

Supporting information

Decisive Role of Heavy-Atom Orientation for Efficient Enhancement of Spin-Orbit Coupling in Organic Thermally Activated Delayed Fluorescence Emitters

Michał Mońska,¹ Daria Grzywacz,² Ester Hoffman,¹ Vladyslav Ievtukhov,² Karol Kozakiewicz,²
Radosław Rogowski,¹ Aleksander Kubicki,¹ Beata Liberek,² Piotr Bojarski,¹ Illia E. Serdiuk^{1*}

¹ Faculty of Mathematics, Physics and Informatics, University of Gdańsk, Wita Stwosza 57, 80-308 Gdańsk, Poland

² Faculty of Chemistry, University of Gdańsk, Wita Stwosza 63, 80-308 Gdańsk, Poland

*Corresponding author. E-mail: illia.serdiuk@ug.edu.pl, phone + 48 58 523 22 44

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Section S1: Photophysical measurements

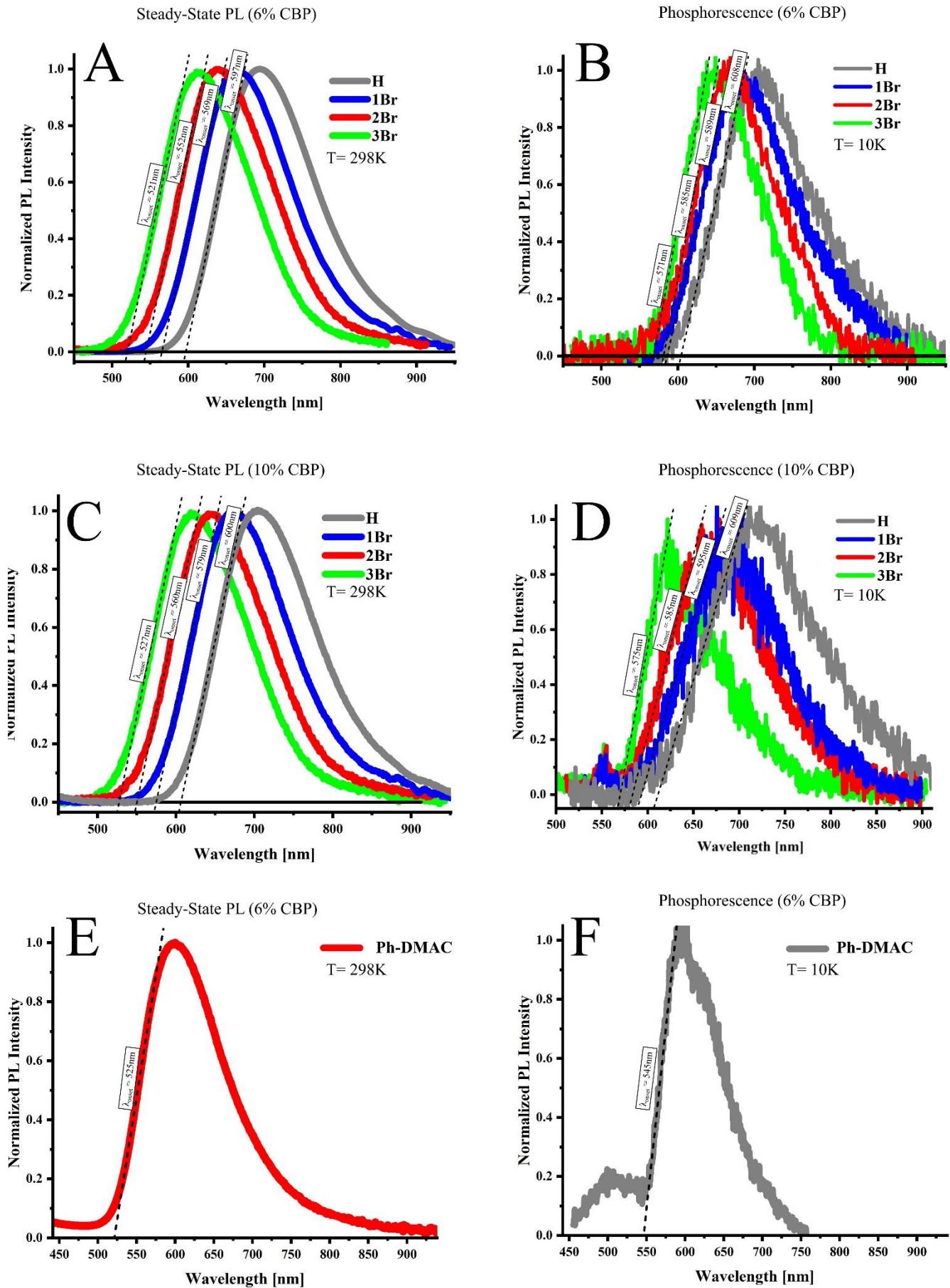


Figure S1. Steady-State PL spectra of investigated compounds: 6% (A, E) and 10% (C) CBP with onsets and phosphorescence spectra measured in 10K, 6% (B, F) and 10% (D) CBP.

Table S1. Experimental determination of ^1CT , ^3CT and ^3LE - onset values.

$w_{X/\text{CBP}}$	Fluorescence		Phosphorescence			Phosphorescence			
	λ_{onset}	^1CT energy	λ_{onset}	^3CT energy	$\Delta E_{^1\text{CT}-^3\text{CT}}$	λ_{onset}	^3LE energy	$\Delta E_{^1\text{CT}-^3\text{LE}}$	
	[nm]	[eV]	[nm]	[eV]	[eV]	[nm]	[eV]	[eV]	
H	6%	597	2.07	608	2.04	0.04	552	2.25	-0.18
	10%	600	2.07	609	2.04	0.03			
1Br	6%	569	2.18	589	2.10	0.08	552	2.25	-0.07
	10%	579	2.14	595	2.08	0.06			
2Br	6%	552	2.25	585	2.12	0.13	552	2.25	0.00
	10%	560	2.21	585	2.11	0.10			
3Br	6%	521	2.37	571	2.17	0.20	552	2.25	0.13
	10%	527	2.35	575	2.16	0.19			

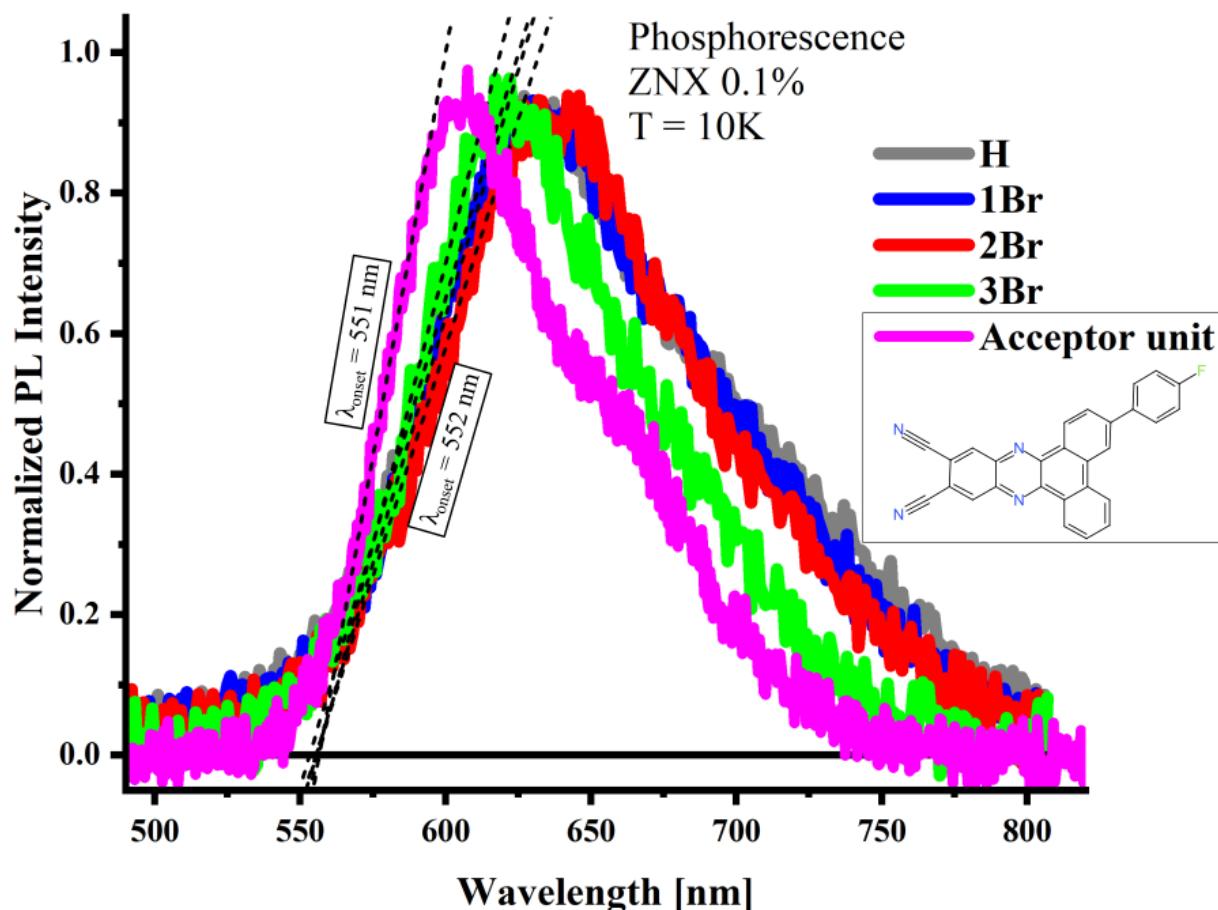


Figure S2. Phosphorescence spectra of investigated compounds dispersed in ZNX, measured in 10K under excitation wavelength $\lambda_{\text{exc}} = 370 \text{ nm}$ with a 20ms delay after excitation pulse.

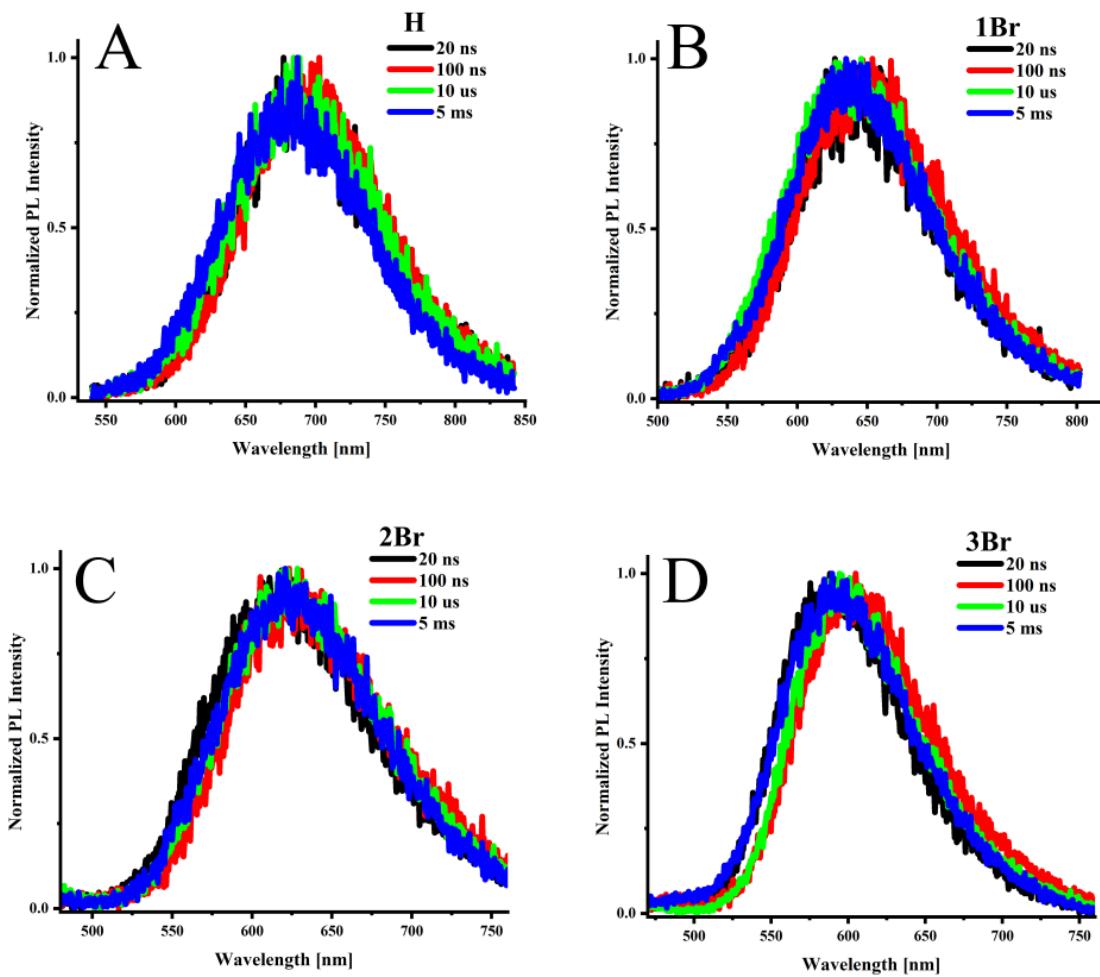


Figure S3. PL spectra of **H** (A), **1Br** (B), **2Br** (C), and **3Br** (D) in 6% CBP, taken at different time delays.

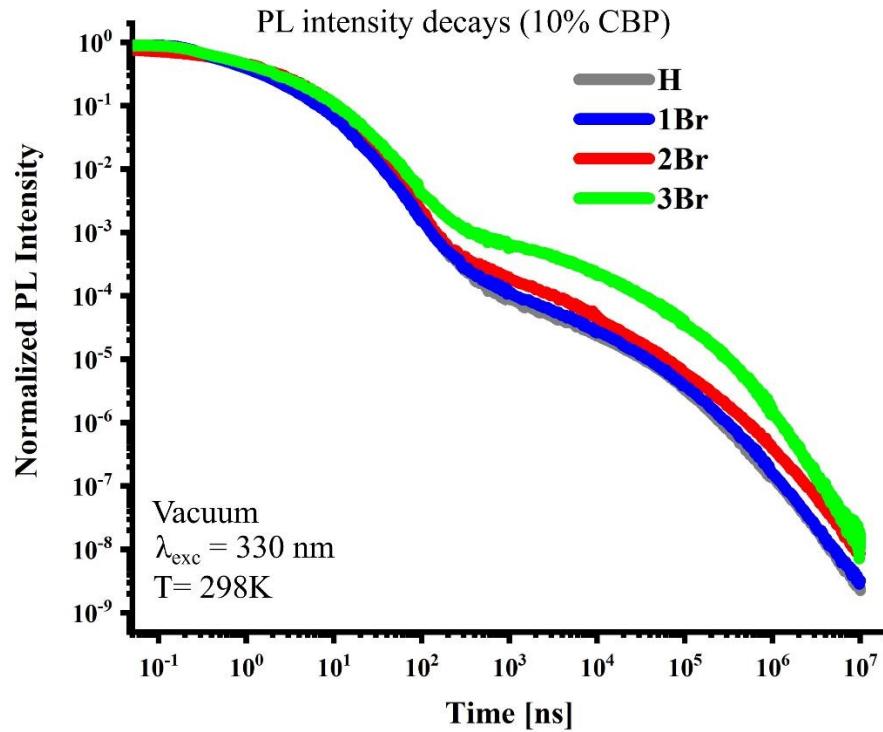


Figure S4. PL intensity decays of investigated compounds (10% CBP) measured in vacuum under excitation wavelength $\lambda_{\text{exc}} = 330 \text{ nm}$.

Section S2: Quantum chemical calculations

Theoretical rate constants of rISC were calculated using Marcus-Hush equation:

$$k_{(r)ISC} = \frac{V^2}{\hbar} \sqrt{\frac{\pi}{k_B T \lambda}} \exp \left[-\frac{(\Delta E_{ST} + \lambda)^2}{4k_B T \lambda} \right], \quad (\text{S1})$$

where V is SOC constant, \hbar is reduced Planck constant, λ is sum of internal (λ_{int}) and external (λ_{solv}) reorganization energies for respective transition (in our calculations we assumed $\lambda_{\text{solv}, i} = 0.3$ eV) ΔE_{ST} is the energy gap between singlet and respective triplet state, k_B stands for Boltzmann constant, T is temperature.

Relative contribution of i -th rotamers in **H**, **1Br**, **2Br** and **3Br**-rotamers was calculated using Boltzmann distribution law:

$$\mu_i [\%] = \frac{\exp \left(-\frac{\Delta E_i}{k_B T} \right)}{\sum_{i=1}^N \exp \left(-\frac{\Delta E_i}{k_B T} \right)}, \quad (\text{S2})$$

where N is the number of existing isomers (for **H**, **1Br** **2Br** and **3Br**-rotamers $N = 2, 16, 16$ and 32 , respectively), ΔE_i denotes the energy difference between i -th rotamer and most stable rotamer (with lowest energy). Procedure for theoretical prediction rate constants described in details in [1].

H - rotamers

1Br - rotamers

2Br - rotamers

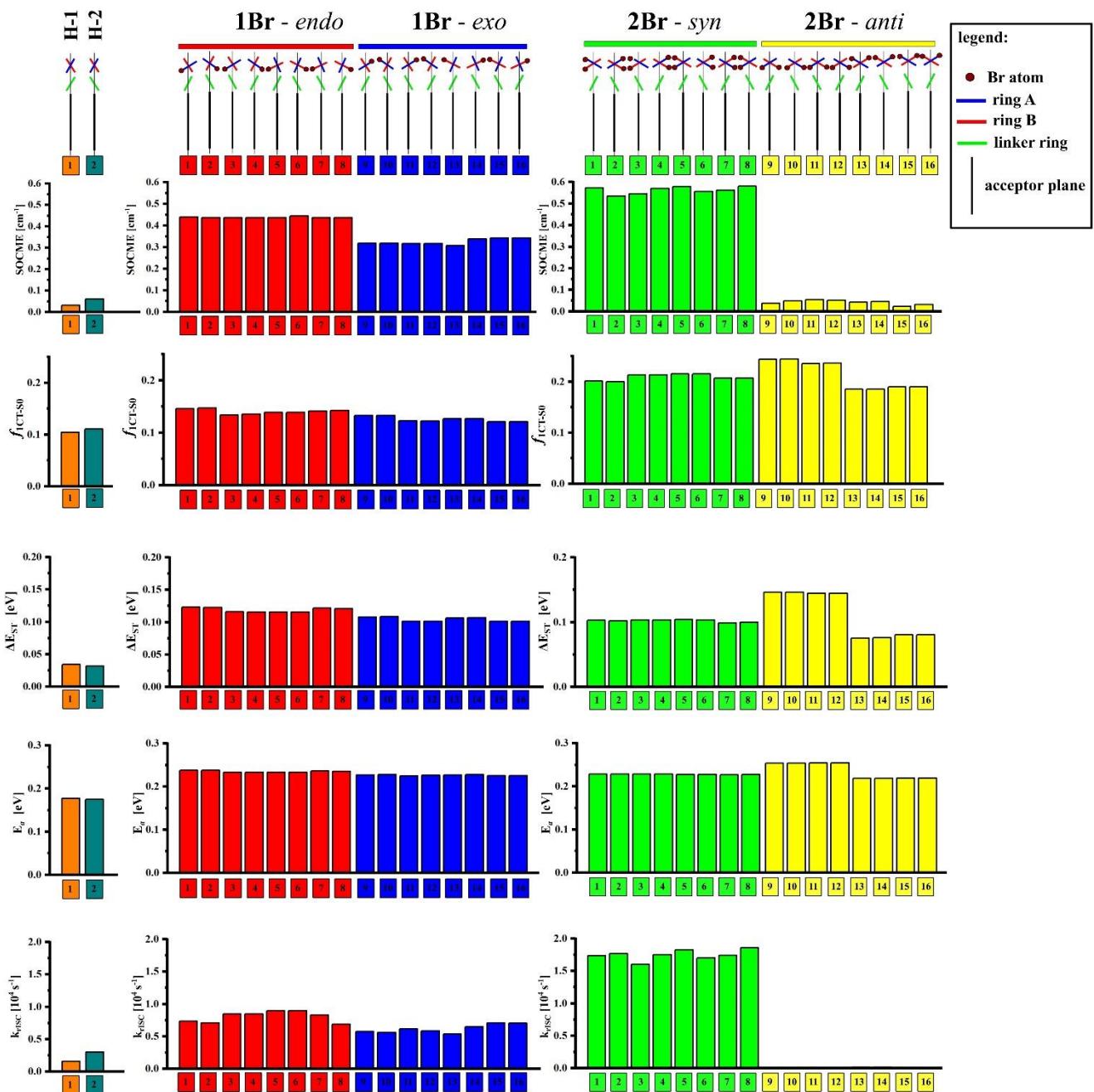


Figure S5. Calculated electronic parameters of all **H**, **1Br** and **2Br** – rotamers. In main text, key rotamers (depicted in **Figure 5**, **Table 3**) from each group: **H-rotamers**: **1** (as **H-1**) and **2** (as **H-2**); **1Br-rotamers**: **3** (as **1Br-endo**) and **12** (as **1Br-exo**); **2Br-rotamers**: **5** (as **2Br-syn**) and **9** (as **2Br-anti**).

3Br - rotamers

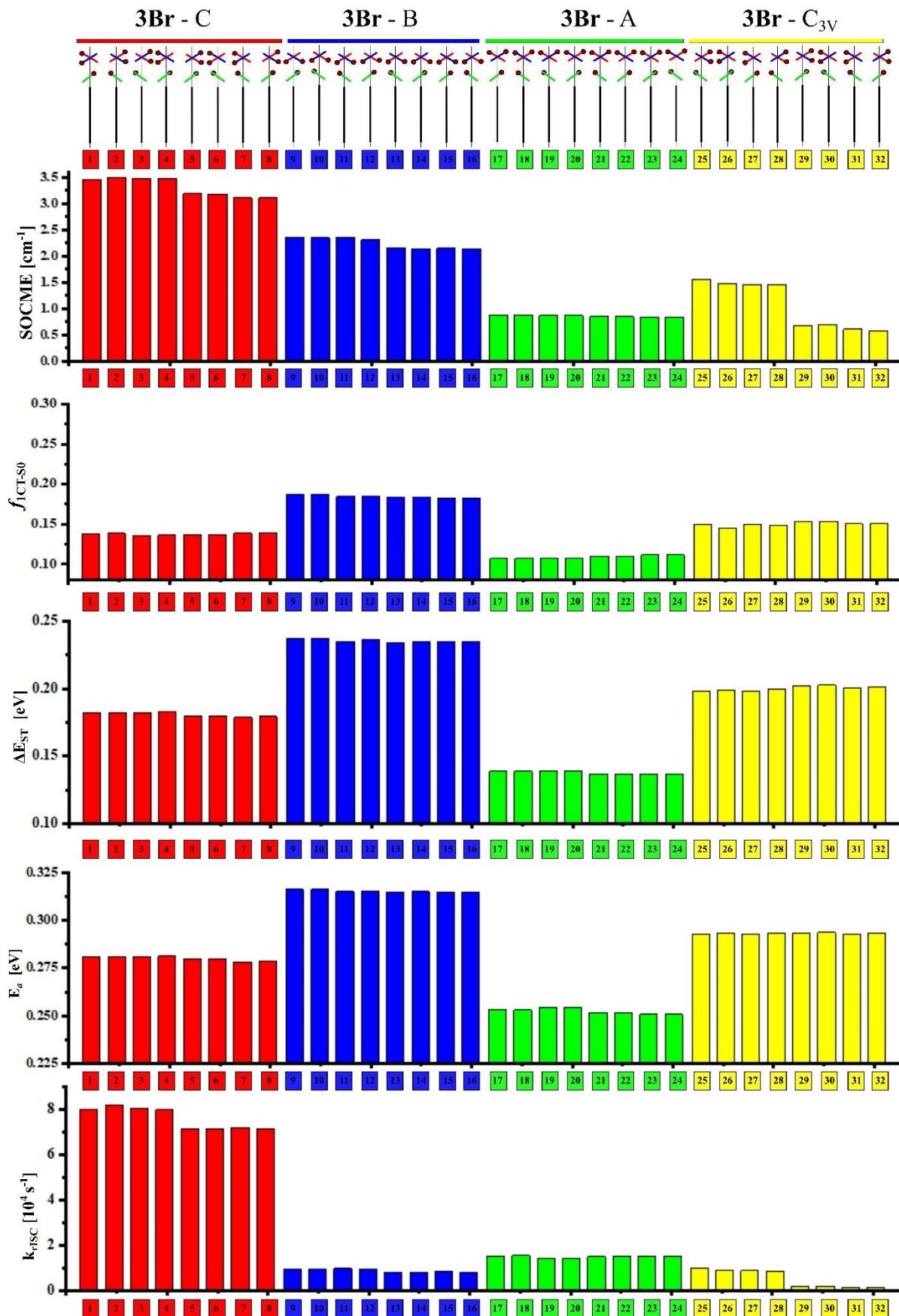


Figure S6. Calculated electronic parameters of all 3Br rotamers. In main text, key rotamers (depicted in **Figure 5, Table 3**) from 3Br-rotamers group: **32** (as 3Br- C_{3V}), **17** (as 3Br-A), **21** (as 3Br-A), **12** (as 3Br-B), **1** (as 3Br-C).

Table S2. Calculated values of different geometry (dihedrals θ_L , θ_A , and θ_B) and electronic parameters (plotted in **Figures S5** and **S6**) of all rotamers.

H - rotamers	1	2	X ^a
$\text{SOC}_{\text{T1-S1}} [\text{cm}^{-1}]$	0.06	0.08	--
$\Delta E_{\text{3CT-1CT}} [\text{eV}]$	0.03	0.03	0.03
$f_{\text{S1-S0}}$	0.10	0.11	0.107
$E_a [\text{eV}]$	0.18	0.17	--
$\mu [\%]$	67.2	32.8	--
$k_{\text{ISC}} [10^4 \text{s}^{-1}]$	0.16	0.30	0.20
θ_L	27.9	-28.8	
θ_A	-64.6	-65.1	
θ_B	-64.8	-65.2	

1Br- rotamers	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	X ^a
$\text{SOC}_{\text{T1-S1}} [\text{cm}^{-1}]$	0.44	0.43	0.44	0.44	0.44	0.44	0.45	0.41	0.32	0.32	0.32	0.32	0.31	0.34	0.34	0.34	--
$\Delta E_{\text{3CT-1CT}} [\text{eV}]$	0.12	0.12	0.12	0.12	0.12	0.12	0.12	0.11	0.11	0.10	0.10	0.11	0.11	0.10	0.10	0.11	
$f_{\text{S1-S0}}$	0.15	0.15	0.13	0.14	0.14	0.14	0.14	0.14	0.13	0.13	0.12	0.12	0.13	0.13	0.12	0.12	0.131
$E_a [\text{eV}]$	0.24	0.24	0.23	0.23	0.23	0.24	0.24	0.23	0.23	0.22	0.23	0.23	0.23	0.23	0.23	--	
$\mu [\%]$	7.0	4.3	1.5	2.0	7.3	10.6	3.9	21.8	7.6	2.1	9.0	3.3	6.6	5.0	3.0	5.1	--
$k_{\text{ISC}} [10^4 \text{s}^{-1}]$	0.7	0.7	0.8	0.8	0.9	0.9	0.8	0.7	0.6	0.6	0.6	0.5	0.6	0.7	0.7	0.71	
θ_L	21.3	-21.1	-22.6	22.5	22.3	-22.3	-22.6	22.7	23.2	-23.1	-23.0	22.4	23.6	-23.6	-21.9	21.9	
θ_A	-74.6	74.7	-74.9	74.9	56.0	-56.0	55.9	-55.9	69.2	-69.1	68.4	-68.8	-55.3	55.2	-55.3	55.2	
θ_B	-56.1	56.5	-56.1	56.2	75.1	-75.1	74.9	-74.9	-55.4	55.3	55.1	69.5	-69.2	69.0	-68.8		

2Br- rotamers	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	X ^a
$\text{SOC}_{\text{T1-S1}} [\text{cm}^{-1}]$	0.57	0.53	0.54	0.57	0.58	0.56	0.56	0.58	0.04	0.05	0.05	0.05	0.04	0.05	0.02	0.03	--
$\Delta E_{\text{3CT-1CT}} [\text{eV}]$	0.10	0.10	0.10	0.10	0.10	0.10	0.10	0.10	0.15	0.15	0.14	0.14	0.08	0.08	0.08	0.08	0.13
$f_{\text{S1-S0}}$	0.20	0.20	0.21	0.21	0.22	0.22	0.21	0.21	0.24	0.24	0.24	0.24	0.19	0.19	0.19	0.19	0.216
$E_a [\text{eV}]$	0.23	0.23	0.23	0.23	0.23	0.23	0.23	0.23	0.25	0.25	0.25	0.25	0.22	0.22	0.22	--	
$\mu [\%]$	0.8	3.3	1.1	1.1	7.2	1.5	1.6	5.0	12.1	8.3	27.5	27.8	1.0	0.8	0.4	0.4	--
$k_{\text{ISC}} [10^4 \text{s}^{-1}]$	1.7	1.8	1.6	1.8	1.8	1.7	1.7	1.9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.40	
θ_L	16.2	-16.1	-18.0	17.8	17.8	-17.9	-16.3	16.8	17.6	-17.7	-15.1	15.1	17.2	-17.1	-18.2	18.2	
θ_A	-77.5	77.5	-77.3	77.2	-63.5	63.5	-63.3	63.4	-68.5	68.5	-68.3	68.3	-78.4	78.2	-78.9	-68.7	
θ_B	63.3	-63.2	63.1	-63.2	77.3	-77.4	77.1	-77.1	-68.4	68.3	-68.3	68.3	-78.1	78.2	-78.4	78.4	

3Br- rotamers	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	X ^a
$\text{SOC}_{\text{T1-S1}} [\text{cm}^{-1}]$	3.45	3.50	3.47	3.48	3.19	3.18	3.11	3.11	2.35	2.35	2.36	2.31	2.15	2.14	2.15	2.14	0.88	0.88	0.87	0.87	0.85	0.84	0.84	1.56	1.48	1.46	1.46	0.68	0.69	0.61	0.58	--	
$\Delta E_{\text{3CT-1CT}} [\text{eV}]$	0.18	0.18	0.18	0.18	0.18	0.18	0.18	0.18	0.24	0.24	0.24	0.24	0.23	0.23	0.23	0.23	0.14	0.14	0.14	0.14	0.14	0.14	0.14	0.20	0.20	0.20	0.20	0.20	0.20	0.20	0.20	0.20	
$f_{\text{S1-S0}}$	0.14	0.14	0.14	0.14	0.14	0.14	0.14	0.14	0.19	0.19	0.18	0.19	0.18	0.18	0.18	0.18	0.11	0.11	0.11	0.11	0.11	0.11	0.11	0.15	0.15	0.15	0.15	0.15	0.15	0.15	0.15	0.118	
$E_a [\text{eV}]$	0.28	0.28	0.28	0.28	0.28	0.28	0.28	0.28	0.32	0.32	0.32	0.32	0.31	0.31	0.31	0.31	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.29	0.29	0.29	0.29	0.29	0.29	0.29	--		
$\mu [\%]$	6.90	0.92	2.12	1.97	1.12	0.95	2.49	2.67	2.18	2.18	3.86	4.14	4.63	4.61	14.68	2.94	0.41	0.46	0.74	0.67	0.56	1.96	0.70	0.65	4.13	3.48	3.53	3.57	5.66	10.15	2.73	2.24	--
$k_{\text{ISC}} [10^4 \text{s}^{-1}]$	8.0	8.2	8.0	8.0	7.1	7.1	7.2	7.1	0.9	0.9	1.0	0.9	0.8	0.8	0.8	0.8	1.5	1.6	1.4	1.5	1.5	1.5	1.5	1.0	0.9	0.9	0.2	0.2	0.2	0.1	0.209		
θ_L	18.1	-18.2	17.5	-17.5	-16.9	16.9	-17.3	17.3	-16.6	16.6	-17.7	17.8	15.5	-15.5	15.7	-15.6	19.5	-19.7	17.5	-17.4	-18.0	18.0	-18.8	18.8	16.1	-15.8	14.5	-14.3	-17.7	17.6	-19.0	18.9	
θ_A	-71.0	71.0	61.2	-61.2	60.9	-60.9	-71.5	71.5	77.8	-77.8	77.8	-77.5	77.9	78.0	-77.9	76.5	-76.4	-83.5	-61.8	75.9	-75.9	61.0	-61.0	75.4	-75.3	-75.8	75.9	75.1	-75.1	74.8	76.6		
θ_B	61.4	-61.3	-70.9	70.9	-71.4	71.4	60.9	-60.9	73.6	-73.6	73.6	-73.6	74.0	-74.0	74.0	-74.0	61.7	76.8	-76.7	60.8	-60.8	75.5	-75.5	-75.7	75.7	75.2	-75.1	-76.2	76.3	-76.6	-74.8		

^a – statistical mean values (weighted using Boltzmann distribution law - equation S2);

$\text{SOC}_{\text{T1-S1}} [\text{cm}^{-1}]$ – SOC constant value for $^3\text{CT} \rightarrow ^1\text{CT}$ transition;

$\Delta E_{\text{3CT-1CT}} [\text{eV}]$ – Energy gap between ^3CT and ^1CT excited states;

$f_{\text{S1-S0}}$ – oscillator strength;

$E_a [\text{eV}]$ – Activation energy;

$\mu [\%]$ – relative contribution estimated using Boltzmann distribution law - equation S2);

$k_{\text{ISC}} [10^4 \text{s}^{-1}]$ – calculated rate constant for $^3\text{CT} \rightarrow ^1\text{CT}$ transition.

$\theta_L \theta_A \theta_B$ – dihedral angles (**Figure 3**)

Calculations of the ${}^3\text{LE} \rightarrow {}^1\text{CT}$ rate constants. To verify whether ${}^3\text{LE} \rightarrow {}^1\text{CT}$ channel has an impact on rISC, we performed theoretical calculations of its rate constants ($k_{3\text{LE}-1\text{CT}}$) using Marcus-Hush equation (S1) and experimental $\Delta E_{3\text{LE}-1\text{CT}}$ values. Since we consider two triplet states from which rISC is potentially possible, at first, we estimated relative population of these levels using Boltzmann distribution law:

$$\chi_i [\%] = \frac{\exp\left(-\frac{\Delta E_i}{k_B T}\right)}{\sum_{i=1}^N \exp\left(-\frac{\Delta E_i}{k_B T}\right)}, \quad (\text{S3})$$

where ΔE_i denotes the energy difference between lowest triplet state (T_1) and respective triplet state (T_i):

$$\Delta E_i = (T_i - T_1), \quad (\text{S4})$$

$$a_i = \exp\left(-\frac{(T_i - T_1)}{k_B T}\right). \quad (\text{S5})$$

Table S3. Theoretical constant rates $k_{3\text{LE}-1\text{CT}}$ with determined population of lowest triplet excited states of emitters.

Alignment of triplet excited states:			CBP 10%					
			$T_2 = {}^3\text{LE(A)}$		${}^3\text{LE(A)} \quad {}^3\text{CT}$			
T_1	T_2	ΔE_{T2-T1}	a_1	a_2	$\chi_{3\text{LE(A)}}$	$\chi_{3\text{CT}}$	$k_{3\text{LE}-1\text{CT}}$	$\chi_{3\text{LE(A)}} k_{3\text{LE}-1\text{CT}}$
	[eV]	[eV]	[eV]		[%]	[%]	[10^4 s $^{-1}$]	[10^4 s $^{-1}$]
H	2.03	2.25	0.22	1	0.0003	0.03	99.97	186.2
1Br	2.08	2.25	0.17	1	0.0018	0.18	99.82	24.4
2Br	2.11	2.25	0.14	1	0.0070	0.70	99.30	1.1
3Br	2.16	2.25	0.09	1	0.0281	2.92	97.08	3.8

From the results included in Table S3, it can be seen, that population of triplet states is strongly dominated by ${}^3\text{CT}$ state due to large difference in energies between ${}^3\text{CT}$ and ${}^3\text{LE}$ levels.

Next, values of $k_{3\text{LE}-1\text{CT}}$ for each emitter were calculated just as $k_{3\text{CT}-1\text{CT}}$, taking into account population of ${}^3\text{LE}$ state $\chi_{3\text{LE(A)}}$. Results are presented in **Table S3** and **Figure 4**.

Since theoretical predictions of rISC constant rates based on exclusively ${}^3\text{LE} \rightarrow {}^1\text{CT}$ channel did not show a good correlation with experimental values, we conclude that ${}^3\text{LE} \rightarrow {}^1\text{CT}$ channel has not considerable impact on rISC in most of the cases except for the rotamers with very low rates of ${}^3\text{CT} \rightarrow {}^1\text{CT}$ transition as **2Br-anti** ones.

$^3\text{CT}-^1\text{CT}$ orbital moment change

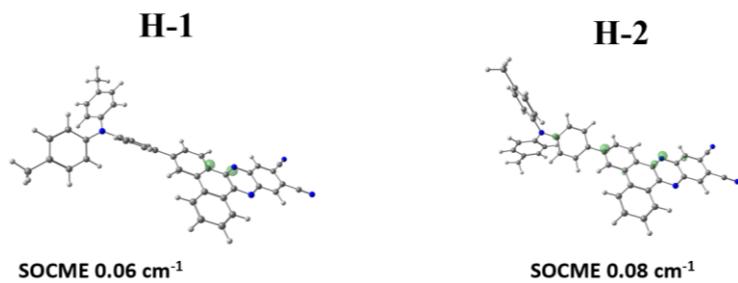


Figure S7. Differences in the orbital transition moment of **H** rotamers. *note that contour value is 0.01 (high)

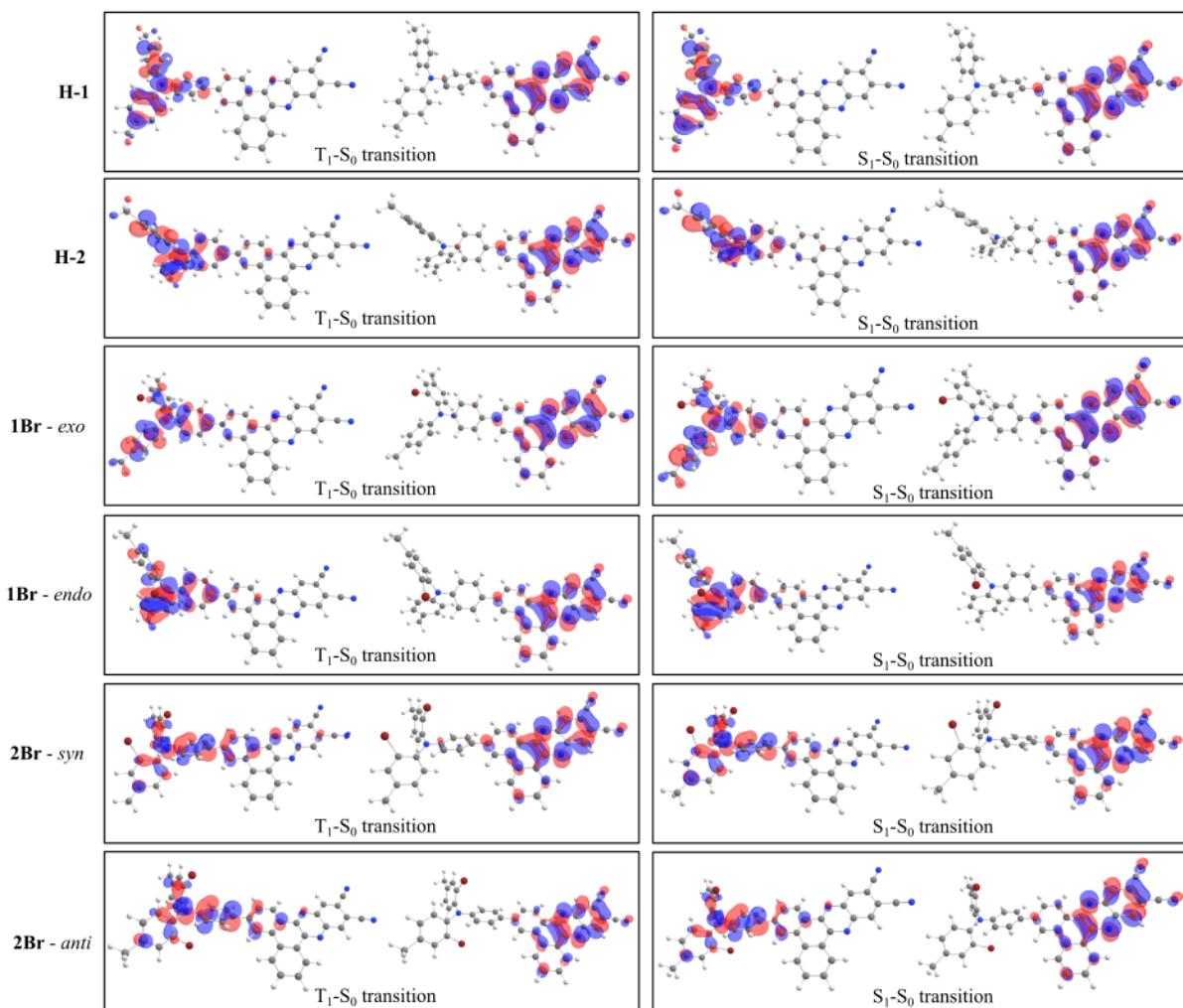


Figure S8. Natural transition orbitals for the S_1-S_0 and T_1-S_0 transitions for selected **H**, **1Br** and **2Br** rotamers. NTO indicate almost negligible role of bromine atoms in the electronic transitions.

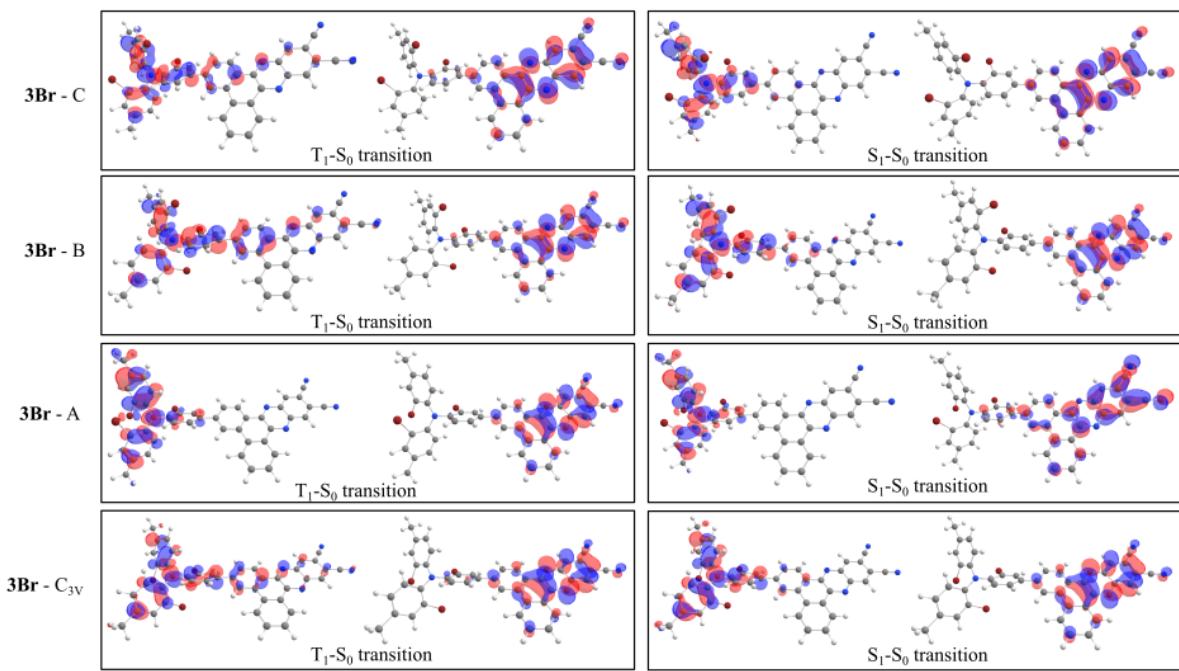


Figure S9. Natural transition orbitals for the S_1 - S_0 and T_1 - S_0 transitions for selected **3Br** rotamers. NTO indicate almost negligible role of bromine atoms in the electronic transitions.

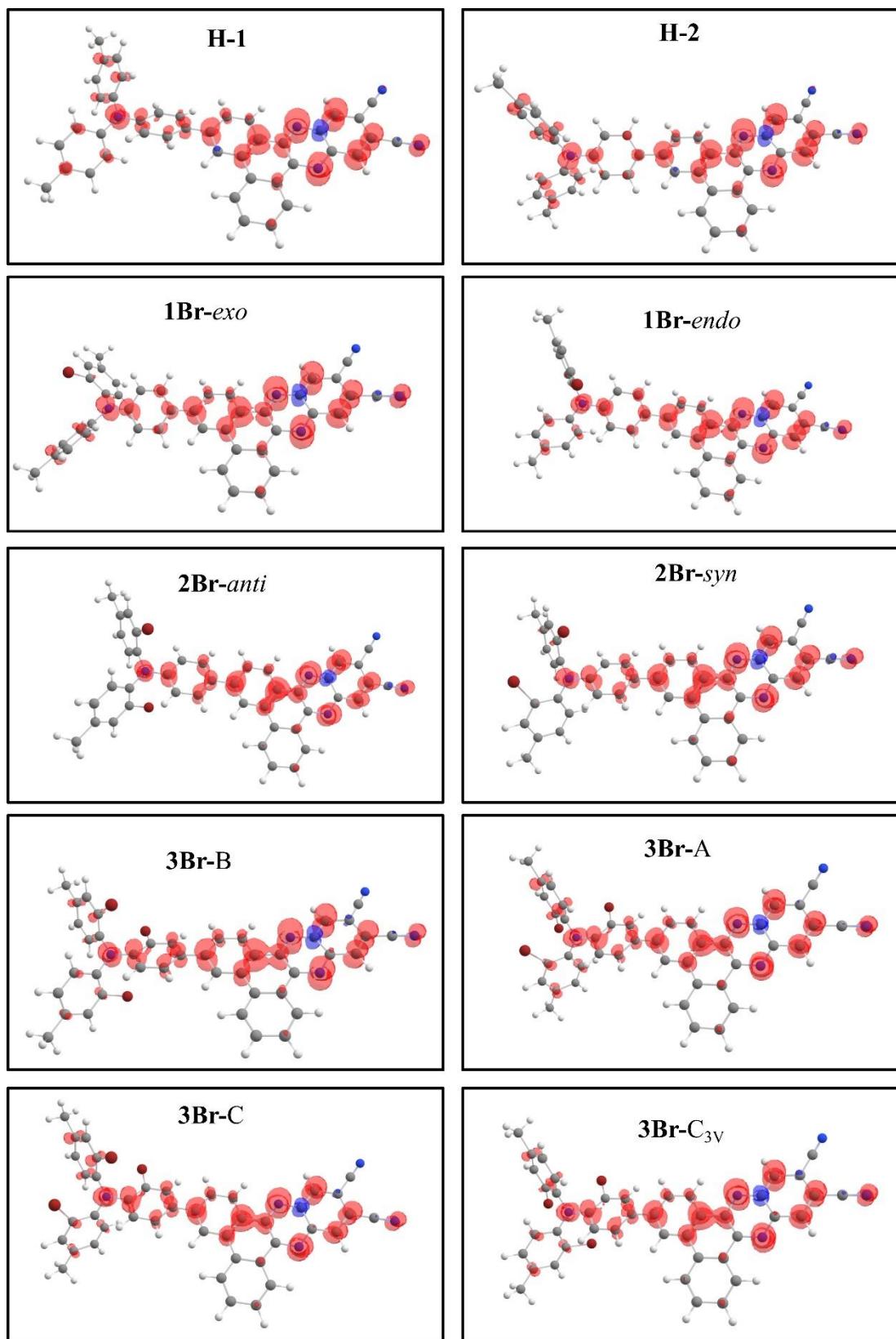
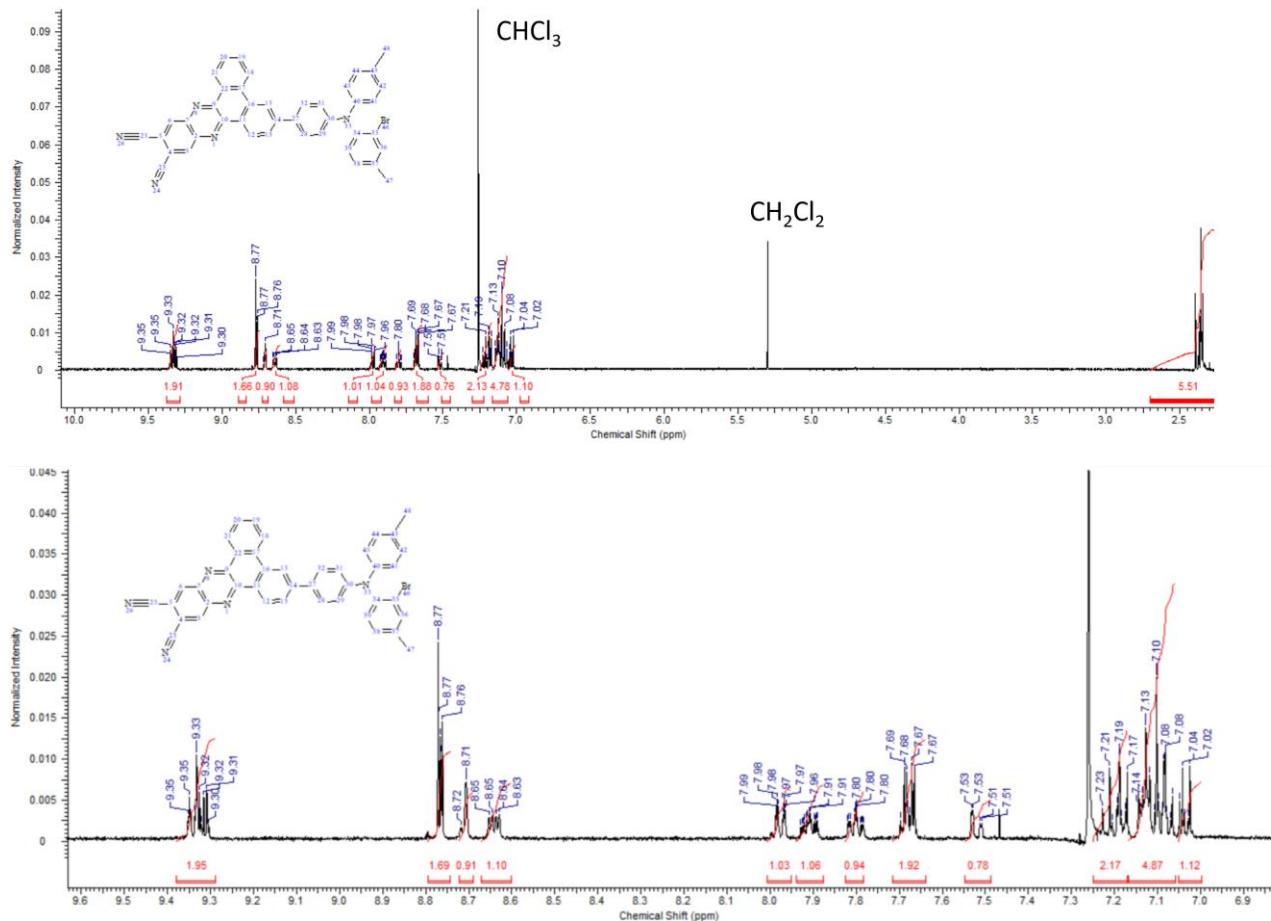


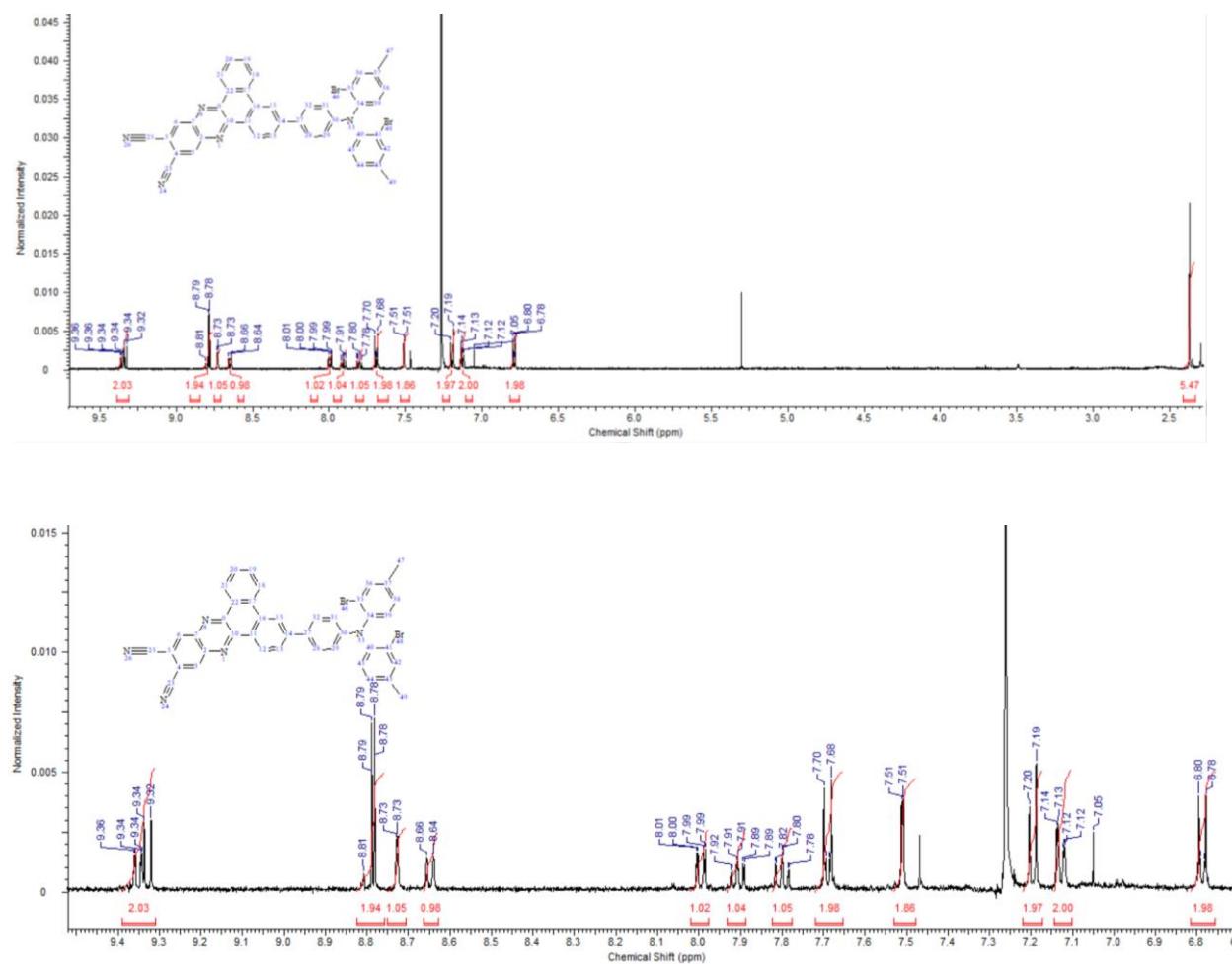
Figure S10. Triplet spin density distribution (TSDD) maps

Section S3: NMR spectra of target emitters

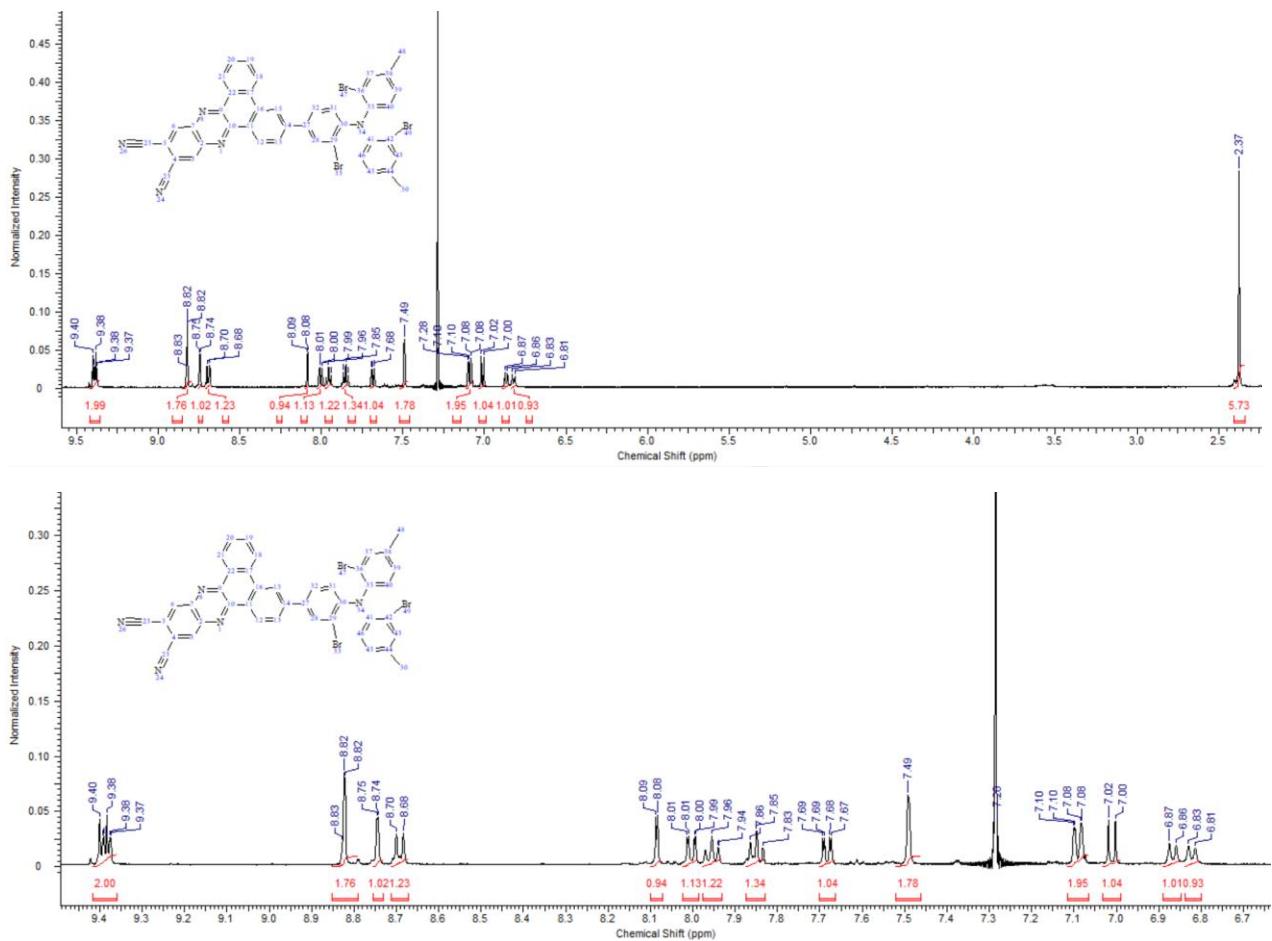
^1H NMR spectrum of 3-(4-((2-bromo-4-methylphenyl)(*p*-tolyl)amino)phenyl)-dibenzo[*a,c*]phenazine-11,12-dicarbonitrile (**1Br**) in CDCl_3



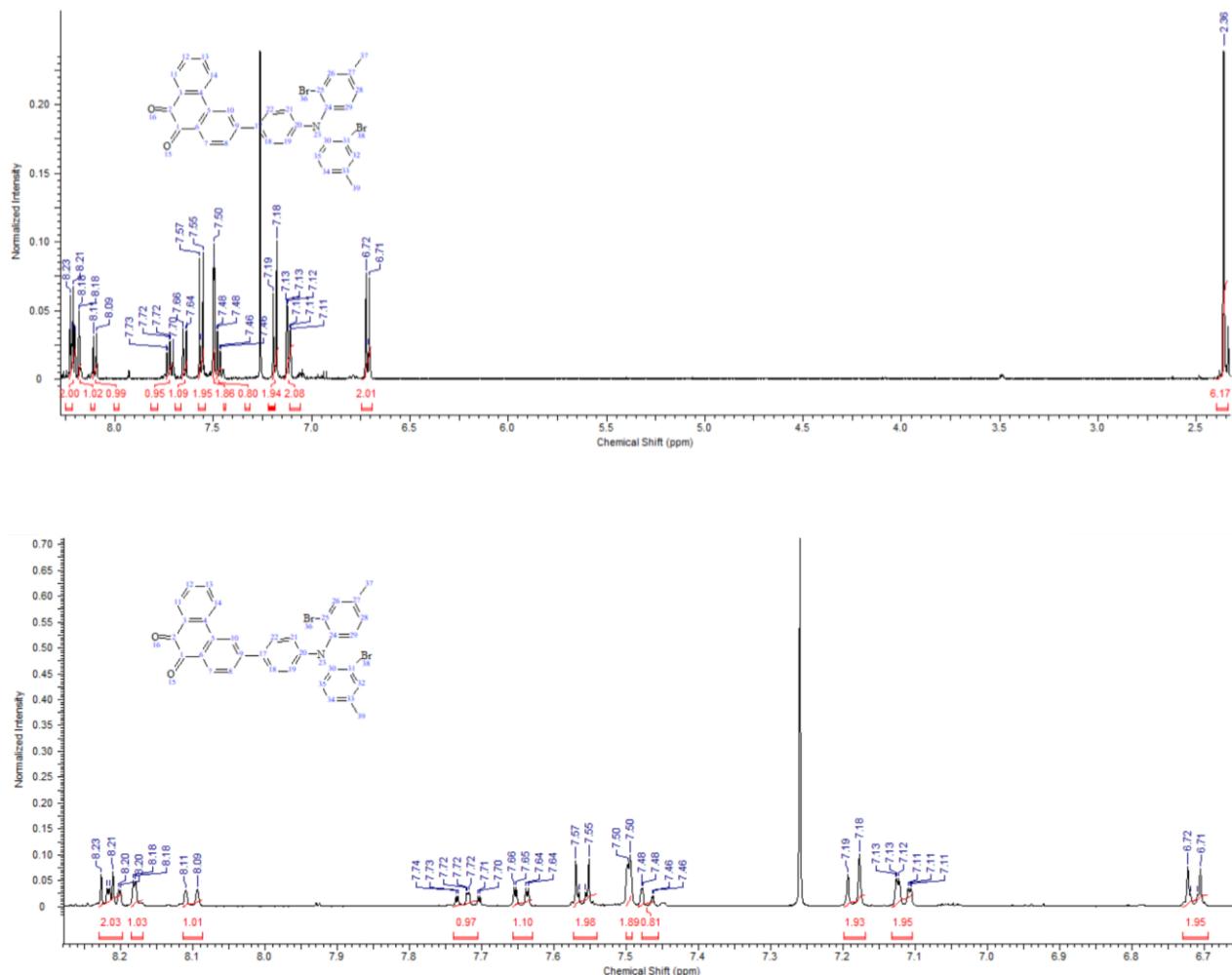
¹H NMR spectrum of 3-(4-(bis(2-bromo-4-methylphenyl)amino)phenyl)dibenzo[*a,c*]phenazine-11,12-dicarbonitrile (**2Br**) in CDCl₃.



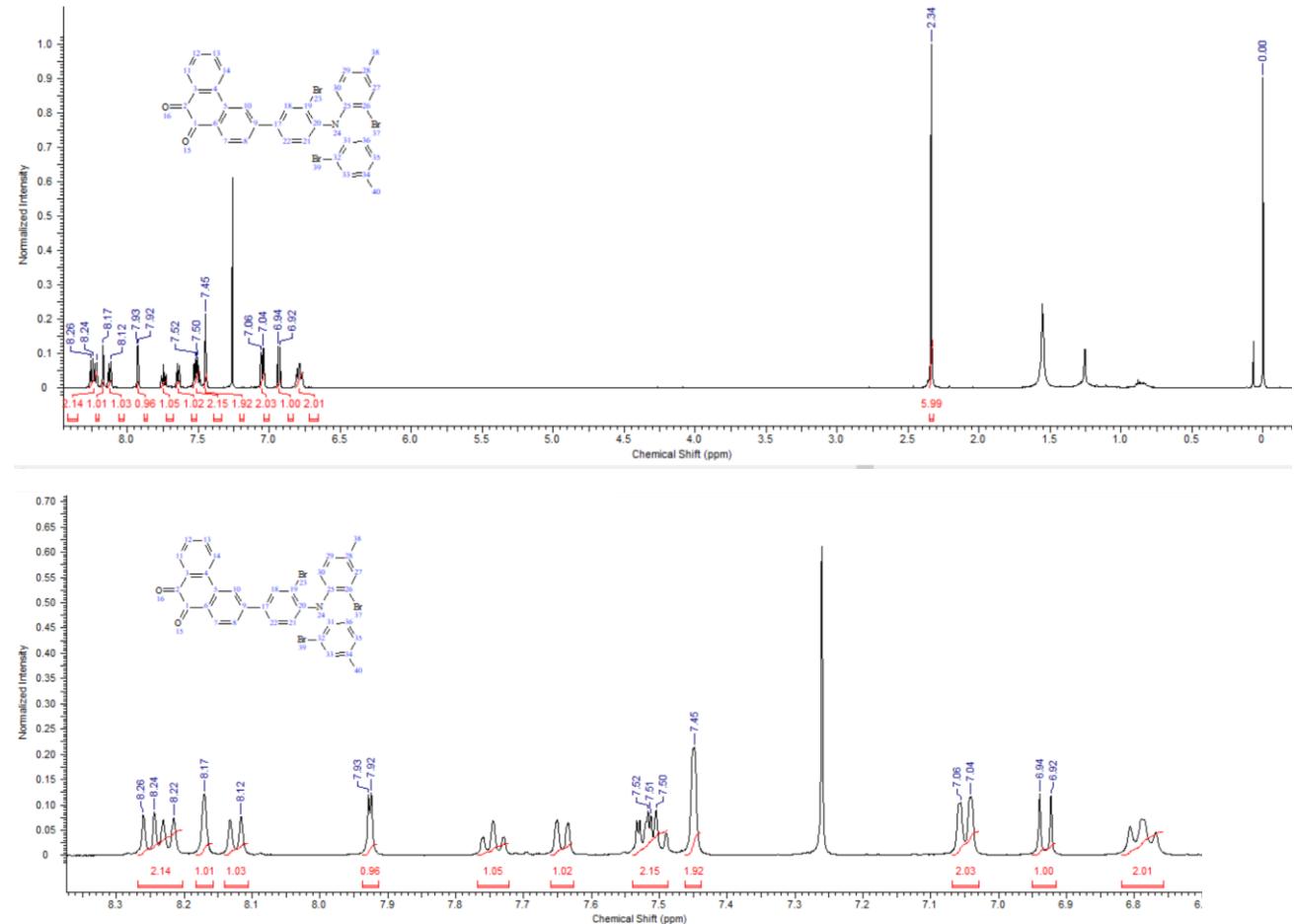
¹H NMR spectrum of 3-(4-(bis(2-bromo-4-methylphenyl)amino)phenyl)dibenzo [*a,c*]phenazine-11,12-dicarbonitrile (**3Br**) in CDCl₃.



¹H NMR spectrum of 3-(4-(bis(2-bromo-4-methylphenyl)amino)phenyl)phenanthrene-9,10-dione (**2**) in CDCl₃.



¹H NMR spectrum of 3-(4-(bis(2-bromo-4-methylphenyl)amino)-3-bromophenyl)-phenanthrene-9,10-dione (**3**) in CDCl₃



Section S4: Determination of photophysical parameters

PL decay curves (presented in **Figures 2F** and **S4**) were fitted with the multiexponential equation:

$$I(t) = A_0 + \sum_{i=1}^n A_i \exp(-t/\tau_i) \quad (\text{S6})$$

where A_i is the pre-exponential factor, τ_i is the decay time and $I(t)$ is emission intensity. Average lifetimes of prompt (τ_{PF}) and delayed fluorescence (τ_{DF}) were determined using the following formula:

$$\tau_{PF}, \tau_{DF} = \sum_{i=1}^n f_i \tau_i, \quad (\text{S7})$$

where f_i is fractional contribution of i -th component expressed as:

$$f_i = \frac{A_i \tau_i}{\sum_{i=1}^n A_i \tau_i} \quad (\text{S8})$$

The ratio of DF and PF quantum yields $\varphi_{DF}/\varphi_{PF}$ was determined as follows:

$$\frac{\varphi_{DF}}{\varphi_{PF}} = \frac{\sum_{i=1}^n \tau_{DF(i)} A_{DF(i)}}{\sum_{j=1}^n \tau_{PF(j)} A_{PF(j)}} \quad (\text{S9})$$

where $A_{DF(i)}$ and $A_{PF(j)}$ is the pre-exponential factor of delayed and prompt fluorescence component, respectively; $\tau_{DF(i)}$ and $\tau_{PF(j)}$ is the lifetime of delayed and prompt fluorescence component, respectively. The rate constants of radiative (k_r) and nonradiative (k_{nr}) decay and intersystem crossing (k_{ISC}) are given by equations[S2]:

$$k_r = \frac{\varphi_{PF}}{\tau_{PF}}, \quad (\text{S10})$$

$$k_{ISC} = \frac{\varphi_{DF}}{\varphi \tau_{PF}}, \quad (\text{S11})$$

$$k_{nr} = \frac{1}{\tau_{PF}} - (k_r + k_{ISC}). \quad (\text{S12})$$

where φ is PLQY ($\varphi_{DF} + \varphi_{PF}$). Further, the quantum yields for ISC and rISC were calculated as

$$\varphi_{ISC} = k_{ISC} \tau_{PF}, \quad (\text{S13})$$

$$\varphi_{rISC} = \frac{1 - \varphi_{PF}/\varphi}{\varphi_{ISC}}. \quad (\text{S14})$$

Finally, the rate constant of rISC (k_{rISC}) was calculated as

$$k_{rISC} = \frac{\varphi_{rISC}}{\tau_{DF}} \left(\frac{\varphi}{\varphi_{PF}} \right). \quad (\text{S15})$$

Thus obtained photophysical parameters are presented in **Table 2** (main text).

References

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