

Supporting Information

2D ZnIn₂S₄ nanosheets in-situ growth on sulfur-doped porous Ti₃C₂T_x MXene 3D multi-functional architectures for photocatalytic H₂ evolution

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Keywords: porous Ti₃C₂T_x, sulfur-doped, Ti-S bonds, 3D architecture, photocatalytic H₂ evolution

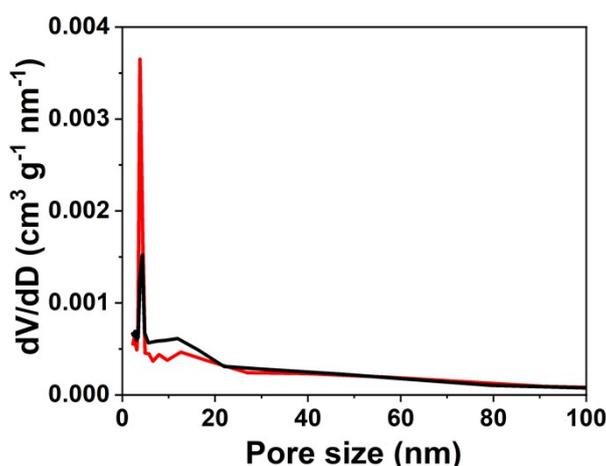


Fig. S1. The pore size distribution curves of the ZnIn₂S₄ and S-doped Ti₃C₂T_x@ZnIn₂S₄ composites

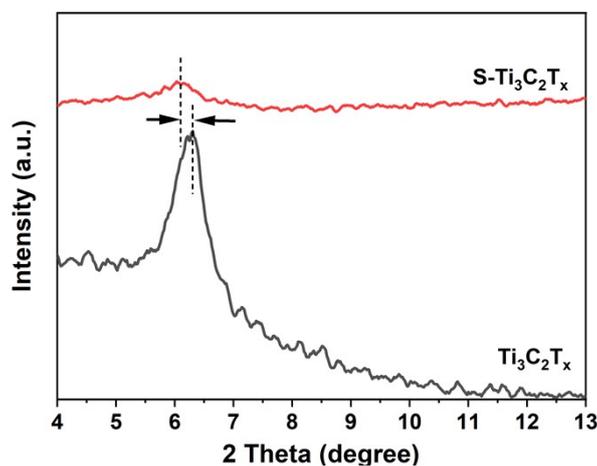


Fig. S2. X-ray diffraction patterns of $\text{Ti}_3\text{C}_2\text{T}_x$ and S-doped $\text{Ti}_3\text{C}_2\text{T}_x$.

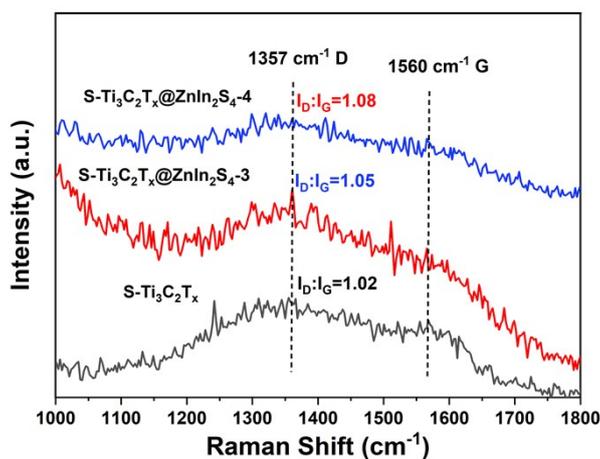


Fig. S3. Raman spectroscopy of S-doped $\text{Ti}_3\text{C}_2\text{T}_x$ and $\text{S-Ti}_3\text{C}_2\text{T}_x@ZnIn_2S_4$ composites.

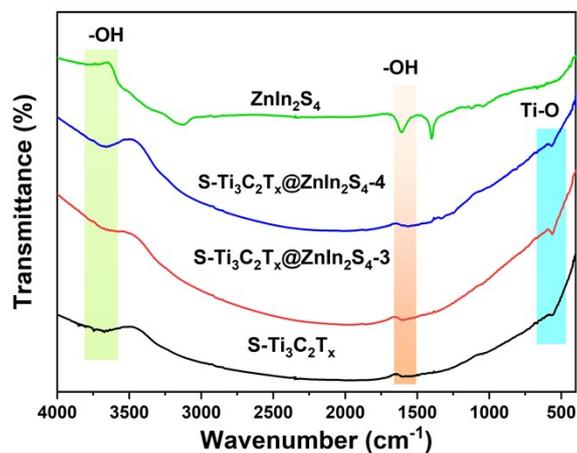


Fig. S4. FT-IR spectra of $ZnIn_2S_4$, $\text{S-Ti}_3\text{C}_2\text{T}_x@ZnIn_2S_4$ composites, and S-doped $\text{Ti}_3\text{C}_2\text{T}_x$.

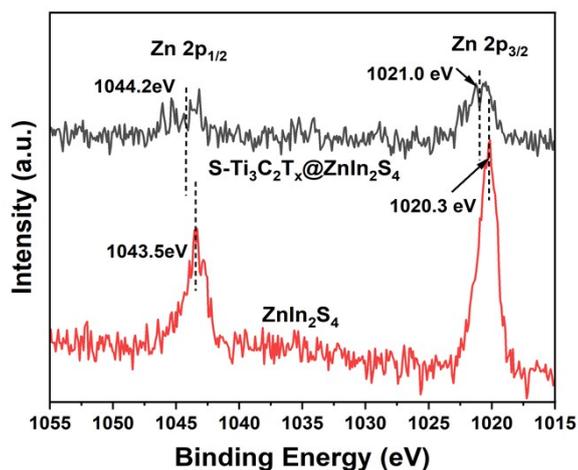


Fig. S5. Zn 2p of $ZnIn_2S_4$ and $\text{S-Ti}_3\text{C}_2\text{T}_x@ZnIn_2S_4$ composites.

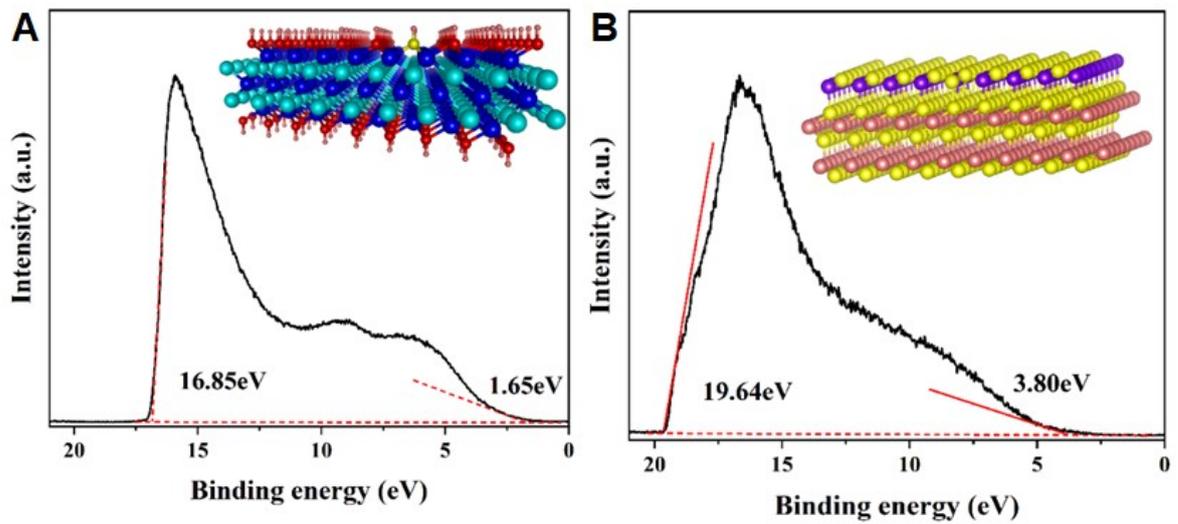


Fig. S6. UPS spectra for A) S doped $\text{Ti}_3\text{C}_2\text{T}_x$ and B) ZnIn_2S_4

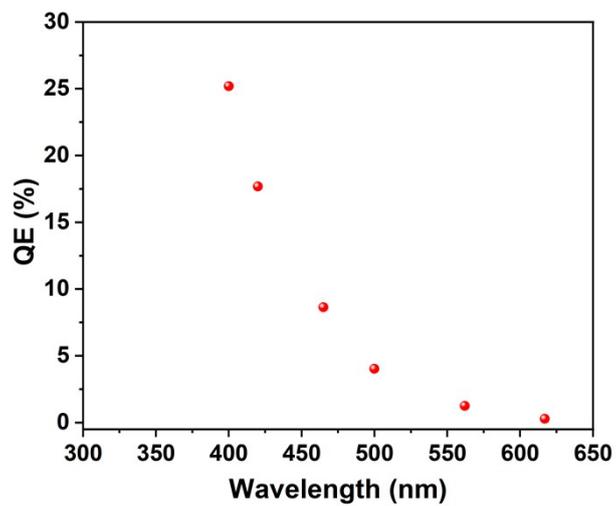


Fig. S7. The QE values of $\text{S-Ti}_3\text{C}_2\text{T}_x@\text{ZnIn}_2\text{S}_4-3$ at different wavelength

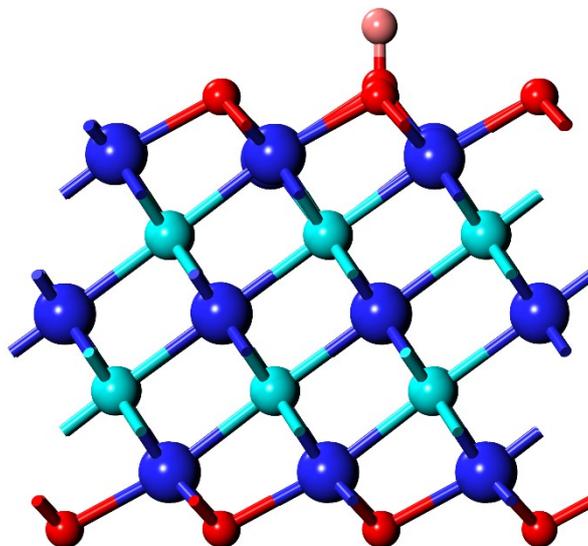


Fig. S8. Side views of the pure $\text{Ti}_3\text{C}_2\text{T}_x$.

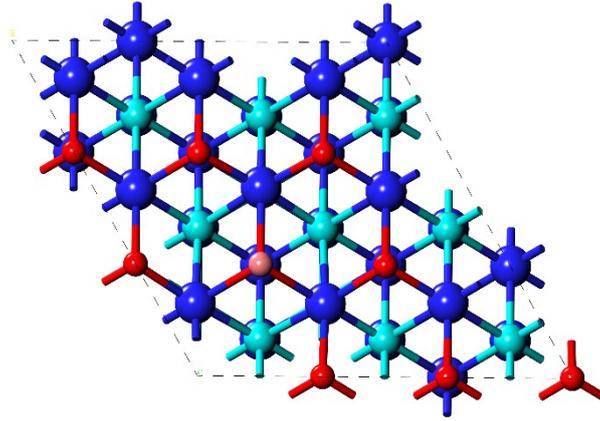


Fig. S9. Top views of the pure $\text{Ti}_3\text{C}_2\text{T}_x$.

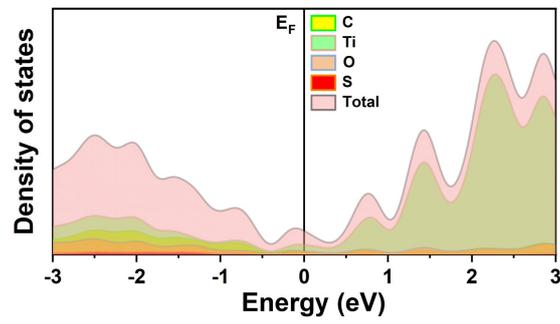


Fig. S10. Density of states of the S doped $\text{Ti}_3\text{C}_2\text{T}_x$.

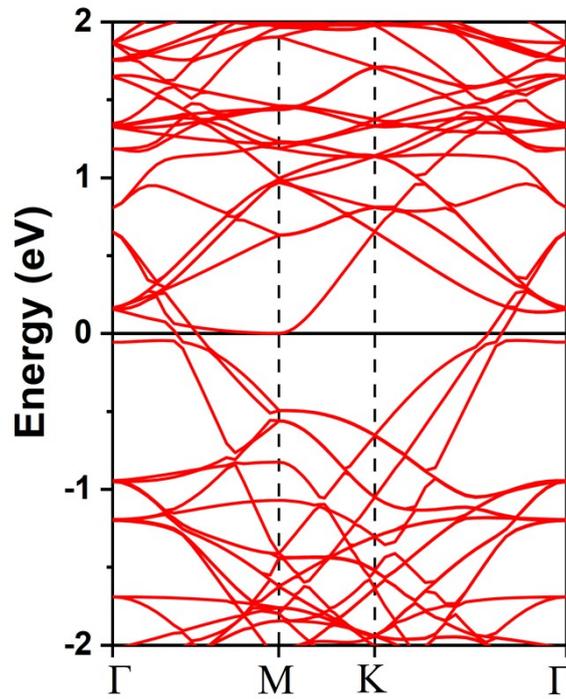


Fig. S11. The electronic band structure of the pure $\text{Ti}_3\text{C}_2\text{T}_x$.

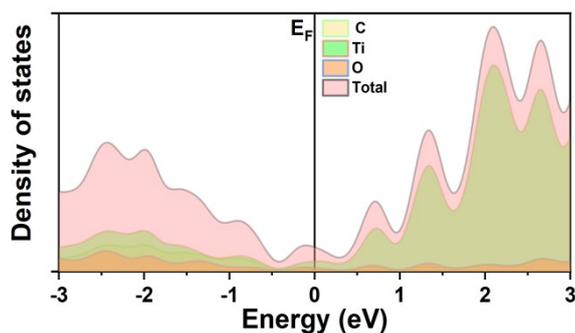


Fig. S12. The density of states of the pure $\text{Ti}_3\text{C}_2\text{T}_x$.

Table S1. Comparison of the photocatalytic H_2 evolution activity for some previous reports photocatalysts

Photocatalyst	Light source	sacrificial agent	H_2 evolution rate ($\text{mmol h}^{-1}\text{g}^{-1}$)	H_2 production QE (%)	Ref.
$\text{CdS}/\text{Ti}_3\text{C}_2$	300 W Xe arc lamp ($\lambda \geq 400$ nm)	TEOA	3.1765	2.28 ($\lambda = 420\text{nm}$)	1
$\text{CuInS}_2/\text{TiO}_2$	300 W Xe lamp ($\lambda \geq 420$ nm)	MEOH	0.655	None	2
$\text{CNFs}/\text{MoS}_2/\text{ZnIn}_2\text{S}_4$	300 W Xe lamp ($\lambda \geq 400$ nm)	aqueous solution	0.151	20.88 ($\lambda = 365\text{nm}$)	3
$\text{NiCo}_2\text{S}_4/\text{ZnIn}_2\text{S}_4$	300 W Xe lamp ($\lambda \geq 400$ nm)	$\text{Na}_2\text{S}/\text{Na}_2\text{SO}_3$	0.77	1.2 ($\lambda = 420\text{nm}$)	4
$\text{MXene}/\text{ZnIn}_2\text{S}_4$	300 W Xe lamp ($\lambda \geq 400$ nm)	TEOA	3.475	11.14 ($\lambda = 420\text{nm}$)	5
$\text{BiVO}_4/\text{Ti}_3\text{C}_2$	300 W Xe arc lamp ($\lambda \geq 420$ nm)	methanol solution	0.196	1.47 ($\lambda = 420\text{nm}$)	6
$\text{ZnIn}_2\text{S}_4/\text{S}$, N co-doped carbon	300 W Xe lamp	TEOA	2.937	19.47 ($\lambda = 435\text{nm}$)	7
$\text{ZnIn}_2\text{S}_4\text{-S}/\text{CNTs}/\text{RP}$	350 W Xe lamp	$\text{Na}_2\text{S}/\text{Na}_2\text{SO}_3$	1.640	None	8
$\text{CuInS}_2@\text{C}_3\text{N}_4$	350 W Xe lamp ($\lambda \geq 420$ nm).	$\text{Na}_2\text{S}/\text{Na}_2\text{SO}_3$	0.373	4.32 ($\lambda = 400\text{nm}$)	9
$\text{TiO}_2/\text{Ti}_3\text{C}_2/\text{g-C}_3\text{N}_4$	300 W Xe lamp ($\lambda \geq 420$ nm).	TEOA	1.409	None	10
$\text{Ti}_3\text{C}_2@\text{TiO}_2/\text{ZnIn}_2\text{S}_4$	300 W Xe lamp ($\lambda \geq 400$ nm)	$\text{Na}_2\text{S}/\text{Na}_2\text{SO}_3$	1.186	None	11
$\text{CdS}/\text{ZnIn}_2\text{S}_4$	300 W Xe lamp	$\text{Na}_2\text{S}/\text{Na}_2\text{SO}_3$	3.072	15.9 ($\lambda = 420\text{nm}$)	12
$\text{NH}_2\text{-MIL-125(Ti)} @\text{ZnIn}_2\text{S}_4/\text{CdS}$	300 W Xe lamp ($\lambda \geq 400$ nm)	lactic acid	2.367	None	13
$\text{WO}_3/\text{ZnIn}_2\text{S}_4$	300 W Xenon arc lamp ($\lambda \geq 420$ nm)	$\text{Na}_2\text{S}/\text{Na}_2\text{SO}_3$	1.945	18.68 ($\lambda = 420\text{nm}$)	14

UiO-66/ZnIn ₂ S ₄	300 W Xe lamp ($\lambda \geq 400$ nm)	TEOA	3.062	19.39 ($\lambda = 400$ nm)	15
S-Ti ₃ C ₂ T _x /ZnIn ₂ S ₄	300 W Xe lamp ($\lambda \geq 400$ nm)	Na ₂ S/Na ₂ SO ₃	3.058	17.68 ($\lambda = 420$ nm)	This work

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