## **Supporting Information**

A solid-state Matryoshka doll-like microwave method for one-step rapid synthesis of composites of NiSe<sub>2</sub> and nitrogen-doped porous carbon for sodium

storage

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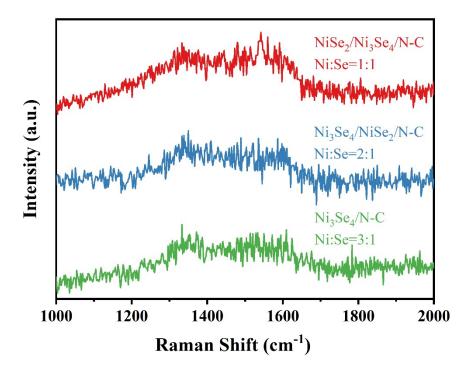
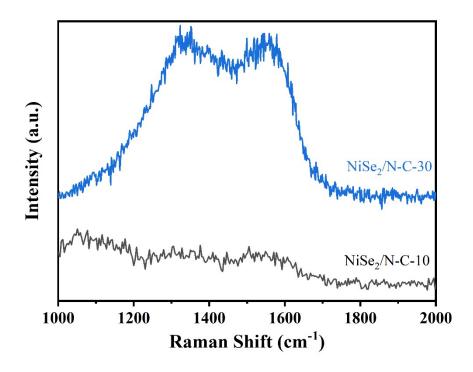
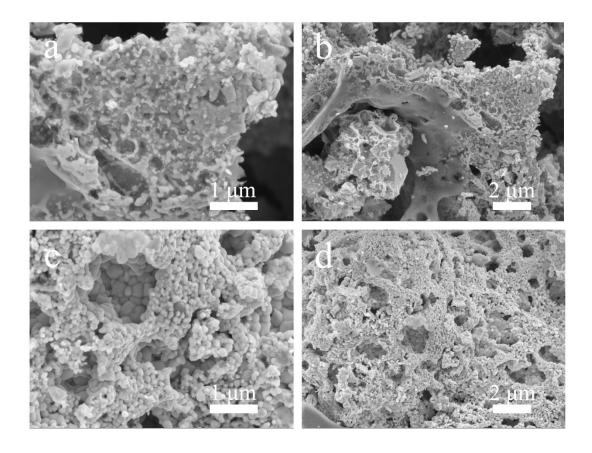


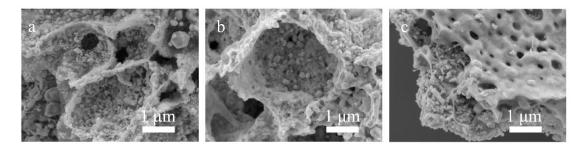
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**Figure S2**. Raman patterns of samples with different microwave time: NiSe<sub>2</sub>/N-C-10 and NiSe<sub>2</sub>/N-C-30.



**Figure S3**. SEM images of samples with different time, a-b) NiSe<sub>2</sub>/N-C-10 and c-d) NiSe<sub>2</sub>/N-C-30.



**Figure S4**. SEM images of samples with different mass ratios, a) NiSe<sub>2</sub>/Ni<sub>3</sub>Se<sub>4</sub>/N-C, b) Ni<sub>3</sub>Se<sub>4</sub>/NiSe<sub>2</sub>/N-C and c) Ni<sub>3</sub>Se<sub>4</sub>/N-C.

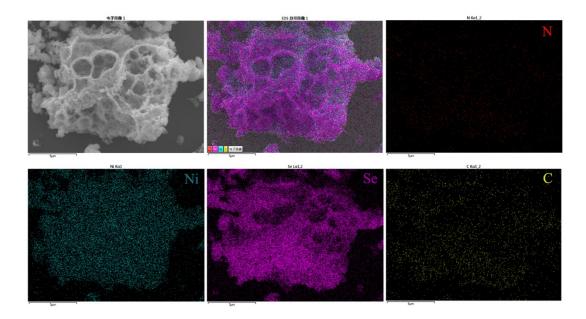


Figure S5. Elemental mapping images of NiSe<sub>2</sub>/N-C.

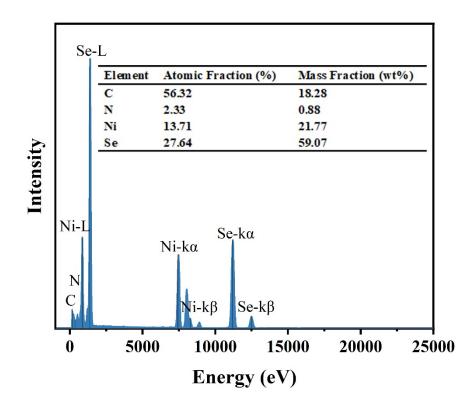


Figure S6. EDS of NiSe<sub>2</sub>/N-C.

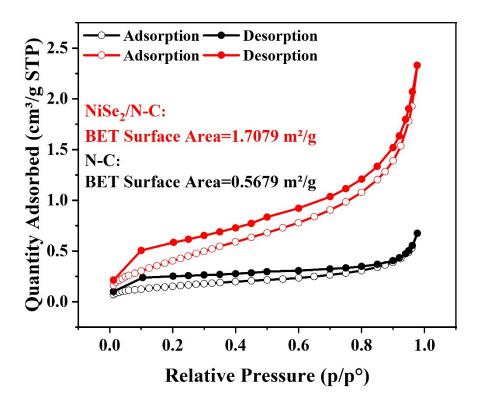


Figure S7. The  $N_2$  adsorption-desorption isotherms.

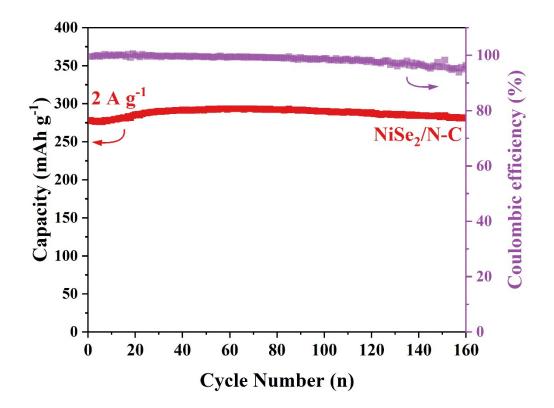
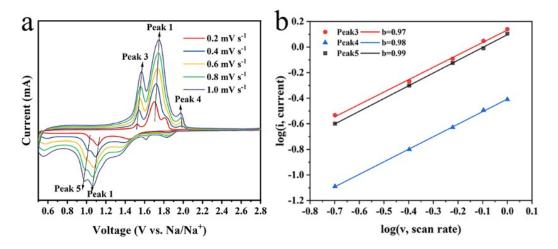


Figure S8. the cycling performance of  $NiSe_2/N-C$  at 2 A  $g^{-1}$ .



**Figure S9**. a) CV curves at different scan rates. b) The linear relationship of log(v) and log(i).

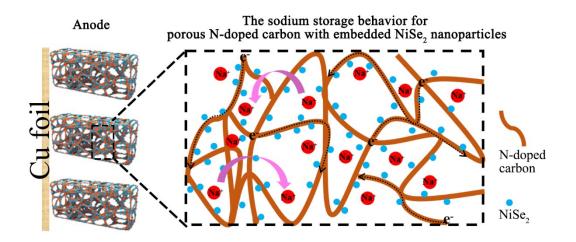


Figure S10. The mechanism of sodium storage behavior.

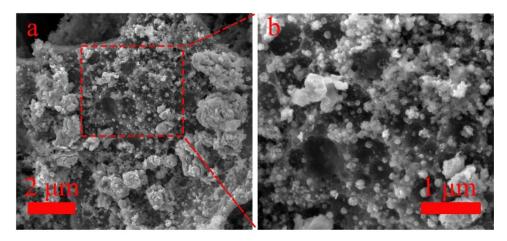


Figure S11. SEM images of the cycled NiSe<sub>2</sub>/N-C electrode.

Sample	$R_{s}\left(\Omega ight)$	$R_1(\Omega)$	$W\left(\Omega ight)$	$R_{2}(\Omega)$
1st cycle	9.481	12.74	0.459	-
25th cycles	8.46	12.79	0.440	-
200th cycles	10.09	18.09	0.378	-
Charge 0.8V	8.499	14.75	0.164	-
Charge 2V	7.689	33.56	0.472	-
Discharge 0.8V	9.471	15.81	0.472	8.884
Discharge 2V	8.93	18.69	0.428	11.66

Table S1.  $R_{s}\left(\Omega\right),R_{1}\left(\Omega\right),W\left(\Omega\right)$  and  $R_{1}\left(\Omega\right)$  of NiSe\_/N-C electrode.

Species	$E_{Na}/eV$	$E_{anode}/eV$	$E_{sodiated anode}/eV$	$E_{adsorption \ enery}/eV$
NiSe <sub>2</sub>	-1299.856444	-164488.3032	-165789.0962	-0.936520295

 Table S2. DFT calculation results of adsorption energies.