

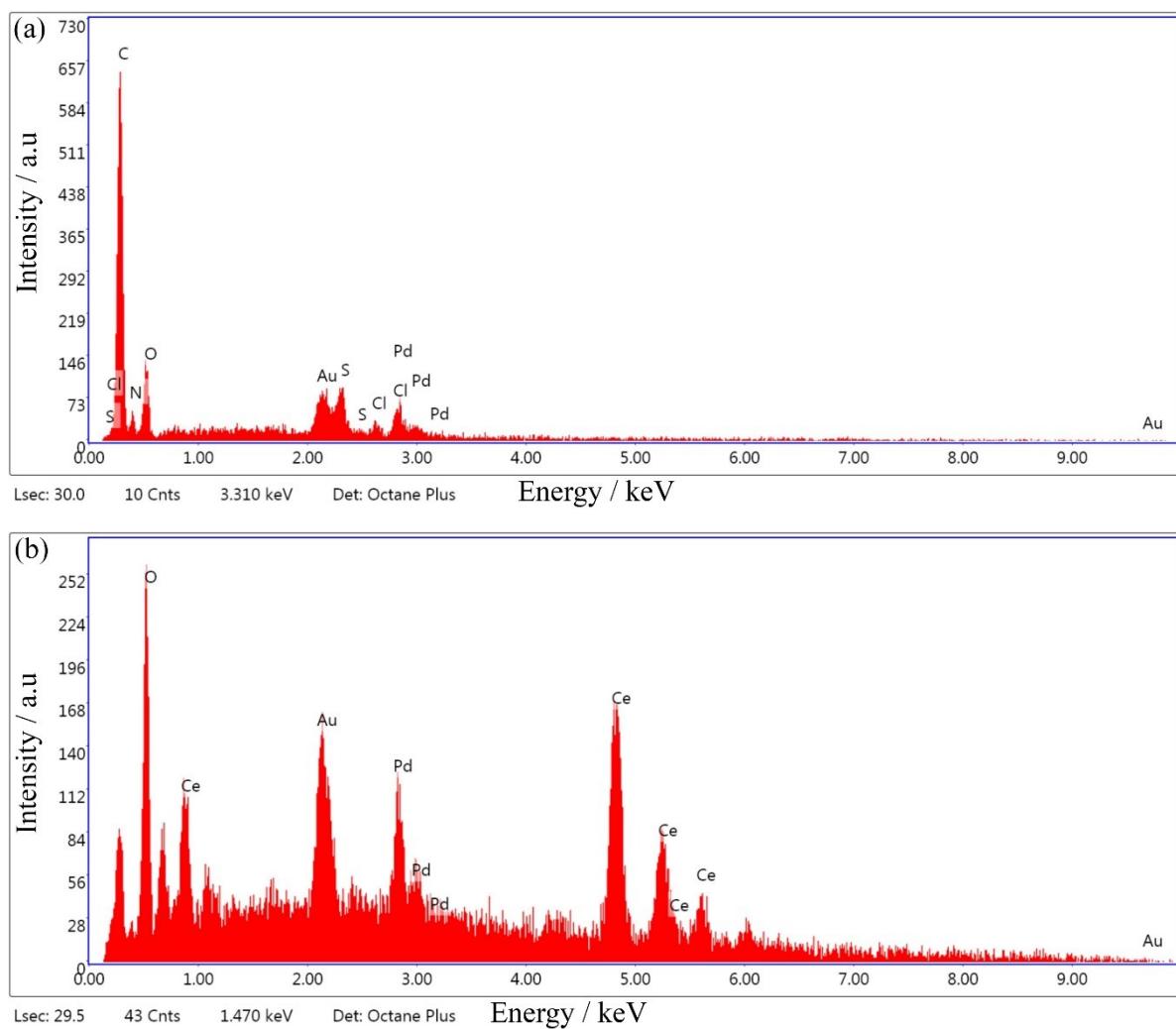
Development of a facile electrochemical sensor based on GCE modified with one-step prepared PNMA-CeO₂-fMWCNTs composite for simultaneous detection of UA and 5-FU

Kübra TURAN¹, Ahmet ÜĞE², Bülent ZEYBEK², Gözde AYDOĞDU TİĞ^{1*}

¹ Ankara University, Faculty of Science, Department of Chemistry, Ankara, 06100, Türkiye

² Kütahya Dumlupınar University, Faculty of Science and Arts, Department of Chemistry, Kütahya, 43100, Türkiye

Supplementary Material



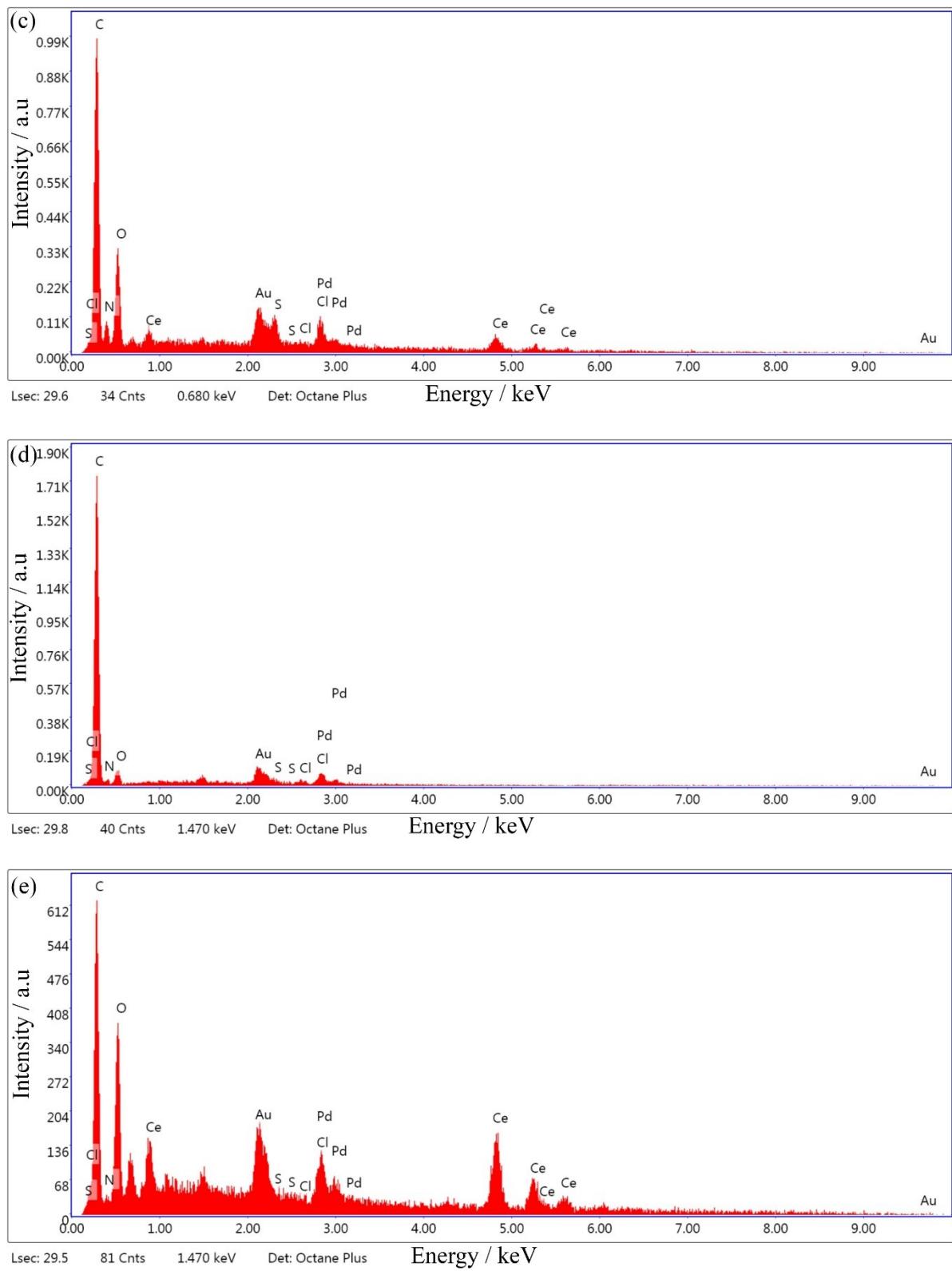


Fig. S1 EDX analysis of (a) PNMA, (b) CeO₂ NPs, (c) PNMA-CeO₂, (d) PNMA-*f*MWCNT, and (e) PNMA-CeO₂-*f*MWCNT composites.

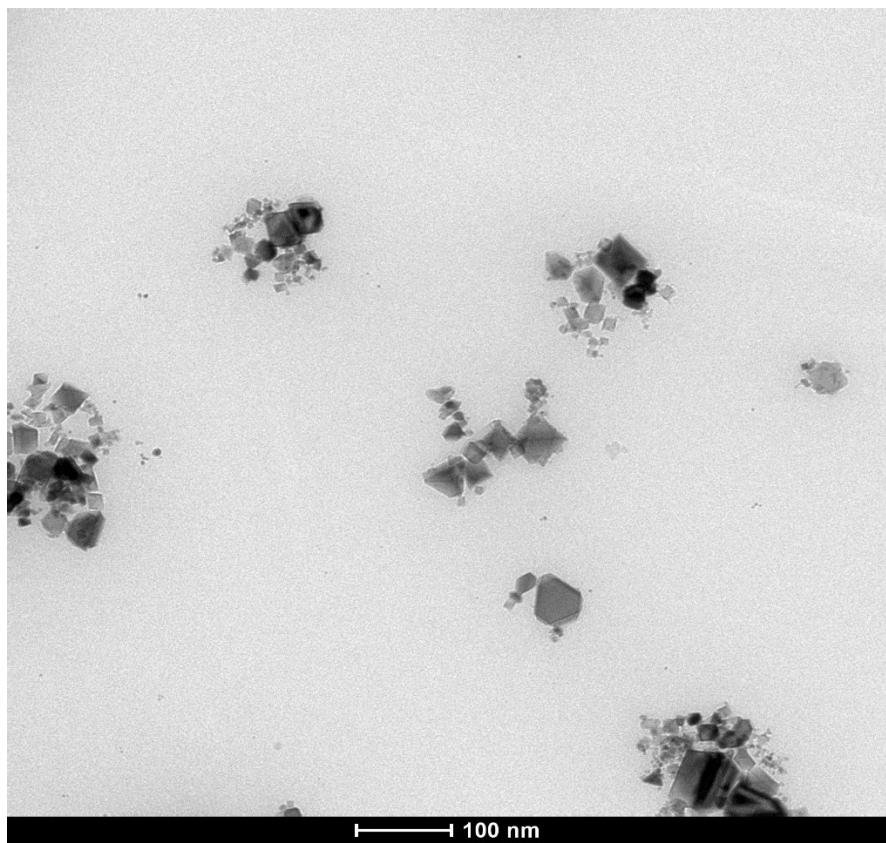


Fig. S2 TEM micrograph of CeO₂ NPs.

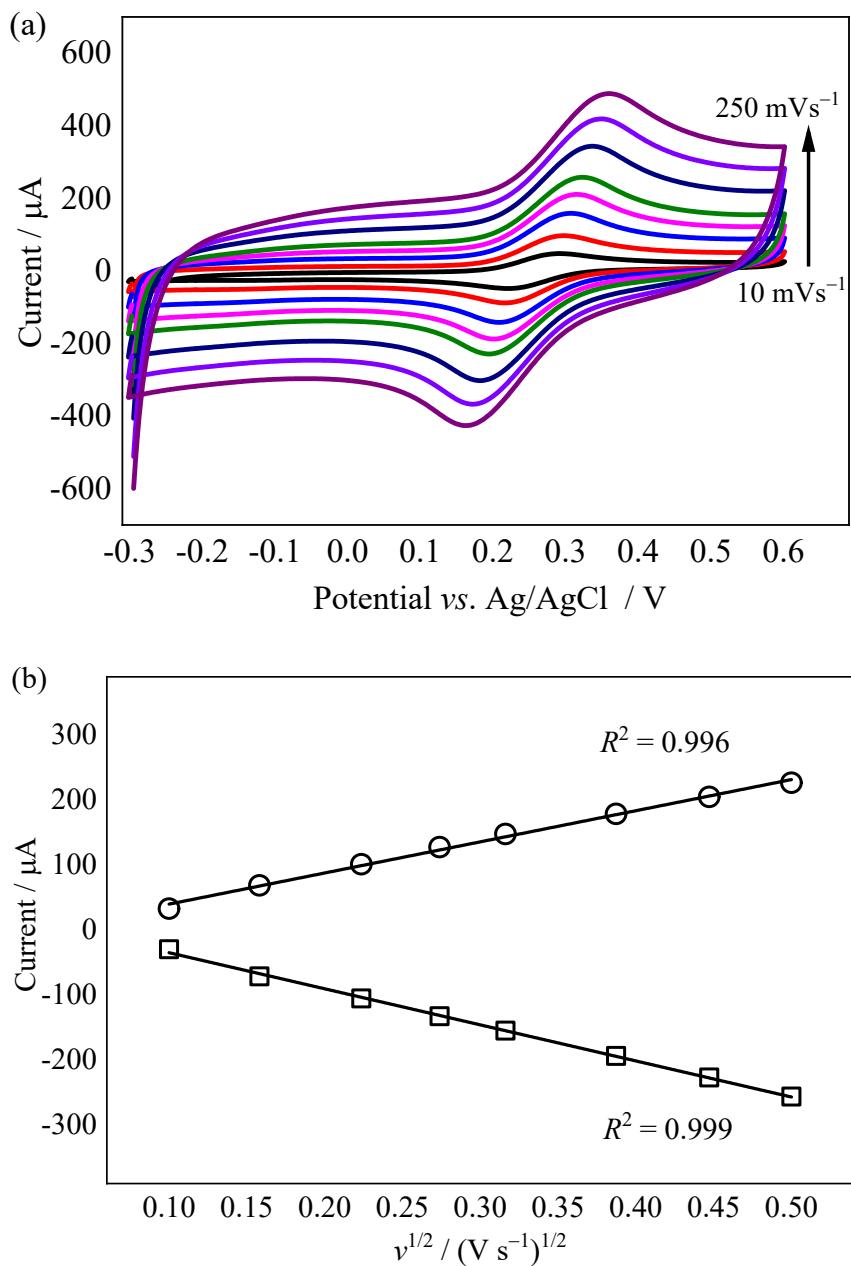
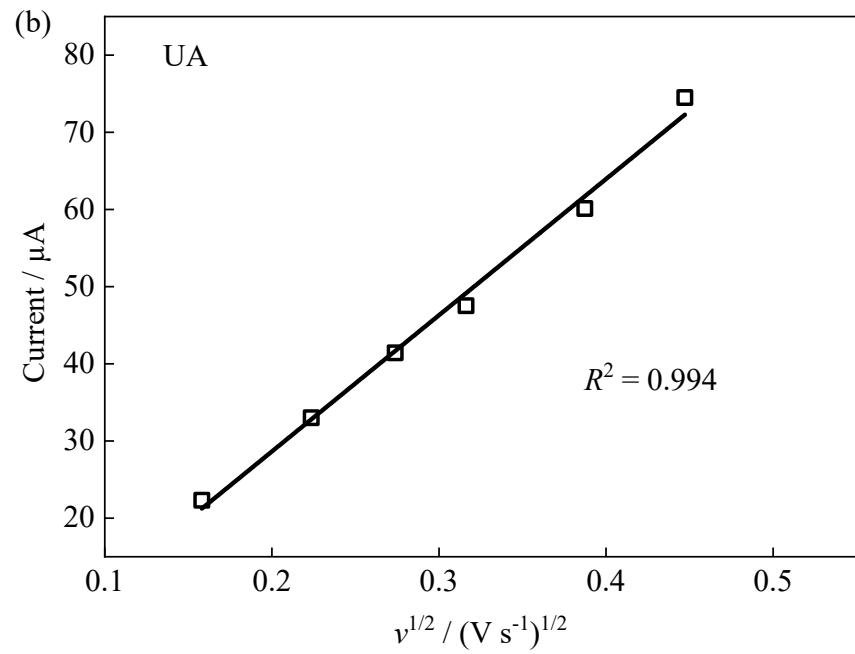
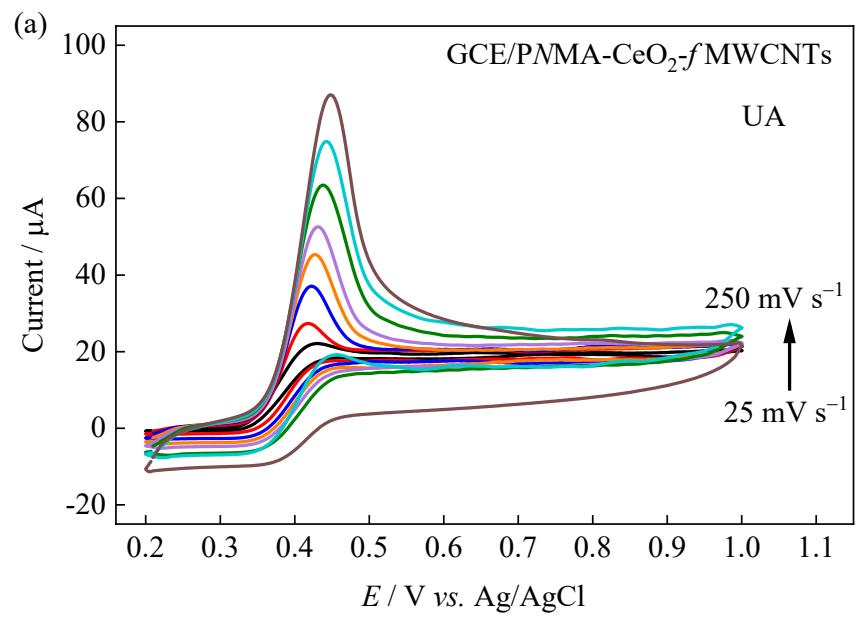


Fig. S3 **(a)** CVs recorded at different scan rates, **(b)** the plot of the peak current against the square root of the scan rate at the GCE/PNMA-CeO₂-fMWCNTs electrode in 0.1 mol L⁻¹ KCl solution containing 5.0 mmol L⁻¹ Fe(CN)₆^{3-/4-}.



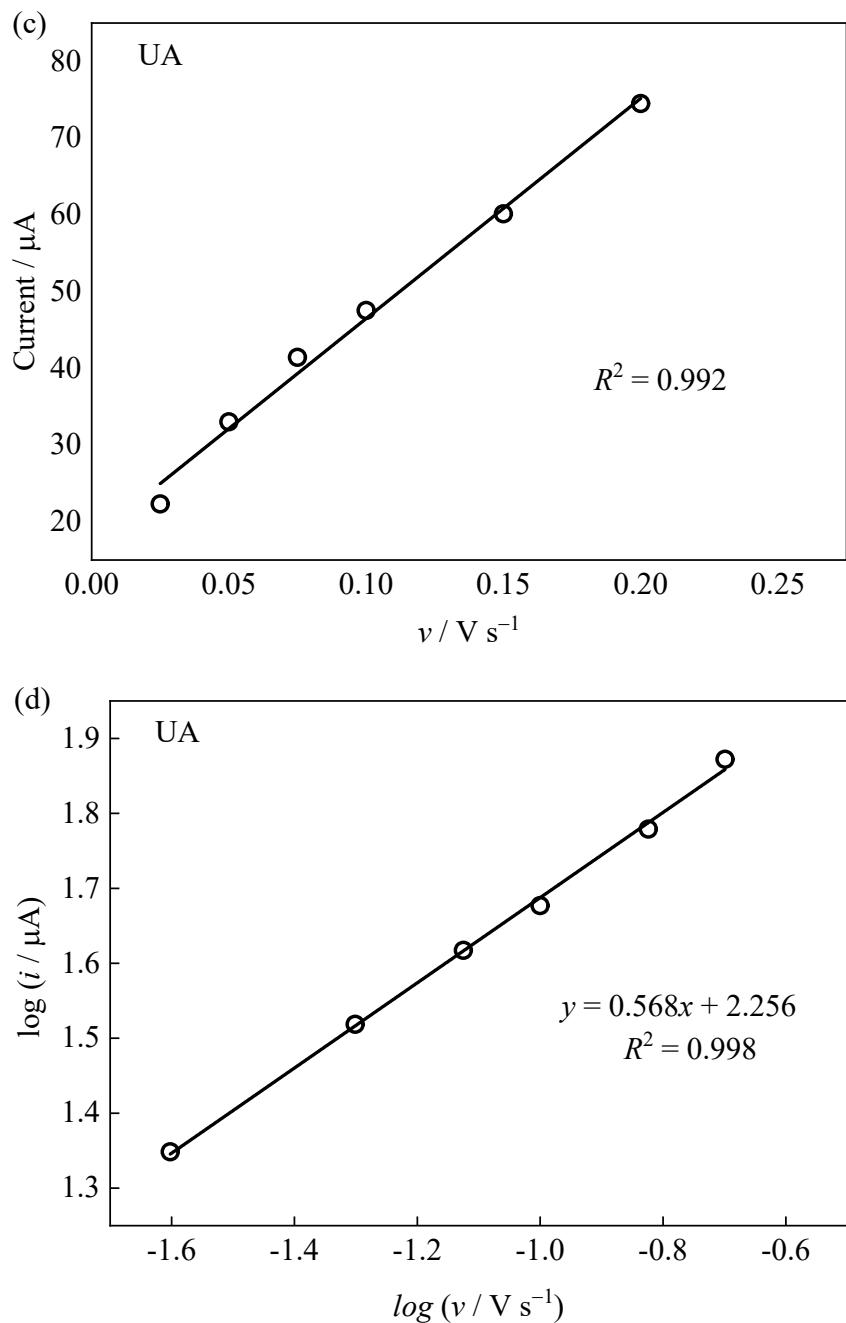
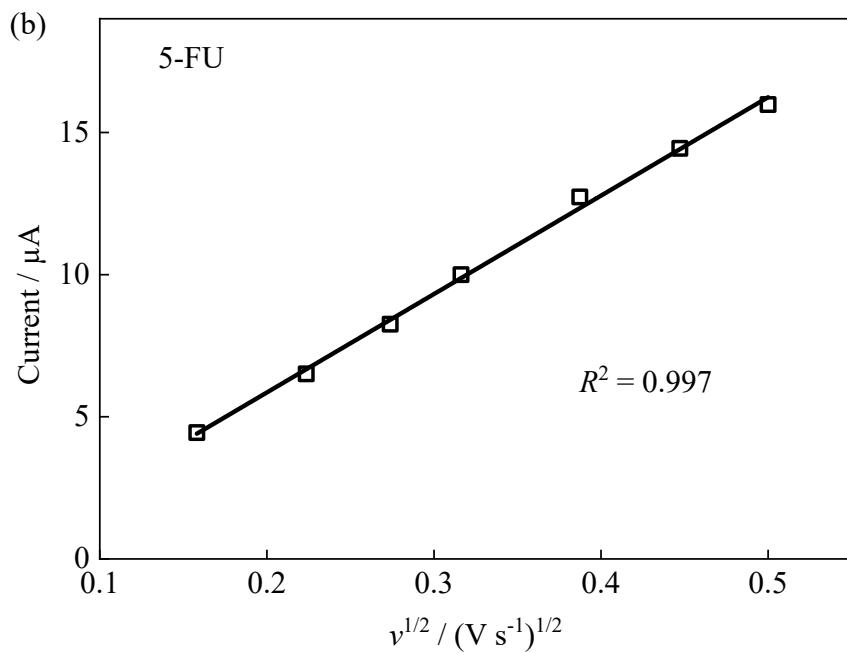
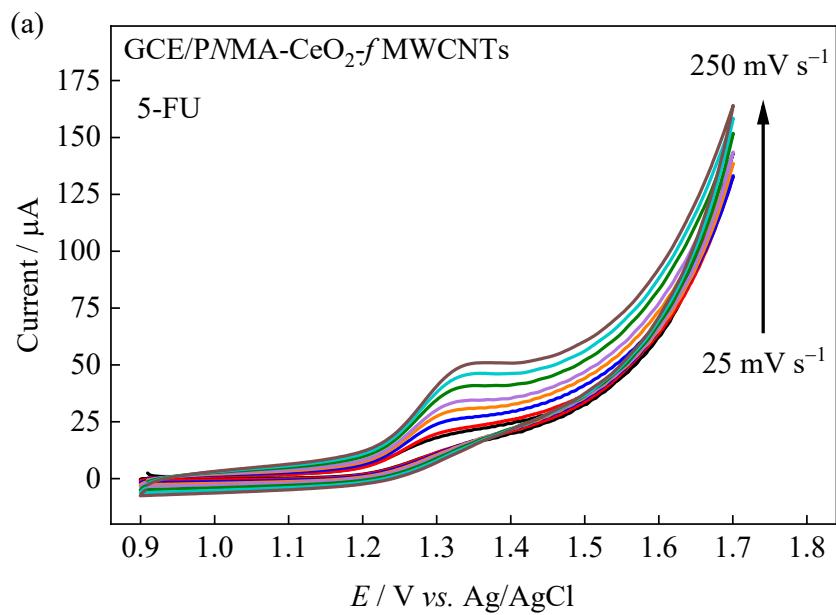


Fig. S4 CVs of (a) GCE/PNMA-CeO₂-*f*MWCNTs electrode at the scanning speeds of 0.025–0.250 V s⁻¹ in 0.1 mol L⁻¹ PBS with 0.3 mmol L⁻¹ UA. The plots of the (b) Current vs. $v^{1/2}$, (c) Current vs. v , and (d) $\log i_p$ vs. $\log v$.



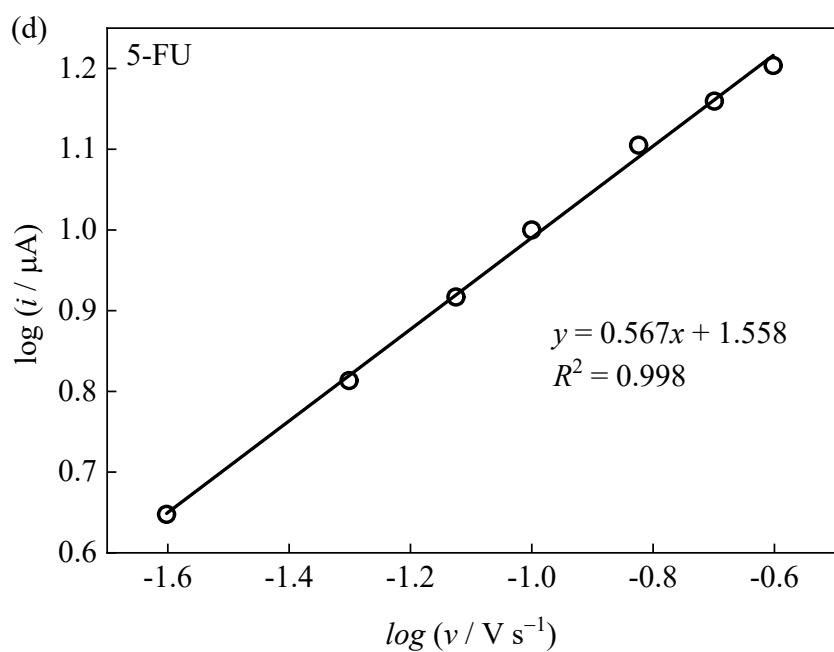
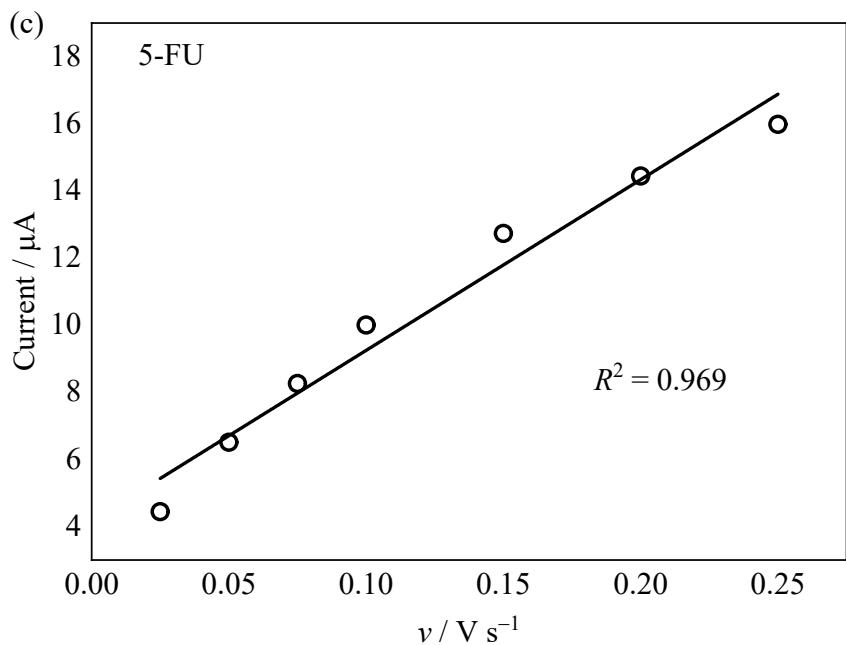


Fig. S5 CVs of (a) GCE/PNMA-CeO₂-fMWCNTs electrode at the scanning speeds of 0.025–0.250 V s⁻¹ in 0.1 mol L⁻¹ PBS with 1.2 mmol L⁻¹ 5-FU. The plots of the (b) Current vs. $\nu^{1/2}$, (c) Current vs. ν , and (d) $\log i_p$ vs. $\log \nu$.

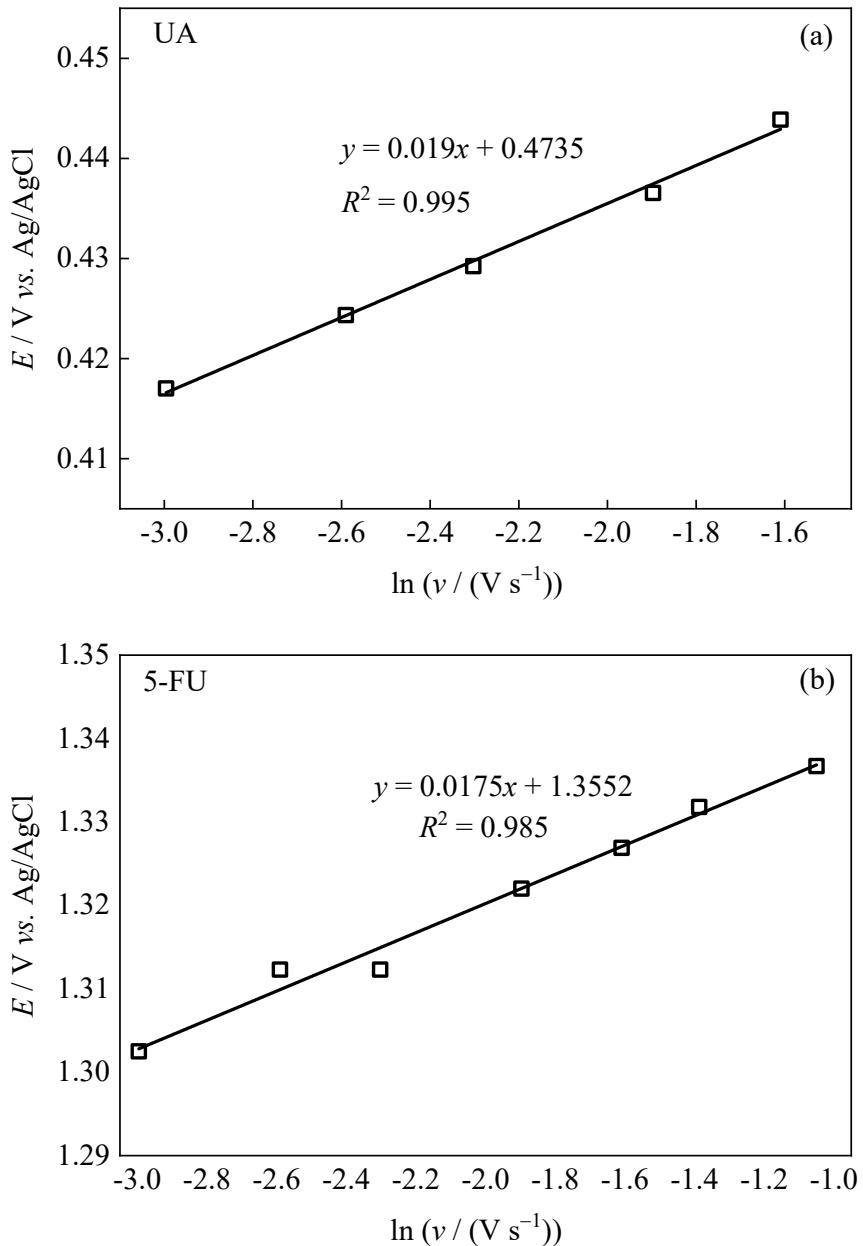


Fig. S6 The plots of the E_p vs. $\ln v$ for (a) UA and (b) 5-FU in 0.1 mol L⁻¹ PBS containing (a) 1.2 mmol L⁻¹ UA and (b) 1.2 mmol L⁻¹ 5-FU at GCE/PNMA-CeO₂-fMWCNTs.