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# Supporting Information

# Maximal Emission beyond 1200 nm Dicyanovinyl-Functionalized Squaraine for in vivo Vascular Imaging

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#### 1. Materials and Methods

All solvents and reagents were purchased from commercial sources and used as received without further purification. The bovine serum albumin was purchase from the Sigma-aldrich. All reactions were carried out under nitrogen atmosphere. All other solvents were purchased from Fisher Scientific.

#### 2. Characterization

<sup>1</sup>H NMR and <sup>13</sup>C NMR spectra were recorded on an AVANCE III HD 400 MHz spectrometer (Bruker, Germany) using TMS as an internal standard. High-resolution mass spectra (HRMS) were recorded using UHPLC30A-Trip TOF 5600+ (SCIEX, Japan). The UV-Vis-NIR absorption spectra were recorded using a Lambda 750 spectrophotometer (PerkinElmer, U. S.). Photoluminescence (PL) spectra were recorded using an FLS1000 fluorescence spectrophotometer (Edinburgh, U. K.). Transmission electron microscopy (TEM) imaging was conducted by a JEM-2100 transmission electron microscope operating at an acceleration voltage of 100 kV. Dynamic light scattering was performed on a particle size analyzer (Mastersizer 3000 system, Malvern, U. K.). The in vivo and in vitro NIR-II fluorescence imaging experiments were operated on a NIR-II imaging system (Wuhan Grand-imaging Technology Co., Ltd) with a 1200 nm LP filter under the 1064 nm laser irradiation. The NIR-II fluorescence imaging system was equipped with an InGaAs camera (Princeton Instruments).





Scheme S1. Synthetic Route of NSQ1270.

Compounds 2 and 6 were prepared according to previously reported methods. [1]



Synthesis of compound **3**. compound **2** (2.48 g, 10 mmol), 1-Bromo-2octyldodecane (10.8 g, 30 mmol), and  $K_2CO_3$  (8.29 g, 60 mmol) were dissolved in acetonitrile (50 ml). After the mixture was refluxed for 72 h, the solvent was evaporated under reduced pressure, and the mixture was extracted with ethyl acetate (2×50 ml) and water. The organic layer was

dried over sodium sulfate and filtered. The solvent was removed under reduced pressure to obtain the crude product and further purified via solid silica gel column chromatography (Petroleum ether / Ethyl acetate = 50/1) to afford compound **3** as a yellow oil (4.03 g, 76.3% yield). 1H NMR (400 MHz, Chloroform-d):  $\delta$  8.10 (d, J = 8.3 Hz, 1H), 8.05 (d, J = 7.0 Hz, 1H), 7.76 (t, J = 7.6 Hz, 1H), 7.61 (d, J = 7.5 Hz, 1H), 6.70 (d, J = 7.5 Hz, 1H), 3.75 (d, J = 7.4 Hz, 2H), 1.93 (p, J = 6.3 Hz, 1H), 1.32 (q, J = 5.9 Hz, 8H), 1.23 (d, J = 16.7 Hz, 23H), 0.86 (q, J = 6.5 Hz, 6H). 13C NMR (101 MHz, Chloroform-d):  $\delta$  167.61, 139.69, 131.10, 130.11, 129.57, 128.82, 126.87, 126.09, 124.99, 113.76, 106.10, 44.76, 37.24, 31.89, 31.84, 31.58, 29.92, 29.59, 29.55, 29.50, 29.31, 29.25, 26.49, 22.67, 22.63, 14.10, 14.09. TOF-MS m/z: calculated for compound 6; [M+H]<sup>+</sup> C<sub>31</sub>H<sub>47</sub>BrNO<sup>+</sup>:528.2836, found:528.2842.



Synthesis of compound **4**. 4,4'-dimethoxy-4-biphenylamin (1.374 g, 6 mmol), compound **3** (2.64 g, 5 mmol), sodium tert-butoxide (1.44 g, 15 mmol), and tri-tert-butylphosphine tetrafluoroborate (4.35 g, 15 mmol) was added to toluene (50 mL). The mixture was then purged with argon for 30 min.  $Pd_2(dba)_3$  (458 mg, 0.50 mmol) was added to the mixture. The reaction was stirred at 110 °C for 16 h under Ar. After cooling to room

temperature, the solvent was evaporated under reduced pressure, and the resulting mixture was extracted with ethyl acetate (2×100 ml) and water. The organic layer was dried over sodium sulfate and filtered. The solvent was removed under reduced pressure to obtain the crude product and further purified by solid silica gel column chromatography (Petroleum ether / Ethyl acetate = 50/1) to afford compound **4** as a clear red oil (2.31 g, 68.3 % yield). <sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 7.99 (d, J = 6.9 Hz, 1H), 7.77 (d, J = 8.3 Hz, 1H), 7.52 – 7.45 (m, 1H), 7.03 (d, J = 7.5 Hz, 1H), 6.97 (d, J = 8.6 Hz, 4H), 6.81 – 6.73 (m, 5H), 3.76 (m, J = 7.4 Hz, 8H), 1.95 (s, 1H), 1.36 (d, J = 5.1 Hz, 6H), 1.23 (s, 25H), 0.86 (d, J = 4.0 Hz, 6H). <sup>13</sup>C NMR (101 MHz, Chloroform-*d*): δ 168.37, 154.85, 142.79, 140.07, 136.50, 128.88, 128.08, 127.18, 126.85, 126.48, 125.62, 124.22, 124.09, 114.50, 106.26, 55.48, 44.66, 37.44, 31.92, 31.89, 31.58, 30.01, 29.64, 29.59, 29.36, 29.31, 26.55, 22.71, 22.68, 14.16.131.52, 131.49, 131.47, 129.71, 129.66, 129.49, 128.37, 125.58, 121.56, 109.99, 106.67, 91.77, 38.59, 13.94. TOF-MS m/z: calculated for compound 4, [M+H]<sup>+</sup> C<sub>45</sub>H<sub>61</sub>N<sub>2</sub>O<sub>3</sub><sup>+</sup>: 677.4667, found: 677.4585.



Synthesis of compound **5**. Methylmagnesium chloride (2.0 mL, 3 M solutions in THF, six mmol) was added to a solution of compound **4** (1.35 mg, two mmol) in anhydrous THF (10 mL) at 0 °C. The mixture was stirred at 60 °C for two hours. After cooling to 0 °C, water (0.5 mL) and then perchloric acid (70% solution in water, 2 mL) were added to the reaction mixture. The orange-red solution obtained was extracted with

dichloromethane (2×50 ml) and water. The organic layer was dried over sodium sulfate and filtered. The solvent was removed under reduced pressure to obtain the crude product as an orange-red oil. The resulting crude product was used directly in the next step without purification.



Synthesis of **NSQ1270**. A solution of compound **5** (2 mmol) and compound **6** (290 mg, one mmol) in toluene (10 mL) and n-butanol (10 mL) was heated to reflux in a Dean-Stark apparatus for one h. After the completion of the reaction, the mixture was concentrated in a vacuum. The residue was purified

by solid silica gel column chromatography (Petroleum ether / Ethyl acetate = 15/1) to afford **NSQ1270** as a black solid (477 mg, 46.7% yield). <sup>1</sup>H NMR (400 MHz, Tetrahydrofuran- $d_8$ ):  $\delta$  8.51 (d, J = 7.2 Hz, 2H), 7.58 (d, J = 8.1 Hz, 2H), 7.41 (t, J = 7.7 Hz, 2H), 7.03 (d, J = 7.8 Hz, 2H), 6.94 – 6.83 (m, 10H), 6.69 (d, J = 8.5 Hz, 10H), 4.01 (d, J = 6.2 Hz, 4H), 3.61 (s, 12H), 2.11 (s, 2H), 1.34 (s, 12H), 1.15 (d, J = 10.6 Hz, 50H), 0.75 (t, J = 6.7 Hz, 12H). <sup>13</sup>C NMR (101 MHz, Tetrahydrofuran- $d_8$ )  $\delta$  173.04, 166.60, 161.29, 155.62, 149.93, 142.44, 142.15, 137.19, 131.08, 129.82, 127.74, 127.61, 126.12, 125.82, 124.60, 123.69, 117.07, 114.17, 113.83, 109.72, 93.25, 54.26, 42.44, 37.46, 31.53, 31.51, 31.01, 29.62, 29.26, 29.17, 29.12, 29.03, 28.97, 28.92, 25.93, 24.38, 23.97, 23.78, 23.58, 22.22, 13.17, 13.16. TOF-MS m/z: calculated for NSQ1270, [M+H] + C<sub>99</sub>H<sub>123</sub>N<sub>6</sub>O<sub>5</sub>+: 1475.9550, found: 1475.9437.

#### 4. Determination of fluorescence quantum yield (QY) of the dyes

The fluorescence quantum yield ( $\phi$ ) was measured following the previous method. [2] Briefly,  $\phi$  was calculated by using the following equation:

$$\phi_X = \phi_S \frac{R_X}{R_S} \left( \frac{\eta_X}{\eta_S} \right)^2$$

where the subscripts S and X mean the standard and test samples, respectively. R is the slope from the integrated emission intensity versus absorbance.  $\eta$  is the refractive indices of solvents. NSQ1270 were measured by using IR-26 ( $\phi = 0.05\%$  in dichloroethane) as a standard. [3]

NSQ1270 and IR-26 were excited at 1064 nm. The absorbance of each dye was below 0.05 at the excited wavelength.

# 5. DFT Calculations

Density functional theory (DFT) calculations were performed on Gaussian 09 program. [4] The single molecular geometries and the frontier orbital energy levels were obtained by the Becke three-parameter hybrid functional combined with Becke-Lee-Yang-Parr correlation functional (B3LYP) with 6-31G(d) basis sets. Theoretical predictions on all long alkyl substituents were replaced with methyl groups in the calculations.

## 6. Fabrication of NSQ1270@BSA NCs.

BSA (100 mg) was dissolved in 5 mL of water, and then one equivalent amount of NSQ1270 methanol solution (1.0 mL) was added to the BSA solution dropwise with vigorous ultrasonic (19 W) at room temperature; consequently, NSQ1270@BSA NCs were obtained. Glutaraldehyde solution (25%, 5  $\mu$ L) was added to cross-link NCs for 12 h at room temperature. An optically clear aqueous dispersion containing NSQ1270@BSA NCs was prepared, which was then purified with a 0.22  $\mu$ m PES membrane syringe-driven filter. The final aqueous dispersions were obtained by ultrafiltration centrifuge filtration (100 kDa filter, 6000 rpm for 20 min) tube for further experiments.

### 7. Cell culture

B16F10 tumor cells were incubated in Dulbecco's Modified Eagle Medium containing 10% fetal bovine serum (FBS) and 1% penicillin/streptomycin at 37 °C in humidified atmosphere of 5% CO<sub>2</sub>, and then the cells were collected for the following cell experiments.

### 8. Animal Experiments.

The use of mice in this study was in accordance with the Regulations for the Administration of Laboratory Animals in China and obtained the ethical approval of the Animal Protection and Utilization Committee of Nanchang University (NCU-ACC-202210134681). C57BL/6 mice (18–20 g, male) were obtained from the Hunan Laboratory Animal Co., Ltd. (Changsha, China). The mice were housed under specific-pathogen-free (SPF) conditions at 24 °C with a 12 h light–dark cycle and fed with standard laboratory water and chow.

### 9. NIR-II Fluorescence Vascular Imaging of Mice

NSQ1270@BSA NCs (100  $\mu$ L, 0.5 mg/mL) were intravenously injected. NIR-II fluorescence imaging was explored at whole-body vasculature of living mice upon exposed to 1064 nm laser (1 W/cm<sup>2</sup>) irradiation, and signals were collected by a 1200 nm LP filter. And then, the mice were sacrificed.

# 10. Supplementary Figures and tables.



Figure S2. <sup>13</sup>C-NMR spectrum of Compound 3 in Chloroform-*d*.



3.75

1.36 1.35 1.23 0.86 0.85

-1.95



7.99 77.98 77.75 77.75 77.75 7.7.749 77.03 7.04 7.03 6.99 6.96 6.76 6.76



Figure S4. <sup>1</sup>H-NMR spectrum of Compound 4 in Chloroform-d.



Figure S5. <sup>13</sup>C-NMR spectrum of Compound 4 in Chloroform-*d*.



Figure S6. TOF-MS mass spectrum of Compound 4.



Figure S7. <sup>1</sup>H-NMR spectrum of NSQ1270 in THF-*d*<sub>8</sub>.





Figure S8. <sup>13</sup>C-NMR spectrum of NSQ1270 in THF- $d_8$ .



Figure S9. TOF-MS mass spectrum of NSQ1270.

Table S1. Th	e photophysical	l data for NSC	)1270 at 298K.
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condition	$\lambda_{ab}$ [nm]	ε [ M <sup>-1</sup> cm <sup>-1</sup> ]	$\lambda_{em}{}^{a)}$ [nm]	ф <sub>РL</sub> ь) [%]
n-Hexane	1091	143,000	1153	0.88
DCM	1074	102,000	1138	0.42
MeOH	1052	101,000	1149	0.12
H <sub>2</sub> O	1124	29000	1270	-

a)  $\lambda_{ex}$ = 1064 nm; b) calculated by indirect method (with IR-26).



Figure S10. a) Normalized absorption, and b) normalized fluorescence spectra of NSQ1270 in FBS, PBS, H<sub>2</sub>O, and DMEM.



Figure S11. Fluorescence intensity of NSQ1270 and IR-26 in PBS under 6000s successively irradiation.



**Figure S12.** a) Normalized absorption spectral of NSQ1270 with variation of  $f_w$ , b) the absorbance trend diagram with different  $f_w$ .



Figure S13. Concentration-dependent absorption spectra of NSQ1270 in a): water and c): methyl cyclohexane, and its normalized absorption spectra in b): water and d): methyl cyclohexane.



Figure S14. Absorption spectra NSQ1270 and in the presence of different nucleophiles.



**Figure S15**. Particle size of NSQ1270@BSA NCs after 6-week storage under 4 °C. (insert: before and after photographs of NSQ1270@BSA NCs dispersion in water)



Figure S16. Normalized absorption of NSQ1270@BSA NCs after 6-week storage under 4 °C.



**Figure S17**. Particle size of NSQ1270@BSA NCs in water, PBS, 0.9% NaCl, and DMEM (insert: from left to right, photographs of NSQ1270@BSA NCs incubated with water, PBS, 0.9% NaCl, and DMEM).



**Figure S18**. NIR-II fluorescence images of whole body treated with NSQ1270@BSA NCs at 5 min post-injection under a) 1100 long-pass filter, b) 1200 long-pass filter, and c) 1300 long-pass filter; d) Cross-sectional intensity profiles (dots) and Gaussian fit (lines) along the white line (hindlimb vasculature) of different long-pass filter.

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