

## Supplementary Material for

Portable smartphone platform based on  $\text{Ti}_3\text{C}_2$  MQDs/CDs assembly for  
ratiometric fluorescence quantitative monitoring of crystal violet

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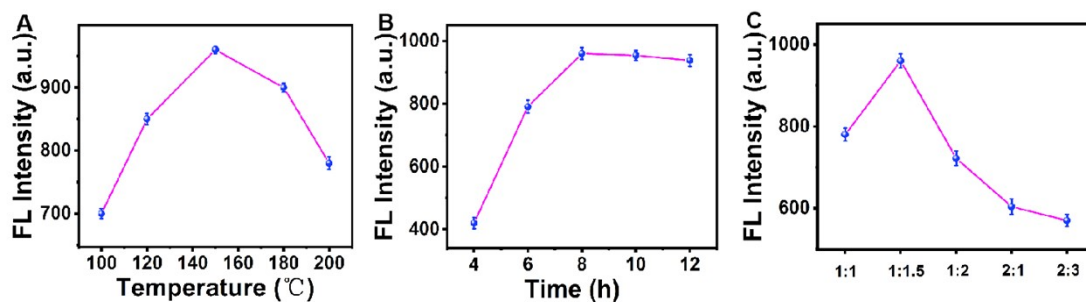


Fig. S1. Optimization of CDs synthesis conditions: (A) reaction temperature, (B) reaction time and (C) ratio of substances of 4-chloro-1,2-diaminobenzene to 3-hydroxytyramine hydrochloride.

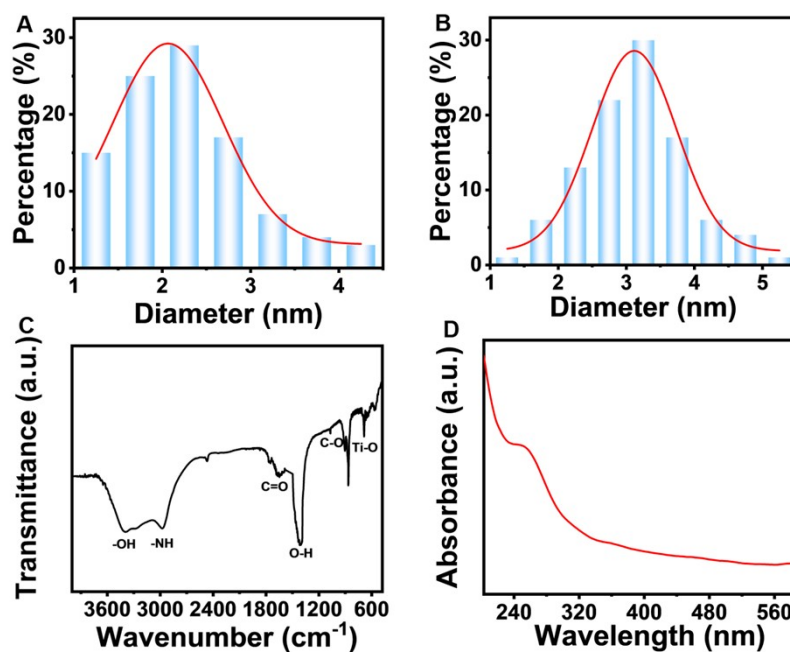


Fig. S2. The size distribution of (A) Ti<sub>3</sub>C<sub>2</sub> MQDs and (B) CDs, (C) FT-IR spectrum and (D) UV-vis absorption spectra of Ti<sub>3</sub>C<sub>2</sub> MQDs.

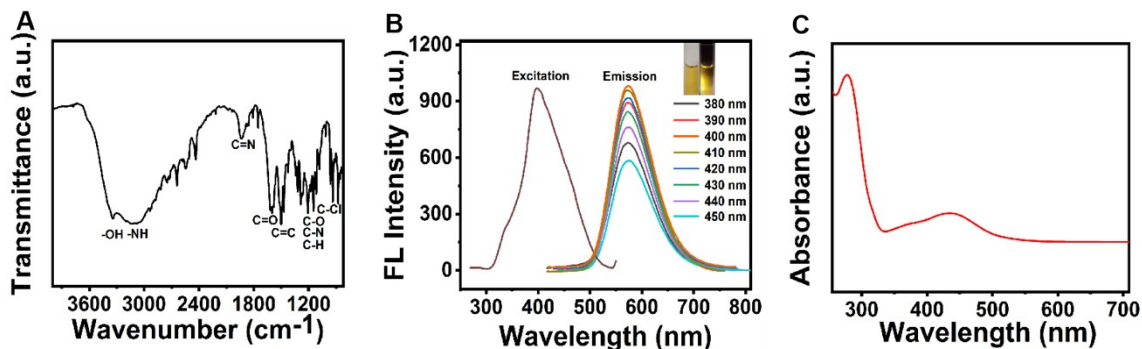


Fig. S3. (A) FT-IR spectrum, (B) fluorescence spectra and (C) UV-vis absorption spectra of CDs.

### 1. The narrow-scan spectra of C, N, Ti in $Ti_3C_2$ MQDs

Specifically, the XPS spectra of C 1s (Fig. S3A) display the peaks for C=C /C-C (284.8 eV), C-O (286.33 eV), C=O (287.05 eV) and N-C=N (288.9 eV). As for the N 1s spectrum (Fig. S3B), the peaks centered at 398.89, 399.63 and 401.56 eV belong to C-N-C, N-(C)<sub>3</sub> and N-H, respectively. As shown in Fig. S3C, the peaks at 458.1, 461.43, 464.94 eV and 468.29 eV correspond to Ti-C 2p<sub>3/2</sub>, Ti-O 2p<sub>3/2</sub>, Ti-O 2p<sub>1/2</sub> and Ti-C 2p<sub>1/2</sub>, respectively, indicating that the synthesized  $Ti_3C_2$  MQDs possess typical characteristic peaks.

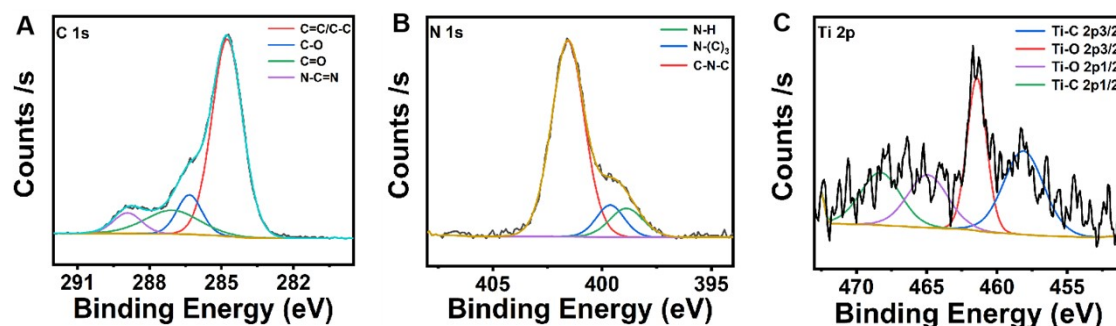


Fig. S4. The high-resolution XPS spectra of (A) C 1s, (B) N 1s, (C) Ti 2p of the  $Ti_3C_2$  MQDs.

### 2. The narrow-scan spectra of C, N, Cl in CDs

As for CDs, further analysis by high-resolution XPS shows that C 1s contains three fitting peaks of 284.8 eV, 285.3 eV and 286.6 eV, which belong to C=C, C-C and C-O, respectively. The peaks of N 1s at 398.8 eV, 399.52 eV and 402.1 eV correspond

to N-C N-(C)<sub>3</sub> and N-H, respectively. The Cl 2p XPS spectrum contains three peaks, which are 196.2 eV (Cl 2p<sub>3/2</sub>) 199.4 eV (Cl 2p) and 201.41 eV (Cl 2p<sub>1/2</sub>) respectively.

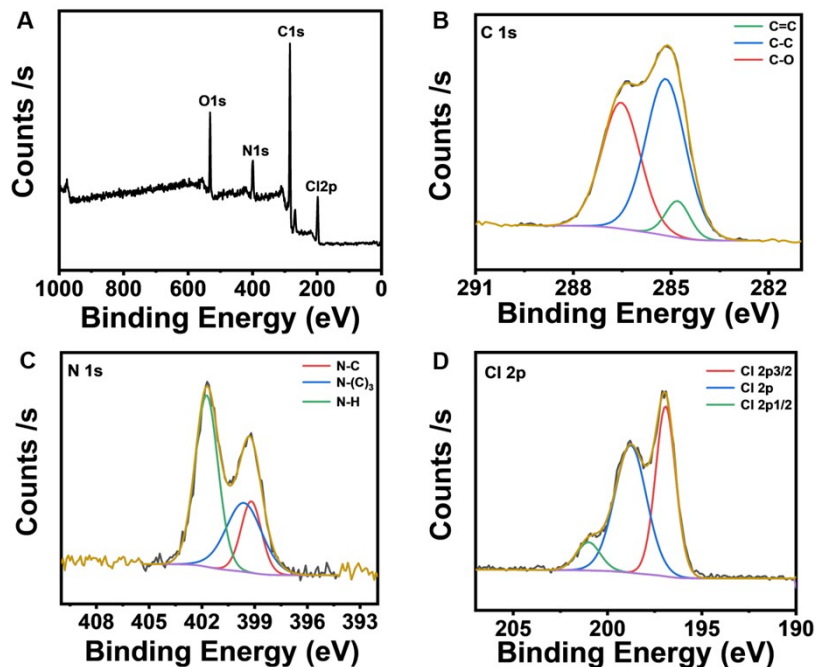


Fig. S5. (A) XPS full spectrum of CDs. The high-resolution XPS spectra of (B) C 1s, (C) N 1s, (D) Cl 2p of the CDs.

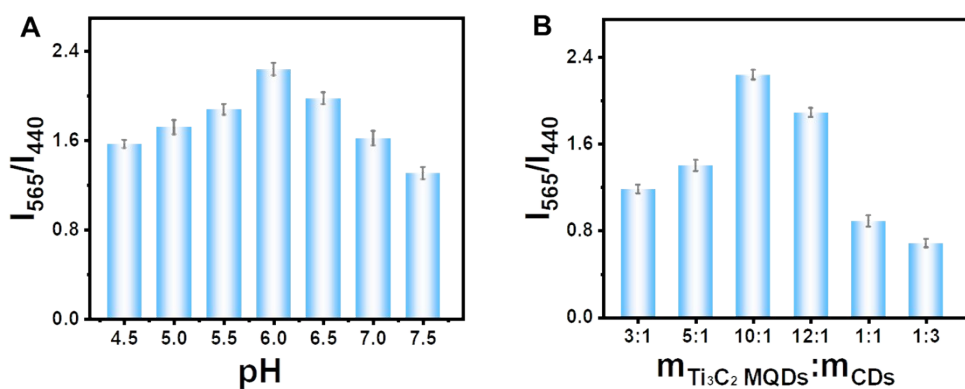


Fig. S6. The effects of (A) pH and (B) coupling mass ratio of Ti<sub>3</sub>C<sub>2</sub> MQDs/CDs.

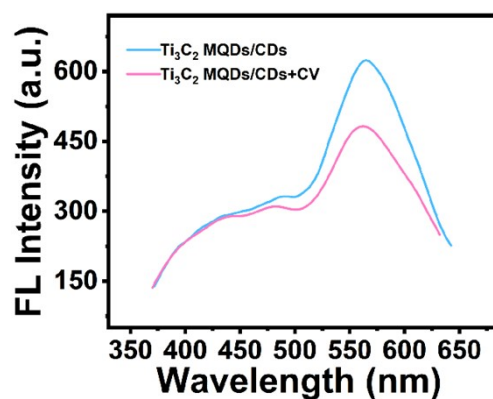


Fig. S7. Feasibility of CV detection with  $\text{Ti}_3\text{C}_2$  MQDs/CDs probe

**Table S1** Fitting fluorescence lifetime parameters for  $\text{Ti}_3\text{C}_2$  MQDs/CDs

Sample	Nanomaterial	Fluorescence lifetime $\tau$ (ns)
1	$\text{Ti}_3\text{C}_2$ MQDs (alone)	4.68
2	$\text{Ti}_3\text{C}_2$ MQDs ( $\text{Ti}_3\text{C}_2$ MQDs/CDs)	3.37
3	CDs (alone)	2.25
4	$\text{Ti}_3\text{C}_2$ MQDs ( $\text{Ti}_3\text{C}_2$ MQDs/CDs)	3.48

**Table S2** Comparison of different fluorescence detection methods for CV

Methods	Linear range	Detection limit	Ref.
Fluorescence	0.02-1.6 $\mu\text{M}$	7.3 nM	[1]
Fluorescence	0.3-12 $\mu\text{M}$	83 nM	[2]
Fluorescence	0.182-37 $\mu\text{M}$	61 nM	[3]
Fluorescence	0.1-11 $\mu\text{M}$	20 nM	[4]
Fluorescence	0.02-2.0 $\mu\text{M}$	8.6 nM	This work

## References

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