

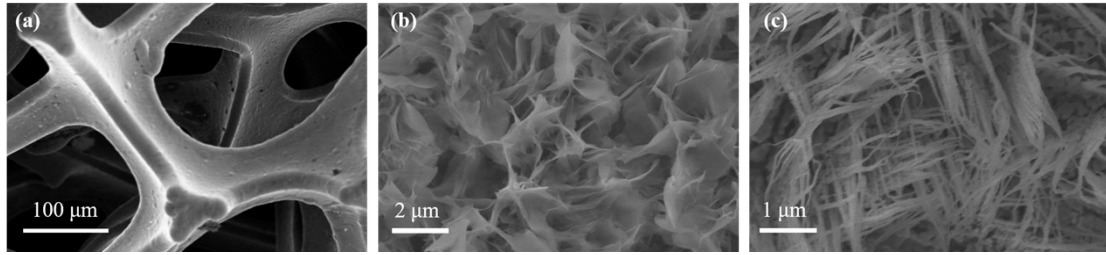
## **Constructing a bifunctional MoO<sub>2</sub>/Co heterojunction for efficient electrocatalytic hydrogen evolution and hydrazine oxidation**

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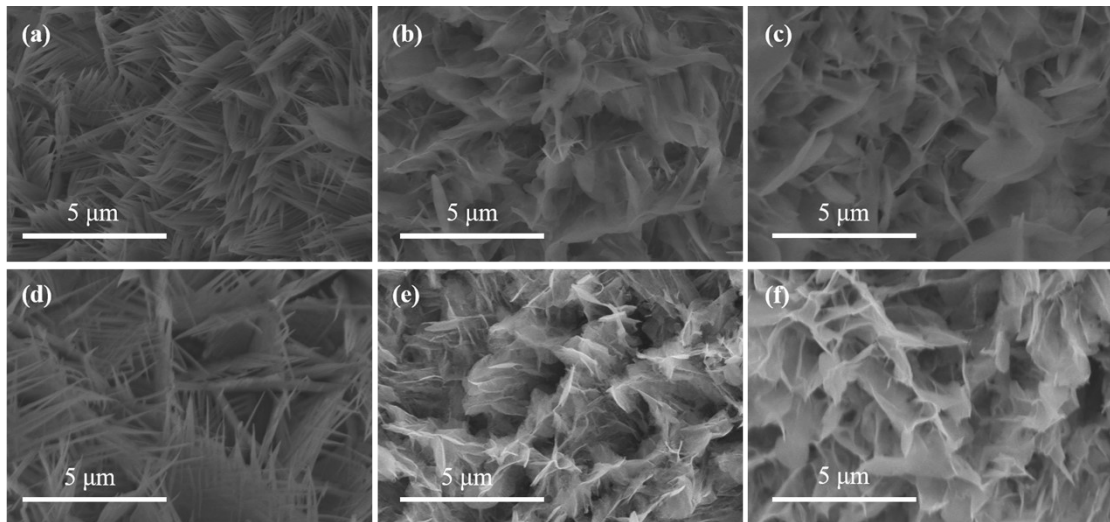
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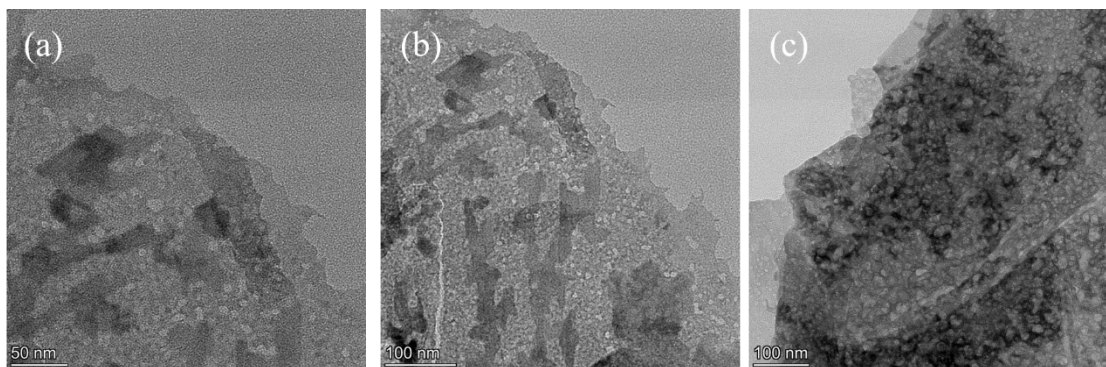
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**Figure. S1** SEM images of bare NF (a), CoMo precursor (b) and Co (c).



**Figure. S2** SEM images of CoMo precursor-1h (a), CoMo precursor-3h (b), CoMo precursor-6h (c), MoO<sub>2</sub>/Co-1h (d), MoO<sub>2</sub>/Co-3h (e) and MoO<sub>2</sub>/Co-6h (f).



**Figure. S3** TEM images of MoO<sub>2</sub>/Co.

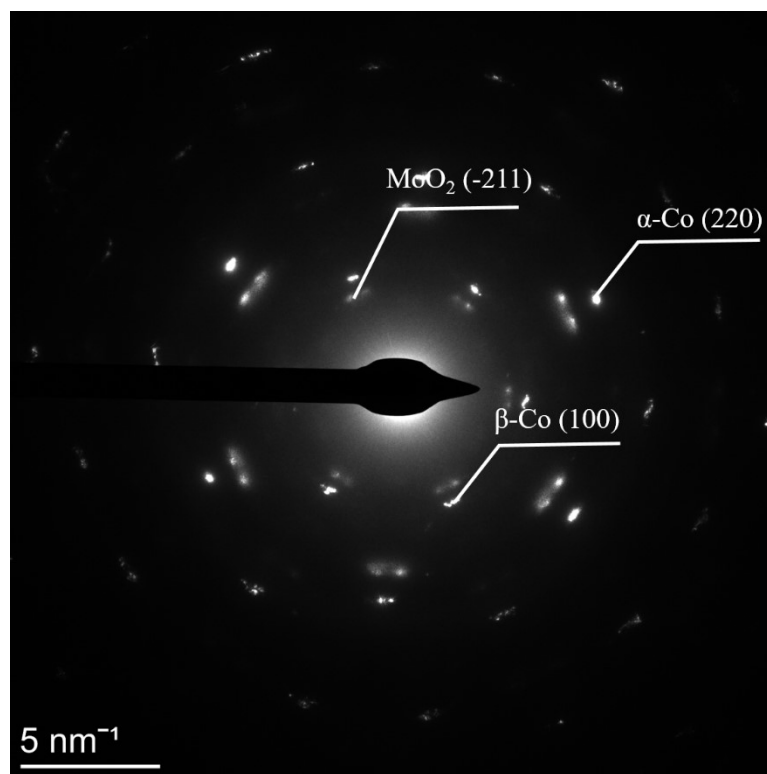


Figure. S4 SAED pattern of MoO<sub>2</sub>/Co.

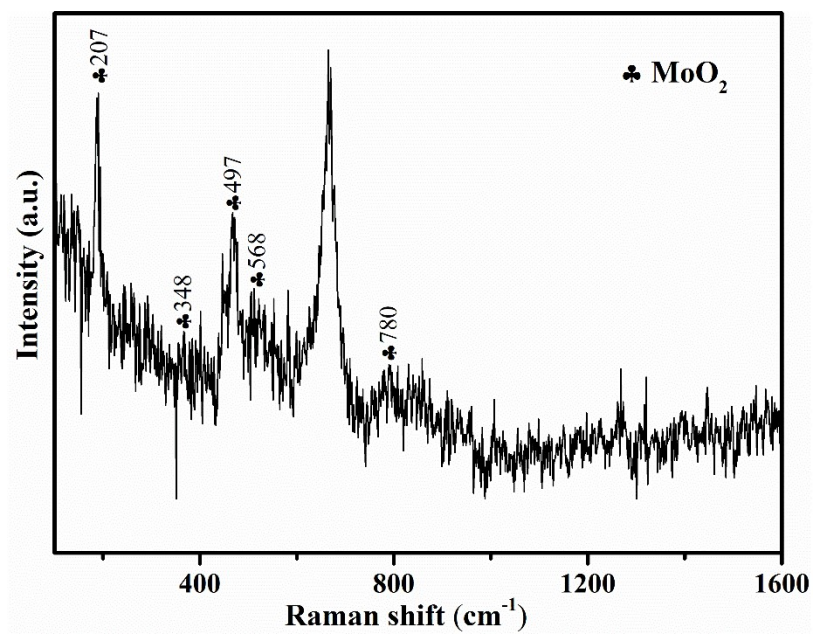
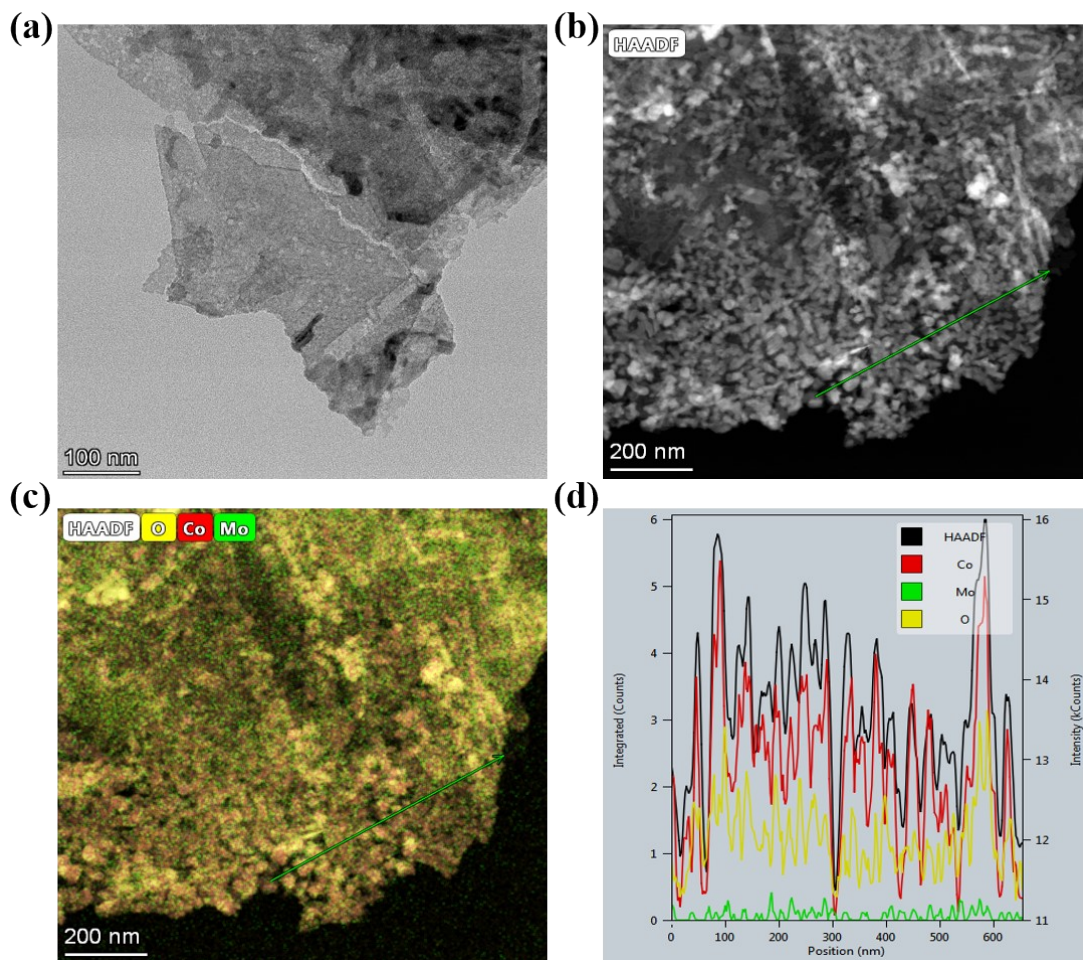
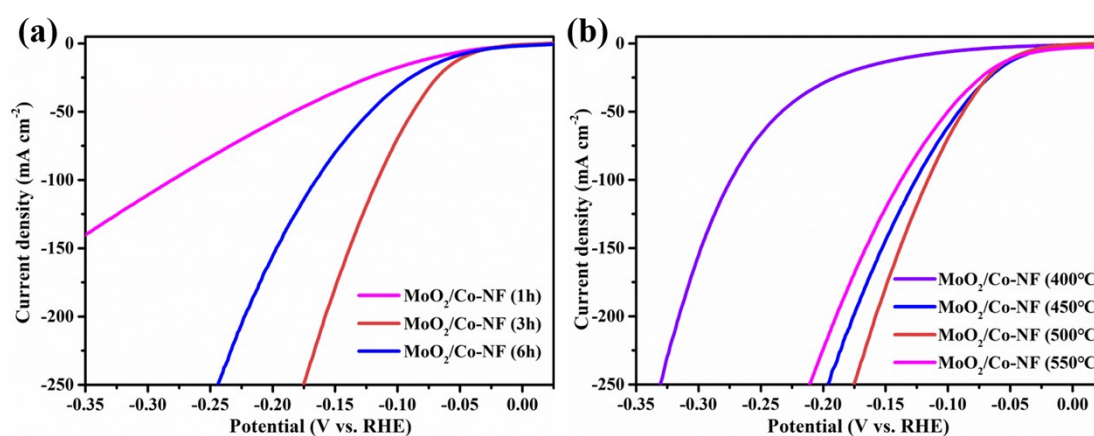


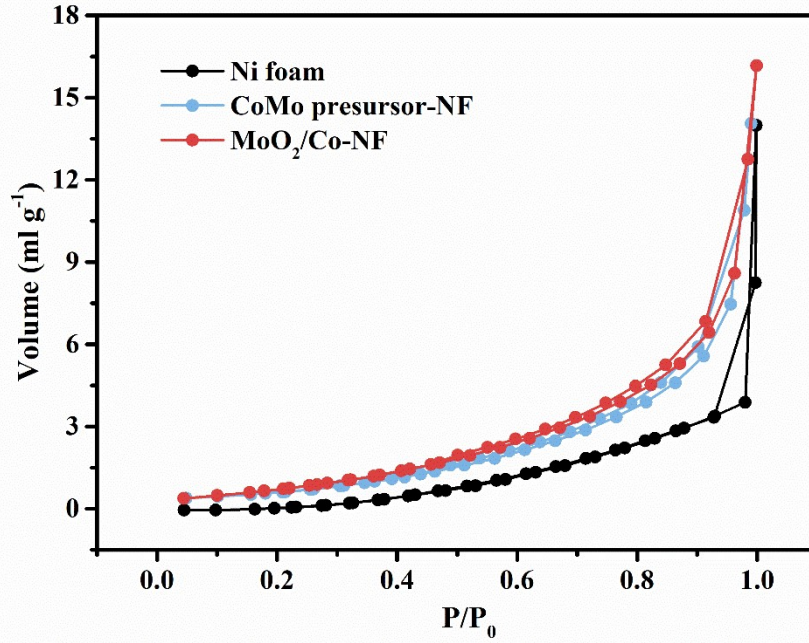
Figure. S5 Raman spectrum of MoO<sub>2</sub>/Co.



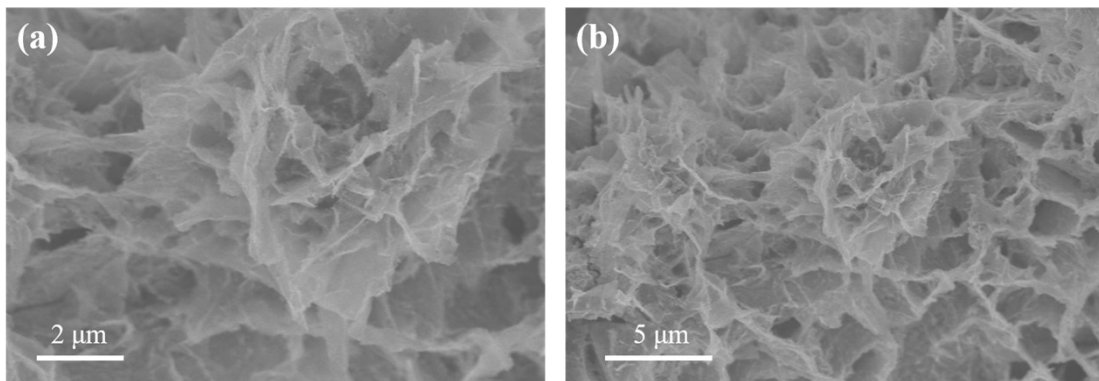
**Figure. S6** (a) TEM image of MoO<sub>2</sub>/Co; (b-d) EDS line scanning images of MoO<sub>2</sub>/Co.



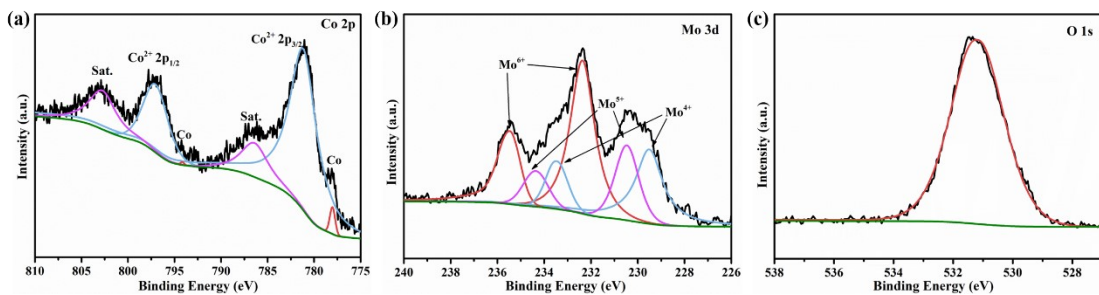
**Figure. S7** LSV curves of MoO<sub>2</sub>/Co-NF with different hydrothermal times (a) and different hydrogenation temperatures (b) in 1.0 M KOH solution.



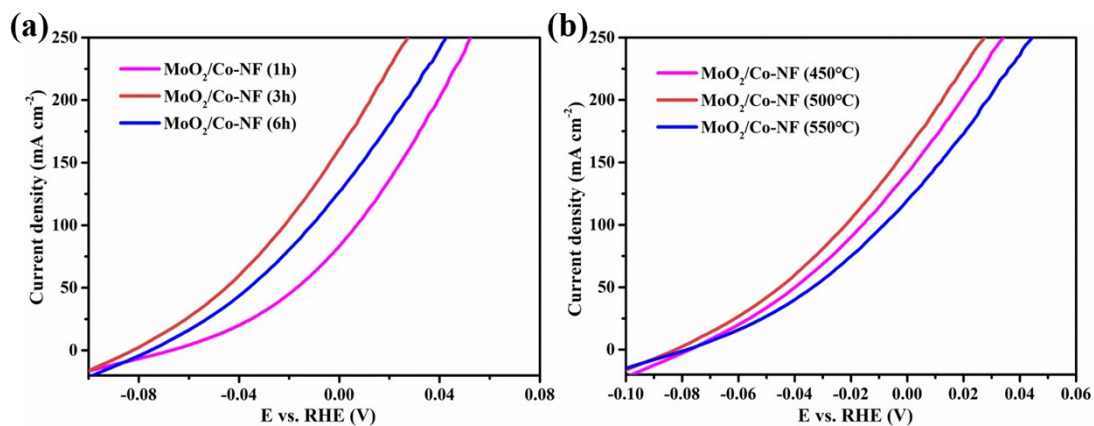
**Figure. S8** BET results of the Ni foam, CoMo precursor-NF and MoO<sub>2</sub>/Co-NF electrodes (sample: active material together with Ni foam substrate).



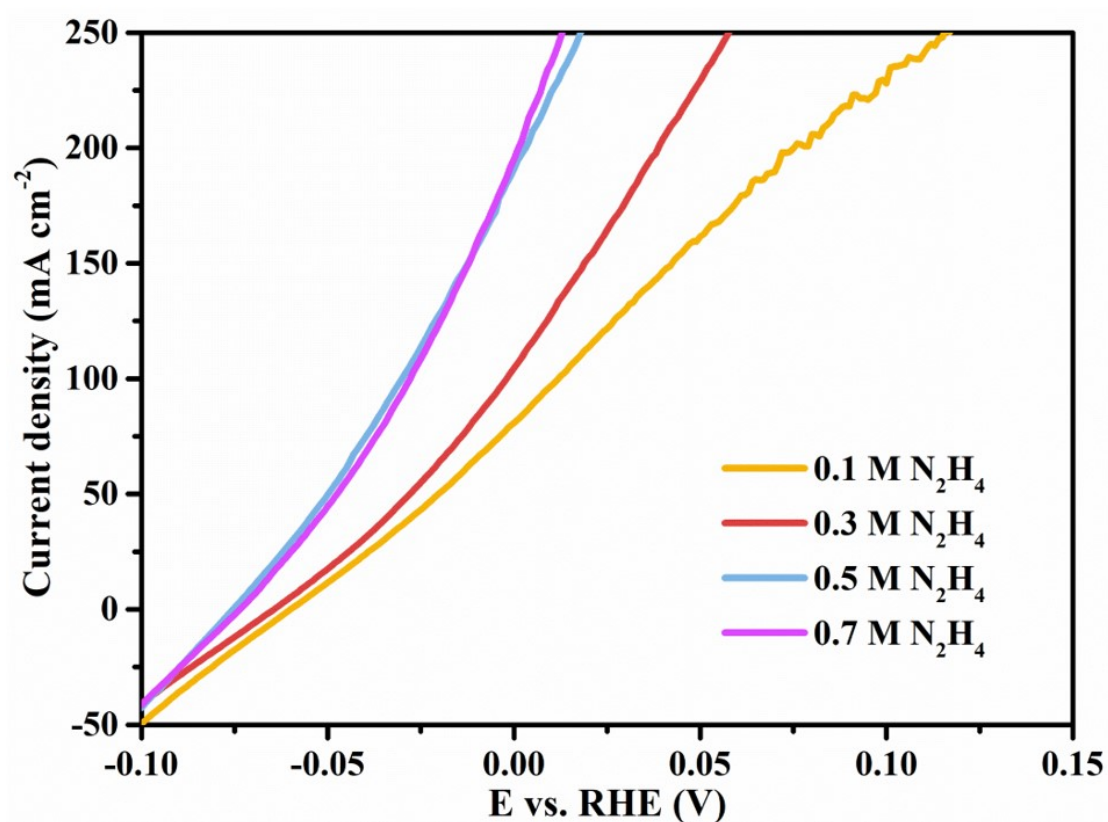
**Figure. S9** SEM images of MoO<sub>2</sub>/Co after 3000 cycles for HER.



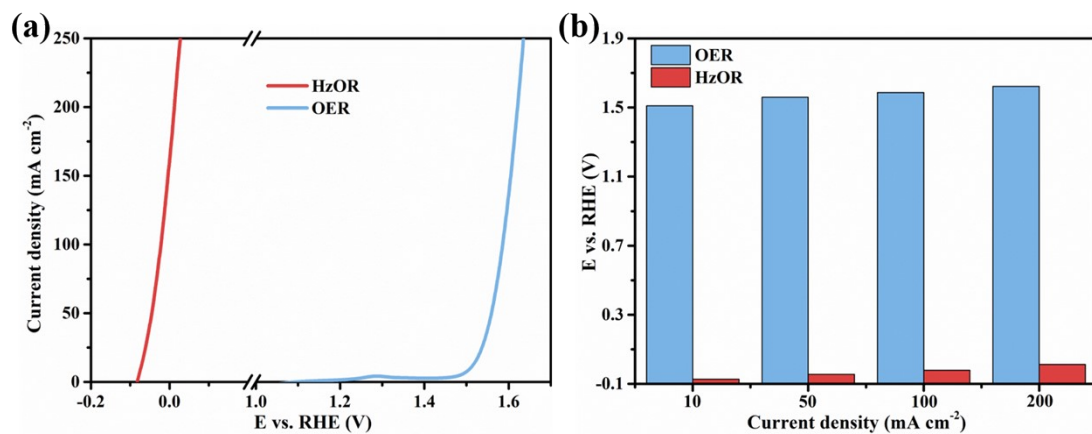
**Figure. S10** XPS survey spectra of MoO<sub>2</sub>/Co after 3000 cycles for HER.



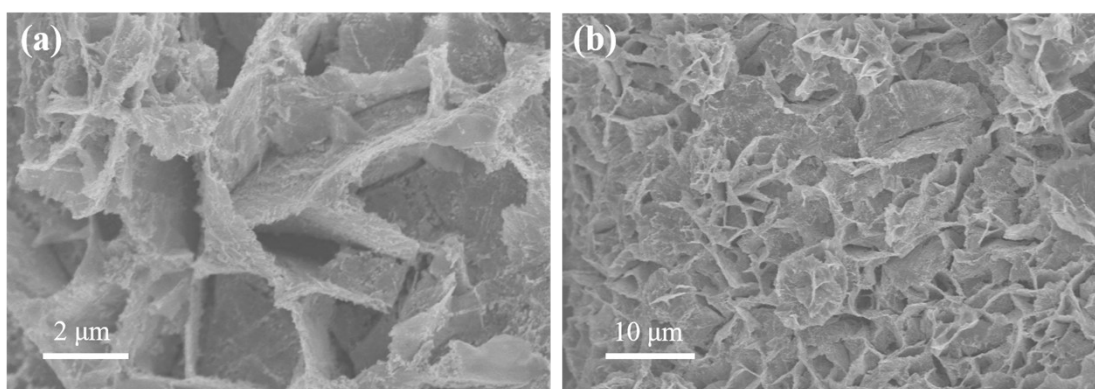
**Figure. S11** LSV curves of MoO<sub>2</sub>/Co-NF with different hydrothermal times (a) and different hydrogenation temperatures (b) in 1.0 M KOH with 0.5 M N<sub>2</sub>H<sub>4</sub> solution.



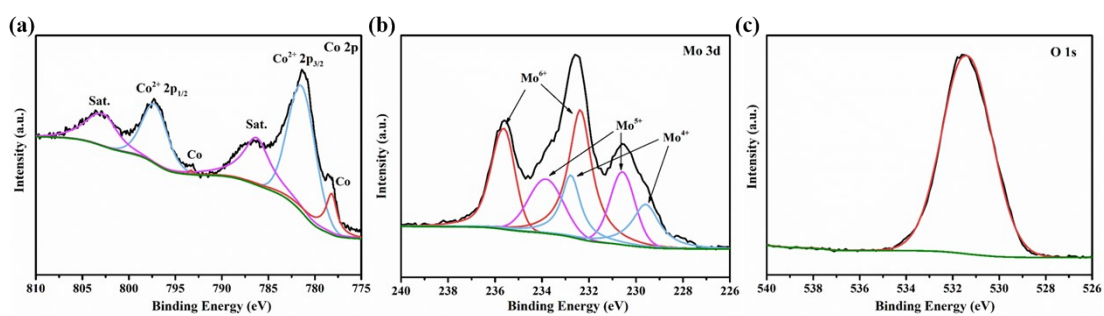
**Figure. S12** LSV curves of MoO<sub>2</sub>/Co-NF with different concentrations of N<sub>2</sub>H<sub>4</sub> in 1M KOH.



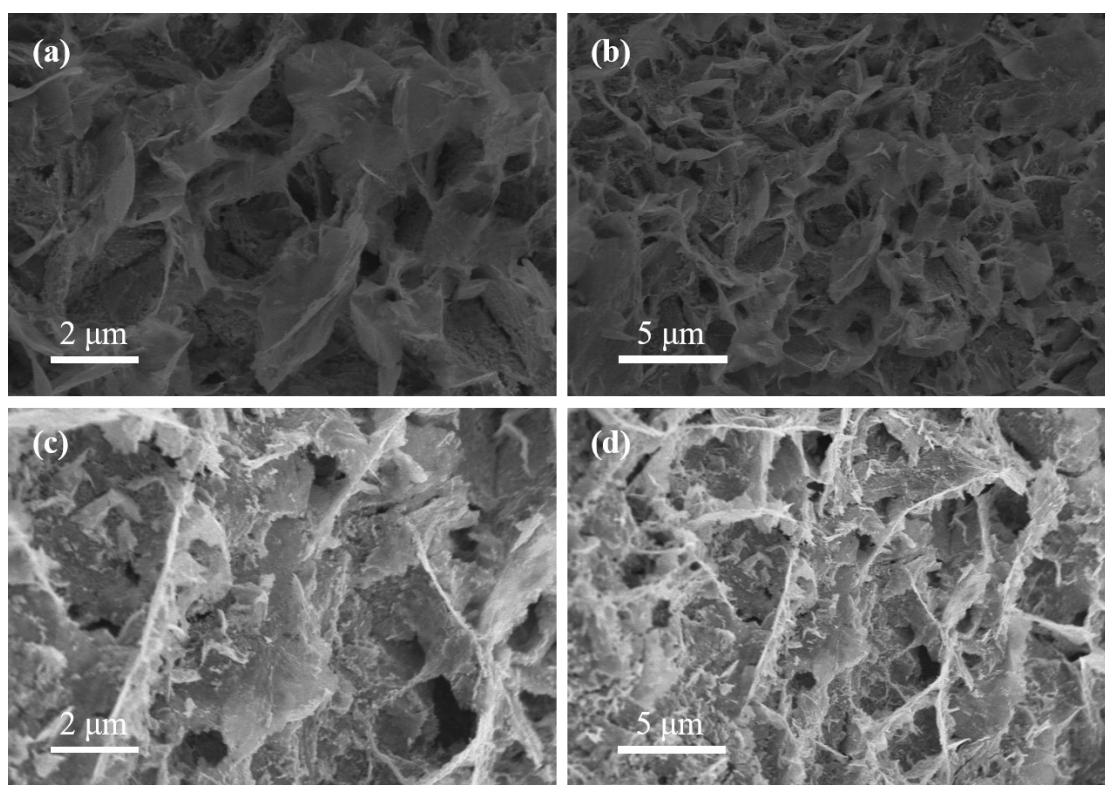
**Figure. S13** LSV curves (a) and potentials (b) of MoO<sub>2</sub>/Co-NF for OER and HzOR.



**Figure. S14** SEM images of MoO<sub>2</sub>/Co after 3000 cycles for HzOR.



**Figure. S15** XPS survey spectra of MoO<sub>2</sub>/Co after 3000 cycles for HzOR.



**Figure. S16** SEM images of MoO<sub>2</sub>/Co after 50 hours stability test for HER (a,b) and HzOR (c,d).

**Table. S1** The element Inductive Coupled Plasma (ICP) characterization after different hydrothermal time.

Sample	Co/Mo (at.)
MoO <sub>2</sub> /Co-1h	19.94:1
MoO <sub>2</sub> /Co-3h	5.558:1
MoO <sub>2</sub> /Co-6h	4.076:1



**Table. S2** Comparison of HER activity in 1 M KOH for MoO<sub>2</sub>/Co with other recently reported HER electrocatalysts.

Catalysts	$\eta_{10}$	Tafel slope (mV dec <sup>-1</sup> )	Reference
MoO <sub>2</sub> /Co-NF	48	49.5	This work
Co <sub>2</sub> P/CoN-in-NCNTs	98	57.0	[1]
Fe-CoS <sub>2</sub>	40	32	[2]
NiCoN/C	103	N/A	[3]
V-Co <sub>4</sub> N	37	44	[4]
PW-Co <sub>3</sub> N	41	40	[5]
CoSe <sub>2</sub>	79	84	[6]
Cr-Co <sub>4</sub> N	21	38.1	[7]
Co-Ni <sub>3</sub> N	194	156	[8]
Ni <sub>3</sub> N/MoO <sub>2</sub>	21	46	[9]
MoO <sub>2</sub> /Ni	50.48	65.52	[10]
MoO <sub>2</sub> -Ni NWs	58.4	36.6	[11]

**Table. S3** Comparison of the HzOR performance of MoO<sub>2</sub>/Co with other recently reported advanced catalysts. E<sub>10</sub> is the HzOR working potential at 10 mA cm<sup>-2</sup>.

Catalysts	Electrolyte	E <sub>10</sub> (mV)	Tafel slope (mV dec <sup>-1</sup> )	Reference
MoO <sub>2</sub> /Co-NF	1.0 M KOH+0.5 M N <sub>2</sub> H <sub>4</sub>	-73	22.9	This work
PW-Co <sub>3</sub> N	1.0 M KOH+0.1 M N <sub>2</sub> H <sub>4</sub>	-55	14	[5]
Fe-CoS <sub>2</sub>	1.0 M KOH+0.1 M N <sub>2</sub> H <sub>4</sub>	-5	48	[2]
CoSe <sub>2</sub>	1.0 M KOH+0.5 M N <sub>2</sub> H <sub>4</sub>	-17	N/A	[6]
Ni NCNAs	1.0 M KOH+0.3 M N <sub>2</sub> H <sub>4</sub>	-26	32.6	[12]
Mo-Ni <sub>3</sub> N/Ni/NF	1.0 M KOH+0.1 M N <sub>2</sub> H <sub>4</sub>	-0.3	48	[13]
Ni <sub>3</sub> N-Co <sub>3</sub> N	1.0 M KOH+0.1 M N <sub>2</sub> H <sub>4</sub>	-88	21.6	[14]
RP-CPM	1.0 M KOH+0.3 M N <sub>2</sub> H <sub>4</sub>	-70	47.6	[15]
Ni <sub>2</sub> P/NF	1.0 M KOH+0.5 M N <sub>2</sub> H <sub>4</sub>	-65	55	[16]
Ni-C HNSA	1.0 M KOH+0.1 M N <sub>2</sub> H <sub>4</sub>	-20	16.2	[17]

**Table. S4** Comparison of OHzS of MoO<sub>2</sub>/Co with other reported catalysts. E<sub>10</sub> is the OHzS working potential at 10 mA cm<sup>-2</sup>.

Catalysts	Electrolyte	E <sub>10</sub> (mV)	Stability	Reference
MoO <sub>2</sub> /Co-NF	1.0 M KOH+0.5 M N <sub>2</sub> H <sub>4</sub>	35	50 h	This work
PW-Co <sub>3</sub> N	1.0 M KOH+0.1 M N <sub>2</sub> H <sub>4</sub>	358	20 h	[5]
CoSe <sub>2</sub>	1.0 M KOH+0.5 M N <sub>2</sub> H <sub>4</sub>	164	14 h	[6]
Ni NCNAs	1.0 M KOH+0.3 M N <sub>2</sub> H <sub>4</sub>	79 (E <sub>50</sub> )	N/A	[12]
Mo-Ni <sub>3</sub> N/Ni/NF	1.0 M KOH+0.1 M N <sub>2</sub> H <sub>4</sub>	55	10 h	[13]
Ni <sub>3</sub> N-Co <sub>3</sub> N	1.0 M KOH+0.1 M N <sub>2</sub> H <sub>4</sub>	71	20 h	[14]
RP-CPM	1.0 M KOH+0.3 M N <sub>2</sub> H <sub>4</sub>	353	20 h	[15]
Ni <sub>2</sub> P/NF	1.0 M KOH+0.5 M N <sub>2</sub> H <sub>4</sub>	1.0V(E <sub>500</sub> )	10 h	[16]
Ni-C HNSA	1.0 M KOH+0.1 M N <sub>2</sub> H <sub>4</sub>	140 (E <sub>50</sub> )	30 h	[17]

## Reference

- [1] Y.Y. Guo, P.F. Yuan, J.A. Zhang, H.C. Xia, F.Y. Cheng, M.F. Zhou, J. Li, Y.Y. Qiao, S.C. Mu, Q. Xu, *Adv Funct Mater*, 28 (2018).
- [2] X.J. Liu, J. He, S.Z. Zhao, Y.P. Liu, Z. Zhao, J. Luo, G.Z. Hu, X.M. Sun, Y. Ding, *Nat Commun*, 9 (2018).
- [3] J.P. Lai, B.L. Huang, Y.G. Chao, X. Chen, S.J. Guo, *Adv Mater*, 31 (2019).
- [4] Z.Y. Chen, Y. Song, J.Y. Cai, X.S. Zheng, D.D. Han, Y.S. Wu, Y.P. Zang, S.W. Niu, Y. Liu, J.F. Zhu, X.J. Liu, G.M. Wang, *Angew Chem Int Edit*, 57 (2018) 5076-5080.
- [5] Y. Liu, J.H. Zhang, Y.P. Li, Q.Z. Qian, Z.Y. Li, Y. Zhu, G.Q. Zhang, *Nat Commun*, 11 (2020).
- [6] J. Qi, W. Zhang, R. Cao, *Adv Energy Mater*, 8 (2018).
- [7] N. Yao, P. Li, Z.R. Zhou, Y.M. Zhao, G.Z. Cheng, S.L. Chen, W. Luo, *Adv Energy Mater*, 9 (2019).
- [8] C.R. Zhu, A.L. Wang, W. Xiao, D.L. Chao, X. Zhang, N.H. Tiep, S. Chen, J.N. Kang, X. Wang, J. Ding, J. Wang, H. Zhang, H.J. Fan, *Adv Mater*, 30 (2018).
- [9] S.W. Niu, Y.Y. Fang, J.B. Zhou, J.Y. Cai, Y.P. Zang, Y.S. Wu, J. Ye, Y.F. Xie, Y. Liu, X.S. Zheng, W.G. Qu, X.J. Liu, G.M. Wang, Y.T. Qian, *J Mater Chem A*, 7 (2019) 10924-10929.
- [10] W.L. Liang, P.Y. Dong, Z.C. Le, X.Y. Lin, X.Y. Gong, F.Y. Xie, H. Zhang, J. Chen, N. Wang, Y.S. Jin, H. Meng, *Acs Appl Mater Inter*, 13 (2021) 39470-39479.
- [11] X. Liu, K. Ni, C.J. Niu, R.T. Guo, W. Xi, Z.Y. Wang, J.S. Meng, J.T. Li, Y.W. Zhu, P.J. Wu, Q. Li, J. Luo, X.J. Wu, L.Q. Mai, *Acs Catal*, 9 (2019) 2275-2285.
- [12] Z.B. Liang, T.J. Qiu, S. Gao, R.Q. Zhong, R.Q. Zou, *Adv Energy Mater*, (2021).
- [13] Y. Liu, J.H. Zhang, Y.P. Li, Q.Z. Qian, Z.Y. Li, G.Q. Zhang, *Adv Funct Mater*, 31 (2021).
- [14] Q.Z. Qian, J.H. Zhang, J.M. Li, Y.P. Li, X. Jin, Y. Zhu, Y. Liu, Z.Y. Li, A. El-Harairy, C. Xiao, G.Q. Zhang, Y. Xie, *Angew Chem Int Edit*, 60 (2021) 5984-5993.
- [15] Y.P. Li, J.H. Zhang, Y. Liu, Q.Z. Qian, Z.Y. Li, Y. Zhu, G.Q. Zhang, *Sci Adv*, 6

(2020).

[16] C. Tang, R. Zhang, W.B. Lu, Z. Wang, D.N. Liu, S. Hao, G. Du, A.M. Asiri, X.P. Sun, *Angew Chem Int Edit*, 56 (2017) 842-846.

[17] Y. Zhu, J.H. Zhang, Q.Z. Qian, Y.P. Li, Z.Y. Li, Y. Liu, C. Xiao, G.Q. Zhang, Y. Xie, *Angew Chem Int Edit*, (2021).