

Supporting information

**An electrochemical ratiometric biosensor for the detection of
dopamine based on MXene-Au nanocomposite**

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1. Experimental sections

Regents and instruments

5 mg·mL⁻¹ multilayer Ti₃C₂T_x MXene solution was purchased from Beike 2D Materials Co. Ltd (Suzhou, China). Methylene blue, HAuCl₄·3H₂O, and dopamine were bought from Aladdin Reagent (Shanghai) Co. Ltd. K₃[Fe(CN)₆], Na₂HPO₄, NaH₂PO₄, and KCl were purchased from Shanghai Macklin Biochemical Co., Ltd. The above reagents were analytically pure and used without further treatment. Milli-Q water purification system was employed to produce the ultrapure water (18.2 MΩ·cm) used in this work.

Transmission electron microscopy (TEM) images were observed on a JEM-2100 instrument. X-ray diffraction (XRD) measurements were conducted on a Rigaku Smartlab diffractometer with Cu Kα radiation. All the electrochemical measurements were performed on a CHI660E electrochemical station (Shanghai CH Instruments Co. Ltd) with three-electrode system: modified glassy carbon electrode as working electrode, platinum wire as counter electrode and Ag/AgCl (3 M KCl) as reference electrode.

Synthesis of MXene-Au nanocomposite

MXene-Au was prepared based on the reported reference [24] with minor modification. First, 0.2 mL MXene solution (5 mg·mL⁻¹) was dropped into 4.8 mL water and sonicated for 30 min. Next, 1.0 mL HAuCl₄ (20 mM) was dropped into the above solution. After reacting for 5 min under continuously stirring, the mixture was centrifugated and washed three times with water. Finally, MXene-Au nanocomposite was obtained by drying the precipitate at 60°C for 12 h in vacuum.

Synthesis of MB-MXene-Au nanocomposite

1.0 mg MXene-Au nanocomposite was added into 1.0 mL 25 μM methylene blue (MB) solution and sonicated for 1 h. The superfluous MB was removed by centrifugation and washing.

Fabrication of MB-MXene-Au/GCE

First, glassy carbon electrode (GCE) was polished with Al₂O₃ suspension on polishing cloth. After being washed with water and ethanol, the performance of GCE

was tested by scanning the cycle voltammetry of $[\text{Fe}(\text{CN})_6]^{3-/4-}$ in 0.10 M KCl solution. Then the electrode was rinsed well with water and dried with nitrogen. Next, 6 μL 1 $\text{mg}\cdot\text{mL}^{-1}$ MB-MXene-Au solution was dropped on the electrode surface and dried naturally. Finally, 6 μL 0.025% Nafion was added on MB-MXene-Au/GCE to immobilize MB-MXene-Au on the electrode firmly.

Electrochemical measurements

Nafion/MB-MXene-Au/GCE was immersed into 5 mL pH 6.0 PBS containing different concentration of dopamine for electrochemical detection.

Differential pulse voltammetry (DPV) was measured with the parameters: +0.5 ~ -0.5V scan range, 4 mV potential increment, 50 mV pulse width, 50 mV amplitude, and 0.5 s pulse period. The electrochemical impedance spectroscopy (EIS) was carried out in 0.1 M KCl solution containing 20 mM $[\text{Fe}(\text{CN})_6]^{3-/4-}$ with a frequency range from 0.1 Hz to 100 kHz. The amplitude of the applied sine wave potential was 5 mV and the formal potential of the system was set at +0.22 V.

2. Figures

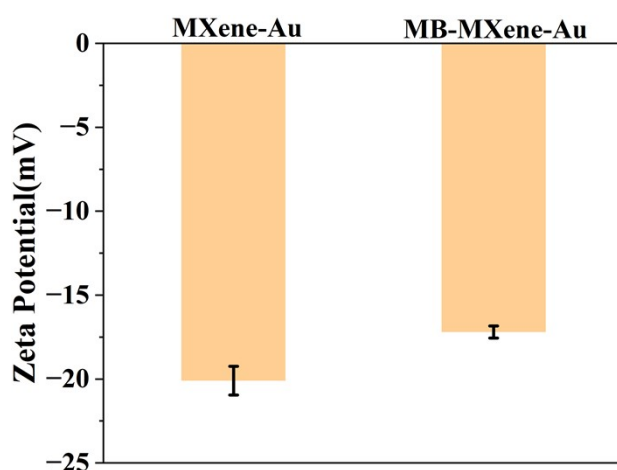


Figure S1. Zeta potential of MXene-Au and MB-MXene-Au

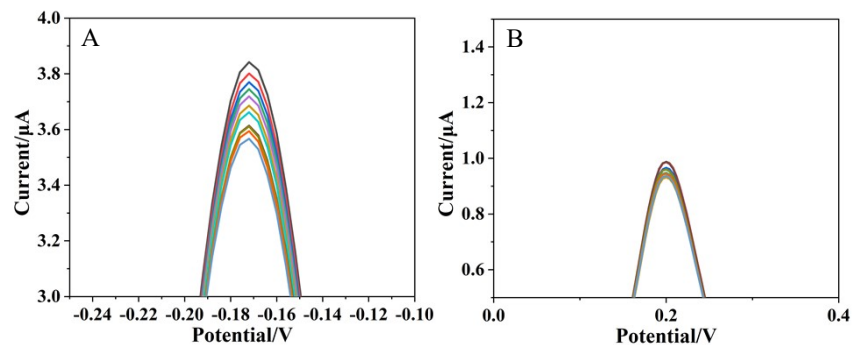


Figure S2. Stability study of the ratiometric biosensor. (A) and (B) are the local enlargement of DPV for MB and DA, respectively.

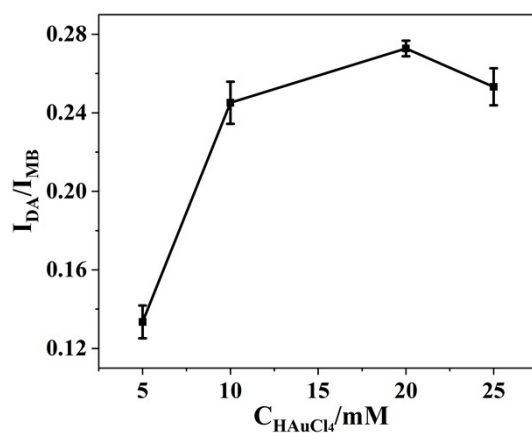


Figure S3. Effect of HAuCl_4 concentration on the performance of the ratiometric biosensor.

3. Tables

Table S1. Comparison of of DA detection based on various modified nanomaterials

Modified material	Detection technique	Linear range	Detection limit	Reference
Perylene diimide-Ti ₃ C ₂ T _x	DPV	100-1000 μ M	0.24 μ M	1
Ionic liquid/Ti ₃ C ₂ Cl ₂	DPV	10-2000 μ M	0.70 μ M	2
Ti ₃ C ₂ T _x /DNA/Pd/Pt	DPV	0.2-1000 μ M	0.03 μ M	3
Ti ₃ C ₂ /G-MWCNTs/ZnO	DPV	0.01-30 μ M	3.3 nM	4
Ti ₃ C ₂ /Holey graphene	DPV	0.5-50 μ M	0.06 μ M	5
SWCNT	DPV	0.4-150.0 μ M	0.22 μ M	6
S-doped graphene	DPV	0.2-12 μ M	0.015 μ M	7
N-doped reduced GO	DPV	0.5-150 μ M	0.41 μ M	8
3D N-doped graphene	DPV	1-1000 μ M	0.26 μ M	9
MXene-ERHG ^a	DPV	0.3-35 μ M	0.071 μ M	10
Cu- MOFs-MWCNTs	Ratiometric DPV	0.3-40 μ M	0.026 μ M	11
MB ^b /BP ^c -CNT	Ratiometric DPV	0.5–350 μ M	0.15 μ M	12
MWCNT	Ratiometric DPV	1.0–20.0 μ M	0.23 μ M	13
MnO ₂ /MWCNT	Ratiometric DPV	1.0–50.0	0.8	14
p(XA) ^d /Au/Cu-TCPP	Ratiometric DPV	5 - 125 μ M	1.0	15
MIPs ^e /pThi ^f /Au-Cu alloy	Ratiometric DPV	0.3–100 μ M	0.1 μ M	16
MB-MXene-Au	Ratiometric DPV	0.1–100 μM	0.04 μM	This work

Note:

^aERHG: electrochemically reduced holey graphene;

^bMB: methylene blue;

^cBP: 4-(pyren-4-yl)-N-(4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl)butanamide;

^dpXA: polyxanthurenic acid;

^eMIPs: molecularly imprinted polymers (MIPs);

^fpThi: polythionine.

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