Controlled Degradation of Polycaprolactone-based Micropillar Arrays

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Supplementary Information



Figure S1: <sup>1</sup>H NMR spectra of PCLDMA



Figure S2: FTIR spectra of PCLDMA



**Figure S3**: FTIR spectra of neat HDDA (1 % PBPO w/w) photopolymerised using the white light irradiation source at different time points from 0 - 120 mins. Top: The full spectra of the timepoints are shown. Middle: A close up of the peaks centred at 1410 cm<sup>-1</sup> related to the vibration of terminal allyl –CH groups. Bottom: 1620 cm<sup>-1</sup> and 1638 cm<sup>-1</sup> describe the stretching of the C=C bonds of acrylate groups.



**Figure S4:** FTIR spectra of 1:9 PCLDMA:HDDA (1 % PBPO w/w) photopolymerised using the white light irradiation source at different time points from 0 - 120 mins. Top: The full spectra of the timepoints are shown. Middle: A close up of the peaks centred at 1410 cm<sup>-1</sup> related to the vibration of terminal allyl –CH groups. Bottom: 1620 cm<sup>-1</sup> and 1638 cm<sup>-1</sup> describe the stretching of the C=C bonds of acrylate groups.



**Figure S5**: FTIR spectra of 1:4 PCLDMA:HDDA (1 % PBPO w/w) photopolymerised using the white light irradiation source at different time points from 0 - 120 mins. Top: The full spectra of the timepoints are shown. Middle: A close up of the peaks centred at 1410 cm<sup>-1</sup> related to the vibration of terminal allyl –CH groups. Bottom: 1620 cm<sup>-1</sup> and 1638 cm<sup>-1</sup> describe the stretching of the C=C bonds of acrylate groups.



Figure S6: SEM images of multiple two-photon polymerised templates 'stitched' together to form a large  $390 \times 390 \ \mu m$  template.



Figure S7: SEM images of large arrays of pillars replicated using 10%PH from the large 390  $\times$  390 µm template *via* NIL.



Figure S8: Average mass vs days in 5 M NaOH of bulk samples of HDDA control (black), 10%PH (red), and 20%PH (blue).



**Figure S9:** FTIR spectra of **20%PH** containing 1 % w/w PBPO before degradation in 5 M NaOH (dry) and after degradation for 160 days in 5 M NaOH. The full spectra are shown (top) as well as a close up of the regions peaks centred at 3400 cm<sup>-1</sup> which described the stretching of the OH bonds (middle) and 1723 cm<sup>-1</sup> and 1560 cm<sup>-1</sup> related to the stretching of C=O bonds and carboxylate anions, respectively (bottom).



**Figure S10:** FTIR spectra of **10%PH** containing 1 % w/w PBPO before degradation in 5 M NaOH (dry) and after degradation for 160 days in 5 M NaOH. The full spectrum is shown (top) as well as a close up of the regions peaks centred at 3400 cm<sup>-1</sup> which described the stretching of the OH bonds (middle) and 1723 cm<sup>-1</sup> and 1560 cm<sup>-1</sup> related to the stretching of C=O bonds and carboxylate anions, respectively (bottom).



**Figure S11:** FTIR spectra of neat HDDA containing 1 % w/w PBPO before degradation in 5 M NaOH (dry) and after degradation for 160 days in 5 M NaOH. The full spectrum is shown (top) as well as a close up of the regions peaks centred at 3400 cm<sup>-1</sup> which described the stretching of the OH bonds (middle) and 1723 cm<sup>-1</sup> and 1560 cm<sup>-1</sup> related to the stretching of C=O bonds and carboxylate anions, respectively (bottom).



**Figure S12:** Atomic force height images of  $2x2x2 \ \mu m$  samples, shows traces used to give height profiles in **Figure 5**. From left to right: HDDA, **10%PH** and **20%PH**, all after 5 days NaOH exposure.