

Supplementary Information

Lateral built-in field of 2D/2D SnS₂/SnSe₂ in-plane heterostructure with boosted interfacial charge transfer

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DFT calculation results:

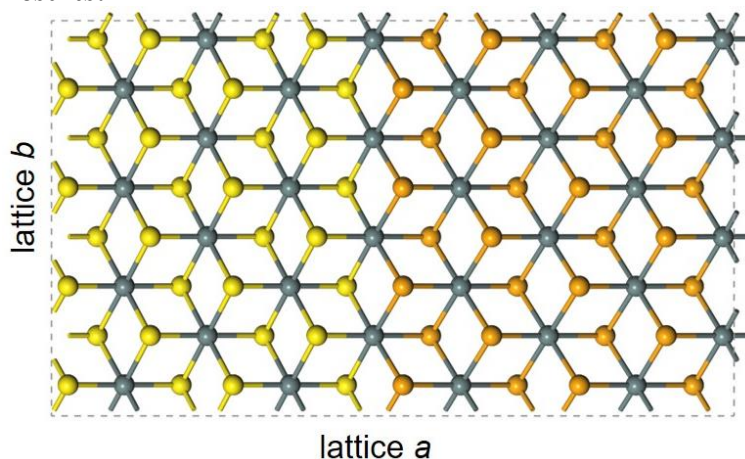


Fig. S1. The slab model of $\text{Sn}_{32}\text{S}_{32}\text{Se}_{32}$ supercell for $\text{SnS}_2/\text{SnSe}_2$ lateral heterojunction, where the lateral heterojunction is $a=26.17 \text{ \AA} / b=15.13 \text{ \AA}$. The orange, yellow, and grey balls are Se, S, and Sn atoms, respectively.

Table S1. Bandgap values derived from the density of states (DOS) in different domains of $\text{SnS}_2/\text{SnSe}_2$ lateral heterojunction.

Domains	SnS_2	Heterointerface	SnSe_2
Bandgap (E_g)	0.83	0.41	0.44

Analyzed results of DFT calculations. The adsorption energies (E_{ads}) for NO_2 molecule on the surface of $\text{SnS}_2/\text{SnSe}_2$ lateral slab shows that the E_{ads} value of interface domain is more negative than that of SnS_2 and SnSe_2 domains (**Fig. 1a**), which the negative value represents the exothermic process, indicating the interfaces could serve as the active sites for NO_2 adsorption. Moreover, the charge density difference of the lateral heterojunctions shows the efficient diffusion of charge carriers across interface (**Fig. 1b**), which originates from the transport bridge of chemical bond S-Sn-Se and is conducive to improving the charge separation and transfer kinetics. The reduced bandgap of the interface domain (**Fig. 1c** and **Table S1**) indicates the improved electrical conductivity in $\text{SnS}_2/\text{SnSe}_2$ lateral heterostructure which could serve as a channel to boost charge transfer along the 2D basal plane to react with electrophilic NO_2 molecules, contributing to enhancing the gas sensing performance. The DOS of interface in the valence band near the Fermi level shows that the holes in the valence band are mainly contributed by Se (**Fig. 1c**), implying that the electrons of SnSe_2 near the interface tend to diffuse into SnS_2 while leaving holes in SnSe_2 .

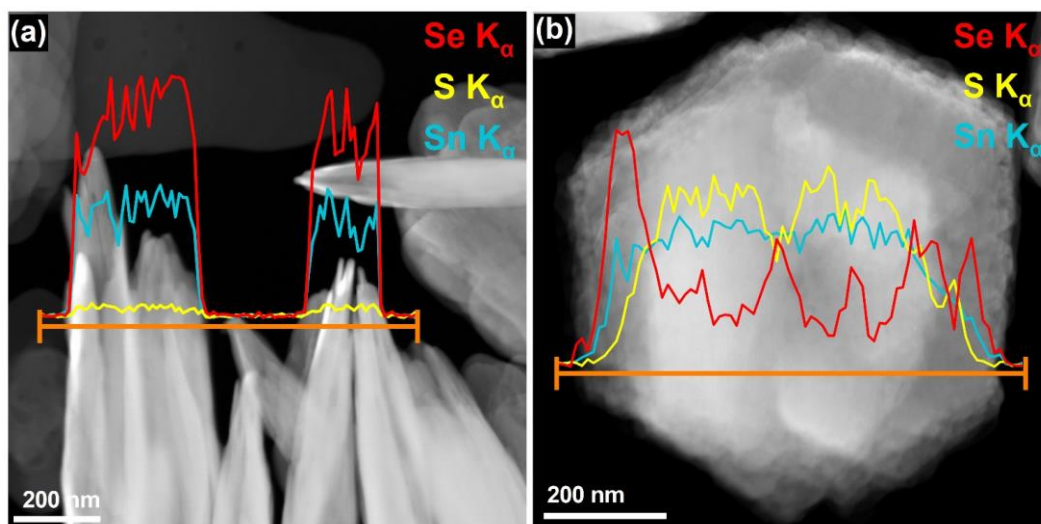


Fig. S2. HAADF-STEM image of SnS₂/SnSe₂-2 nanosheets, with composition determined from the EDX profile along orange line of Se K α (red line), S K α (yellow line) and Sn K α (cyan line) signals: (a) side view, (b) top view. The results show that the spatial distribution of Se and S elements have the highest and lowest concentration at the edges, respectively, indicating that the *in-situ* grown SnSe₂ interfaces with the SnS₂ to form a lateral rather than vertical 2D/2D heterostructure at low concentration of Se precursor.

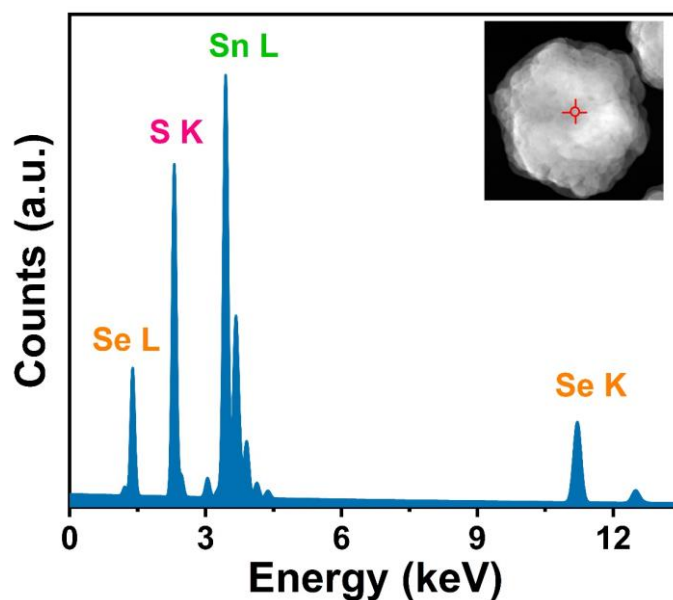


Fig. S3. EDX spot analysis in the core region of SnS₂/SnSe₂-3 nanosheet. The result shows the coexistence of Se and S elements in the core region of SnS₂/SnSe₂-3, further confirming the formation of the core-shell heterostructure for SnS₂/SnSe₂-3.

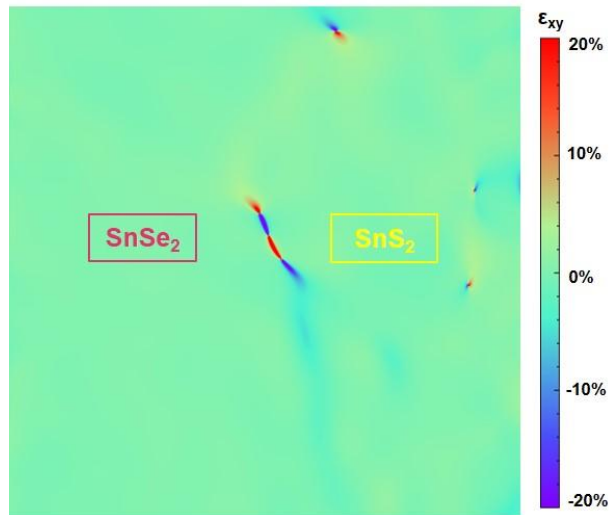


Fig. S4. geometric phase analysis (GPA) map deriving from the HRTEM image of SnS₂/SnSe₂-2 heterostructure. The result indicates the low density of stacking faults between the parent SnS₂ and derived SnSe₂, confirming the formation of sharp lattice-matched heterointerface.

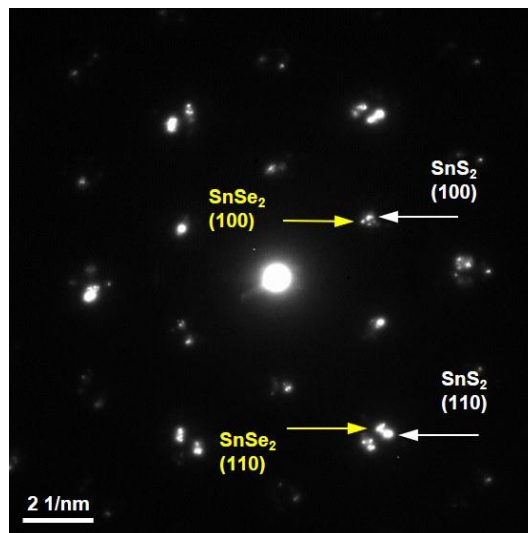


Fig. S5. SAED pattern of SnS₂/SnSe₂-2 heterostructure. The result shows an approximate hexagon in distribution with two identifiable groups of diffraction spots in the same orientation, corresponding to the SnSe₂ and SnS₂, respectively.

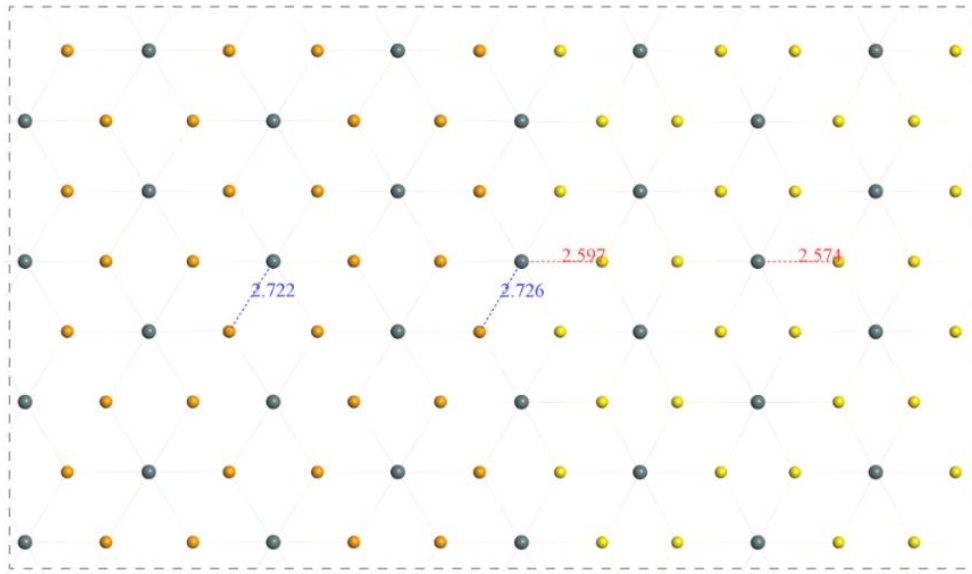


Fig. S6. Atomic arrangement and bond lengths of Se-Sn and S-Sn bonds in SnSe₂ region, interface, and SnS₂ region of 2D SnS₂/SnSe₂ lateral heterostructure. The orange, yellow, and grey balls represent the Se, S, and Sn atoms, respectively.

Table S2. The bond lengths and bond orders around the interface of SnS₂/SnSe₂ lateral heterostructure. The number after the element is the serial number of the element in the SnS₂/SnSe₂ lateral heterostructure.

	Bond Length (Å)	Bond Order
S7-Sn22	2.574	0.39
S18-Sn24-Se29	2.597, 2.726	0.35, 0.03
Se12-Sn6	2.722	0.20

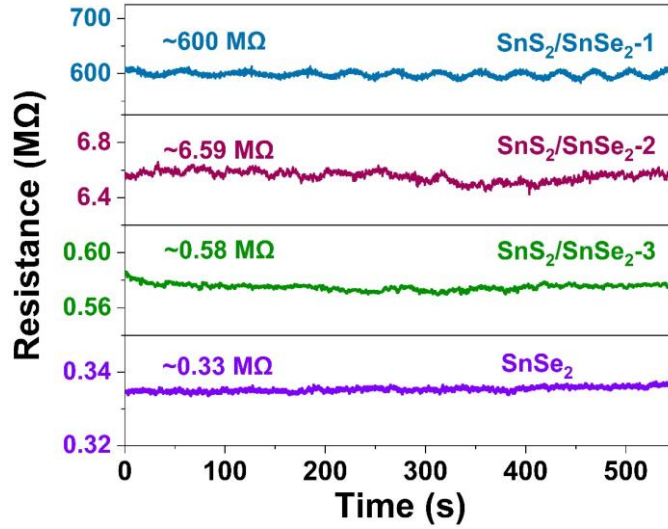


Fig. S7. Baseline resistance of SnS₂ template, SnS₂/SnSe₂-1, SnS₂/SnSe₂-2, SnS₂/SnSe₂-3, and SnSe₂. The baseline resistance sharply declines after the formation of SnS₂/SnSe₂ heterostructure, from ~600 MΩ of SnS₂/SnSe₂-1 to ~0.58 MΩ of SnS₂/SnSe₂-3 that is close to ~0.33 MΩ of SnSe₂, indicating that the formation of 2D/2D seamless stitching SnS₂/SnSe₂ heterostructure significantly modulates the electronic structure and improves the conductivity of SnS₂. This is due to that the electrons of SnSe₂ diffuse into the SnS₂ to increase the electron density of SnS₂, contributing to the lower resistance of heterostructure than that of pristine SnS₂.

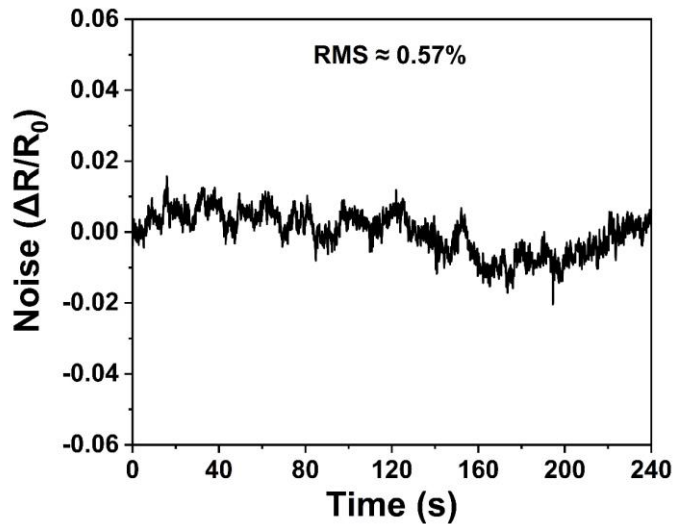


Fig. S8. Experimentally recorded noise of response for the SnS₂/SnSe₂-2 sensor. The result indicates that the root-mean-square (RMS) value of background noise is about 0.57%.

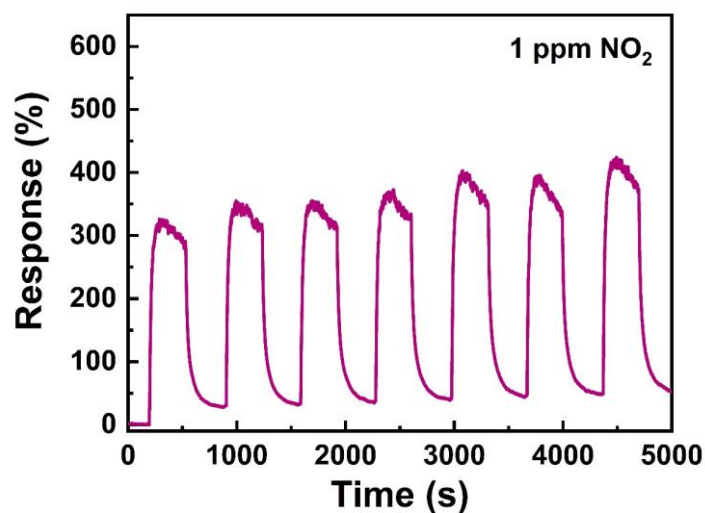


Fig. S9. Repeatability test of SnS₂/SnSe₂-2 conducted by seven successive cycles upon the exposure of 1 ppm NO₂ at room temperature. The result shows the good reversible response and recovery with the nearly same response value.

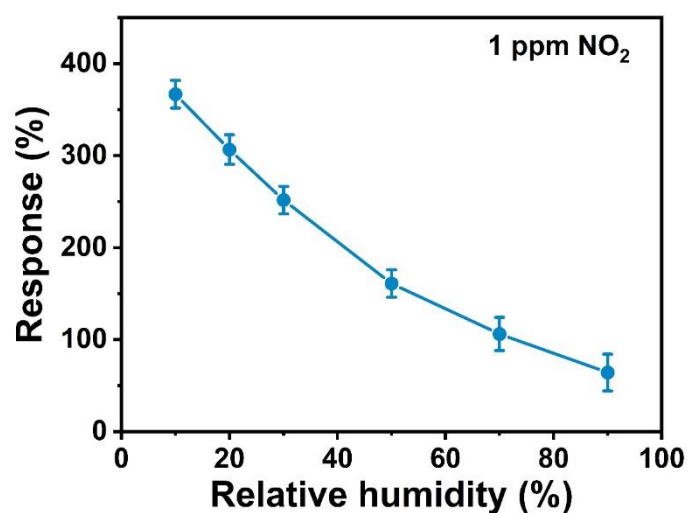


Fig. S10. Evolution of response with the relative humidity increasing from 10% to 90% for SnS₂/SnSe₂-2. The result shows that the response of SnS₂/SnSe₂-2 decreases with the relative humidity changing from 10% to 90%, which may be ascribe to the strong competitive adsorption between H₂O and NO₂ molecules.

Table S3. The sensing properties of SnS₂/SnSe₂-2 sensor compared with the state-of-the-art resistive-type SnS₂-based NO₂ sensors in literature.

t_{res} = Response time, t_{rec} = Recovery time, RT* = Light illumination at room temperature.

Sensing materials	Condition (°C)	NO ₂ (ppm)	Response (%)	t _{res} /t _{rec} (s/s)	Sensitivity (% ppm ⁻¹)	Detection Limit (ppb)	Reference
SnS ₂ flakes	150	10	2000	- / -	~570	300	[1]
2D SnS ₂ /rGO	150	5	3200	50 / 48	/	500	[2]
Mo-doped SnS ₂	150	100	4751	33.5 / 21	/	/	[3]
3D SnS ₂ /rGO	RT	8	49.8	153 / 76	6.1	8.7	[4]
SnS ₂ /MoS ₂	RT	5	270	2.9 / 24.5	/	50	[5]
SnS ₂ /SnS	RT	4	660	365 / 1216	~200	75	[6]
S-rich SnS ₂	RT	10	701	272 / 3800	0.5	1000	[7]
SnS ₂ flowers	RT*	5	1428	400 / 1100	251	400	[8]
SnSe ₂ /SnSe	RT*	1	~600	- / >300	6.1	100	[9]
SnS₂/SnSe₂-2	RT	4	1165.2	80 / 78	322.0	50	This work

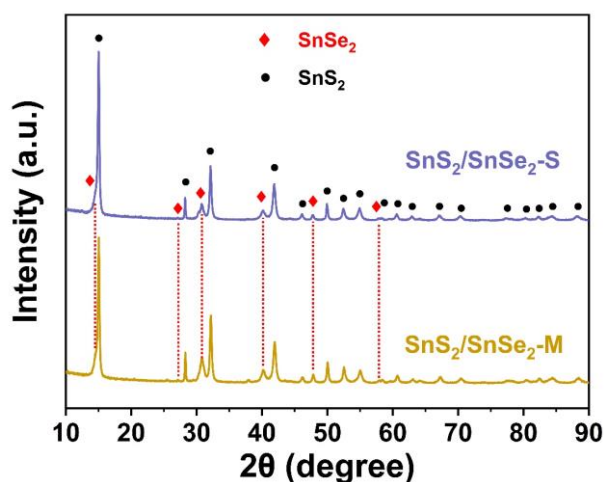


Fig. S11. XRD patterns of SnS₂/SnSe₂-M and SnS₂/SnSe₂-S heterostructures. The SnS₂/SnSe₂-M and SnS₂/SnSe₂-S heterostructures were synthesized by physical mixing and solvothermal deposition approaches, respectively. The XRD results show the coexistence of SnS₂ and SnSe₂ crystals without any other impurities, indicating the formation of SnS₂/SnSe₂-M and SnS₂/SnSe₂-S heterostructures.

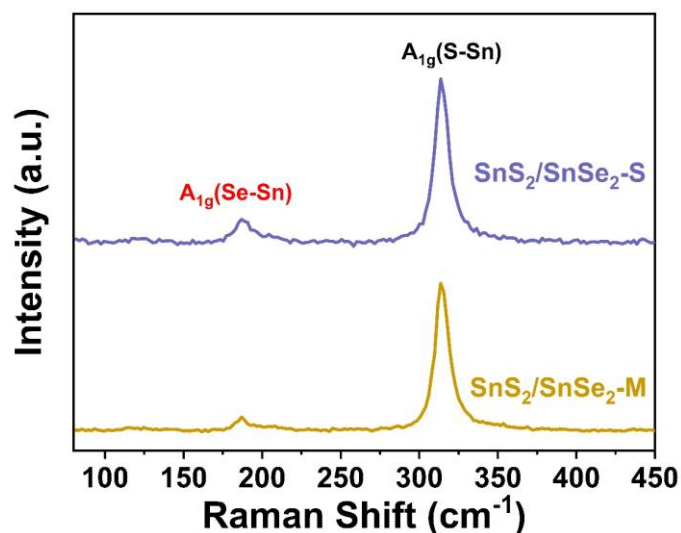


Fig. S12. Raman spectra of SnS₂/SnSe₂-M and SnS₂/SnSe₂-S heterostructures. The results show the vibration modes of layered SnS₂ and SnSe₂ crystals, which is consistent with the XRD result, verifying that the formation of SnS₂/SnSe₂-M and SnS₂/SnSe₂-S heterostructures.

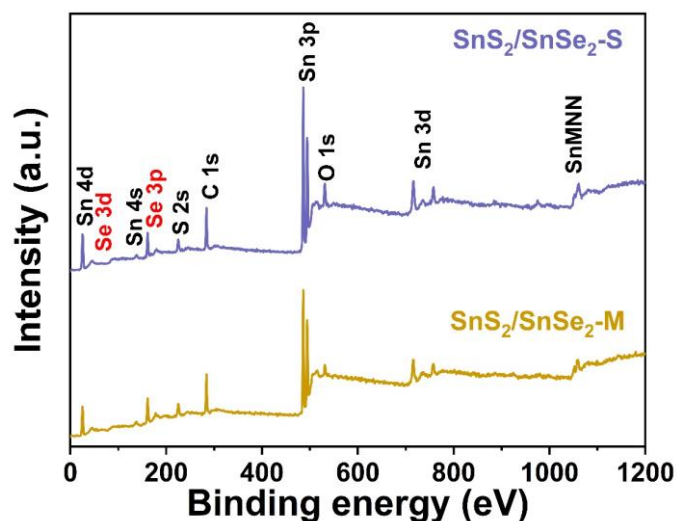


Fig. S13. XPS elemental survey spectra of SnS₂/SnSe₂-M and SnS₂/SnSe₂-S heterostructures. The results show the coexistence of S and Se elements without impurities in the SnS₂/SnSe₂-M and SnS₂/SnSe₂-S heterostructures, confirming that the formation of SnS₂/SnSe₂-M and SnS₂/SnSe₂-S heterostructures.

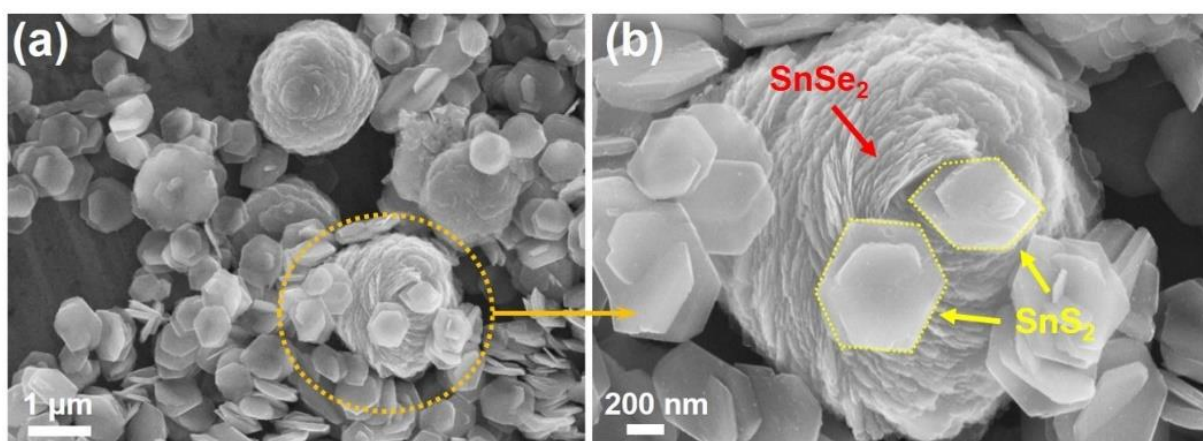


Fig. S14. SEM images of SnS₂/SnSe₂-M heterostructure. The SEM images of SnS₂/SnSe₂-M display that the hexagonal SnS₂ crystals (marked by yellow) randomly and loosely stack on the SnSe₂ crystals (marked by red), indicating a poor-quality heterointerface of SnS₂/SnSe₂-M.

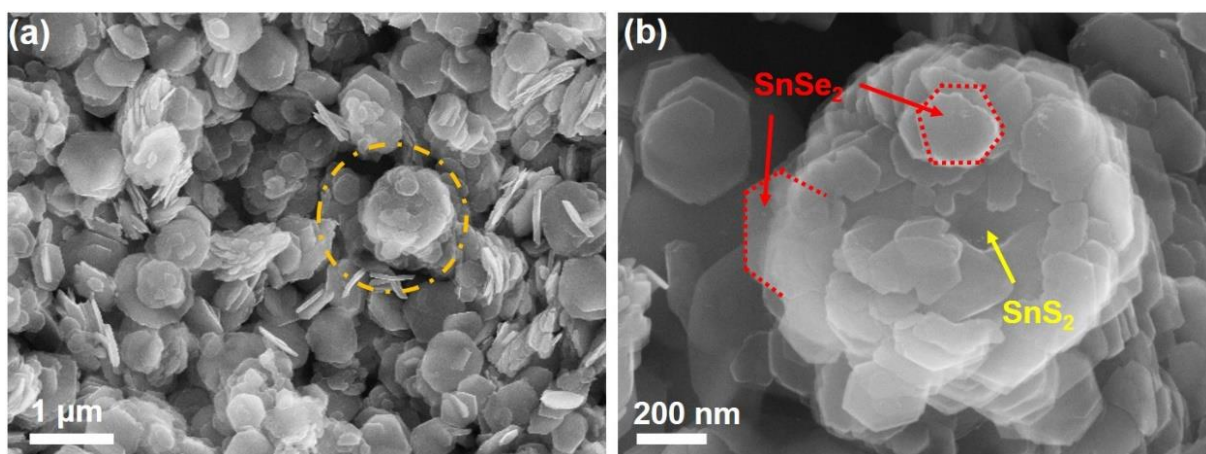


Fig. S15. SEM images of SnS₂/SnSe₂-S heterostructure. The SEM images of SnS₂/SnSe₂-S show that the SnSe₂ crystals (marked by red) grow on the surface of SnS₂ nanosheets (marked by yellow) with a relatively intimate contact compared to the SnS₂/SnSe₂-M.

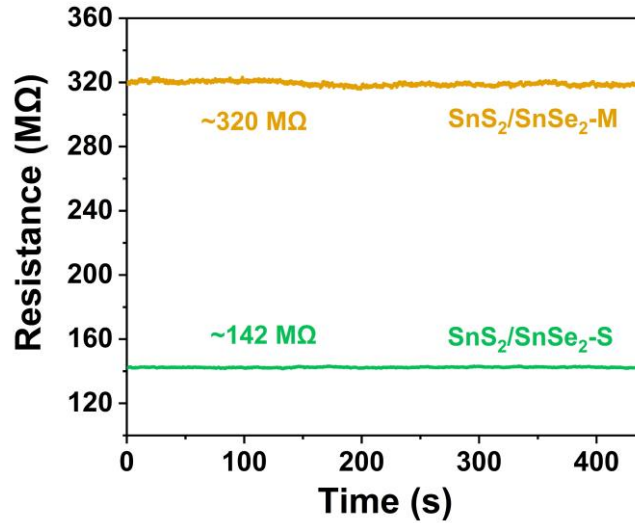


Fig. S16. The initial resistance of the sensors based on the SnS₂/SnSe₂-M and SnS₂/SnSe₂-S heterostructures. The smaller resistance of SnS₂/SnSe₂-S sensor may be due to the better contact and uniform dispersion between SnS₂ and SnSe₂, as compared with the SnS₂/SnSe₂-M.

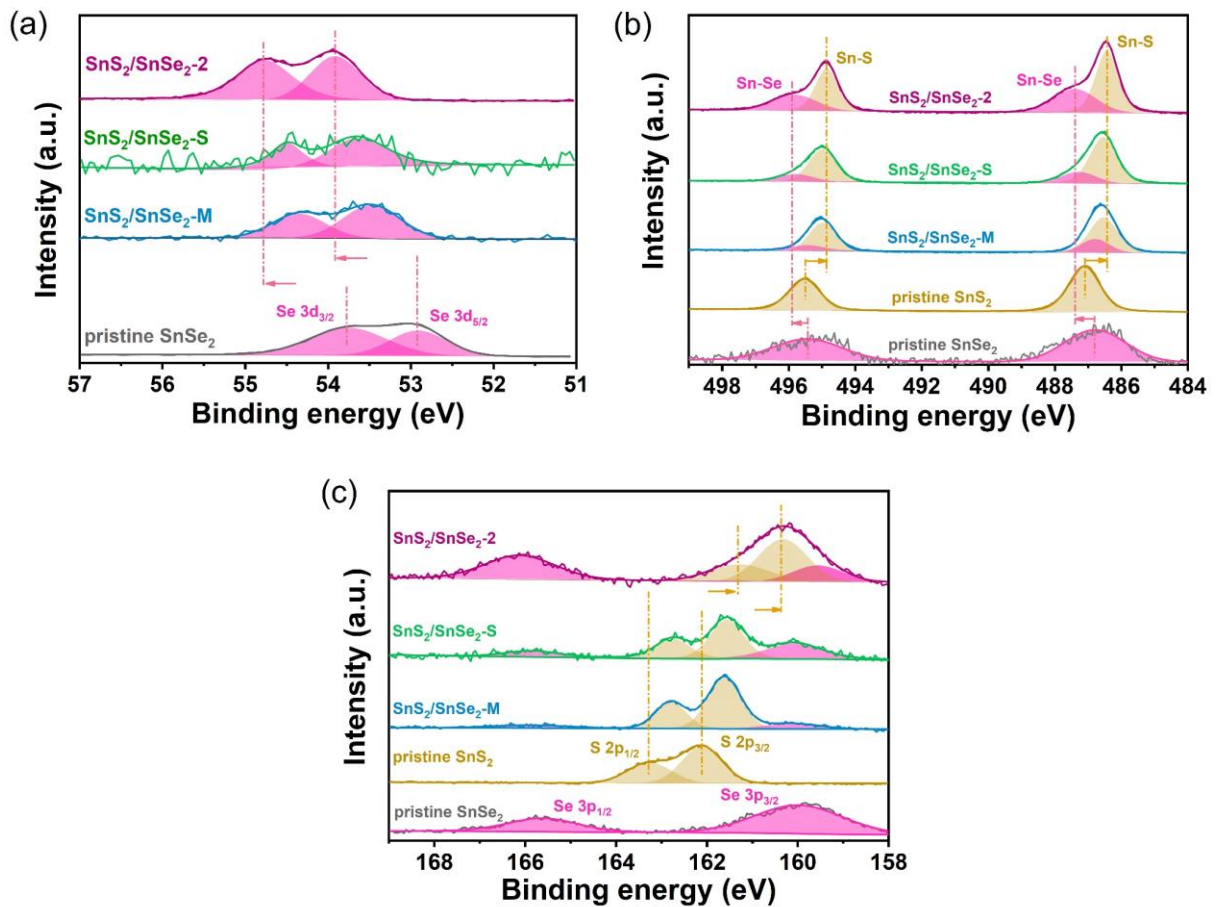


Fig. S17. High-resolution (a) Se 3d, (b) Sn 3d, and (c) S 2p & Se 3p XPS spectra for the pristine SnSe₂, pristine SnS₂, SnS₂/SnSe₂-M, SnS₂/SnSe₂-S, and SnS₂/SnSe₂-2.

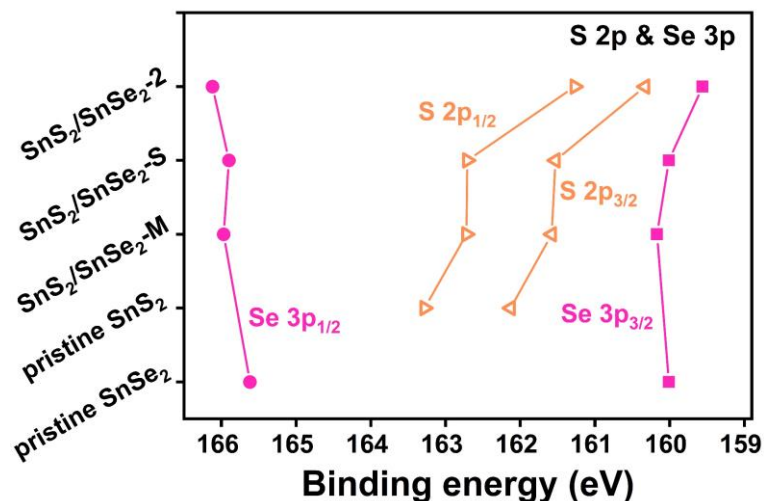


Fig. S18. The summarized S 2p and Se 3p peak positions for the pristine SnSe₂, pristine SnS₂, SnS₂/SnSe₂-M, SnS₂/SnSe₂-S, and SnS₂/SnSe₂-2 samples.

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