

Indirect-to-Direct Bandgap Transition in Layered Metal

Halide Perovskite - CsPb₂Br₅

–Electronic Supplementary Information

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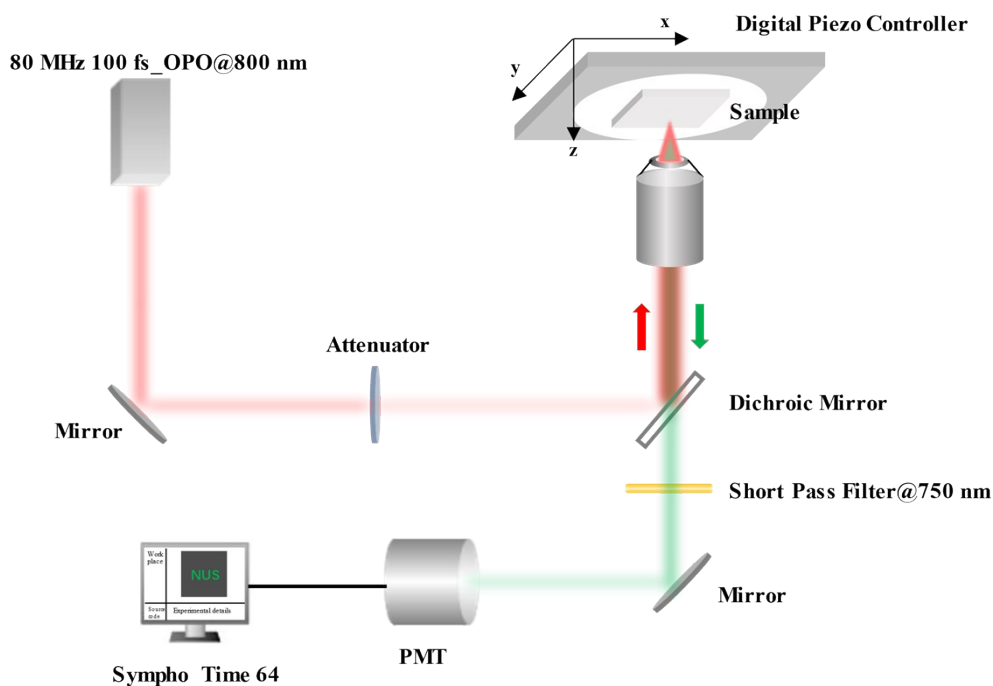


Figure S1 Schematic diagram of the optical system for laser writing and reading.

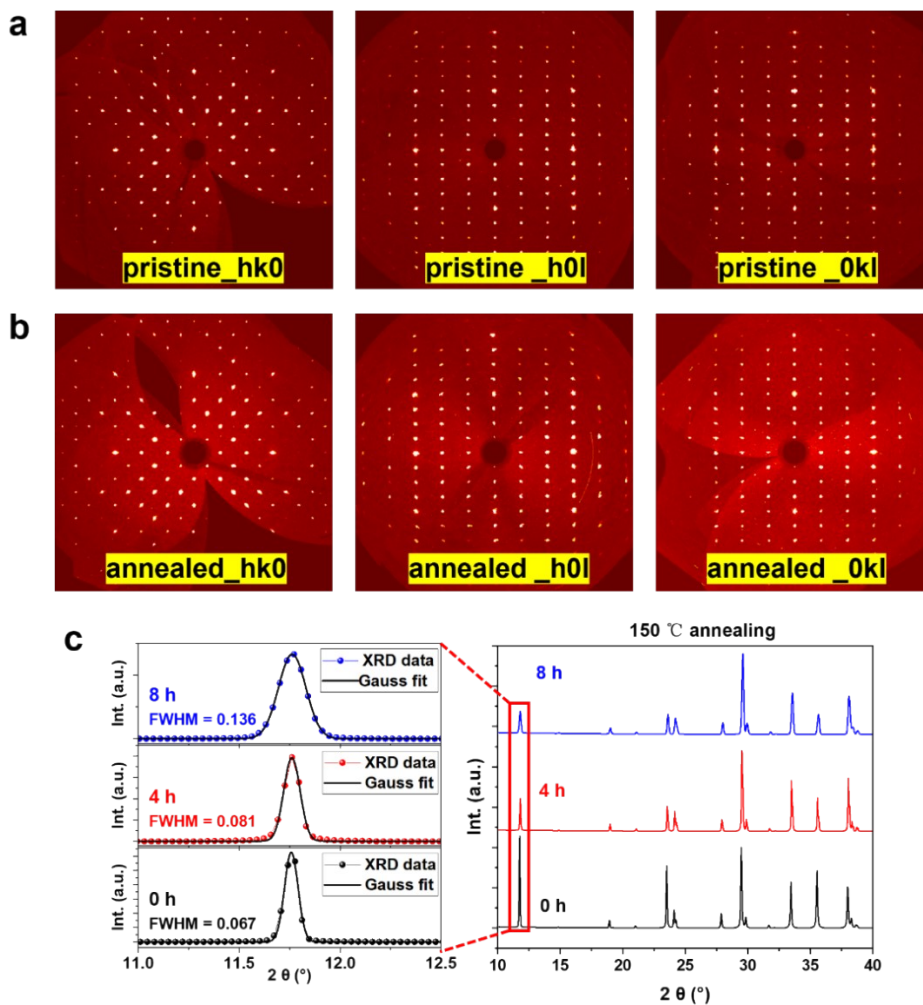


Figure S2 (a) Precession images derived from single crystal X-ray diffraction patterns of the pristine CsPb_2Br_5 SCs and (b) annealed CsPb_2Br_5 SCs (150 °C, 8 h); (c) Powder-XRD of CsPb_2Br_5 SC-powder with annealing time of 0/4/8 h at 150 °C.

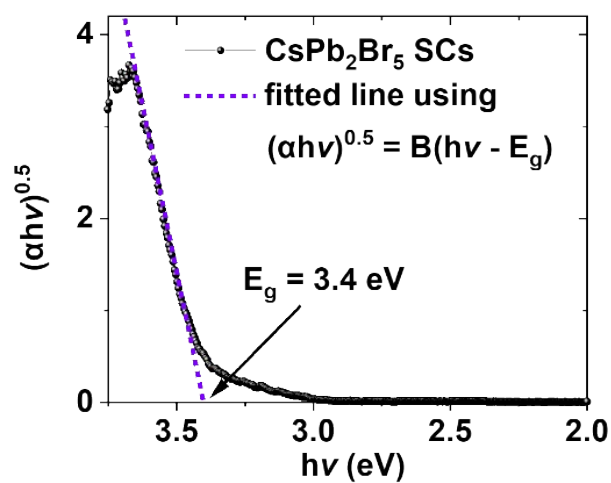


Figure S3 Indirect bandgap of CsPb_2Br_5 SCs obtained by using the Tauc method.

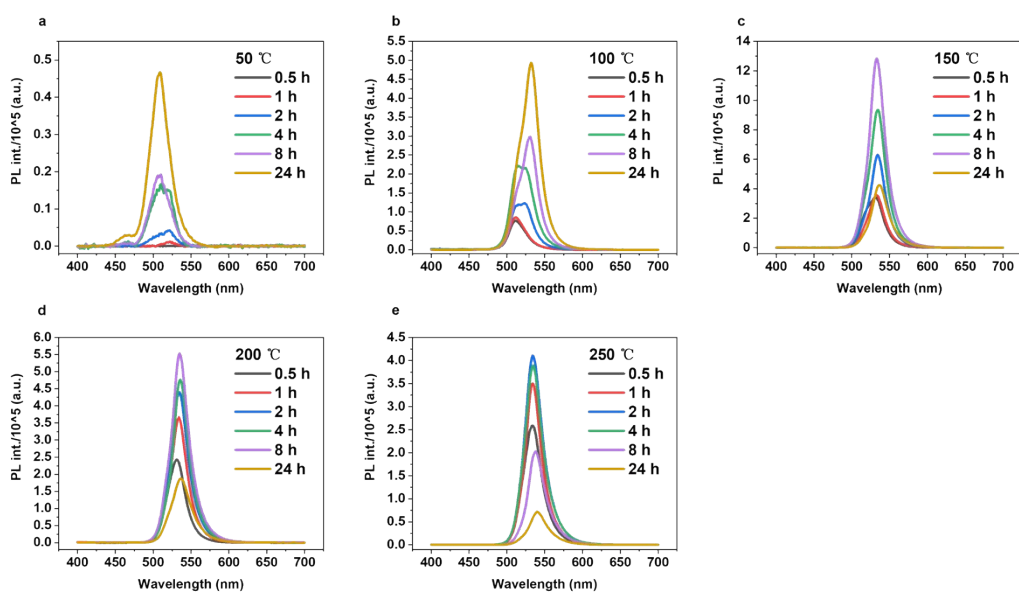


Figure S4 PL spectra of CsPb₂Br₅ SCs under thermal annealing at different temperatures for different time periods.

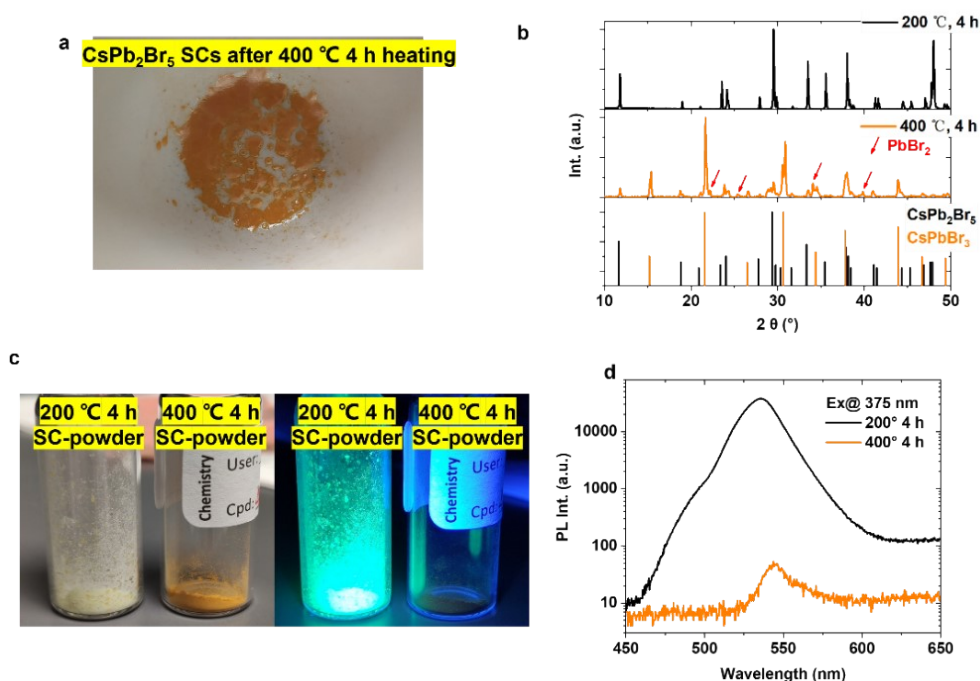


Figure S5 (a) Photograph of CsPb₂Br₅ SCs after thermal annealing at 400 °C for 4 hrs. (b) Powder-XRD patterns of CsPb₂Br₅ SCs after thermal annealing at 200/400 °C for 4 hrs and the comparison with JCPDS card of CsPb₂Br₅ (25-0211) and CsPbBr₃ (54-0752). (c) Photograph of CsPb₂Br₅ SC-powder after thermal annealing at 200/400 °C for 4 hrs under daylight and UV light. (d) PL spectra of CsPb₂Br₅ SC-powder after thermal annealing at 200/400 °C for 4 hrs.

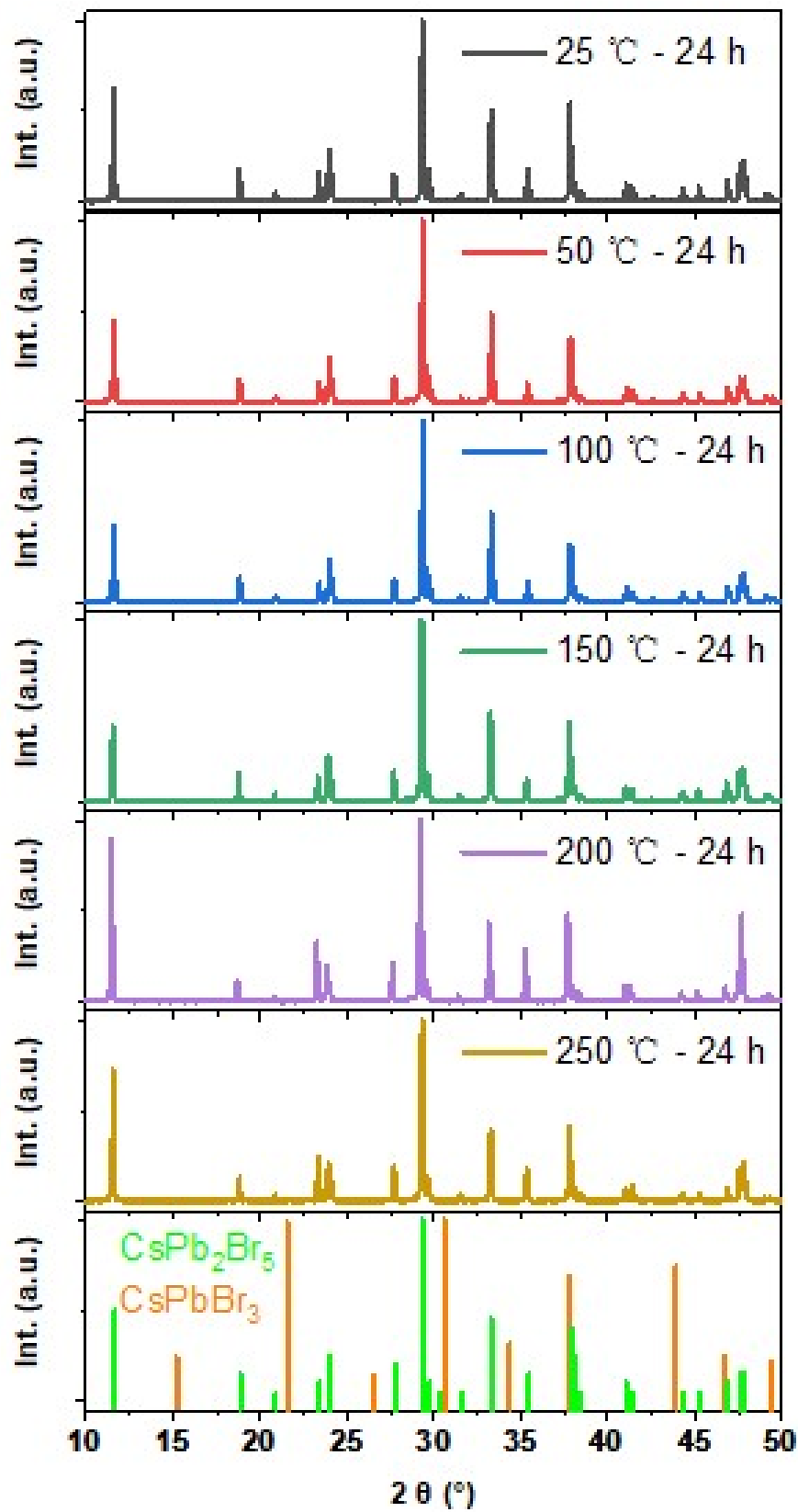


Figure S6 Powder-XRD patterns of CsPb₂Br₅ SCs with different annealing temperatures for 24 hours.

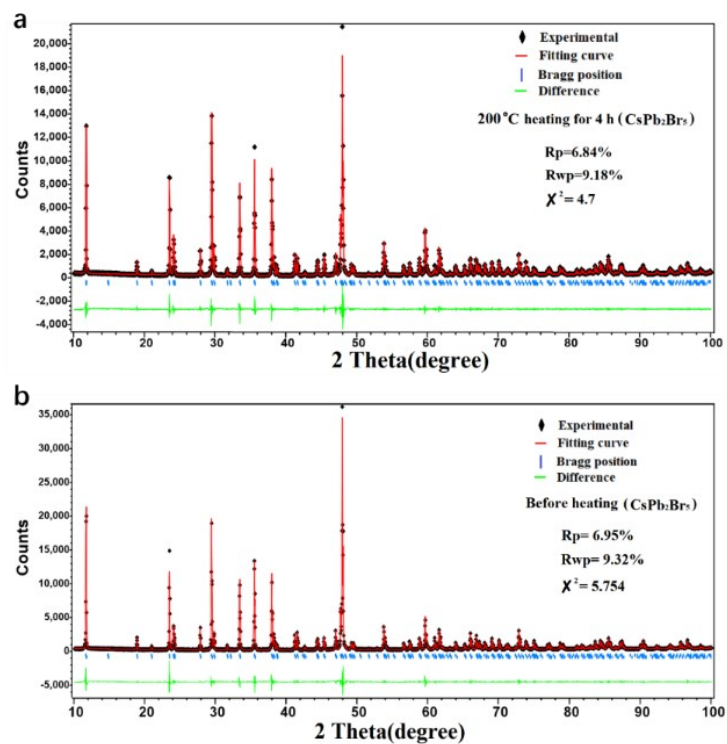


Figure S7 (a, b) Rietveld refinement pattern of powder XRD data for CsPb_2Br_5 SCs before (a) and after (b) thermal annealing.

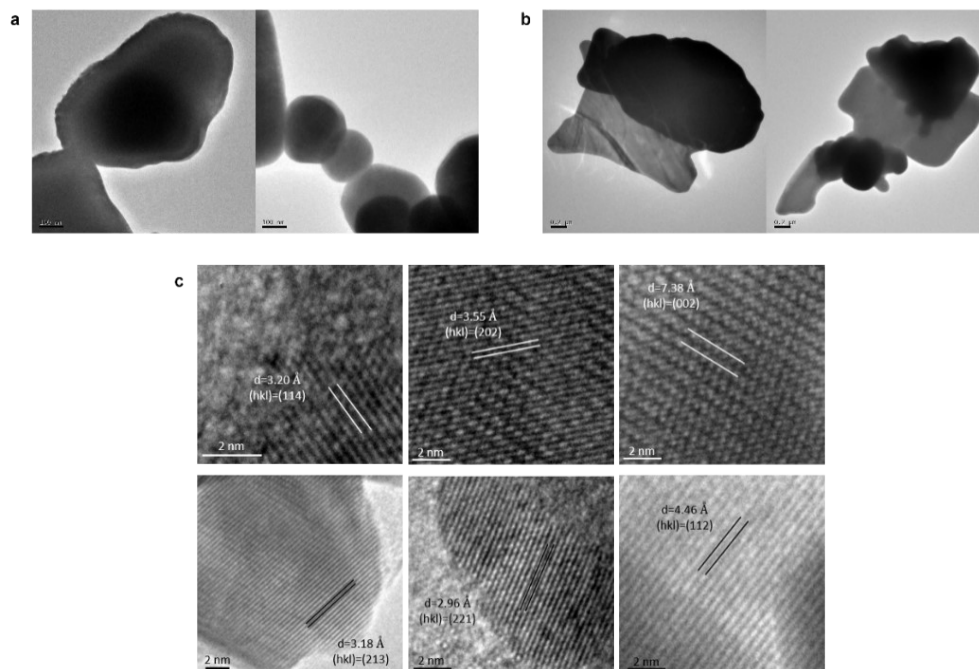


Figure S8 TEM (a, b) and HRTEM (c) images of CsPb_2Br_5 SC-powder before (a) and after (b, c) thermal annealing treatment.

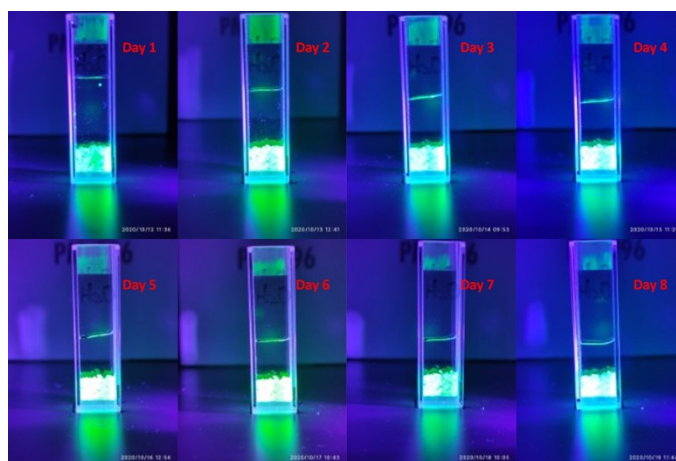


Figure S9 Photograph of the annealed CsPb_2Br_5 SCs immersed in water for 8 days under UV light illumination.

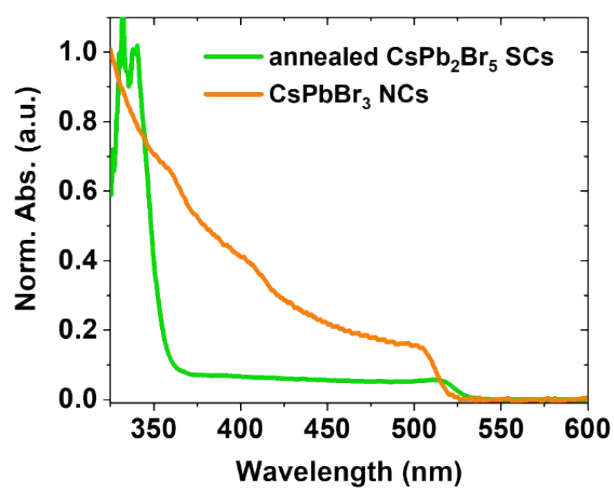


Figure S10 Absorption spectra of the annealed CsPb_2Br_5 SCs and pure CsPbBr_3 NCs.

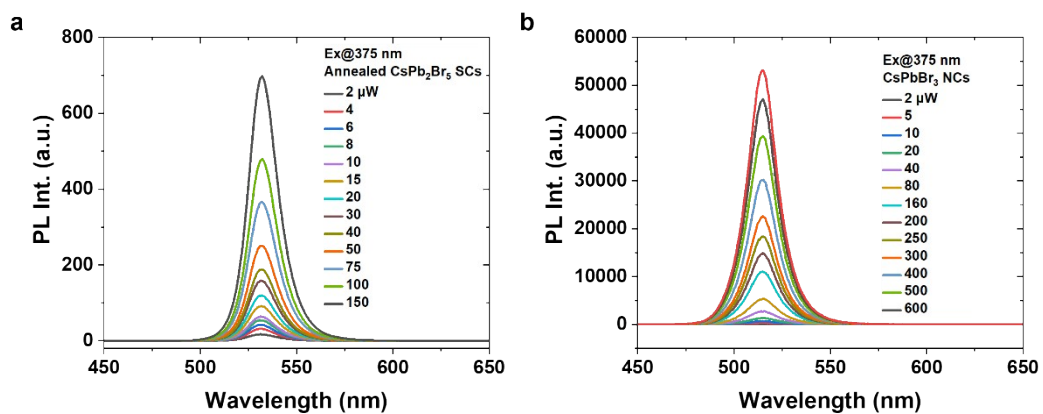


Figure S11 Excitation power dependent PL spectra of the annealed CsPb_2Br_5 SCs (a) and pure CsPbBr_3 NCs (b).

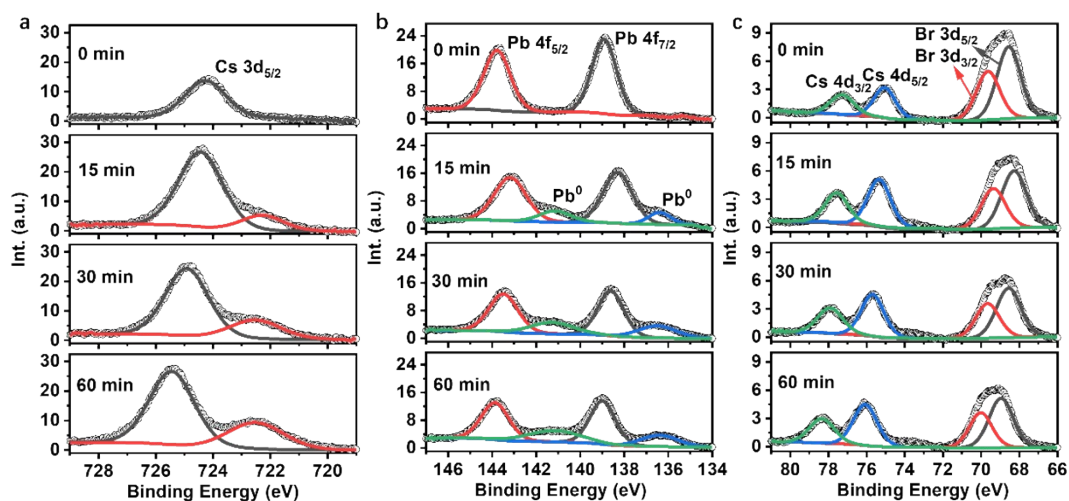


Figure S12 *In-situ* XPS spectra of CsPb₂Br₅ SCs during thermal annealing.

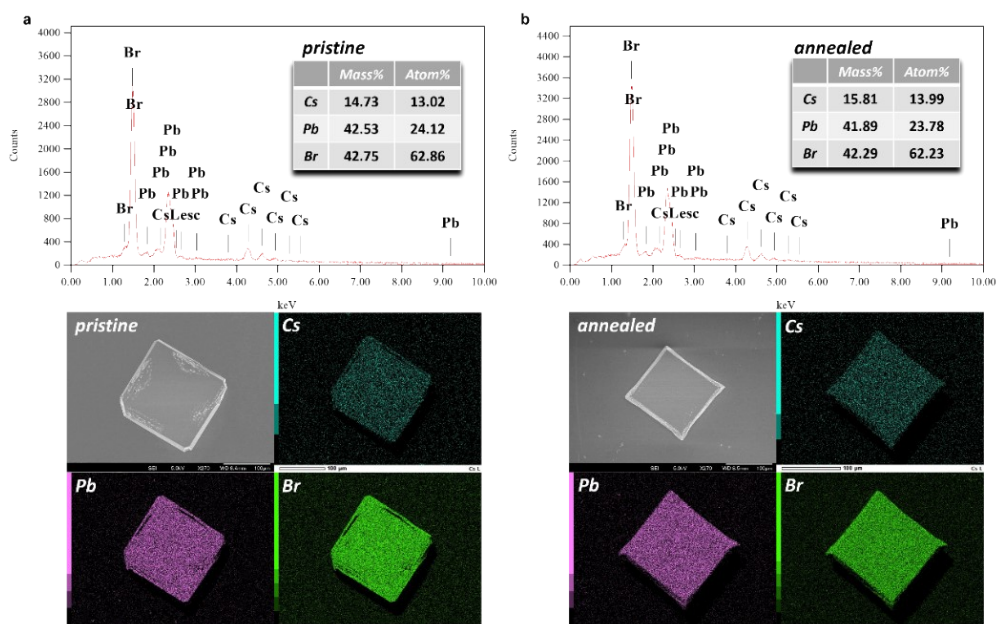


Figure S13 SEM-EDS spectra and mapping of the pristine (a) and annealed (b) CsPb₂Br₅ SCs (200°C, 4 h).

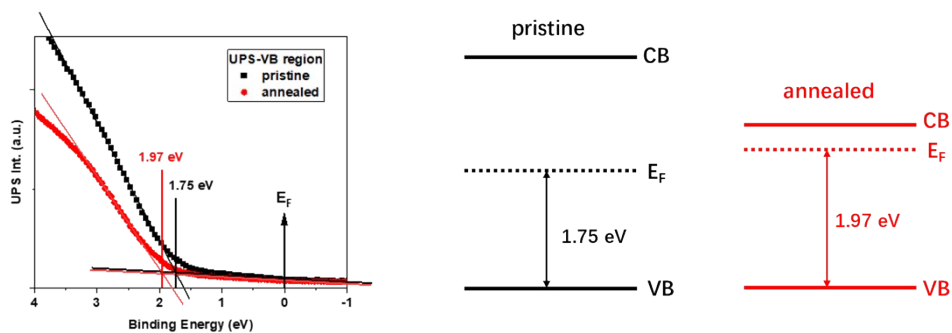


Figure S14 UPS spectra of the pristine and annealed CsPb_2Br_5 (200 °C, 1 hr). Position of valence band (VB) edges are marked with vertical bars (black for the pristine sample and red for the annealed sample), and Fermi level (E_F) is marked with a vertical arrow.

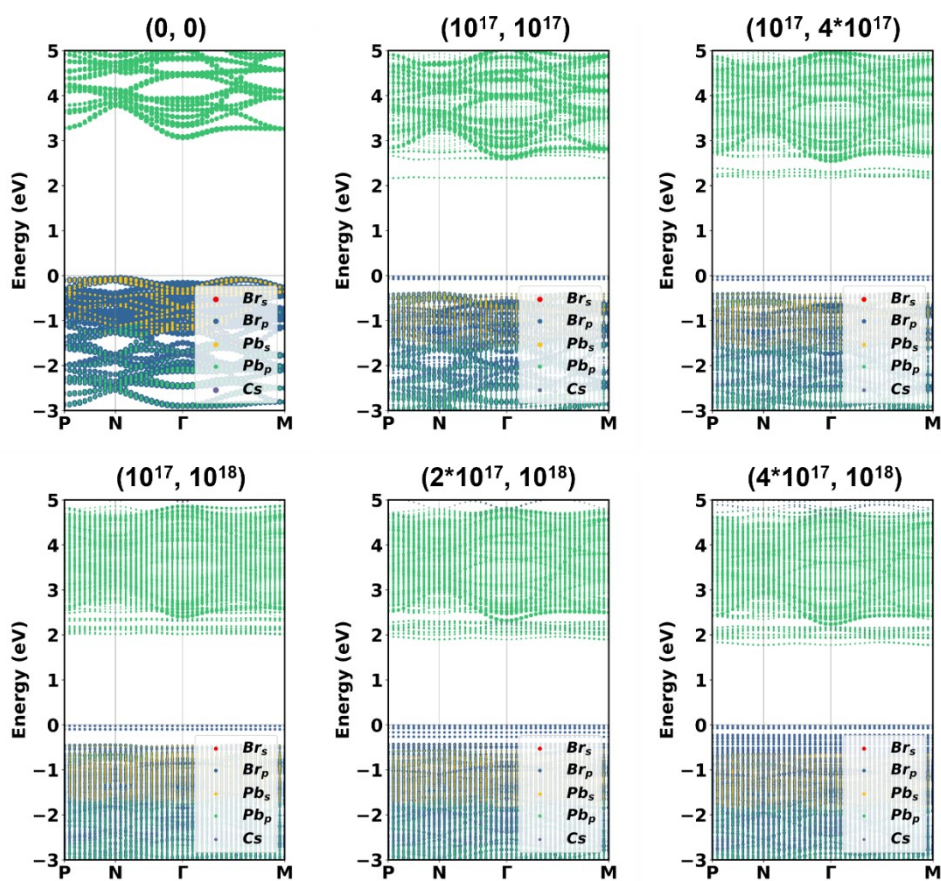


Figure S15 Band structures of CsPb_2Br_5 with different defect concentrations. Defect concentrations of $(C_{\text{Spb}}, V_{\text{Br}})$ are marked on the top of each panel in cm^{-3} .

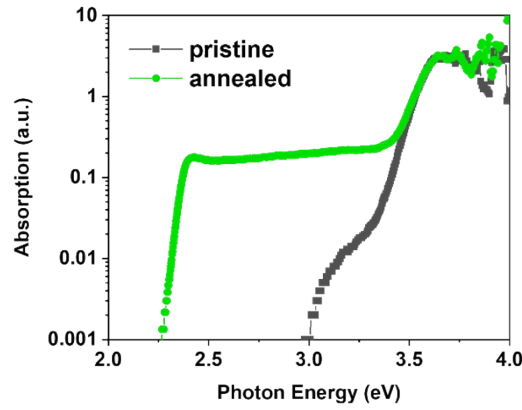


Figure S16 Absorption spectra of the pristine and annealed (150 °C, 8 hrs) CsPb₂Br₅ SCs. The data are the same as in Fig. 1d with the y-axis plotted in logarithms for a clearer comparison. A dramatic increase of the absorption near the band edge with a factor of more than 200 within 0.1 eV photon energy for the annealed CsPb₂Br₅. In contrast, the absorbance near the band edge of the pristine CsPb₂Br₅ increases by a factor of only 30 within 0.4 eV. Such dramatic increase in the absorbance slope near the band edge suggests a transition from indirect to direct bandgap after thermal annealing as the absorbance (A) near band edge for direct and indirect bandgap semiconductors has very different formula (ref. 65 and 66).

Direct bandgap: $A \propto A' * \sqrt{h\nu - E_g}$, where A' is a certain frequency-independent constant, $h\nu$ is photon energy, and E_g is bandgap.

Indirect bandgap: $A \propto \frac{(h\nu - E_g + E_p)^2}{\exp\left(\frac{E_p}{kT}\right) - 1} + \frac{(h\nu - E_g - E_p)^2}{1 - \exp\left(-\frac{E_p}{kT}\right)}$, where E_p is phonon energy, k is Boltzmann's constant, and T is temperature.

Direct bandgap semiconductors show a dramatic increase in absorbance when photon energy is just above the bandgap. In contrast, the absorbance near the band edge increases slowly with photon energy for indirect bandgap semiconductors, followed by a speedy increase due to photon energy being higher enough to allow direct optical absorption.

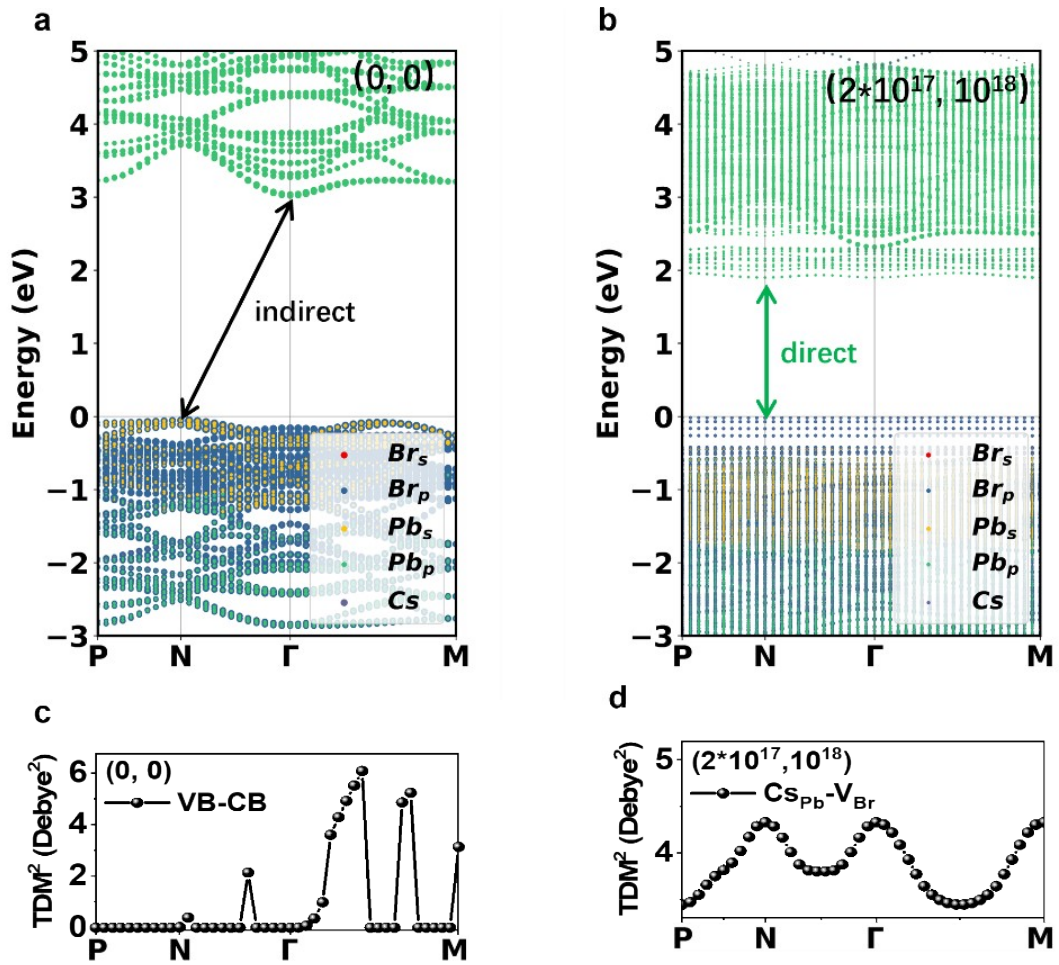


Figure S17 (a-d) Band structure and TDM^2 of $CsPb_2Br_5$ without defects (a, c) and $CsPb_2Br_5$ with defect concentration of $(2 \cdot 10^{17}, 10^{18})$ (b, d).

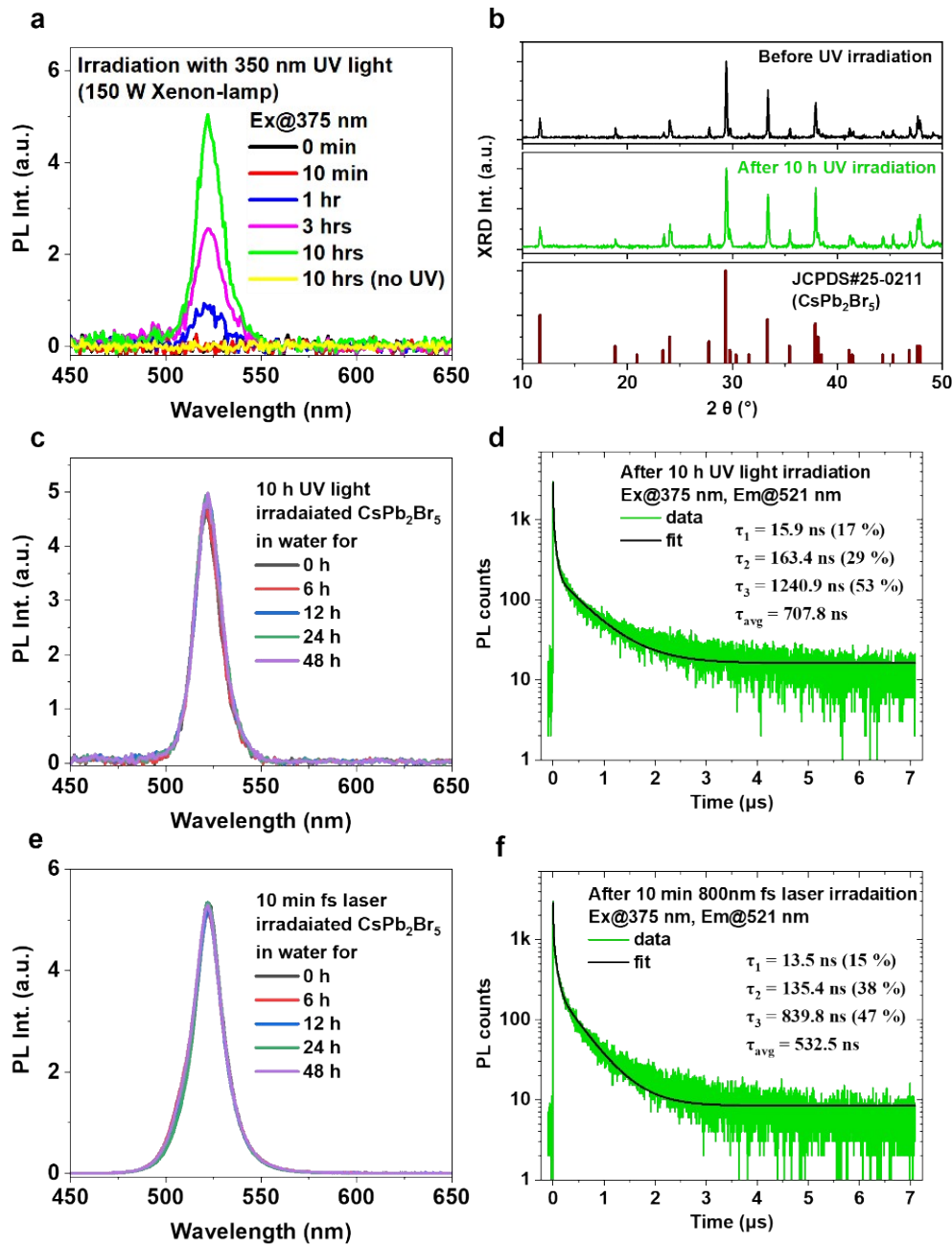


Figure S18 (a) Photoluminescence evolution of CsPb_2Br_5 SCs under 350 nm UV light irradiation; (b) powder-XRD of pristine and UV-light irradiated CsPb_2Br_5 SCs; Water stability measurements for (c) UV light irradiated and (e) fs-laser irradiated CsPb_2Br_5 SCs; Time-resolved PL measurements for (d) UV light irradiated and (f) fs-laser irradiated CsPb_2Br_5 SCs.

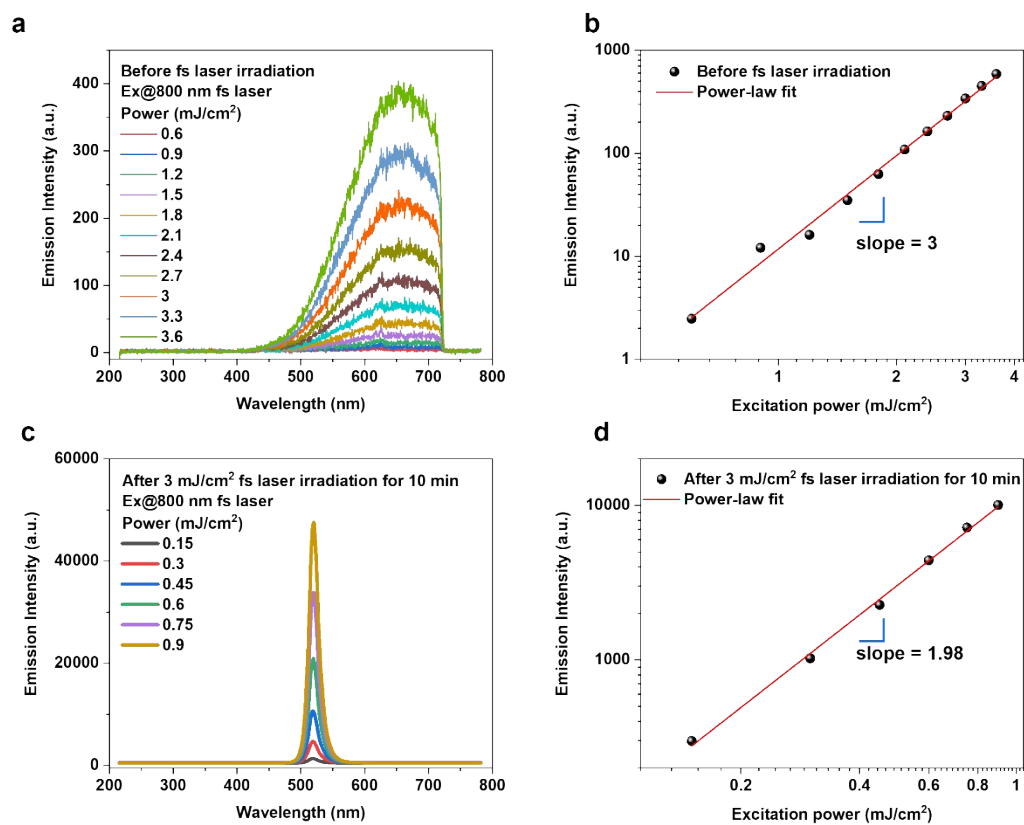


Figure S19 Excitation power-dependent emission spectra (**a, c**) and intensities (**b, d**) of CsPb₂Br₅ SCs before (**a, b**) and after (**c, d**) fs laser irradiation for 10 min (800 nm, 3mJ/cm²). The excitation source for the emission spectra is 800 nm fs laser pulses.