

Supplementary Information

Multifunctional, Superhydrophobic and Highly Elastic

MXene/Bacterial Cellulose Hybrid Aerogels Enabled via Silylation

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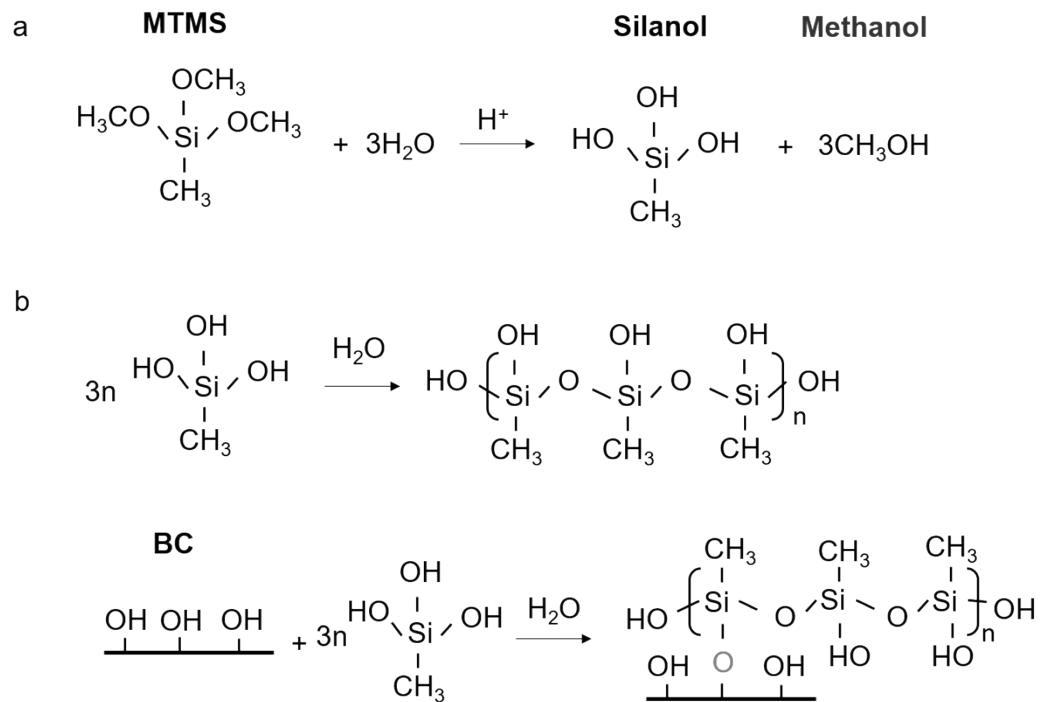


Figure S1. Schematic illustration of the reaction of MTMS in aqueous BC suspension.

(a) Chemical reaction between MTMS and H₂O. (b) self-polymerization among unreacted silanols and *in situ* polymerization between unreacted silanols and hydroxyl groups.

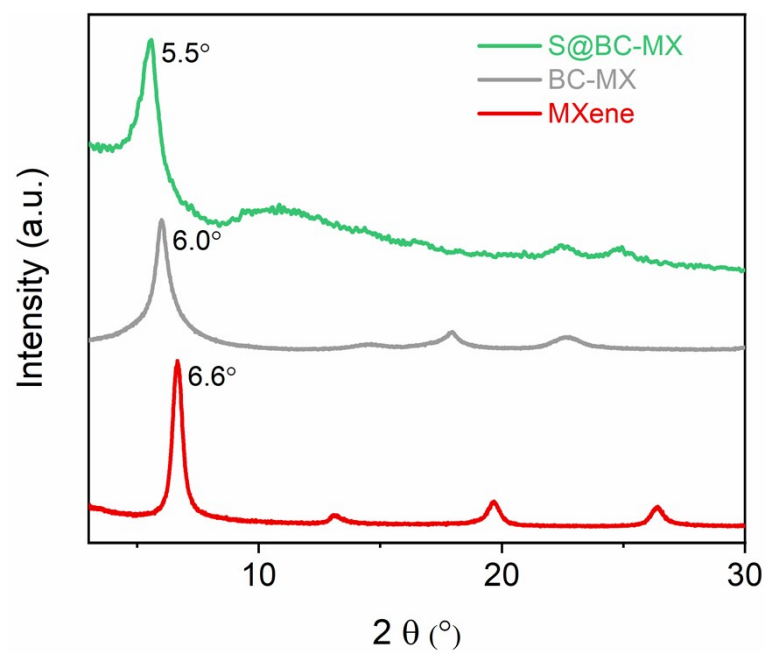


Figure S2. XRD patterns of pure MXene, BC-MX50%, and S@BC-MX50% aerogels.

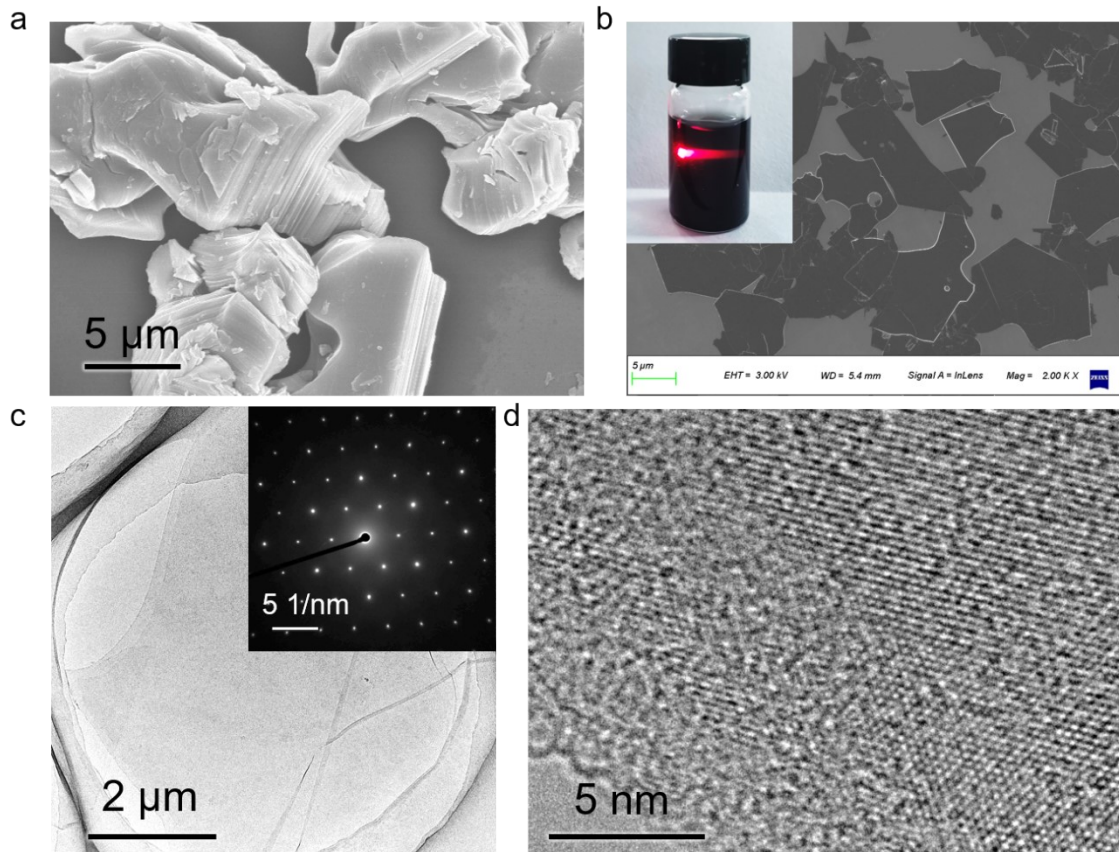


Figure S3. (a) SEM image of Ti_3AlC_2 powder showing the compact layered structure. (b) SEM image of $\text{Ti}_3\text{C}_2\text{T}_x$ nanosheets and inserted optical image of the MXene aqueous dispersion showing a typical Tyndall scattering effect. (c) TEM image and inserted electron diffraction of single-layer $\text{Ti}_3\text{C}_2\text{T}_x$. (d) HRTEM image of MXene nanosheets.

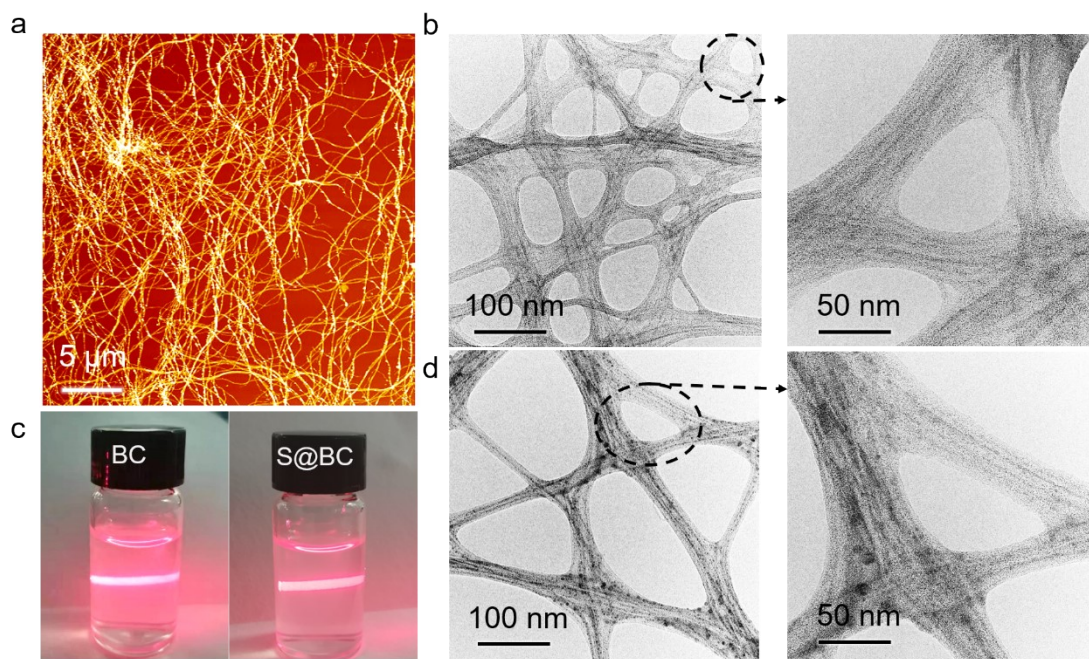


Figure S4. Morphologies of BC and Silyated BC. (a) AFM image of BC showing a quite high aspect ratio. (b) TEM images of BC networks. (c) Stable colloidal suspension of BC and silylated BC. (d) TEM images of silylated BC networks.

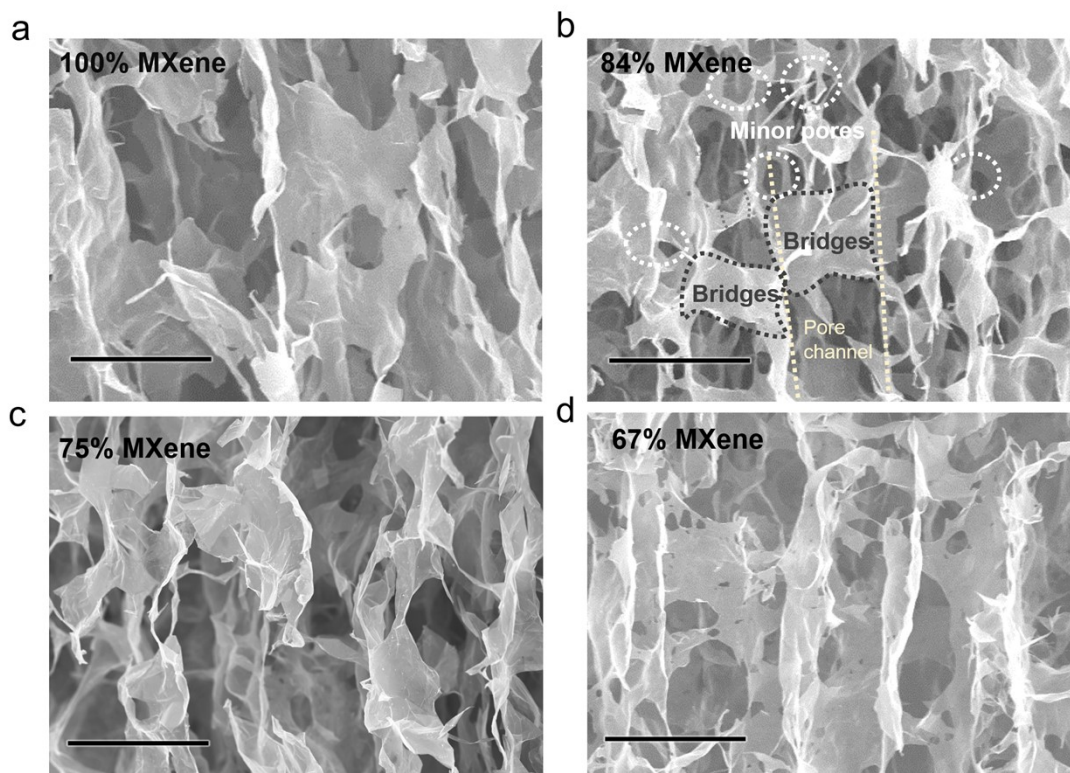


Figure S5. Cross-sectional SEM images of (a) pure MXene aerogel and (b-d) S@BC-MX84%, S@BC-MX75%, and S@BC-MX67% aerogels. Scale bar is 50 μm and applies to all images.

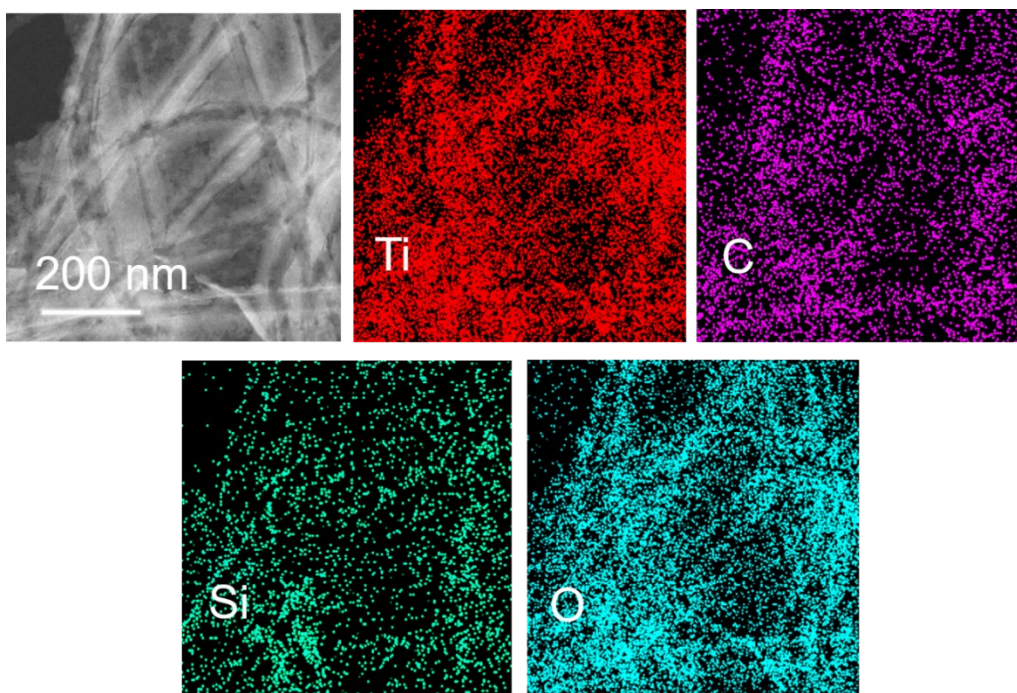


Figure S6. HADF-STEM image of S@BC-MX50% aerogel and corresponding EDS mapping images of Ti, C, Si, and O.

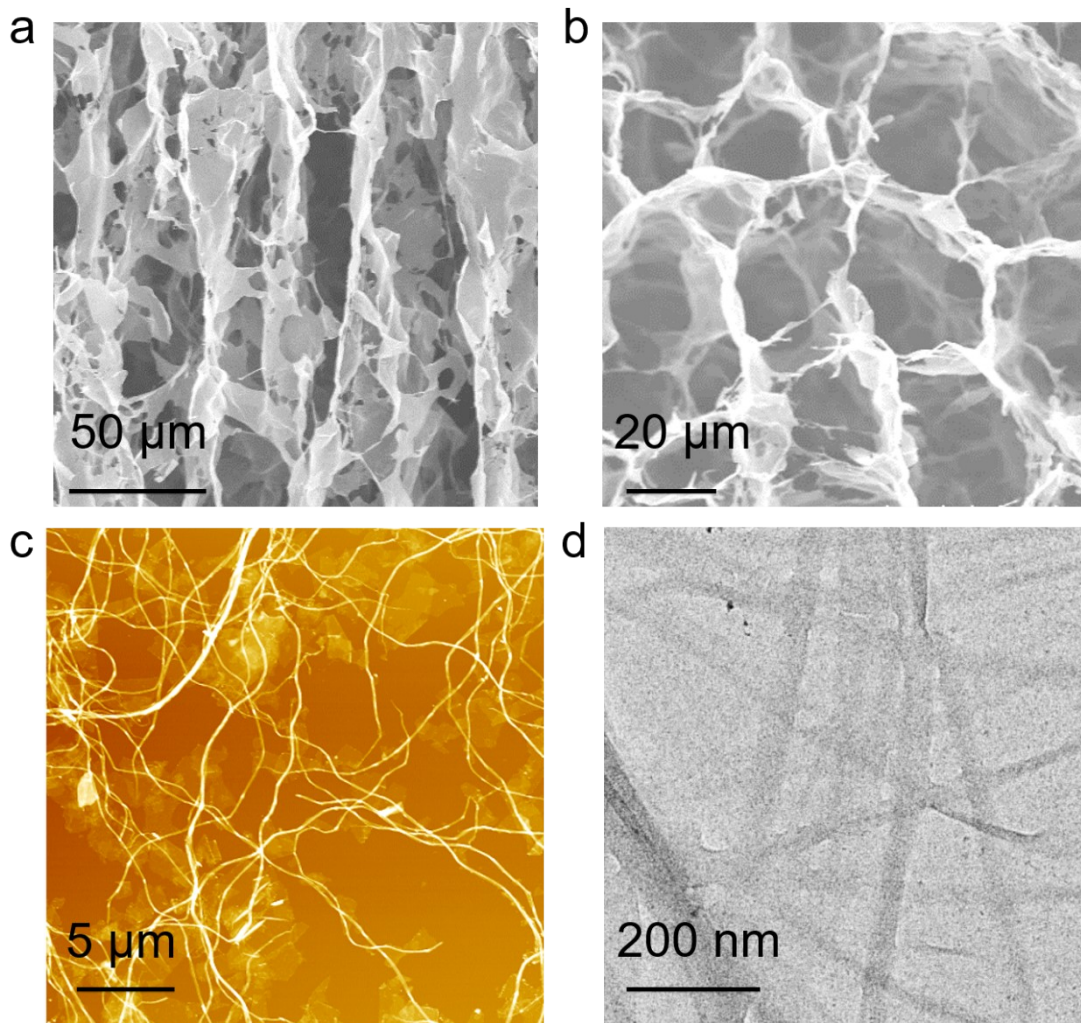


Figure S7 Morphologies of BC-MX50% hybrid aerogel. (a) cross-sectional SEM image. (b) Top-view SEM image. (c) AFM image of MXene nanosheets assembled with BC nanofibers. (d) TEM image of cell wall.

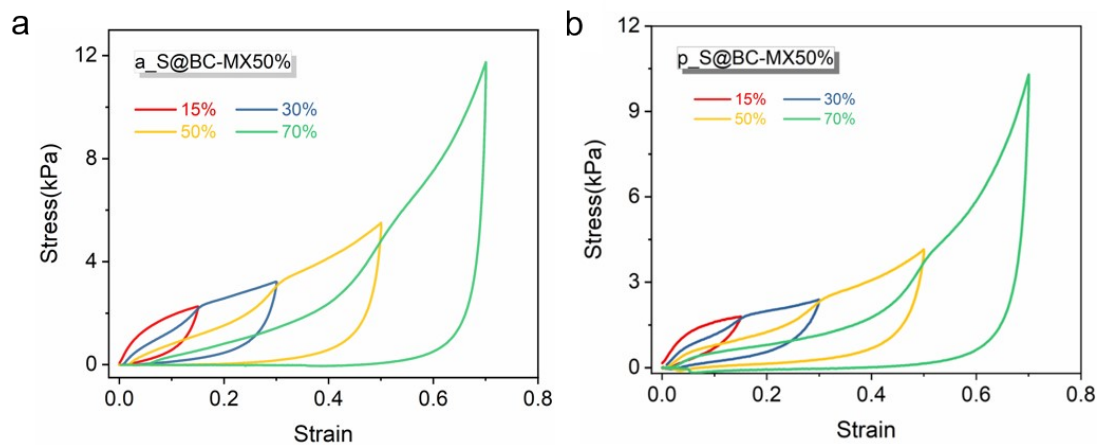


Figure S8. Stress-strain curves of S@BC-MX50% aerogel compressed along (a) “a” and (b) “p_” directions.

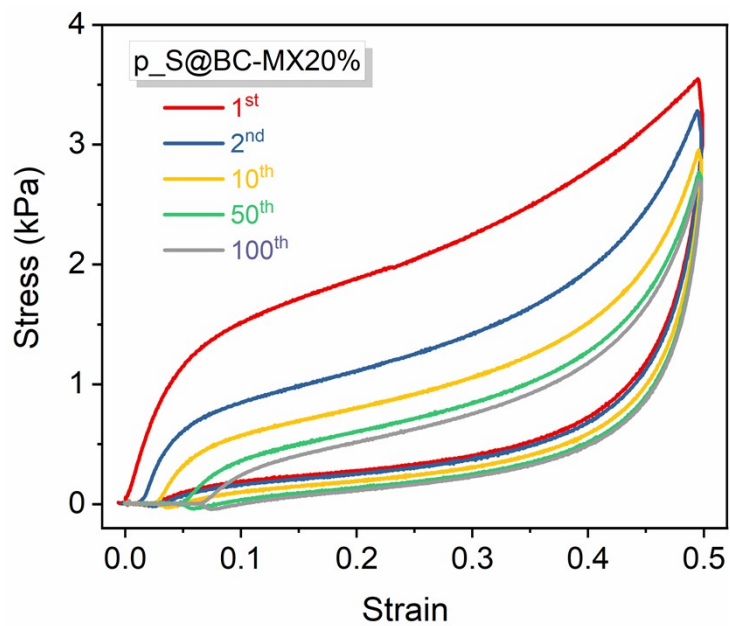


Figure S9. Stress-strain curves of S@BC-MX20% aerogel along “p_” direction at 50% strain for 100 cycles.

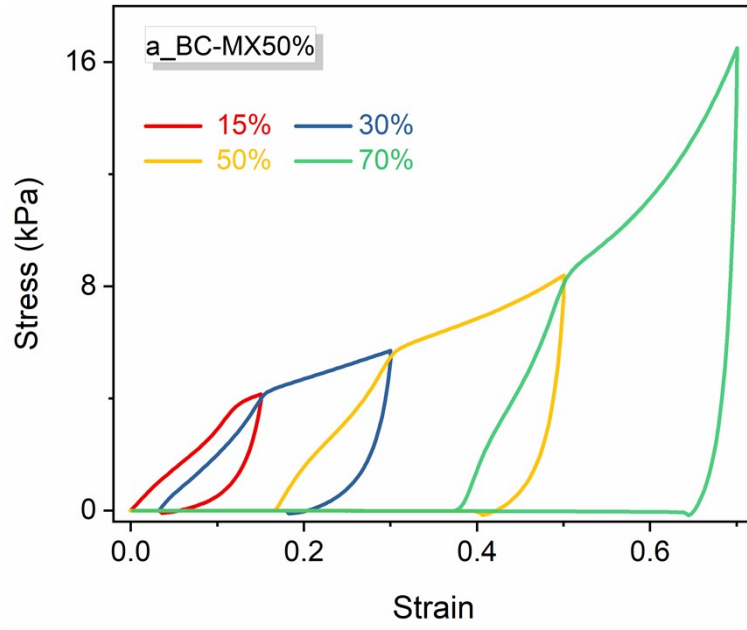


Figure S10. Stress-strain curves of unmodified BC-MX50% aerogel compressed at “a_” direction.

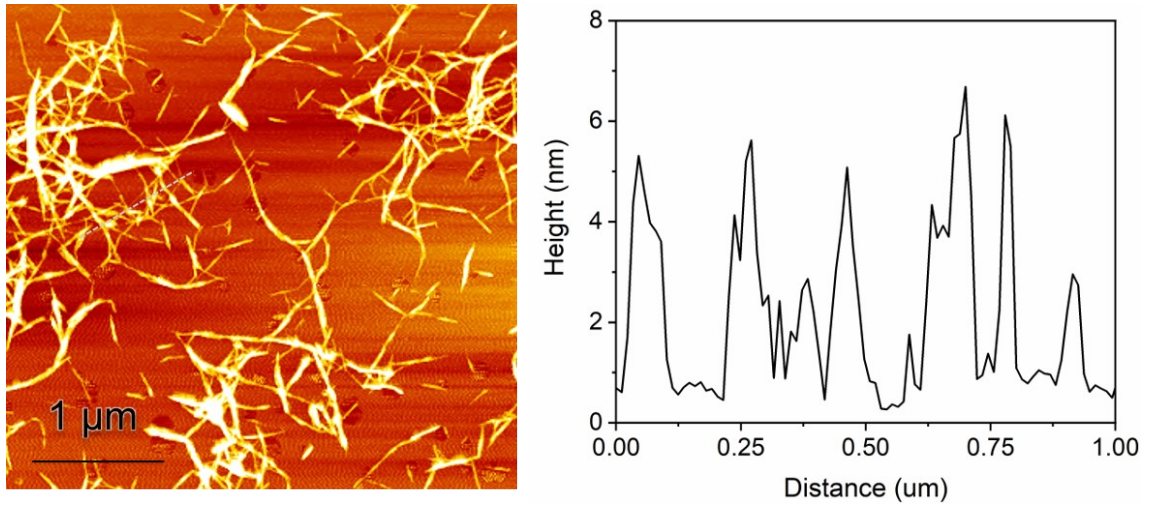


Figure S11. AFM image and height profile of CNF.

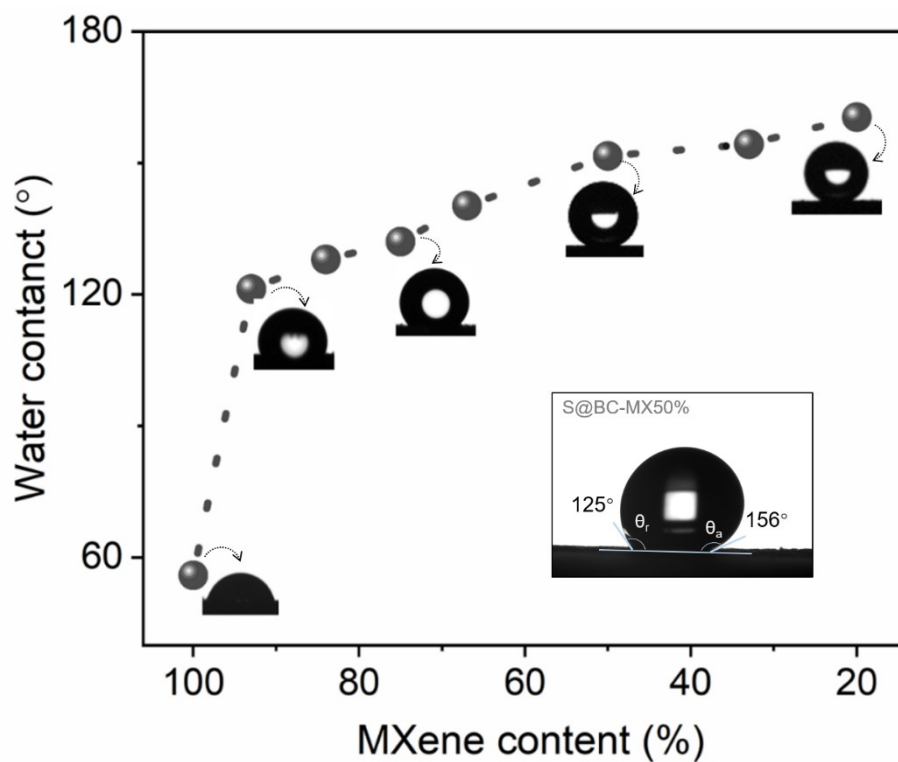


Figure S12. Water contact angle of S@BC-MX aerogels with different MXene contents, with the bottom right inset showing the advancing and receding contact angles of S@BC-MX50% aerogel measured by tilted plate experiments.

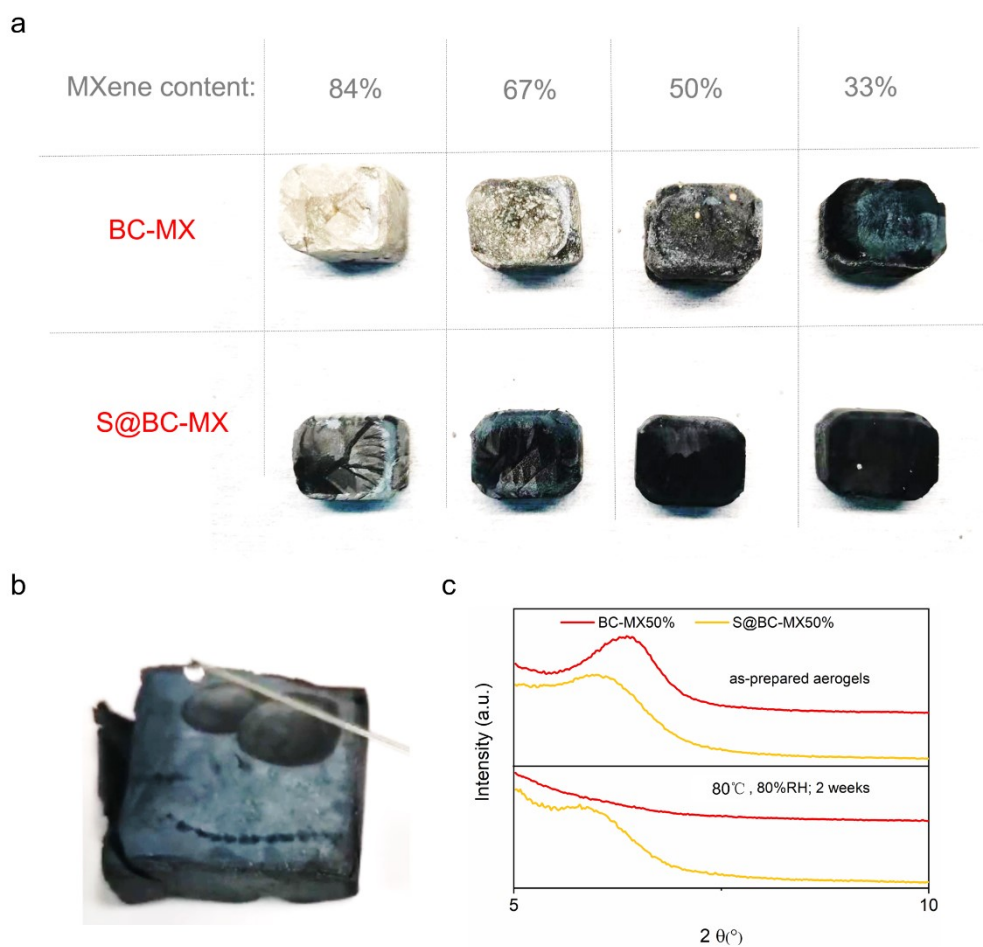


Figure S13. (a) The real-object pictures of BC-MX aerogels and S@BC-MX aerogels after being exposed to humid air for 2 weeks at 80°C under a constant relative humidity of 80%. (b) The structural collapse of unmodified BC-MX50% aerogel touched with water droplets. (c) XRD patterns of BC-MX50% aerogels and S@BC-MX50% aerogels before and after aging tests in humid air.

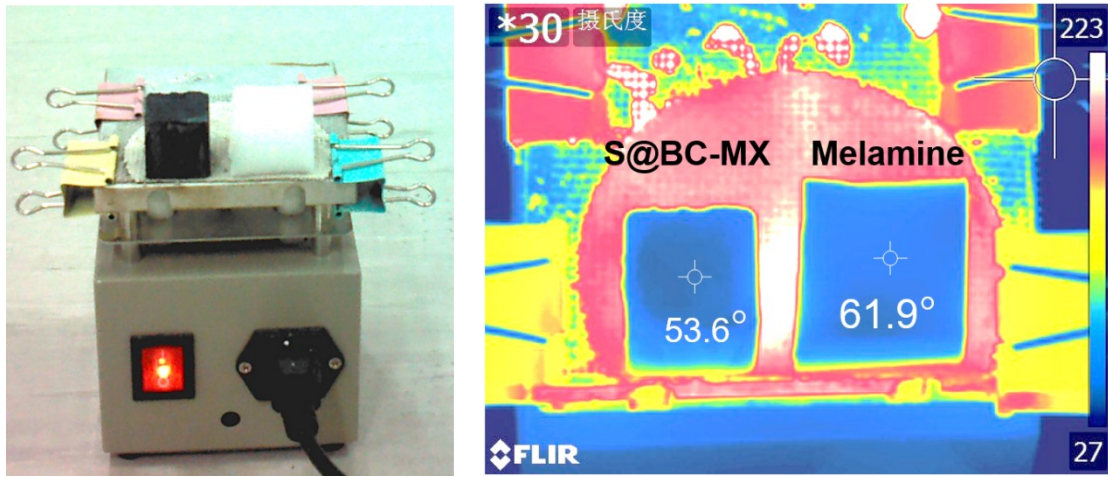


Figure S14. Physical and thermographic images of S@BC-MX50% aerogel and melamine foam (density: $9.1\text{mg}/\text{cm}^3$, thickness: 25mm) on a heating plate of $\sim 235\text{ }^\circ\text{C}$.

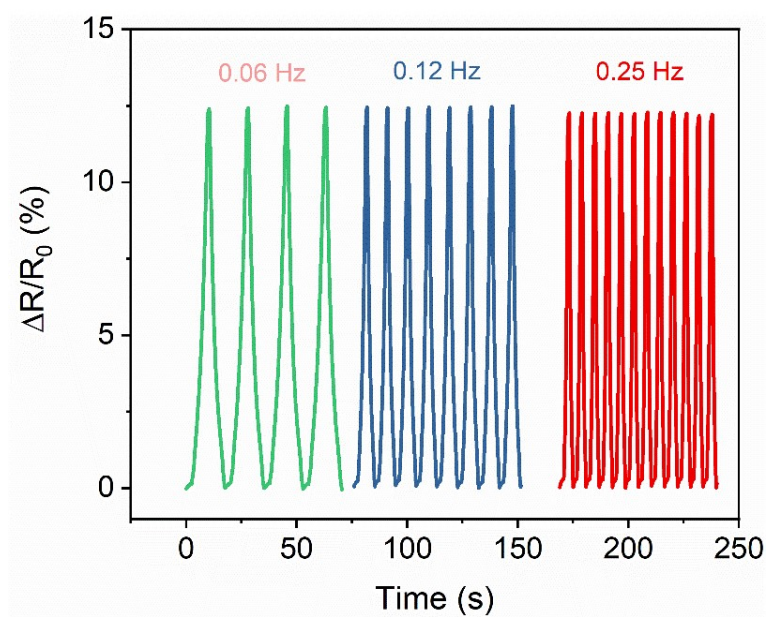


Figure S15. Normalized resistance changes of the S@BC-MX67% aerogel at 30% strain with different deformation frequencies.

Table S1. Comparison of hydrophobicity, electrical conductivity and compression property of nanocellulose- and MXene- based elastic aerogel materials.

Materials	Density (mg/cm ³)	Water contact angle (°)	Electrical conductivity (S/m)	Compression properties				Ref.
				Strain (%)	Cycle	Height retention (%)	Stress retention (%)	
S@BC- MX50%	6.2	152	7	50	500	85	60	This work
S@BC- MX84%	6.2	128	118	50	100	87.5	61	
Pure MXene	11.0	Hydrophilic	-	50	20	84	95	1
Silylated CNF	17.3	138	Insulating	50	1	96	-	2
Silylated PVA/CNF	10.2	142	Insulating	70	100	82	86	3
MXene/CNF/ PU	-	152	-	40	100	82	76	4
MXene/ANF	25.0	Hydrophilic	-	30	1000	96	92	5
MXene/CNT	9.1	Hydrophilic	447	50	100	93	79	6
Carbonized BC/MXene	26.2	Hydrophobic	0.013	50	5000	98	91	7
MXene/GO	12.2	99	31.6	50	100	99	93	8
MXene/PI	13.8	-	0.4	50	1000	86	90	9

Table S2. Elemental content of S@BC-MX aerogels with different concentrations of MXene.

MXene content	Ti (at%)	C (at%)	Si (at%)	O (at%)	F (at%)	Cl (at%)
100%	38.1	28.0	-	21.2	11.5	1.2
93%	27.6	31.7	1.0	26.4	11.5	1.8
84%	25.1	27.5	2.1	34.1	9.3	1.9
75%	17.6	38.6	2.8	34.1	5.4	1.5
67%	14.5	39.9	3.8	34.4	5.9	1.5
50%	9.1	37.0	5.1	41.2	6.7	0.9
33%	5.0	42.0	9.0	39.7	3.7	0.6

Table S3. Comparison of the absorbent capacity of various hydrophobic aerogel materials towards organic solvents and oils.

Absorbent materials	Density (mg/cm ³)	Water contact angle	Absorbent capacity (g/g)	Ref.
Silylated BC/MXene	6.2	152	88-210	This work
Silyated cellulose	2.9-16.6	142	51-143	10
Methyltrichlorosilane modified PVA-CNF	13	150	44-96	11
Trimethylsilylation modified bacterial cellulose	6.7	147	80-185	12
Carbonaceous fiber from raw cotton	12	120	55-183	13
Carbonaceous fiber from bacterial cellulose	4-6	129	106-321	14
CNT	5-10	156	87-176	15
Graphene-CNT	2.1-4.3	92-140	125-533	16
GO	4.4-7.9	155	100-250	17
APTES-modified GO-GO nanoribbon	2.3-4.1	142	98-447	18
GO-GO nanoribbon	2.2-6.8	-	100-350	19

Table S4. Electrical conductivities of S@BC-MX aerogels and BC-MX aerogels with various MXene contents at the density of 6.2 mg/cm³.

MXene content (wt%)	Electrical conductivity of S@BC-MX aerogels (S/m)	Electrical conductivities of BC-MX aerogels(S/m)
100	163	163
93	132.1	118.5
84	117.6	108.4
75	61.6	48.8
67	38.9	33.7
50	7.8	7.0
33	0.07	1.3

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