

*Electronic Supplementary Information for*

**Design of potential singlet fission chromophores based on diketofurofuran: an alternative to diketopyrrolopyrrole**

Li Shen,<sup>a</sup> Qian Wu,<sup>a</sup> Jitao Lu,<sup>a</sup> Haitao Zhao,<sup>b</sup> Heyuan Liu,<sup>c</sup> Qingguo Meng,<sup>a,\*</sup> Xiyu Li<sup>d,\*</sup>

<sup>a</sup>College of Chemical Engineering and Environmental Chemistry, Weifang University, Weifang 261061, China

<sup>b</sup>Shandong Provincial Key Laboratory of Chemical Energy Storage and Novel Cell Technology, Liaocheng University, Liaocheng, 252059, China

<sup>c</sup>College of New Energy, China University of Petroleum (East China), Qingdao, 266580, China

<sup>d</sup>School of Materials Science and Engineering, China University of Petroleum (East China), Qingdao, 266580, China

To whom correspondence should be addressed: [mengqg@wfu.edu.cn](mailto:mengqg@wfu.edu.cn),  
[xiyouli@upc.edu.cn](mailto:xiyouli@upc.edu.cn)

## Contents

1. Comparisons of the excitation energies calculated with different functionals in toluene (Fig. S1, Table S1-2).
2. Solvation effects and basis set effect on  $E(S_1)$  and  $E(T_1)$  of DFF derivatives (Fig. S1, Table S3-4)
3. Effect of electronic coupling in 1B-DFF dimers on  $E(S_1)_v$  and  $E(T_1)_v$  (Fig. S1, Table S5)
4. Bond lengths (Å) and torsion Angles (deg) of along the conjugated backbones of  $\alpha$ Me<sub>2</sub>N-1F-DFF and  $\alpha$ MeO-1F-DFF (Fig. S1, Table S6).
5. Comparisons of the calculated results for F3B-DFF, Pa-DFF and 3B-DFF (Table S7)
6. Hammett constant ( $\sigma_p$ ) of the introduced terminal substituents (Table S8)
7.  $E(S_1)_v$ ,  $E(S_1)$  and  $E(T_1)$  as a function of HLG in DFF derivatives with different aromatic substituents in toluene (Fig. S2).
8. Diradical character  $y_0$  as a function of HLG in DFF derivatives with different aromatic substituents (Fig. S3).
9. Energy differences between 1F-DFF derivatives and the pristine 1F-DFF in  $E(S_1)$ ,  $2E(T_1)$  and  $E(T_2)$  with respect to the Hammett constant ( $\sigma_p$ ) of the attached terminal substituents in toluene (Fig. S4).
10. HOMO and LUMO energies for  $\alpha$ - and  $\beta$ -substitutions in 1F-DFF derivatives with respect to the Hammett constant ( $\sigma_p$ ) of the attached terminal substituents (Fig. S5).
11. Electron density distributions of HOMO and LUMO in  $\alpha$ NO<sub>2</sub>-1F-DFF,  $\alpha$ MeO-1F-DFF,  $\beta$ NO<sub>2</sub>-1F-DFF and  $\beta$ MeO-1F-DFF (Fig. S6).
12. HOMO, LUMO energy levels and HOMO-LUMO gap (HLG) values (in eV), diradical character  $y_0$  and tetraradical character  $y_1$  (Table S9).
13. Vertical excitation energies ( $E(S_1)_v$  and  $E(T_2)_v$ ), adiabatic excitation energies ( $E(S_1)$ ,  $E(T_1)$  and  $E(T_2)$ ) and the corresponding  $\Delta E_{SF(v)}$ ,  $\Delta E_{SF}$ ,  $\Delta E_{TTA(v)}$ ,  $\Delta E_{TTA}$  in toluene (Table S10).
14. Coordinates of the four dimers.

Because of the donor-acceptor-donor structures for all the compounds, different functionals were used to evaluate the effect of charge-transfer character on SF relevant excited states. These functionals include B3LYP, CAM-B3LYP,  $\omega$ B97XD, M06-2X and tuned LC-BLYP. The solvation effects were evaluated by polarizable continuum model (PCM) with toluene and chloroform used as solvents. Compared to available experimental results, the functional CAM-B3LYP,  $\omega$ B97XD, M06-2X and tuned LC-BLYP overestimate  $E(S_1)_v$  and give larger  $E(S_1)$  simultaneously (Table S1). The range separating parameters  $\mu$  were tuned for DFF derivatives (Table S2) with Quantum program optDFTw.<sup>a</sup> The functional CAM-B3LYP and  $\omega$ B97XD show similar results and larger  $E(T_1)$  than the functional B3LYP. The calculated  $E(S_1)_v$  with functional B3LYP are more close to the experimental values for both **1B-DFF** and **Br-1F-DFF**.<sup>b</sup> Moreover, the basis set was extended to larger basis set, including 6-311G\*\* and cc-pVTZ. The calculated results show that solvation effects affect the SF relevant excited state energies (Table S3), while the excitation energies are insensitive to the extension of basis set (Table S4).

Singlet fission is an intrinsically multi-molecular process, and electronic coupling among adjacent molecules may affect the SF relevant excited energies. We took four dimers in which the two monomer units show different distances and relative orientations in 1B-DFF crystal. We calculated  $E(S_1)_v$  and  $E(T_1)_v$  for these dimers at the theoretical level of  $\omega$ B97xd/6-311G\* based on the original geometries without optimizations. Due to electronic coupling,  $S_1$  state in the monomer splits into a higher and a lower excited states in dimers. Compared to 1B-DFF monomer, the energy shift caused by molecular interactions is rather small (Table S5). The maximum energy shifts are  $\sim 0.09$  eV for  $E(S_1)_v$  and  $\sim 0.03$  eV for  $E(T_1)_v$ . Likewise, in a covalent TDPP dimer with H-aggregate character, the energy difference of  $S_1$  state between monomer and dimer is only 0.03 eV.<sup>c</sup> It is reasonable that molecular interactions make negligible effect on SF relevant excited energies in DFF dimers. In addition, the functional  $\omega$ B97XD was also used to obtain the relaxed geometries of  $S_1$ ,  $T_1$  and  $T_2$  states of  $\alpha$ MeO-1F-DFF and  $\alpha$ Me<sub>2</sub>N-1F-DFF in toluene. Changing from B3LYP to  $\omega$ B97XD, there exists negligible variations in bond length ( $< 0.007$  Å) and dihedral angles ( $< 0.3^\circ$ ) (Fig. 1S, Table S6). As a result, it is reasonable to evaluate SF feasibility of DFF derivatives at the theoretical level of B3LYP/6-311G\*.

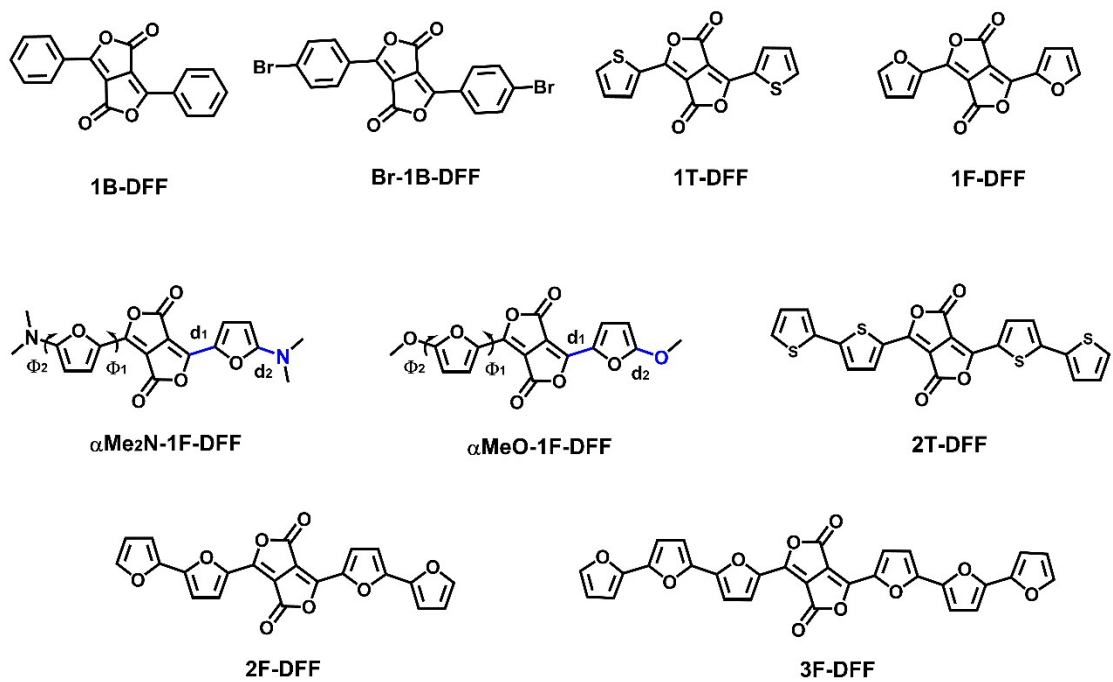


Fig. S1 Molecular structures of the involved molecules for test.

**Table S1** Vertical and adiabatic excitation energies with different functionals in toluene.

		B3LYP	Cam- B3LYP	M06-2X	$\omega$ B97XD	Tuned-LC- BLYP <sup>1</sup>	Exp <sup>b</sup>
<b><i>1B</i>-DFF</b>	E(S <sub>1</sub> ) <sub>v</sub>	2.720	3.051	3.055	3.062	2.849	2.725
	E(S <sub>1</sub> )	2.548	2.804	2.808	2.815		
	E(T <sub>1</sub> )	1.385	1.419	1.599	1.457		
	E(T <sub>2</sub> ) <sub>v</sub>	2.739	2.808	3.115	2.902		
	E(T <sub>2</sub> )	2.529	2.600	2.862	2.692		
<b><i>Br-1B</i>- DFF</b>	E(S <sub>1</sub> ) <sub>v</sub>	2.610	2.888	2.968	2.994	2.743	2.638
	E(S <sub>1</sub> )	2.447	2.726	2.722	2.743		
	E(T <sub>1</sub> )	1.355	1.394	1.572	1.433		
	E(T <sub>2</sub> ) <sub>v</sub>	2.674	2.647	3.055	2.845		
	E(T <sub>2</sub> )	2.481	2.554	2.819	2.643		
<b><i>1F</i>-DFF</b>	E(S <sub>1</sub> ) <sub>v</sub>	2.565	2.853	2.848	2.858	2.706	
	E(S <sub>1</sub> )	2.435	2.650	2.652	2.658		
	E(T <sub>1</sub> )	1.210	1.234	1.417	1.279		
	E(T <sub>2</sub> ) <sub>v</sub>	2.476	2.545	2.816	2.615		
	E(T <sub>2</sub> )	2.284	2.327	2.593	2.405		
<b><i>1T</i>-DFF</b>	E(S <sub>1</sub> ) <sub>v</sub>	2.519	2.836	2.833	2.850	2.656	
	E(S <sub>1</sub> )	2.370	2.612	2.616	2.629		
	E(T <sub>1</sub> )	1.216	1.252	1.443	1.301		
	E(T <sub>2</sub> ) <sub>v</sub>	2.392	2.476	2.782	2.558		
	E(T <sub>2</sub> )	2.191	2.248	2.549	2.342		
<b><i>2T</i>-DFF</b>	E(S <sub>1</sub> ) <sub>v</sub>	2.027	2.466	2.432	2.529	2.187	
	E(S <sub>1</sub> )	1.880	2.209	2.195	2.266		
	E(T <sub>1</sub> )	1.000	1.092	1.262	1.162		
	E(T <sub>2</sub> ) <sub>v</sub>	1.906	2.067	2.347	2.184		
	E(T <sub>2</sub> )	1.706	1.731	2.086	1.838		

<sup>1</sup>Calculated  $E(S_1)_v$  based on the optimized S<sub>0</sub> geometries at the theoretical level of B3LYP/6-311G\*

**Table S2** Tuned range separating parameters  $\mu$  [bohr<sup>-1</sup>] in LC-BLYP/6-31G\* for DPP derivatives with Quantum program optDFTw.

	<b><i>I</i>B-DFE</b>	<b>Br-<i>I</i>B-DFE</b>	<b><i>I</i>F-DFE</b>	<b><i>I</i>T-DFE</b>	<b>2T-DFE</b>
$\mu$	0.2160	0.2012	0.2324	0.2181	0.1754

**Table S3** Solvation effects on  $E(S_1)$  and  $E(T_1)$  of DFE derivatives (in eV).

	vacuum	toluene	CHCl <sub>3</sub>
<b><i>I</i>F-DFE</b>	2.618/1.211	2.435/1.210	2.341/1.209
<b>2F-DFE</b>	2.181/1.029	1.967/1.023	1.862 /1.018
<b>3F-DFE</b>	1.929/0.970	1.725/0.961	1.626/0.953

**Table S4** Basis set effects on  $E(S_1)$  and  $E(T_1)$  of DFE derivatives in toluene (in eV).

	6-311G*	6-311G**	cc-pVTZ
<b><i>I</i>F-DFE</b>	2.435/1.210	2.433/1.210	2.414/1.200
<b>2F-DFE</b>	1.967/1.023	1.967/1.024	1.949/1.012
<b>3F-DFE</b>	1.725/0.961	1.725/0.961	1.709/0.948

**Table S5.** Effect of electronic coupling in ***I*B-DFE** dimers on  $E(S_1)_v$  and  $E(T_1)_v$  in vacuum (oscillator strengths are listed in parentheses) (in eV).

	Monomer	Dimer1	Dimer2	Dimer3	Dimer4
S <sub>1</sub>	3.208(0.750)	3.160(1.477)	3.117(0.000)	3.171(1.635)	3.192(1.455)
S <sub>2</sub>	4.388(0.001)	3.218(0.071)	3.227(1.298)	3.237(0.000)	3.207(0.083)
S <sub>3</sub>	4.473(0.000)	4.263(0.003)	4.059(0.030)	4.382(0.001)	4.432(0.000)
S <sub>4</sub>	4.677(0.000)	4.318(0.003)	4.060(0.000)	4.382(0.000)	4.434(0.001)
T <sub>1</sub>	1.284	1.294	1.258	1.282	1.296
T <sub>2</sub>	2.895	1.300	1.263	1.282	1.300
T <sub>3</sub>	3.745	2.893	2.883	2.895	2.889
T <sub>4</sub>	3.891	2.896	2.890	2.895	2.893

**Table S6** Bond lengths (Å) and torsion Angles (deg) of along the conjugated backbones of  $\alpha$ Me<sub>2</sub>N-*I*F-*D*FF and  $\alpha$ MeO-*I*F-*D*FF.

		$\alpha$ Me <sub>2</sub> N- <i>I</i> F- <i>D</i> FF				$\alpha$ MeO- <i>I</i> F- <i>D</i> FF			
		d <sub>1</sub>	d <sub>2</sub>	$\Phi_1$	$\Phi_2$	d <sub>1</sub>	d <sub>2</sub>	$\Phi_1$	$\Phi_2$
B3LYP	S <sub>0</sub>	1.405	1.353	-0.582	-9.176	1.412	1.321	0	0
	S <sub>1</sub>	1.392	1.351	-0.193	-7.524	1.393	1.318	0	0
	T <sub>1</sub>	1.384	1.352	-0.316	-7.959	1.386	1.320	0	0
	T <sub>2</sub>	1.378	1.347	-0.126	-7.208	1.381	1.317	0	0
$\omega$ B97XD	S <sub>0</sub>	1.409	1.352	-0.535	-9.298	1.416	1.316	0	0
	S <sub>1</sub>	1.388	1.345	0.090	-7.366	1.390	1.311	0	0
	T <sub>1</sub>	1.379	1.350	-0.249	-8.019	1.383	1.315	0	0
	T <sub>2</sub>	1.383	1.344	0.131	-7.109	1.385	1.311	0	0

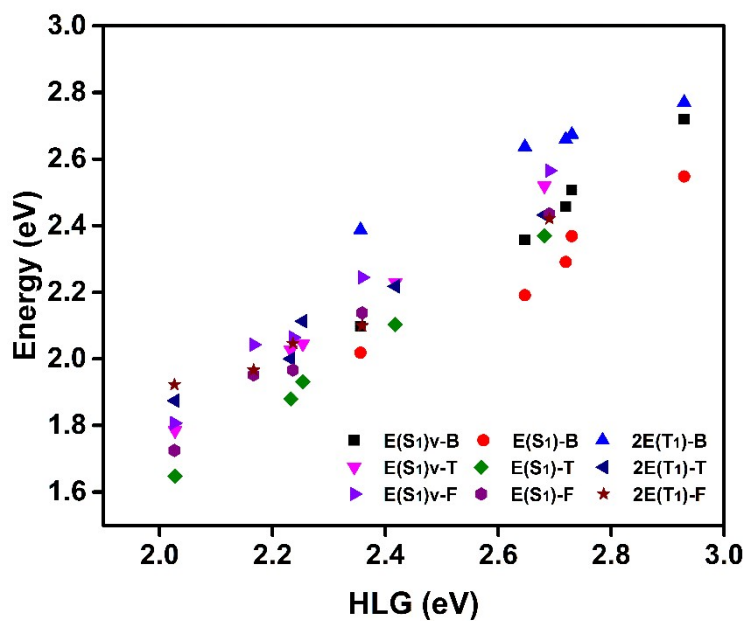
**Table S7** Comparisons of the calculated results for **F3B-DFF**, **Pa-DFF** and **3B-DFF**.<sup>1</sup>

	HOMO	LUMO	HLG	y <sub>0</sub>	y <sub>1</sub>	E(S <sub>1</sub> )	E(T <sub>1</sub> )	E(T <sub>2</sub> )
<b>F3B-DFF</b>	-5.606	-3.249	2.356	0.273	0.141	2.018	1.194	1.520
<b>Pa-DFF</b>	-5.893	-3.183	2.710	0.193	0.056	2.290	1.347	2.322
<b>3B-DFF</b>	-5.822	-3.175	2.647	0.186	0.044	2.191	1.319	2.276

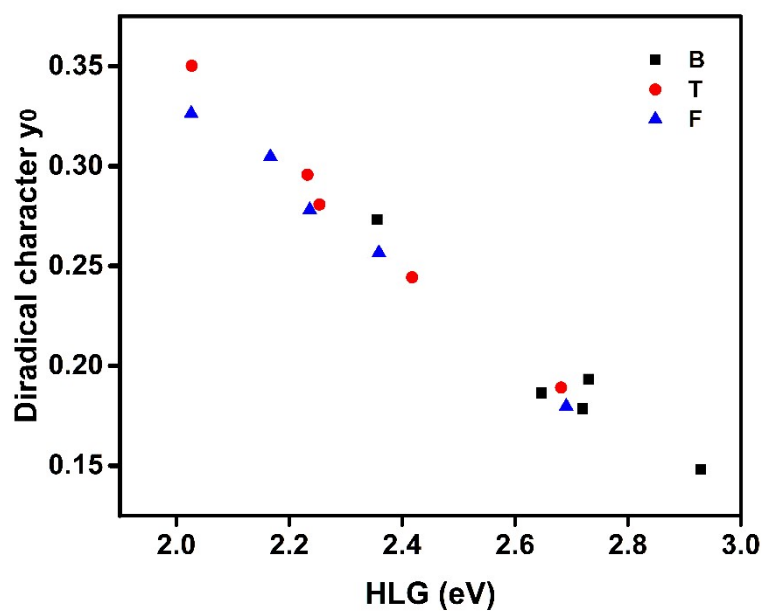
<sup>1</sup> Energies are evaluated in toluene (in eV).

**Table S8** Hammett constant ( $\sigma_p$ ) of the introduced terminal substituents.<sup>d</sup>

	-F	-Cl	-Br	-COOMe	-COOH	-CF <sub>3</sub>	-CHO
$\sigma_p$	0.06	0.23	0.23	0.45	0.45	0.54	0.42
	-CN	-NO <sub>2</sub>	-Me	-OH	-OMe	-NH <sub>2</sub>	-NMe <sub>2</sub>
$\sigma_p$	0.66	0.78	-0.17	-0.37	-0.27	-0.66	-0.83

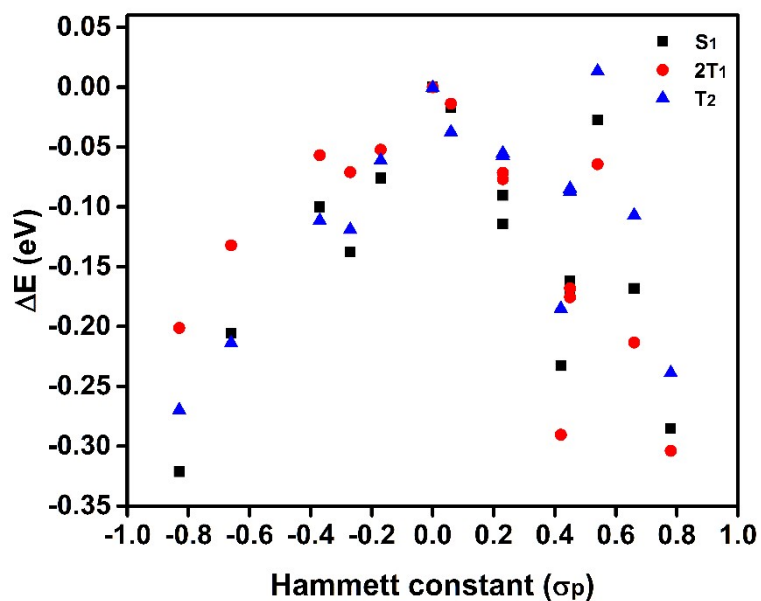


**Fig. S2**  $E(S_1)v$ ,  $E(S_1)$  and  $E(T_1)$  as a function of HLG in DFF derivatives with different aromatic substituents in toluene.

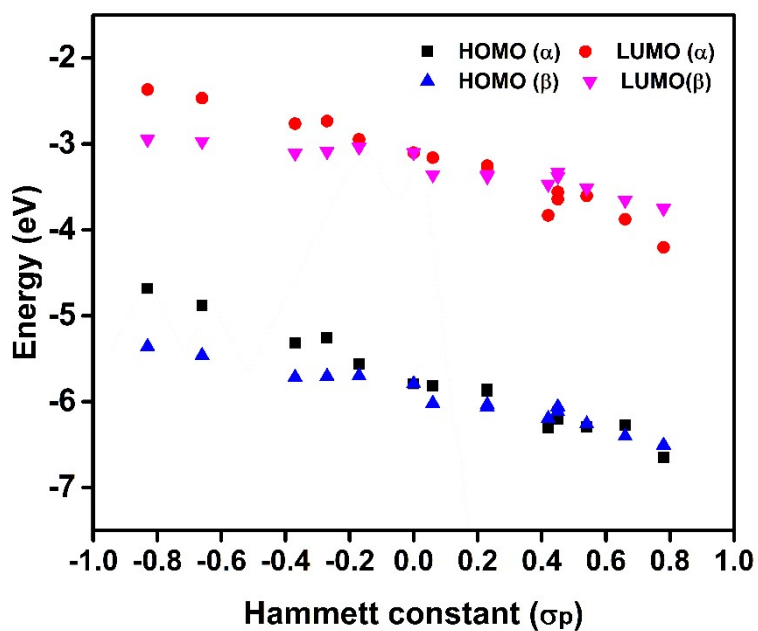


**Fig. S3** Diradical character  $y_0$  as a function of HLG in DFF derivatives with different aromatic substituents.

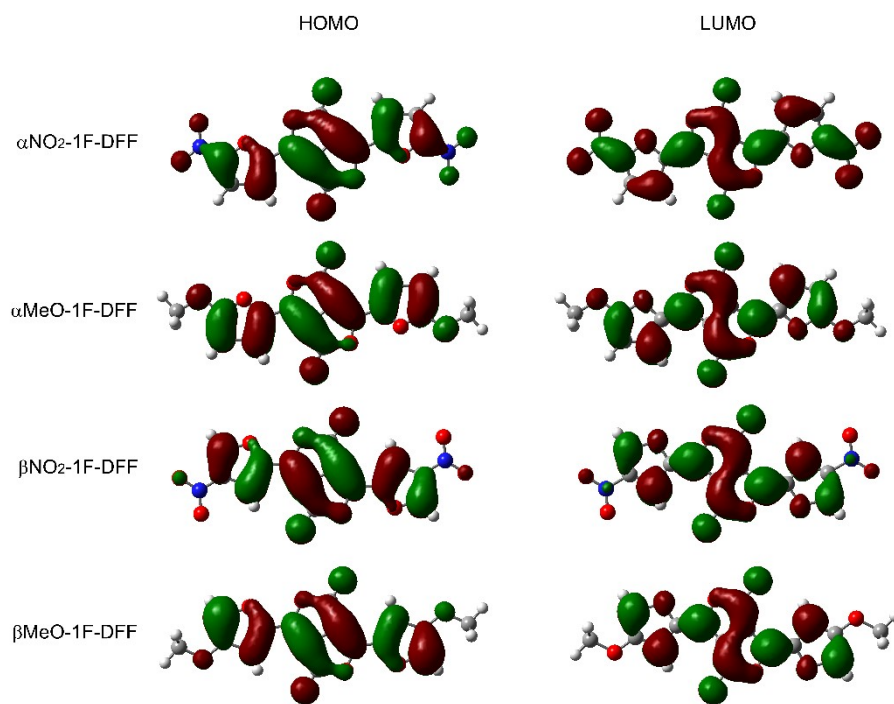




**Fig. S4** Energy differences between 1F-DFF derivatives and the pristine 1F-DFF in  $E(S_1)$ ,  $2E(T_1)$  and  $E(T_2)$  with respect to the Hammett constant ( $\sigma_p$ ) of the attached terminal substituents in toluene.



**Fig. S5** HOMO and LUMO energies for  $\alpha$ - and  $\beta$ -substitutions in 1F-DFF derivatives with respect to the Hammett constant ( $\sigma_p$ ) of the attached terminal substituents in toluene.



**Fig. S6** Electron density distributions of HOMO and LUMO in  $\alpha$ NO<sub>2</sub>-1F-DFF,  $\alpha$ MeO-1F-DFF,  $\beta$ NO<sub>2</sub>-1F-DFF and  $\beta$ MeO-1F-DFF.

**Table S9** HOMO、LUMO energy levels and HOMO-LUMO gap (HLG) values in toluene (in eV), diradical character  $y_0$  and tetraradical character  $y_1$ .

Compound	HOMO	LUMO	HLG	$y_0$	$y_1$	$y_1/y_0$
1B-DFF	-6.087	-3.157	2.929	0.148	0.015	0.101
2B-DFF	-5.886	-3.167	2.720	0.179	0.031	0.172
F2B-DFF	-5.920	-3.189	2.730	0.193	0.052	0.271
3B-DFF	-5.822	-3.175	2.647	0.186	0.044	0.237
F3B-DFF	-5.605	-3.249	2.356	0.273	0.141	0.515
1T-DFF	-5.848	-3.166	2.682	0.189	0.020	0.104
2T-DFF	-5.466	-3.234	2.232	0.296	0.063	0.214
F2T-DFF	-5.634	-3.216	2.418	0.244	0.035	0.143
3T-DFF	-5.293	-3.266	2.027	0.350	0.114	0.326
F3T-DFF	-5.492	-3.238	2.254	0.281	0.052	0.186
1F-DFF	-5.792	-3.101	2.691	0.180	0.015	0.086
2F-DFF	-5.342	-3.106	2.236	0.278	0.050	0.179
F2F-DFF	-5.501	-3.142	2.359	0.257	0.032	0.125
3F-DFF	-5.141	-3.115	2.026	0.326	0.093	0.285
F3F-DFF	-5.307	-3.141	2.167	0.305	0.050	0.163
1F-DPP	-5.188	-2.679	2.509	0.208	0.014	0.068
1F-DTT	-5.819	-3.179	2.640	0.206	0.018	0.088
1F-DSS	-5.798	-3.083	2.715	0.194	0.019	0.097

Substituents		HOMO	LUMO	HLG	y <sub>0</sub>	y <sub>1</sub>	y <sub>1</sub> /y <sub>0</sub>
-NMe <sub>2</sub>	α	-4.683	-2.366	2.317	0.137	0.009	0.062
	β	-5.385	-2.946	2.412	0.196	0.018	0.091
-NH <sub>2</sub>	α	-4.882	-2.467	2.415	0.146	0.010	0.066
	β	-5.462	-2.970	2.492	0.201	0.019	0.093
-OMe	α	-5.258	-2.733	2.525	0.151	0.010	0.068
	β	-5.704	-3.087	2.617	0.187	0.016	0.086
-OH	α	-5.314	-2.764	2.550	0.153	0.010	0.069
	β	-5.715	-3.108	2.607	0.193	0.017	0.089
-Me	α	-5.558	-2.946	2.612	0.180	0.015	0.084
	β	-5.700	-3.039	2.661	0.182	0.016	0.085
-F	α	-5.815	-3.158	2.656	0.167	0.013	0.076
	β	-6.020	-3.359	2.660	0.193	0.017	0.090
-Cl	α	-5.876	-3.262	2.614	0.189	0.017	0.088
	β	-6.061	-3.380	2.681	0.187	0.017	0.089
-Br	α	-5.853	-3.253	2.600	0.191	0.017	0.089
	β	-6.037	-3.357	2.680	0.186	0.016	0.088
-COOMe	α	-6.118	-3.558	2.559	0.209	0.021	0.100
	β	-6.060	-3.330	2.730	0.168	0.013	0.079
-COOH	α	-6.205	-3.643	2.562	0.211	0.021	0.101
	β	-6.117	-3.380	2.737	0.167	0.013	0.079
-CF <sub>3</sub>	α	-6.293	-3.604	2.688	0.194	0.018	0.091
	β	-6.259	-3.514	2.746	0.172	0.014	0.079
-CHO	α	-6.309	-3.832	2.476	0.243	0.033	0.134
	β	-6.195	-3.468	2.728	0.180	0.017	0.095
-CN	α	-6.276	-3.876	2.400	0.236	0.030	0.126
	β	-6.403	-3.656	2.747	0.182	0.017	0.092
-NO <sub>2</sub>	α	-6.648	-4.203	2.445	0.237	0.079	0.333
	β	-6.507	-3.752	2.755	0.175	0.058	0.333

**Table S10** Vertical excitation energies ( $E(S_1)_v$  and  $E(T_2)_v$ ), adiabatic excitation energies ( $E(S_1)$ ,  $E(T_1)$  and  $E(T_2)$ ) and the corresponding  $\Delta E_{SF(v)}$ ,  $\Delta E_{SF}$ ,  $\Delta E_{TTA(v)}$ ,  $\Delta E_{TTA}$  in toluene. (in eV).

	$E(S_1)_v$	$E(S_1)$	$E(T_2)_v$	$E(T_2)$	$E(T_1)$	$\Delta E_{SF(v)}$	$\Delta E_{SF}$	$\Delta E_{TTA(v)}$	$\Delta E_{TTA}$
<i>I</i> B-DFE	2.720	2.548	2.739	2.529	1.385	-0.050	-0.223	-0.031	-0.241
2B-DFE	2.457	2.291	2.542	2.368	1.330	-0.203	-0.369	-0.117	-0.292
F2B-DFE	2.507	2.369	2.306	2.190	1.337	-0.168	-0.305	-0.368	-0.484
3B-DFE	2.357	2.191	2.445	2.276	1.319	-0.280	-0.446	-0.192	-0.362
F3B-DFE	2.098	2.018	1.633	1.520	1.194	-0.290	-0.369	-0.755	-0.867
<i>I</i> T-DFE	2.519	2.370	2.392	2.191	1.216	0.087	-0.062	-0.040	-0.241
2T-DFE	2.027	1.880	1.906	1.706	1.000	0.026	-0.120	-0.095	-0.294
F2T-DFE	2.229	2.103	2.150	1.994	1.109	0.011	-0.115	-0.068	-0.225
3T-DFE	1.785	1.648	1.650	1.452	0.937	-0.090	-0.227	-0.225	-0.423
F3T-DFE	2.045	1.931	2.005	1.868	1.057	-0.068	-0.182	-0.108	-0.245
<i>I</i> F-DFE	2.565	2.435	2.476	2.284	1.210	0.144	0.014	0.056	-0.137
2F-DFE	2.064	1.967	2.004	1.881	1.023	0.018	-0.080	-0.042	-0.165
F2F-DFE	2.245	2.138	2.155	2.004	1.050	0.145	0.039	0.056	-0.096
3F-DFE	1.807	1.725	1.736	1.634	0.961	-0.116	-0.197	-0.187	-0.288
F3F-DFE	2.043	1.952	1.980	1.862	0.983	0.077	-0.014	0.014	-0.104
<i>I</i> F-DPP	2.362	2.236	2.636	2.487	1.053	0.256	0.129	0.529	0.380
<i>I</i> F-DTT	2.470	2.346	2.354	2.180	1.202	0.066	-0.058	-0.050	-0.224
<i>I</i> F-DSS	2.503	2.372	2.336	2.163	1.273	-0.043	-0.174	-0.210	-0.383

		$E(S_1)_v$	$E(S_1)$	$E(T_2)_v$	$E(T_2)$	$E(T_1)$	$\Delta E_{SF(v)}$	$\Delta E_{SF}$	$\Delta E_{TTA(v)}$	$\Delta E_{TTA}$
-NMe <sub>2</sub>	$\alpha$	2.184	2.114	2.146	2.014	1.110	-0.036	-0.106	-0.073	-0.206
	$\beta$	2.230	2.127	1.934	1.765	1.014	0.022	-0.081	-0.274	-0.444
-NH <sub>2</sub>	$\alpha$	2.310	2.229	2.223	2.070	1.144	0.021	-0.060	-0.066	-0.219
	$\beta$	2.374	2.291	2.138	1.979	1.121	0.132	0.049	-0.105	-0.263
-OMe	$\alpha$	2.397	2.297	2.339	2.165	1.175	0.047	-0.053	-0.011	-0.185
	$\beta$	2.490	2.379	2.386	2.241	1.180	0.130	0.019	0.026	-0.119
-OH	$\alpha$	2.438	2.335	2.351	2.172	1.182	0.073	-0.029	-0.013	-0.192
	$\beta$	2.495	2.383	2.372	2.221	1.167	0.161	0.049	0.037	-0.113
-Me	$\alpha$	2.478	2.359	2.405	2.223	1.184	0.109	-0.010	0.036	-0.146
	$\beta$	2.527	2.404	2.456	2.271	1.201	0.124	0.001	0.053	-0.132
-F	$\alpha$	2.541	2.418	2.445	2.246	1.204	0.134	0.011	0.038	-0.161
	$\beta$	2.548	2.418	2.453	2.267	1.182	0.184	0.055	0.089	-0.096
-Cl	$\alpha$	2.471	2.344	2.417	2.229	1.175	0.121	-0.005	0.067	-0.121
	$\beta$	2.540	2.408	2.475	2.287	1.196	0.149	0.017	0.083	-0.105
-Br	$\alpha$	2.445	2.321	2.411	2.227	1.172	0.101	-0.023	0.067	-0.117
	$\beta$	2.532	2.401	2.474	2.287	1.197	0.137	0.006	0.079	-0.108
-COOMe	$\alpha$	2.410	2.265	2.372	2.197	1.126	0.157	0.013	0.119	-0.056
	$\beta$	2.575	2.436	2.529	2.328	1.228	0.118	-0.021	0.072	-0.129
-COOH	$\alpha$	2.419	2.273	2.374	2.199	1.123	0.174	0.027	0.129	-0.046
	$\beta$	2.583	2.443	2.538	2.335	1.229	0.124	-0.016	0.079	-0.124
-CF <sub>3</sub>	$\alpha$	2.555	2.408	2.494	2.297	1.178	0.198	0.051	0.137	-0.059
	$\beta$	2.604	2.460	2.555	2.348	1.224	0.156	0.012	0.107	-0.099
-CHO	$\alpha$	2.346	2.202	2.258	2.099	1.065	0.215	0.072	0.127	-0.032
	$\beta$	2.572	2.432	2.508	2.314	1.225	0.123	-0.018	0.059	-0.135
-CN	$\alpha$	2.414	2.267	2.348	2.177	1.104	0.206	0.059	0.140	-0.031
	$\beta$	2.588	2.440	2.553	2.350	1.223	0.142	-0.006	0.107	-0.096
-NO <sub>2</sub>	$\alpha$	2.299	2.150	2.214	2.045	1.059	0.182	0.033	0.097	-0.072
	$\beta$	2.590	2.435	2.562	2.360	1.222	0.146	-0.009	0.118	-0.084

### Coordinates of the four dimers.

#### Dimer 1

O	0.51230000	6.25870000	8.25060000
O	1.59510000	8.12320000	8.93380000
C	0.65150000	7.66590000	8.33940000
C	-0.48810000	8.20270000	7.63320000
C	-0.63420000	5.93140000	7.54440000
C	-0.94050000	4.53500000	7.39260000
C	-2.11810000	4.12810000	6.80340000
C	-2.42590000	2.81280000	6.65980000
C	-1.51650000	1.87360000	7.11210000
C	-0.34470000	2.23610000	7.67690000
C	-0.04500000	3.57290000	7.82570000
H	-2.71930000	4.71090000	6.53620000
H	-3.24290000	2.64600000	6.32150000
H	-1.70490000	1.28480000	6.95070000
H	0.22870000	1.72830000	7.95000000
H	0.74050000	3.79320000	8.21650000
C	-1.24840000	7.09230000	7.17130000
C	-1.10230000	9.36360000	7.26010000
C	-2.38800000	7.62910000	6.46510000
O	-2.24880000	9.03630000	6.55400000
C	-0.79600000	10.76000000	7.41190000
O	-3.33160000	7.17180000	5.87070000
C	0.38150000	11.16690000	8.00110000
C	-1.69150000	11.72210000	6.97880000
C	0.68940000	12.48220000	8.14470000
H	0.98280000	10.58410000	8.26830000
C	-1.39180000	13.05890000	7.12760000
H	-2.47700000	11.50180000	6.58800000
C	-0.22000000	13.42140000	7.69240000
H	1.50640000	12.64900000	8.48300000
H	-1.96520000	13.56670000	6.85450000
H	-0.03160000	14.01020000	7.85380000
O	-1.81470000	-1.38880000	2.85280000
O	-2.89750000	0.47570000	2.16960000
C	-1.95390000	0.01840000	2.76400000
C	-0.81430000	0.55520000	3.47020000
C	-0.66820000	-1.71610000	3.55900000
C	-0.36190000	-3.11250000	3.71070000
C	0.81570000	-3.51940000	4.30000000
C	1.12350000	-4.83470000	4.44360000

C	0.21410000	-5.77390000	3.99130000
C	-0.95770000	-5.41140000	3.42650000
C	-1.25740000	-4.07460000	3.27770000
H	1.41690000	-2.93660000	4.56720000
H	1.94050000	-5.00150000	4.78190000
H	0.40260000	-6.36270000	4.15270000
H	-1.53110000	-5.91920000	3.15340000
H	-2.04280000	-3.85430000	2.88690000
C	-0.05390000	-0.55520000	3.93210000
C	-0.20010000	1.71610000	3.84320000
C	1.08560000	-0.01840000	4.63830000
O	0.94640000	1.38880000	4.54940000
C	-0.50640000	3.11250000	3.69150000
O	2.02920000	-0.47570000	5.23270000
C	-1.68390000	3.51940000	3.10230000
C	0.38910000	4.07460000	4.12450000
C	-1.99180000	4.83470000	2.95870000
H	-2.28520000	2.93660000	2.83510000
C	0.08940000	5.41140000	3.97570000
H	1.17460000	3.85430000	4.51540000
C	-1.08240000	5.77390000	3.41100000
H	-2.80880000	5.00150000	2.62040000
H	0.66280000	5.91920000	4.24890000
H	-1.27080000	6.36270000	3.24960000

#### Dimer 2

O	0.51230000	6.25870000	8.25060000
O	1.59510000	8.12320000	8.93380000
C	0.65150000	7.66590000	8.33940000
C	-0.48810000	8.20270000	7.63320000
C	-0.63420000	5.93140000	7.54440000
C	-0.94050000	4.53500000	7.39260000
C	-2.11810000	4.12810000	6.80340000
C	-2.42590000	2.81280000	6.65980000
C	-1.51650000	1.87360000	7.11210000
C	-0.34470000	2.23610000	7.67690000
C	-0.04500000	3.57290000	7.82570000
H	-2.71930000	4.71090000	6.53620000
H	-3.24290000	2.64600000	6.32150000
H	-1.70490000	1.28480000	6.95070000
H	0.22870000	1.72830000	7.95000000
H	0.74050000	3.79320000	8.21650000



C	-1.24840000	7.09230000	7.17130000
C	-1.10230000	9.36360000	7.26010000
C	-2.38800000	7.62910000	6.46510000
O	-2.24880000	9.03630000	6.55400000
C	-0.79600000	10.76000000	7.41190000
O	-3.33160000	7.17180000	5.87070000
C	0.38150000	11.16690000	8.00110000
C	-1.69150000	11.72210000	6.97880000
C	0.68940000	12.48220000	8.14470000
H	0.98280000	10.58410000	8.26830000
C	-1.39180000	13.05890000	7.12760000
H	-2.47700000	11.50180000	6.58800000
C	-0.22000000	13.42140000	7.69240000
H	1.50640000	12.64900000	8.48300000
H	-1.96520000	13.56670000	6.85450000
H	-0.03160000	14.01020000	7.85380000
O	6.28930000	6.25870000	8.25060000
O	7.37210000	8.12320000	8.93380000
C	6.42850000	7.66590000	8.33940000
C	5.28890000	8.20270000	7.63320000
C	5.14280000	5.93140000	7.54440000
C	4.83650000	4.53500000	7.39260000
C	3.65890000	4.12810000	6.80340000
C	3.35110000	2.81280000	6.65980000
C	4.26050000	1.87360000	7.11210000
C	5.43230000	2.23610000	7.67690000
C	5.73200000	3.57290000	7.82570000
H	3.05770000	4.71090000	6.53620000
H	2.53410000	2.64600000	6.32150000
H	4.07210000	1.28480000	6.95070000
H	6.00570000	1.72830000	7.95000000
H	6.51750000	3.79320000	8.21650000
C	4.52860000	7.09230000	7.17130000
C	4.67470000	9.36360000	7.26010000
C	3.38900000	7.62910000	6.46510000
O	3.52820000	9.03630000	6.55400000
C	4.98100000	10.76000000	7.41190000
O	2.44540000	7.17180000	5.87070000
C	6.15850000	11.16690000	8.00110000
C	4.08550000	11.72210000	6.97880000
C	6.46640000	12.48220000	8.14470000
H	6.75980000	10.58410000	8.26830000
C	4.38520000	13.05890000	7.12760000
H	3.30000000	11.50180000	6.58800000

C	5.55700000	13.42140000	7.69240000
H	7.28340000	12.64900000	8.48300000
H	3.81180000	13.56670000	6.85450000
H	5.74540000	14.01020000	7.85380000

### Dimer3

O	-1.81470000	-1.38880000	2.85280000
O	-2.89750000	0.47570000	2.16960000
C	-1.95390000	0.01840000	2.76400000
C	-0.81430000	0.55520000	3.47020000
C	-0.66820000	-1.71610000	3.55900000
C	-0.36190000	-3.11250000	3.71070000
C	0.81570000	-3.51940000	4.30000000
C	1.12350000	-4.83470000	4.44360000
C	0.21410000	-5.77390000	3.99130000
C	-0.95770000	-5.41140000	3.42650000
C	-1.25740000	-4.07460000	3.27770000
H	1.41690000	-2.93660000	4.56720000
H	1.94050000	-5.00150000	4.78190000
H	0.40260000	-6.36270000	4.15270000
H	-1.53110000	-5.91920000	3.15340000
H	-2.04280000	-3.85430000	2.88690000
C	-0.05390000	-0.55520000	3.93210000
C	-0.20010000	1.71610000	3.84320000
C	1.08560000	-0.01840000	4.63830000
O	0.94640000	1.38880000	4.54940000
C	-0.50640000	3.11250000	3.69150000
O	2.02920000	-0.47570000	5.23270000
C	-1.68390000	3.51940000	3.10230000
C	0.38910000	4.07460000	4.12450000
C	-1.99180000	4.83470000	2.95870000
H	-2.28520000	2.93660000	2.83510000
C	0.08940000	5.41140000	3.97570000
H	1.17460000	3.85430000	4.51540000
C	-1.08240000	5.77390000	3.41100000
H	-2.80880000	5.00150000	2.62040000
H	0.66280000	5.91920000	4.24890000
H	-1.27080000	6.36270000	3.24960000
O	-1.81470000	13.90620000	2.85280000
O	-2.89750000	15.77070000	2.16960000
C	-1.95390000	15.31340000	2.76400000

C	-0.81430000	15.85020000	3.47020000
C	-0.66820000	13.57890000	3.55900000
C	-0.36190000	12.18250000	3.71070000
C	0.81570000	11.77560000	4.30000000
C	1.12350000	10.46030000	4.44360000
C	0.21410000	9.52110000	3.99130000
C	-0.95770000	9.88360000	3.42650000
C	-1.25740000	11.22040000	3.27770000
H	1.41690000	12.35840000	4.56720000
H	1.94050000	10.29350000	4.78190000
H	0.40260000	8.93230000	4.15270000
H	-1.53110000	9.37580000	3.15340000
H	-2.04280000	11.44070000	2.88690000
C	-0.05390000	14.73980000	3.93210000
C	-0.20010000	17.01110000	3.84320000
C	1.08560000	15.27660000	4.63830000
O	0.94640000	16.68380000	4.54940000
C	-0.50640000	18.40750000	3.69150000
O	2.02920000	14.81930000	5.23270000
C	-1.68390000	18.81440000	3.10230000
C	0.38910000	19.36960000	4.12450000
C	-1.99180000	20.12970000	2.95870000
H	-2.28520000	18.23160000	2.83510000
C	0.08940000	20.70640000	3.97570000
H	1.17460000	19.14930000	4.51540000
C	-1.08240000	21.06890000	3.41100000
H	-2.80880000	20.29650000	2.62040000
H	0.66280000	21.21420000	4.24890000
H	-1.27080000	21.65770000	3.24960000

#### Dimer4

O	6.28930000	6.25870000	8.25060000
O	7.37210000	8.12320000	8.93380000
C	6.42850000	7.66590000	8.33940000
C	5.28890000	8.20270000	7.63320000
C	5.14280000	5.93140000	7.54440000
C	4.83650000	4.53500000	7.39260000
C	3.65890000	4.12810000	6.80340000
C	3.35110000	2.81280000	6.65980000
C	4.26050000	1.87360000	7.11210000
C	5.43230000	2.23610000	7.67690000
C	5.73200000	3.57290000	7.82570000

H	3.05770000	4.71090000	6.53620000
H	2.53410000	2.64600000	6.32150000
H	4.07210000	1.28480000	6.95070000
H	6.00570000	1.72830000	7.95000000
H	6.51750000	3.79320000	8.21650000
C	4.52860000	7.09230000	7.17130000
C	4.67470000	9.36360000	7.26010000
C	3.38900000	7.62910000	6.46510000
O	3.52820000	9.03630000	6.55400000
C	4.98100000	10.76000000	7.41190000
O	2.44540000	7.17180000	5.87070000
C	6.15850000	11.16690000	8.00110000
C	4.08550000	11.72210000	6.97880000
C	6.46640000	12.48220000	8.14470000
H	6.75980000	10.58410000	8.26830000
C	4.38520000	13.05890000	7.12760000
H	3.30000000	11.50180000	6.58800000
C	5.55700000	13.42140000	7.69240000
H	7.28340000	12.64900000	8.48300000
H	3.81180000	13.56670000	6.85450000
H	5.74540000	14.01020000	7.85380000
O	-1.81470000	-1.38880000	2.85280000
O	-2.89750000	0.47570000	2.16960000
C	-1.95390000	0.01840000	2.76400000
C	-0.81430000	0.55520000	3.47020000
C	-0.66820000	-1.71610000	3.55900000
C	-0.36190000	-3.11250000	3.71070000
C	0.81570000	-3.51940000	4.30000000
C	1.12350000	-4.83470000	4.44360000
C	0.21410000	-5.77390000	3.99130000
C	-0.95770000	-5.41140000	3.42650000
C	-1.25740000	-4.07460000	3.27770000
H	1.41690000	-2.93660000	4.56720000
H	1.94050000	-5.00150000	4.78190000
H	0.40260000	-6.36270000	4.15270000
H	-1.53110000	-5.91920000	3.15340000
H	-2.04280000	-3.85430000	2.88690000
C	-0.05390000	-0.55520000	3.93210000
C	-0.20010000	1.71610000	3.84320000
C	1.08560000	-0.01840000	4.63830000
O	0.94640000	1.38880000	4.54940000
C	-0.50640000	3.11250000	3.69150000
O	2.02920000	-0.47570000	5.23270000
C	-1.68390000	3.51940000	3.10230000

C	0.38910000	4.07460000	4.12450000
C	-1.99180000	4.83470000	2.95870000
H	-2.28520000	2.93660000	2.83510000
C	0.08940000	5.41140000	3.97570000
H	1.17460000	3.85430000	4.51540000
C	-1.08240000	5.77390000	3.41100000
H	-2.80880000	5.00150000	2.62040000
H	0.66280000	5.91920000	4.24890000
H	-1.27080000	6.36270000	3.24960000

## References

- (a) T. Lu, optDFTw program v1.0, webpage: <http://sobereva.com/346>.
- (b) K. Zhang, B. Tieke, *Macromolecules*, 2008, **41**, 7287-7295.
- (c) C. M. Mauck, Y. J. Bae, M. Chen, N. Powers-Riggs, Y.-L. Wu, M. R. Wasielewski, *ChemPhotoChem*, 2018, **2**, 223-233.
- (d) C. Hansch, A. Leo, R. W. Taft, *Chem. Rev.*, 1991, **91**, 165-195.