

1 Electronic Supplementary Information for  
2 Theoretical Prediction of the Carrier  
3 Mobilities for  $M^{II}_2M^{III}-Cl$ -Layered Double  
4 Hydroxides in the Three-Dimensional  
5 Directions

6 Si-Min Xu,<sup>‡\*a</sup> Yu-Quan Zhu,<sup>‡a</sup> Zi-Ru Chen,<sup>\*a</sup> Jiang-Rong Yang,<sup>a</sup> Xudong Chen<sup>\*b</sup>  
7 and Hong Yan<sup>a</sup>

8 <sup>a</sup> State Key Laboratory of Chemical Resource Engineering, Beijing University of  
9 Chemical Technology, Beijing 100029, P. R. China.

10 <sup>b</sup> Key Laboratory of Weak Light Nonlinear Photonics, Ministry of Education, School  
11 of Physics, Nankai University, Tianjin 300071, P. R. China.

12 <sup>‡</sup> S.-M. Xu, and Y.-Q. Zhu contributed equally to this work.

13 Corresponding authors

14 [2019210630@mail.buct.edu.cn](mailto:2019210630@mail.buct.edu.cn) (Z.-R. Chen),

15 [chenxd@email.tjut.edu.cn](mailto:chenxd@email.tjut.edu.cn) (X. Chen),

16 [xsm713@sina.com](mailto:xsm713@sina.com), [Si-MinXu@mail.buct.edu.cn](mailto:Si-MinXu@mail.buct.edu.cn) (S.-M. Xu).

## Contents

Item	Title	Pagination
Supplementary Table 1	Band Gap Energy, Work Function, and Band Edge Placements of $M^{II}_2M^{III}-Cl-LDHs$ under Lattice Dilation and Compression in $x$ and $y$ Directions	1
Supplementary Table 2	Transfer Integrals, Reorganization Energies, Gibbs Free Energy Changes, Transfer Rates, and Carrier Mobilities for $[Mg_2Ga(OH)_6(OH_2)_7]^+$ using Different Exchange-Correlation Functionals	3
Supplementary Table 3	Transfer Integrals, Reorganization Energies, Gibbs Free Energy Changes, Transfer Rates, and Carrier Mobilities for $Mg_2Ga-Cl-LDH$ Matrix Clusters with Different Sizes	4
Supplementary Figure 1	Phonon dispersion curves of $M^{II}_2M^{III}-Cl-LDHs$	5
Supplementary Figure 2	Band structures of $Mg_2Fe-Cl-LDH$ , $Ni_2Fe-Cl-LDH$ , $Ni_2Ga-Cl-LDH$ , and $Zn_2Fe-Cl-LDH$ without the Hubbard correction	6
Supplementary Figure 3	Linear fitting of the conduction band minimum and valence band maximum shift with respect to the lattice dilation and compression in the $y$ direction for $M^{II}_2M^{III}-Cl-LDHs$	7

**Table S1.** Band Gap Energies ( $E_g$ ), Work Functions ( $W$ ), Valence Band Maximum ( $E_{\text{VBM}}$ ), and Conduction Band Minimum ( $E_{\text{CBM}}$ ) of  $\text{Mg}_2\text{Fe-Cl-LDH}$ ,  $\text{Mg}_2\text{Ga-Cl-LDH}$ ,  $\text{Ni}_2\text{Fe-Cl-LDH}$ ,  $\text{Ni}_2\text{Ga-Cl-LDH}$ ,  $\text{Zn}_2\text{Fe-Cl-LDH}$ , and  $\text{Zn}_2\text{Ga-Cl-LDH}$  under Lattice Dilation and Compression in  $x$  and  $y$  Directions

model	direction	$\Delta l / l_0$	$E_g$ (eV)	$W$ (eV)	$E_{\text{VBM}}$ (eV)	$E_{\text{CBM}}$ (eV)
$\text{Mg}_2\text{Fe-Cl-LDH}$	$x$	-1%	1.819	4.685	-5.459	-3.637
		-0.5%	1.823	4.548	-5.529	-3.709
		0%	1.832	4.753	-5.695	-3.775
		0.5%	1.879	4.589	-5.769	-3.887
		1%	1.889	4.919	-5.864	-3.974
	$y$	-1%	1.791	4.672	-5.631	-3.819
		-0.5%	1.800	4.613	-5.567	-3.777
		0%	1.812	4.725	-5.537	-3.713
		0.5%	1.827	4.553	-5.513	-3.671
		1%	1.827	4.624	-5.466	-3.640
$\text{Mg}_2\text{Ga-Cl-LDH}$	$x$	-1%	4.169	4.753	-4.844	-0.885
		-0.5%	4.159	4.764	-4.838	-0.878
		0%	4.146	4.729	-4.802	-0.868
		0.5%	4.127	4.728	-4.792	-0.865
		1%	4.116	4.726	-4.784	-0.856
	$y$	-1%	4.167	4.759	-4.843	-0.876
		-0.5%	4.143	4.749	-4.820	-0.878
		0%	4.125	4.742	-4.804	-0.879
		0.5%	4.097	4.739	-4.788	-0.882
		1%	4.075	4.717	-4.755	-0.891
$\text{Ni}_2\text{Fe-Cl-LDH}$	$x$	-1%	1.910	4.660	-5.615	-3.421
		-0.5%	1.887	4.433	-5.499	-3.489
		0%	1.875	4.359	-5.407	-3.605
		0.5%	1.859	4.669	-5.377	-3.639
		1%	1.843	4.665	-5.297	-3.743
	$y$	-1%	1.695	4.684	-5.531	-3.837
		-0.5%	1.695	4.631	-5.478	-3.784
		0%	1.669	4.487	-5.421	-3.703
		0.5%	1.662	4.459	-5.361	-3.653
		1%	1.638	4.482	-5.290	-3.628
$\text{Ni}_2\text{Ga-Cl-LDH}$	$x$	-1%	2.554	4.746	-6.024	-4.169
		-0.5%	2.556	4.746	-5.973	-4.108
		0%	2.571	4.556	-5.941	-4.071
		0.5%	2.572	4.554	-5.860	-4.008

		1%	2.573	4.542	-5.829	-3.955
		-1%	2.531	4.598	-6.286	-4.048
		-0.5%	2.545	4.576	-6.273	-4.033
	<i>y</i>	0%	2.545	4.576	-6.261	-4.024
		0.5%	2.538	4.617	-6.259	-4.014
		1%	2.547	4.578	-6.248	-4.005
		-1%	2.187	4.724	-5.818	-3.630
		-0.5%	2.176	4.763	-5.839	-3.645
	<i>x</i>	0%	2.197	4.772	-5.851	-3.653
		0.5%	2.184	4.747	-5.861	-3.665
		1%	2.174	4.764	-5.871	-3.677
Zn <sub>2</sub> Fe-Cl-LDH		-1%	1.485	4.696	-5.399	-4.003
		-0.5%	1.355	4.747	-5.424	-4.025
	<i>y</i>	0%	1.375	4.712	-5.439	-4.070
		0.5%	1.390	4.789	-5.466	-4.070
		1%	1.396	4.768	-5.484	-4.094
		-1%	3.131	4.813	-4.879	-1.518
		-0.5%	3.135	4.793	-4.861	-1.547
	<i>x</i>	0%	3.093	4.798	-4.852	-1.536
		0.5%	3.131	4.785	-4.850	-1.520
		1%	3.132	4.786	-4.844	-1.520
Zn <sub>2</sub> Ga-Cl-LDH		-1%	3.109	4.803	-4.857	-1.549
		-0.5%	3.118	4.797	-4.856	-1.538
	<i>y</i>	0%	3.116	4.788	-4.846	-1.535
		0.5%	3.111	4.790	-4.845	-1.530
		1%	3.109	4.783	-4.838	-1.528

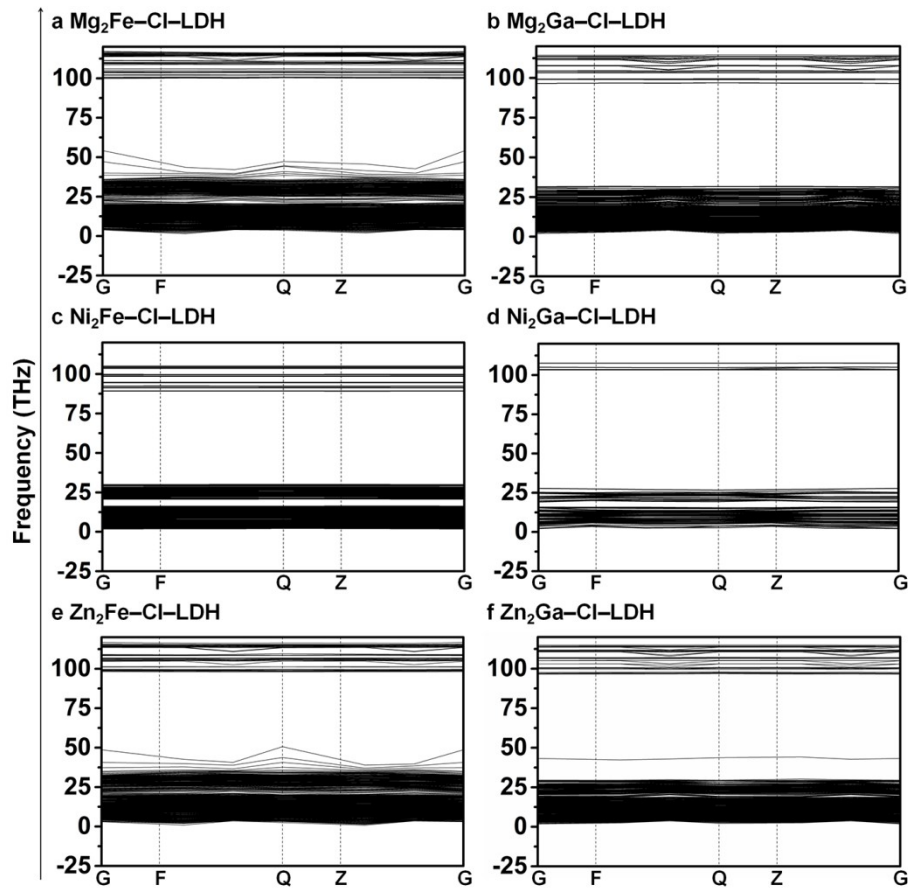
---

**Table S2.** Transfer Integral  $V$ , Reorganization Energy  $\lambda$ , Gibbs Free Energy Change  $\Delta G$ , Transfer Rate  $k$ , and Carrier Mobility  $\mu$  for  $[\text{Mg}_2\text{Ga}(\text{OH})_6(\text{OH}_2)_7]^+$  Using Different Exchange-Correlation Functionals

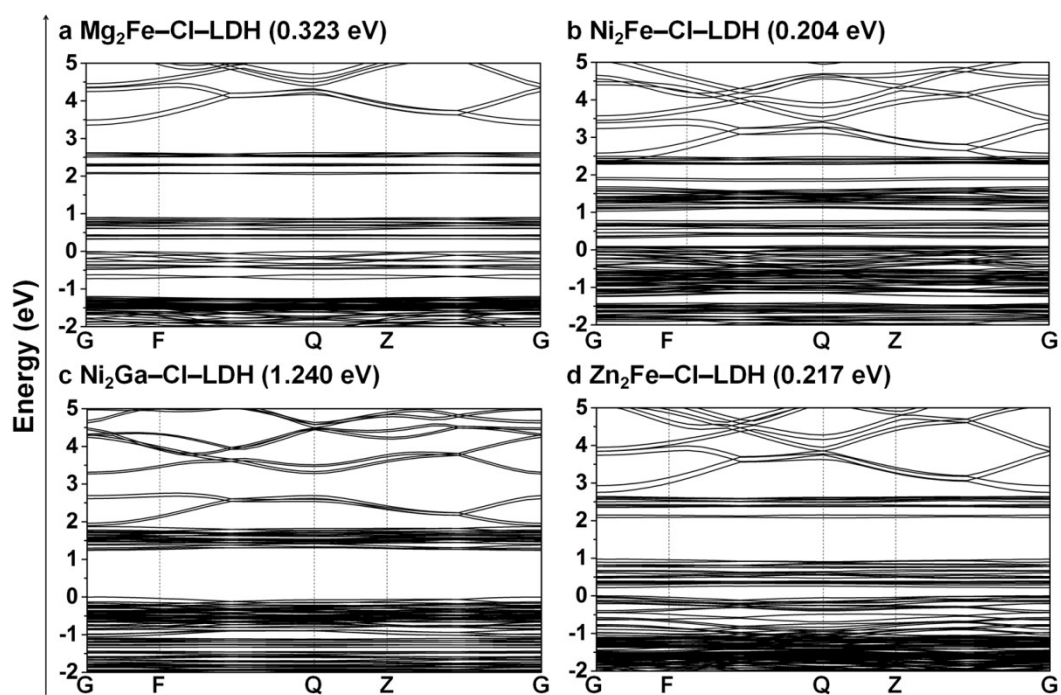
exchange-correlation functional	charge carrier	$V$ (eV)	$\lambda$ (eV)	$\Delta G$ (eV)	$k$ ( $\text{s}^{-1}$ )	$\mu$ ( $\text{cm}^2 \cdot \text{V}^{-1} \cdot \text{s}^{-1}$ )
B3LYP	e h	0.049	2.20	6.54	$4 \times 10^{-133}$	$5 \times 10^{-164}$
M06	e h	0.030	2.30	4.56	$2 \times 10^{-73}$	$2 \times 10^{-104}$
PBE	e h	0.088	2.61	3.75	$6 \times 10^{-52}$	$9 \times 10^{-83}$

**Table S3.** Transfer Integrals  $V$ , Reorganization Energies  $\lambda$ , Gibbs Free Energy Changes  $\Delta G$ , Transfer Rates  $k$ , and Carrier Mobilities  $\mu$  for Carrier Transportation in Mg<sub>2</sub>Ga-Cl-LDH Matrix Clusters with Different Sizes

chemical formula	charge carrier	$V$ (eV)	$\lambda$ (eV)	$\Delta G$ (eV)	$k$ (s <sup>-1</sup> )	$\mu$ (cm <sup>2</sup> ·V <sup>-1</sup> ·s <sup>-1</sup> )
[Mg <sub>2</sub> Al(OH) <sub>6</sub> (OH <sub>2</sub> ) <sub>7</sub> ]·Cl	e h	0.049	2.20	6.54	$4 \times 10^{-133}$	$5 \times 10^{-164}$
[Mg <sub>4</sub> Al <sub>2</sub> (OH) <sub>12</sub> (OH <sub>2</sub> ) <sub>10</sub> ]·2C 1	e h	0.693	1.75	3.44	$1 \times 10^{-49}$	$2 \times 10^{-80}$
[Mg <sub>6</sub> Al <sub>3</sub> (OH) <sub>18</sub> (OH <sub>2</sub> ) <sub>12</sub> ]·3C 1	e h	0.715	3.27	3.18	$2 \times 10^{-38}$	$2 \times 10^{-69}$

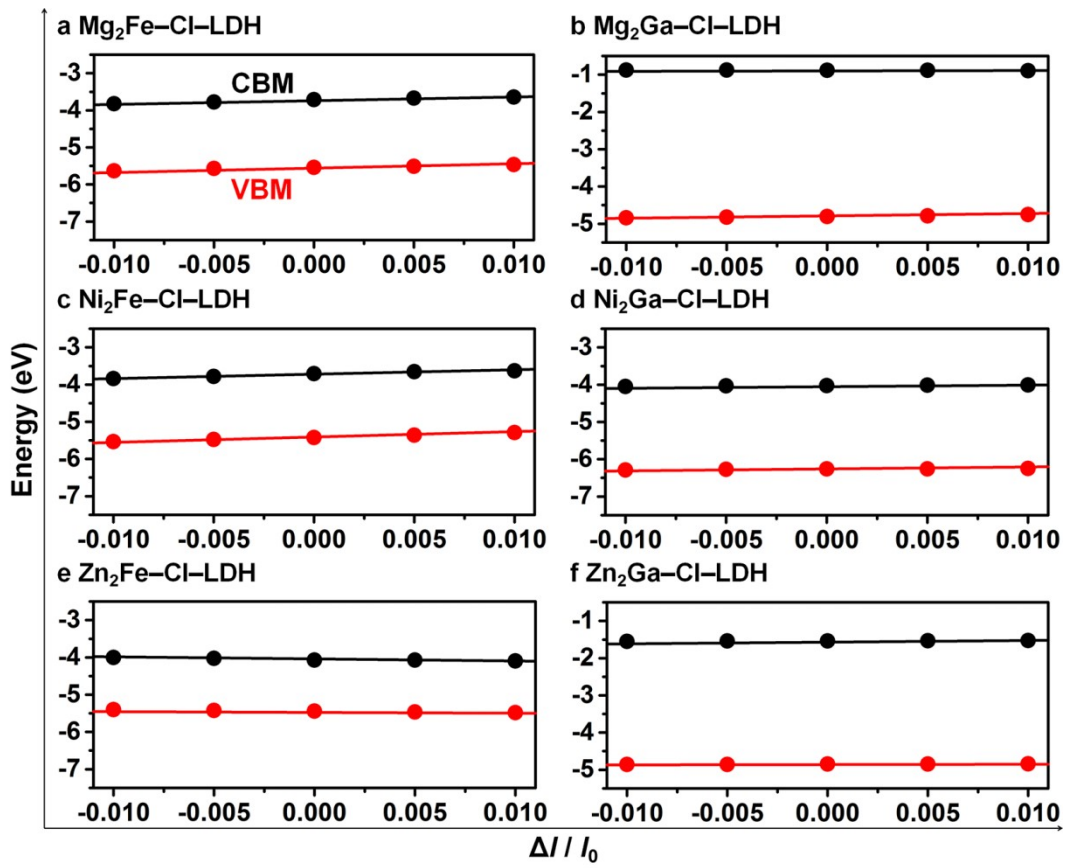


**Figure S1.** Phonon dispersion curves of (a)  $\text{Mg}_2\text{Fe}-\text{Cl}-\text{LDH}$ , (b)  $\text{Mg}_2\text{Ga}-\text{Cl}-\text{LDH}$ , (c)  $\text{Ni}_2\text{Fe}-\text{Cl}-\text{LDH}$ , (d)  $\text{Ni}_2\text{Ga}-\text{Cl}-\text{LDH}$ , (e)  $\text{Zn}_2\text{Fe}-\text{Cl}-\text{LDH}$ , and (f)  $\text{Zn}_2\text{Ga}-\text{Cl}-\text{LDH}$ , respectively.



**Figure S2.** Band structures of  $\text{Mg}_2\text{Fe-Cl-LDH}$ ,  $\text{Ni}_2\text{Fe-Cl-LDH}$ ,  $\text{Ni}_2\text{Ga-Cl-LDH}$ , and  $\text{Zn}_2\text{Fe-Cl-LDH}$  without the Hubbard correction. The band gap energy of each LDH is listed in the bracket.





**Figure S3.** Linear fitting of the conduction band minimum and valence band maximum shift ( $\Delta\nu$ ) with respect to the lattice dilation and compression ( $\Delta/l_0$ ) in the  $y$  direction for (a) Mg<sub>2</sub>Fe-Cl-LDH, (b) Mg<sub>2</sub>Ga-Cl-LDH, (c) Ni<sub>2</sub>Fe-Cl-LDH, (d) Ni<sub>2</sub>Ga-Cl-LDH, (e) Zn<sub>2</sub>Fe-Cl-LDH, and (f) Zn<sub>2</sub>Ga-Cl-LDH, respectively.