

Supporting Information

Thermal Stable Zinc-Based Hybrid Halides with High External Quantum Efficiency as Temperature Detectors

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Table S1. Crystal data and structure refinement for (C₉H₁₅N₃)ZnCl₄ at 293(2) K.

| | |
|---|---|
| Empirical formula | (C ₉ H ₁₅ N ₃)ZnCl ₄ |
| Formula weight | 1111.17 |
| Temperature | 150.0 K |
| Crystal system | monoclinic |
| Space group | <i>P2</i> ₁ |
| Unit-cell dimensions | $a = 10.350(2) \text{ \AA}$, $\alpha = 90^\circ$ $b = 20.076(4) \text{ \AA}$, $\beta = 91.748(7)^\circ$ $c = 20.679(4) \text{ \AA}$, $\gamma = 90^\circ$ |
| Volume | 4294.9(14) Å ³ |
| <i>Z</i> | 4 |
| Density (calculated) | 1.718 g/cm ³ |
| Absorption coefficient | 2.422 mm ⁻¹ |
| <i>F</i> (000) | 2236 |
| θ range for data collection | 1.969 to 24.999° |
| Index ranges | $-12 \leq h \leq 12$, $-23 \leq k \leq 23$, $-24 \leq l \leq 24$ |
| Reflections collected | 60098 |
| Independent reflections | 14944 [<i>R</i> _{int} = 0.0754] |
| Completeness to $\theta = 25.242^\circ$ | 98.8% |
| Refinement method | Full-matrix least-squares on <i>F</i> ² |
| Data/restraints/parameters | 14944/37/953 |
| Goodness-of-fit | 1.076 |
| Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)] | <i>R</i> _{obs} = 0.0588, <i>wR</i> _{obs} = 0.1488 |
| <i>R</i> indices [all data] | <i>R</i> _{all} = 0.0774, <i>wR</i> _{all} = 0.1615 |
| Largest diff. peak and hole | 0.617 and -0.693 e ⁻ Å ⁻³ |

$$R = \frac{\sum ||F_o| - |F_c||}{\sum |F_o|}, wR = \left\{ \frac{\sum [w(|F_o|^2 - |F_c|^2)^2]}{\sum [w(|F_o|^4)]} \right\}^{1/2} \text{ and } w = 1/[\sigma^2(F_o^2) + (0.0778P)^2 + 2.7836P] \text{ where } P = (F_o^2 + 2F_c^2)/3$$

Table S2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for (C₉H₁₅N₃)ZnCl₄ at 150.0 K with estimated standard deviations in parentheses.

| Label | x | y | z | Occupancy | U _{eq} * |
|--------|----------|---------|---------|-----------|-------------------|
| Zn(02) | 6590(2) | 7010(1) | 7131(1) | 1 | 36(1) |
| Zn(03) | 10520(2) | 7060(1) | 486(1) | 1 | 45(1) |
| Zn(04) | 6428(8) | 4961(2) | 1177(3) | 0.69(2) | 31(2) |
| Zn(05) | 9662(10) | 4966(9) | 4533(7) | 0.423(17) | 33(2) |
| Zn(06) | 4058(10) | 7046(6) | 3846(5) | 0.548(16) | 40(2) |

| | | | | | |
|--------|-----------|----------|---------|-----------|-------|
| CI(07) | 8431(3) | 10881(2) | 1871(2) | 1 | 40(1) |
| CI(08) | 5212(3) | 10076(2) | 2062(2) | 1 | 39(1) |
| CI(09) | 8071(3) | 9064(2) | 1560(2) | 1 | 44(1) |
| CI(0A) | 4436(3) | 6829(2) | 7023(2) | 1 | 42(1) |
| CI(0B) | 7819(4) | 9776(2) | 3245(2) | 1 | 55(1) |
| CI(0C) | 8326(4) | 6903(2) | 401(2) | 1 | 47(1) |
| CI(0D) | 7185(3) | 6974(2) | 8210(2) | 1 | 46(1) |
| CI(0E) | 9227(5) | 4003(2) | 4957(2) | 1 | 66(2) |
| CI(0F) | 5754(17) | 4071(5) | 1723(5) | 0.70(4) | 44(3) |
| CI(0G) | 5909(3) | 4912(2) | 120(2) | 1 | 53(1) |
| CI(0H) | 12005(4) | 5151(2) | 4652(2) | 1 | 53(1) |
| CI(0I) | 8722(4) | 5128(2) | 1292(2) | 1 | 51(1) |
| CI(0J) | 7258(4) | 7948(2) | 6683(2) | 1 | 51(1) |
| CI(0K) | 11183(4) | 7901(2) | -114(2) | 1 | 52(1) |
| CI(0L) | 11535(4) | 6092(2) | 202(2) | 1 | 49(1) |
| CI(0M) | 9449(4) | 5005(2) | 3447(2) | 1 | 57(1) |
| CI(0N) | 1762(4) | 6886(2) | 3704(2) | 1 | 59(2) |
| CI(0O) | 5353(15) | 5921(4) | 1495(7) | 0.63(6) | 38(3) |
| CI(0P) | 8778(5) | 5810(3) | 4990(3) | 1 | 81(2) |
| CI(0Q) | 7716(16) | 6102(8) | 6770(8) | 0.56(2) | 45(2) |
| CI(0R) | 11073(4) | 7179(3) | 1567(2) | 1 | 63(2) |
| CI(0S) | 4757(8) | 7878(3) | 3248(4) | 0.655(16) | 46(2) |
| CI(0T) | 4664(15) | 6184(6) | 3328(7) | 0.346(13) | 48(2) |
| CI(0U) | 4198(4) | 7064(3) | 4918(2) | 1 | 84(2) |
| N(00V) | 4753(11) | 10229(7) | 3618(6) | 1 | 44(3) |
| N(00W) | 12318(13) | 5372(8) | 3095(6) | 1 | 55(4) |
| C(00X) | 10669(12) | 7627(8) | 7005(7) | 1 | 37(2) |
| N(00Y) | 11217(12) | 10307(8) | 270(7) | 1 | 61(4) |
| C(00Z) | 4351(11) | 7554(7) | 1545(7) | 1 | 39(3) |
| N(010) | 10548(10) | 7290(6) | 7574(6) | 1 | 37(2) |
| C(011) | 11308(13) | 9291(8) | 904(7) | 1 | 40(3) |
| N(012) | 11380(11) | 9629(6) | 1499(6) | 1 | 41(3) |
| C(013) | 4499(12) | 7602(6) | 375(7) | 1 | 34(3) |
| C(014) | 4333(15) | 6225(9) | -779(8) | 1 | 52(4) |

| | | | | | |
|--------|-----------|-----------|----------|---|-------|
| C(015) | 10558(12) | 7597(8) | 8189(7) | 1 | 37(2) |
| C(016) | 11338(12) | 8679(6) | 2178(7) | 1 | 32(3) |
| C(017) | 12234(13) | 3352(7) | 733(7) | 1 | 46(3) |
| C(32) | 12371(14) | 3651(7) | 1300(8) | 1 | 48(4) |
| N(01A) | 12542(12) | 4711(6) | 1893(6) | 1 | 40(3) |
| C(01C) | 7287(14) | 7541(8) | 4921(7) | 1 | 46(3) |
| C(01D) | 11948(15) | 10346(9) | 1442(7) | 1 | 56(4) |
| N(01E) | 10718(11) | 7282(6) | 6479(5) | 1 | 39(3) |
| N(01F) | 4364(13) | 7272(8) | 936(7) | 1 | 57(4) |
| C(01G) | 10994(14) | 6566(7) | 6427(7) | 1 | 39(3) |
| N(01H) | 4501(11) | 7256(6) | -185(6) | 1 | 37(3) |
| C(01K) | 12370(12) | 3741(8) | 141(7) | 1 | 42(3) |
| C(01L) | 3721(14) | 7246(8) | -1335(8) | 1 | 49(4) |
| H01N() | 3752.33 | 7481.98 | -1755.03 | 1 | 59 |
| C(01M) | 5354(14) | 10541(9) | 4221(6) | 1 | 47(4) |
| N(01N) | 4385(14) | 9616(7) | 4823(7) | 1 | 65(4) |
| C(01O) | 11275(15) | 8399(8) | 2738(9) | 1 | 57(5) |
| N(30) | 11164(12) | 9755(7) | 2632(6) | 1 | 50(3) |
| C(01Q) | 4487(12) | 8238(8) | 1585(7) | 1 | 45(4) |
| C(01R) | 4420(20) | 9570(10) | 3666(8) | 1 | 63(5) |
| C(01S) | 7494(14) | 7239(8) | 4326(7) | 1 | 46(4) |
| N(32) | 12540(11) | 4698(6) | 744(5) | 1 | 40(3) |
| C(01U) | 10699(12) | 8314(7) | 8196(7) | 1 | 37(2) |
| C(01V) | 11771(14) | 6416(8) | 5882(8) | 1 | 52(4) |
| C(01W) | 11001(17) | 7410(9) | 5293(8) | 1 | 57(5) |
| C(01X) | 7310(15) | 8240(10) | 4986(9) | 1 | 61(4) |
| C(01Z) | 10130(15) | 7550(9) | 5853(7) | 1 | 55(4) |
| N(020) | 7832(13) | 8286(7) | 3856(6) | 1 | 57(4) |
| C(021) | 11305(11) | 9356(9) | 2086(7) | 1 | 42(4) |
| C(022) | 4619(12) | 8279(7) | 435(7) | 1 | 37(3) |
| C(023) | 4989(14) | 6612(7) | -271(6) | 1 | 39(3) |
| C(025) | 11063(15) | 8778(8) | 3298(8) | 1 | 52(4) |
| C(026) | 4502(19) | 10384(10) | 4822(8) | 1 | 69(5) |
| C(027) | 10731(12) | 8311(7) | 7047(7) | 1 | 37(2) |

| | | | | | |
|--------|-----------|----------|----------|-----------|--------|
| N(029) | 4323(9) | 6585(5) | -1395(5) | 1 | 28(2) |
| C(02A) | 10591(14) | 9676(8) | 383(7) | 1 | 42(3) |
| C(02B) | 3710(20) | 9386(11) | 4245(8) | 1 | 80(6) |
| C(02C) | 11195(14) | 10739(8) | 884(6) | 1 | 39(3) |
| C(02D) | 12513(12) | 4365(8) | 187(6) | 1 | 41(3) |
| C(0) | 12380(20) | 4350(11) | 2479(10) | 0.53(2) | 31(4) |
| C(02F) | 4596(12) | 8603(8) | 1043(7) | 1 | 42(4) |
| C(02H) | 7596(15) | 8604(9) | 4435(7) | 1 | 59(4) |
| C(02I) | 12110(20) | 5443(9) | 1883(8) | 1 | 73(6) |
| C(02J) | 10772(12) | 8651(8) | 7628(6) | 1 | 37(2) |
| N(02L) | 11214(12) | 6643(6) | 5269(6) | 1 | 46(3) |
| C02N() | 4395(18) | 7644(9) | -831(8) | 1 | 59(4) |
| C(02O) | 10986(17) | 9458(10) | 3219(8) | 1 | 64(5) |
| C(02Q) | 12525(13) | 4373(8) | 1315(7) | 1 | 41(3) |
| C(2B) | 12510(30) | 4635(13) | 3051(14) | 0.52(3) | 29(5) |
| C(4) | 13260(20) | 4502(12) | 2480(11) | 0.47(2) | 31(4) |
| C(1) | 11610(30) | 5620(15) | 2454(12) | 0.376(18) | 27(3) |
| Cl(1) | 5780(40) | 5825(13) | 1740(30) | 0.40(6) | 63(10) |
| Zn(8) | 7396(2) | 9915(1) | 2165(1) | 1 | 33(1) |
| Zn(2) | 6792(13) | 4936(6) | 1270(6) | 0.29(2) | 31(2) |
| Zn(1) | 3639(10) | 7021(8) | 3820(7) | 0.425(15) | 40(2) |
| Zn(0A) | 10053(11) | 4955(6) | 4514(5) | 0.540(17) | 33(2) |
| Cl(1A) | 4278(16) | 7974(7) | 3409(8) | 0.333(14) | 46(2) |
| Cl(2A) | 7550(20) | 6194(10) | 6561(9) | 0.47(2) | 45(2) |
| Cl(2) | 6180(30) | 3979(12) | 1708(13) | 0.30(4) | 44(3) |
| Cl(3) | 5074(8) | 6095(3) | 3553(4) | 0.643(14) | 48(2) |
| C(3AA) | 3821(14) | 9382(10) | 6498(8) | 1 | 54(4) |
| C(1AA) | 3896(13) | 8739(8) | 6583(7) | 1 | 45(3) |
| C(0AA) | 4027(14) | 8357(8) | 6047(9) | 1 | 57(4) |
| N(0AA) | 4153(13) | 8639(8) | 5475(8) | 1 | 76(5) |
| C(5) | 3951(12) | 9690(8) | 5959(6) | 1 | 37(3) |
| C(6) | 4100(13) | 9344(9) | 5416(7) | 1 | 49(3) |
| N(1) | 8107(10) | 6696(5) | 1967(5) | 1 | 31(2) |
| N(2AA) | 8014(13) | 7315(6) | 3205(5) | 1 | 44(3) |

| | | | | | |
|--------|-----------|----------|----------|-----------|-------|
| C(3) | 7786(14) | 7600(7) | 3791(7) | 1 | 45(3) |
| C(5AA) | 8090(30) | 7747(13) | 2610(12) | 0.57(2) | 51(5) |
| C(4AA) | 7690(20) | 7449(10) | 2043(9) | 1 | 87(8) |
| C(6AA) | 8030(40) | 6297(14) | 2580(13) | 0.57(3) | 64(4) |
| C(7) | 8317(19) | 6658(7) | 3113(8) | 1 | 64(4) |
| C(8) | 8850(50) | 6520(16) | 2558(15) | 0.47(3) | 64(4) |
| C(9) | 7210(40) | 7660(16) | 2592(15) | 0.45(2) | 51(5) |
| C(10) | 12562(17) | 5796(9) | 2462(7) | 0.624(18) | 27(3) |
| C(2A) | 13140(30) | 4746(13) | 2994(13) | 0.48(3) | 29(5) |

* U_{eq} is defined as one third of the trace of the orthogonalized U_{ij} tensor.

Table S3. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $(\text{C}_9\text{H}_{15}\text{N}_3)\text{ZnCl}_4$ at 150.0 K with estimated standard deviations in parentheses.

| Label | U_{11} | U_{22} | U_{33} | U_{12} | U_{13} | U_{23} |
|--------|----------|----------|----------|----------|----------|----------|
| Zn(02) | 33(1) | 36(1) | 39(1) | -1(1) | 3(1) | 0(1) |
| Zn(03) | 61(1) | 37(1) | 36(1) | 5(1) | 0(1) | -1(1) |
| Zn(04) | 39(3) | 31(1) | 23(2) | 7(2) | -8(2) | 3(1) |
| Zn(05) | 34(4) | 29(1) | 37(2) | -3(4) | 6(3) | 0(1) |
| Zn(06) | 45(4) | 40(2) | 34(2) | 5(4) | -5(3) | -2(1) |
| Cl(07) | 38(2) | 38(2) | 45(2) | -7(2) | 10(2) | -2(2) |
| Cl(08) | 30(2) | 46(2) | 40(2) | -3(2) | 1(2) | -4(2) |
| Cl(09) | 45(2) | 37(2) | 51(2) | 2(2) | 6(2) | -8(2) |
| Cl(0A) | 33(2) | 55(2) | 38(2) | -2(2) | -2(2) | 0(2) |
| Cl(0B) | 57(2) | 76(3) | 34(2) | -4(2) | 0(2) | 7(2) |
| Cl(0C) | 63(2) | 43(2) | 36(2) | 2(2) | -1(2) | -2(2) |
| Cl(0D) | 37(2) | 53(2) | 47(2) | 4(2) | -6(2) | 8(2) |
| Cl(0E) | 96(3) | 43(2) | 58(2) | -17(2) | -4(2) | 9(2) |
| Cl(0F) | 47(7) | 32(3) | 52(2) | -1(4) | -17(4) | 13(2) |
| Cl(0G) | 48(2) | 73(2) | 38(2) | 13(2) | 4(2) | 7(2) |
| Cl(0H) | 84(2) | 40(2) | 36(2) | 0(2) | -5(2) | 0(2) |
| Cl(0I) | 70(2) | 47(2) | 36(2) | -2(2) | 2(2) | 6(2) |
| Cl(0J) | 50(2) | 43(2) | 60(2) | -6(2) | 11(2) | 6(2) |
| Cl(0K) | 71(2) | 39(2) | 47(2) | -5(2) | -1(2) | -1(2) |
| Cl(0L) | 64(2) | 36(2) | 47(2) | 5(2) | 11(2) | 5(2) |
| Cl(0M) | 78(2) | 53(2) | 39(2) | -6(2) | -14(2) | 4(2) |
| Cl(0N) | 99(3) | 41(2) | 37(2) | 12(2) | 9(2) | -1(2) |
| Cl(0O) | 43(4) | 38(3) | 31(5) | 5(3) | -2(4) | 0(3) |

| | | | | | | |
|--------|--------|--------|--------|--------|--------|--------|
| Cl(OP) | 68(3) | 72(3) | 104(4) | -18(2) | 24(2) | -38(3) |
| Cl(OQ) | 47(4) | 34(5) | 55(8) | 12(3) | 7(5) | 1(5) |
| Cl(OR) | 65(2) | 87(3) | 37(2) | 17(2) | -16(2) | -5(2) |
| Cl(OS) | 56(5) | 34(3) | 47(4) | 4(3) | 2(3) | -1(2) |
| Cl(OT) | 69(5) | 33(3) | 42(4) | 6(3) | 12(3) | 11(3) |
| Cl(OU) | 70(3) | 138(5) | 43(2) | 44(3) | 0(2) | 8(3) |
| N(00V) | 36(6) | 62(9) | 35(6) | -4(5) | -1(4) | -13(6) |
| N(00W) | 54(7) | 81(10) | 29(6) | 15(7) | -8(5) | 12(6) |
| C(00X) | 38(2) | 38(3) | 34(2) | 9(2) | -6(2) | -2(2) |
| N(00Y) | 42(7) | 92(11) | 51(8) | -10(7) | 4(5) | -23(7) |
| C(00Z) | 31(6) | 37(7) | 49(8) | -3(5) | -2(5) | -13(6) |
| N(010) | 38(2) | 38(3) | 34(2) | 9(2) | -6(2) | -2(2) |
| C(011) | 43(7) | 38(7) | 40(7) | -6(5) | 5(5) | -6(6) |
| N(012) | 37(6) | 44(7) | 42(7) | -10(5) | 4(5) | -9(5) |
| C(013) | 37(6) | 21(6) | 44(7) | -6(5) | 3(5) | -18(5) |
| C(014) | 50(8) | 45(9) | 61(10) | 12(7) | 9(7) | 1(8) |
| C(015) | 38(2) | 38(3) | 34(2) | 9(2) | -6(2) | -2(2) |
| C(016) | 36(6) | 15(6) | 46(8) | 8(5) | -9(5) | 1(5) |
| C(017) | 60(8) | 25(6) | 53(8) | 1(6) | 12(6) | -13(6) |
| C(32) | 66(9) | 25(6) | 54(8) | 0(6) | 22(7) | 15(5) |
| N(01A) | 63(7) | 26(6) | 30(6) | 14(5) | 0(5) | -4(4) |
| C(01C) | 68(9) | 41(8) | 30(7) | -9(6) | 0(6) | 8(6) |
| C(01D) | 56(8) | 69(10) | 41(8) | -32(7) | -16(6) | 15(7) |
| N(01E) | 45(6) | 50(7) | 22(5) | 6(5) | 8(4) | 6(5) |
| N(01F) | 55(7) | 46(8) | 68(10) | -3(6) | -2(6) | 0(7) |
| C(01G) | 59(8) | 27(7) | 32(7) | 24(6) | -9(6) | 10(5) |
| N(01H) | 52(6) | 24(5) | 35(6) | 0(4) | 11(5) | 7(4) |
| C(01K) | 43(7) | 46(8) | 39(7) | 13(6) | 8(5) | -8(6) |
| C(01L) | 47(8) | 49(9) | 51(9) | 8(6) | -13(6) | 5(7) |
| C(01M) | 56(8) | 60(9) | 26(6) | -18(7) | 2(5) | 11(6) |
| N(01N) | 93(10) | 46(7) | 57(8) | -52(7) | 19(7) | -9(6) |
| C(01O) | 64(9) | 29(7) | 77(12) | 5(6) | -30(8) | 5(8) |
| N(30) | 58(7) | 40(7) | 52(8) | 6(5) | -10(5) | 9(6) |
| C(01Q) | 29(6) | 61(10) | 45(8) | -15(6) | -4(5) | -16(7) |
| C(01R) | 93(13) | 57(11) | 38(9) | -36(9) | 2(8) | 7(8) |
| C(01S) | 67(9) | 30(7) | 43(8) | -1(6) | 8(6) | 5(6) |
| N(32) | 51(6) | 42(6) | 27(5) | -9(5) | -3(4) | 0(4) |
| C(01U) | 38(2) | 38(3) | 34(2) | 9(2) | -6(2) | -2(2) |

| | | | | | | |
|--------|---------|--------|--------|---------|--------|---------|
| C(01V) | 48(8) | 46(8) | 60(9) | 17(6) | -8(6) | -5(7) |
| C(01W) | 61(9) | 67(11) | 44(9) | 28(8) | 2(7) | 14(8) |
| C(01X) | 58(9) | 68(11) | 56(10) | 7(8) | -2(7) | -6(8) |
| C(01Z) | 64(9) | 60(10) | 40(8) | 32(7) | -3(6) | 16(7) |
| N(020) | 89(9) | 53(8) | 30(6) | -12(7) | 4(6) | -11(6) |
| C(021) | 24(6) | 71(10) | 32(7) | -3(6) | 0(5) | -18(7) |
| C(022) | 29(6) | 39(8) | 44(8) | -6(5) | 8(5) | -12(6) |
| C(023) | 51(7) | 32(7) | 33(7) | 7(5) | -9(5) | 3(5) |
| C(025) | 68(9) | 47(8) | 41(8) | -13(7) | -6(7) | 7(7) |
| C(026) | 100(13) | 73(12) | 36(8) | -17(10) | 19(8) | 3(8) |
| C(027) | 38(2) | 38(3) | 34(2) | 9(2) | -6(2) | -2(2) |
| N(029) | 34(5) | 19(5) | 31(5) | 4(4) | 9(4) | -12(4) |
| C(02A) | 52(7) | 44(8) | 32(7) | -11(6) | 12(5) | -10(6) |
| C(02B) | 104(14) | 90(14) | 47(9) | -59(11) | 20(8) | -1(9) |
| C(02C) | 49(7) | 41(8) | 26(6) | -1(6) | 1(5) | -1(5) |
| C(02D) | 40(7) | 56(8) | 29(6) | 0(6) | 13(5) | 11(6) |
| C(0) | 41(9) | 23(8) | 30(8) | -3(7) | -2(8) | -5(7) |
| C(02F) | 29(6) | 50(9) | 47(8) | -1(5) | -2(5) | -17(7) |
| C(02H) | 73(10) | 62(10) | 42(8) | -10(8) | 6(7) | -15(7) |
| C(02I) | 133(17) | 46(9) | 38(9) | -3(10) | -4(9) | -14(7) |
| C(02J) | 38(2) | 38(3) | 34(2) | 9(2) | -6(2) | -2(2) |
| N(02L) | 69(8) | 28(6) | 40(7) | 23(5) | 4(5) | -17(5) |
| C(02N) | 83(11) | 45(9) | 47(9) | 6(8) | 8(7) | 6(7) |
| C(02O) | 80(11) | 78(12) | 33(8) | -8(9) | -8(7) | -12(8) |
| C(02Q) | 40(7) | 44(7) | 40(7) | 5(5) | 8(5) | 8(5) |
| C(2B) | 38(14) | 21(9) | 28(8) | 3(10) | 3(11) | 8(7) |
| C(4) | 41(9) | 23(8) | 30(8) | -3(7) | -2(8) | -5(7) |
| C(1) | 45(8) | 28(8) | 8(6) | -5(6) | -4(6) | 1(5) |
| Cl(1) | 60(13) | 54(7) | 76(19) | -13(8) | 30(14) | -18(10) |
| Zn(8) | 34(1) | 34(1) | 31(1) | -4(1) | 3(1) | -1(1) |
| Zn(2) | 39(3) | 31(1) | 23(2) | 7(2) | -8(2) | 3(1) |
| Zn(1) | 45(4) | 40(2) | 34(2) | 5(4) | -5(3) | -2(1) |
| Zn(0A) | 34(4) | 29(1) | 37(2) | -3(4) | 6(3) | 0(1) |
| Cl(1A) | 56(5) | 34(3) | 47(4) | 4(3) | 2(3) | -1(2) |
| Cl(2A) | 47(4) | 34(5) | 55(8) | 12(3) | 7(5) | 1(5) |
| Cl(2) | 47(7) | 32(3) | 52(2) | -1(4) | -17(4) | 13(2) |
| Cl(3) | 69(5) | 33(3) | 42(4) | 6(3) | 12(3) | 11(3) |
| C(3AA) | 34(7) | 85(11) | 45(8) | 5(7) | 1(5) | -1(7) |

| | | | | | | |
|--------|---------|--------|---------|--------|---------|--------|
| C(1AA) | 51(8) | 51(9) | 34(7) | 16(6) | -1(5) | 7(6) |
| C(0AA) | 47(8) | 29(7) | 96(13) | -9(6) | 7(7) | 23(8) |
| N(0AA) | 70(8) | 41(7) | 121(13) | 4(6) | 51(8) | -7(8) |
| C(5) | 34(6) | 42(7) | 34(7) | 9(5) | 1(5) | 4(5) |
| C(6) | 44(7) | 52(8) | 53(8) | -11(6) | 21(6) | -2(6) |
| N(1) | 41(5) | 26(5) | 28(5) | 8(4) | 8(4) | -10(4) |
| N(2AA) | 78(8) | 34(6) | 20(5) | -10(5) | 2(5) | 0(4) |
| C(3) | 63(8) | 21(6) | 50(8) | -13(6) | -1(6) | -7(6) |
| C(5AA) | 57(12) | 51(11) | 43(10) | 3(11) | -11(11) | 20(8) |
| C(4AA) | 154(19) | 64(12) | 40(10) | 52(13) | -27(11) | 7(8) |
| C(6AA) | 123(12) | 25(5) | 45(6) | 10(6) | 4(7) | 9(4) |
| C(7) | 123(12) | 25(5) | 45(6) | 10(6) | 4(7) | 9(4) |
| C(8) | 123(12) | 25(5) | 45(6) | 10(6) | 4(7) | 9(4) |
| C(9) | 57(12) | 51(11) | 43(10) | 3(11) | -11(11) | 20(8) |
| C(10) | 45(8) | 28(8) | 8(6) | -5(6) | -4(6) | 1(5) |
| C(2A) | 38(14) | 21(9) | 28(8) | 3(10) | 3(11) | 8(7) |

The anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^*U_{11} + \dots + 2hka^*b^*U_{12}]$.

Table S4. Bond lengths [Å] for (C₉H₁₅N₃)ZnCl₄ at 150.0 K with estimated standard deviations in parentheses.

| Label | Distances | | |
|---------------|-----------|---------------|-----------|
| Zn(02)-Cl(0A) | 2.264(3) | Zn(06)-Cl(0U) | 2.217(11) |
| Zn(02)-Cl(0D) | 2.297(4) | Zn(06)-Cl(1A) | 2.086(18) |
| Zn(02)-Cl(0J) | 2.218(4) | Zn(06)-Cl(3) | 2.271(13) |
| Zn(02)-Cl(0Q) | 2.301(16) | Cl(07)-Zn(8) | 2.306(4) |
| Zn(02)-Cl(2A) | 2.27(2) | Cl(08)-Zn(8) | 2.287(3) |
| Zn(03)-Cl(0C) | 2.294(4) | Cl(09)-Zn(8) | 2.242(4) |
| Zn(03)-Cl(0K) | 2.217(5) | Cl(0B)-Zn(8) | 2.279(4) |
| Zn(03)-Cl(0L) | 2.295(4) | Cl(0E)-Zn(0A) | 2.297(13) |
| Zn(03)-Cl(0R) | 2.303(4) | Cl(0F)-Zn(2) | 2.260(16) |
| Zn(04)-Cl(0F) | 2.237(10) | Cl(0G)-Zn(2) | 2.522(12) |
| Zn(04)-Cl(0G) | 2.237(8) | Cl(0H)-Zn(0A) | 2.069(12) |
| Zn(04)-Cl(0I) | 2.403(9) | Cl(0I)-Zn(2) | 2.034(14) |
| Zn(04)-Cl(0O) | 2.329(10) | Cl(0M)-Zn(0A) | 2.276(12) |
| Zn(04)-Cl(1) | 2.205(13) | Cl(0N)-Zn(1) | 1.969(11) |
| Zn(04)-Cl(2) | 2.27(2) | Cl(0O)-Cl(1) | 0.69(5) |
| Zn(05)-Cl(0E) | 2.176(17) | Cl(0O)-Zn(2) | 2.527(16) |

| | | | |
|---------------|-----------|---------------|-----------|
| Zn(05)-Cl(0H) | 2.459(11) | Cl(0P)-Zn(0A) | 2.394(13) |
| Zn(05)-Cl(0M) | 2.250(15) | Cl(0S)-Zn(1) | 2.405(17) |
| Zn(05)-Cl(0P) | 2.157(16) | Cl(0S)-Cl(1A) | 0.636(13) |
| Zn(06)-Cl(0N) | 2.407(11) | Cl(0T)-Zn(1) | 2.25(2) |
| Zn(06)-Cl(0S) | 2.213(13) | Cl(0T)-Cl(3) | 0.646(12) |
| Zn(06)-Cl(0T) | 2.140(17) | Cl(0U)-Zn(1) | 2.328(13) |

Table S5. Bond angles [°] for (C₉H₁₅N₃)ZnCl₄ at 150.0 K with estimated standard deviations in parentheses.

| Label | Angles | Label | Angles |
|----------------------|------------|----------------------|------------|
| Cl(0A)-Zn(02)-Cl(0D) | 108.97(14) | Cl(1)-Cl(0O)-Zn(2) | 63.3(14) |
| Cl(0A)-Zn(02)-Cl(0Q) | 110.3(5) | Zn(05)-Cl(0P)-Zn(0A) | 8.4(5) |
| Cl(0A)-Zn(02)-Cl(2A) | 106.2(6) | Zn(06)-Cl(0S)-Zn(1) | 9.8(4) |
| Cl(0D)-Zn(02)-Cl(0Q) | 99.6(4) | Cl(1A)-Cl(0S)-Zn(06) | 70.2(16) |
| Cl(0J)-Zn(02)-Cl(0A) | 114.40(16) | Cl(1A)-Cl(0S)-Zn(1) | 64.2(16) |
| Cl(0J)-Zn(02)-Cl(0D) | 110.81(17) | Zn(06)-Cl(0T)-Zn(1) | 11.1(4) |
| Cl(0J)-Zn(02)-Cl(0Q) | 111.7(5) | Cl(3)-Cl(0T)-Zn(06) | 93.3(16) |
| Cl(0J)-Zn(02)-Cl(2A) | 104.4(5) | Cl(3)-Cl(0T)-Zn(1) | 100.9(17) |
| Cl(2A)-Zn(02)-Cl(0D) | 111.9(4) | Zn(06)-Cl(0U)-Zn(1) | 10.7(4) |
| Cl(2A)-Zn(02)-Cl(0Q) | 12.4(4) | Zn(04)-Cl(1)-Zn(2) | 10.4(2) |
| Cl(0C)-Zn(03)-Cl(0L) | 108.88(16) | Cl(0O)-Cl(1)-Zn(04) | 92(2) |
| Cl(0C)-Zn(03)-Cl(0R) | 107.78(16) | Cl(0O)-Cl(1)-Zn(2) | 101(2) |
| Cl(0K)-Zn(03)-Cl(0C) | 112.61(16) | Cl(08)-Zn(8)-Cl(07) | 108.80(14) |
| Cl(0K)-Zn(03)-Cl(0L) | 110.47(16) | Cl(09)-Zn(8)-Cl(07) | 109.79(15) |
| Cl(0K)-Zn(03)-Cl(0R) | 113.10(18) | Cl(09)-Zn(8)-Cl(08) | 112.30(14) |
| Cl(0L)-Zn(03)-Cl(0R) | 103.54(17) | Cl(09)-Zn(8)-Cl(0B) | 113.59(17) |
| Cl(0F)-Zn(04)-Cl(0G) | 112.9(4) | Cl(0B)-Zn(8)-Cl(07) | 106.53(16) |
| Cl(0F)-Zn(04)-Cl(0I) | 112.6(6) | Cl(0B)-Zn(8)-Cl(08) | 105.53(14) |
| Cl(0F)-Zn(04)-Cl(0O) | 111.1(5) | Cl(0F)-Zn(2)-Cl(0G) | 102.3(5) |
| Cl(0F)-Zn(04)-Cl(2) | 12.2(5) | Cl(0F)-Zn(2)-Cl(0O) | 103.5(7) |
| Cl(0G)-Zn(04)-Cl(0I) | 108.1(2) | Cl(0F)-Zn(2)-Cl(1) | 101.1(8) |
| Cl(0G)-Zn(04)-Cl(0O) | 102.1(5) | Cl(0G)-Zn(2)-Cl(0O) | 89.4(5) |
| Cl(0G)-Zn(04)-Cl(2) | 114.0(7) | Cl(0I)-Zn(2)-Cl(0F) | 128.0(7) |
| Cl(0O)-Zn(04)-Cl(0I) | 109.6(4) | Cl(0I)-Zn(2)-Cl(0G) | 110.6(5) |
| Cl(1)-Zn(04)-Cl(0F) | 104.9(10) | Cl(0I)-Zn(2)-Cl(0O) | 115.6(6) |
| Cl(1)-Zn(04)-Cl(0G) | 118.8(18) | Cl(0I)-Zn(2)-Cl(1) | 107.7(10) |
| Cl(1)-Zn(04)-Cl(0I) | 98.9(11) | Cl(0I)-Zn(2)-Cl(2) | 116.5(9) |
| Cl(1)-Zn(04)-Cl(0O) | 17.1(13) | Cl(1)-Zn(2)-Cl(0G) | 104.8(17) |

| | | | |
|----------------------|-----------|----------------------|-----------|
| Cl(1)-Zn(04)-Cl(2) | 112.8(13) | Cl(1)-Zn(2)-Cl(0O) | 15.4(13) |
| Cl(2)-Zn(04)-Cl(0I) | 101.4(7) | Cl(2)-Zn(2)-Cl(0F) | 12.3(5) |
| Cl(2)-Zn(04)-Cl(0O) | 121.3(8) | Cl(2)-Zn(2)-Cl(0G) | 105.6(8) |
| Cl(0E)-Zn(05)-Cl(0H) | 108.1(6) | Cl(2)-Zn(2)-Cl(0O) | 115.1(9) |
| Cl(0E)-Zn(05)-Cl(0M) | 114.7(7) | Cl(2)-Zn(2)-Cl(1) | 111.0(12) |
| Cl(0M)-Zn(05)-Cl(0H) | 99.2(5) | Cl(0N)-Zn(1)-Cl(0S) | 121.8(7) |
| Cl(0P)-Zn(05)-Cl(0E) | 115.2(6) | Cl(0N)-Zn(1)-Cl(0T) | 108.6(8) |
| Cl(0P)-Zn(05)-Cl(0H) | 105.5(6) | Cl(0N)-Zn(1)-Cl(0U) | 109.8(6) |
| Cl(0P)-Zn(05)-Cl(0M) | 112.3(7) | Cl(0N)-Zn(1)-Cl(1A) | 112.3(7) |
| Cl(0S)-Zn(06)-Cl(0N) | 111.8(5) | Cl(0N)-Zn(1)-Cl(3) | 118.1(7) |
| Cl(0S)-Zn(06)-Cl(0U) | 122.3(6) | Cl(0S)-Zn(1)-Cl(3) | 97.3(5) |
| Cl(0S)-Zn(06)-Cl(3) | 108.8(5) | Cl(0T)-Zn(1)-Cl(0S) | 94.1(6) |
| Cl(0T)-Zn(06)-Cl(0N) | 97.7(6) | Cl(0T)-Zn(1)-Cl(0U) | 111.3(7) |
| Cl(0T)-Zn(06)-Cl(0S) | 102.9(6) | Cl(0T)-Zn(1)-Cl(3) | 15.0(3) |
| Cl(0T)-Zn(06)-Cl(0U) | 120.2(7) | Cl(0U)-Zn(1)-Cl(0S) | 110.2(6) |
| Cl(0T)-Zn(06)-Cl(3) | 16.5(3) | Cl(0U)-Zn(1)-Cl(3) | 96.5(5) |
| Cl(0U)-Zn(06)-Cl(0N) | 99.1(4) | Cl(1A)-Zn(1)-Cl(0S) | 15.1(4) |
| Cl(0U)-Zn(06)-Cl(3) | 105.3(5) | Cl(1A)-Zn(1)-Cl(0T) | 108.9(8) |
| Cl(1A)-Zn(06)-Cl(0N) | 100.8(6) | Cl(1A)-Zn(1)-Cl(0U) | 106.0(7) |
| Cl(1A)-Zn(06)-Cl(0S) | 16.7(4) | Cl(1A)-Zn(1)-Cl(3) | 112.3(7) |
| Cl(1A)-Zn(06)-Cl(0T) | 117.9(7) | Cl(0E)-Zn(0A)-Cl(0P) | 102.5(5) |
| Cl(1A)-Zn(06)-Cl(0U) | 114.5(7) | Cl(0H)-Zn(0A)-Cl(0E) | 118.5(6) |
| Cl(1A)-Zn(06)-Cl(3) | 125.3(7) | Cl(0H)-Zn(0A)-Cl(0M) | 111.3(5) |
| Cl(3)-Zn(06)-Cl(0N) | 108.6(5) | Cl(0H)-Zn(0A)-Cl(0P) | 110.8(5) |
| Zn(05)-Cl(0E)-Zn(0A) | 10.0(4) | Cl(0M)-Zn(0A)-Cl(0E) | 109.2(5) |
| Zn(04)-Cl(0F)-Zn(2) | 10.7(2) | Cl(0M)-Zn(0A)-Cl(0P) | 103.2(5) |
| Zn(04)-Cl(0G)-Zn(2) | 7.5(3) | Zn(06)-Cl(1A)-Zn(1) | 11.3(3) |
| Zn(0A)-Cl(0H)-Zn(05) | 3.2(7) | Cl(0S)-Cl(1A)-Zn(06) | 93.1(17) |
| Zn(2)-Cl(0I)-Zn(04) | 5.2(4) | Cl(0S)-Cl(1A)-Zn(1) | 100.8(17) |
| Zn(05)-Cl(0M)-Zn(0A) | 10.3(3) | Zn(2)-Cl(2)-Zn(04) | 10.7(2) |
| Zn(1)-Cl(0N)-Zn(06) | 0.2(8) | Zn(06)-Cl(3)-Zn(1) | 9.7(4) |
| Zn(04)-Cl(0O)-Zn(2) | 8.8(2) | Cl(0T)-Cl(3)-Zn(06) | 70.2(15) |
| Cl(1)-Cl(0O)-Zn(04) | 71.2(14) | Cl(0T)-Cl(3)-Zn(1) | 64.1(15) |

Table S6. The manganese content for the metal components in the raw material and

that obtained from EDS experiment.

| Raw Material (Mn%) | | 1 | 5 | 10 | 20 | 30 | 40 | 50 |
|---------------------|---------|------|------|------|-----|------|-----|------|
| Experiment (Mn%) | first | 4.7 | 9.2 | 3.1 | 3.6 | 0.21 | 7 | 11.3 |
| | second | 0 | 1.2 | 0 | 11 | 14.6 | 4.1 | 6.1 |
| | third | 0.39 | 0 | 0 | 4.9 | 0 | 8.1 | 14.4 |
| | average | 1.69 | 3.47 | 1.03 | 6.5 | 4.94 | 6.4 | 10.6 |

Table S7. CIE coordinates of $(\text{C}_9\text{H}_{15}\text{N}_3)\text{ZnCl}_4:\text{Mn}^{2+}$ with different Mn^{2+} feed ratio (0~50%).

| Mn^{2+} (%) | 0 | 1 | 5 | 10 | 20 | 30 | 40 | 50 |
|-------------------------|-------------------------|-------------------------|-------------------------|-------------------------|-------------------------|------------------------|-------------------------|-------------------------|
| CIE | (0.162 7,0.09 72) | (0.173 6,0.19 86) | (0.192 4,0.36 86) | (0.197 9,0.40 98) | (0.206 7,0.48 44) | (0.211 9,0.52 6) | (0.215 6,0.55 64) | (0.218 7,0.56 47) |

Table S8. CIE coordinates of $(\text{C}_9\text{H}_{15}\text{N}_3)\text{ZnCl}_4:5\%\text{Mn}^{2+}$ in the temperature range of 110 to 290 K

| Temperature (K) | 110 K | 140 K | 170 K | 200 K | 230 K | 260 K | 290 K |
|--------------------|---------------------|---------------------|---------------------|--------------------|---------------------|---------------------|---------------------|
| CIE | (0.1885, 0.3067) | (0.1897, 0.3157) | (0.1921, 0.3257) | (0.1967, 0.348) | (0.1999, 0.3761) | (0.2028, 0.4052) | (0.2053, 0.4208) |

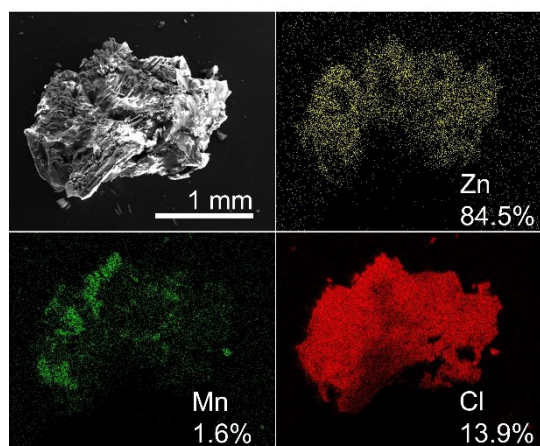


Figure S1. SEM images of $(\text{C}_9\text{H}_{15}\text{N}_3)_2\text{ZnCl}_4: \text{Mn}^{2+}$ (50%) and element mapping images of Zn, Mn, and Cl.

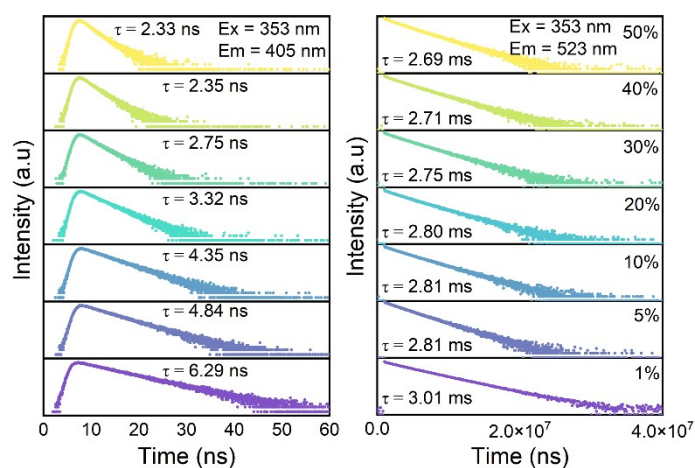


Figure S2. The fluorescence decay lifetime of $(\text{C}_9\text{H}_{15}\text{N}_3)_2\text{ZnCl}_4: \text{Mn}^{2+}$ with different Mn^{2+} feed ratio (1% \sim 50%) recorded at 405 and 523 nm at room temperature.

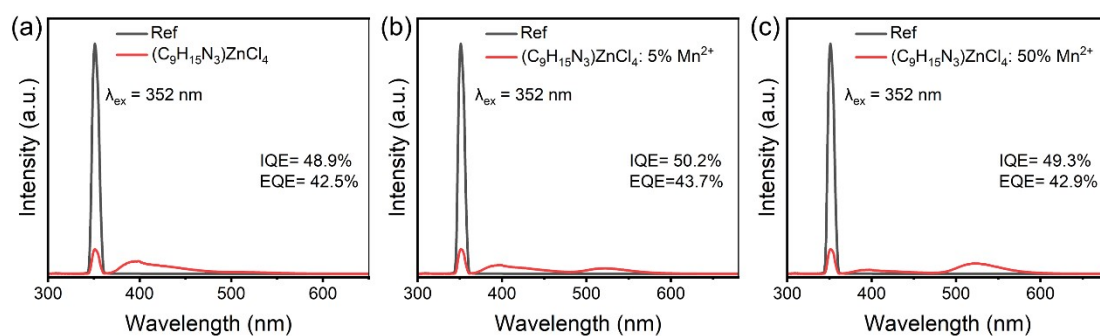


Figure S3. The PL spectra of (a) $(\text{C}_9\text{H}_{15}\text{N}_3)_2\text{ZnCl}_4$, (b) $(\text{C}_9\text{H}_{15}\text{N}_3)_2\text{ZnCl}_4: 5\%\text{Mn}^{2+}$ and (c) $(\text{C}_9\text{H}_{15}\text{N}_3)_2\text{ZnCl}_4: 50\%\text{Mn}^{2+}$ with Al_2O_3 as reference.

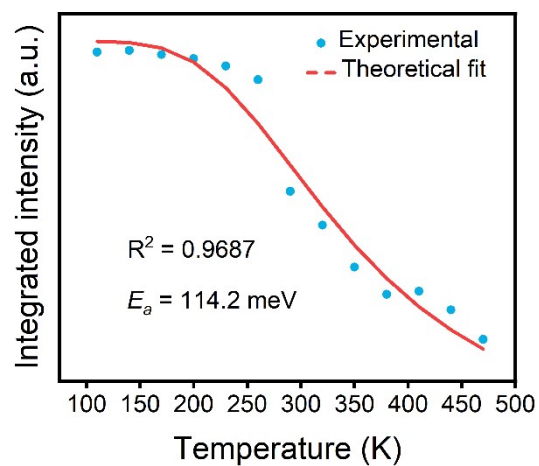


Figure S4. The integrated PL intensity of $(\text{C}_9\text{H}_{15}\text{N}_3)\text{ZnCl}_4$ in the range of 110 ~ 470 K as a function of temperature.