	4f	0.5	0.333	0.667	0.415
O2	4e	0.5	0	0	0.296

Space group: *P63/mmc*, $a = 2.952 \text{ \AA}$, $b = 2.952 \text{ \AA}$, $c = 22.188 \text{ \AA}$, $\alpha = \beta = 90^\circ$, $\gamma = 120^\circ$, $V = 167.449 \text{ \AA}^3$, $\Delta V\% = 0.95\%$, $R_{wp} = 1.9\%$

2

1 **Table S5.** Refined structural parameters of P''2 phase in the discharged NSMO and NMO electrodes.

Sample	Atom	Site	Occupancy	x	y	z
NSMO	Na _f	4c	0.45(2)	0	1.712	0.250
	Na _e	4c	0.47(6)	0	-4.154	0.250
	Mn	4a	0.94(3)	0	0	0
	Sn	4a	0.05(7)	0	0	0
	O	8f	1	0	1.473	-2.387
NMO	Na _f	4c	0.45(9)	0	0.645	0.250
	Na _e	4c	0.51(1)	0	0.646	0.250
	Mn	4a	1	0	0	0
	O	8f	1	0	0.199	-0.047

Space group: NSMO: *Cmcm*, a = 2.89 Å, b = 5.42 Å, c = 10.74 Å, $\alpha = \beta = \gamma = 90^\circ$, V = 168.23 Å³, $\Delta V\% = 0.49\%$, Rwp = 4.5%; **NMO:** *Cmcm*, a = 2.89 Å, b = 5.66 Å, c = 10.89 Å, $\alpha = \beta = \gamma = 90^\circ$, V = 178.13 Å³, $\Delta V\% = 7.1\%$, Rwp = 8%

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1 **Table S6.** Refined structural parameters of OP4 and P2 phases in the charged NMO electrode.

Phase	Atom	Site	Occupancy	x	y	z
OP4	Na_f	2d	0.02(7)	0.667	0.333	0.250
	Na_e	2c	0.06(8)	0.333	0.667	0.250
	Na₁	2a	0.03(5)	0	0	0.5
	Mn	4f	1	0.667	0.333	0.373
	O₁	4f	0.5	0.333	0.667	0.402
	O₂	4e	0.5	0	0	0.328
P2	Na_f	4f	0.00(5)	0.333	0.667	0.582
	Na_e	2b	0.00(5)	0	0	0.250
	Mn	2a	1	0	0	0
	O	4f	1	0.333	0.667	0.582

Space group: OP4: *P63/mmc*, a = 2.99 Å, b = 2.99 Å, c = 18.87 Å, $\alpha = \beta = 90^\circ$, $\gamma = 120^\circ$, V = 146.30 Å³, $\Delta V\% = 12\%$, Rwp = 3.6%, wt% = 83.8%; **P2:** *P63/mmc*, a = 2.82 Å, b = 2.82 Å, c = 11.24 Å, $\alpha = \beta = \gamma = 90^\circ$, V = 77.32 Å³, $\Delta V\% = 53.5\%$, Rwp = 3.6%, wt% = 16.2%

2

1 **Table S7.** Distances and angles between Na_f and Na_e in Na layers for P2-NMO, P'2-NMO and P'2-NSMO
2 based on Rietveld refinement.

Sample	a (°)	L1 (Å)	L2 (Å)	ΔL (Å)
P2-NMO	60.0	1.6	3.3	1.7
P'2-NMO	20.5	2.1	3.2	1.1
P'2-NSMO	14.7	2.3	3.0	0.7

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1 **Table S8.** Na⁺ diffusion coefficients of peak 1 and 2 obtained by CV.

Peak	NSMO (cm²s⁻¹)	NMO (cm²s⁻¹)
1	2.25×10 ⁻¹⁰	2.97×10 ⁻¹¹
2	2.26×10 ⁻¹¹	1.20×10 ⁻¹¹

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1 **Table S9.** Electrochemical performance comparison of some reported cathodes in SIBs.

Materials	Cyclability (capacity retention)	Rate Capability	Ref
NSMO	131.2 mAh g ⁻¹ (71.63%) after 200 cycles at 50 mA g ⁻¹	218.4 mAh g ⁻¹ at 10 mA g ⁻¹ 143.2 mAh g ⁻¹ at 500 mA g ⁻¹	This work
poly(diphenylaminesulfonic acid sodium) (PDS)	72 mAh g ⁻¹ (70%) after 100 cycles at 50 mA g ⁻¹	92 mAh g ⁻¹ at 100 mA g ⁻¹ 75 mAh g ⁻¹ at 200 mA g ⁻¹	50
NaFePO ₄	100 mAh g ⁻¹ (90%) after 240 cycles at 15.4 mA g ⁻¹	120 mAh g ⁻¹ at 7.7 mA g ⁻¹ 46 mAh g ⁻¹ at 308 mA g ⁻¹	51
Na _{0.7} Mn _{0.6} Ni _{0.3} Co _{0.1} O ₂	69.6 mAh g ⁻¹ (58%) after 20 cycles at 9 mA g ⁻¹	120 mAh g ⁻¹ at 9 mA g ⁻¹ 80 mAh g ⁻¹ at 185 mA g ⁻¹	52
Na _{0.66} (Mn _{0.54} Co _{0.13} Ni _{0.13})O ₂	60.5 mA g ⁻¹ (50%) after 100 cycles at 160 mA g ⁻¹	121 mAh g ⁻¹ at 160 mA g ⁻¹ 94 mAh g ⁻¹ at 800 mA g ⁻¹	53
Na _{0.44} MnO ₂	75.0 mA g ⁻¹ (63%) after 60 cycles at 24.2 mA g ⁻¹	102 mAh g ⁻¹ at 60 mA g ⁻¹ 98 mAh g ⁻¹ at 121 mA g ⁻¹	54

2