

Supporting Information

One-step solvothermal synthesis of topological insulator Bi₂Te₃ nanorod-modified TiO₂ photocatalyst for enhanced H₂-evolution activity

Lin Dong,^a Xuefei Wang,^a Ping Wang,^{*a} and Huogen Yu ^{*ab}

^a School of Chemistry, Chemical Engineering and Life Sciences, Wuhan University of Technology, Wuhan 430070, PR China

^b Laboratory of Solar Fuel, Faculty of Materials Science and Chemistry, China University of Geosciences, Wuhan, 430074, PR China

*Corresponding authors. Tel: +86(27)87749379;

E-mail: wangping0904@whut.edu.cn (Ping Wang);

huogenyu@163.com (Huogen Yu)

Figure captions

Fig. S1. The photocatalytic H₂-evolution activity of Bi₂Te₃/TiO₂(0.5 wt%) in lactic acid (CA, 10 vol%), triethanolamine (TEOA, 10 vol%), and ethyl alcohol (EA, 25 vol%) of (a) TiO₂ and (b) Bi₂Te₃/TiO₂(0.5 wt%).

Fig. S2. (A) The XRD patterns and (B) UV-vis spectra of (a) TiO₂, (b) Bi₂Te₃/TiO₂(5.0 wt%) before photocatalytic H₂ evolution, and (c) Bi₂Te₃/TiO₂(5.0 wt%) after photocatalytic H₂ evolution.

Table S1. The ICP-OES results of Bi₂Te₃/TiO₂ samples.

Samples	Bi (wt%)	Te (wt%)	Molar ratio of Te/Bi	Actual ratio of Bi ₂ Te ₃ /TiO ₂ (wt%)
Bi ₂ Te ₃ /TiO ₂ (0.5 wt%)	0.2404	0.2165	1.47	0.45
Bi ₂ Te ₃ /TiO ₂ (5.0 wt%)	2.9553	2.5986	1.44	5.43

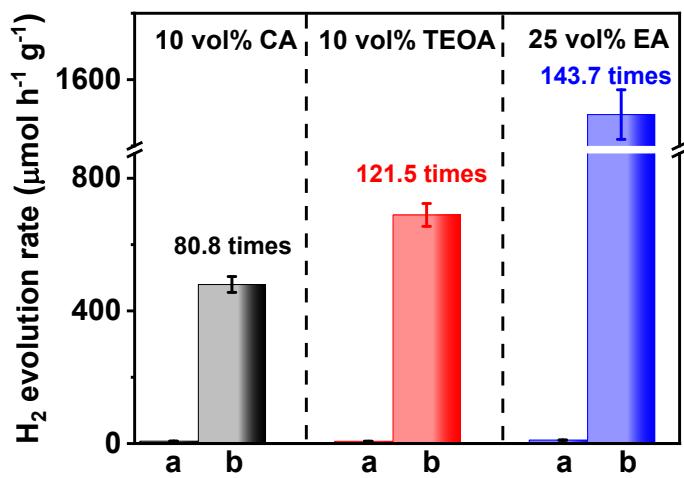


Fig. S1. The photocatalytic H_2 -evolution activity of $\text{Bi}_2\text{Te}_3/\text{TiO}_2$ (0.5 wt%) in lactic acid (CA, 10 vol%), triethanolamine (TEOA, 10 vol%), and ethyl alcohol (EA, 25 vol%) of (a) TiO_2 and (b) $\text{Bi}_2\text{Te}_3/\text{TiO}_2$ (0.5 wt%).

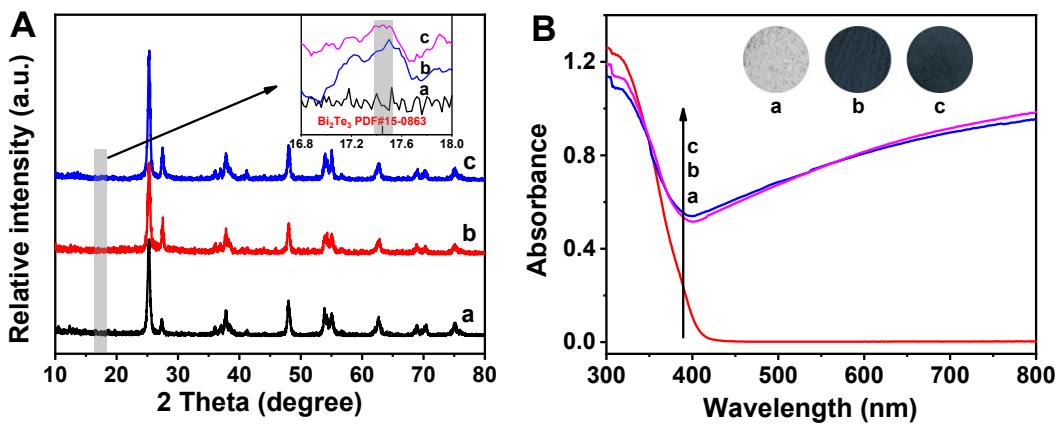


Fig. S2. (A) The XRD patterns and (B) UV-vis spectra of (a) TiO₂, (b) Bi₂Te₃/TiO₂(5.0 wt%) before photocatalytic H₂ evolution, and (c) Bi₂Te₃/TiO₂(5.0 wt%) after photocatalytic H₂ evolution.