Supporting information

Ultrasmall superparamagnetic iron oxide nanoparticles for enhanced tumor penetration

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Figure S1. Standard curve of iron in 6M HCL via Prussian blue reaction.



Figure S2. TEM image of typical octahedral 15 nm Fe_3O_4 -OA nanoparticles. Scale bar, 10 nm.



Figure S3. The cell viability of MCF-7 monolayer cells after treatment with different types of Fe_3O_4 nanoparticles at (A) 24 h and (B) 48 h. (Mean values ± standard deviation, n=5).



Figure S4. (A) Distribution of different types of nanoparticles in MCF-7 cells for incubation times of 1 h and 3 h. The cells were observed by an objective lens of 63X. Scale bar, 30 μ m. (B) Pictures illustrated the process of calculating the relative area of the Fe₃O₄ nanoparticles to cells in ImageJ. The yellow lines indicate the cellular areas and the red dots indicate the accumulation of Fe₃O₄ nanoparticles.



Figure S5. Picture displaying the Fe₃O₄-DMSA nanoparticles possessing higher dye loading efficiency than Fe₃O₄-PEI nanoparticles, showing deeper purple colour when the iron concentrations of the two types of nanoparticles are the same.

Sample name	Dynamic light scattering		Zeta potential (mV
	Size (Number mean, nm) (n=3)	Polydispersity index (PDI) (n=3)	(n=3)
O-Fe ₃ O ₄ -15 (-)	14.4 ± 0.2	0.42 ± 0.01	-30.8 ± 2.3
S-Fe ₃ O ₄ -21 (-)	23 ± 5.2	0.487 ± 0.007	-37 ± 1.4
S-Fe ₃ O ₄ -10 (+)	59.5 ± 1.7	0.209 ± 0.01	34.8 ± 0.6
O-Fe ₃ O ₄ -15 (+)	122 ± 14.8	0.179 ± 0.02	22.7 ± 2.6
S-Fe ₃ O ₄ -21 (+)	306.8 ± 8.1	0.594 ± 0.02	30.5 ± 0.2