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3 **Supplementary Materials**

4 **for**

5 **Turn-off/turn-on Biosensing of Tetracycline and Ciprofloxacin**
6 **Antibiotics by Fluorescent Iron Oxide Quantum Dots**
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38 **Table S1.** Determination results of TCy and CPx in the real sample (n=3)

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Antibiotic	Real Samples	Amount found (μM)	Added (μM)	Total found (μM)	Recovery (%)	RSD (%)
TCy	Drinking water	0	1	1.14 ± 0.01	114.33 ± 0.57	0.50
	Honey	0	10	11.77 ± 0.09	117.7 ± 0.95	0.81
CPx	Drinking water	0	1	1.12 ± 0.03	98.03 ± 0.75	0.76
	Honey	0	10	9.80 ± 0.08	112.67 ± 2.51	2.23

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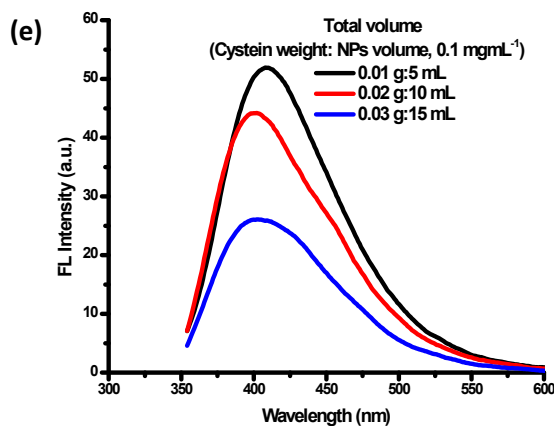
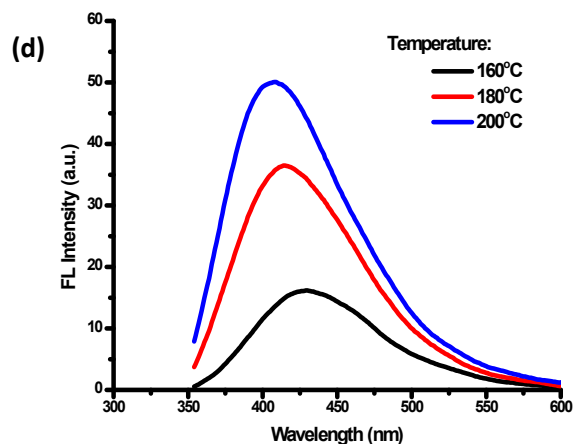
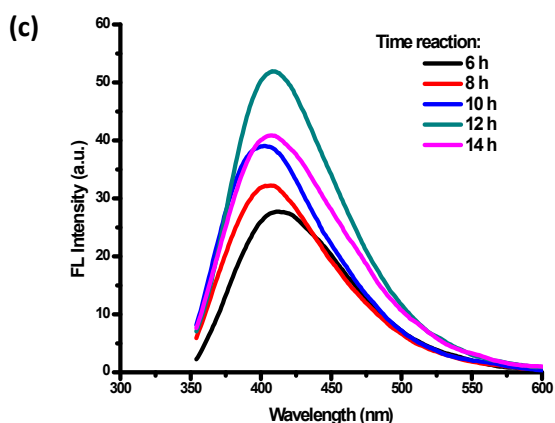
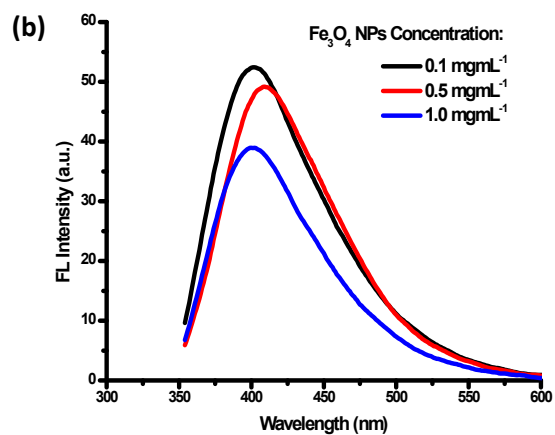
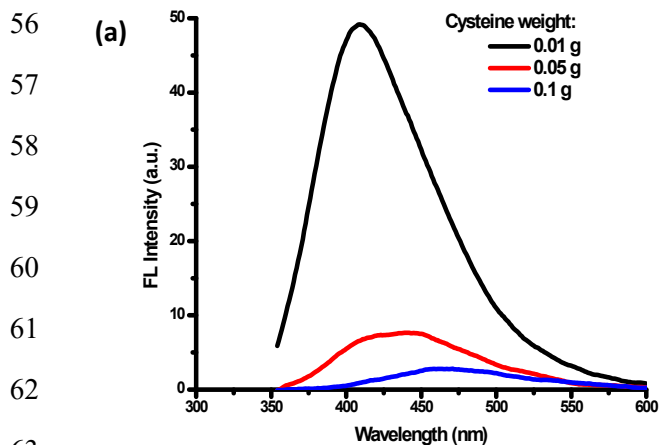
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54 **Figure S1**

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79 **Figure S1.** Synthetic condition of (a) cysteine weight (Fe_3O_4 NPs=0.5 mgmL^{-1} , 5 mL, $t=12\text{h}$,
80 $T=200^\circ\text{C}$). (b) Fe_3O_4 NPs (cysteine weight=0.01 g, $t=12$ h, $T=200^\circ\text{C}$). (c) reaction time (cysteine
81 weight=0.01 g, Fe_3O_4 NPs=0.1 mgmL^{-1} , 5 mL, $T=200^\circ\text{C}$). (d) temperature (cysteine weight=0.01
82 g, Fe_3O_4 NPs=0.1 mgmL^{-1} , 5 mL, $t=12$ h). (e) total volume ($t=12$ h, $T=200^\circ\text{C}$)

83 **Figure S2**

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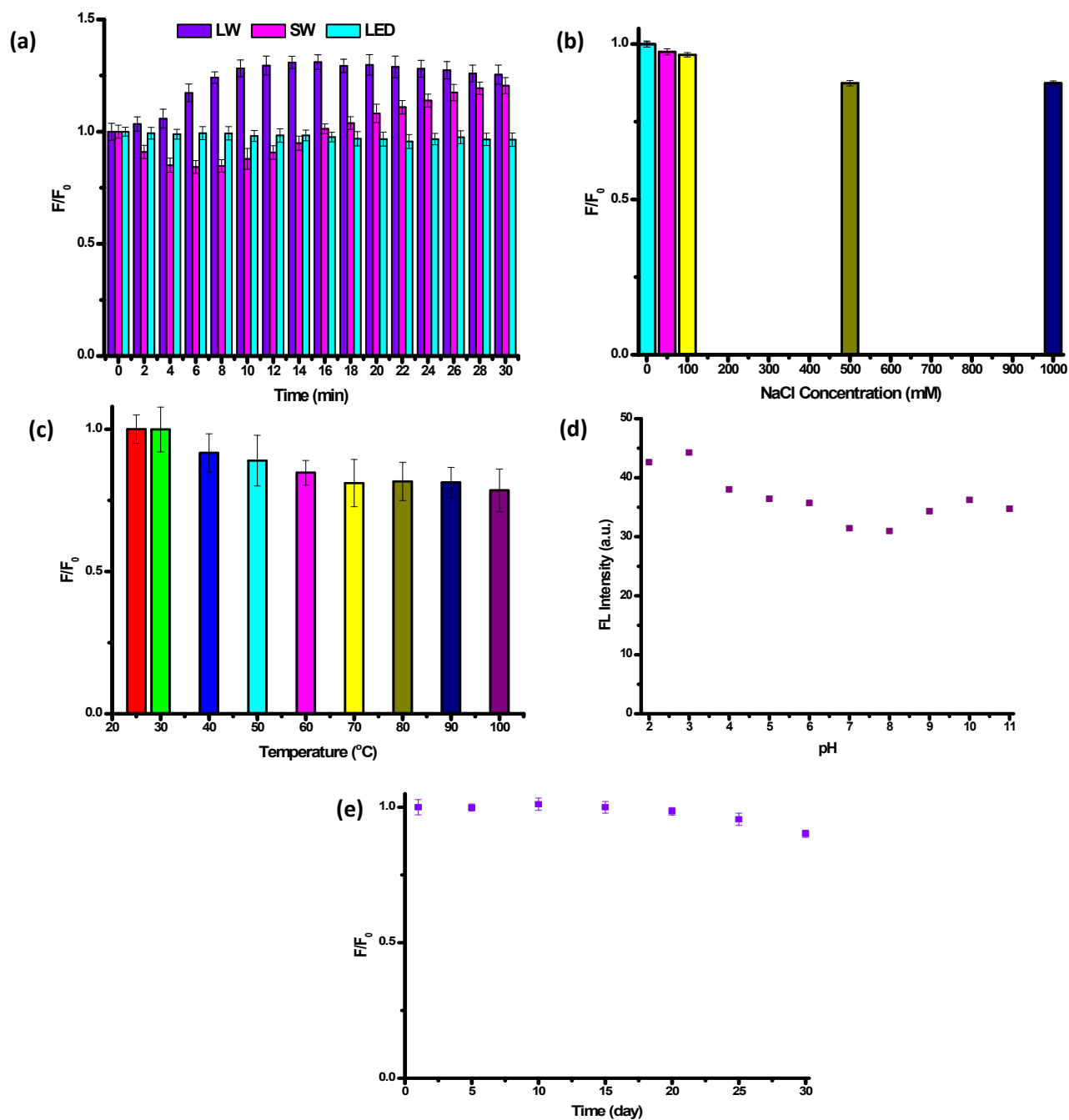
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108 **Figure S2.** (a) photostability of IO-QDs. (b) the ionic strength of IO-QDs. (c) thermal stability of
 109 IO-QDs. (d) pH stability of IO-QDs. (e) long term stability of IO-QDs

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112 **Figure S3**

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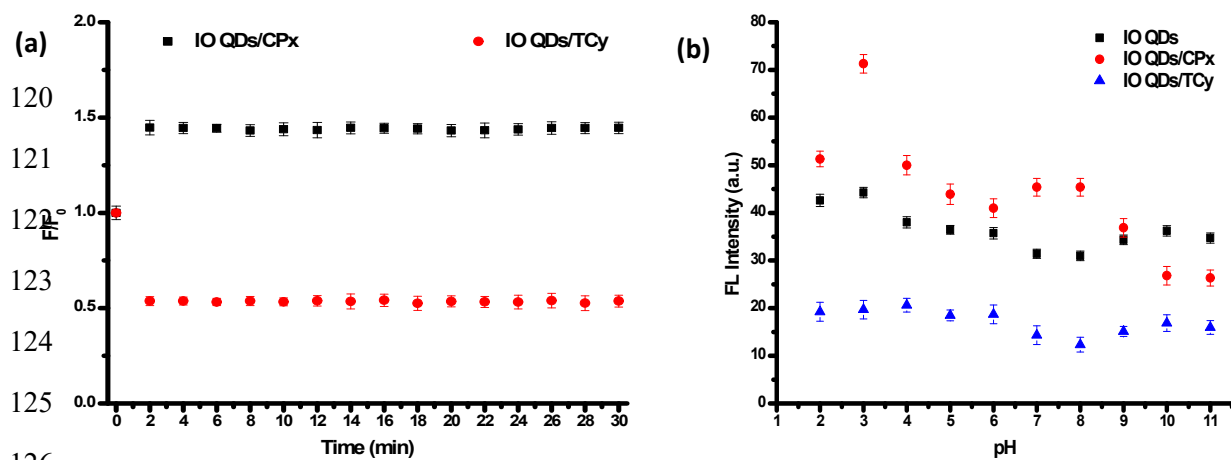
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128 **Figure S3.** Effect of (a) time reaction (0-30 min) after adding TCy and CPx at a concentration of
129 500 μ M, respectively. (b) pH value (2-11) of IO-QDs, IO-QDs/CPx, and IO-QDs/TCy system on
130 fluorescence intensity at a concentration of 500 μ M.

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141 **Figure S4**

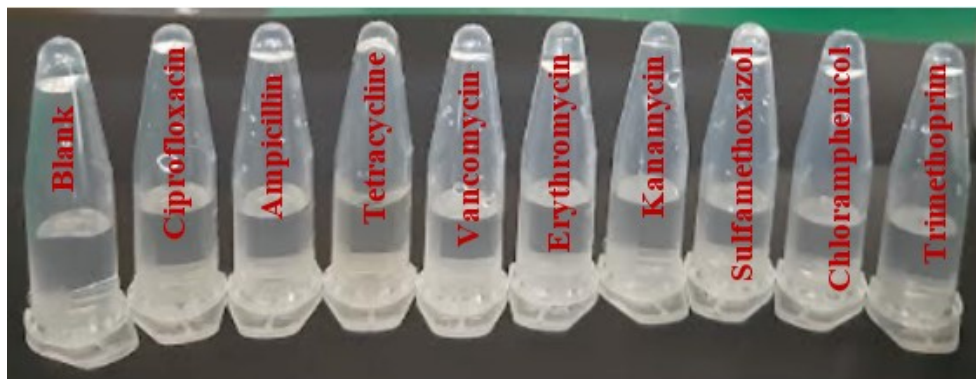
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(a)



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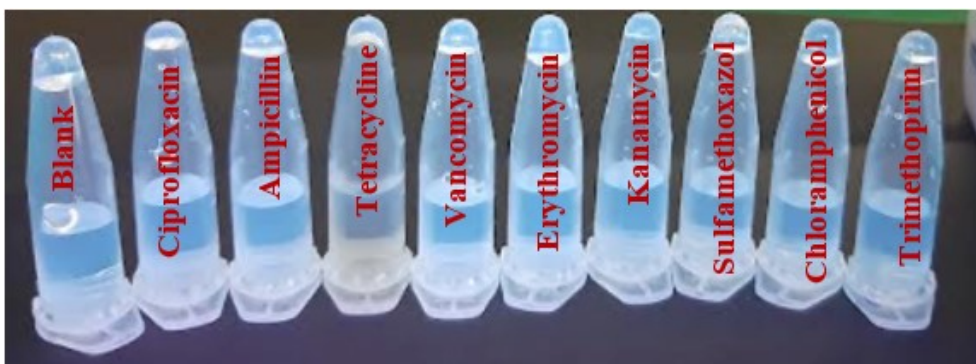
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(b)



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160 **Figure S4.** IO-QDs solution in the absence and presence of various antibiotics at 200 μ M under

161 (a) daylight. (b) UV light ($\lambda=365$ nm)

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170 **Figure S5**

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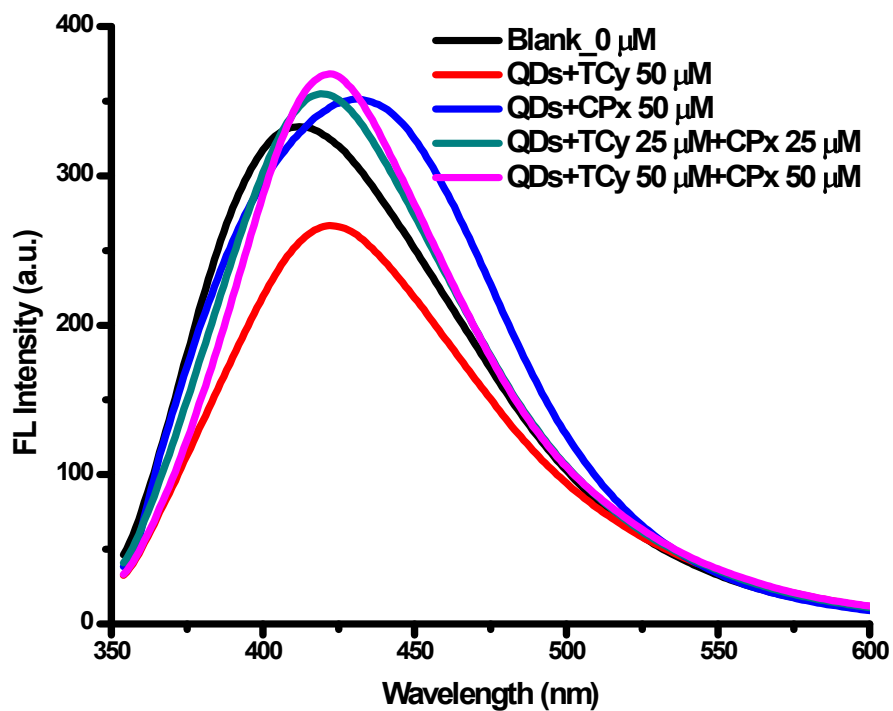


Figure S5. Spectra of IO-QDs probe with TCy, CPx, and TCy/CPx mixture, respectively

199 **Figure S6**

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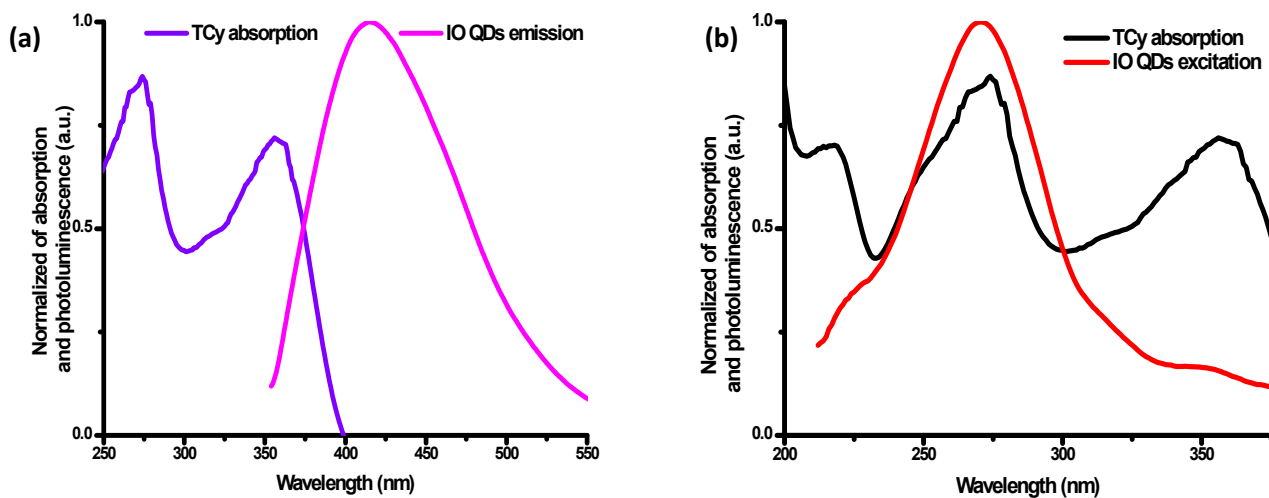


Figure S6. Spectral overlap in (a) FRET. (b) IFE

227 **Figure S7**

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245 **Figure S7.** The fluorescence spectrum of ciprofloxacin at an excitation wavelength of 330 nm

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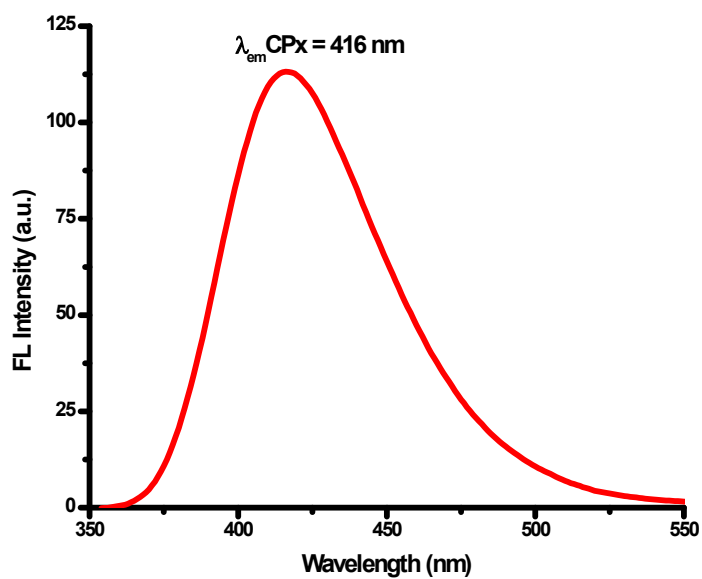
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256 **Förster resonance energy transfer (FRET) calculation**

257 Förster distance (R_0) was calculated using the formula [1]:

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$$R_0 = 0.02108(k^2 \phi_D \eta^{-4} J)^{1/6} \quad (1)$$

259 Where R_0 is the Förster distance (in nm), k^2 denotes dipole orientation factor (2/3) [2], ϕ_D
260 denotes fluorescence quantum yield of IO QDs, η denotes the refractive index of the solvent, and
261 J denotes the overlap integral ($\text{nm}^4 \text{M}^{-1} \text{cm}^{-1}$).

262 Overlap integral (J) was calculated [1]:

263
$$J = \int_0^\infty F_D(\lambda) \epsilon_A(\lambda) \lambda^4 d\lambda \quad (2)$$

264 Energy transfer efficiency (E) was calculated as follows [1]:

265
$$E = 1 - \frac{\tau_{DA}}{\tau_D} = 1 - \frac{IDA}{ID} \quad (3)$$

266 Where E refers to energy transfer efficiency, τ_{DA} and τ_D refer to the fluorescence lifetime of IO
267 QDs in the presence and absence of TCy, respectively.

268 Donor-acceptor distance (r) was estimated by engaging the formula [1]:

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$$E = \frac{1}{[1 + (\frac{r}{R_0})^6]} \quad (4)$$

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271 **Stern-Volmer plot**

272 The quenching efficiency represents the Stern-Volmer quenching constant (K_{SV}) which can be
273 calculated as follow [3]:

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$$F_0/F = K_{sv}[Q] + 1 \quad (5)$$

275 where F and F_0 denote the fluorescence intensity of the IO QDs in the presence and absence TCy,
276 K_{SV} denotes the Stern-Volmer quenching constant, and Q denotes the concentration of TCy.

277 The quenching rate constant was calculated according to the equation as follows [4]:

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$$K_{SV} = K_q \tau_0 \quad (6)$$

279 K_q is the quenching rate constant, and τ_0 is the fluorescence lifetime of IO QDs.

280 **References:**

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