

## Electronic Supplementary Information (ESI)

# Local Structural Distortions and Reduced Thermal Conductivity in Ge-Substituted Chalcopyrite

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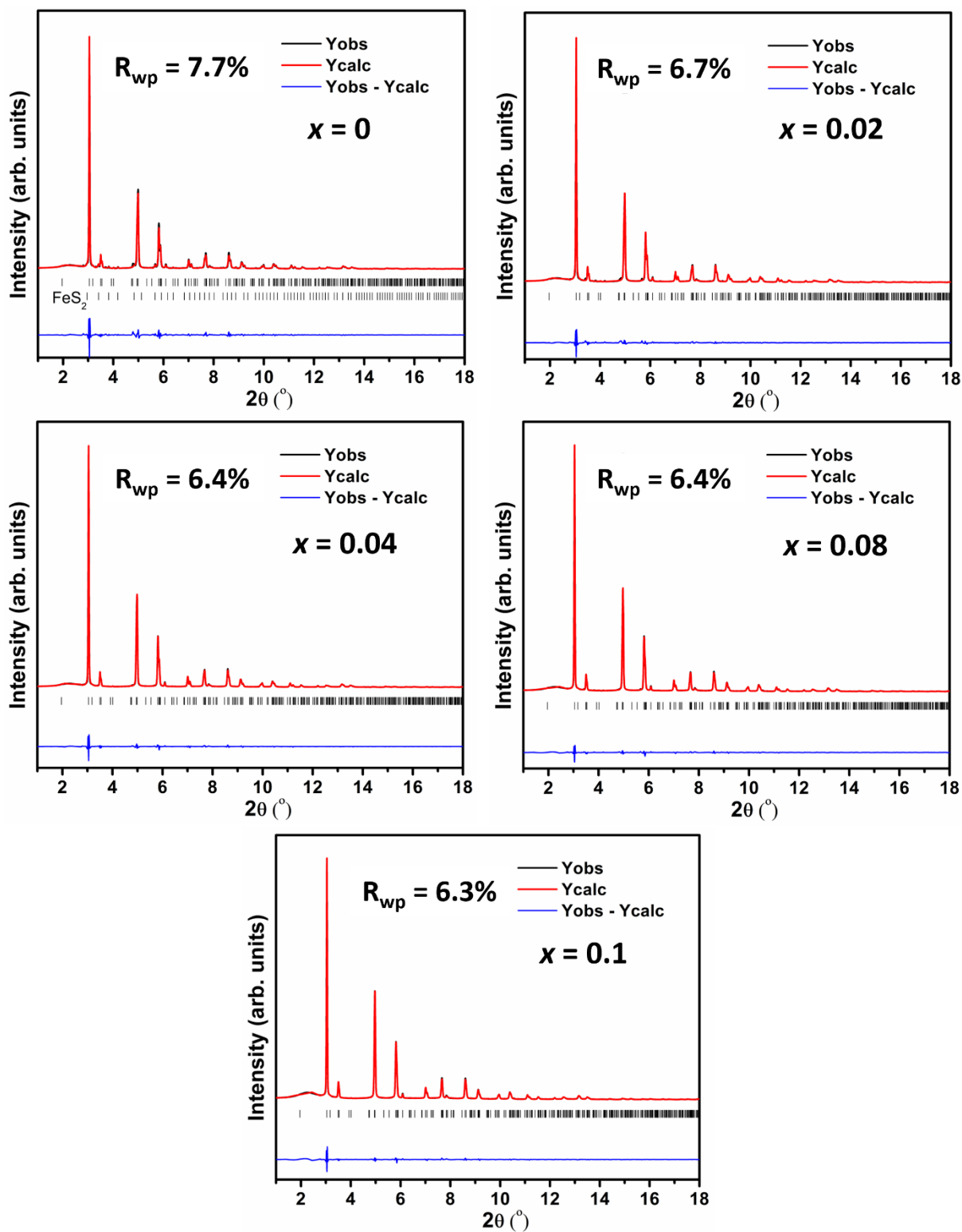
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**Figure S1.** Rietveld refinement using synchrotron X-ray powder diffraction data ( $\lambda = 0.161669 \text{ \AA}$ ) for  $\text{CuFe}_{1-x}\text{Ge}_x\text{S}_2$  ( $0 \leq x \leq 0.1$ ).

**Table S1.** Lattice parameters of  $\text{CuFe}_{1-x}\text{Ge}_x\text{S}_2$  ( $0 \leq x \leq 0.1$ ) described in the space group  $I\bar{4}2d$ .

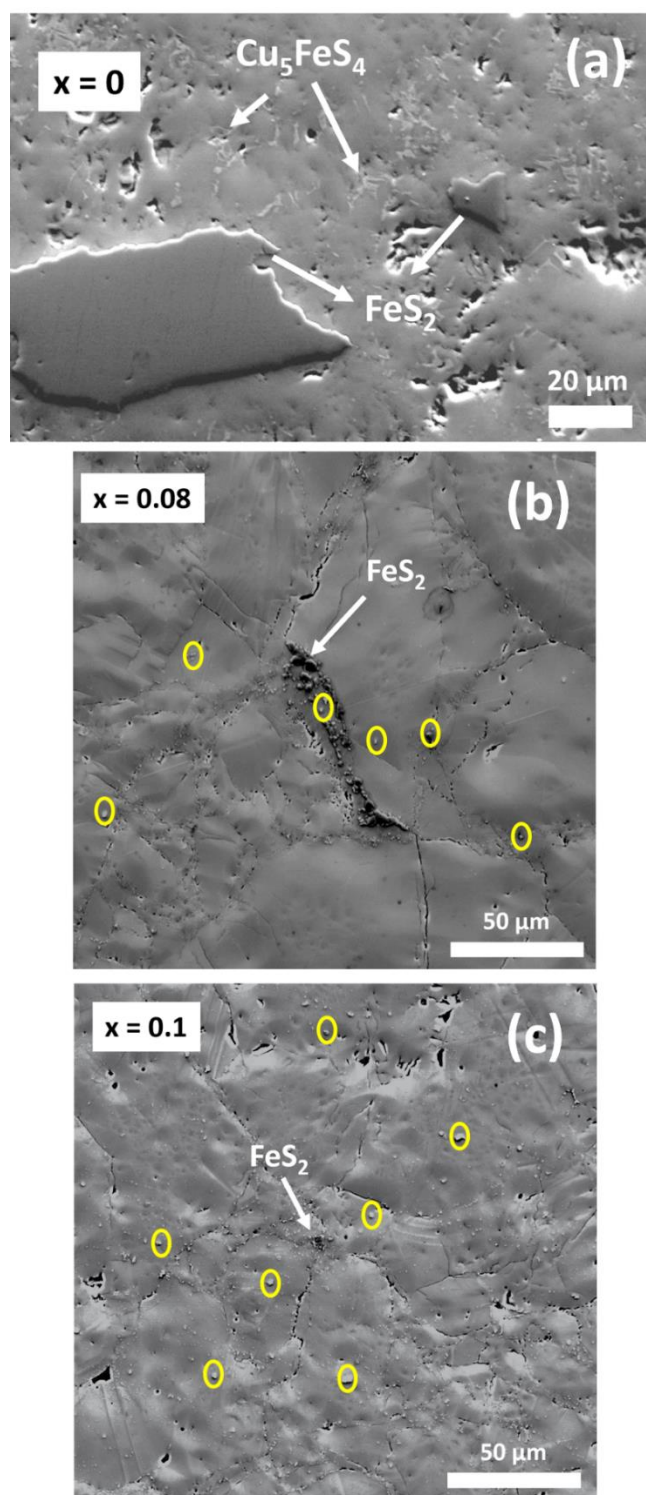
Sample	$a/\text{\AA}$	$c/\text{\AA}$
$\text{CuFeS}_2$	5.2909(3)	10.4319(7)
$\text{CuFe}_{0.98}\text{Ge}_{0.02}\text{S}_2$	5.2912(2)	10.4374(6)
$\text{CuFe}_{0.96}\text{Ge}_{0.04}\text{S}_2$	5.2913(2)	10.4529(5)
$\text{CuFe}_{0.94}\text{Ge}_{0.06}\text{S}_2$	5.2918(2)	10.4709(6)
$\text{CuFe}_{0.92}\text{Ge}_{0.08}\text{S}_2$	5.2913(2)	10.4790(6)
$\text{CuFe}_{0.9}\text{Ge}_{0.1}\text{S}_2$	5.2903(3)	10.4911(7)

**Table S2:** Compositions<sup>#</sup> of the chalcopyrite-type phase determined by EDS.

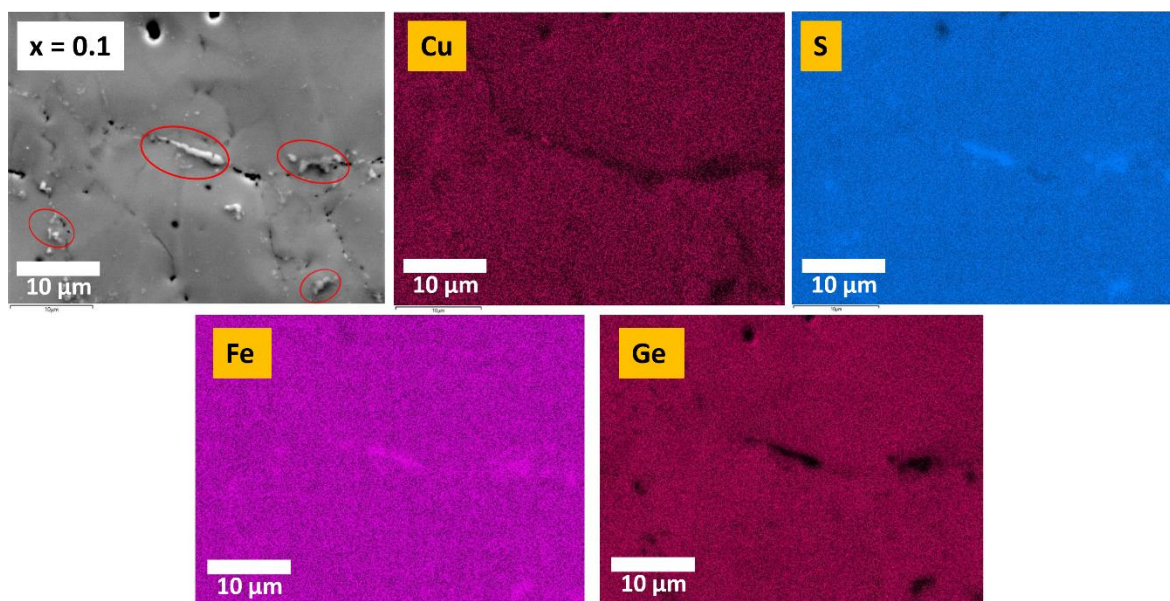
Sample (Ge concentration)	Nominal composition	EDS composition of the main phase*
$x = 0$	$\text{CuFeS}_2$	$\text{Cu}_{1.117(9)}\text{Fe}_{1.018(5)}\text{S}_2$
$x = 0.02$	$\text{CuFe}_{0.98}\text{Ge}_{0.02}\text{S}_2$	$\text{Cu}_{1.146(5)}\text{Fe}_{0.984(5)}\text{Ge}_{0.024(3)}\text{S}_2$
$x = 0.04$	$\text{CuFe}_{0.96}\text{Ge}_{0.04}\text{S}_2$	$\text{Cu}_{1.146(8)}\text{Fe}_{0.971(9)}\text{Ge}_{0.042(4)}\text{S}_2$
$x = 0.06$	$\text{CuFe}_{0.94}\text{Ge}_{0.06}\text{S}_2$	$\text{Cu}_{1.118(4)}\text{Fe}_{0.957(5)}\text{Ge}_{0.062(3)}\text{S}_2$
$x = 0.08$	$\text{CuFe}_{0.92}\text{Ge}_{0.08}\text{S}_2$	$\text{Cu}_{1.102(2)}\text{Fe}_{0.936(9)}\text{Ge}_{0.078(3)}\text{S}_2$
$x = 0.1$	$\text{CuFe}_{0.9}\text{Ge}_{0.1}\text{S}_2$	$\text{Cu}_{1.098(3)}\text{Fe}_{0.912(2)}\text{Ge}_{0.094(2)}\text{S}_2$

\*Normalized to 2 sulfur atoms per formula unit in  $\text{CuFeS}_2$ .

<sup>#</sup>Due to the overlap of the  $K_\alpha$  and  $L_\alpha$  characteristic lines of Cu and Fe, there may be uncertainties in the quantitative at.% determined by EDS.



**Figure S2.** SEM images of  $\text{CuFe}_{1-x}\text{Ge}_x\text{S}_2$  ( $x = 0, 0.08$  and  $0.1$ ) showing the presence of trace amounts of secondary phases. Substituted samples with higher Ge content ( $x \geq 0.08$ ) contain another chalcopyrite-like phase (yellow circles) identified from EDS as  $\text{Cu}_{21.69}\text{Fe}_{25.74}\text{Ge}_{0.94}\text{S}_{51.64}$  and  $\text{Cu}_{21.91}\text{Fe}_{25.46}\text{Ge}_{1.83}\text{S}_{50.79}$  in materials with compositions  $x = 0.08$  and  $0.1$ , respectively.



**Figure S3.** SEM-EDS elemental mapping of  $\text{CuFe}_{0.9}\text{Ge}_{0.1}\text{S}_2$  showing the presence of copper-poor chalcopyrite-like secondary phase.

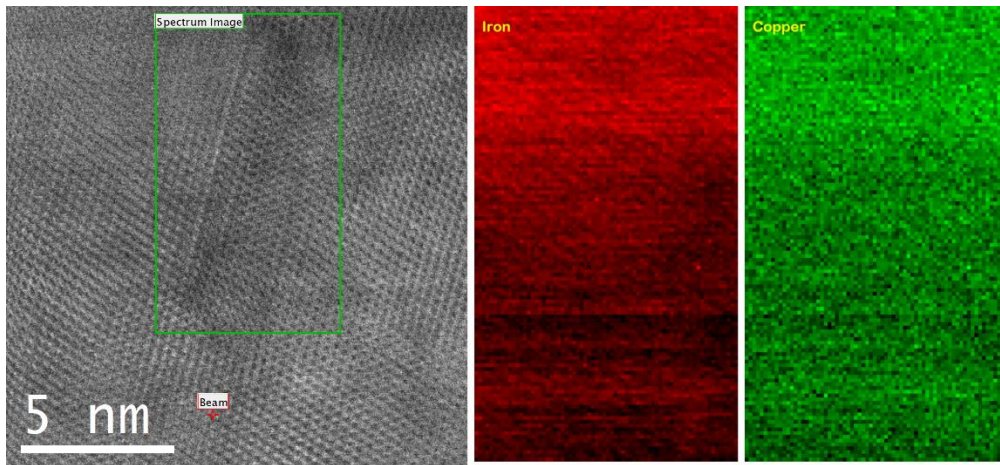
**Table S3.** XPS binding energies and corresponding oxidation states of individual elements in Ge-substituted  $\text{CuFe}_{0.94}\text{Ge}_{0.06}\text{S}_2$ .

Element	Peak	B.E.(eV)*	Oxidation state
Cu	2p <sub>3/2</sub>	932.3	+1
	2p <sub>1/2</sub>	952.1	+1
Fe	2p <sub>3/2</sub>	710.2	+3
	2p <sub>1/2</sub>	723.7	+3
	2p	707.9	+2
S	2p <sub>3/2</sub>	161.4	-2
	2p <sub>1/2</sub>	162.6	-2
Ge	3d <sub>5/2</sub>	31.2	+4
	3d <sub>3/2</sub>	31.8	+4
	3d <sub>5/2</sub>	30.5	+2
	3d <sub>3/2</sub>	31.1	+2

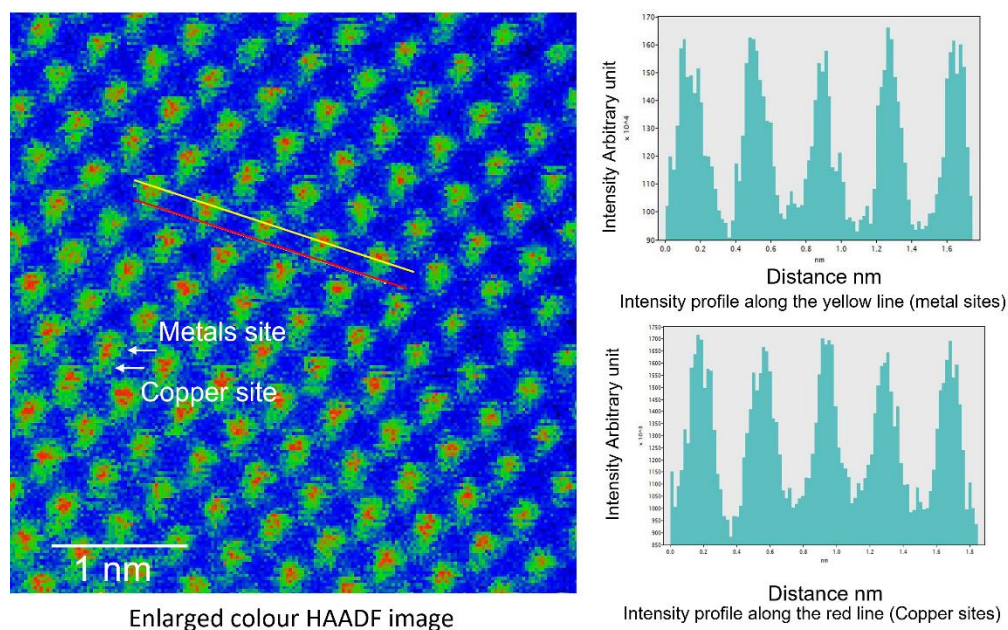
\*The binding energy of the XPS peaks is assigned on the basis of data in the literature<sup>1-5</sup> and in the NIST database.<sup>6</sup>

**Table S4.** Hall coefficient data for  $\text{CuFe}_{1-x}\text{Ge}_x\text{S}_2$  ( $0 \leq x \leq 0.1$ ) at room temperature.

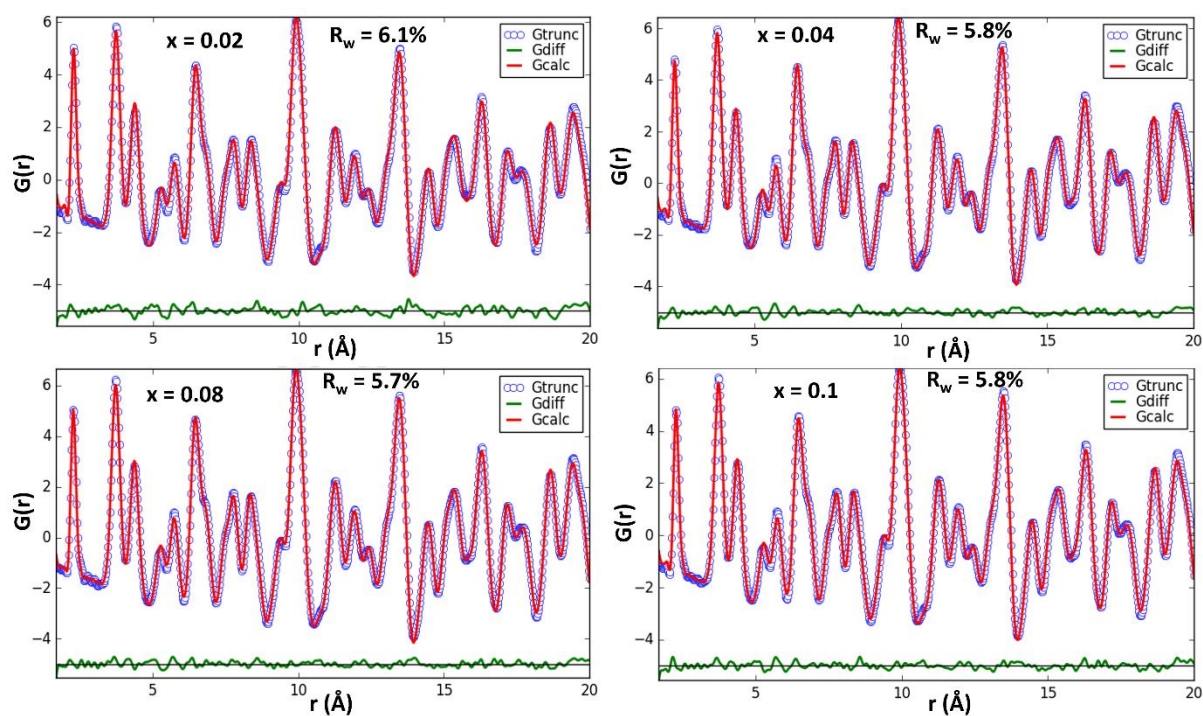
Sample	Charge carrier concentration ( $\times 10^{19} \text{ cm}^{-3}$ )	Charge carrier mobility ( $\text{cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ )
$\text{CuFeS}_2$	1.4(2)	15(3)
$\text{CuFe}_{0.98}\text{Ge}_{0.02}\text{S}_2$	2.3(2)	11(1)
$\text{CuFe}_{0.96}\text{Ge}_{0.04}\text{S}_2$	3.1(3)	9(2)
$\text{CuFe}_{0.94}\text{Ge}_{0.06}\text{S}_2$	4.7(3)	8(2)
$\text{CuFe}_{0.92}\text{Ge}_{0.08}\text{S}_2$	2.8(4)	11(1)
$\text{CuFe}_{0.9}\text{Ge}_{0.1}\text{S}_2$	1.9(2)	15(2)



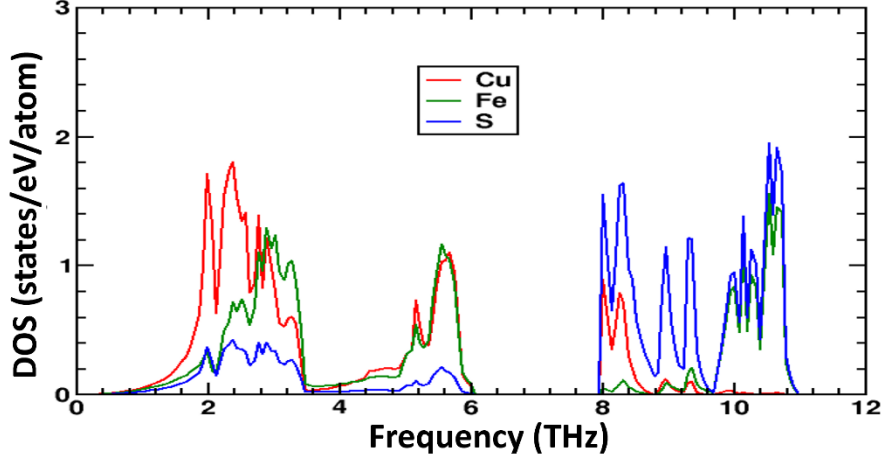
**Figure S4.** STEM image illustrating the chemical homogeneity of the cations in the defect regions in  $\text{CuFe}_{0.94}\text{Ge}_{0.06}\text{S}_2$ .



**Figure S5.** Enlarged HAADF-STEM image showing the uniform distribution of cations in  $\text{CuFe}_{0.94}\text{Ge}_{0.06}\text{S}_2$ .



**Figure S6.** Experimental, calculated and difference pair distribution function,  $G(r)$  as a function of atomic-pair distance ( $r$ ) for  $\text{CuFe}_{1-x}\text{Ge}_x\text{S}_2$  ( $0.02 \leq x \leq 0.1$ ).



**Figure S7.** Phonon density of states (DOS) of CuFeS<sub>2</sub>.

**Table S5.** Phonon group velocities of the LA, LA' and TO modes in CuFeS<sub>2</sub> and in CuFeS<sub>2</sub> with 6.25 at.% of Ge substitution.

	CuFeS <sub>2</sub> (m s <sup>-1</sup> )	Ge-substituted CuFeS <sub>2</sub> (m s <sup>-1</sup> )
Γ – X	2307, 2911, 5127	2470, 2553, 4723
Γ – Z	2009, 2921, 5271	2300, 2585, 5142
Γ – N	2352, 5354	2597, 4888

### Calculation Details:

- **Lorenz number and electronic thermal conductivity:**

The electronic part of the thermal conductivity ( $\kappa_e$ ) was calculated from the Wiedemann-Franz relation:

$$\kappa_e = L\sigma T \quad (\text{S1})$$

$L$  is the temperature-dependent Lorenz number and  $T$  is the temperature. The temperature-dependent Lorenz number was evaluated from the following relation:

$$L = \left(\frac{k_B}{e}\right)^2 \left( \frac{\left(\frac{r+7}{2}\right)F_{r+5/2}(\eta)}{\left(\frac{r+3}{2}\right)F_{r+1/2}(\eta)} - \left[ \frac{\left(\frac{r+5}{2}\right)F_{r+3/2}(\eta)}{\left(\frac{r+3}{2}\right)F_{r+1/2}(\eta)} \right]^2 \right) \quad (\text{S2})$$



Where  $k_B$  is the Boltzmann's constant,  $\eta$  is the reduced Fermi energy that is obtained from Seebeck coefficient values via the relation:

$$S = \pm \frac{k_B}{e} \left( \frac{\left(\frac{r+5}{2}\right) F_{r+3/2}(\eta)}{\left(\frac{r+3}{2}\right) F_{r+1/2}(\eta)} - \eta \right) \quad (\text{S3})$$

Here,  $F(\eta)$  is the reduced Fermi integral given by:

$$F_n(\eta) = \int_0^\infty \frac{x^n}{1+e^{x-\eta}} dx \quad (\text{S4})$$

And  $\eta = E_F/k_B T$  where  $E_F$  denotes the Fermi level. Assuming that the main scattering mechanism is acoustic phonon scattering, the value of  $r$  is taken as  $-1/2$ . The Lorenz number at each temperature value is therefore obtained by substituting  $\eta$  and  $r$  in equation (S2).

- **Mean sound velocity ( $v_m$ ):**

$$v_m = \left[ \frac{1}{3} \left( \frac{2}{v_t^3} + \frac{1}{v_l^3} \right) \right]^{-\frac{1}{3}}$$

- and **average sound velocity ( $v_{\text{avg}}$ ):**

$$v_{\text{avg}} = (2v_t + v_l)/3$$

where,  $v_l$  and  $v_t$  are the longitudinal and transverse sound velocities, respectively.

- **Shear modulus ( $G$ ):**

$$G = d v_t^2$$

Where  $d$  and  $v_t$  are the density and transverse velocities respectively.

- **Young's modulus ( $E$ ):**

$$E = \frac{d v_t^2 (3v_l^2 - 4v_t^2)}{(v_l^2 - v_t^2)}$$

- **Debye temperature ( $\theta_D$ ):**

$$\theta_D = \frac{h}{k_B} \left( \frac{3N}{4\pi V} \right)^{1/3} v_m$$

where  $h$  is the Plank's constant,  $k_B$  is the Boltzmann's constant,  $N$  is the number of atoms in the unit cell,  $V$  is the volume of the unit cell and  $v_m$  is the mean sound velocity.

- **Distortion Parameters:**

Bond distortion parameter:

$$\gamma_x = \frac{1}{4} \sum_{i=1}^4 \left[ \frac{\langle x_i \rangle - x_i}{\langle x_i \rangle} \right]^2$$

where  $\langle x_i \rangle$  and  $x_i$  denote the mean and individual Ge-S bond lengths in the GeS<sub>4</sub> tetrahedra.

Angle distortion parameter:

$$\gamma_\theta = \frac{1}{6} \sum_{i=1}^6 \left[ \frac{\langle \theta_i \rangle - \theta_i}{\langle \theta_i \rangle} \right]^2$$

where  $\langle \theta_i \rangle$  and  $\theta_i$  denote the mean and individual S-Ge-S bond angles in the GeS<sub>4</sub> tetrahedra.

- **Scattering Parameters:**

The point defect parameter ( $A$ ) can be written as:

$$A = \frac{\Omega_o}{4\pi v_m^3} \Gamma$$

where,  $\Omega_o$  is volume of the primitive unit cell, and  $\Gamma = \Gamma_M$  (mass-difference fluctuation) +  $\Gamma_S$  (strain-field fluctuation) given by:

$$\Gamma_M = \frac{\sum_{i=1}^n c_i \left(\frac{M_i}{M}\right)^2 f_i^2 f_i^2 \left(\frac{M_i^1 - M_i^2}{M_i}\right)^2}{\sum_{i=1}^n c_i}$$

and,

$$\Gamma_S = \frac{\sum_{i=1}^n c_i \left(\frac{M_i}{M}\right)^2 f_i^2 f_i^2 \varepsilon \left(\frac{r_i^1 - r_i^2}{r_i}\right)^2}{\sum_{i=1}^n c_i}$$

where,  $\overline{M}_l = \sum_k f_i^k M_i^k$  ;  $\overline{r}_l = \sum_k f_i^k r_i^k$  ;  $\overline{M} = \frac{\sum_{i=1}^n c_i \overline{M}_l}{\sum_{i=1}^n c_i}$ . Here  $n$  is the number of crystallographic sub-lattices which is 3 for CuFeS<sub>2</sub> and  $c_i$  is the degeneracy of each site in the primitive unit cell ( $c_1 = c_2 = 1$ ,  $c_3 = 2$  corresponding to Cu, Fe and S sites respectively).  $\overline{M}_l$  and  $\overline{r}_l$  denote the average mass and radius of the atoms on the  $l^{\text{th}}$  sublattice, respectively.  $M_i^k$  and  $f_i^k$  are the atomic mass and fractional occupation of the  $k^{\text{th}}$  atom on the  $i^{\text{th}}$  sublattice, respectively.  $\overline{M}$  is the total average atomic mass of the compound.  $\varepsilon$  is a phenomenological adjustable parameter that was evaluated from fitting Equation 1 to the experimental  $\kappa_L$ . For CuFe<sub>1-x</sub>Ge<sub>x</sub>S<sub>2</sub> samples, on the basis of germanium substitution occurring at the iron site, the total scattering parameter can be, thus, written as:

$$\Gamma = \frac{1}{4} \left( \frac{\overline{M}_{Fe}}{\overline{M}} \right)^2 x(1-x) \left[ \left( \frac{M_{Fe}^{Fe} - M_{Fe}^{Ge}}{\overline{M}_{Fe}} \right)^2 + \varepsilon \left( \frac{r_{Fe}^{Fe} - r_{Fe}^{Ge}}{\overline{r}_{Fe}} \right)^2 \right]$$

#### References:

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