

Supporting Information (SI)

Modulating Electronic and Optical Properties of Monolayered MoS₂ by Covalent Mono- and Bisfunctionalization

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Fig. S1: (a) The QP band gaps of the functionalized 2H-MoS₂ with -CH₃ group as a function of vacuum thickness with 50 eV G₀W₀ self-energy and 12×12 k-point sampling (2×2 supercell). (b) The QP band gaps of the functionalized 2H-MoS₂ with -CH₃ group as a function of k-point sampling with 50 eV G₀W₀ self-energy and 8 Å vacuum thickness (2×2 supercell). (c) Optical absorption spectra of the functionalized 2H-MoS₂ with -CH₃ group as a function of the number of valence and conduction bands with 50 eV self-energy and 8 Å vacuum thickness (2×2 supercell).

Fig. S2: Optimized structures of functionalized 2H-MoS₂ and 1T'-MoS₂ with various groups.

Fig. S3: Average energy difference ΔE between functionalized 2H- and 1T'-MoS₂ as a function of the group coverage. Here $\Delta E = (E_{1T'} - E_{2H})/m$, where m is the number of MoS₂ units, $E_{1T'}$ and E_{2H} are the total energy of functionalized 2H- and 1T'-MoS₂ for a given coverage, respectively. The positive ΔE indicates that functionalized 2H-MoS₂ is more stable, while the negative ΔE indicates that functionalized 1T'-MoS₂ is more stable.

Fig. S4: Average Bader charge transfer ΔQ as a function of the group coverage with respect to the 2H- and 1T'-MoS₂ monolayers. Here $\Delta Q = \sum_i^N (Z_i - Q_i^{Bader})/n$, where N is the total number of atoms in the attached groups, Z_i is the number of electrons of atom i , Q_i^{Bader} is the Bader charge of atom i and n is the number of functional groups. A positive charge transfer value indicates electron flow from the substrate to the functional group.

Fig. S5: DFT band gaps as a function of the group coverage with respect to the functionalized 2H-MoS₂ and 1T'-MoS₂ monolayers.

Fig. S6: Total density of states and the projected density of states for the functionalized (a) 2H-MoS₂ and (b) 1T'-MoS₂ calculated by PBE functional.

Fig. S7: Optical absorption spectra of the functionalized 1T'-MoS₂ monolayer.

Table S1: The optical band gap (E_{opt}) for the functionalized 2H-MoS₂ at 6.25% coverage. Energies are in eV.

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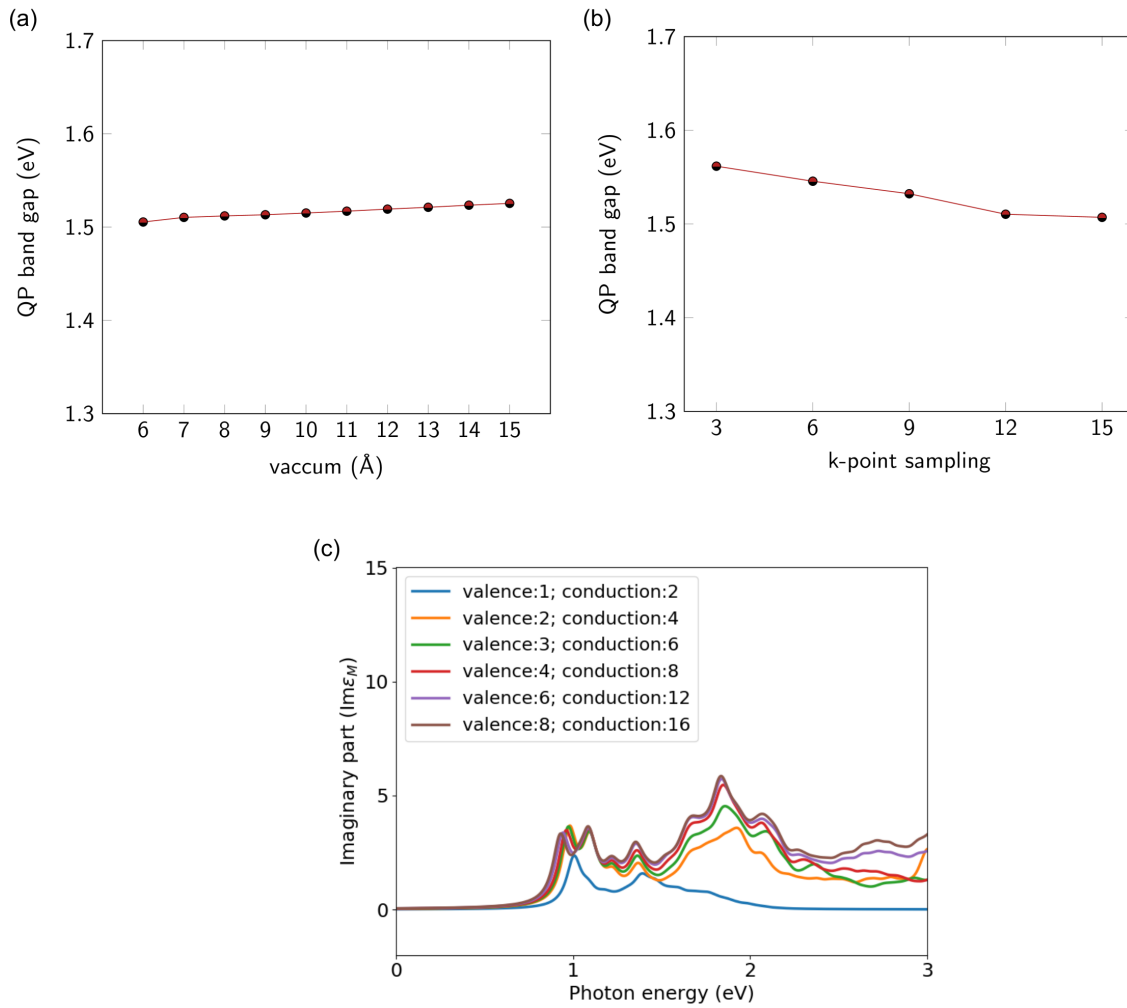
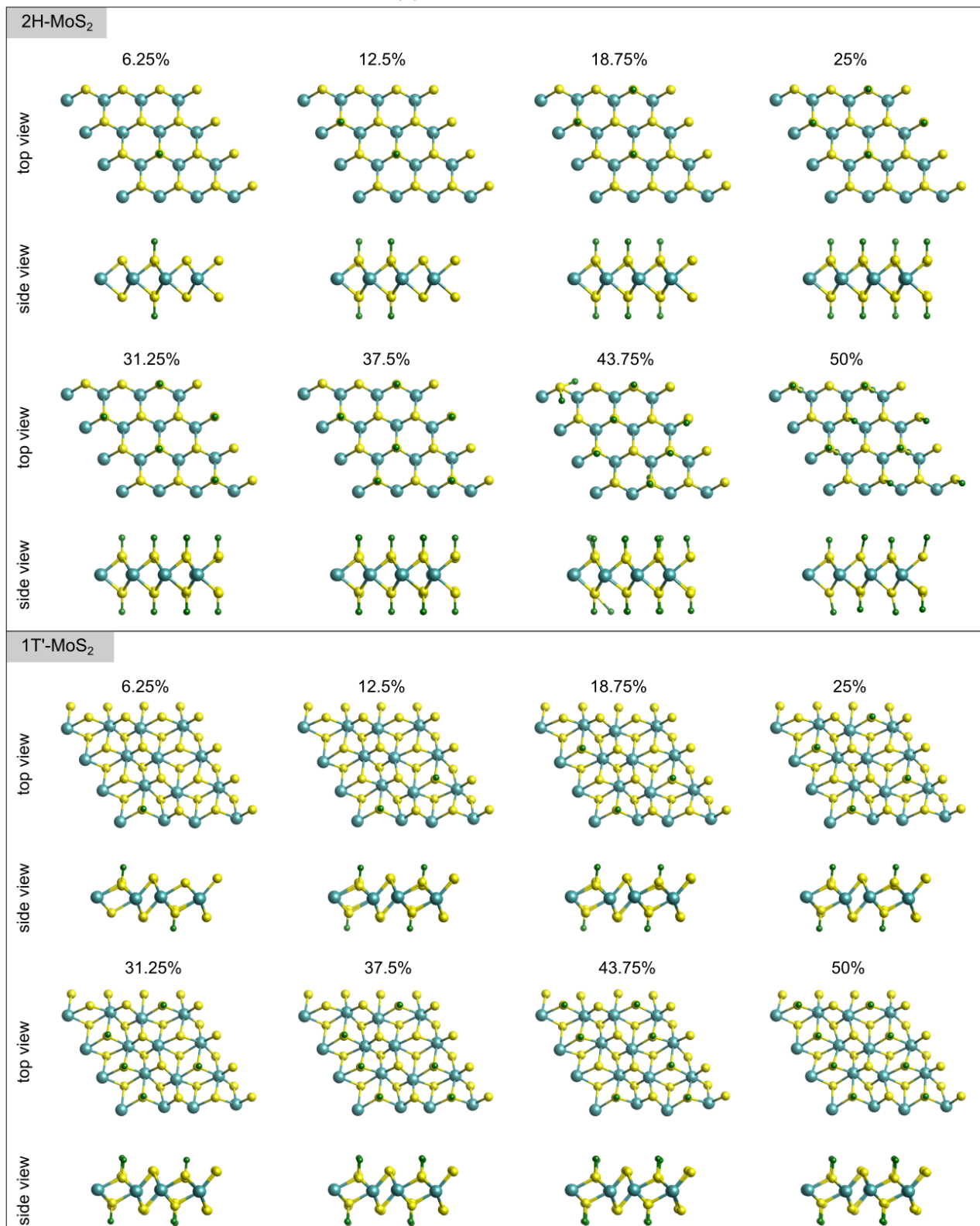
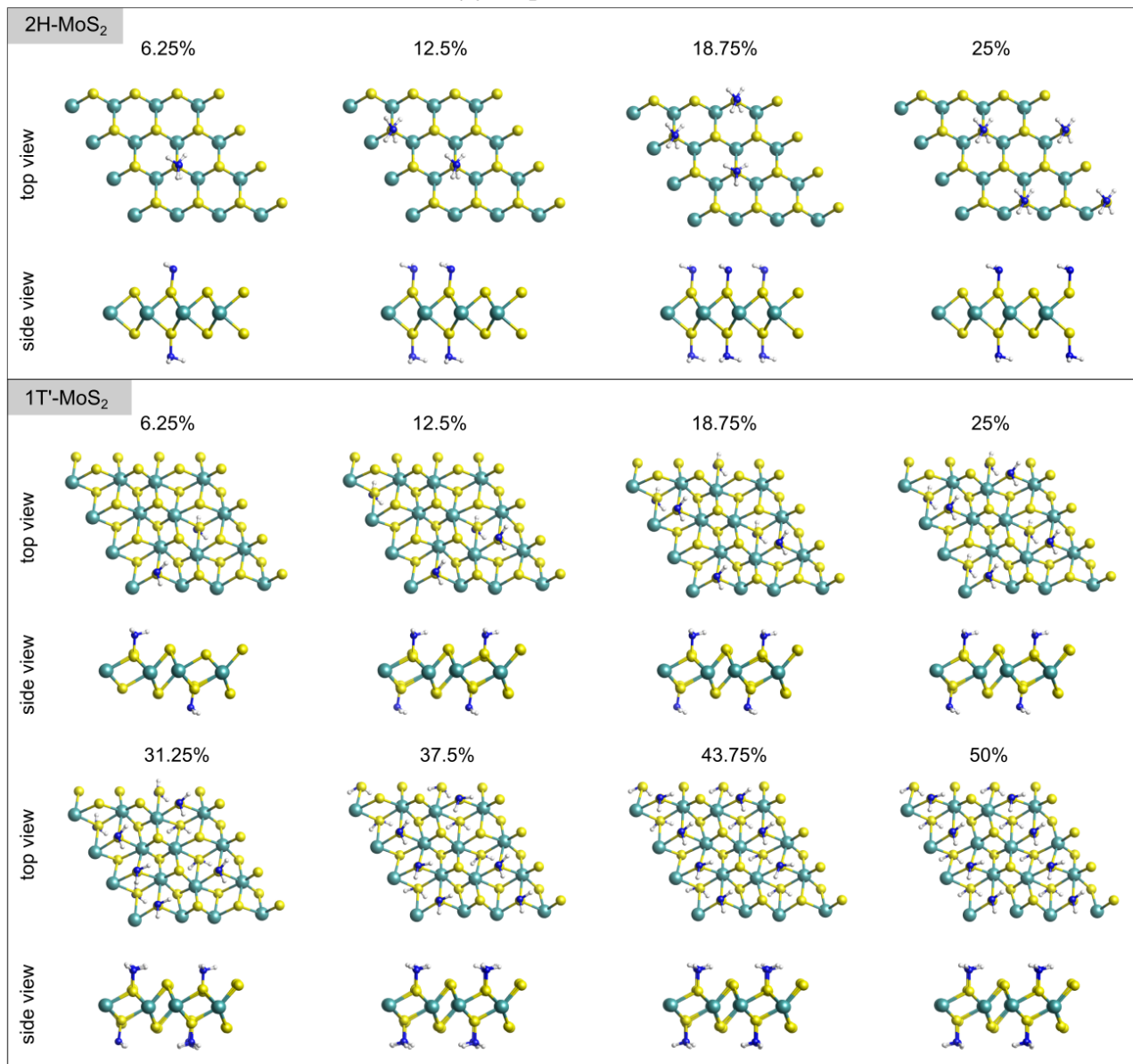


Fig. S2: Optimized structures of functionalized 2H-MoS₂ and 1T'-MoS₂ with various groups.

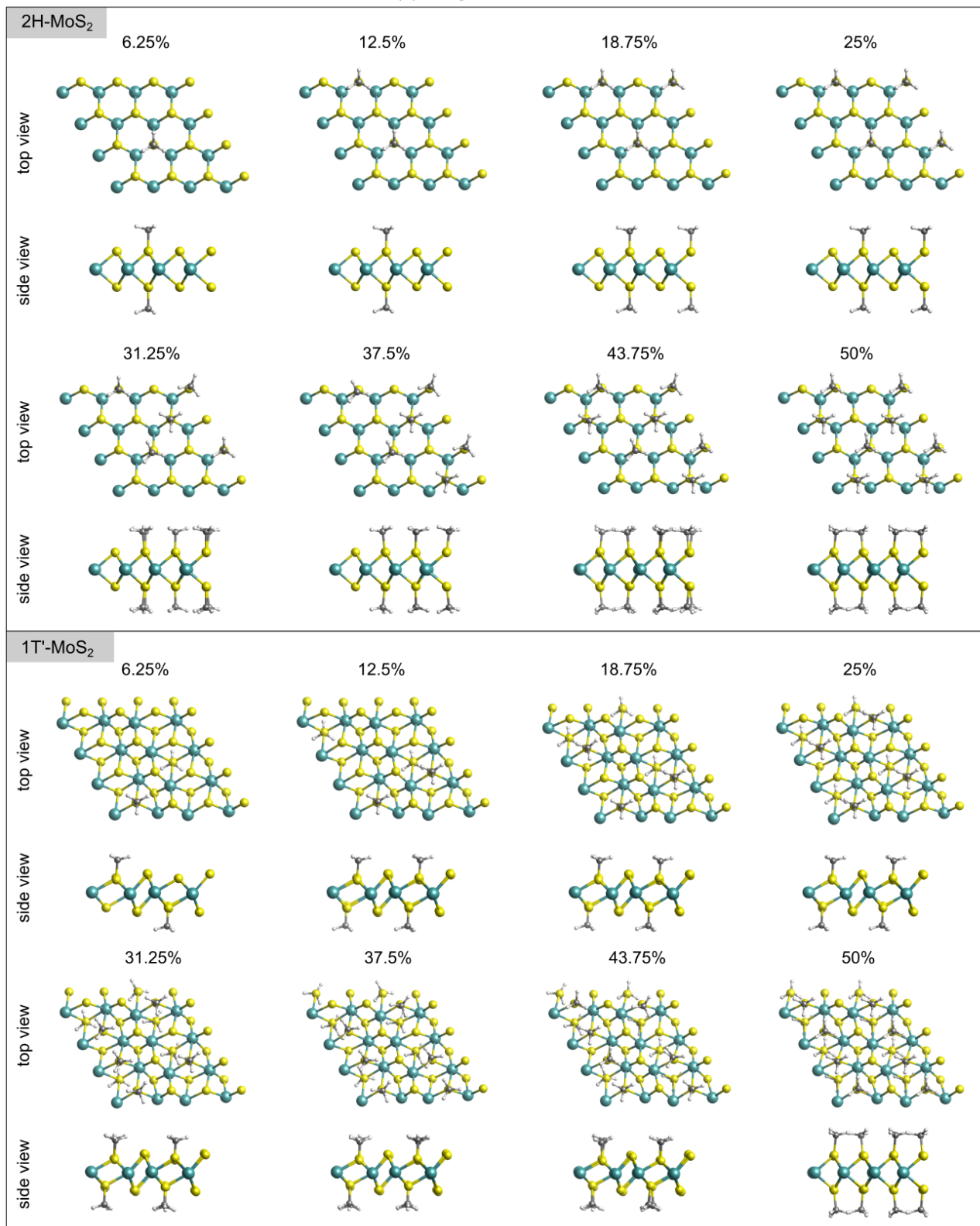
(a) -F



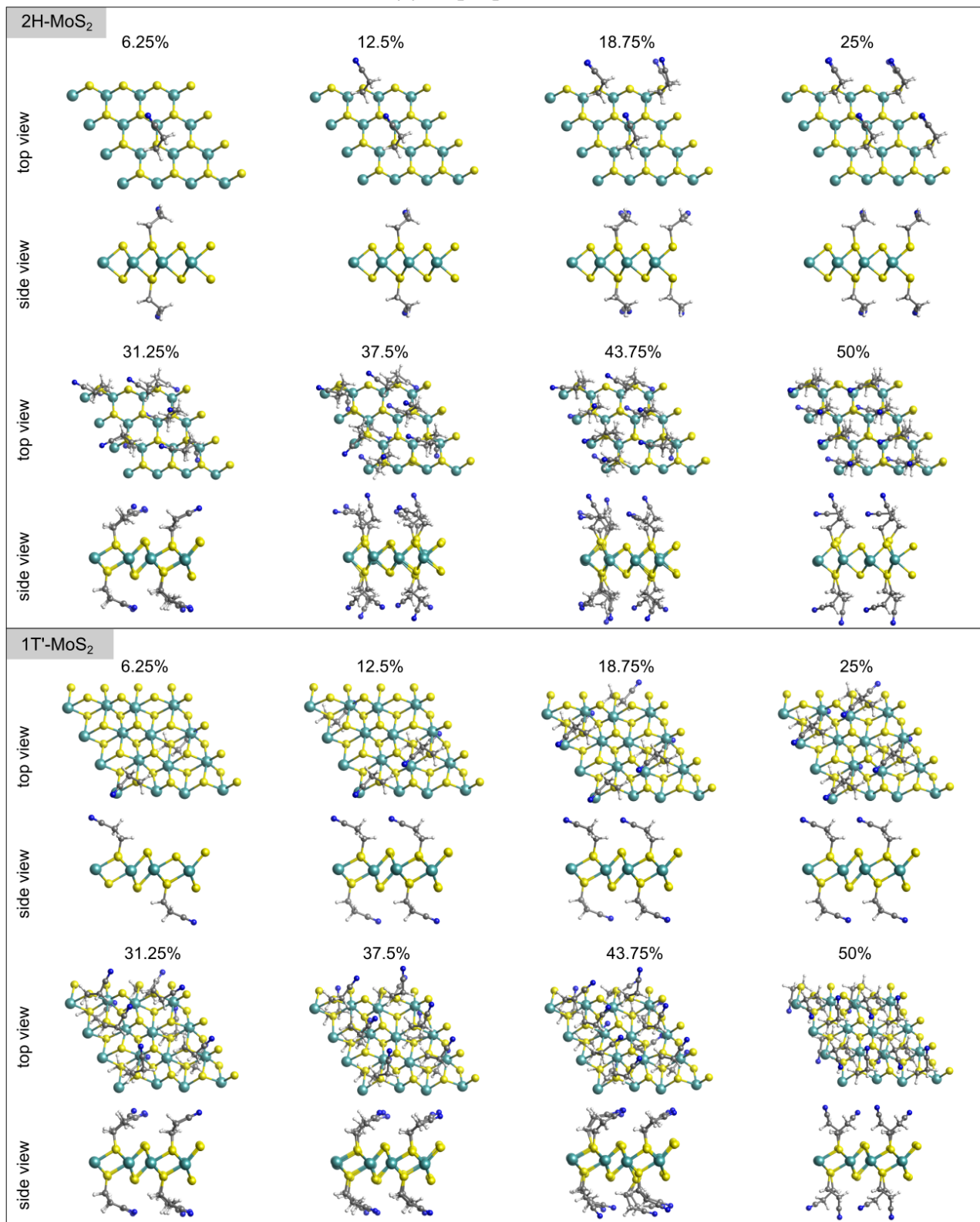
(b) -NH₂



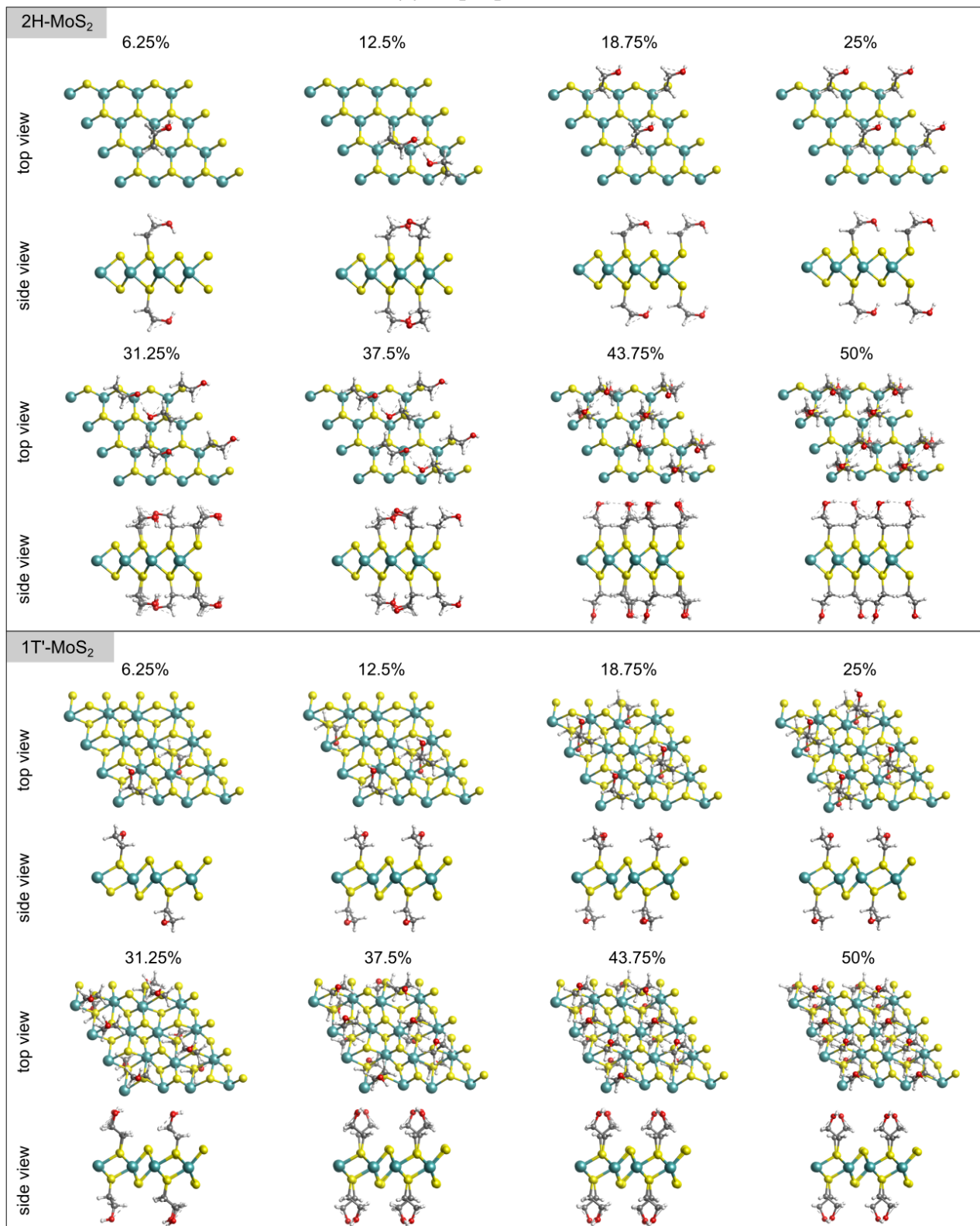
(c) -CH₃



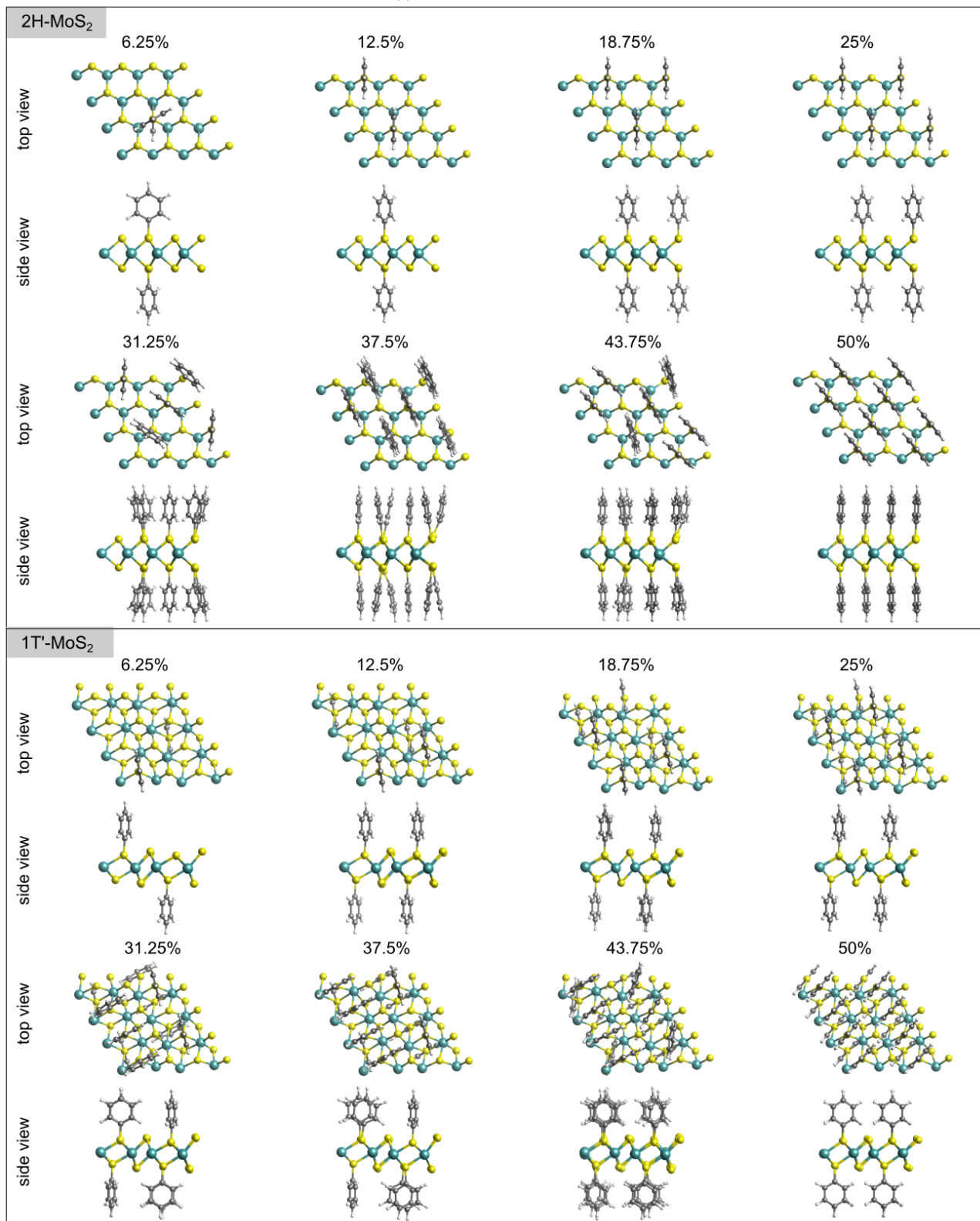
(d) -CH₂CH₂CN



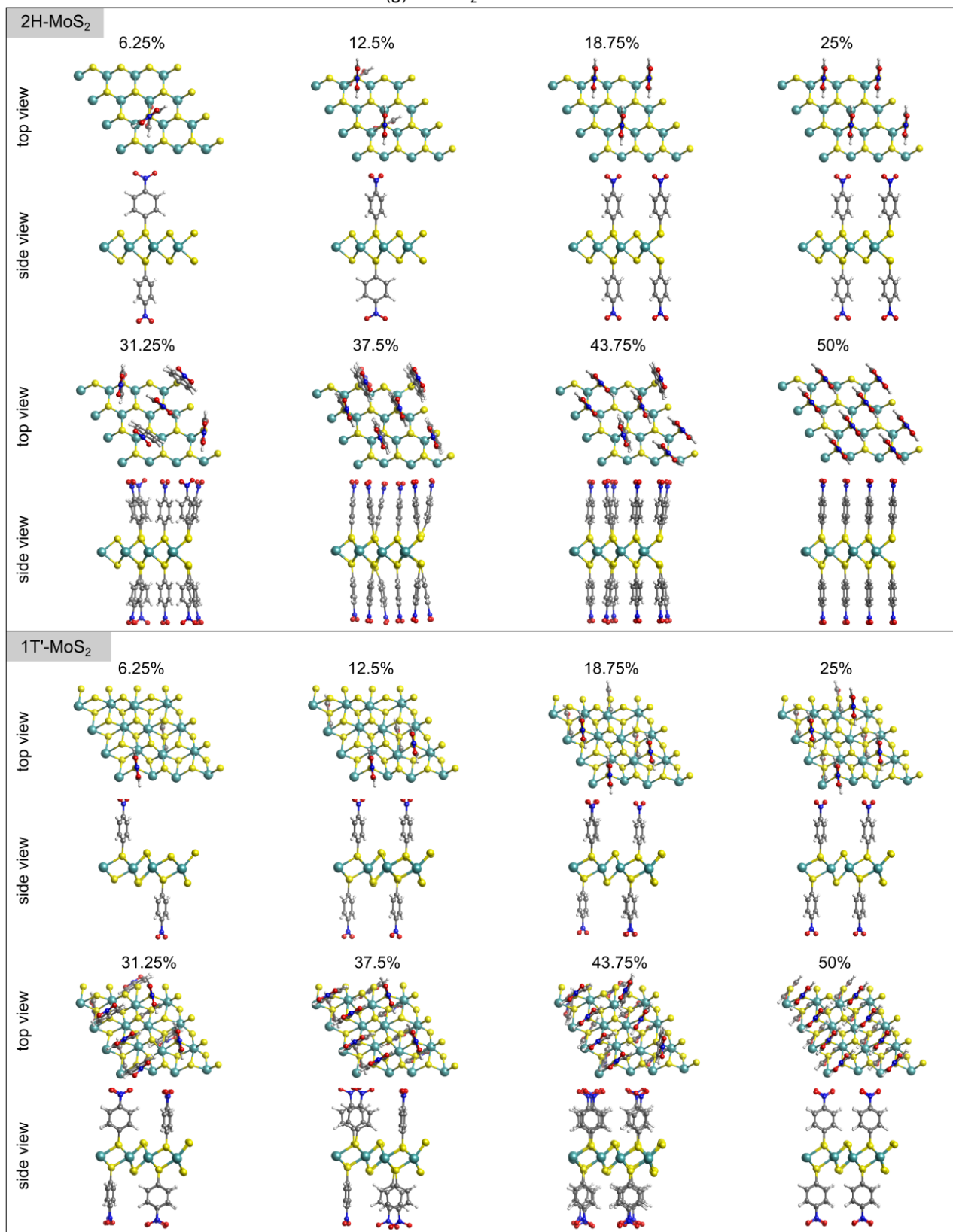
(e) -CH₂CH₂OH



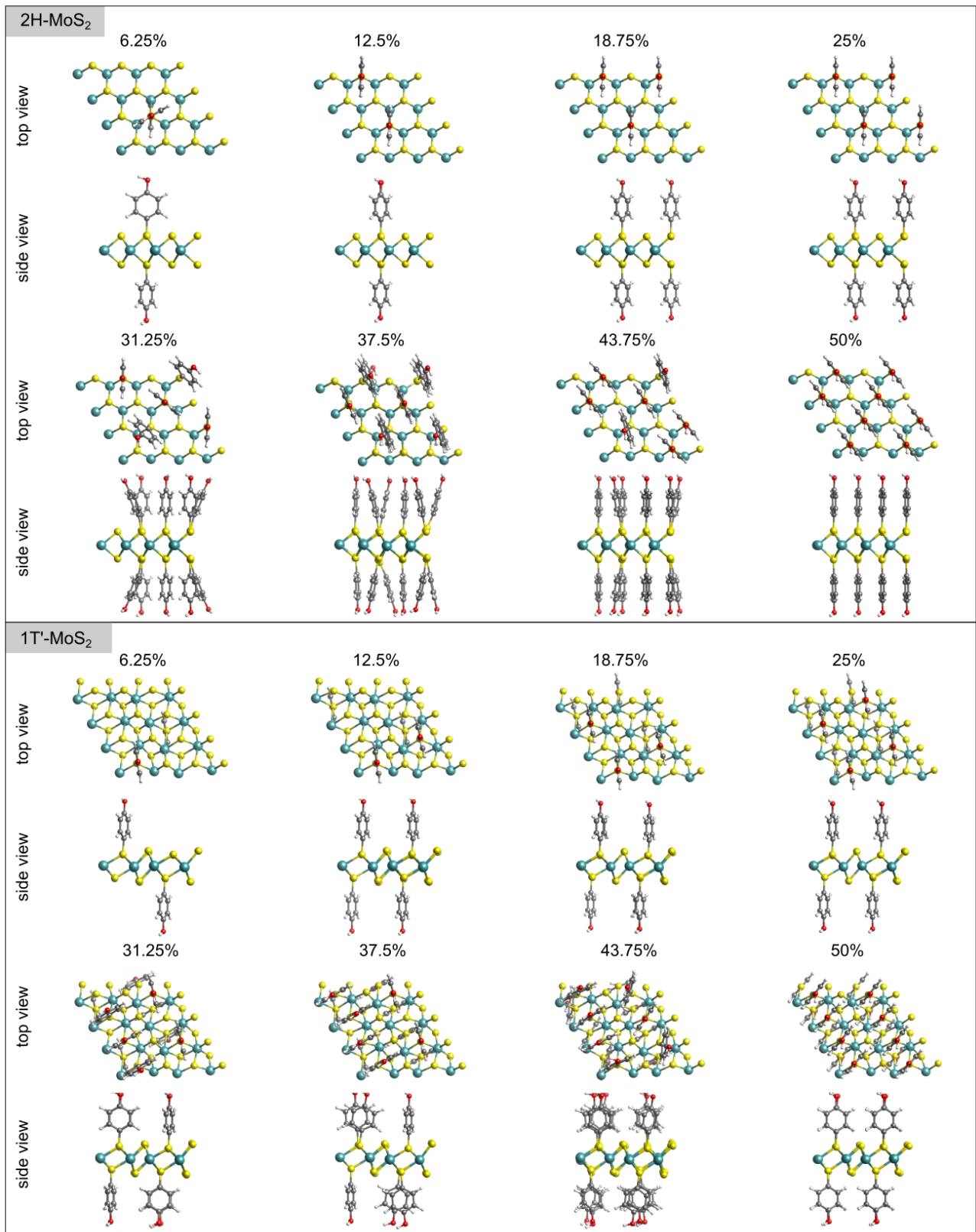
(f) -Ph



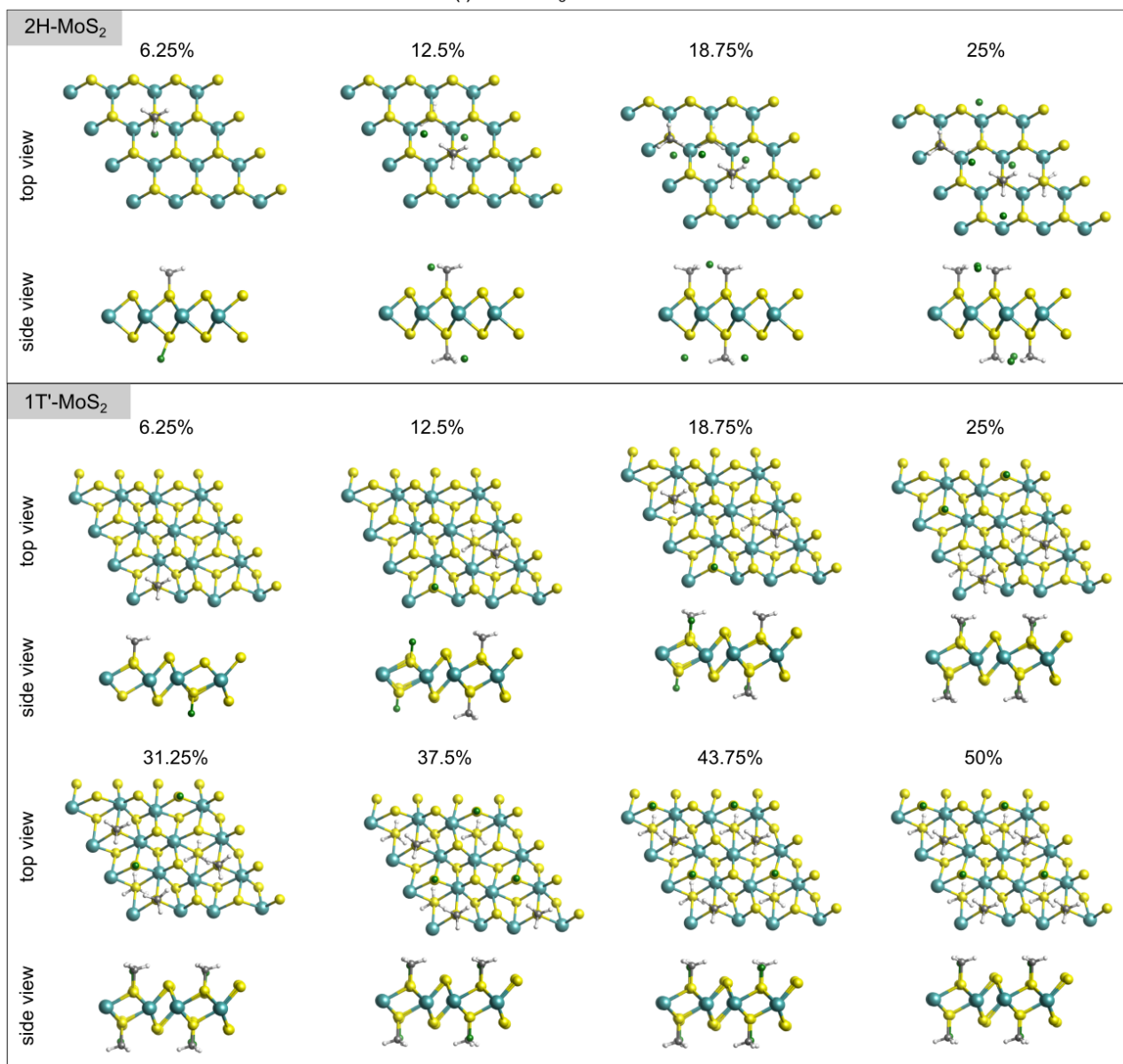
(g) -PhNO₂



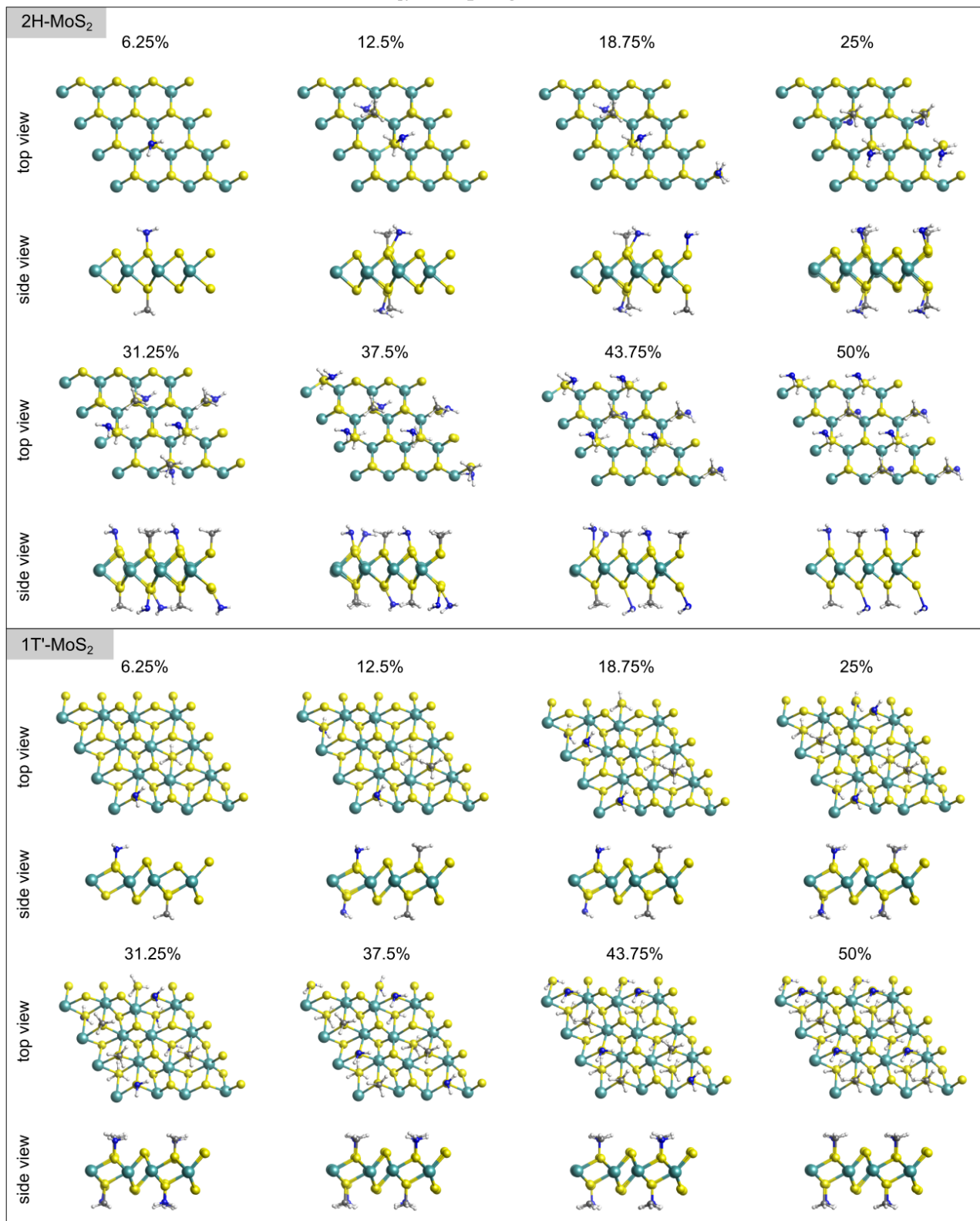
(h) -PhOH



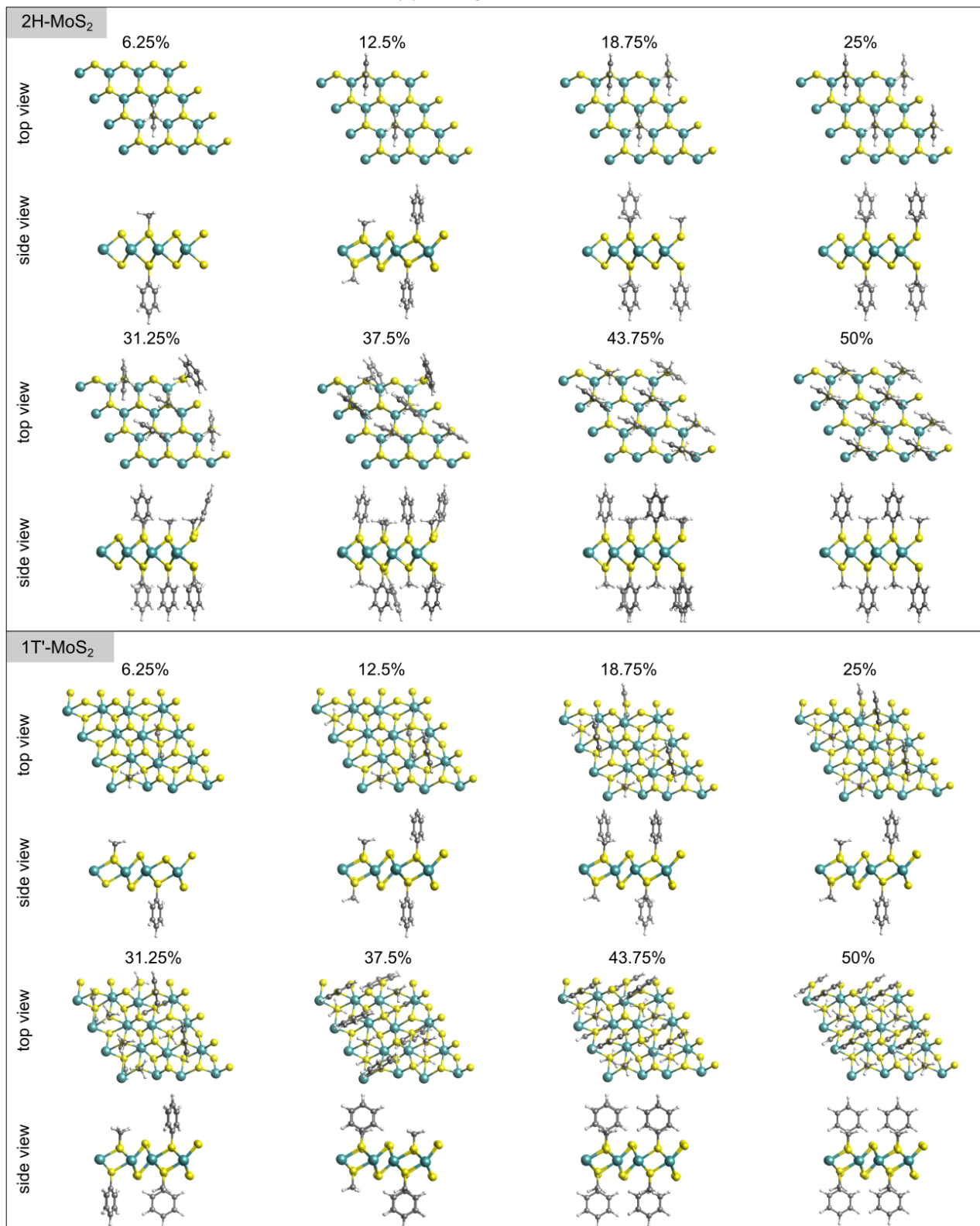
(i) m-F-CH₃



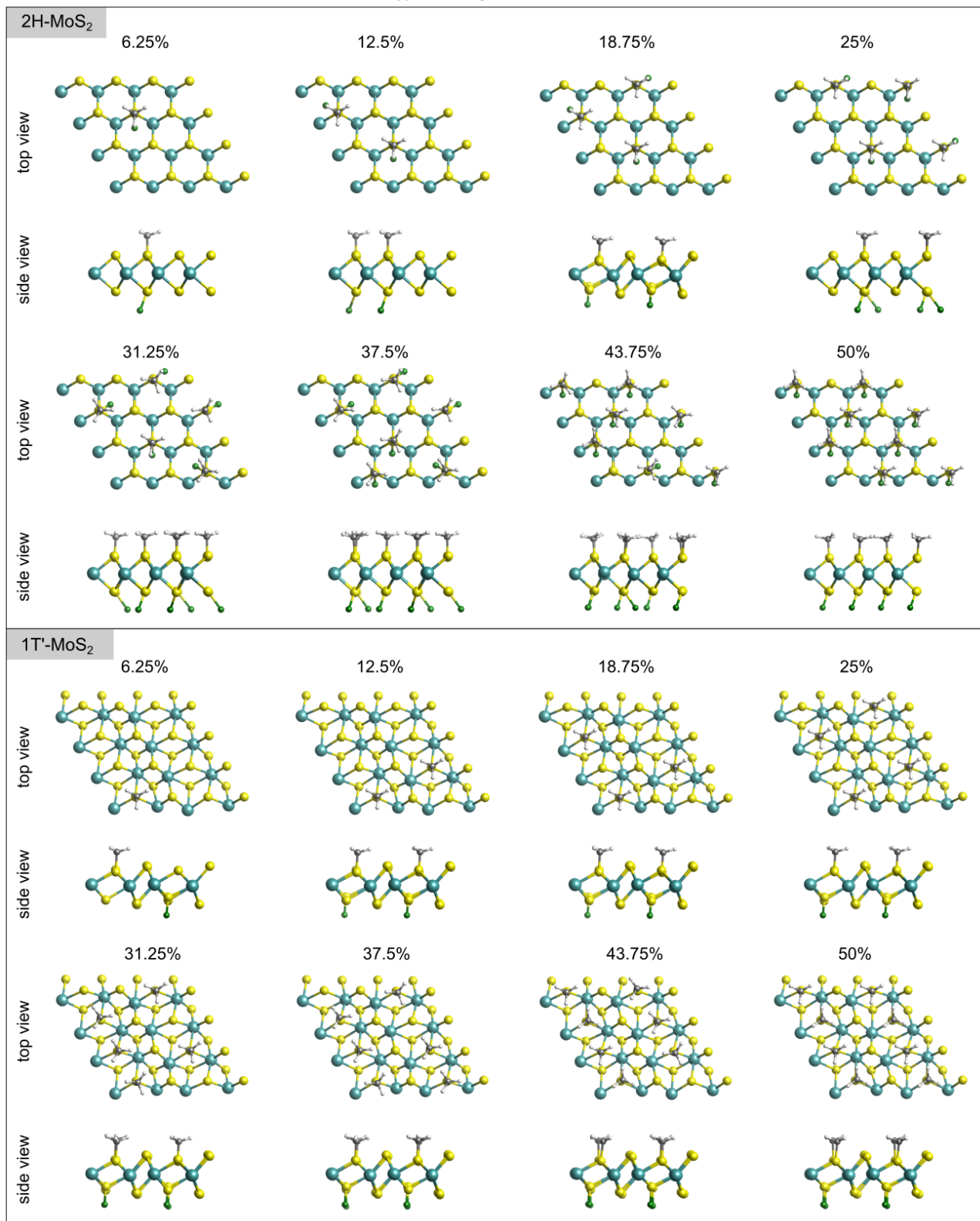
(j) m-NH₂-CH₃



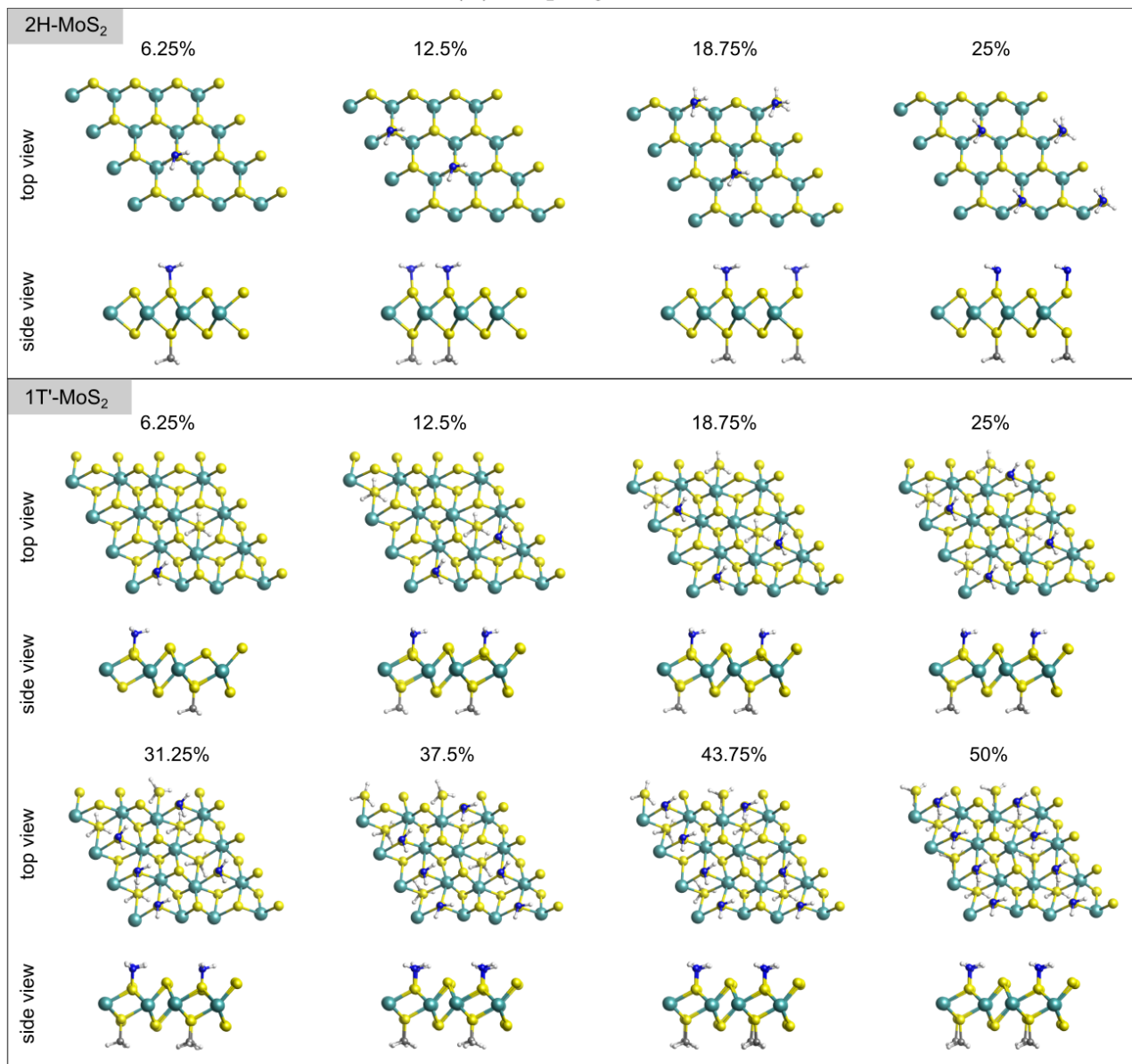
(k) m-CH₃-Ph



(I) s-F-CH₃



(m) s-NH₂-CH₃



(n) s-CH₃-Ph

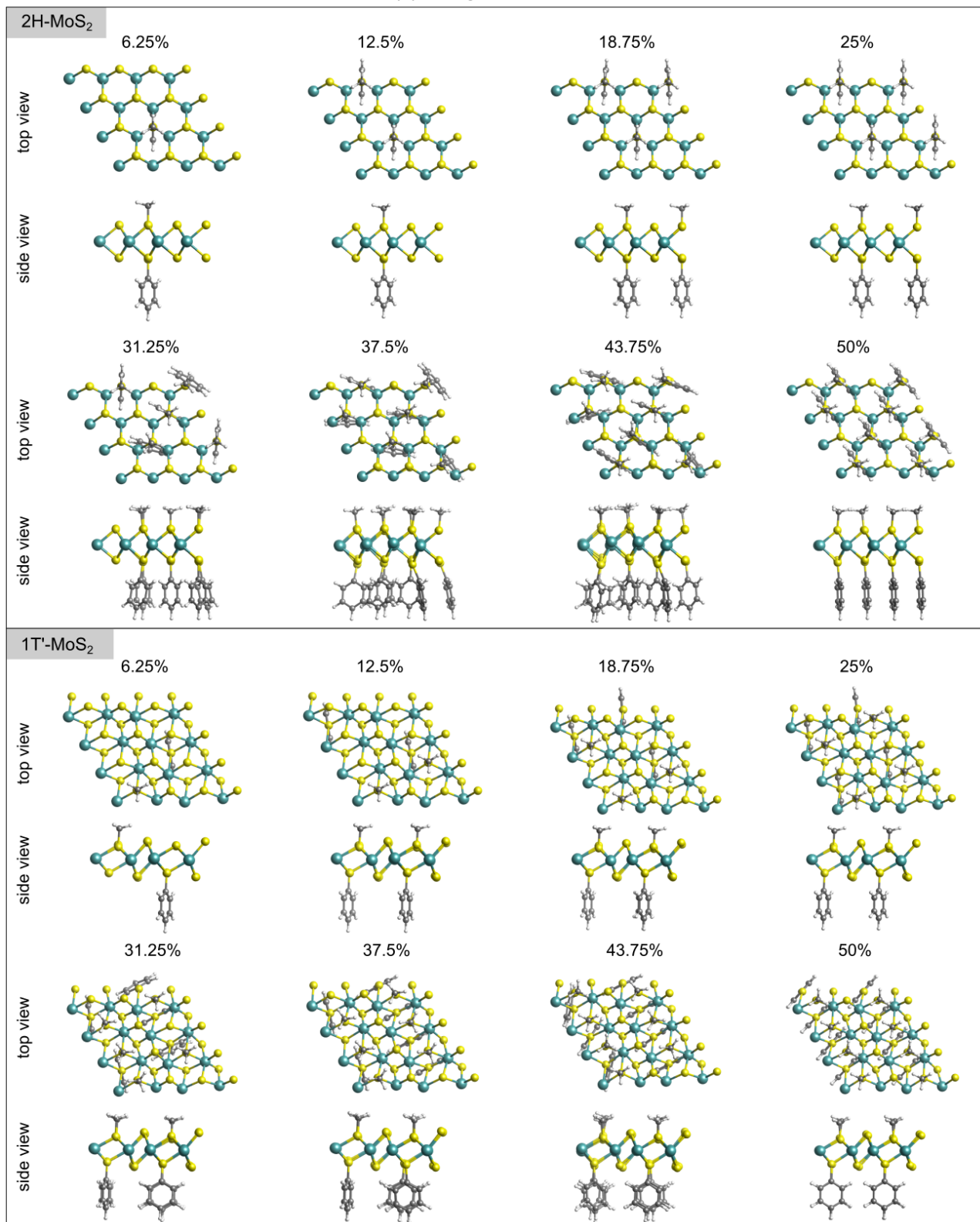


Fig. S3: Average energy difference ΔE between functionalized 2H- and 1T'-MoS₂ as a function of the group coverage. Here $\Delta E = (E_{1T'} - E_{2H})/m$, where m is the number of MoS₂ units, $E_{1T'}$ and E_{2H} are the total energy of functionalized 2H- and 1T'-MoS₂ for a given coverage, respectively. The positive ΔE indicates that functionalized 2H-MoS₂ is more stable, while the negative ΔE indicates that functionalized 1T'-MoS₂ is more stable.

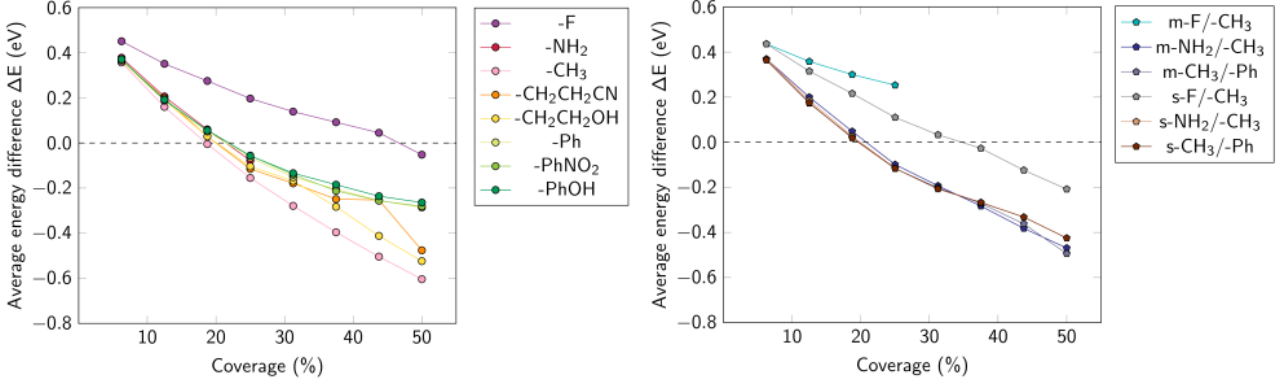


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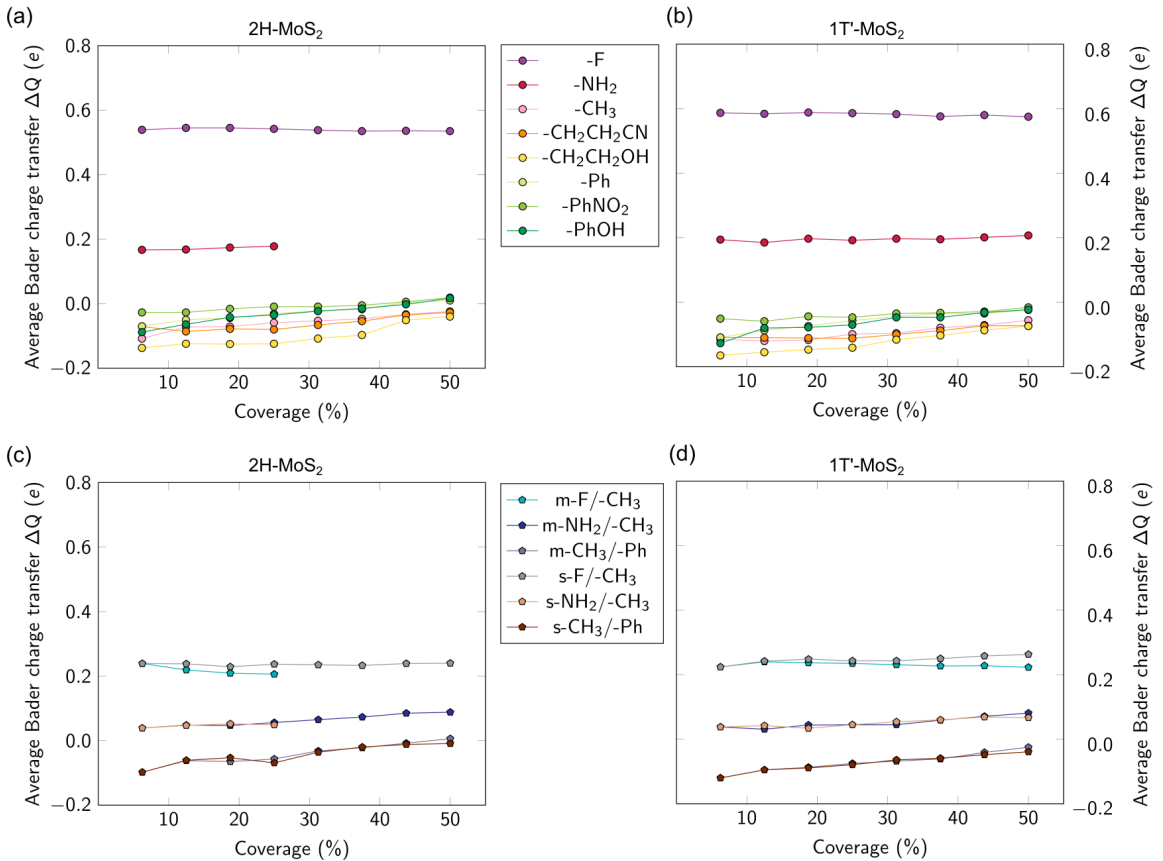


Fig. S5: DFT band gaps as a function of the group coverage with respect to the functionalized 2H-MoS₂ and 1T'-MoS₂ monolayers.

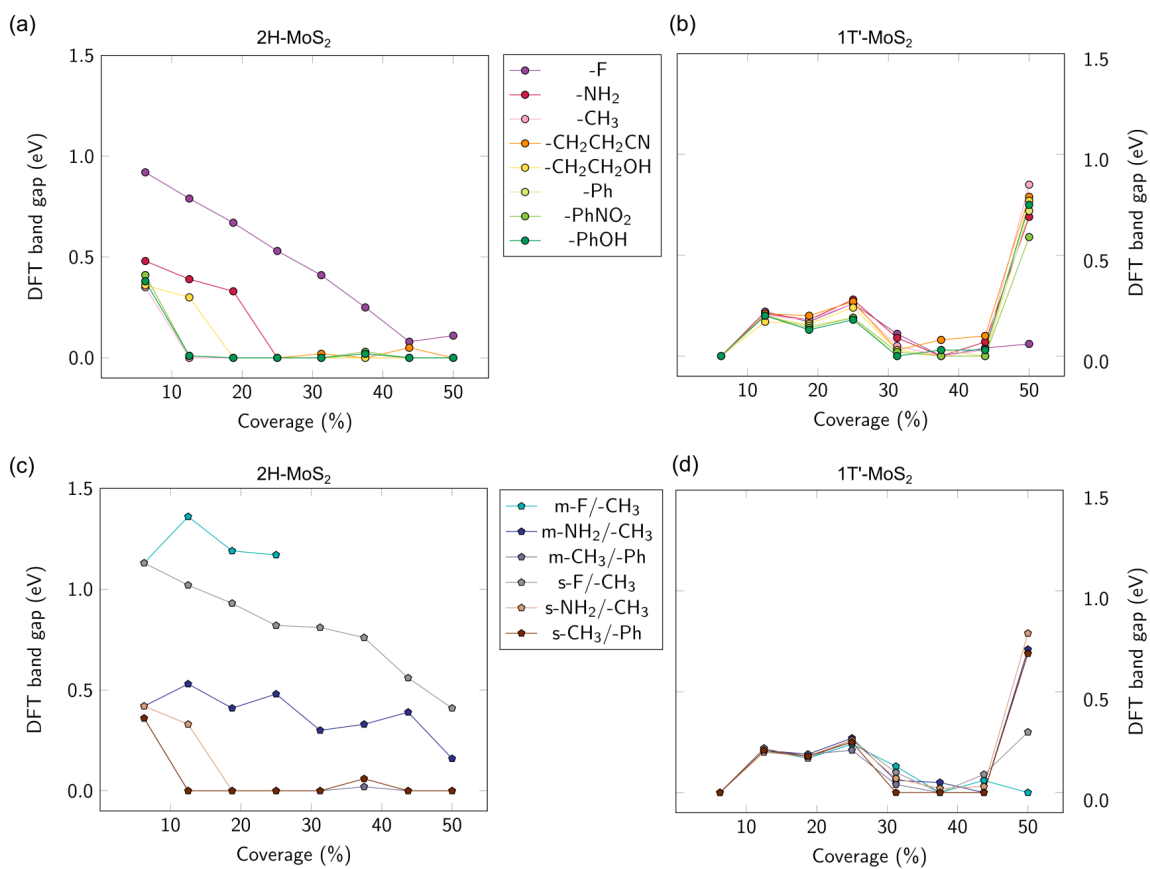
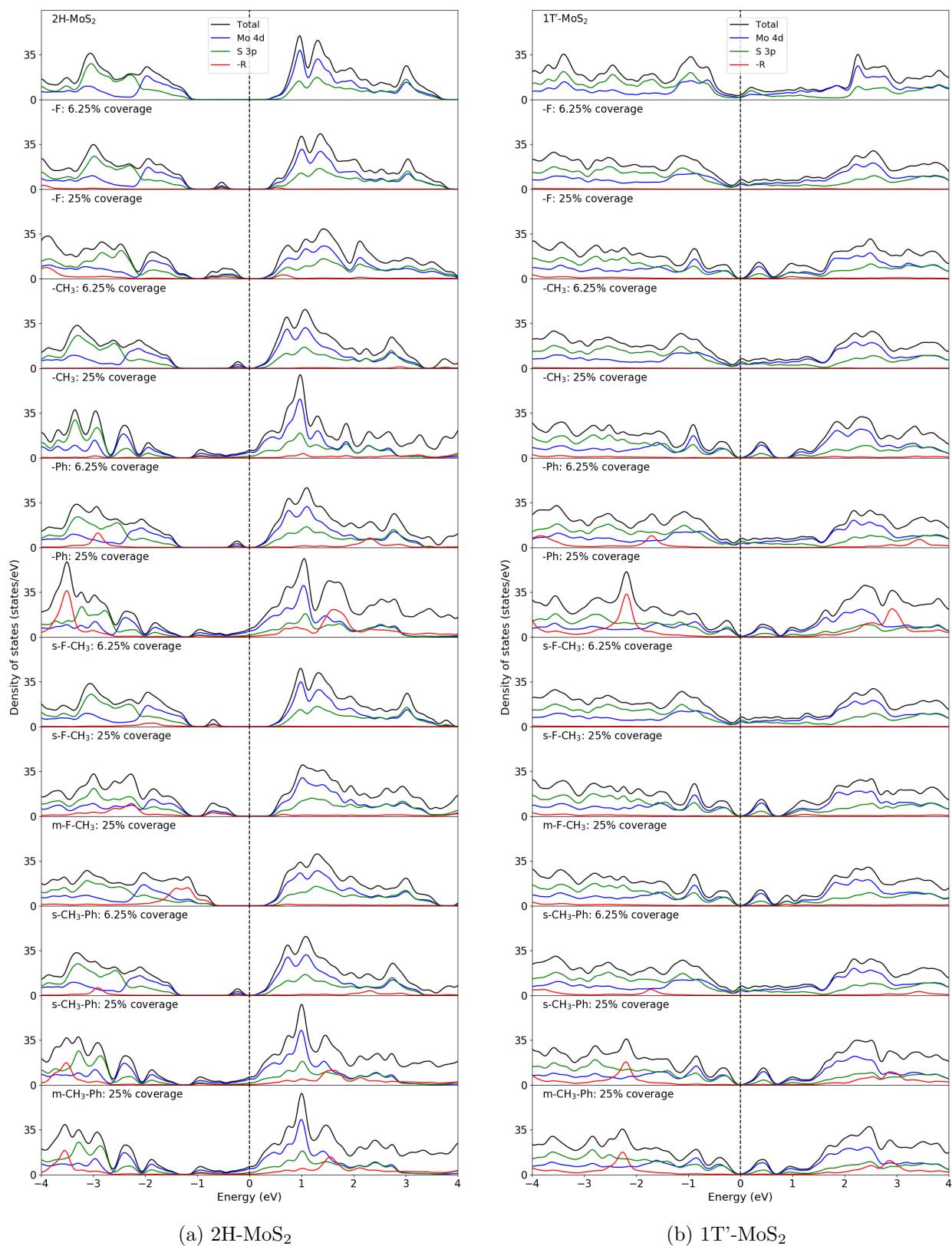


Fig. S6: Total density of states and the projected density of states for the functionalized (a) 2H-MoS₂ and (b) 1T'-MoS₂ calculated by PBE functional.



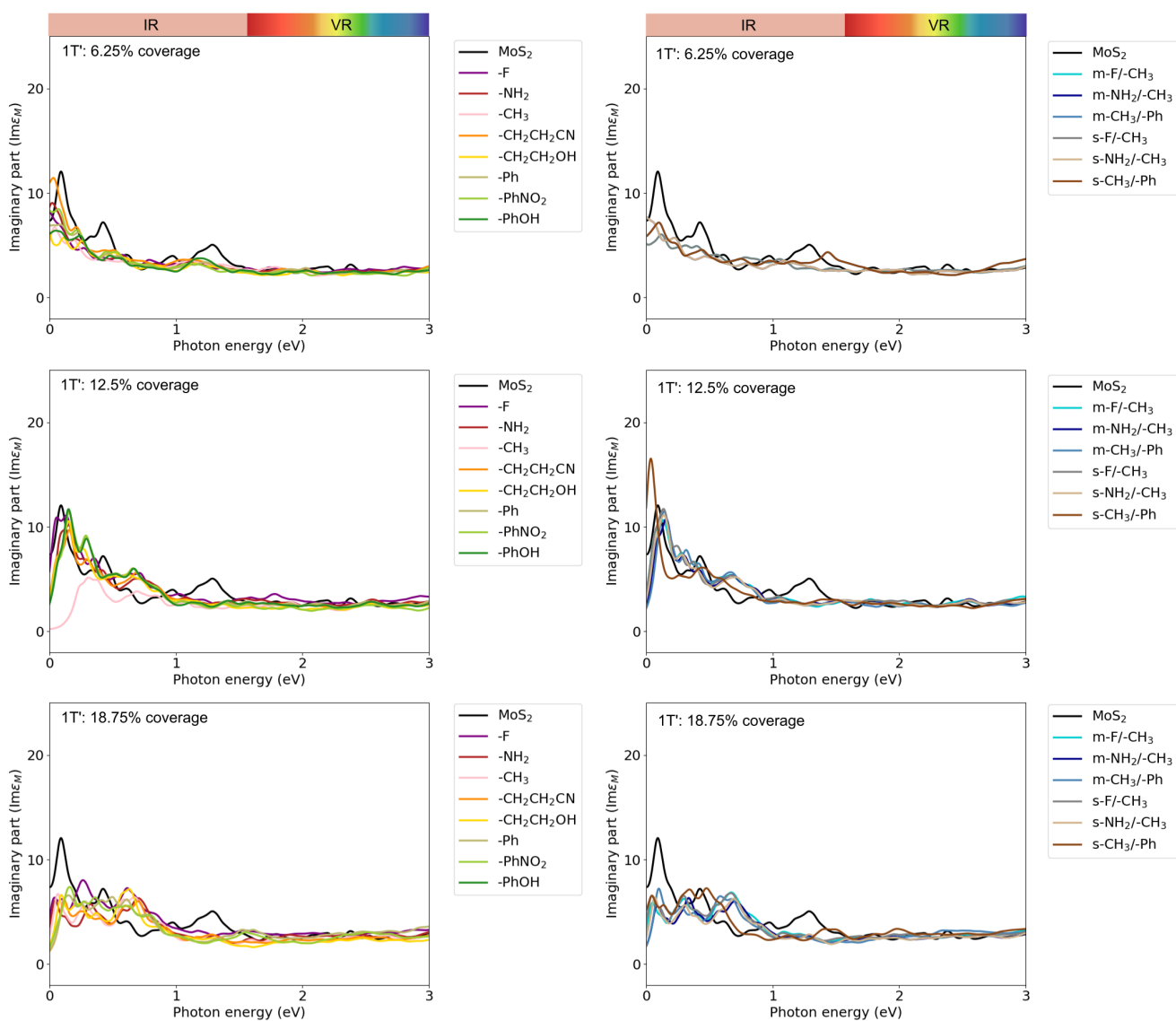


Fig. S7: Optical absorption spectra of the functionalized 1T'-MoS₂ monolayer.

Table S1: The optical band gap (E_{opt}) for the functionalized 2H-MoS₂ at 6.25% coverage. Energies are in eV.

	2H-MoS ₂	-F	-NH ₂	-CH ₃	-CH ₂ CH ₂ CN	-CH ₂ CH ₂ OH
E_{opt}	1.84	1.14	0.55	0.37	0.49	0.37
E_{ext}	0.67	0.67	0.65	0.63	0.60	0.64
	-Ph	-PhNO ₂	PhOH	m-F/-NH ₂	m-NH ₂ /-CH ₃	m-CH ₃ /-Ph
E_{opt}	0.37	0.43	0.37	1.49	0.48	0.17
E_{ext}	0.63	0.59	0.62	0.73	0.63	0.82
	s-F/-NH ₂	s-NH ₂ /-CH ₃	s-CH ₃ /-Ph			
E_{opt}	1.49	0.48	0.17			
E_{ext}	0.73	0.63	0.82			