Supporting Information

A Serial of Ultrasensitive Electrocatalysts Fe-MOF/MWCNTs for

Fentanyl Determination

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Synthesis of the five Fe-MOFs

Preparation of Fe-MOF-235 : H₂BDC (1.23 mmol) was added to 60 mL DMF and stirred for 10 min to form a clear solution. Then $FeCl₃·6H₂O$ (0.738 mmol) was added to the solution and stirred for 5 min. The reactant mixture of 30 mL and 30 mL of ethanol were loaded into a Teflon-lined autoclave, heated at 80 °C for 24 h. After cooling to room temperature, the product was collected by centrifugation, washed several times each with ethanol and DMF, and dried at 60 °C for 12 h in a vacuum.[1]

Preparation of Fe-MIL-88B : FeCl₃·6H₂O (1 mmol) and H₂BDC (1 mmol) were dissolved in a mixed solution of DMF (5 mL) and NaOH (2 M, 0.4 mL). After stirring for 30 minutes, the mixture was transferred to a Teflon-lined stainless steel autoclave, heated at 100 ℃ for 12 h. After cooling to room temperature, the product was collected by centrifugation, washed many times each with ethanol and DMF, and dried at 60 °C for 12 h in a vacuum.^[2]

Preparation of Fe-MIL-53 : FeCl₃·6H₂O (5 mmol) and H₂BDC (5 mmol) were dissolved in DMF (25 mL) and stirred for 30 min. The mixture was transferred to a Teflon-lined stainless steel autoclave, heated at 150 ℃ for 15 h. After cooling to room temperature, the product was collected by centrifugation, washed several times each with ethanol and DMF, and dried at 60 ℃ for 12 h in a vacuum.[3]

Preparation of Fe-MIL-68 : FeCl₃·6H₂O (1 mmol) and H₂BDC (2 mmol) were dissolved into DMF (12 mL). Then HF (5 M, 120 μL) and HCl (1 M, 120 μL) were added to the above solution. After stirring for 30 minutes, the mixture was transferred to a Teflon-lined stainless steel autoclave, heated at 100 ℃ for 120 h. After cooling to room temperature, the product was

collected by centrifugation, washed many times each with deionized water and acetone, and dried at 60 °C for 12 h in a vacuum.^[4]

Preparation of Fe-MIL-101 : FeCl₃·6H₂O (2.45 mmol) and H₂BDC (1.24 mmol) were added to DMF (15 mL) and stirred for 30 min. The mixture was transferred to a Teflon-lined stainless steel autoclave, heated at 110 ℃ for 24 h. After cooling to room temperature, the product was collected by centrifugation, washed several times each with ethanol and DMF, and dried at 60 ℃ for 12 h in a vacuum.[5]

Fig.S1 TEM images and EDS spectrum of prepared Fe-MOF-235.

Fig.S2 TEM images and EDS spectrum of prepared Fe-MIL-88B.

Fig.S3 TEM images and EDS spectrum of prepared Fe-MIL-53.

Fig.S4 TEM images and EDS spectrum of prepared Fe-MIL-68.

Fig.S5 TEM images and EDS spectrum of prepared Fe-MIL-101.

Fig.S6 CV curves of different scan rate values in the range of 10–600 mV·s−1(a) and the Linear relationship of Ipa and \sqrt{v} of bare CGE.

Fig.S7 CV curves of different scan rate values in the range of 10–600 mV·s−1(a) and the Linear

relationship of Ipa and \sqrt{v} of MWCNTs/CGE.

Fig.S8 CV curves of different scan rate values in the range of 10–600 mV·s−1(a) and the Linear relationship of Ipa and $\sqrt{\nu}$ of Fe-MOF235/MWCNTs/CGE.

Fig.S9 CV curves of different scan rate values in the range of 10–600 mV·s−1(a) and the Linear relationship of Ipa and $\sqrt{\nu}$ of Fe-MIL-88B/MWCNTs/CGE.

Fig.S10 CV curves of different scan rate values in the range of 10–600 mV·s−1(a) and the Linear relationship of Ipa and \sqrt{v} of Fe-MIL-53/MWCNTs/CGE.

Fig.S11 CV curves of different scan rate values in the range of 10–600 mV·s−1(a) and the Linear relationship of Ipa and \sqrt{v} of Fe-MIL-68/MWCNTs/CGE.

Fig.S12 CV curves of different scan rate values in the range of 10–600 mV·s−1(a) and the Linear relationship of Ipa and $\sqrt{\nu}$ of Fe-MIL-101/MWCNTs/CGE.

Fig.S13 The Linear relationship of Log Ipa and Log v of, Fe-MOF-235/MWCNTs(a) ,Fe-MIL-88B/MWCNTs(b) ,Fe-MIL-53/MWCNTs(c) ,Fe-MIL-68/MWCNTs(d) ,Fe-MIL-101/MWCNTs(e).

Fig.S14 CV curves of fentanyl recorded by Fe-MOF-235/MWCNTs(a) ,Fe-MIL-88B/MWCNTs(b) ,Fe-MIL-53/MWCNTs(c) ,Fe-MIL-68/MWCNTs(d) ,Fe-MIL-101/MWCNTs(e) repeated for 10 times using a single modified GCE .

Reference

[1] Wang, Q.; Astruc, D., State of the Art and Prospects in Metal-Organic Framework (MOF)- Based and MOF-Derived Nanocatalysis. Chemical Reviews 2020, 120 (2), 1438-1511.

[2] Yu, X. Y.; Feng, Y.; Guan, B. Y.; Lou, X. W.; Paik, U., Carbon coated porous nickel phosphides nanoplates for highly efficient oxygen evolution reaction. Energy & Environmental Science 2016, 9 (4), 1246-1250.

[3] Yan, L. T.; Cao, L.; Dai, P. C.; Gu, X.; Liu, D. D.; Li, L. J.; Wang, Y.; Zhao, X. B., Metal-Organic Frameworks Derived Nanotube of Nickel-Cobalt Bimetal Phosphides as Highly Efficient Electrocatalysts for Overall Water Splitting. Advanced Functional Materials 2017, 27 (40).

[4] Zhao, S. L.; Wang, Y.; Dong, J. C.; He, C. T.; Yin, H. J.; An, P. F.; Zhao, K.; Zhang, X. F.; Gao, C.; Zhang, L. J.; Lv, J. W.; Wang, J. X.; Zhang, J. Q.; Khattak, A. M.; Khan, N. A.; Wei, Z. X.; Zhang, J.; Liu, S. Q.; Zhao, H. J.; Tang, Z. Y., Ultrathin metal-organic framework nanosheets for electrocatalytic oxygen evolution. Nature Energy 2016, 1, 1-10.

[5] Jahan, M.; Liu, Z. L.; Loh, K. P., A Graphene Oxide and Copper-Centered Metal Organic Framework Composite as a Tri-Functional Catalyst for HER, OER, and ORR. Advanced Functional Materials 2013, 23 (43), 5363-5372.