

# Supplementary Information

## Design of multi-functional gel polymer electrolyte with 3D compact stacked polymer micro-sphere matrix for high- performance lithium metal batteries

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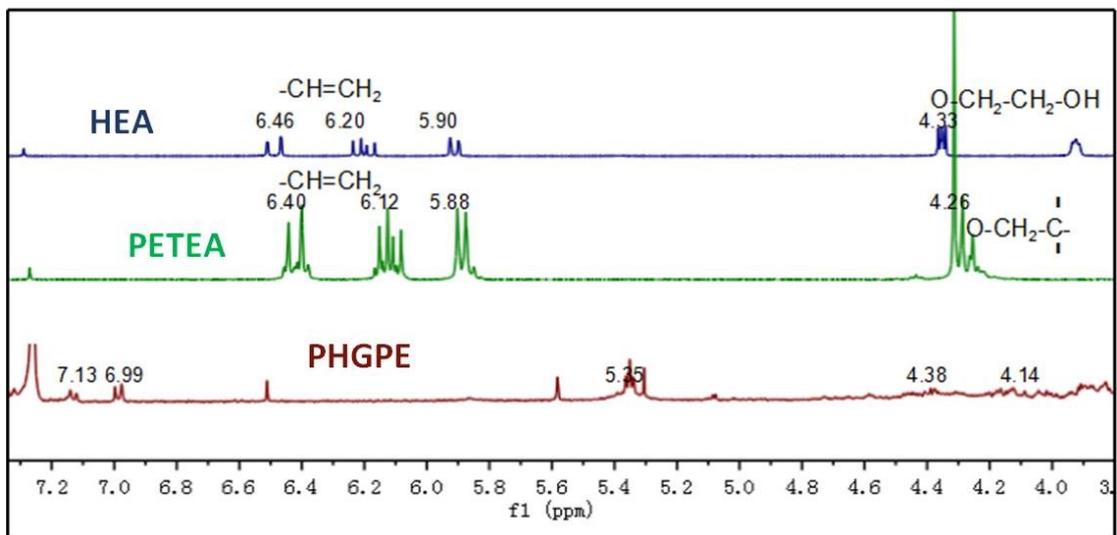
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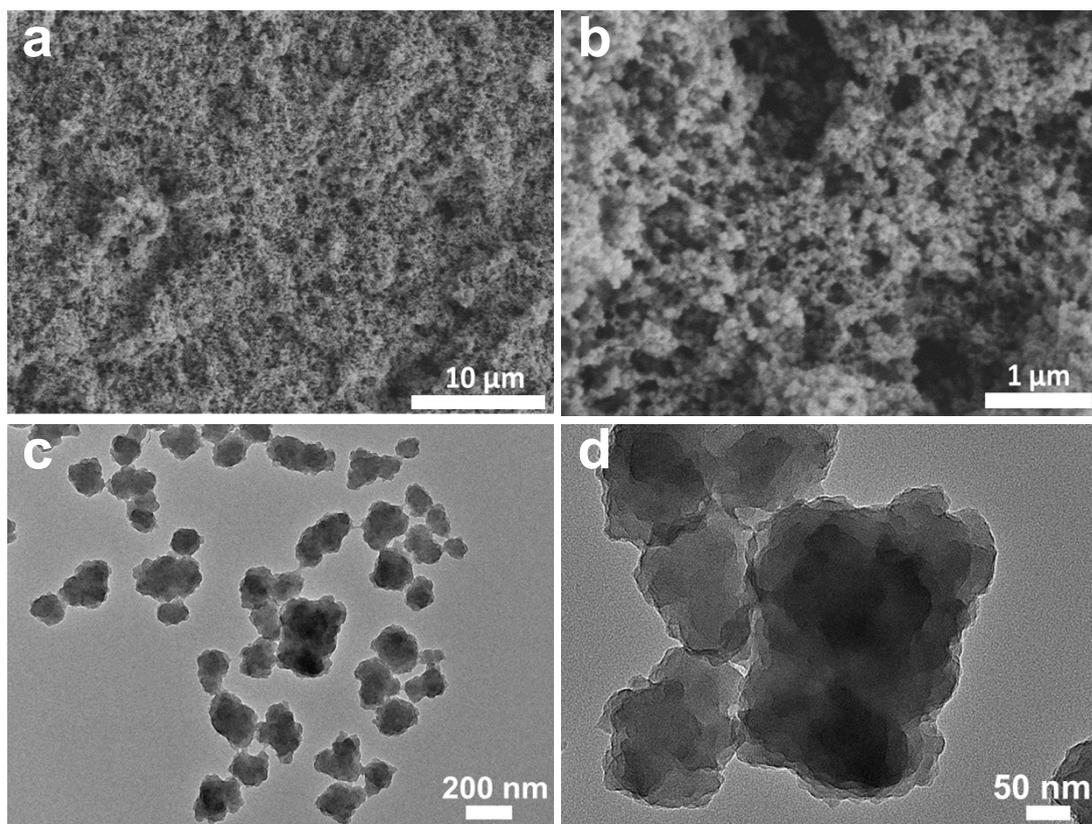
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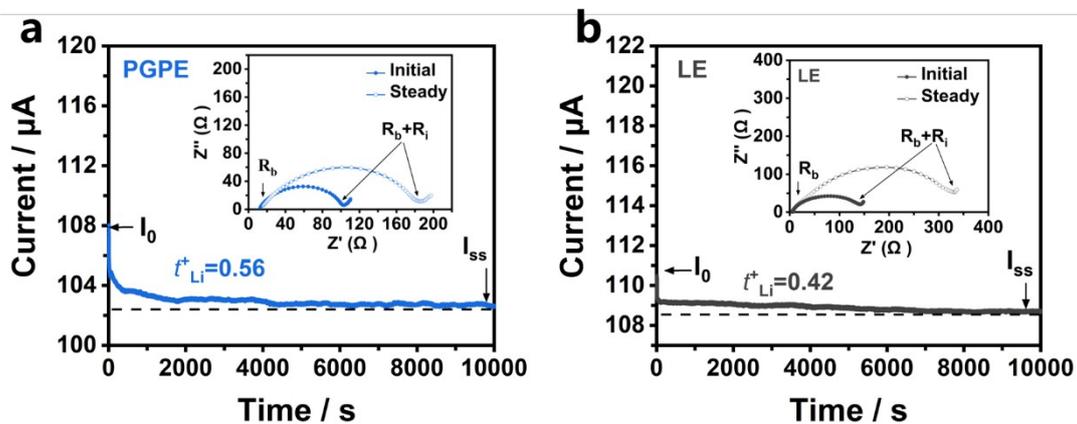
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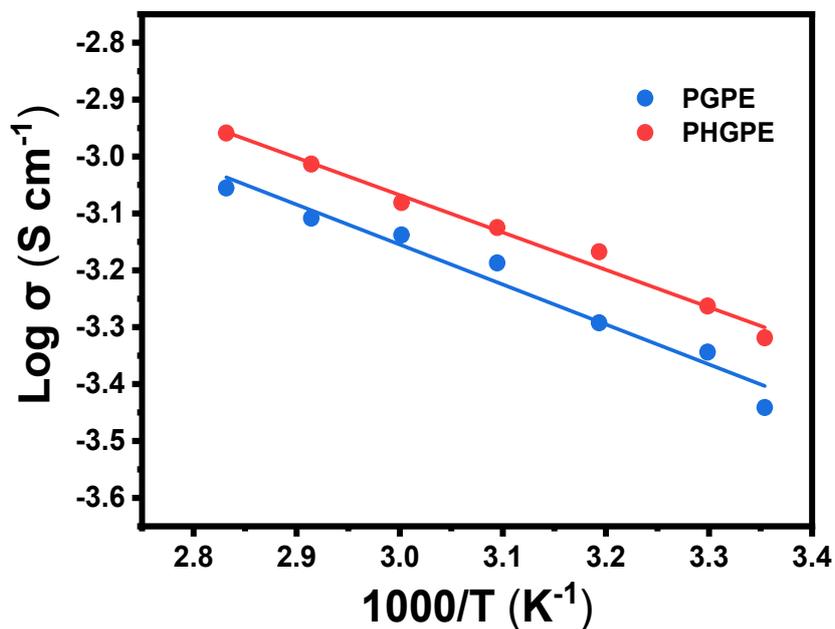
**Fig. S1.** <sup>1</sup>H NMR spectra of HEA, PETEA and PHGPE



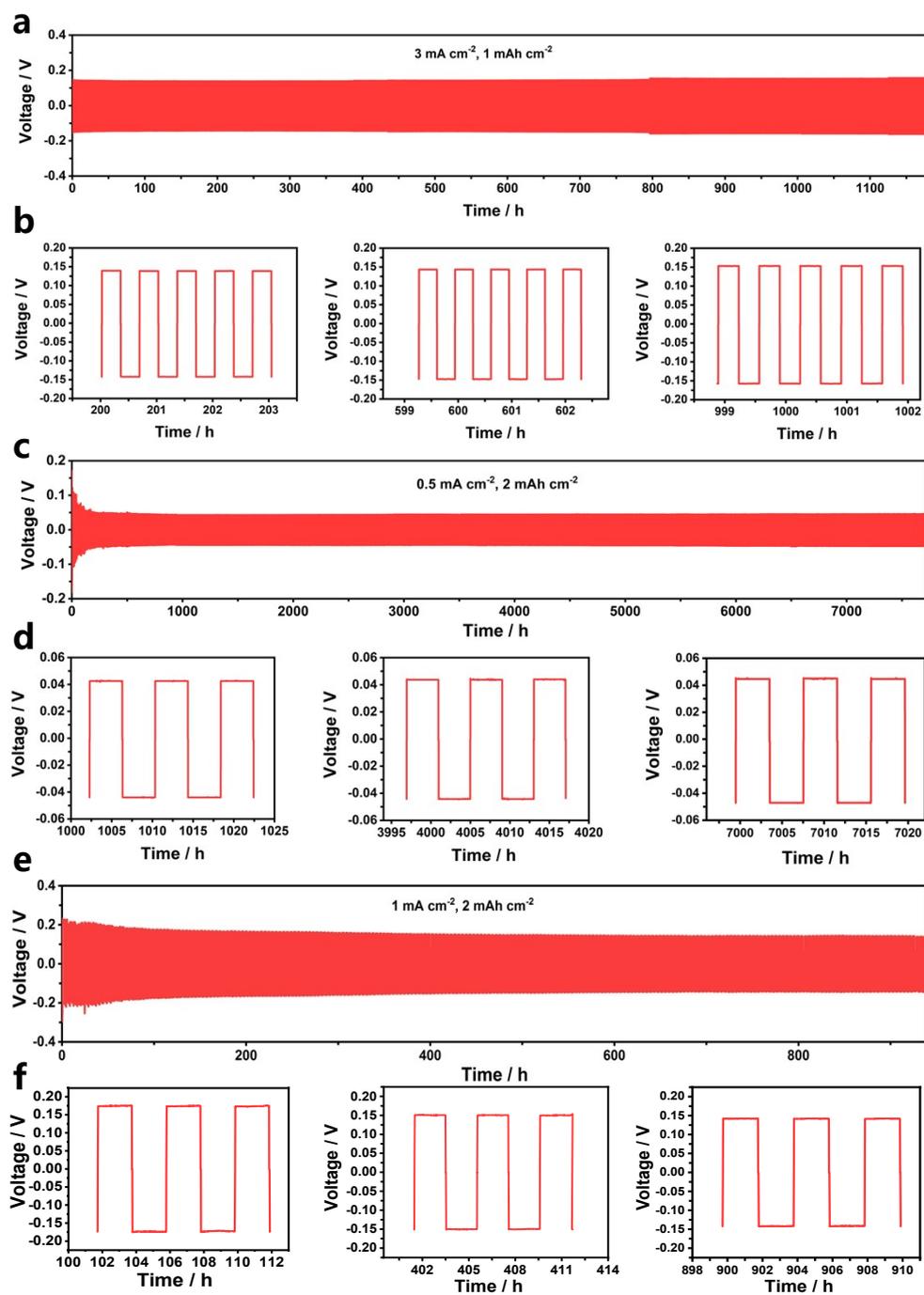
**Fig. S2.** (a, b) SEM images of PGPE matrix. (c, d) TEM images of PGPE matrix.



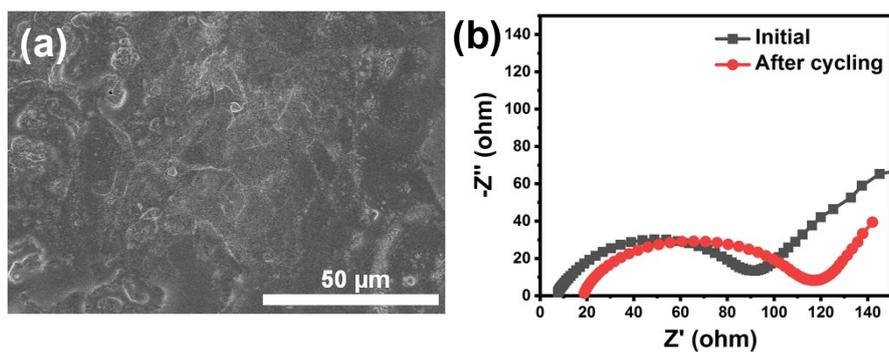
**Fig. S3.** The chronoamperometry profiles of (a) Li||PGPE||Li and (b) Li||LE||Li cells under a polarization voltage of 10 mV. The corresponding EISs before and after polarization are shown in the insets.



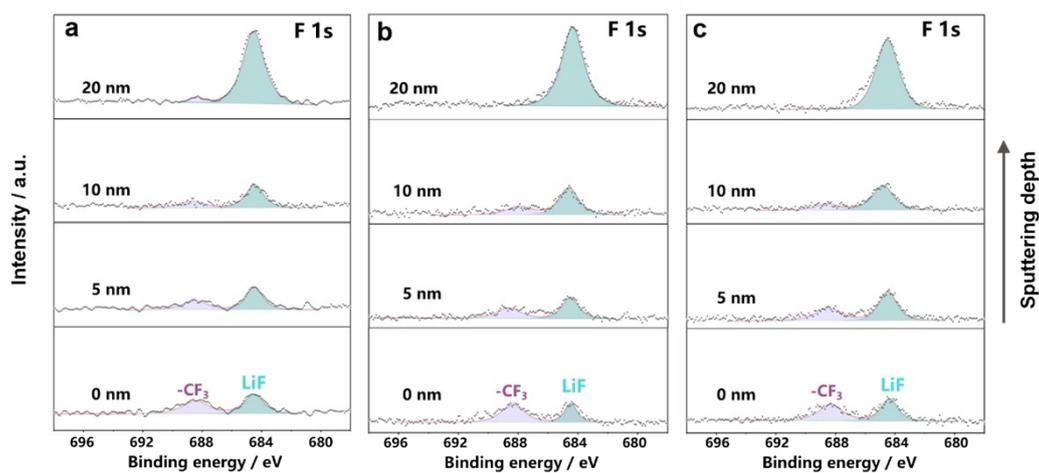
**Fig. S4.** Ionic conductivities of PGPE and PHGPE as a function of temperature.



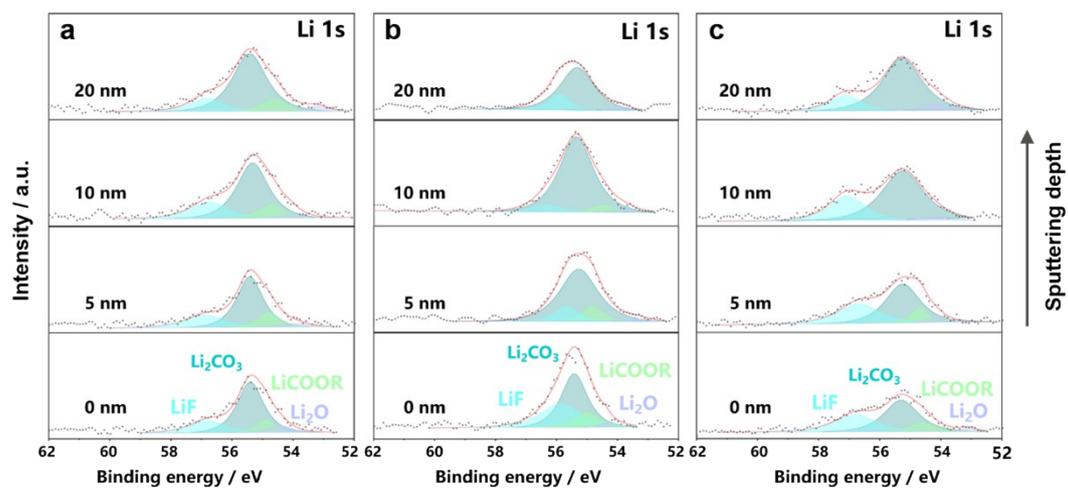
**Fig. S5.** Cycling performance of Li||PHGPE||Li symmetric cells at (a-b)  $3 \text{ mA cm}^{-2}$ ,  $1 \text{ mAh cm}^{-2}$ , (c-d)  $0.5 \text{ mA cm}^{-2}$ ,  $2 \text{ mAh cm}^{-2}$ , (e-f)  $1 \text{ mA cm}^{-2}$ ,  $2 \text{ mAh cm}^{-2}$ .



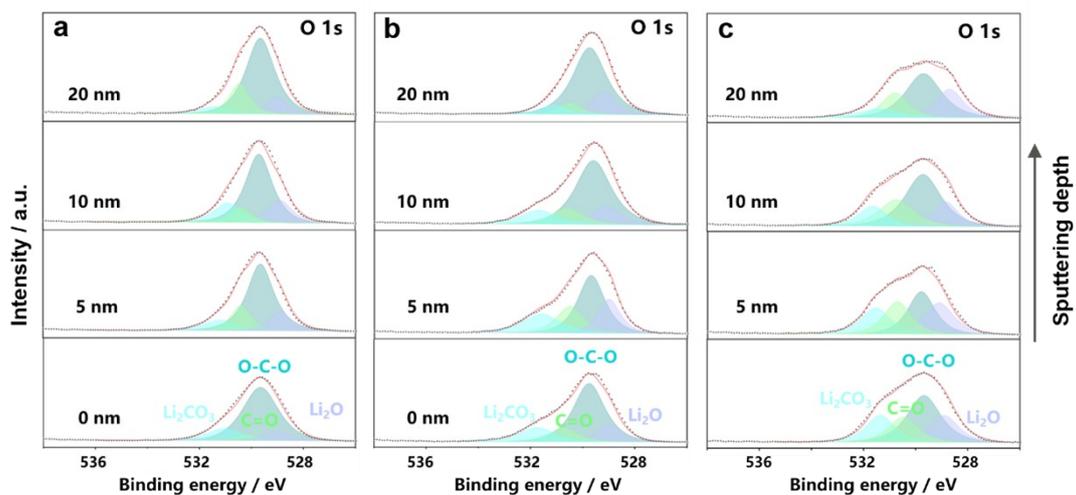
**Fig. S6.** (a) SEM image of Lithium metal in PHGPE based symmetric cell after 960 h cycling with a cycling capacity of 2 mAh cm<sup>-2</sup> at a current rate of 1 mA cm<sup>-2</sup>; (b) Nyquist plots of Li||PHGPE||Li symmetric cell before and after cycling.



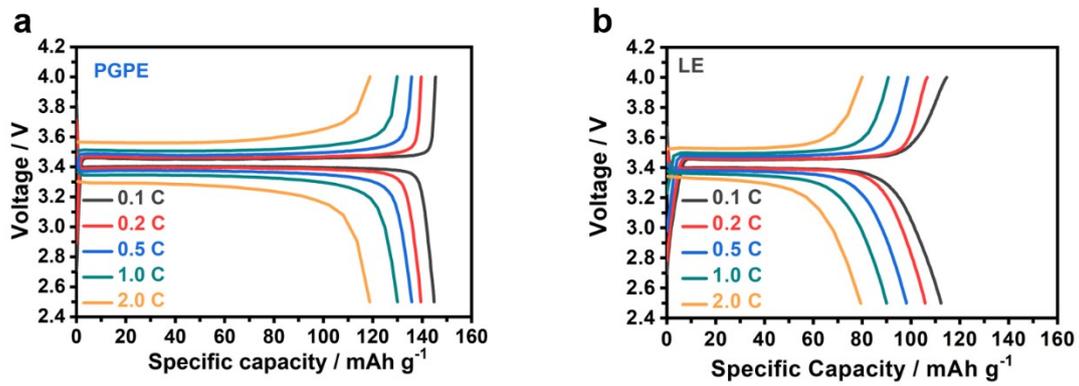
**Fig. S7.** F-1s HRXPS spectra of (a) LE, (b) PGPE and (c) PHGPE at different sputtering depths after 10 cycles of symmetric Li-ion battery. C-F: 688.7 eV, LiF: 684.5 eV.



**Fig. S8.** Li-1s HRXPS spectra of (a) LE, (b) PGPE, (c) PHGPE at different sputtering depths after 10 cycles of symmetric Li-ion battery. LiF: 56.88 eV, Li<sub>2</sub>CO<sub>3</sub>: 55.38 eV, LiCOOR: 54.88 eV, Li<sub>2</sub>O: 54.28 eV.



**Fig. S9.** O-1s HRXPS spectra of (a) LE, (b) PGPE, (c) PHGPE at different sputtering depths after 10 cycles of symmetric Li-ion battery.



**Fig. S10.** Capacity voltage distribution of Li||electrolyte||LFP at different multipliers, (a) PGPE; (b) LE.



**Fig. S11.** Digital images of Li foil for nickel tabs

**Table S1.** The comparison of Li||Li symmetrical cell cycling performances between this work and other reported literatures.

Electrolyte	Current density / areal capacity (mA cm <sup>-2</sup> / mAh cm <sup>-2</sup> )	Plating/ stripping (hours)	Ref.
SPE1-PI-ZIF8	0.1/0.1	800 h	1
	0.3/0.3	300 h	
Li/CF composites	0.25/0.5	750 h	2
	0.5/0.5	560 h	
PEO-SCN-LiTFSI	0.05/0.05	600 h	3
Al <sub>2</sub> O <sub>3</sub> -GPE	1/1	1000 h	4
CSE-B- 71515	0.2/0.2	333 h	5
PDADMA-FSI	0.1/0.1	225 h	6
IPN9-10PPC	1.5/1.5	300 h	7
PIL-IPN	0.5/1.5	1800 h	8
POSS-4PEG2K	0.3/0.9	2600 h	9
GPR <sub>5</sub> -LiTFSI	0.1/0.1	600 h	10
PVDF film	0.5/0.5	1400 h	11
	1/1	650 h	
	1/2	370 h	
Li <sub>3</sub> PO <sub>4</sub> /PVA layer	1/2	1000 h	12
PU-LiF	1/1	1000 h	13
	1/2	800 h	
PHGPE	1/1	6000 h	This work
	0.5/2	7700 h	
	1/2	7700 h	

**Table S2.** The specifications of the Li||LFP pouch cell.

Cell components	Specification	Parameters
Cathode (LFP)	Areal mass loading ( $\text{mg cm}^{-2}$ )	11.71
	The area of cathode ( $\text{mm} \times \text{mm}$ )	$50 \times 83$
	The content of LFP (wt.%)	96
	Folds	12
Anode (Lithium foil)	Thickness ( $\mu\text{m}$ )	60
	The area of lithium foil ( $\text{mm}^2$ )	$55 \times 88$
	Folds	13
Electrolyte	1.0 M $\text{LiPF}_6$ in EC : DMC=1 : 1 V/V (contain 1.5% of the mass fraction with PHGPE)	
	The mass of electrolyte (g): capacity of pouch cell (Ah)	2:1
Pouch cell	Specific capacity (Ah)	1

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