

SUPPORTING INFORMATION

Surface stability of ionic-liquid-passivated mixed-cation perovskite probed with in-situ photoelectron spectroscopy

Suresh Maniyarasu^a, Ben F. Spencer^{bc}, Hongbo Mo^b, Alex S. Walton^d, Andrew G. Thomas^{*bc} and Wendy R. Flavell^{*ac}

^aPhoton Science Institute, Department of Physics and Astronomy, School of Natural Sciences, The University of Manchester, M13 9PL, U.K.

^bPhoton Science Institute, Department of Materials, School of Natural Sciences, The University of Manchester, M13 9PL, U.K.

^cHenry Royce Institute, The University of Manchester, M13 9PL, U.K.

^dPhoton Science Institute, Department of Chemistry, School of Natural Sciences, The University of Manchester, M13 9PL, U.K.

*E-mail: wendy.flavell@manchester.ac.uk and andrew.g.thomas@manchester.ac.uk

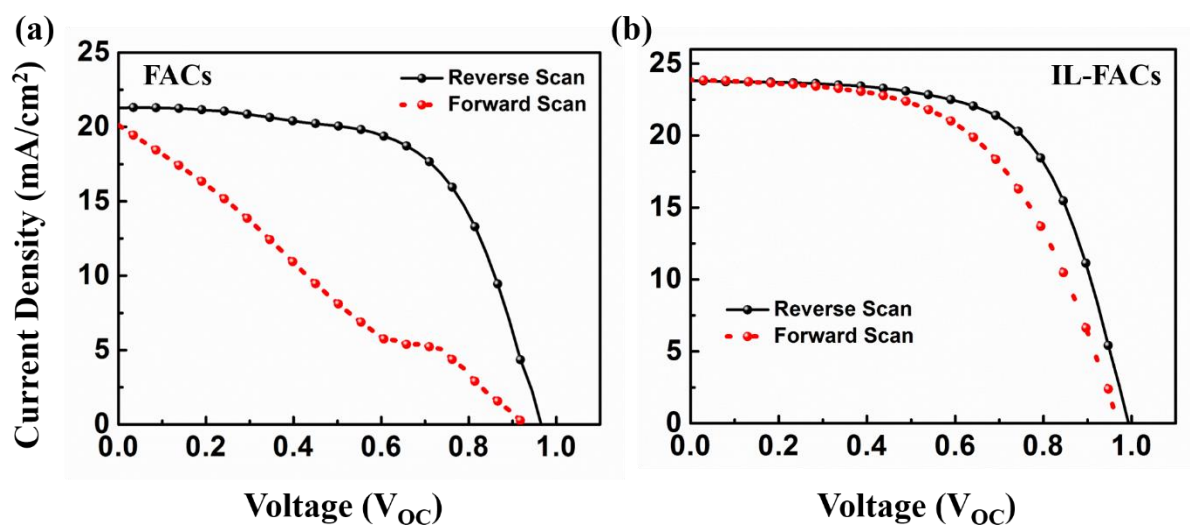


Figure. S1 (a, b) Reverse scan (black) and forward scan (red) J - V curves of devices fabricated with FACs and IL-FACs films, respectively.

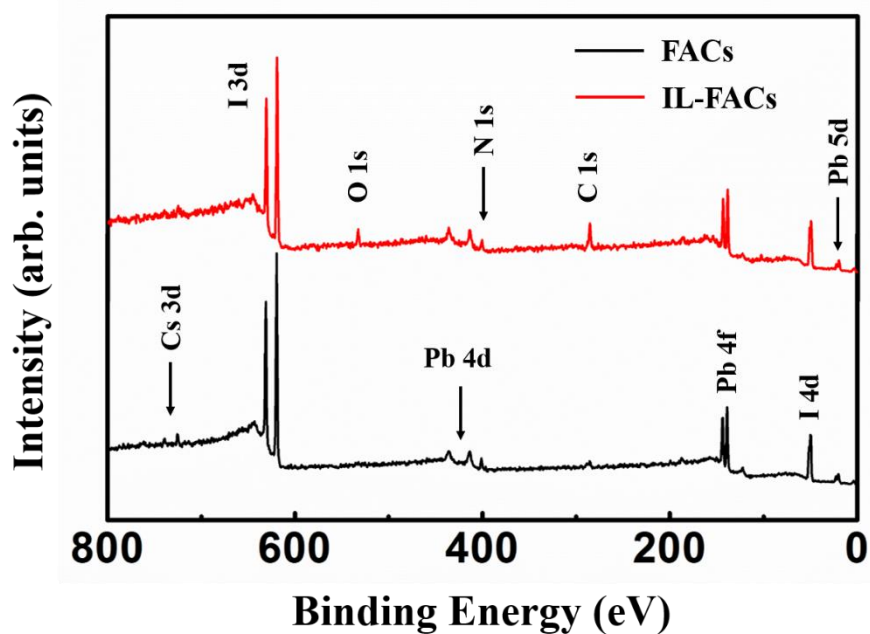


Figure S2. The overview spectra of FACs (black) and IL-FACs (red) measured using XPS at room temperature under UHV conditions.

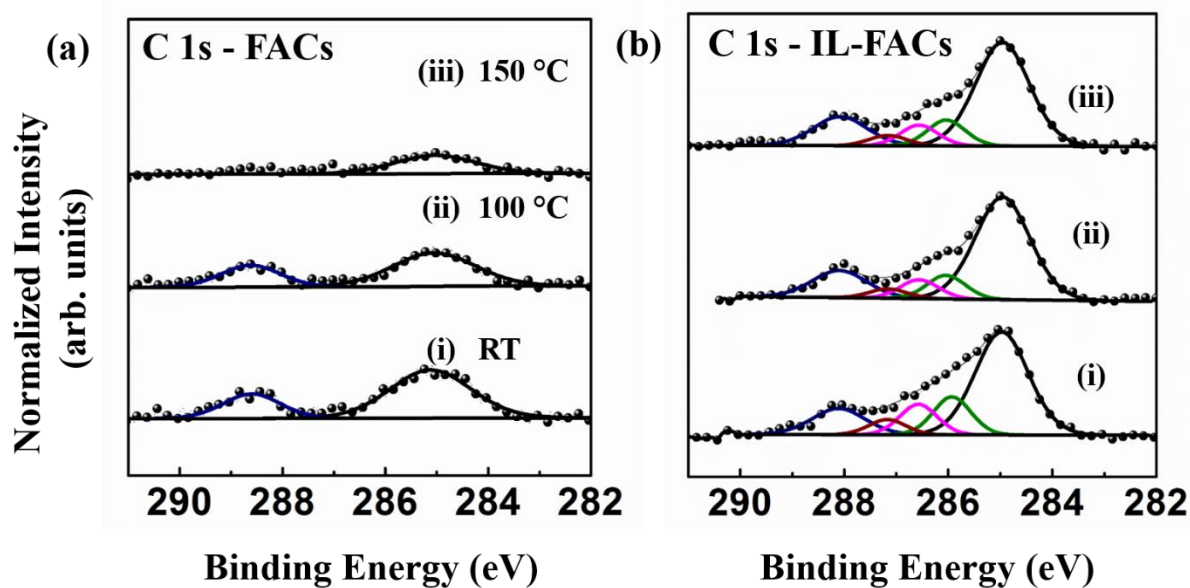


Figure S3. High-resolution C 1s spectra of FACs and IL-FACs measured under UHV conditions during thermal annealing. Spectra labelled as (i), (ii), and (iii) correspond to room temperature, 100 °C and 150 °C, respectively. Spectra have been normalised to the intensity of the Pb 4f_{7/2} signal from Pb²⁺ for each condition.

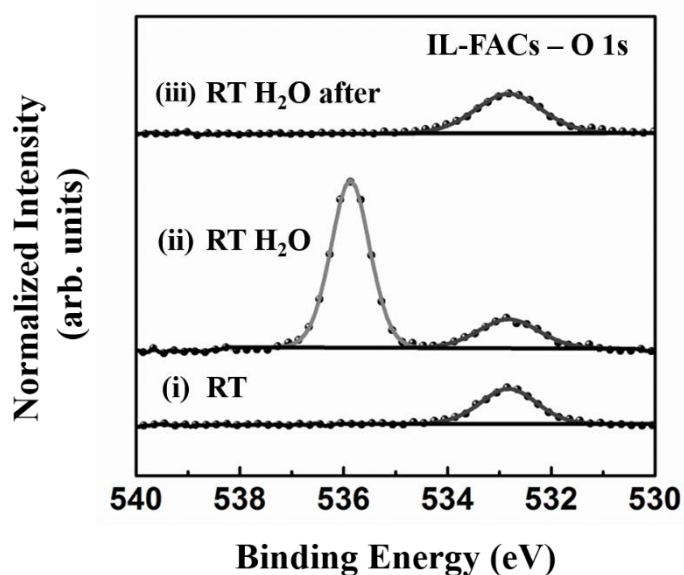


Figure S4. High-resolution O 1s spectra of IL-FACs measured at UHV conditions, during exposure and after exposure to 9 mbar water vapour at room temperature. Spectra labelled (i), (ii), and (iii) correspond to the sample measured before, during, and after exposure, respectively. Spectra are normalized to the intensity of Pb 4f_{7/2} core level from Pb²⁺ for each condition.

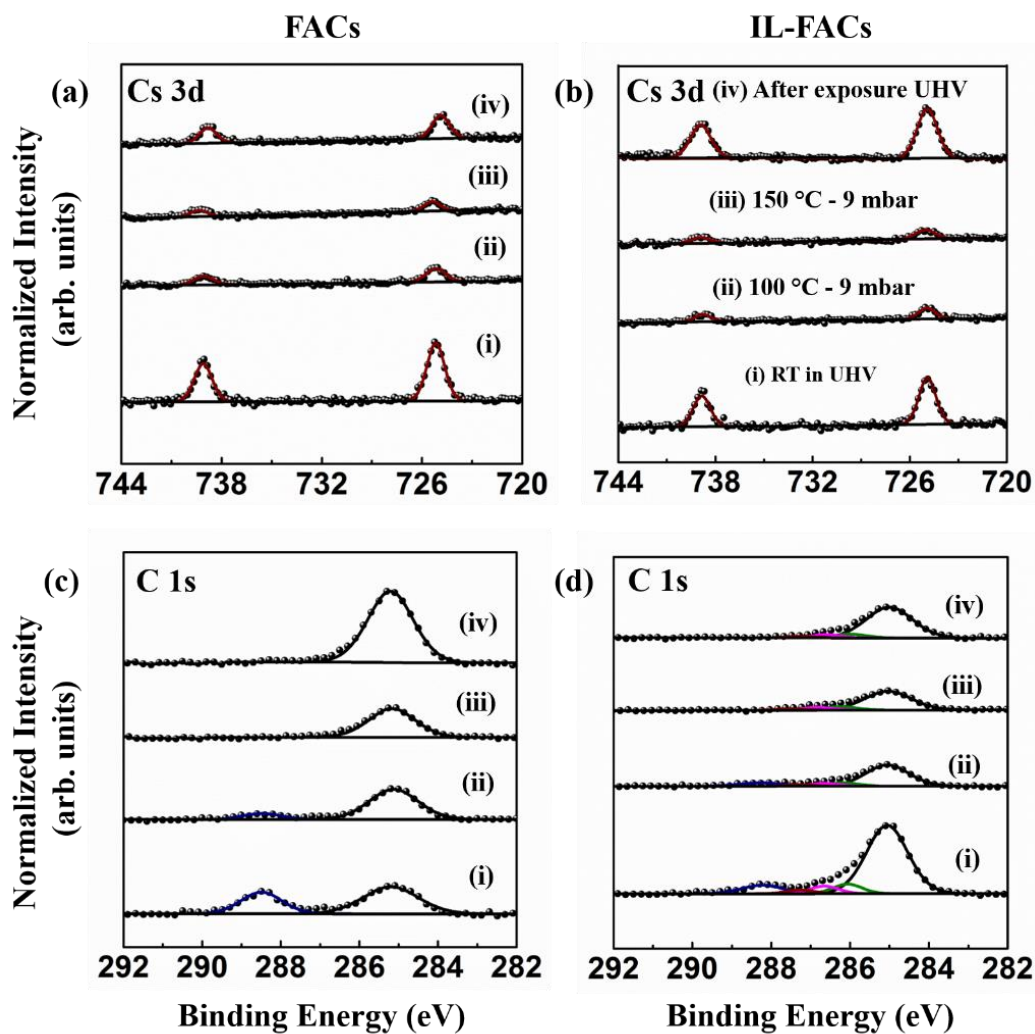


Figure S5. High-resolution core level NAPXPS spectra of (a,b) Cs 3d, (c,d) C 1s obtained from FACs (a,c) and IL-FACs (b,d) measured at different conditions as noted. Spectra labelled (i), (ii), (iii) and (iv) correspond to the sample measured under UHV conditions, at 100 °C in 9 mbar water vapour, 150 °C in 9 mbar water vapour and at room temperature in UHV conditions (after exposure), respectively. For each condition, all spectra are normalized to the intensity of the Pb 4f_{7/2} core level from Pb²⁺.

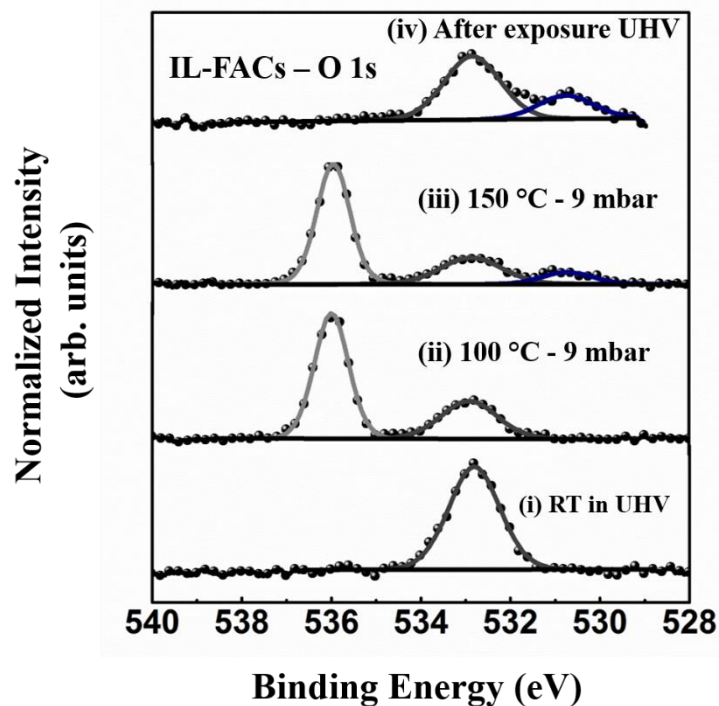


Figure S6. High-resolution O 1s spectra of IL-FACs measured at different conditions as noted. Spectra labelled (i), (ii), (iii) and (iv) correspond to the sample measured under UHV conditions, at 100 °C in 9 mbar water vapour, 150 °C in 9 mbar water vapour and at room temperature in UHV conditions (after exposure), respectively. Spectra are normalized to the intensity of Pb 4f_{7/2} core level from Pb²⁺ for each condition.

Table S1. Elemental stoichiometry calculated from XPS (and normalised to $[\text{Pb}^{2+}]=1.0$, i.e. I/Pb^{2+} , N/Pb^{2+} , C/Pb^{2+} , Cs/Pb^{2+} , and $\text{Pb}^0/\text{Pb}^{2+}$) of the FACs¹ and IL-FACs sample as a function of temperature.

Sample conditions	Pb²⁺	I	N (FA)	N (IL)	C-N(FA)	C-N (IL)	C-H	Cs	Metallic Pb⁰
Nominal	1.0	3.0	1.8		0.9			0.1	0
FACs pristine	1.0 ± 0.1	2.7 ± 0.2	1.2 ± 0.2	-	0.7 ± 0.2	-	1.9 ± 0.2	0.07 ± 0.02	0.07 ± 0.02
FACs 100 °C	1.0 ± 0.1	2.6 ± 0.2	1.0 ± 0.2	-	0.6 ± 0.2	-	1.3 ± 0.2	0.05 ± 0.02	0.11 ± 0.02
FACs 150 °C	1.0 ± 0.1	2.1 ± 0.2	0.4 ± 0.2	-	0.2 ± 0.2	-	0.7 ± 0.2	0.04 ± 0.02	0.17 ± 0.02
IL-FACs pristine	1.0 ± 0.1	3.2 ± 0.2	1.4 ± 0.2	0.5 ± 0.1	0.8 ± 0.2	1.6 ± 0.2	3.2 ± 0.2	0.12 ± 0.02	-
IL-FACs 100 °C	1.0 ± 0.1	3.5 ± 0.2	1.6 ± 0.2	0.5 ± 0.1	0.9 ± 0.2	1.4 ± 0.2	3.9 ± 0.2	0.08 ± 0.02	-
IL-FACs 100 °C	1.0 ± 0.1	3.5 ± 0.2	1.6 ± 0.2	0.5 ± 0.1	0.9 ± 0.2	1.4 ± 0.2	3.8 ± 0.2	0.05 ± 0.02	0.02

Table S2. Elemental stoichiometry calculated from XPS (and normalised to $[\text{Pb}^{2+}]=1.0$, *i.e.* I/Pb^{2+} , N/Pb^{2+} , C/Pb^{2+} , Cs/Pb^{2+} , and $\text{Pb}^0/\text{Pb}^{2+}$) of the FACs¹ and IL-FACs samples calculated from XPS before and after exposure to 9 mbar water vapour.

Sample conditions	Pb²⁺	I	O 1s	N (FA)	N (IL)	C-N(FA)	C-N (IL)	C-H	Cs	Metallic Pb⁰
Nominal	1.0	3.0		1.8		0.9			0.1	0
FACs before exposure	1.0 ± 0.1	2.6 ± 0.2	0.6 ± 0.1	1.2 ± 0.2	-	0.8 ± 0.2	-	2.3 ± 0.2	0.04 ± 0.03	0.06 ± 0.02
FACs after exposure	1.0 ± 0.1	2.4 ± 0.2	0.6 ± 0.1	1.0 ± 0.2	-	0.7 ± 0.2	-	3.6 ± 0.2	0.01 ± 0.03	0.06 ± 0.02
IL-FACs before exposure	1.0 ± 0.1	2.7 ± 0.2	1.5 ± 0.1	1.5 ± 0.2	0.5 ± 0.1	0.9 ± 0.2	1.0 ± 0.2	5.9 ± 0.2	0.1 ± 0.02	0.02 ± 0.02
IL-FACs after exposure	1.0 ± 0.1	2.6 ± 0.2	2.1 ± 0.1	1.5 ± 0.2	0.4 ± 0.1	0.9 ± 0.2	1.0 ± 0.2	5.8 ± 0.2	0.03 ± 0.02	0.02 ± 0.02

Table S3. Elemental stoichiometry calculated from XPS (and normalised to $[\text{Pb}^{2+}]=1.0$, *i.e.* I/Pb^{2+} , N/Pb^{2+} , C/Pb^{2+} , Cs/Pb^{2+} , and $\text{Pb}^0/\text{Pb}^{2+}$) of the FACs and IL-FACs samples calculated from NAPXPS before and after annealing to 100 and 150 °C in 9 mbar water vapour.

Sample conditions	Pb²⁺	I	N (FA)	N (IL)	C-N(FA)	C-N (IL)	C-H	Cs	Metallic Pb⁰
Nominal	1.0	3.0	1.8	0.9				0.1	0
FACs before exposure	1.0 ± 0.1	2.8 ± 0.2	1.8 ± 0.2	-	0.9 ± 0.2	-	1.7 ± 0.2	0.08 ± 0.02	0.1 ± 0.02
FACs after exposure and annealing	1.0 ± 0.1	1.7 ± 0.2	0.1 ± 0.2	-	0 ± 0.2	-	3.2 ± 0.2	0.03 ± 0.02	0.1 ± 0.02
IL-FACs before exposure	1.0 ± 0.1	2.7 ± 0.2	1.4 ± 0.2	0.4 ± 0.1	0.9 ± 0.2	1.1 ± 0.2	7.0 ± 0.2	0.09 ± 0.02	0.1 ± 0.02
IL-FACs after exposure and annealing	1.0 ± 0.1	2.0 ± 0.2	0.2 ± 0.2	0.2 ± 0.1	0.0 ± 0.2	0.8 ± 0.2	4.2 ± 0.2	0.12 ± 0.02	0.2 ± 0.02

Table S4. Stoichiometry ratio of O 1s calculated from XPS (and normalised to $[\text{Pb}^{2+}]=1.0$, *i.e.* O/ Pb^{2+}) of the FACs and IL-FACs samples calculated from XPS before and after annealing to 100 and 150 °C in 9 mbar water

Sample Conditions	O_{ads}	O_{lattice}
FACs before exposure	0.4 ± 0.1	
FACs after exposure and annealing	0.4 ± 0.1	0.2 ± 0.1
IL-FACs before exposure	1.8 ± 0.1	
IL-FACs after exposure and annealing	1.4 ± 0.1	0.5 ± 0.1

Reference

- 1 S. Maniyarasu, J. C. R. Ke, B. F. Spencer, A. S. Walton, A. G. Thomas and W. R. Flavell, *ACS Appl. Mater. Interfaces*, 2021, **13(36)**, 43573-43586.