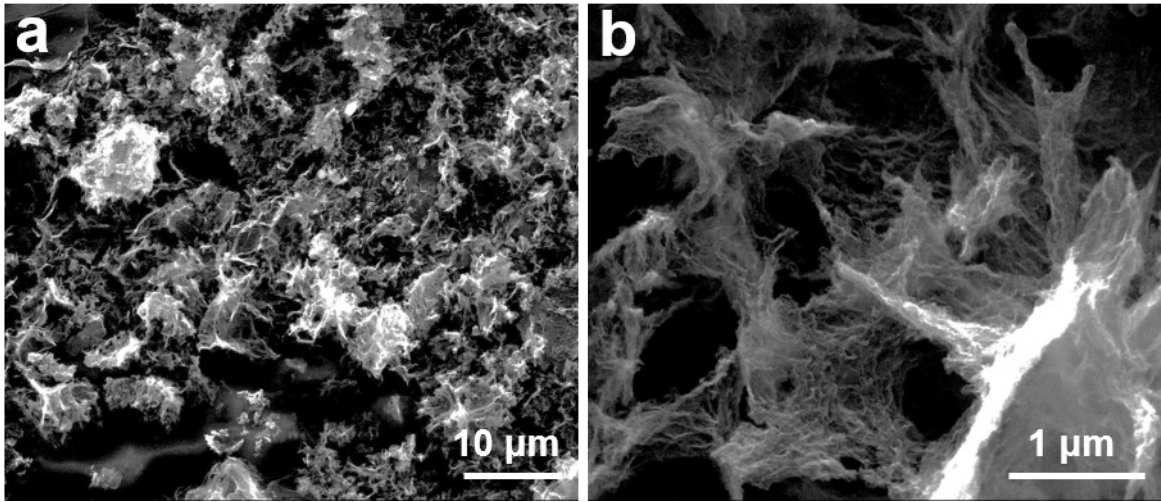


## **NbN Nanodots Decorated N-doped Graphene as Multifunctional Interlayer for High-Performance Lithium-Sulfur Batteries**

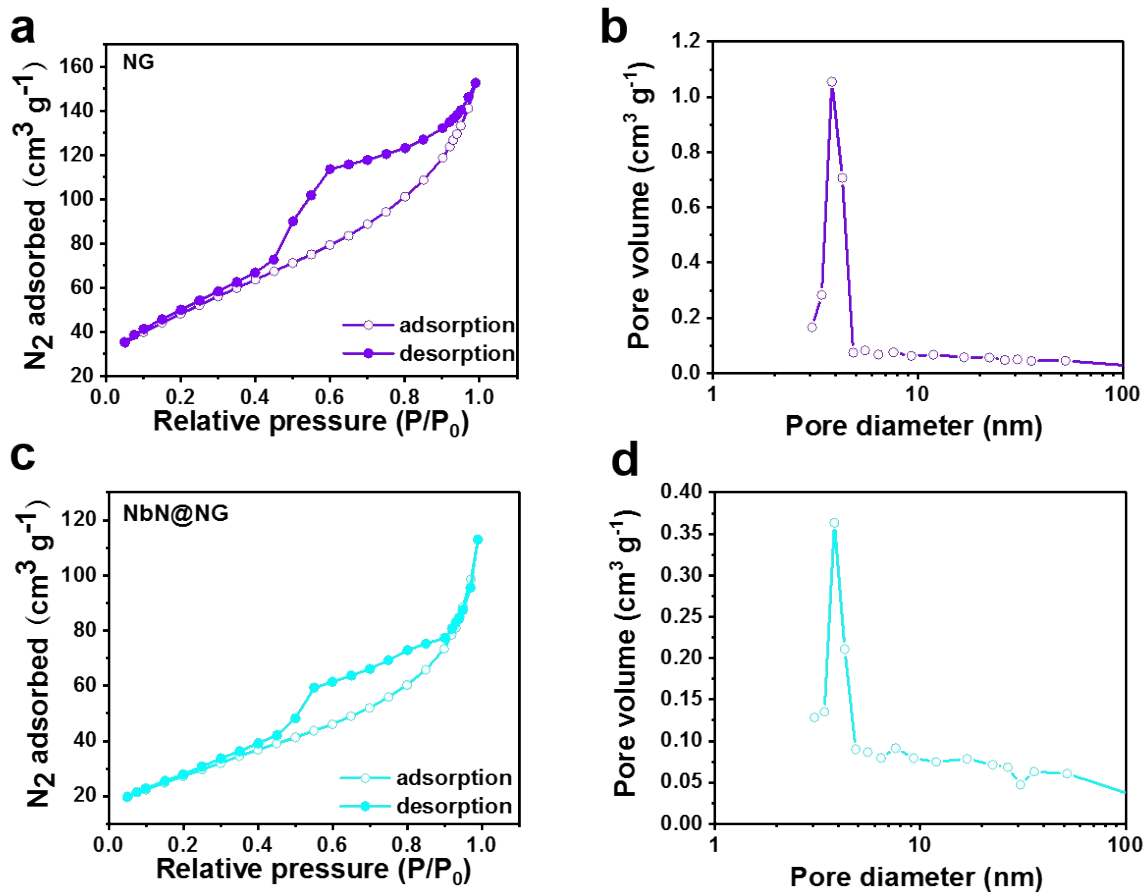
*Fei Ma<sup>a#</sup>, Xiaojuan Zhang<sup>a#</sup>, katam Srinivas<sup>a</sup>, Dawei Liu<sup>a</sup>, Ziheng Zhang<sup>a</sup>, Xin Chen<sup>a</sup>, Wanli Zhang<sup>a</sup>, Qi Wu<sup>\*b</sup>, Yuanfu Chen<sup>\*a,b</sup>*

<sup>a</sup> School of Electronic Science and Engineering, and State Key Laboratory of Electronic Thin Films and Integrated Devices, University of Electronic Science and Technology of China, Chengdu 610054, PR China.

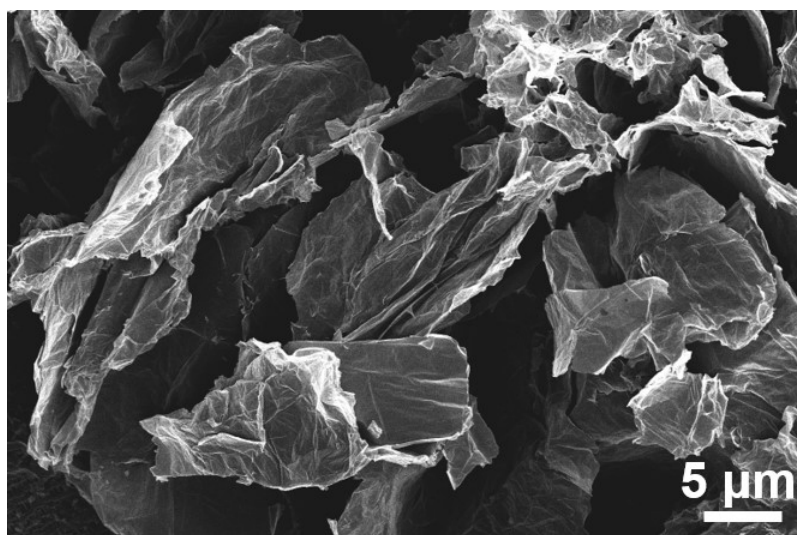
<sup>b</sup> College of Science and Institute of Oxygen Supply, Center of Tibetan Studies (Everest Research Institute), Tibet University, Lhasa, 850000, PR China.



**Fig. S1** FESEM image of the NG at different magnifications.

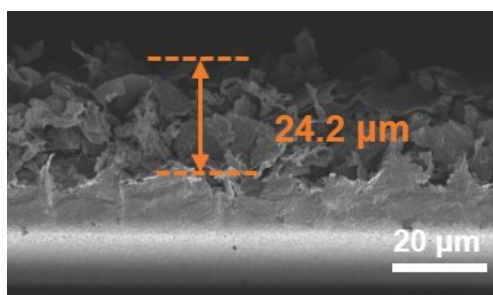


**Fig. S2**  $N_2$  adsorption and desorption isotherms of (a) NG and (b) NbN@NG. Calculated pore size distribution (c) and (d) of NG and (c) NbN@NG.



**Fig. S3** SEM images of the top surface for NG modified separators.





**Fig. S4** Cross-section of NG modified separators.

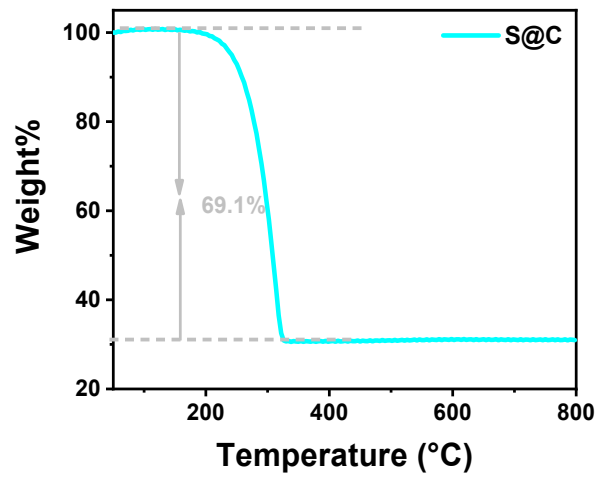


Fig. S5 TGA patterns of S@C.

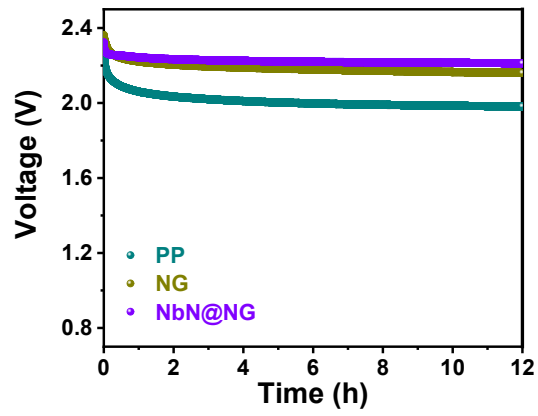
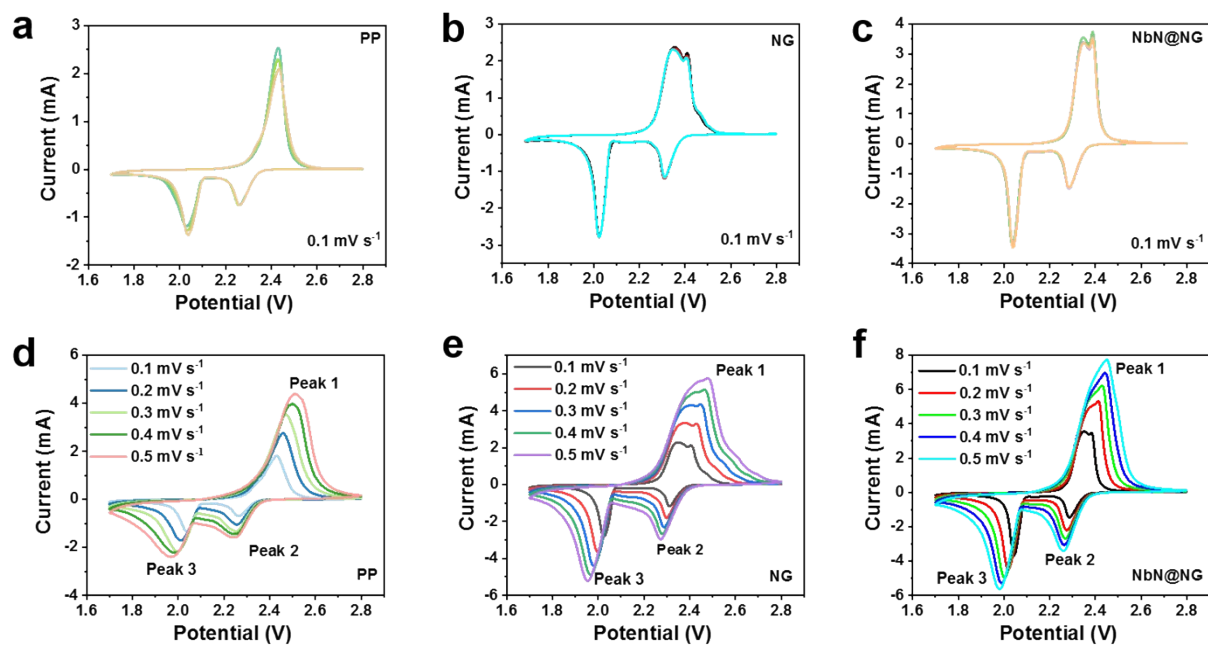
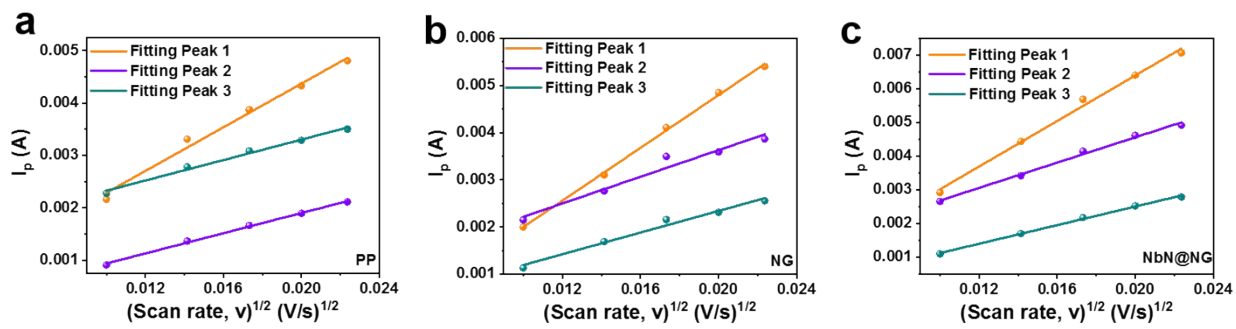


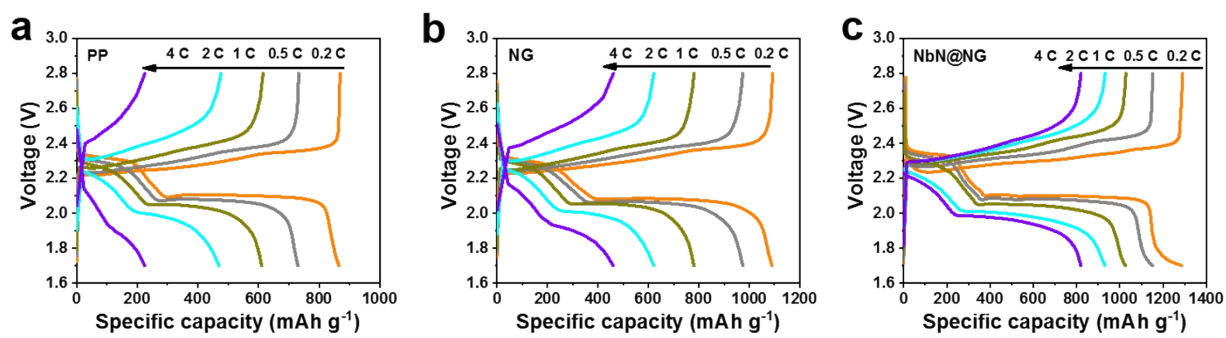
Fig. S6 Self-discharge behavior of the batteries.



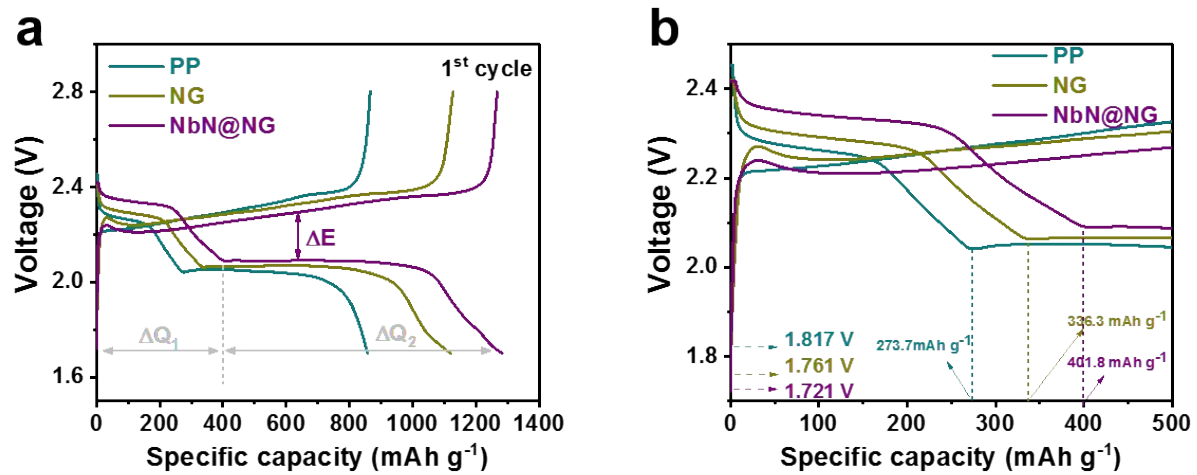
**Fig. S7** CV curves of the batteries with (a) PP, (b) NG, and (c) NbN@NG separators at 0.1 mV s<sup>-1</sup>. CV curves of the batteries with (d) PP, (e) NG, and (f) NbN@NG separators in the range of 0.1-0.5 mV s<sup>-1</sup>.



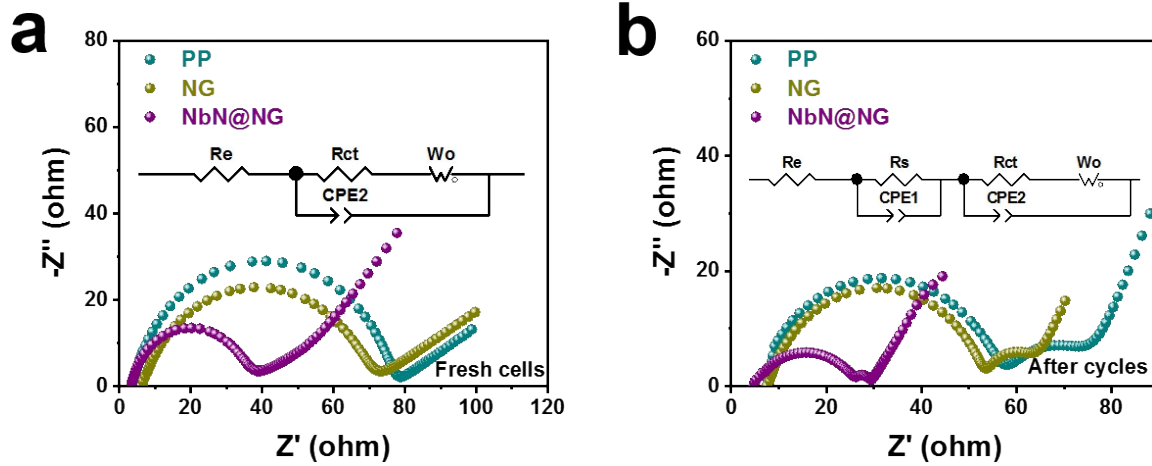
**Fig. S8** (a-c) The linear fitting curves of the peak current as a function of scan rate with different separators.



**Fig. S9** Galvanostatic discharging/charging curves of the batteries with (a) PP, (b) NG, and (c) NbN@NG separator at various rates.

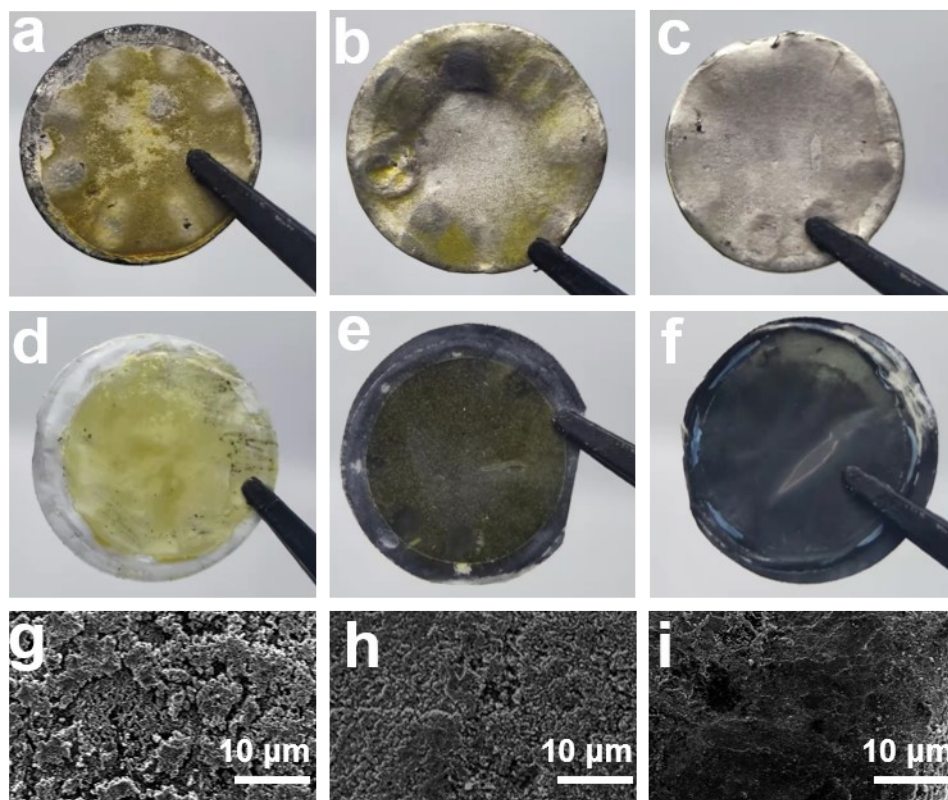


**Fig. S10** (a) Galvanostatic charge/discharge curves of different separators at 0.2 C. (b) The details of charge/discharge curves.

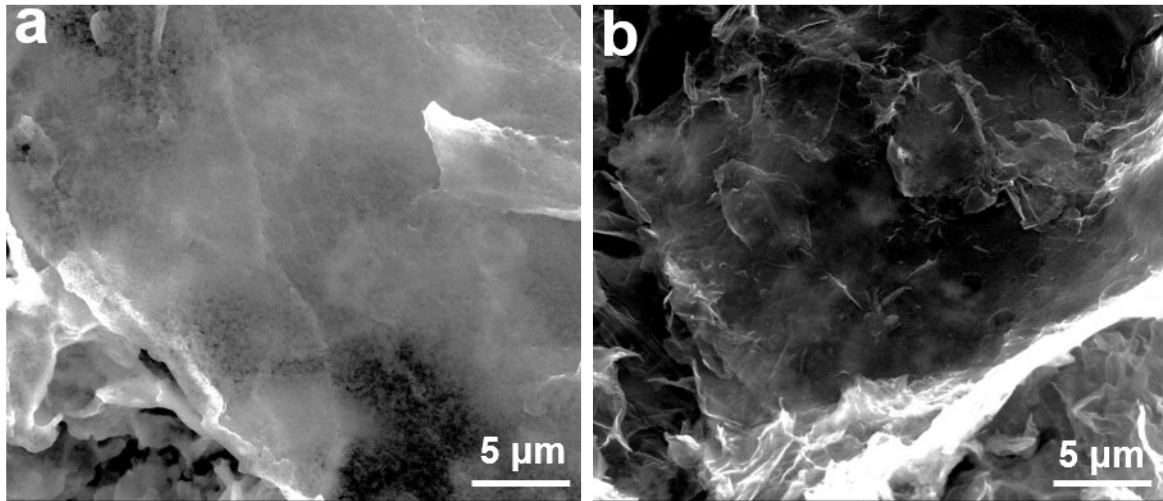


**Fig. S11** Nyquist plots of the batteries with different materials (a) before and (b) after 200 cycles at 0.2 C, insets: the corresponding equivalent circuits.

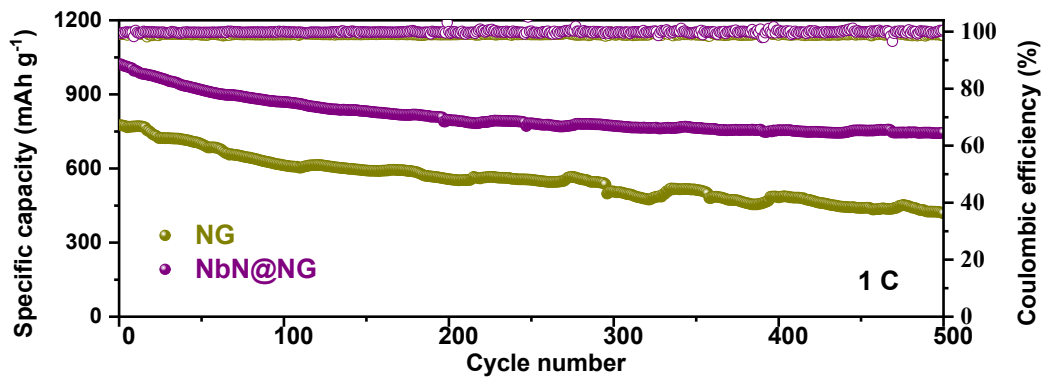




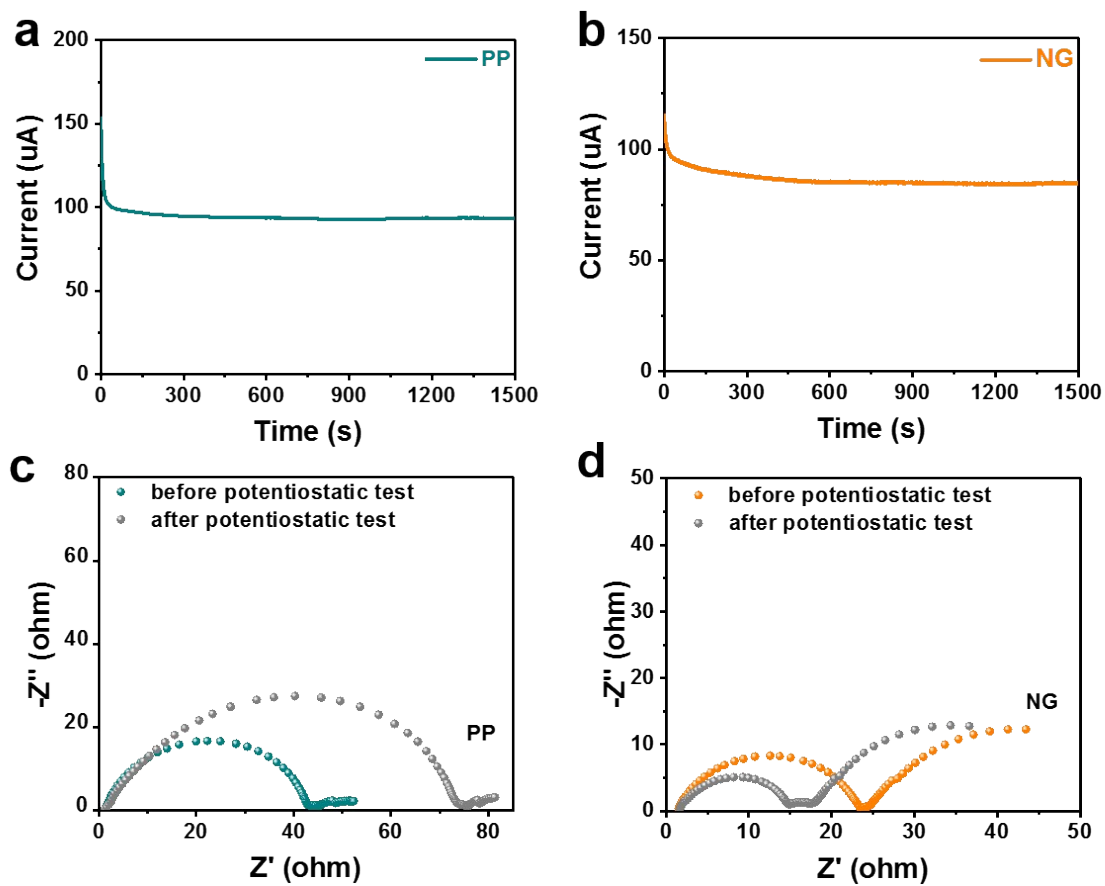
**Fig. S12** Digital images of Li-S batteries after 200 cycles at 0.2 C when paired with (a) PP, (b) NG, (c) NbN@NG separators and (d-f) corresponding Li-metal anodes. (g-i) SEM images of corresponding Li-metal anodes.



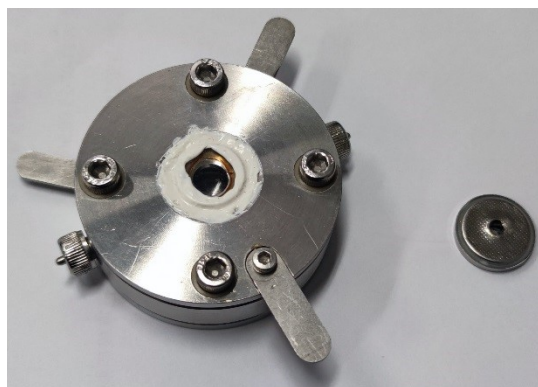
**Fig. S13** FESEM images of the (a) NG and (b) NbN@NG modified separators after 200 cycles at 0.2 C.



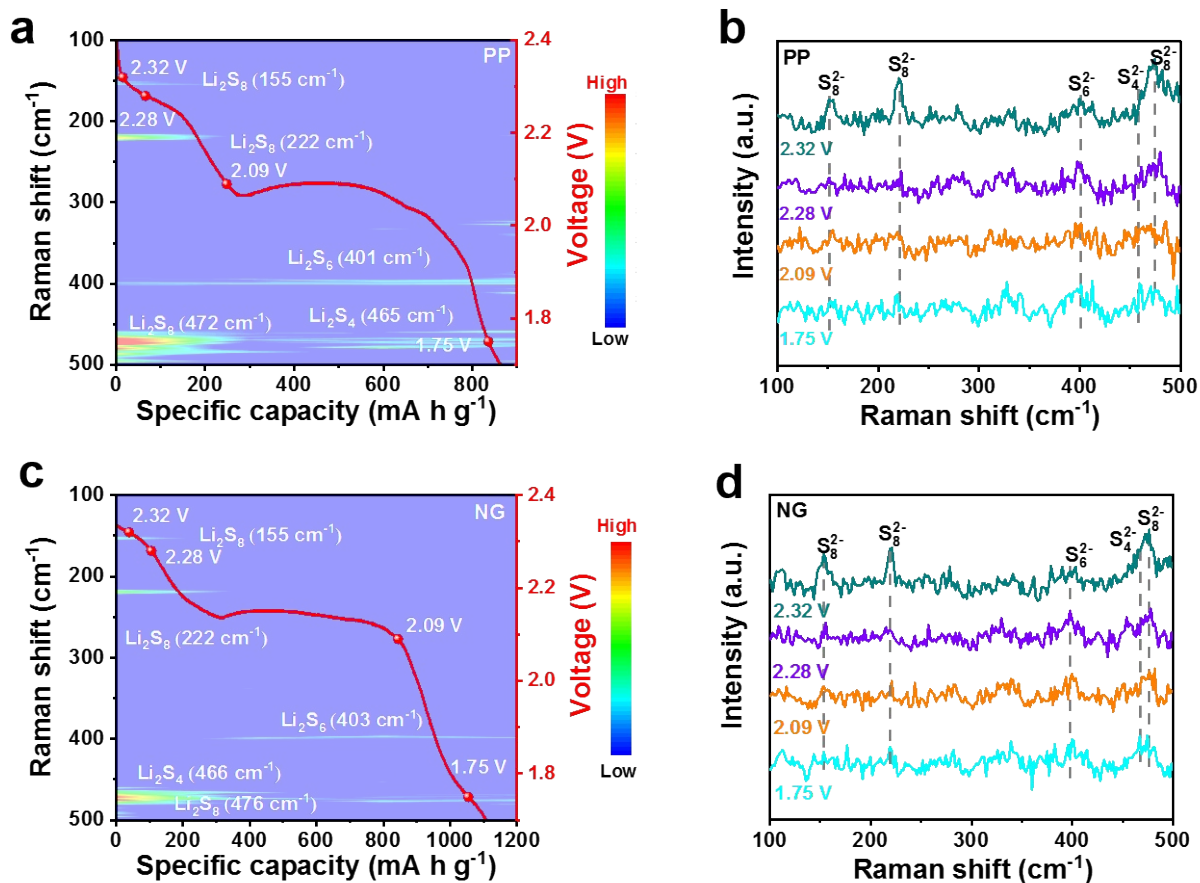
**Fig. S14** Long cycle performance test for the batteries at 1C.



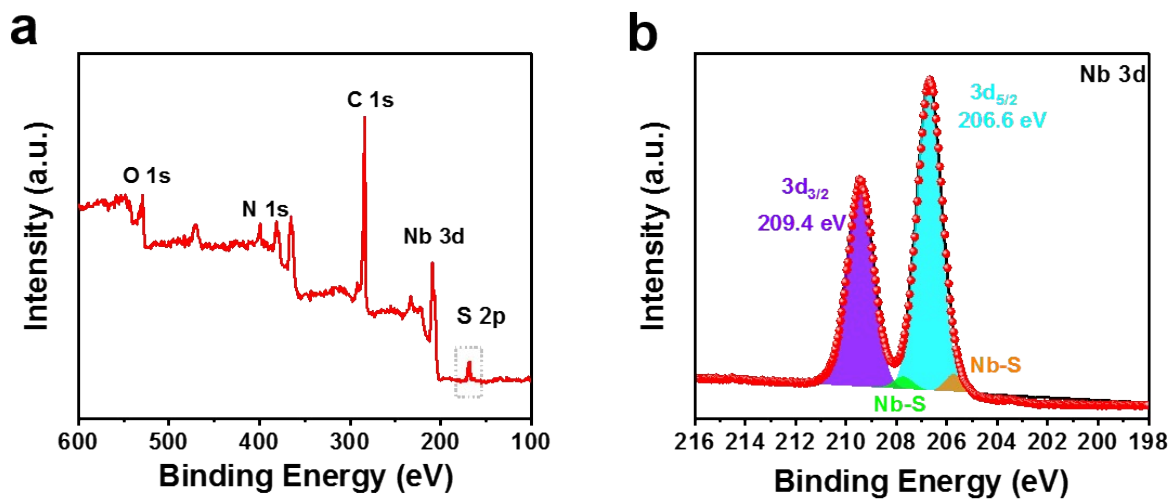
**Fig. S15** I-t curves and the corresponding EIS plots of the batteries with separators of (a, c) PP and (b, d) NG.



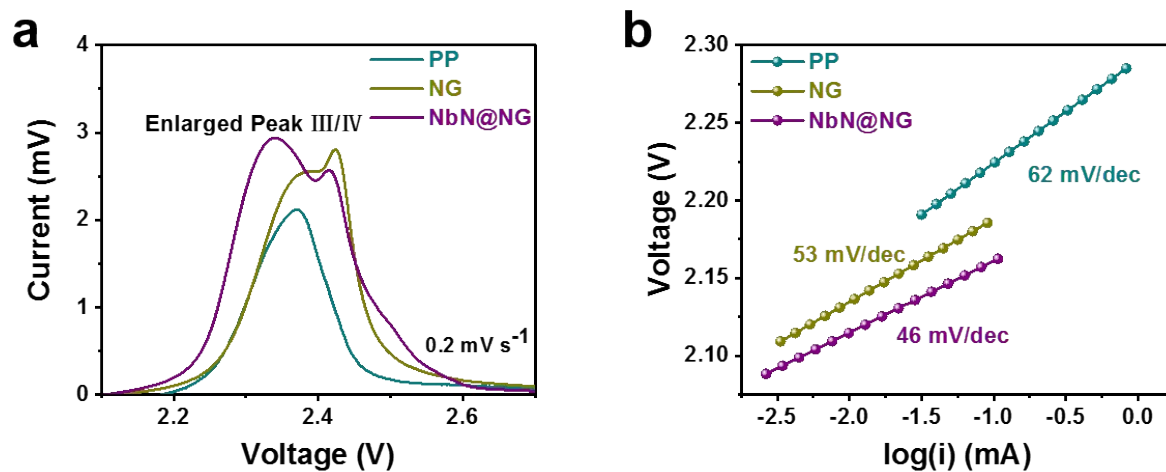
**Fig. S16** Digital photographs of an in-situ Raman test configuration.



**Fig. S17** (a, c) In situ time-resolved Raman spectra and (b, d) selected Raman spectroscopy of Li-S cells based on PP and NG modified separators during the discharging processes.



**Fig. S18** (a) XPS spectrum of the NbN@NG + Li<sub>2</sub>S<sub>6</sub> sample. (b) The corresponding Nb 3d XPS spectrum after LiPSs adsorption test.



**Fig. S19** LSV profiles of an enlarged view in (a) peak III and peak IV of Li-S batteries with different separators. (b) Tafel plots calculated from reduction peak II and oxidation peak III.



**Table S1.** Summary of electrochemical parameters of different separators

Modified separator	PP	NG	NbN@NG
$D_{\text{Li}^+}$ at peak 1 ( $\text{cm}^2 \text{s}^{-1}$ )	$6.779 \times 10^{-8}$	$1.211 \times 10^{-7}$	$2.227 \times 10^{-7}$
$D_{\text{Li}^+}$ at peak 2 ( $\text{cm}^2 \text{s}^{-1}$ )	$1.444 \times 10^{-8}$	$3.115 \times 10^{-8}$	$5.538 \times 10^{-8}$
$D_{\text{Li}^+}$ at peak 3 ( $\text{cm}^2 \text{s}^{-1}$ )	$1.504 \times 10^{-8}$	$2.072 \times 10^{-8}$	$2.984 \times 10^{-8}$
$\text{Li}^+$ conductivity ( $\text{mS cm}^{-1}$ )	0.869	0.374	0.765
$\text{Li}^+$ transfer number	0.623	0.747	0.876

**Table S2.** EIS fitting results of Li-S batteries paired with different separators before cycling

Modified separator	PP	NG	NbN@NG
$R_e (\Omega)$	3.61	6.47	3.38
$R_{ct} (\Omega)$	72.64	62.96	30.57

**Table S3.** EIS fitting results of Li-S batteries paired with different separators after cycling

Modified separator	PP	NbN	NbN@NG
$R_e$ ( $\Omega$ )	7.55	7.38	4.63
$R_{ct}$ ( $\Omega$ )	50.88	45.61	22.15
$R_s$ ( $\Omega$ )	13.77	9.02	2.17

**Table S4.** Electrochemical performances of this work compared with previous works involving different separators in recently reported literature.

Modifid separator	S loading (mg cm <sup>-2</sup> )	Coating loading (mg cm <sup>-2</sup> )	Rate capacity (mA h g <sup>-1</sup> )	Capacity decay rate/cycle number/ C rate	Ref.
TiN	1.3	1	782 (2C)	0.085%/300/2C	1
VN	1.6	1.52	760 (2C)	0.077%/800/1C	2
CNT/MoP <sub>2</sub>	1.2	0.58	1223 (0.2C)	0.152%/100/0.2C	3
NbN	2.0	0.2	800 (1C)	0.08%/300/1C	4
VSe <sub>2</sub> NC@NG	--	0.07-0.11	6000 (8C)	0.052%/550/2C	5
Co <sub>9</sub> S <sub>8</sub> /CoO	1.0	0.3	536 (5C)	0.014%/300/1C	6
CoNi@MP C	1.0-1.5	--	665.3 (4C)	0.087%/500/4C	7
MoS <sub>2</sub>	1.6	--	550 (C)	0.083%/600/0.5C	8
Sb <sub>2</sub> S <sub>3</sub> /CNT	1.0	--	770 (2C)	0.05%/200/2C	9
Co/mSiO <sub>2</sub> - NCNTs	1.15	0.25	552 (5C)	0.09%/250/1C	10
<b>NbN@NG</b>	<b>1.3-1.6</b>	<b>0.25-0.28</b>	<b>819 (4C)</b>	<b>0.036%/500/2C</b>	<b>This work</b>

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