

Electronic Supporting Information
Iodine(III)-Promoted Oxidative
Carbotrifluoromethylation of Maleimides with
Imidazopyridines and Langlois' Reagent

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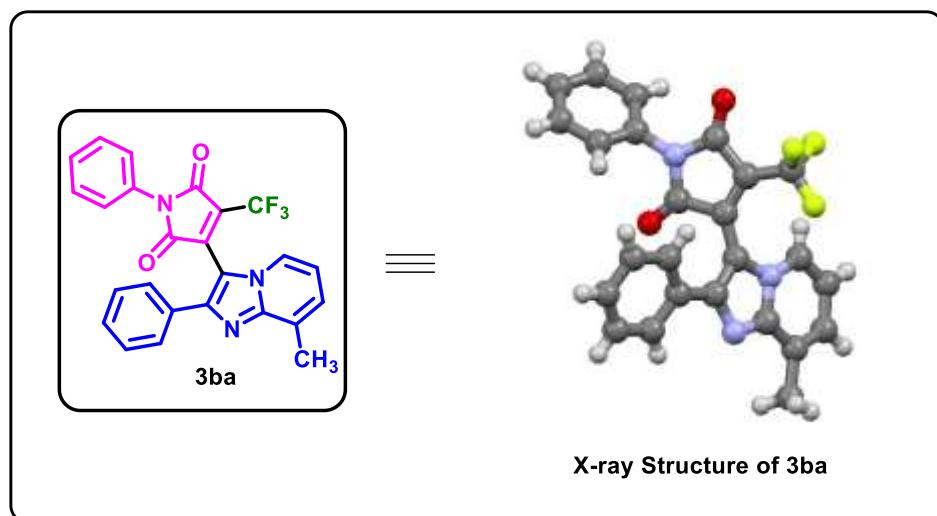
1. General information:

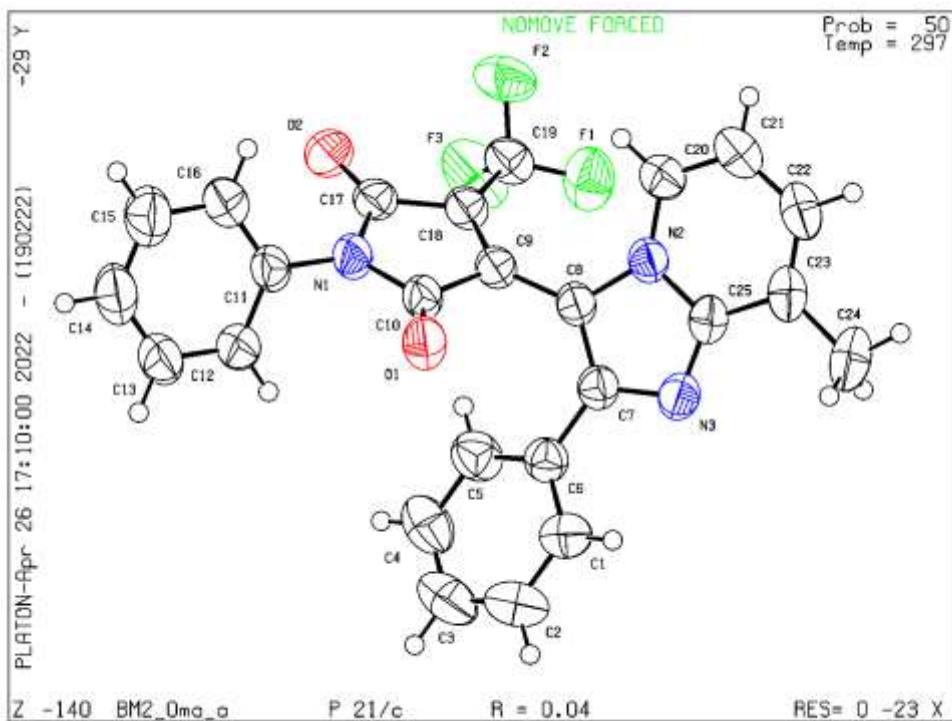
All reagents were purchased from commercial sources and used without further purification. ^1H NMR spectra were determined on a 400 MHz spectrometer as solutions in CDCl_3 . Chemical shifts are expressed in parts per million (δ) and the signals were reported as s (singlet), d (doublet), t (triplet), q (quartet), m (multiplet) and coupling constants (J) were given in Hz. $^{13}\text{C}\{\text{H}\}$ NMR and ^{19}F NMR spectra were recorded at 100 MHz and at 376 MHz in CDCl_3 solution respectively. Chemical shifts are referenced to CDCl_3 ($\delta = 7.26$ for ^1H and $\delta = 77.16$ for $^{13}\text{C}\{\text{H}\}$ NMR) as internal standard. TLC was done on a silica gel-coated glass slide. All solvents were dried and distilled before use. Commercially available solvents were freshly distilled before the reaction. Melting points (M.p.) were determined after the recrystallization of solid compounds from a solution of dichloromethane/petroleum ether (1:3). The crystallographic data for the compound **3ba** was collected by SCXRD-BRUKER D8QUEST and the crystal data was solved by APEX4 software.

Starting Materials: All the imidazopyridines (**1**)¹ and *N*-alkyl/aryl maleimides (**2**)² were prepared by previously reported methods.

2. Structure determination (X-ray crystallographic data for **3ba**):

The brown block shaped crystal of **3ba** was obtained by crystallization from a solution in dichloromethane/petroleum ether after purification by column chromatography. The chemical formula of compound **3ba**: $\text{C}_{25}\text{H}_{16}\text{F}_3\text{N}_3\text{O}_2$.





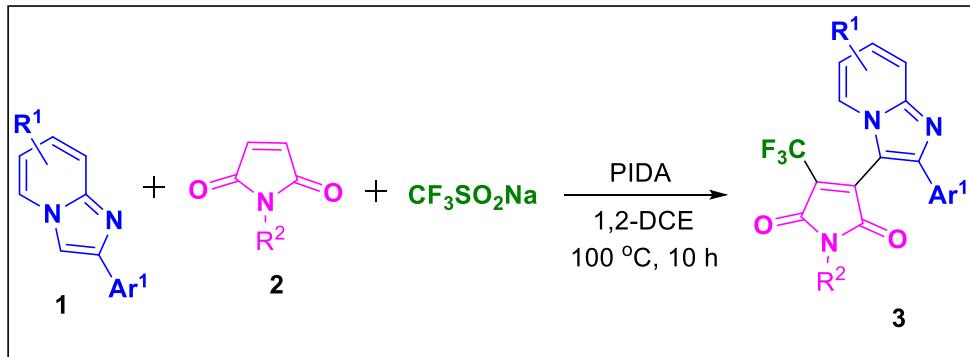
View of ORTEP diagram for the crystal structure of the compound **3-(8-Methyl-2-phenylimidazo[1,2-*a*]pyridin-3-yl)-1-phenyl-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3ba) (Thermal ellipsoid contour at 50% probability level).**

Wavelength	0.71073 Å	
Formula	$C_{25}H_{16}F_3N_3O_2$	
Crystal system	Monoclinic	
Space group	P 2 ₁ /c	
Unit cell dimensions	$a = 11.571(4) \text{ \AA}$	$\alpha = 90^\circ$
	$b = 21.464(8) \text{ \AA}$	$\beta = 107.385(9)^\circ$
	$c = 8.872(3) \text{ \AA}$	$\gamma = 90^\circ$
Volume	2102.8 \AA^3	
Z	4	

R-factor (%)	4.1
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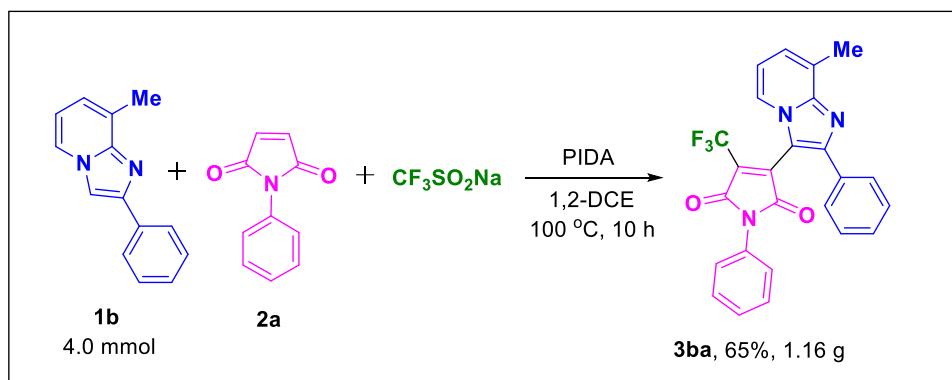
The crystallographic data have been deposited with the Cambridge Crystallographic Data Centre as a supplementary publication with a CCDC reference number CCDC 2232229.

3. General experimental procedure for the synthesis of 3aa-3bh:



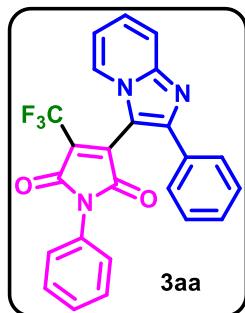
A mixture of 2-arylimidazo[1,2-*a*]pyridine (0.2 mmol) (**1**), *N*-aryl/alkyl maleimide (0.2 mmol) (**2**), CF₃SO₂Na (0.2 mmol, 31.2 mg) and PIDA (0.4 mmol, 128.8 mg) were taken in an oven-dried screw tube. Solvent 1,2-DCE (2 ml) was then added to the screw tube and the resultant mixture was stirred at 100 °C for 10 h. Progress of the reaction was monitored by TLC in regular intervals. After completion, the reaction mixture was allowed to cool to room temperature and then extracted with DCM (10 mL). The organic phase was dried over anhydrous Na₂SO₄ and concentrated under reduced pressure to get the crude residue which was further purified by column chromatography using silica gel (100–200 mesh) and using mixture of petroleum ether and ethyl acetate as an eluent to afford the corresponding products **3**.

4. Gram scale synthesis of compound **3ba**:

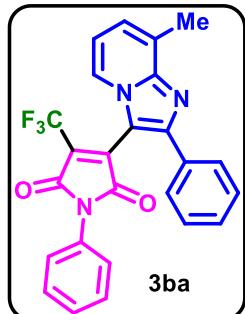


A mixture of 8-methyl-2-phenylimidazo[1,2-*a*]pyridine (4.0 mmol, 0.83 g) (**1b**), *N*-phenyl maleimide (4.0 mmol, 0.69 g) (**2a**), $\text{CF}_3\text{SO}_2\text{Na}$ (4.0 mmol, 0.62 g) and PIDA (8.0 mmol, 2.57 g) were taken in an oven-dried round bottom flask. Then, 1,2-DCE (40 ml) was added to it and the resultant mixture was then stirred at 100°C for 10 h. The progress of the reaction was monitored by TLC. After completion, the reaction mixture was allowed to cool to room temperature and then extracted with DCM (50 mL). The organic phase was dried over anhydrous Na_2SO_4 and concentrated under reduced pressure to get the crude residue which was further purified by column chromatography using silica gel (100–200 mesh) and using a mixture of petroleum ether and ethyl acetate (93:07) as an eluent to afford the corresponding product **3ba** (65%, 1.16 g) as a red solid.

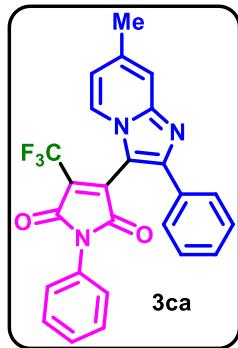
5. Characterization data of the synthesized compounds (3aa-3bh):



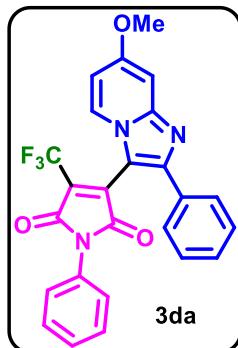
1-Phenyl-3-(2-phenylimidazo[1,2-a]pyridin-3-yl)-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3aa): Red solid (57.2 mg, 66%); M.p. 110–111 °C; $R_f = 0.50$ (PE : EA = 80 : 20); ^1H NMR (CDCl_3 , 400 MHz): δ 8.07 (d, $J = 7.2$ Hz, 1H), 7.78 (d, $J = 8.8$ Hz, 1H), 7.66–7.63 (m, 2H), 7.47–7.42 (m, 5H), 7.41–7.37 (m, 2H), 7.35–7.33 (m, 2H), 7.02–6.98 (m, 1H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 165.5, 164.3, 152.6, 148.7, 134.8, 133.6, 130.7, 129.8, 129.4, 129.0, 128.7, 128.6, 128.1, 126.4, 125.9, 125.5, 120.1 (q, $J_{\text{C}-\text{F}} = 275.0$ Hz), 118.4, 114.0, 108.1; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.52; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for $[\text{C}_{24}\text{H}_{15}\text{F}_3\text{N}_3\text{O}_2]^+$: 434.1111; found: 434.1109.



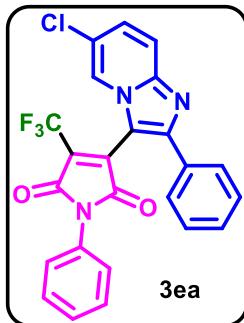
3-(8-Methyl-2-phenylimidazo[1,2-a]pyridin-3-yl)-1-phenyl-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3ba): Red solid (65.3 mg, 73%); M.p. 190–191 °C; $R_f = 0.50$ (PE : EA = 90 : 10); ^1H NMR (CDCl_3 , 400 MHz): δ 7.94 (d, $J = 6.8$ Hz, 1H), 7.66–7.64 (m, 2H), 7.48–7.43 (m, 5H), 7.40–7.36 (m, 1H), 7.33–7.31 (m, 2H), 7.21 (d, $J = 6.8$ Hz, 1H), 6.91 (t, $J = 7.2$ Hz, 1H), 2.71 (s, 3H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 165.5, 164.3, 152.1, 149.1, 135.0, 134.9, 133.8, 130.7, 129.4, 129.2, 128.9, 128.8, 128.6, 126.9, 125.9, 123.7, 123.3, 120.1 (q, $J_{\text{C}-\text{F}} = 270.0$ Hz), 114.0, 108.6, 17.2; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.43; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for $[\text{C}_{25}\text{H}_{17}\text{F}_3\text{N}_3\text{O}_2]^+$: 448.1267; found: 448.1287.



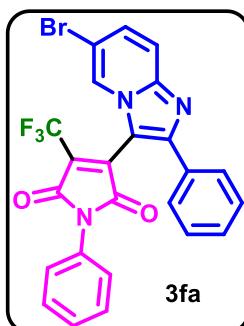
3-(7-Methyl-2-phenylimidazo[1,2-a]pyridin-3-yl)-1-phenyl-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3ca): Red solid (66.2 mg, 74%); M.p. 188-189 °C; $R_f = 0.55$ (PE : EA = 75 : 25); ^1H NMR (CDCl_3 , 400 MHz): δ 7.96 (d, $J = 7.2$ Hz, 1H), 7.64-7.62 (m, 2H), 7.52 (s, 1H), 7.48-7.41 (m, 5H), 7.40-7.36 (m, 1H), 7.34-7.32 (m, 2H), 6.84-6.82 (m, 1H), 2.48 (s, 3H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 165.6, 164.4, 153.0, 149.2, 139.9, 134.7, 133.8, 132.3, 130.7, 129.4, 129.3, 128.9, 128.7, 128.6, 125.9, 124.8, 120.2 (q, $J_{\text{C}-\text{F}} = 272.0$ Hz), 116.8, 116.6, 107.9, 21.6; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.30; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for $[\text{C}_{25}\text{H}_{17}\text{F}_3\text{N}_3\text{O}_2]^+$: 448.1267; found: 448.1275.



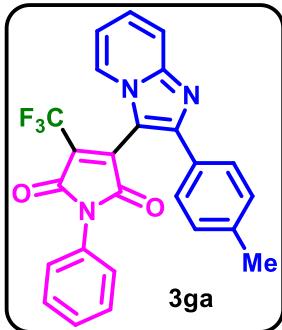
3-(7-Methoxy-2-phenylimidazo[1,2-a]pyridin-3-yl)-1-phenyl-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3da): Purple solid (70.4 mg, 76%); M.p. 190-191 °C; $R_f = 0.45$ (PE : EA = 70 : 30); ^1H NMR (CDCl_3 , 400 MHz): δ 7.90 (d, $J = 7.6$ Hz, 1H), 7.63-7.61 (m, 2H), 7.47-7.37 (m, 6H), 7.32 (d, $J = 7.6$ Hz, 2H), 7.03 (d, $J = 2.4$ Hz, 1H), 6.70-6.68 (m, 1H), 3.93 (s, 3H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 165.7, 164.5, 160.4, 153.6, 150.8, 134.4, 133.8, 130.7, 129.38, 129.36, 128.9, 128.7, 128.5, 126.2, 125.9, 121.9, 120.3 (q, $J_{\text{C}-\text{F}} = 272.0$ Hz), 108.7, 107.6, 95.7, 56.0; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.09; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for $[\text{C}_{25}\text{H}_{17}\text{F}_3\text{N}_3\text{O}_3]^+$: 464.1217; found: 464.1221.



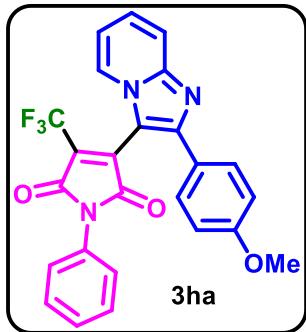
3-(6-Chloro-2-phenylimidazo[1,2-a]pyridin-3-yl)-1-phenyl-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3ea): Red solid (59.8 mg, 64%); M.p. 184-185 °C; $R_f = 0.50$ (PE : EA = 80 : 20); ^1H NMR (CDCl_3 , 400 MHz): δ 8.10 (s, 1H), 7.70 (d, $J = 9.6$ Hz, 1H), 7.63-7.61 (m, 2H), 7.49-7.41 (m, 6H), 7.39-7.34 (m, 3H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 165.3, 164.0, 152.8, 146.8, 138.1, 134.3, 133.1, 130.5, 129.6, 129.4, 129.3, 129.0, 128.7, 128.6, 125.9, 123.4, 122.3, 119.8 (q, $J_{\text{C}-\text{F}} = 274.0$ Hz), 114.6, 108.2; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.77; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for $[\text{C}_{24}\text{H}_{14}^{35}\text{ClF}_3\text{N}_3\text{O}_2]^+$: 468.0721; found: 468.0705.



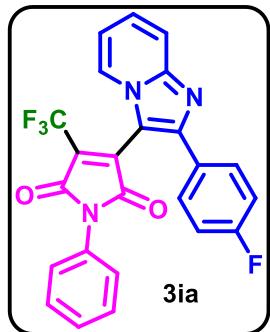
3-(6-Bromo-2-phenylimidazo[1,2-a]pyridin-3-yl)-1-phenyl-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3fa): Red solid (66.5 mg, 65%); M.p. 180-181 °C; $R_f = 0.50$ (PE : EA = 78 : 22); ^1H NMR (CDCl_3 , 400 MHz): δ 8.19 (s, 1H), 7.67-7.61 (m, 3H), 7.49-7.44 (m, 6H), 7.40-7.34 (m, 3H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 165.3, 164.0, 152.6, 148.6, 146.9, 134.3, 133.0, 131.4, 130.5, 129.6, 129.4, 129.0, 128.7, 128.6, 125.9, 125.6, 119.8 (q, $J_{\text{C}-\text{F}} = 272.0$ Hz), 118.8, 108.7, 108.0; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.77; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for $[\text{C}_{24}\text{H}_{14}^{79}\text{BrF}_3\text{N}_3\text{O}_2]^+$: 512.0216; found: 512.0202.



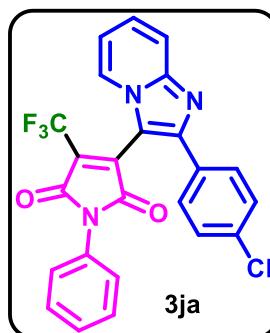
1-Phenyl-3-(*p*-tolyl)imidazo[1,2-*a*]pyridin-3-yl)-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3ga): Red solid (62.6 mg, 70%); M.p. 190-191 °C; $R_f = 0.50$ (PE : EA = 75 : 25); ^1H NMR (CDCl_3 , 400 MHz): δ 8.06 (d, $J = 7.2$ Hz, 1H), 7.76 (d, $J = 8.8$ Hz, 1H), 7.54 (d, $J = 8.0$ Hz, 2H), 7.49-7.44 (m, 2H), 7.42-7.39 (m, 1H), 7.37-7.33 (m, 2H), 7.27-7.25 (m, 3H), 6.99 (t, $J = 7.2$ Hz, 1H), 2.40 (s, 3H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 165.4, 164.2, 152.6, 148.6, 139.3, 134.87, 134.84, 130.6, 129.6, 129.2, 128.5, 127.9, 125.8, 125.7, 125.3, 123.8, 120.0 (q, $J_{\text{C}-\text{F}} = 277.0$ Hz), 118.1, 113.8, 107.7, 21.4; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.45; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for $[\text{C}_{25}\text{H}_{17}\text{F}_3\text{N}_3\text{O}_2]^+$: 448.1267; found: 448.1257.



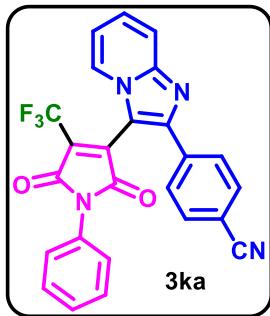
3-(4-Methoxyphenyl)imidazo[1,2-*a*]pyridin-3-yl)-1-phenyl-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3ha): Red solid (66.7 mg, 72%); M.p. 128-129 °C; $R_f = 0.45$ (PE : EA = 70 : 30); ^1H NMR (CDCl_3 , 400 MHz): δ 8.05 (d, $J = 6.8$ Hz, 1H), 7.75 (d, $J = 9.2$ Hz, 1H), 7.59 (d, $J = 8.8$ Hz, 2H), 7.48-7.33 (m, 6H), 7.00-6.96 (m, 3H), 3.85 (s, 3H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 165.5, 164.3, 160.6, 152.5, 148.7, 134.9, 130.7, 130.1, 129.4, 128.6, 128.5, 128.1, 126.1, 125.9, 125.4, 120.2 (q, $J_{\text{C}-\text{F}} = 271.0$ Hz), 118.1, 114.5, 113.8, 107.7, 55.4; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.38; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for $[\text{C}_{25}\text{H}_{17}\text{F}_3\text{N}_3\text{O}_3]^+$: 464.1217; found: 464.1210.



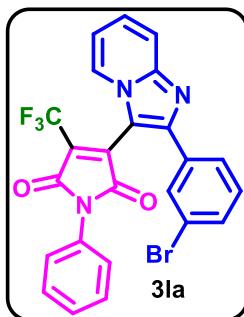
3-(2-(4-Fluorophenyl)imidazo[1,2-*a*]pyridin-3-yl)-1-phenyl-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3ia): Red solid (47.8 mg, 53%); M.p. 155-156 °C; $R_f = 0.50$ (PE : EA = 80 : 20); ^1H NMR (CDCl_3 , 400 MHz): δ 8.07 (d, $J = 6.8$ Hz, 1H), 7.77 (d, $J = 9.2$ Hz, 1H), 7.64-7.61 (m, 2H), 7.48-7.45 (m, 2H), 7.43-7.38 (m, 2H), 7.35-7.33 (m, 2H), 7.16 (t, $J = 8.8$ Hz, 2H), 7.02 (t, $J = 7.2$ Hz, 1H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 165.4, 164.2, 163.5 (d, $J_{\text{C}-\text{F}} = 248$ Hz), 151.5, 148.6, 134.4, 130.5 (d, $J_{\text{C}-\text{F}} = 9$ Hz), 129.7, 129.4, 128.7, 128.4, 126.4 (d, $J_{\text{C}-\text{F}} = 23$ Hz), 125.9, 125.5, 120.0 (q, $J_{\text{C}-\text{F}} = 275$ Hz), 118.2, 116.2, 116.0, 114.2, 107.9; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.51, -111.50; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for $[\text{C}_{24}\text{H}_{14}\text{F}_4\text{N}_3\text{O}_2]^+$: 452.1017; found: 452.1012.



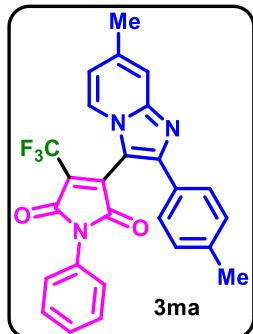
3-(2-(4-Chlorophenyl)imidazo[1,2-*a*]pyridin-3-yl)-1-phenyl-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3ja): Red solid (62.6 mg, 67%); M.p. 135-136 °C; $R_f = 0.50$ (PE : EA = 80 : 20); ^1H NMR (CDCl_3 , 400 MHz): δ 8.06 (d, $J = 4.8$ Hz, 1H), 7.78 (d, $J = 8.0$ Hz, 1H), 7.59 (d, $J = 8.0$ Hz, 2H), 7.50-7.43 (m, 6H), 7.34 (d, $J = 7.6$ Hz, 2H), 7.03-7.00 (m, 1H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 165.4, 164.1, 151.1, 148.6, 135.6, 134.5, 132.1, 130.6, 129.9, 129.4, 129.3, 128.7, 128.4, 125.9, 125.5, 124.9, 120.0 (q, $J_{\text{C}-\text{F}} = 275.0$ Hz), 118.4, 114.3, 108.0; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.46; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for $[\text{C}_{24}\text{H}_{14}^{35}\text{ClF}_3\text{N}_3\text{O}_2]^+$: 468.0721; found: 468.0717.



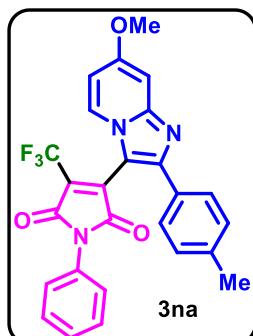
4-(3-(2,5-Dioxo-1-phenyl-4-(trifluoromethyl)-2,5-dihydro-1*H*-pyrrol-3-yl)imidazo[1,2-*a*]pyridin-2-yl)benzonitrile (3ka): Yellow solid (56.8 mg, 62%); M.p. 180-181 °C; $R_f = 0.45$ (PE : EA = 75 : 25); ^1H NMR (CDCl_3 , 400 MHz): δ 8.08 (d, $J = 6.8$ Hz, 1H), 7.80-7.72 (m, 5H), 7.50-7.46 (m, 3H), 7.42-7.38 (m, 1H), 7.33 (d, $J = 7.6$ Hz, 2H), 7.05 (t, $J = 7.2$ Hz, 1H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 165.3, 163.9, 150.0, 148.6, 138.1, 133.9, 132.6, 130.4, 129.5, 129.2, 128.8, 128.6, 125.8, 125.5, 125.3, 124.9, 119.9 (q, $J_{\text{C}-\text{F}} = 272.0$ Hz), 118.5, 114.6, 112.7, 108.5; ^{19}F NMR (376 MHz, CDCl_3): δ -61.52; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for $[\text{C}_{25}\text{H}_{14}\text{F}_3\text{N}_4\text{O}_2]^+$: 459.1063; found: 459.1048.



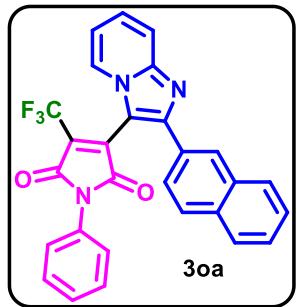
3-(2-(3-Bromophenyl)imidazo[1,2-*a*]pyridin-3-yl)-1-phenyl-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3la): Red solid (59.4 mg, 58%); M.p. 140-141 °C; $R_f = 0.55$ (PE : EA = 80 : 20); ^1H NMR (CDCl_3 , 400 MHz): δ 8.06 (d, $J = 7.2$ Hz, 1H), 7.87 (s, 1H), 7.78 (d, $J = 8.8$ Hz, 1H), 7.56-7.50 (m, 2H), 7.48-7.43 (m, 3H), 7.42-7.38 (m, 1H), 7.36-7.29 (m, 3H), 7.02 (t, $J = 6.8$ Hz, 1H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 165.4, 164.1, 150.5, 148.6, 137.4, 135.5, 134.3, 132.3, 131.6, 130.5, 130.3, 129.4, 128.8, 128.4, 127.2, 126.0, 125.5, 123.1, 122.7 (q, $J_{\text{C}-\text{F}} = 272.0$ Hz), 118.4, 114.3, 108.2; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.45; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for $[\text{C}_{24}\text{H}_{14}^{79}\text{BrF}_3\text{N}_3\text{O}_2]^+$: 512.0216; found: 512.0201.



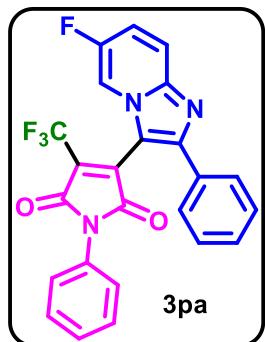
3-(7-Methyl-2-(*p*-tolyl)imidazo[1,2-*a*]pyridin-3-yl)-1-phenyl-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3ma): Red solid (71.9 mg, 78%); M.p. 198-199 °C; $R_f = 0.50$ (PE : EA = 80 : 20); ^1H NMR (CDCl_3 , 400 MHz): δ 7.94 (d, $J = 7.2$ Hz, 1H), 7.53-7.51 (m, 3H), 7.48-7.44 (m, 2H), 7.39 (d, $J = 7.2$ Hz, 1H), 7.33 (d, $J = 7.6$ Hz, 2H), 7.25-7.24 (m, 2H), 6.83-6.81 (m, 1H), 2.47 (s, 3H), 2.39 (s, 3H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 165.6, 164.5, 153.1, 149.2, 139.8, 139.4, 134.8, 130.9, 130.8, 129.7, 129.4, 128.6, 128.5, 126.0, 124.7, 122.7, 120.3 (q, $J_{\text{C}-\text{F}} = 273.0$ Hz), 116.7, 116.5, 107.7, 21.6, 21.5; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.23; HRMS (ESI-TOF) m/z: $[\text{M} + \text{H}]^+$ Calcd for $[\text{C}_{26}\text{H}_{19}\text{F}_3\text{N}_3\text{O}_2]^+$: 462.1424; found: 462.1428.



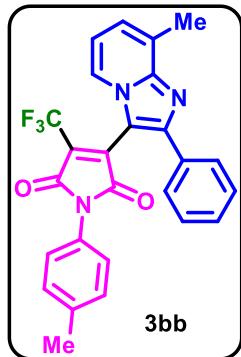
3-(7-Methoxy-2-(*p*-tolyl)imidazo[1,2-*a*]pyridin-3-yl)-1-phenyl-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3na): Purple solid (76.3 mg, 80%); M.p. 195-196 °C; $R_f = 0.45$ (PE : EA = 75 : 25); ^1H NMR (CDCl_3 , 400 MHz): δ 7.85 (d, $J = 7.6$ Hz, 1H), 7.49 (d, $J = 7.6$ Hz, 2H), 7.44-7.40 (m, 2H), 7.36-7.29 (m, 3H), 7.22 (d, $J = 8.0$ Hz, 2H), 6.98-6.97 (m, 1H), 6.65-6.61 (m, 1H), 3.87 (d, $J = 3.6$ Hz, 3H), 2.36 (s, 3H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 165.7, 164.5, 160.3, 153.7, 150.8, 139.4, 134.5, 130.9, 130.7, 129.6, 129.3, 128.55, 128.50, 126.1, 125.9, 123.0 (q, $J_{\text{C}-\text{F}} = 271.0$ Hz), 121.5, 108.5, 107.4, 95.6, 55.9, 21.5; ^{19}F NMR (CDCl_3 , 376 MHz): δ -60.96; HRMS (ESI-TOF) m/z: $[\text{M} + \text{H}]^+$ Calcd for $[\text{C}_{26}\text{H}_{19}\text{F}_3\text{N}_3\text{O}_3]^+$: 478.1373; found: 478.1375.



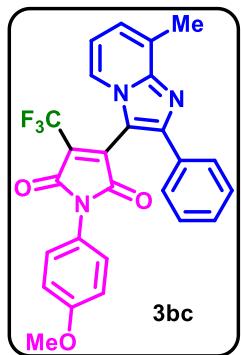
3-(2-(Naphthalen-2-yl)imidazo[1,2-*a*]pyridin-3-yl)-1-phenyl-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3oa): Red solid (55.1 mg, 57%); M.p. 215-216 °C; $R_f = 0.50$ (PE : EA = 70 : 30); ^1H NMR (CDCl_3 , 400 MHz): δ 8.18 (s, 1H), 8.10 (d, $J = 6.8$ Hz, 1H), 7.94-7.86 (m, 3H), 7.82 (d, $J = 9.2$ Hz, 1H), 7.75-7.73 (m, 1H), 7.54-7.51 (m, 2H), 7.49-7.45 (m, 1H), 7.43-7.41 (m, 2H), 7.37 (d, $J = 7.2$ Hz, 1H), 7.28 (d, $J = 7.6$ Hz, 2H), 7.03 (t, $J = 7.2$ Hz, 1H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 165.4, 164.3, 152.3, 148.7, 134.8, 133.6, 133.4, 131.0, 130.5, 129.4, 128.8, 128.6, 128.5, 128.4, 128.3, 127.9, 126.9, 126.7, 125.94, 125.91, 125.4, 124.1, 120.1 (q, $J_{\text{C}-\text{F}} = 271$ Hz), 118.3, 114.1, 108.3; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.33; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for $[\text{C}_{28}\text{H}_{17}\text{F}_3\text{N}_3\text{O}_2]^+$: 484.1267; found: 484.1262.



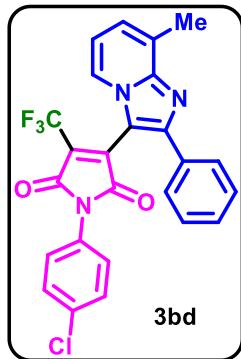
3-(6-Fluoro-2-phenylimidazo[1,2-*a*]pyridin-3-yl)-1-phenyl-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3pa): Red solid (49.6 mg, 55%); M.p. 187-188 °C; $R_f = 0.50$ (PE : EA = 80 : 20); ^1H NMR (CDCl_3 , 400 MHz): δ 8.06 (s, 1H), 7.63 (s, 2H), 7.47-7.36 (m, 10H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 165.1, 163.8, 152.8, 151.1, 144.9, 139.3, 136.8, 133.9, 131.7, 130.3, 129.9 (d, $J_{\text{C}-\text{F}} = 10.0$ Hz), 129.4, 129.2, 128.7, 128.5, 128.4 (d, $J_{\text{C}-\text{F}} = 7.0$ Hz), 126.0, 122.4 (q, $J_{\text{C}-\text{F}} = 277.0$ Hz), 113.4 (d, $J_{\text{C}-\text{F}} = 34.0$ Hz), 109.0; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.68, -136.31; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for $[\text{C}_{24}\text{H}_{14}\text{F}_4\text{N}_3\text{O}_2]^+$: 452.1017; found: 452.1011.



3-(8-Methyl-2-phenylimidazo[1,2-*a*]pyridin-3-yl)-1-(*p*-tolyl)-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3bb): Red solid (68.2 mg, 74%); M.p. 150-151 °C; $R_f = 0.45$ (PE : EA = 92 : 8); ^1H NMR (CDCl_3 , 400 MHz): δ 7.93 (d, $J = 7.2$ Hz, 1H), 7.67-7.65 (m, 2H), 7.48-7.41 (m, 3H), 7.27-7.19 (m, 5H), 6.89 (t, $J = 6.8$ Hz, 1H), 2.71 (s, 3H), 2.37 (s, 3H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 165.6, 164.4, 152.0, 149.0, 138.7, 134.9, 133.8, 130.0, 129.1, 128.9, 128.8, 128.5, 126.9, 126.3, 125.8, 123.7, 123.3, 120.1 (q, $J_{\text{C}-\text{F}} = 271.0$ Hz), 113.9, 108.5, 21.2, 17.2; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.46; HRMS (ESI-TOF) m/z: [M + H] $^+$ Calcd for $[\text{C}_{26}\text{H}_{19}\text{F}_3\text{N}_3\text{O}_2]^+$: 462.1424; found: 462.1427.

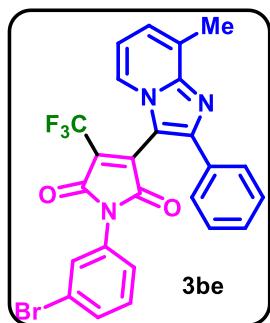


1-(4-Methoxyphenyl)-3-(8-methyl-2-phenylimidazo[1,2-*a*]pyridin-3-yl)-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3bc): Red solid (71.6 mg, 75%); M.p. 140-141 °C; $R_f = 0.50$ (PE : EA = 80 : 20); ^1H NMR (CDCl_3 , 400 MHz): δ 7.93 (d, $J = 6.8$ Hz, 1H), 7.66-7.64 (m, 2H), 7.47-7.40 (m, 3H), 7.23-7.20 (m, 3H), 6.97-6.94 (m, 2H), 6.89 (t, $J = 7.2$ Hz, 1H), 3.81 (s, 3H), 2.71 (s, 3H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): 165.8, 164.6, 159.5, 152.0, 149.0, 134.9, 133.8, 129.2, 128.9, 128.8, 128.5, 127.6, 127.4, 126.9, 123.3, 123.1, 120.1 (q, $J_{\text{C}-\text{F}} = 271.0$ Hz), 114.6, 113.9, 108.6, 55.6, 17.2; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.45; HRMS (ESI-TOF) m/z: [M + H] $^+$ Calcd for $[\text{C}_{26}\text{H}_{19}\text{F}_3\text{N}_3\text{O}_3]^+$: 478.1373; found: 478.1378.



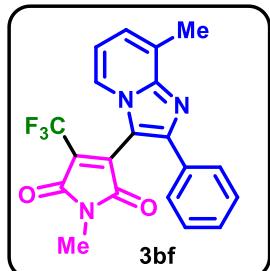
1-(4-Chlorophenyl)-3-(8-methyl-2-phenylimidazo[1,2-*a*]pyridin-3-yl)-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3bd):

Red solid (65.5 mg, 68%); M.p. 123-124 °C; $R_f = 0.50$ (PE : EA = 85 : 15); ^1H NMR (CDCl_3 , 400 MHz): δ 7.93 (d, $J = 6.8$ Hz, 1H), 7.65-7.63 (m, 2H), 7.48-7.41 (m, 5H), 7.30-7.24 (m, 3H), 6.93 (t, $J = 6.8$ Hz, 1H), 2.73 (s, 3H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 165.2, 164.0, 152.0, 148.9, 134.98, 134.96, 134.4, 133.5, 129.6, 129.4, 129.2, 129.0, 128.9, 128.6, 127.4, 127.0, 123.3, 120.1 (q, $J_{\text{C}-\text{F}} = 271.0$ Hz), 114.3, 108.5, 17.3; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.42; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for $[\text{C}_{25}\text{H}_{16}^{35}\text{ClF}_3\text{N}_3\text{O}_2]^+$: 482.0878; found: 482.0896.

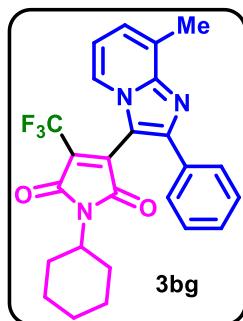


1-(3-Bromophenyl)-3-(8-methyl-2-phenylimidazo[1,2-*a*]pyridin-3-yl)-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3be):

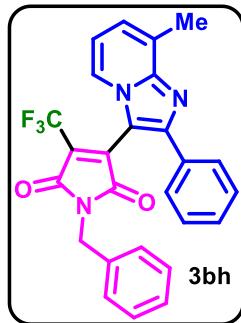
Red solid (63.1 mg, 60%); M.p. 160-161 °C; $R_f = 0.45$ (PE : EA = 90 : 10); ^1H NMR (CDCl_3 , 400 MHz): δ 7.93 (d, $J = 6.8$ Hz, 1H), 7.65-7.63 (m, 2H), 7.54-7.50 (m, 2H), 7.48-7.42 (m, 3H), 7.35-7.28 (m, 2H), 7.24 (s, 1H), 6.93 (t, $J = 7.2$ Hz, 1H), 2.72 (s, 3H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 165.1, 163.9, 149.0, 134.9, 133.6, 131.8, 131.6, 130.6, 129.5, 129.4, 129.0, 128.9, 128.6, 127.8, 127.3, 125.0, 124.4, 123.3, 122.6, 119.8 (q, $J_{\text{C}-\text{F}} = 271.0$ Hz), 114.2, 108.5, 17.3; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.43; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for $[\text{C}_{25}\text{H}_{16}^{79}\text{BrF}_3\text{N}_3\text{O}_2]^+$: 526.0373; found: 526.0377.



1-Methyl-3-(8-methyl-2-phenylimidazo[1,2-*a*]pyridin-3-yl)-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3bf): Red solid (56.2 mg, 73%); M.p. 152-153 °C; $R_f = 0.40$ (PE : EA = 90 : 10); ^1H NMR (CDCl_3 , 400 MHz): δ 7.85 (d, $J = 6.8$ Hz, 1H), 7.60-7.58 (m, 2H), 7.44-7.39 (m, 3H), 7.19 (d, $J = 6.8$ Hz, 1H), 6.87 (t, $J = 7.2$ Hz, 1H), 3.11 (s, 3H), 2.70 (s, 3H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 166.7, 165.6, 151.7, 148.9, 135.28, 135.23, 133.8, 129.1, 128.9, 128.8, 128.5, 126.7, 123.2, 120.0 (q, $J_{\text{C}-\text{F}} = 271.0$ Hz), 113.9, 108.4, 24.9, 17.2; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.77; HRMS (ESI-TOF) m/z: $[\text{M} + \text{H}]^+$ Calcd for $[\text{C}_{20}\text{H}_{15}\text{F}_3\text{N}_3\text{O}_2]^+$: 386.1111; found: 386.1109.



1-Cyclohexyl-3-(8-methyl-2-phenylimidazo[1,2-*a*]pyridin-3-yl)-4-(trifluoromethyl)-1*H*-pyrrole-2,5-dione (3bg): Red solid (67.1 mg, 74%); M.p. 110-111 °C; $R_f = 0.50$ (PE : EA = 90 : 10); ^1H NMR (CDCl_3 , 400 MHz): δ 7.83 (d, $J = 6.8$ Hz, 1H), 7.60-7.58 (m, 2H), 7.44-7.36 (m, 3H), 7.18 (d, $J = 6.8$ Hz, 1H), 6.86 (t, $J = 7.2$ Hz, 1H), 4.04-3.93 (m, 1H), 2.69 (s, 3H), 2.07-1.91 (m, 2H), 1.83-1.80 (m, 2H), 1.69-1.62 (m, 3H), 1.34-1.27 (m, 2H), 1.21-1.13 (m, 1H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 166.3, 165.5, 151.3, 148.7, 134.8, 133.9, 129.0, 128.8, 128.7, 128.4, 126.6, 123.9, 123.1, 120.2 (q, $J_{\text{C}-\text{F}} = 271.0$ Hz), 113.8, 108.5, 52.0, 29.8, 29.5, 25.88, 25.86, 24.8, 17.2; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.48; HRMS (ESI-TOF) m/z: $[\text{M} + \text{H}]^+$ Calcd for $[\text{C}_{25}\text{H}_{23}\text{F}_3\text{N}_3\text{O}_2]^+$: 454.1737; found: 454.1735.



1-Benzyl-3-(8-methyl-2-phenylimidazo[1,2-a]pyridin-3-yl)-4-(trifluoromethyl)-1H-pyrrole-2,5-dione (3bh): Red solid (68.2 mg, 74%); M.p. 158-159 °C; $R_f = 0.45$ (PE : EA = 90 : 10); ^1H NMR (CDCl_3 , 400 MHz): δ 7.81 (d, $J = 6.8$ Hz, 1H), 7.59-7.57 (m, 2H), 7.39-7.32 (m, 8H), 7.19 (d, $J = 6.8$ Hz, 1H), 6.87 (t, $J = 7.2$ Hz, 1H), 4.77-4.66 (m, 2H), 2.70 (s, 3H); $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 100 MHz): δ 166.3, 165.1, 151.7, 149.0, 135.4, 135.2, 133.9, 129.0, 128.96, 128.90, 128.8, 128.5, 128.3, 126.8, 123.8, 123.1, 120.1 (q, $J_{\text{C}-\text{F}} = 271.0$ Hz), 113.9, 113.8, 108.6, 42.5, 17.2; ^{19}F NMR (CDCl_3 , 376 MHz): δ -61.38; HRMS (ESI-TOF) m/z: [M + H] $^+$ Calcd for $[\text{C}_{26}\text{H}_{19}\text{F}_3\text{N}_3\text{O}_2]^+$: 462.1424; found: 462.1429.

6. Control Experiments:

The productivity of this reaction was not found at all in the presence of radical scavengers like 2,2,6,6-tetramethylpiperidine-1-oxyl (TEMPO), 2,6-di-*tert*-butyl-4-methyl phenol (BHT). Additionally, by employing *p*-benzoquinone (BQ) in the reaction, a trace amount of product was observed (Figure S1, eq A). Notably, the formation of radical-adduct (**4**) was detected by GC-MS in the presence of 1,1-DPE (Figure S1, eq B and ESI, S1 7). These observations clearly highlight that the reaction likely proceeds through a radical pathway.

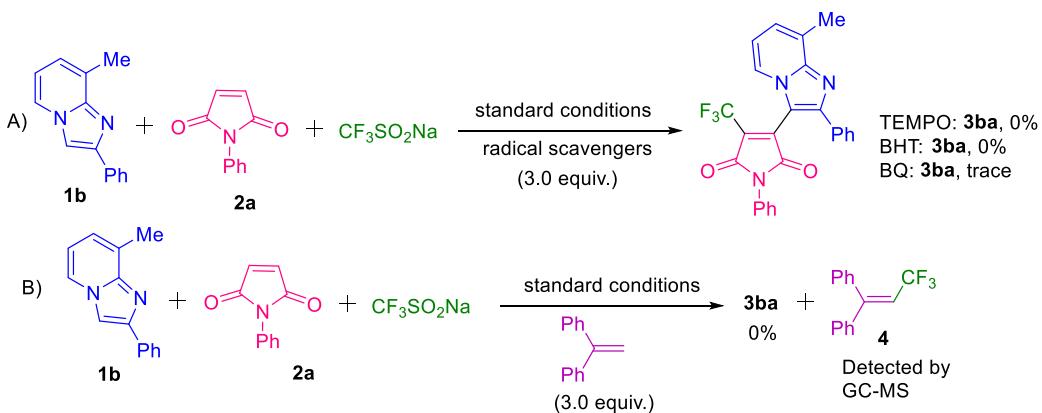
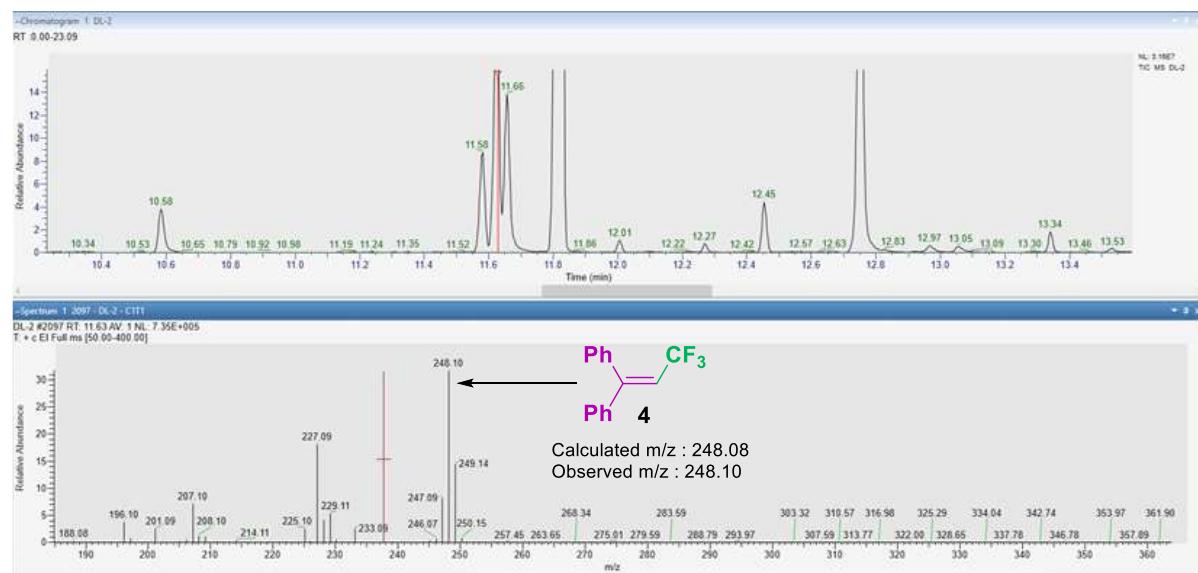


Figure: S1

7. GC-MS spectra of radical-adduct 4:



8. Computational Studies:

Geometry optimisation and energy calculations for all systems have been executed utilising Gaussian 16 software.³ The resultant data were visualised, and the progression of calculations was meticulously tracked with various integrated scripts and graphic tools, including Gauss View. The electronic structure calculations were done using the UB3LYP hybrid density functional method.⁴ The 6-31+G(d,p) basis set was utilized for all elements, with the LANL2DZ basis set specifically chosen for the iodine atom. Notably, unrestricted spin was taken into account for calculating the radical fragments. The influence of the bulk solvent was approximated by utilising the Polarization Continuum Model (PCM).⁵ As dichloroethane (DCE) was employed as the solvent in the experimental process, established solvent-specific parameters were meticulously applied during the computational calculations.

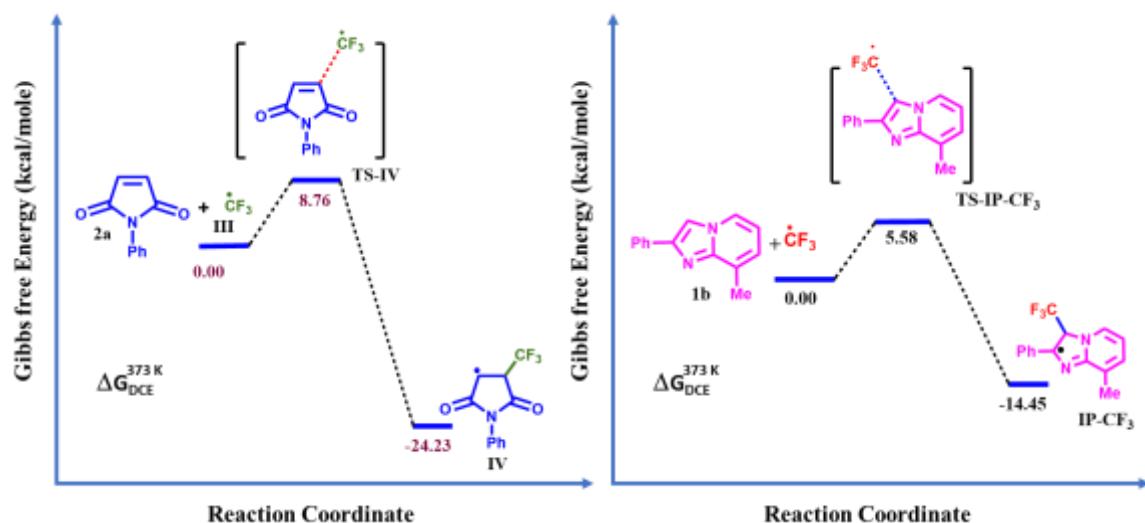


Figure 1: Potential energy surface for generated CF_3 maleimide radical intermediate **IV** and CF_3 imidazopyridine radical intermediate **IP-CF₃**.

St.Pt.	General Structure	Ball & Stick model				
2a + III	<p>2a + III</p>					
	Cartesian co-ordinate	Frequencies				
Atoms	X	Y	Z			
				8.7077	10.9563	16.5160
				27.4395	32.1755	42.4595
				46.1288	79.9807	109.1719
				150.7642	265.6715	284.4427
F	5.06483	-1.51440	-0.47176	289.6126	379.5341	415.4450
O	-2.55091	2.66738	-0.08826	463.5633	492.9081	494.1369
F	4.18569	0.48143	-0.24291	507.4806	587.9322	621.3115
O	0.92566	-0.35485	-0.12155	628.3370	633.6826	680.9610
F	3.97795	-0.99891	1.36117	694.4788	705.0926	720.0996
N	-1.09246	0.83274	-0.09230	761.5616	767.2098	838.0491
C	-0.13395	2.95648	-0.15732	843.5036	921.3780	963.1437
C	-1.43002	2.20392	-0.10874	974.2283	978.3447	1001.0964
C	0.87463	2.07982	-0.16787	1017.2564	1041.2980	1051.7157
C	0.30854	0.69175	-0.12644	1061.3125	1084.1651	1107.0871
C	-2.02984	-0.24607	-0.04704	1153.8202	1186.0317	1202.8888
C	4.05832	-0.81482	0.04901	1215.3536	1225.4573	1234.9062
C	-3.08878	-0.20761	0.86646	1330.8118	1345.6794	1361.8750
H	-3.19978	0.64027	1.53202	1406.6524	1490.5406	1535.6662
C	-1.88179	-1.33180	-0.91687	1637.7254	1649.4164	1654.9784
H	-1.05551	-1.35606	-1.61751	1777.3580	1834.7937	3185.6460
C	-3.86050	-2.35184	0.04129	3196.1170	3207.3004	3226.4083
H	-4.57254	-3.17091	0.07570	3227.4625	3250.1525	3270.7380
C	-4.00559	-1.25952	0.90042			
H	-4.83101	-1.22478	1.60502			
C	-2.79543	-2.38562	-0.86249			
H	-2.67500	-3.23091	-1.53329			
H	1.94209	2.25377	-0.19810			
H	-0.10877	4.03802	-0.18004			

Statistical Thermodynamic Analysis

Temperature=373.15 K

Pressure=1 atm

Zero-point correction= 0.160857

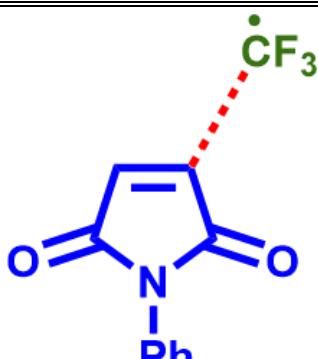
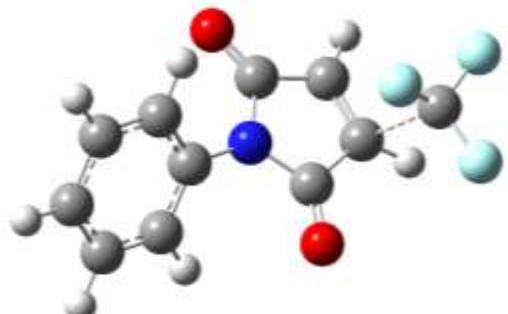
Electronic Energy = -928.093644723

Internal Energy (E)= -927.909635723

Enthalpy (H)= -927.908454723

Gibbs Free Energy (G)=-928.003931723

Gibbs Free Energy of Solvation=-928.013016551

St.Pt.	General Structure	Ball & Stick model				
TS-IV						
Cartesian co-ordinate		Frequencies				
Atoms	X	Y	Z	-223.6579	12.9585	29.4648
				37.0144	57.8892	88.1554
				102.4562	120.0199	129.1387
				153.0693	268.0145	284.4574
F	3.37976	-1.75682	0.37328	301.8135	383.3039	415.1145
O	-0.47708	2.71471	-0.07913	463.1334	496.8327	499.8923
F	3.99592	0.30956	0.75079	508.2710	591.0941	619.1579
O	0.34093	-1.65284	-1.31367	627.5235	640.0983	662.7620
F	2.17061	-0.47216	1.67448	697.1706	707.7923	721.4549
N	-0.44439	0.41550	-0.54358	752.6871	768.5493	825.6253
C	1.47797	1.63275	-1.03076	843.2621	920.0514	922.4525
C	0.09589	1.72352	-0.48394	960.3614	978.6297	1001.3097
C	1.75225	0.33640	-1.32085	1010.3650	1017.5782	1038.7750
C	0.50238	-0.47274	-1.08408	1051.3248	1071.6865	1106.7248
C	-1.76228	0.05418	-0.12119	1151.7778	1186.1283	1202.5570
C	2.96077	-0.51784	0.60435	1217.4976	1222.1808	1235.6626
C	-2.24349	0.50902	1.11137	1327.9333	1346.0123	1362.3588
H	-1.62117	1.13636	1.73834	1399.0419	1490.6583	1533.7549
C	-2.55970	-0.75007	-0.94274	1535.7526	1637.5169	1648.9310
H	-2.17633	-1.10364	-1.89238	1775.0820	1830.4571	3185.7911
				3196.4014	3207.5138	3228.2410

C	-4.33368	-0.64870	0.70562		3229.2821	3244.0723	3265.2093
H	-5.33405	-0.92220	1.02737				
C	-3.53281	0.16024	1.51599				
H	-3.90737	0.51902	2.46988				
C	-3.84204	-1.10420	-0.52036				
H	-4.45746	-1.73392	-1.15571				
H	2.58190	-0.06627	-1.88735				
H	2.10625	2.50654	-1.13999				

Statistical Thermodynamic Analysis

Temperature=373.15 K

Pressure=1 atm

Zero-point correction= 0.160718

Electronic Energy = -928.086799116

Internal Energy (E)= -927.904513116

Enthalpy (H)= -927.903332116

Gibbs Free Energy (G)=-927.988701116 Gibbs Free Energy of Solvation=-927.999062591

St.Pt.	General Structure			Ball & Stick model		
IV						
<u>Cartesian co-ordinate</u>				<u>Frequencies</u>		
Atoms	X	Y	Z			
F	3.34522	-1.61105	0.27009	33.2174	41.4952	53.6343
O	-0.56343	2.62389	-0.51267	76.7865	111.8773	139.3383
F	4.08561	0.42622	0.06053	161.3490	247.7525	266.2823
O	0.38847	-1.89995	-0.29758	273.4620	315.1406	334.7365
F	2.57646	-0.03603	1.56857	388.6575	412.9861	472.6318
N	-0.43677	0.28033	-0.33141	500.7311	517.2110	530.8678
C	1.51236	1.44112	-0.76889	535.2749	582.4381	611.5025
C	0.09184	1.59435	-0.53422	628.4519	660.6114	688.1634
C	1.88924	0.00250	-0.72049	697.4095	710.5244	729.8511
C	0.54863	-0.70611	-0.41670	767.3343	796.3255	842.2694
				855.9367	924.6466	956.6881
				979.3642	1002.8481	1017.1785
				1040.6580	1050.6256	1059.5237
				1111.1407	1120.3973	1166.3240

C	-1.82017	0.00957	-0.08119	1175.8280	1187.1632	1202.8249
C	2.97842	-0.31952	0.30856	1215.0471	1244.9477	1269.0831
C	-2.51963	0.79013	0.84624	1305.4003	1347.5032	1349.8157
H	-2.01847	1.59838	1.36441	1363.4977	1380.3395	1491.1100
C	-2.46086	-1.03012	-0.76505	1532.7357	1636.5304	1646.9514
H	-1.91011	-1.63589	-1.47426	1725.5736	1825.0838	3036.4656
C	-4.51616	-0.51243	0.40949	3187.5439	3198.5199	3209.0698
H	-5.56552	-0.71566	0.60060	3231.7088	3233.9280	3266.8205
C	-3.86879	0.52565	1.08449			
H	-4.41179	1.13330	1.80183			
C	-3.80848	-1.28830	-0.51234			
H	-4.30391	-2.09749	-1.04004			
H	2.27034	-0.36588	-1.68380			
H	2.17264	2.27613	-0.95892			

Statistical Thermodynamic Analysis

Temperature=373.15 K

Pressure=1 atm

Zero-point correction= 0.163293

Electronic Energy = -928.144653920

Internal Energy (E)= -927.96047192

Enthalpy (H)= -927.95929092

Gibbs Free Energy (G)=-928.03992992

Gibbs Free Energy of Solvation=-928.051626089

St.Pt.	General Structure	Ball & Stick model				
1b + III	<p>The general structure shows a substituted imidazo[1,2-b]pyridine ring system. It features a phenyl group (Ph) at position 2 and a methyl group (Me) at position 4. The reaction is with a trifluoromethyl radical (·CF3).</p>	<p>The ball-and-stick model illustrates the spatial arrangement of atoms in the reactant complex. Carbon atoms are represented by grey spheres, hydrogen atoms by small white spheres, and the nitrogen atoms by blue spheres. The trifluoromethyl radical is shown with its three fluorine atoms (cyan spheres) and central carbon atom (grey sphere).</p>				
Cartesian co-ordinate		Frequencies				
Atoms	X	Y	Z	7.0034	13.5465	15.9725
C	-1.52497	-1.14078	-0.02286	19.7101	26.1772	33.9452
C	-2.76687	-1.65085	0.44918	37.8828	68.6086	101.3131
C	-3.92055	-1.13982	-0.10854	121.4241	170.0364	188.0726
C	-3.87788	-0.13653	-1.11905	234.8406	261.4254	297.1023
C	-2.68053	0.34906	-1.56513	328.6377	394.4036	412.0698
				491.6932	492.5374	497.4423
				518.2035	533.5024	558.6179

H	-4.88466	-1.50814	0.22913	573.3778	614.0455	633.4955
H	-4.79544	0.25373	-1.54420	661.0242	664.9156	680.8146
H	-2.57270	1.11195	-2.32621	698.4981	719.9539	750.9075
N	-1.52184	-0.15132	-1.01997	753.1758	767.2725	791.2114
N	-0.28282	-1.45699	0.33543	855.6937	881.1814	891.9966
C	-0.19900	0.14872	-1.26849	933.3937	944.9569	952.1203
C	0.54428	-0.67613	-0.42687	984.8204	986.5358	1004.2768
C	2.00852	-0.76353	-0.31055	1014.0908	1016.4742	1049.5948
C	2.58462	-1.63211	0.63215	1063.9327	1084.4383	1097.8573
C	2.85858	0.00308	-1.12711	1104.9123	1112.1248	1179.4239
C	3.97092	-1.72785	0.75427	1183.6221	1204.1224	1204.4854
H	1.92941	-2.22479	1.26093	1223.0760	1235.5637	1273.4122
C	4.24424	-0.09356	-1.00316	1288.5648	1316.2249	1360.3224
H	2.43951	0.67919	-1.86690	1370.2369	1381.7207	1399.9452
C	4.80760	-0.95998	-0.06108	1425.2418	1471.2156	1477.5893
H	4.39936	-2.40401	1.48880	1482.3669	1499.9771	1513.0100
H	4.88448	0.50717	-1.64290	1535.0405	1571.3032	1595.3755
H	5.88683	-1.03528	0.03534	1626.9665	1653.7647	1671.6055
C	-2.75594	-2.70806	1.51752	3044.2654	3101.9126	3129.2630
H	-2.20056	-3.59188	1.18545	3176.1308	3183.5817	3193.3040
H	-2.25288	-2.34423	2.41999	3193.3816	3203.9789	3218.3941
H	-3.77388	-3.00879	1.77867	3220.3176	3237.5308	3288.2300
C	-0.11311	2.63858	0.68041			
F	-1.13392	2.40322	1.50271			
F	-0.44975	3.55720	-0.23226			
F	0.96267	3.02776	1.35957			
H	0.08272	0.86825	-2.02058			

Statistical Thermodynamic Analysis

Temperature=373.15 K Pressure=1 atm
 Zero-point correction= 0.237945 Electronic Energy = -987.853776661
 Internal Energy (E)= -987.588568661 Enthalpy (H)= -987.587386661
 Gibbs Free Energy (G)=-987.691652661 Gibbs Free Energy of Solvation=-987.699686476

St.Pt.	General Structure	Ball & Stick model	Cartesian co-ordinate			Frequencies		
			Atoms	X	Y	Z	-151.9895	10.8048
TS-IP-CF ₃							40.9625	29.4205
							49.6907	63.9248

C	-1.51371	-1.09089	-0.07455	90.0906	97.4975	113.2188
C	-2.74381	-1.71321	0.26442	119.0606	167.4150	194.0354
C	-3.90418	-1.11672	-0.19085	235.5470	253.0059	295.7120
C	-3.87561	0.07895	-0.96208	331.3041	394.7391	412.0256
C	-2.68400	0.67546	-1.27071	486.6681	487.9700	490.0815
H	-4.86270	-1.56800	0.04744	518.0475	523.1921	561.9413
H	-4.79737	0.53342	-1.30615	567.2181	614.8660	624.4226
H	-2.58448	1.59400	-1.83477	633.3108	661.4798	696.0217
N	-1.52351	0.08686	-0.83366	700.8381	749.3505	752.4767
N	-0.26338	-1.46340	0.21492	765.5543	789.0516	790.2576
C	-0.20202	0.49421	-0.95564	854.5600	887.0200	894.4824
C	0.55453	-0.53335	-0.34592	933.8993	942.5875	947.0770
C	2.01474	-0.63870	-0.27365	948.8826	985.5762	985.8064
C	2.60506	-1.77119	0.31725	1004.9038	1013.6670	1048.6913
C	2.85185	0.36663	-0.79135	1064.2323	1082.7639	1094.2292
C	3.99183	-1.89355	0.38249	1103.5679	1110.4570	1129.6040
H	1.95965	-2.54452	0.71880	1159.3890	1181.2227	1184.3545
C	4.23883	0.24045	-0.72536	1205.5651	1236.5460	1273.2610
H	2.42253	1.25884	-1.23642	1286.9960	1313.2587	1358.7519
C	4.81542	-0.89035	-0.13904	1366.1578	1376.1246	1399.9058
H	4.43222	-2.77406	0.84165	1425.2578	1460.7384	1476.6034
H	4.86956	1.02831	-1.12673	1478.1480	1492.8781	1499.2437
H	5.89595	-0.98713	-0.08667	1526.0460	1558.0638	1593.9500
C	-2.72197	-2.97167	1.08634	1624.6892	1649.8475	1665.6094
H	-2.13593	-3.75203	0.58928	3045.9196	3104.3569	3131.0865
H	-2.24726	-2.79385	2.05726	3179.3513	3187.8650	3194.8645
H	-3.73556	-3.34480	1.25362	3198.1714	3206.9453	3220.9139
C	-0.19414	2.31596	0.61930	3222.3945	3244.8688	3275.5252
F	-0.69118	1.92104	1.80332			
F	-1.01502	3.26371	0.10249			
F	1.01869	2.86624	0.81385			
H	0.08224	1.21443	-1.70686			

Statistical Thermodynamic Analysis

Temperature=373.15 K

Pressure=1 atm

Zero-point correction= 0.237722

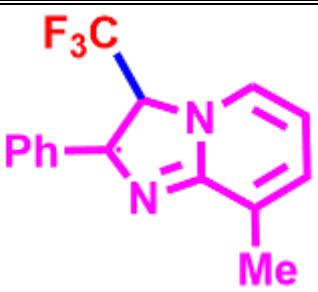
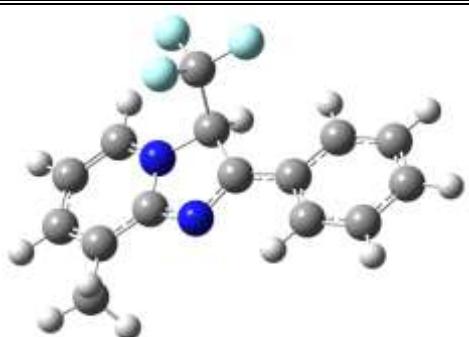
Electronic Energy = -987.852370110

Internal Energy (E)= -987.58885611

Enthalpy (H)= -987.58767411

Gibbs Free Energy (G)=-987.68211811

Gibbs Free Energy of Solvation=-987.690787183

St.Pt.	General Structure	Ball & Stick model				
IP-CF ₃						
	<u>Cartesian co-ordinate</u>	<u>Frequencies</u>				
Atoms	X	Y	Z			
C	-1.53362	-0.98932	-0.03061	40.1176	49.9608	60.4769
C	-2.74268	-1.72551	0.09979	62.7298	100.5241	113.2499
C	-3.91204	-1.11383	-0.30680	124.1980	176.7730	193.5846
C	-3.92323	0.21214	-0.84398	231.2985	234.0801	266.2759
C	-2.75712	0.91656	-0.93910	282.9604	292.2997	327.2430
H	-4.84900	-1.65703	-0.22607	399.0307	417.5894	421.7466
H	-4.84907	0.66824	-1.17308	486.8662	510.1859	515.4336
H	-2.68996	1.92783	-1.31830	526.8223	536.4464	569.9476
N	-1.59173	0.32795	-0.52086	615.9451	622.5159	631.5075
N	-0.28666	-1.38512	0.20508	653.8201	694.6264	695.1894
C	-0.23916	0.87888	-0.50904	712.9430	745.9600	766.3718
C	0.55151	-0.37364	-0.11391	772.3169	838.7338	843.5376
C	1.98362	-0.52748	-0.15932	884.9075	891.0797	899.0308
C	2.57340	-1.71717	0.34121	912.5701	934.5582	968.3344
C	2.84081	0.44649	-0.72662	977.9781	995.4208	1001.0669
C	3.94810	-1.90907	0.28428	1016.7564	1049.3775	1063.0261
H	1.92659	-2.47409	0.77033	1067.6106	1091.1091	1106.7433
C	4.21732	0.24085	-0.78532	1113.0211	1166.2298	1175.1983
H	2.43246	1.36830	-1.12605	1181.2192	1200.2902	1211.4717
C	4.78344	-0.93343	-0.27849	1240.8197	1247.0093	1251.4528
H	4.37633	-2.82592	0.68011	1274.4258	1336.3937	1336.6804
H	4.85173	1.00297	-1.22915	1361.1306	1364.3752	1370.3122
H	5.85745	-1.08764	-0.32112	1401.5049	1426.0673	1467.5261
C	-2.68611	-3.12404	0.64915	1476.7312	1479.2164	1490.5093
H	-2.02795	-3.75411	0.04140	1510.7773	1543.6349	1582.1338
H	-2.27394	-3.12847	1.66398	1602.6400	1628.2739	1644.3008
H	-3.68222	-3.57329	0.67262	1697.4101	3045.1372	3103.2872
C	-0.15422	2.04612	0.50711	3130.4803	3177.7313	3185.3783
F	-0.52325	1.67200	1.74646	3193.1793	3201.4628	3211.3466
F	-0.97598	3.06836	0.13414	3221.1403	3226.0141	3250.3797
H	0.03986	1.30419	-1.48517			

Statistical Thermodynamic Analysis

Temperature=373.15 K

Pressure=1 atm

Zero-point correction= 0.240163

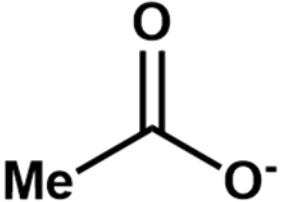
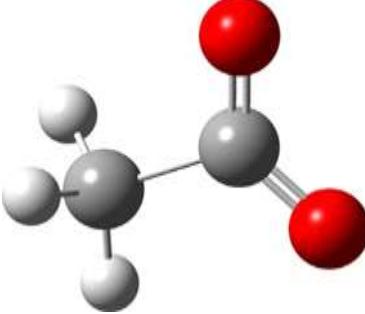
Electronic Energy = -987.892383235

Internal Energy (E)= -987.627128235

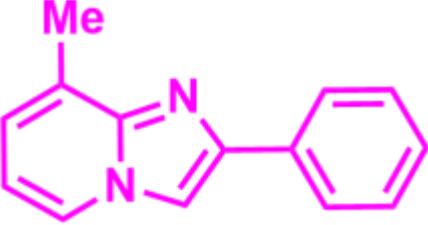
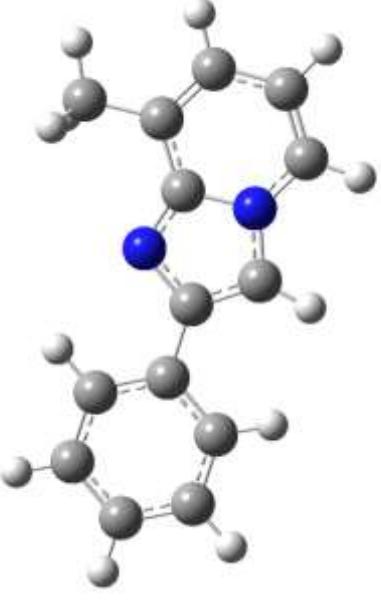
Enthalpy (H)= -987.625946235

Gibbs Free Energy (G)=-987.714673235

Gibbs Free Energy of Solvation=-987.722713249

St.Pt.	General Structure	Ball & Stick model																															
Acetate ion																																	
<u>Cartesian co-ordinate</u>		<u>Frequencies</u>																															
<table border="1"> <thead> <tr> <th>Atoms</th> <th>X</th> <th>Y</th> <th>Z</th> </tr> </thead> <tbody> <tr> <td>C</td> <td>-0.00028</td> <td>-1.35371</td> <td>0.00000</td> </tr> <tr> <td>H</td> <td>-0.48399</td> <td>-1.75537</td> <td>0.89495</td> </tr> <tr> <td>H</td> <td>1.04105</td> <td>-1.69996</td> <td>0.00000</td> </tr> <tr> <td>H</td> <td>-0.48399</td> <td>-1.75537</td> <td>-0.89495</td> </tr> <tr> <td>C</td> <td>-0.01115</td> <td>0.18745</td> <td>-0.00000</td> </tr> <tr> <td>O</td> <td>-0.00028</td> <td>0.76301</td> <td>-1.12926</td> </tr> <tr> <td>O</td> <td>-0.00028</td> <td>0.76301</td> <td>1.12926</td> </tr> </tbody> </table>		Atoms	X	Y	Z	C	-0.00028	-1.35371	0.00000	H	-0.48399	-1.75537	0.89495	H	1.04105	-1.69996	0.00000	H	-0.48399	-1.75537	-0.89495	C	-0.01115	0.18745	-0.00000	O	-0.00028	0.76301	-1.12926	O	-0.00028	0.76301	1.12926
Atoms	X	Y	Z																														
C	-0.00028	-1.35371	0.00000																														
H	-0.48399	-1.75537	0.89495																														
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C	-0.01115	0.18745	-0.00000																														
O	-0.00028	0.76301	-1.12926																														
O	-0.00028	0.76301	1.12926																														
<u>Statistical Thermodynamic Analysis</u>																																	
Temperature=373.15 K	Pressure=1 atm																																
Zero-point correction= 0.048358	Electronic Energy = -228.638898750																																
Internal Energy (E)= -228.58446375	Enthalpy (H)= -228.58328175																																
Gibbs Free Energy (G)= -228.62644775	Gibbs Free Energy of Solvation=-228.530426694																																

St.Pt.	General Structure	Ball & Stick model																																			
Acetic Acid																																					
<u>Cartesian co-ordinate</u>		<u>Frequencies</u>																																			
<table border="1"> <thead> <tr> <th>Atoms</th> <th>X</th> <th>Y</th> <th>Z</th> </tr> </thead> <tbody> <tr><td>C</td><td>1.39616</td><td>-0.12116</td><td>-0.00001</td></tr> <tr><td>H</td><td>1.92586</td><td>0.83051</td><td>-0.00028</td></tr> <tr><td>H</td><td>1.67670</td><td>-0.70460</td><td>-0.88204</td></tr> <tr><td>H</td><td>1.67668</td><td>-0.70400</td><td>0.88243</td></tr> <tr><td>C</td><td>-0.08735</td><td>0.12034</td><td>-0.00002</td></tr> <tr><td>O</td><td>-0.79111</td><td>-1.03517</td><td>0.00002</td></tr> <tr><td>O</td><td>-0.63301</td><td>1.20978</td><td>0.00001</td></tr> <tr><td>H</td><td>-1.73908</td><td>-0.81383</td><td>-0.00009</td></tr> </tbody> </table>		Atoms	X	Y	Z	C	1.39616	-0.12116	-0.00001	H	1.92586	0.83051	-0.00028	H	1.67670	-0.70460	-0.88204	H	1.67668	-0.70400	0.88243	C	-0.08735	0.12034	-0.00002	O	-0.79111	-1.03517	0.00002	O	-0.63301	1.20978	0.00001	H	-1.73908	-0.81383	-0.00009
Atoms	X	Y	Z																																		
C	1.39616	-0.12116	-0.00001																																		
H	1.92586	0.83051	-0.00028																																		
H	1.67670	-0.70460	-0.88204																																		
H	1.67668	-0.70400	0.88243																																		
C	-0.08735	0.12034	-0.00002																																		
O	-0.79111	-1.03517	0.00002																																		
O	-0.63301	1.20978	0.00001																																		
H	-1.73908	-0.81383	-0.00009																																		
<u>Statistical Thermodynamic Analysis</u>																																					
Temperature=373.15 K	Pressure=1 atm																																				
Zero-point correction= 0.061421	Electronic Energy = -229.113005988																																				
Internal Energy (E)= -229.045102988	Enthalpy (H)= -229.043921988																																				
Gibbs Free Energy (G)= -229.087420988	Gibbs Free Energy of Solvation= -229.080133806																																				

St.Pt.	General Structure	Ball & Stick model				
1b	 <p>1b</p>					
<u>Cartesian co-ordinate</u>						
Atoms	X	Y	Z	4.5664	66.8222	98.7844
				126.1191	167.5276	185.7034
				233.1027	263.1321	295.8058
				329.2325	392.9829	414.0278
N	-1.55859	-1.07318	-0.00223	501.1839	517.4345	536.6705
N	-0.35612	0.83043	0.00207	558.1522	574.7442	612.9205
C	-0.22935	-1.43829	-0.00296	631.9747	659.8748	678.6087
C	0.49194	-0.25209	-0.00026	701.9766	727.4434	749.0132
C	1.95601	-0.08772	-0.00006	755.3150	773.5618	796.3666
C	-2.70056	-1.84082	-0.00432	855.4013	890.2523	890.6279
H	-2.56627	-2.91511	-0.00697	933.3259	951.6478	955.0253
C	-1.59209	0.32980	0.00092	984.5697	988.2613	1005.9437
C	-2.85047	0.99740	0.00248	1011.0777	1046.3964	1060.9804
C	2.81826	-1.20045	0.00754	1084.1540	1096.4795	1101.4852
H	2.40947	-2.20638	0.01405	1109.2776	1177.9482	1179.4450
C	-2.88361	2.50100	0.00613	1199.8662	1235.4400	1276.2173
H	-2.37069	2.90821	-0.87238	1287.5531	1313.9404	1356.5599
H	-2.37093	2.90384	0.88679	1368.8820	1377.8249	1398.1511
H	-3.91378	2.86534	0.00689	1417.8880	1464.8161	1468.7152
C	-3.98683	0.21465	0.00048	1479.5425	1494.2093	1513.5945
H	-4.96242	0.69130	0.00156	1530.7618	1574.5195	1594.5275
C	2.52446	1.19878	-0.00762	1622.6037	1649.4062	1669.6825
H	1.86847	2.06232	-0.01372	3040.7400	3095.1763	3131.0563
C	-3.91240	-1.20870	-0.00298	3180.5580	3187.2012	3196.8176
H	-4.81734	-1.80517	-0.00460	3197.3058	3206.0134	3216.2595
C	4.20337	-1.03103	0.00747	3223.2914	3243.0896	3290.3759
H	4.85024	-1.90365	0.01355			

C	3.91067	1.36677	-0.00759
H	4.32943	2.36920	-0.01354
C	4.75788	0.25394	-0.00005
H	5.83599	0.38492	0.00002
H	0.06703	-2.47502	-0.00562

Statistical Thermodynamic Analysis

Temperature=373.15 K

Pressure=1 atm

Zero-point correction= 0.225944

Electronic Energy = -650.282146028

Internal Energy (E)= -650.036939028

Enthalpy (H)= -650.035757028

Gibbs Free Energy (G)=-650.112155028

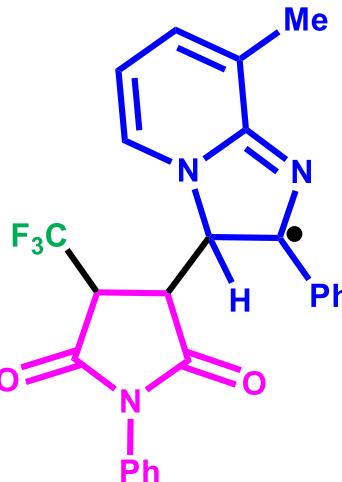
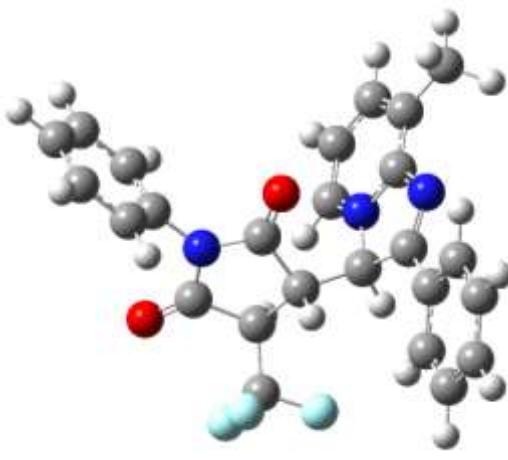
Gibbs Free Energy of Solvation=-650.10363081

St.Pt.	General Structure	Ball & Stick model																																																																																																																																				
II	$\text{Ph}-\overset{\bullet}{\underset{\text{OAc}}{\text{i}}}$																																																																																																																																					
Cartesian co-ordinate		Frequencies																																																																																																																																				
<table border="1" style="width: 100%; border-collapse: collapse;"> <thead> <tr> <th>Atoms</th> <th>X</th> <th>Y</th> <th>Z</th> <th></th> <th></th> <th></th> </tr> </thead> <tbody> <tr><td>C</td><td>-3.43343</td><td>0.79548</td><td>1.21188</td><td>16.8618</td><td>24.4084</td><td>32.9743</td></tr> <tr><td>C</td><td>-2.16460</td><td>0.20845</td><td>1.22137</td><td>45.9228</td><td>85.9184</td><td>91.0605</td></tr> <tr><td>C</td><td>-1.54847</td><td>-0.08319</td><td>0.00004</td><td>146.6395</td><td>176.0562</td><td>226.3260</td></tr> <tr><td>C</td><td>-2.16476</td><td>0.20774</td><td>-1.22138</td><td>253.8506</td><td>410.5271</td><td>432.6667</td></tr> <tr><td>C</td><td>-3.43359</td><td>0.79476</td><td>-1.21206</td><td>453.7175</td><td>611.6293</td><td>617.7218</td></tr> <tr><td>C</td><td>-4.06717</td><td>1.08914</td><td>-0.00014</td><td>619.1477</td><td>657.3856</td><td>680.8767</td></tr> <tr><td>H</td><td>-3.92164</td><td>1.02298</td><td>2.15464</td><td>746.5423</td><td>846.8819</td><td>901.2724</td></tr> <tr><td>H</td><td>-1.67386</td><td>-0.01732</td><td>2.16155</td><td>933.8484</td><td>991.6345</td><td>995.2729</td></tr> <tr><td>H</td><td>-1.67414</td><td>-0.01858</td><td>-2.16149</td><td>1009.6645</td><td>1013.9339</td><td>1024.0068</td></tr> <tr><td>H</td><td>-3.92193</td><td>1.02170</td><td>-2.15489</td><td>1042.6070</td><td>1068.2818</td><td>1099.5903</td></tr> <tr><td>H</td><td>-5.05204</td><td>1.54596</td><td>-0.00020</td><td>1182.1742</td><td>1202.1460</td><td>1248.1913</td></tr> <tr><td>I</td><td>0.37632</td><td>-0.98503</td><td>0.00015</td><td>1328.8207</td><td>1347.6985</td><td>1369.1912</td></tr> <tr><td>O</td><td>1.81284</td><td>1.27527</td><td>-0.00037</td><td>1456.4479</td><td>1470.7867</td><td>1473.4494</td></tr> <tr><td>C</td><td>3.02986</td><td>0.83013</td><td>-0.00028</td><td>1500.0783</td><td>1599.3258</td><td>1612.4737</td></tr> <tr><td>O</td><td>3.37179</td><td>-0.36083</td><td>-0.00013</td><td>1617.8166</td><td>3063.8230</td><td>3141.3196</td></tr> <tr><td>C</td><td>4.07880</td><td>1.94822</td><td>-0.00010</td><td>3170.4517</td><td>3196.3065</td><td>3204.3921</td></tr> <tr><td>H</td><td>4.70890</td><td>1.82384</td><td>-0.88531</td><td>3214.4623</td><td>3223.6012</td><td>3225.5666</td></tr> <tr><td>H</td><td>3.62840</td><td>2.94089</td><td>-0.00302</td><td></td><td></td><td></td></tr> </tbody> </table>		Atoms	X	Y	Z				C	-3.43343	0.79548	1.21188	16.8618	24.4084	32.9743	C	-2.16460	0.20845	1.22137	45.9228	85.9184	91.0605	C	-1.54847	-0.08319	0.00004	146.6395	176.0562	226.3260	C	-2.16476	0.20774	-1.22138	253.8506	410.5271	432.6667	C	-3.43359	0.79476	-1.21206	453.7175	611.6293	617.7218	C	-4.06717	1.08914	-0.00014	619.1477	657.3856	680.8767	H	-3.92164	1.02298	2.15464	746.5423	846.8819	901.2724	H	-1.67386	-0.01732	2.16155	933.8484	991.6345	995.2729	H	-1.67414	-0.01858	-2.16149	1009.6645	1013.9339	1024.0068	H	-3.92193	1.02170	-2.15489	1042.6070	1068.2818	1099.5903	H	-5.05204	1.54596	-0.00020	1182.1742	1202.1460	1248.1913	I	0.37632	-0.98503	0.00015	1328.8207	1347.6985	1369.1912	O	1.81284	1.27527	-0.00037	1456.4479	1470.7867	1473.4494	C	3.02986	0.83013	-0.00028	1500.0783	1599.3258	1612.4737	O	3.37179	-0.36083	-0.00013	1617.8166	3063.8230	3141.3196	C	4.07880	1.94822	-0.00010	3170.4517	3196.3065	3204.3921	H	4.70890	1.82384	-0.88531	3214.4623	3223.6012	3225.5666	H	3.62840	2.94089	-0.00302			
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C	4.07880	1.94822	-0.00010	3170.4517	3196.3065	3204.3921																																																																																																																																
H	4.70890	1.82384	-0.88531	3214.4623	3223.6012	3225.5666																																																																																																																																
H	3.62840	2.94089	-0.00302																																																																																																																																			

H	4.70463	1.82736	0.88865	
Statistical Thermodynamic Analysis				
Temperature=373.15 K		Pressure=1 atm		
Zero-point correction= 0.139209		Electronic Energy = -471.479851960		
Internal Energy (E)= -471.32304896		Enthalpy (H)= -471.32186696		
Gibbs Free Energy (G)= -471.40054796		Gibbs Free Energy of Solvation= -471.390118143		

St.Pt.	General Structure	Ball & Stick model																																																																																																																																											
PhI	PhI																																																																																																																																												
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Temperature=373.15 K		Pressure=1 atm																																																																																																																																											
Zero-point correction= 0.089876		Electronic Energy = -243.042537542																																																																																																																																											

Internal Energy (E)= -242.943828542 Enthalpy (H)= -242.942646542
 Gibbs Free Energy (G)=-242.993698542 Gibbs Free Energy of Solvation=-242.990574237

St.Pt.	General Structure	Ball & Stick model				
V						
Cartesian co-ordinate		Frequencies				
Atoms	X	Y	Z	17.0464	17.9415	31.0234
				38.5408	45.2839	53.7071
F	-1.01560	-3.32526	-2.78864	56.0052	57.1100	72.6274
O	-0.51894	0.84328	1.37103	80.9530	87.4428	93.3673
F	1.01780	-2.81035	-2.19834	114.6232	128.5130	150.8126
O	-3.11758	-1.83077	-1.31017	169.3934	198.4161	203.1475
F	-0.37800	-3.60429	-0.72249	210.1300	239.1631	256.8821
N	0.94661	1.61398	-1.30241	267.9829	288.4277	310.5119
N	-2.08362	-0.42084	0.21321	314.4177	336.0229	377.2680
N	2.46624	1.86236	0.37794	407.5424	412.0594	418.3232
C	0.18741	-0.66890	-0.42186	419.4886	466.5760	487.6172
C	-0.80268	0.03926	0.50824	497.7437	503.7304	511.4577
C	1.34510	0.23966	-0.91416	530.1258	530.6442	547.7264
C	2.36568	0.57802	0.17902	567.2813	606.5916	615.3715
C	-0.70662	-1.35973	-1.47517	621.1899	628.4306	629.5060
C	3.14463	-0.40737	0.91485	631.1001	653.0341	665.3837
C	0.11806	2.01298	-2.29678	689.2002	695.7355	704.6364
H	-0.33516	1.25601	-2.92093	723.8481	739.2704	761.5308
C	-2.12757	-1.27937	-0.88588	771.9751	785.2929	788.0734
C	-3.25436	-0.04381	0.96638	804.9577	839.7218	852.7699
C	1.61464	2.49875	-0.49757	858.5073	879.6238	888.7426
C	1.43029	3.88203	-0.63439	911.3968	912.9265	939.1396
C	-0.28253	-2.78859	-1.79803	939.7189	957.5277	966.8463
C	3.33557	-1.71458	0.42178	990.5872	1000.9888	1001.5572
H	2.91746	-2.01622	-0.53290	1010.5603	1011.3733	1015.8487
C	2.16441	4.84054	0.26037	1016.3053	1022.7133	1034.7622
H	3.24685	4.71414	0.16040	1047.0601	1055.6486	1058.3515
H	1.91237	4.66292	1.31041	1067.8918	1084.5778	1096.6522
				1102.1680	1113.8222	1117.3836

H	1.90811	5.87113	0.00902	1122.5593	1153.1003	1183.7687
C	0.55118	4.28884	-1.63928	1186.9642	1188.3773	1199.0472
H	0.37471	5.34932	-1.78797	1202.8326	1207.0771	1216.1055
C	3.73878	-0.03726	2.14039	1223.5644	1236.4575	1253.4874
H	3.58518	0.96608	2.52134	1260.3971	1277.0841	1291.8747
C	-0.09786	3.36416	-2.47336	1294.9493	1311.6935	1335.6130
H	-0.76219	3.69357	-3.26229	1338.4862	1351.1921	1356.3031
C	-3.55508	-0.72726	2.14591	1356.8727	1365.9654	1370.8731
H	-2.91260	-1.53279	2.48645	1398.8094	1403.1061	1420.9239
C	-4.06309	0.99337	0.49824	1457.8668	1472.1518	1479.4662
H	-3.80778	1.50888	-0.42197	1486.7877	1495.2276	1520.1288
C	4.10436	-2.62868	1.13965	1528.8071	1532.3245	1580.4694
H	4.25510	-3.62906	0.74764	1617.5776	1625.8338	1637.3905
C	-5.51029	0.67746	2.41507	1642.0163	1643.6651	1662.1116
H	-6.39258	0.95913	2.98136	1762.9075	1845.2291	3057.1968
C	-4.69027	-0.35914	2.87202	3073.5179	3092.6959	3113.6003
H	-4.93294	-0.88383	3.79052	3119.0694	3150.7694	3194.8194
C	4.49812	-0.95772	2.85460	3198.3375	3202.3510	3205.4144
H	4.94484	-0.66947	3.80063	3208.3708	3211.9445	3214.9940
C	4.68238	-2.25434	2.35684	3215.9161	3220.0327	3220.3456
H	5.27644	-2.97013	2.91645	3227.1721	3245.5732	3270.1260
C	-5.19759	1.35168	1.23033			
H	-5.83389	2.15583	0.87488			
H	-0.72803	-0.83001	-2.43098			
H	0.65642	-1.44110	0.19920			
H	1.84127	-0.23385	-1.76745			

Statistical Thermodynamic Analysis

Temperature=373.15 K

Pressure=1 atm

Zero-point correction= 0.395871

Electronic Energy = -1578.30912445

Internal Energy (E)= -1577.87201745

Enthalpy (H)= -1577.87083545

Gibbs Free Energy (G)=-1577.99844345

Gibbs Free Energy of Solvation=-1577.92691342

St.Pt.	General Structure	Ball & Stick model
VI		
<u>Cartesian co-ordinate</u>		<u>Frequencies</u>

Atoms	X	Y	Z			
F	-1.01560	-3.32526	-2.78864	17.0464	17.9415	31.0234
O	-0.51894	0.84328	1.37103	38.5408	45.2839	53.7071
F	1.01780	-2.81035	-2.19834	56.0052	57.1100	72.6274
O	-3.11758	-1.83077	-1.31017	80.9530	87.4428	93.3673
F	-0.37800	-3.60429	-0.72249	114.6232	128.5130	150.8126
N	0.94661	1.61398	-1.30241	169.3934	198.4161	203.1475
N	-2.08362	-0.42084	0.21321	210.1300	239.1631	256.8821
N	2.46624	1.86236	0.37794	267.9829	288.4277	310.5119
C	0.18741	-0.66890	-0.42186	314.4177	336.0229	377.2680
C	-0.80268	0.03926	0.50824	407.5424	412.0594	418.3232
C	1.34510	0.23966	-0.91416	419.4886	466.5760	487.6172
C	2.36568	0.57802	0.17902	497.7437	503.7304	511.4577
C	-0.70662	-1.35973	-1.47517	530.1258	530.6442	547.7264
C	3.14463	-0.40737	0.91485	567.2813	606.5916	615.3715
C	0.11806	2.01298	-2.29678	621.1899	628.4306	629.5060
H	-0.33516	1.25601	-2.92093	631.1001	653.0341	665.3837
C	-2.12757	-1.27937	-0.88588	689.2002	695.7355	704.6364
C	-3.25436	-0.04381	0.96638	723.8481	739.2704	761.5308
C	1.61464	2.49875	-0.49757	771.9751	785.2929	788.0734
C	1.43029	3.88203	-0.63439	804.9577	839.7218	852.7699
C	-0.28253	-2.78859	-1.79803	858.5073	879.6238	888.7426
C	3.33557	-1.71458	0.42178	911.3968	912.9265	939.1396
H	2.91746	-2.01622	-0.53290	939.7189	957.5277	966.8463
C	2.16441	4.84054	0.26037	990.5872	1000.9888	1001.5572
H	3.24685	4.71414	0.16040	1010.5603	1011.3733	1015.8487
H	1.91237	4.66292	1.31041	1016.3053	1022.7133	1034.7622
H	1.90811	5.87113	0.00902	1047.0601	1055.6486	1058.3515
C	0.55118	4.28884	-1.63928	1067.8918	1084.5778	1096.6522
H	0.37471	5.34932	-1.78797	1102.1680	1113.8222	1117.3836
C	3.73878	-0.03726	2.14039	1122.5593	1153.1003	1183.7687
H	3.58518	0.96608	2.52134	1186.9642	1188.3773	1199.0472
C	-0.09786	3.36416	-2.47336	1202.8326	1207.0771	1216.1055
H	-0.76219	3.69357	-3.26229	1223.5644	1236.4575	1253.4874
C	-3.55508	-0.72726	2.14591	1260.3971	1277.0841	1291.8747
H	-2.91260	-1.53279	2.48645	1294.9493	1311.6935	1335.6130
C	-4.06309	0.99337	0.49824	1338.4862	1351.1921	1356.3031
H	-3.80778	1.50888	-0.42197	1356.8727	1365.9654	1370.8731
C	4.10436	-2.62868	1.13965	1398.8094	1403.1061	1420.9239
H	4.25510	-3.62906	0.74764	1457.8668	1472.1518	1479.4662
C	-5.51029	0.67746	2.41507	1486.7877	1495.2276	1520.1288
H	-6.39258	0.95913	2.98136	1528.8071	1532.3245	1580.4694
C	-4.69027	-0.35914	2.87202	1617.5776	1625.8338	1637.3905
H	-4.93294	-0.88383	3.79052	1642.0163	1643.6651	1662.1116
C	4.49812	-0.95772	2.85460	1762.9075	1845.2291	3057.1968
H	4.94484	-0.66947	3.80063	3073.5179	3092.6959	3113.6003
C	4.68238	-2.25434	2.35684	3119.0694	3150.7694	3194.8194
H	5.27644	-2.97013	2.91645	3198.3375	3202.3510	3205.4144
C	-5.19759	1.35168	1.23033	3208.3708	3211.9445	3214.9940
H	-5.83389	2.15583	0.87488	3215.9161	3220.0327	3220.3456
H	-0.72803	-0.83001	-2.43098	3227.1721	3245.5732	3270.1260
H	0.65642	-1.44110	0.19920			
H	1.84127	-0.23385	-1.76745			

Statistical Thermodynamic Analysis

Temperature=373.15 K

Pressure=1 atm

Zero-point correction= 0.395871

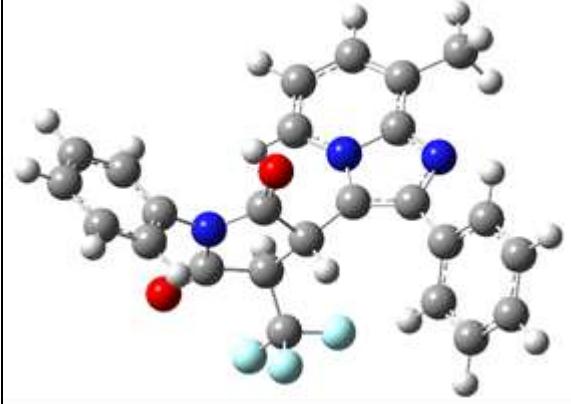
Electronic Energy = -1578.30912445

Internal Energy (E)= -1577.87201745

Enthalpy (H)= -1577.87083545

Gibbs Free Energy (G)= -1577.99844345

Gibbs Free Energy of Solvation=-1577.92691342

St.Pt.	General Structure	Ball & Stick model		
VII				
	Cartesian co-ordinate	Frequencies		
Atoms	X	Y	Z	
F	1.00589	-1.64625	3.63644	15.4306
O	1.03591	0.36514	-1.92744	33.0457
F	-0.94531	-1.67629	2.67840	58.9741
O	3.19702	-0.66622	1.98694	77.3776
F	0.69405	-2.84774	1.84336	127.4160
N	-1.26857	1.76565	0.29895	175.6950
N	2.39763	-0.11227	-0.11432	225.0018
N	-3.29569	1.12804	-0.44984	267.1813
C	0.05081	-0.40893	0.19482	319.6657
C	1.17153	0.01413	-0.77510	397.0783
C	-1.20192	0.39419	0.07185	433.2583
C	-2.46744	0.03963	-0.39681	504.6737
C	0.77382	-0.47436	1.57170	527.3246
C	-2.94665	-1.27871	-0.86680	584.7264
C	-0.31881	2.67252	0.71662	623.2764
H	0.67995	2.30862	0.90946	633.9851
C	2.27300	-0.45685	1.22979	662.1599
C	3.66976	0.08393	-0.76198	688.9116
C	-2.57290	2.16478	-0.02401	726.0471
C	-2.95550	3.52732	0.12459	759.5621
C	0.39031	-1.66613	2.43665	788.2259
C	-2.69953	-2.46847	-0.16091	853.8543
H	-2.17219	-2.43807	0.78548	890.1600
C	-4.36405	3.93169	-0.21170	939.0650
H	-5.08741	3.38316	0.40147	975.8142
H	-4.59815	3.70454	-1.25757	1009.4887
H	-4.50809	5.00207	-0.04729	1016.3378

C	-1.99634	4.41500	0.56604	1184.9863	1190.0958	1197.9191
H	-2.25167	5.46301	0.68824	1200.2102	1204.1503	1215.8181
C	-3.69507	-1.34502	-2.05669	1236.8043	1253.6281	1264.8155
H	-3.89776	-0.43029	-2.60424	1276.7193	1284.0482	1317.2725
C	-0.67154	3.98580	0.85487	1334.6382	1337.7425	1356.3100
H	0.07798	4.69576	1.18432	1358.8392	1371.6043	1382.2795
C	4.24089	-0.96946	-1.47877	1391.8354	1398.4108	1413.9493
H	3.72853	-1.92416	-1.53996	1418.9106	1463.3804	1469.5923
C	4.30514	1.32336	-0.66405	1477.0853	1485.8614	1491.4815
H	3.84117	2.12807	-0.10298	1521.1217	1528.8239	1529.7419
C	-3.17002	-3.69361	-0.64292	1574.3261	1595.4444	1621.2377
H	-2.97405	-4.60138	-0.07976	1636.7613	1642.7141	1649.8851
C	6.12016	0.46136	-2.01736	1670.5539	1756.0064	1836.6557
H	7.07753	0.60878	-2.50767	3043.6611	3072.0709	3097.7963
C	5.47231	-0.77469	-2.10908	3099.4301	3133.9128	3184.1173
H	5.92335	-1.58798	-2.66872	3192.5876	3193.1161	3201.1645
C	-4.16614	-2.56825	-2.53514	3201.3724	3203.9611	3207.7201
H	-4.73735	-2.60039	-3.45845	3211.8525	3214.5722	3219.2179
C	-3.90128	-3.74916	-1.83228	3227.3768	3227.4385	3264.0720
H	-4.26739	-4.70124	-2.20499			
C	5.53758	1.50794	-1.29552			
H	6.03891	2.46797	-1.22388			
H	0.56706	0.41352	2.17774			
H	-0.19794	-1.42884	-0.12129			
H	1.84127	-0.23385	-1.76745			

Statistical Thermodynamic Analysis

Temperature=373.15 K

Pressure=1 atm

Zero-point correction= 0.382956

Electronic Energy = -1577.88607903

Internal Energy (E)= -1577.46212303

Enthalpy (H)= -1577.46094203

Gibbs Free Energy (G)=-1577.58816603

Gibbs Free Energy of Solvation=-1577.56736877

St.Pt.	General Structure	Ball & Stick model				
3ba						
<u>Cartesian co-ordinate</u>		<u>Frequencies</u>				
Atoms	X	Y	Z	18.0711	20.5340	31.9157
				36.2504	48.2713	51.6377
				55.3903	64.7396	71.6857

F	0.26656	-3.78911	1.17537	83.5022	107.5791	121.0982
O	-1.14825	1.12645	-1.35398	125.9346	143.3735	161.0673
F	1.50511	-2.15404	1.91841	177.2778	196.3227	214.2103
O	-2.71244	-2.47840	1.01310	226.0767	239.0952	260.7668
F	-0.40278	-2.59400	2.86860	270.0745	287.6936	305.8162
N	2.44051	-0.73185	-0.55189	323.1058	341.2687	347.6496
N	-2.30243	-0.54904	-0.23005	384.1388	415.5310	419.0912
N	3.14601	1.40539	-0.37647	425.3715	443.2829	493.2453
C	0.02312	-0.48627	0.08858	497.9293	508.7027	521.5025
C	-1.17453	0.18302	-0.59132	524.0559	538.1320	552.2547
C	1.34747	0.02926	-0.12150	578.4646	591.3437	626.1537
C	1.83845	1.34844	-0.02583	627.3327	628.8943	635.8297
C	-0.46602	-1.57158	0.74846	640.8805	666.0559	699.3292
C	1.13705	2.54270	0.47487	702.9145	705.9662	722.7041
C	2.54197	-2.05592	-0.90153	737.6734	759.5559	762.6714
H	1.65930	-2.66978	-0.79636	769.2632	775.0600	785.4821
C	-1.94202	-1.64823	0.56712	792.0683	807.0306	837.6265
C	-3.63587	-0.26806	-0.67655	851.8828	858.6350	861.0688
C	3.50484	0.15691	-0.71172	895.6238	906.0090	936.2579
C	4.75704	-0.31112	-1.18850	939.6846	967.1284	968.8684
C	0.22625	-2.52116	1.67104	985.8107	988.7481	991.2592
C	0.19701	2.46156	1.51719	1009.9994	1011.6682	1012.1386
H	-0.01152	1.50613	1.98851	1015.8029	1038.8966	1047.6959
C	5.89605	0.65890	-1.33931	1055.5836	1063.1947	1084.8260
H	6.13783	1.13248	-0.38172	1089.8710	1099.1955	1102.1008
H	5.63633	1.46324	-2.03597	1105.7350	1135.0307	1155.3195
H	6.78794	0.15023	-1.71202	1168.0890	1180.2402	1181.3632
C	4.85219	-1.65261	-1.50496	1189.2722	1195.1823	1198.7455
H	5.79307	-2.04900	-1.87356	1200.7973	1234.8076	1283.5646
C	1.43699	3.80286	-0.07285	1286.6807	1322.9082	1338.1879
H	2.16838	3.87201	-0.87128	1342.7872	1356.6034	1358.0448
C	3.73872	-2.52308	-1.37205	1362.3959	1382.7104	1407.9110
H	3.82121	-3.56912	-1.64206	1418.8771	1437.1101	1461.5794
C	-4.28266	0.88531	-0.22445	1470.2506	1477.0996	1486.1197
H	-3.77687	1.55420	0.46406	1492.1559	1501.6582	1523.6361
C	-4.26898	-1.14870	-1.55801	1531.4848	1548.9996	1598.8301
H	-3.74936	-2.03656	-1.90282	1620.7836	1635.2812	1643.2755
C	-0.44574	3.61065	1.98330	1647.0821	1654.6329	1668.2901
H	-1.16453	3.53235	2.79353	1746.6240	1815.3573	3045.6890
C	-6.22372	0.28327	-1.54378	3102.4127	3136.7682	3185.5046
H	-7.23295	0.49800	-1.88158	3191.3921	3192.4425	3199.8733
C	-5.57836	1.16157	-0.66744	3200.8759	3204.0912	3207.2503
H	-6.08338	2.05826	-0.32216	3208.7557	3214.3511	3214.6464
C	0.79230	4.94847	0.39313	3218.3668	3230.8312	3269.0713
H	1.02604	5.91364	-0.04651			
C	-0.15401	4.85652	1.42076			
H	-0.65362	5.74963	1.78408			
C	-5.56980	-0.87152	-1.98518			
H	-6.06743	-1.55465	-2.66640			
H	0.56706	0.41352	2.17774			
H	-0.19794	-1.42884	-0.12129			
H	1.84127	-0.23385	-1.76745			

Statistical Thermodynamic Analysis

Temperature=373.15 K

Pressure=1 atm

Zero-point correction= 0.358756

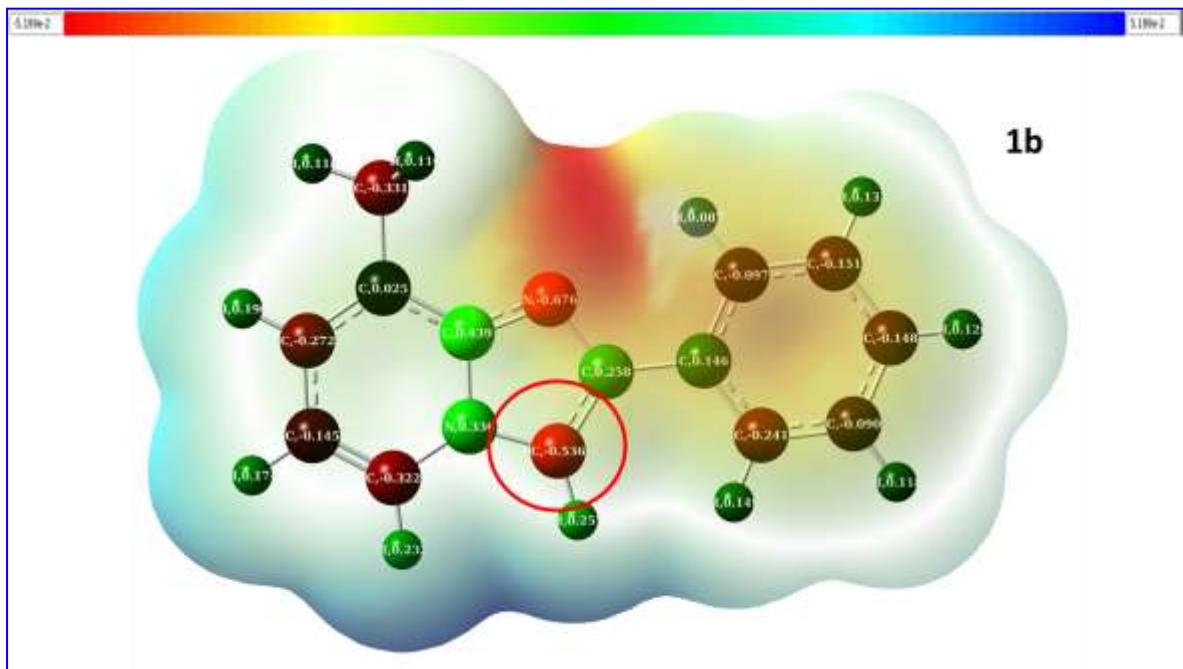
Electronic Energy = -1576.65655713

Internal Energy (E)= -1576.25716213

Enthalpy (H)= -1576.25598013

Gibbs Free Energy (G)=-1576.38175113 Gibbs Free Energy of Solvation=-1576.36371172

Merz Kollman Charges and Electron density mapped over electrostatic potential surface



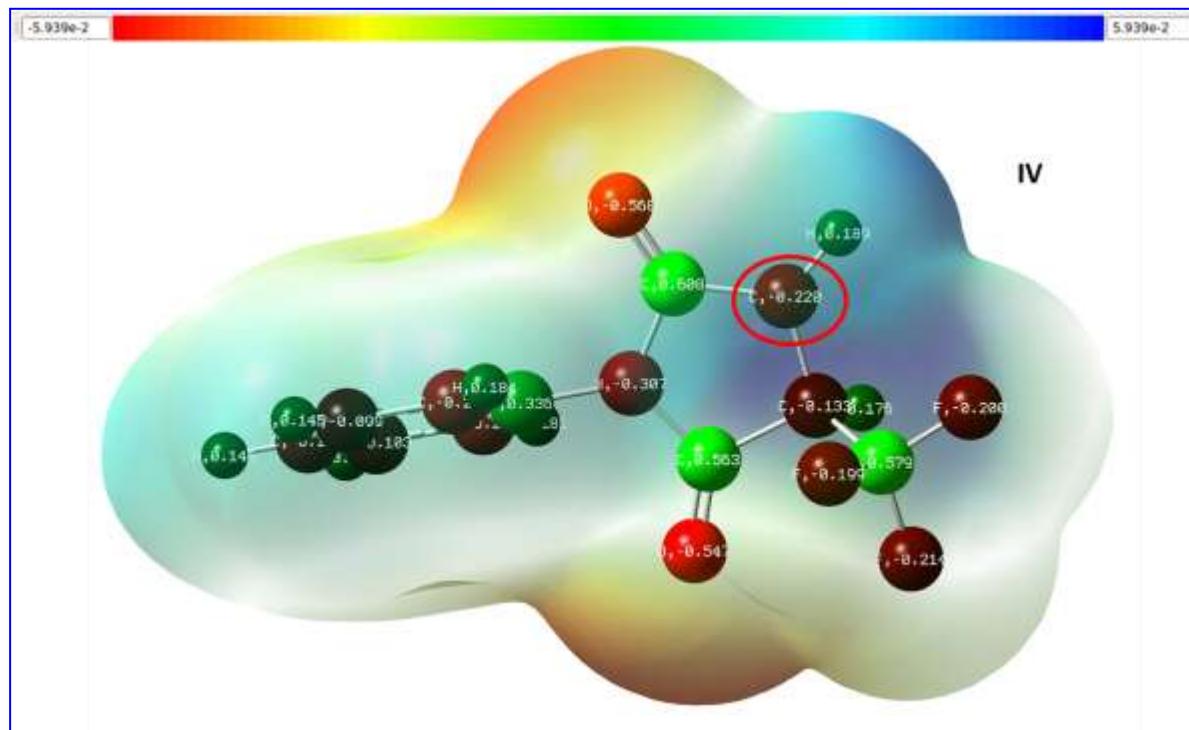


Figure 2: Merz Kollman Charges and Electron density mapped over electrostatic potential surface of intermediate **IV** and **1b** (DFT, Functional UB3LYP, Basis 6-31+G(d,p))

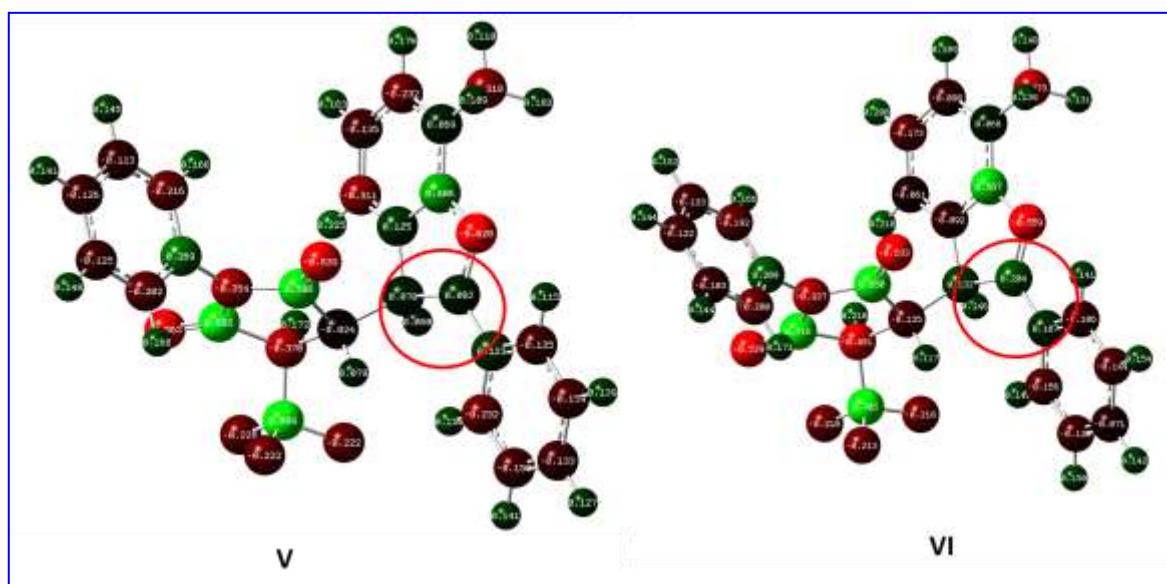


Figure 3: Increase in Merz Kollman charges from radical intermediate **V** to cationic intermediate **VI**, (DFT, Functional UB3LYP, Basis-6-31+G(d,p)). Increase in charge signifies single electron transfer.

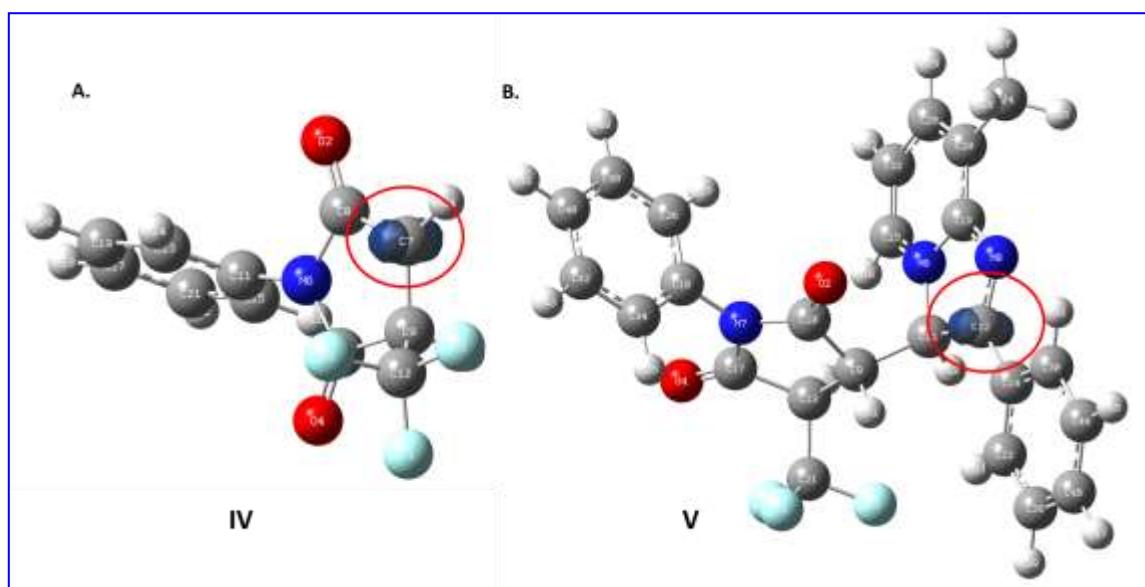
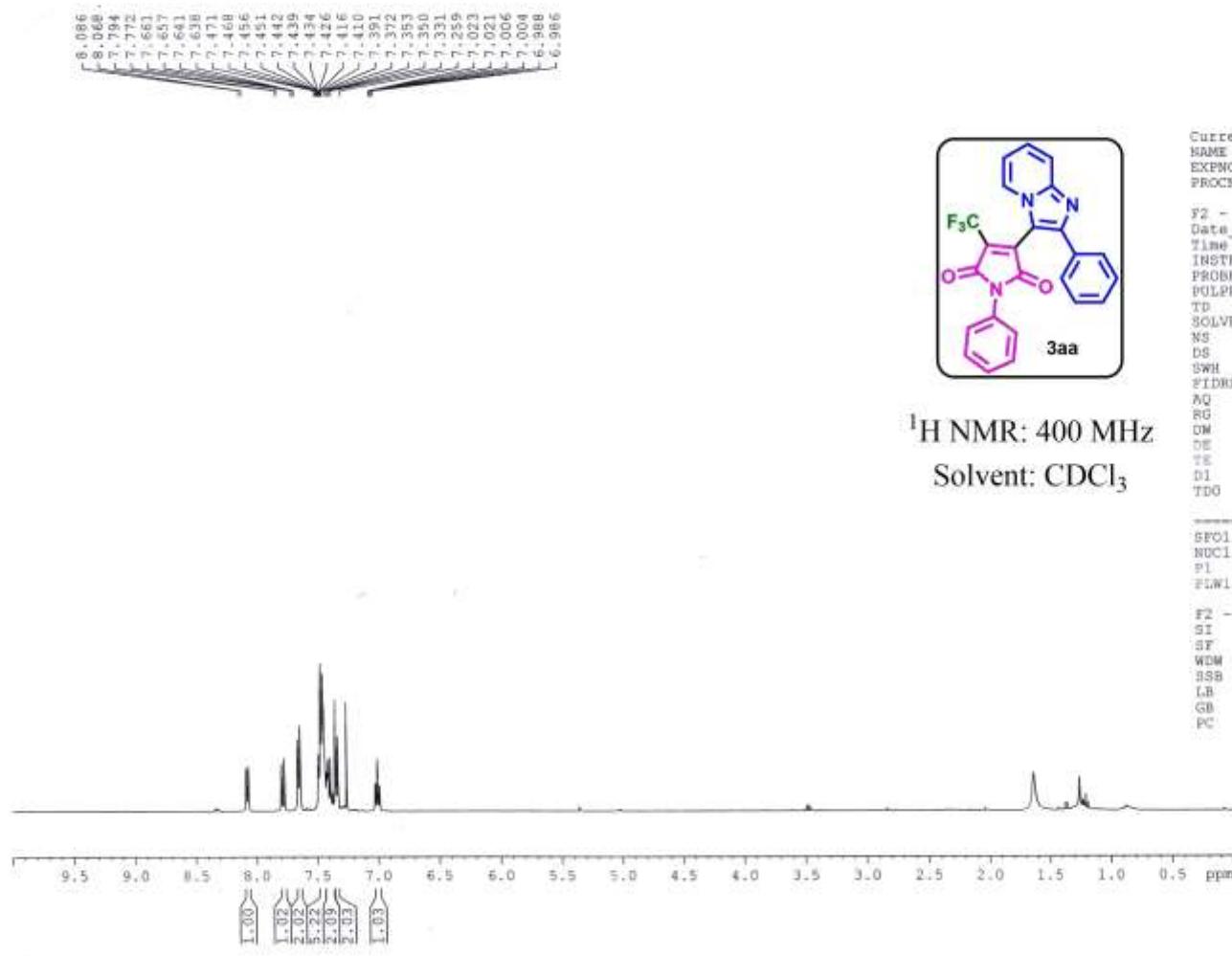


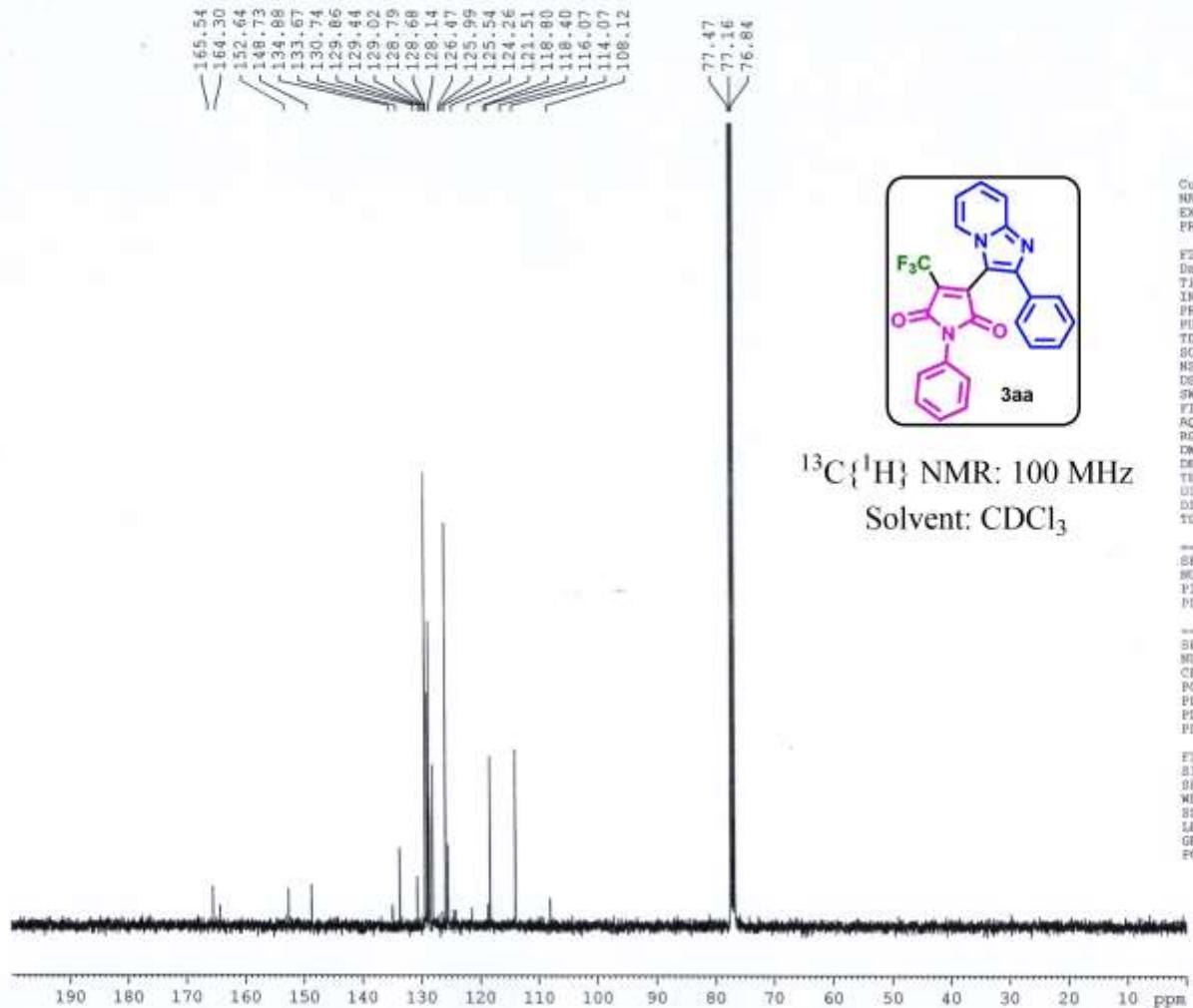
Figure 4: Spin density calculation for intermediate **IV** (iso value 0.05) and **V** (iso value 0.03), DFT-NBO calculation, Functional UB3LYP, Basis-6-31+G(d,p). Spin density signifies localization of the radical.

9. References:

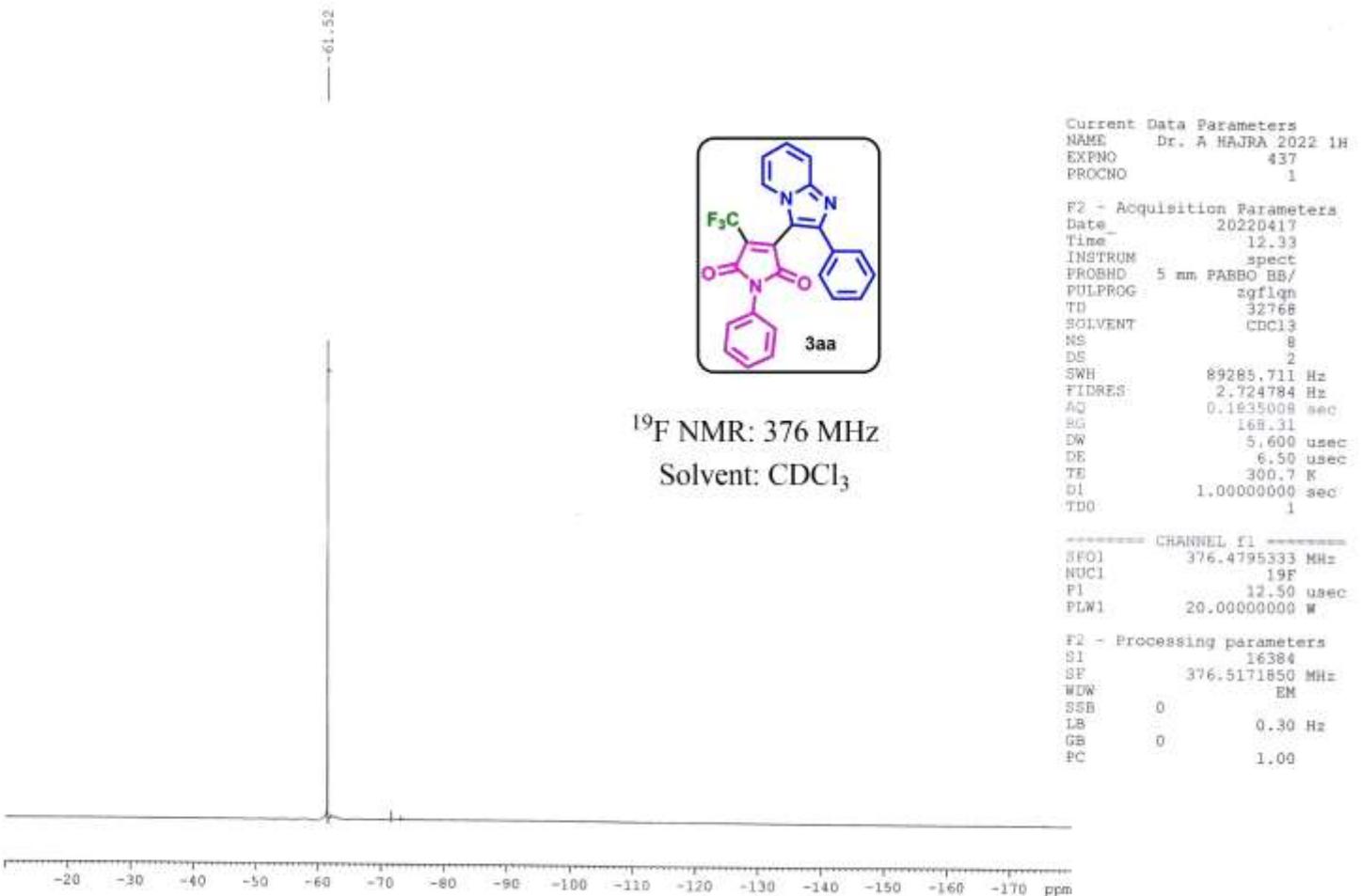
- (1) (a) A. K. Bagdi, M. Rahman, S. Santra, A. Majee and A. Hajra, *Adv. Synth. Catal.*, 2013, **355**, 1741-1747; (b) B. S. Santaniello , M. J. Price and J. K. Murray Jr , *J. Chem. Educ.*, 2017, **94** , 388-391.
- (2) (a) G. B. Deshmukh, N. S. Patil, V. B. Gaikwad, A. D. Bhole and S. V. Patil, *J. Chem. Pharm. Res.*, 2014, **6**, 393-399; (b) A. K. Ghosh, S. Samanta, P. Ghosh, S. Neogi and A. Hajra, *Org. Biomol. Chem.*, 2020, **18**, 3093-3097.
- (3) Gaussian 16, Revision C.01, M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, G. A. Petersson, H. Nakatsuji, X. Li, M. Caricato, A. V. Marenich, J. Bloino, B. G. Janesko, R. Gomperts, B. Mennucci, H. P. Hratchian, J. V. Ortiz, A. F. Izmaylov, J. L. Sonnenberg, D. Williams-Young, F. Ding, F. Lipparini, F. Egidi, J. Goings, B. Peng, A. Petrone, T. Henderson, D. Ranasinghe, V. G. Zakrzewski, J. Gao, N. Rega, G. Zheng, W. Liang, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, K. Throssell, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. J. Bearpark, J. J. Heyd, E. N. Brothers, K. N. Kudin, V. N. Staroverov, T. A. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. P. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, J. M. Millam, M. Klene, C. Adamo, R. Cammi, J. W. Ochterski, R. L. Martin, K. Morokuma, O. Farkas, J. B. Foresman, and D. J. Fox, Gaussian, Inc., Wallingford CT, 2016.
- (4) A. D. Becke, *J. Chem. Phys.*, 1993, **98**, 5648-5652.
- (5) J. Tomasi and M. Persico, *Chem. Rev.*, 1994, **94**, 2027–2094.

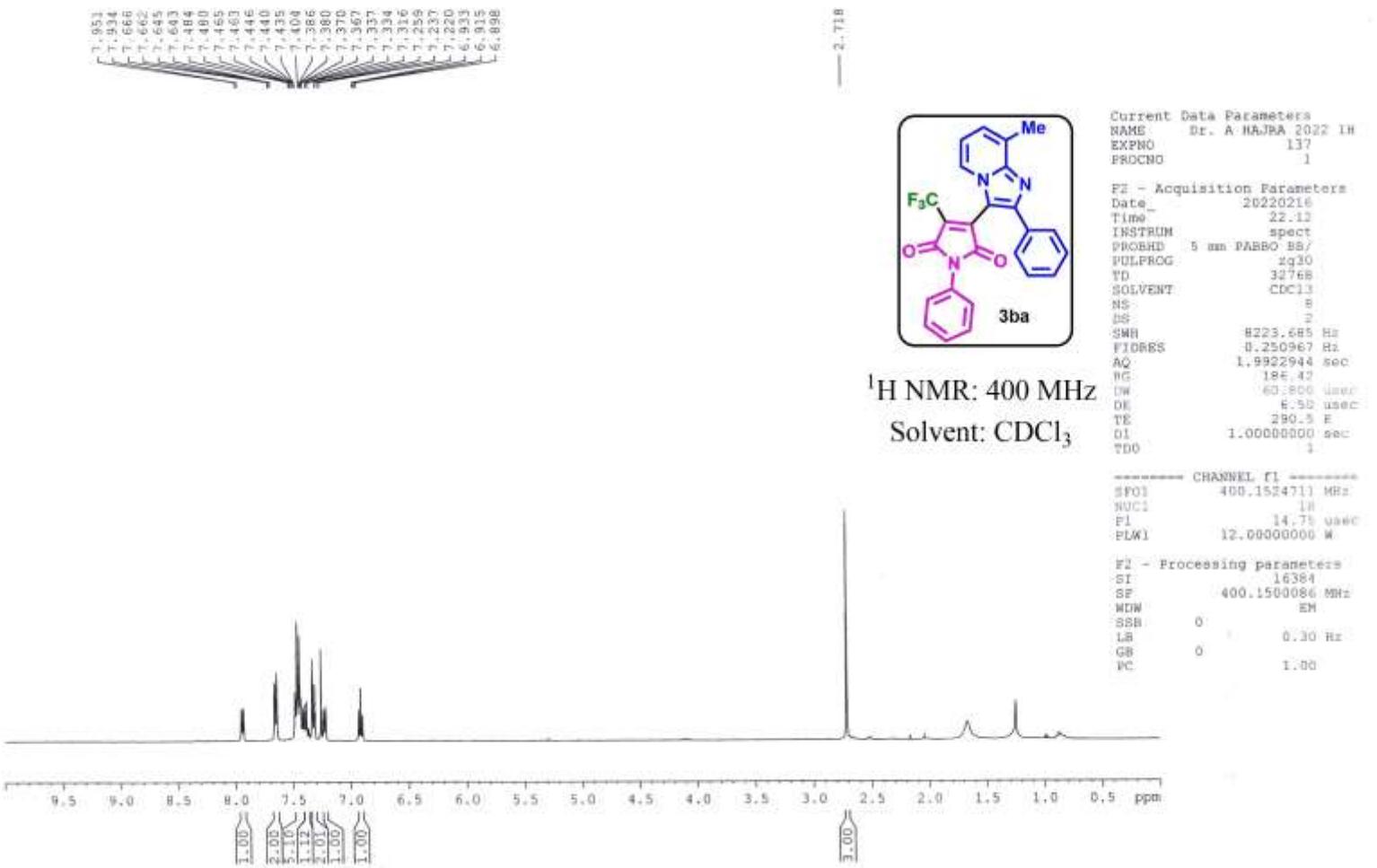
10. NMR spectra [^1H , $^{13}\text{C}\{^1\text{H}\}$ and ^{19}F] of synthesized products

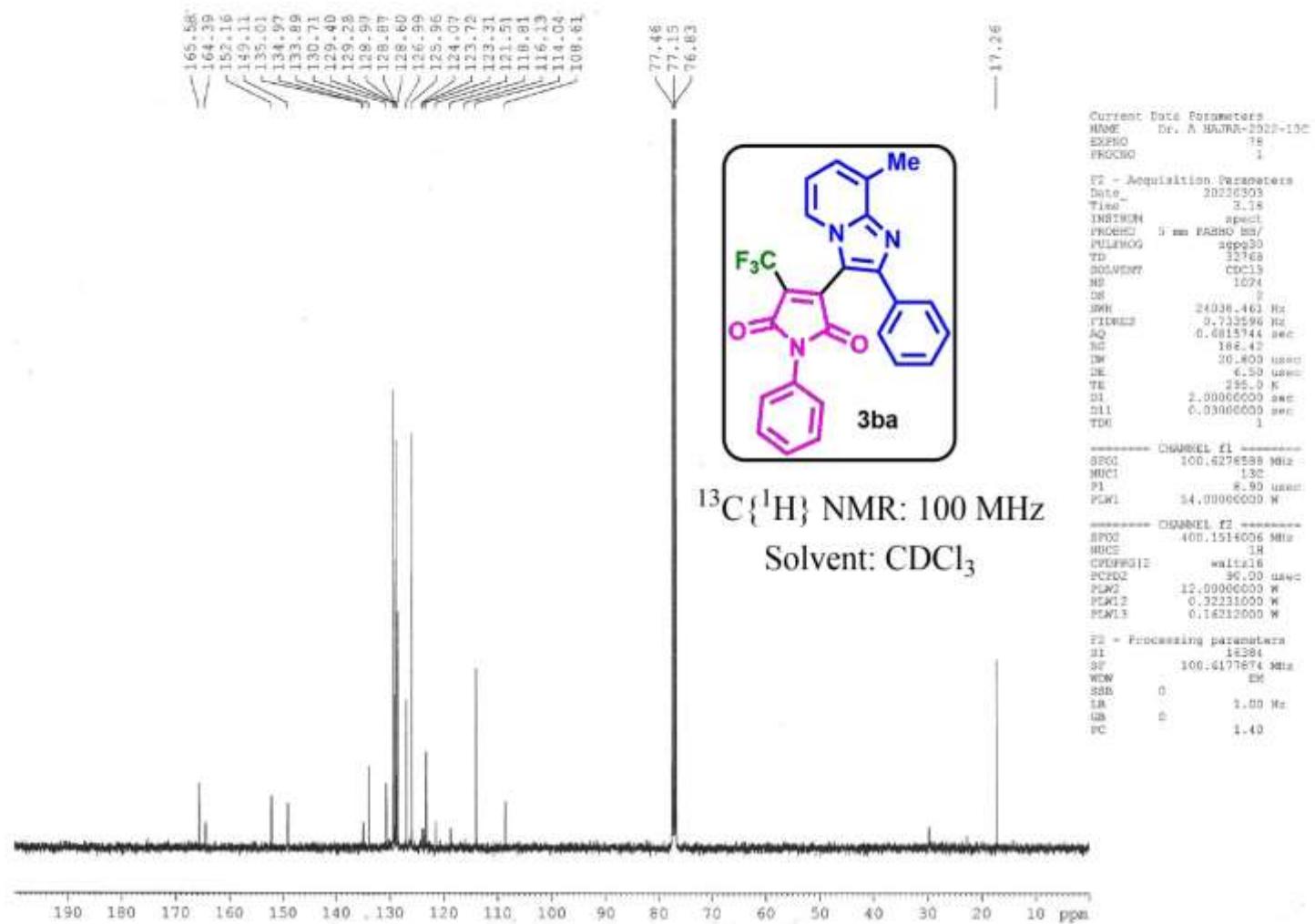


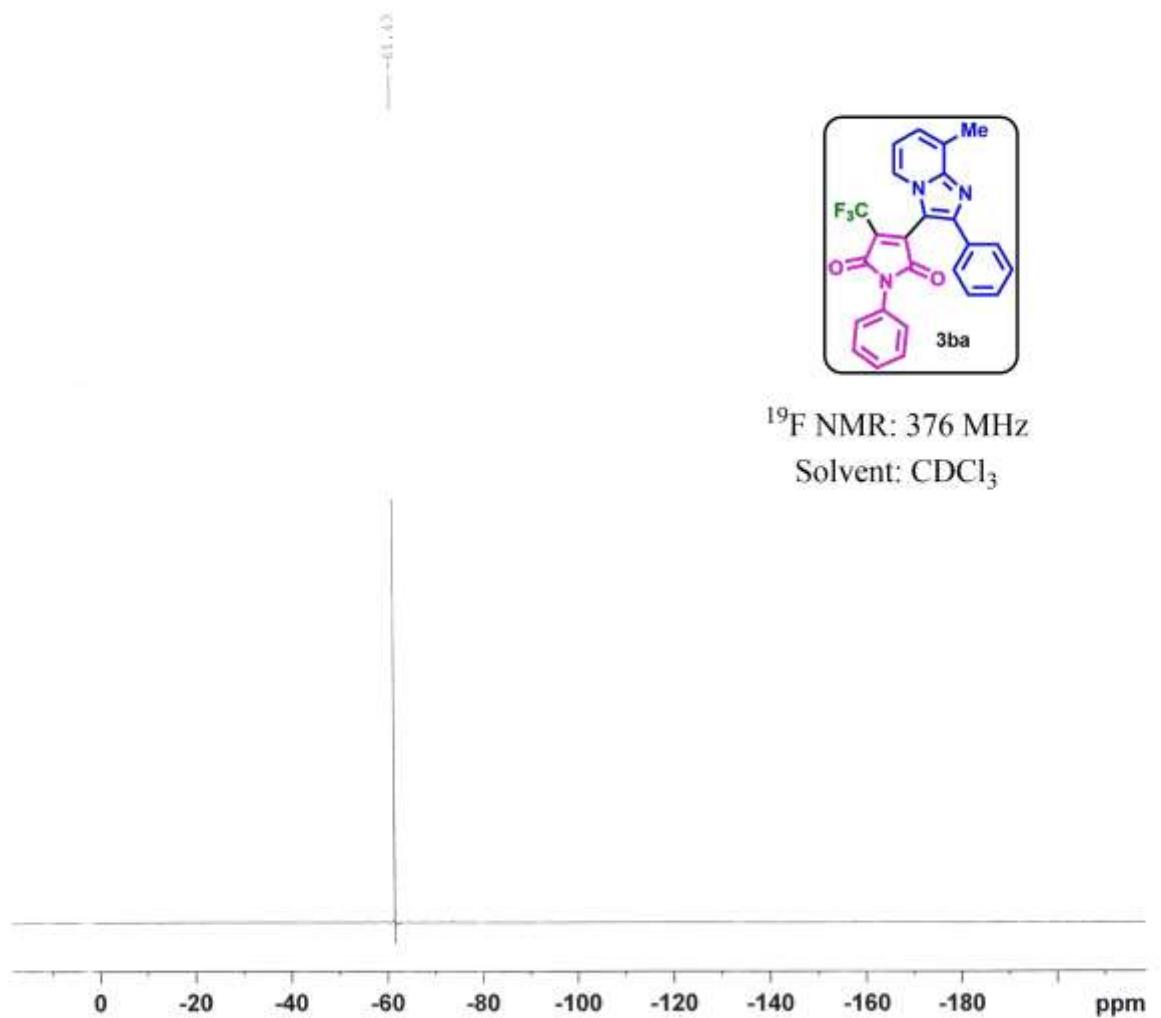


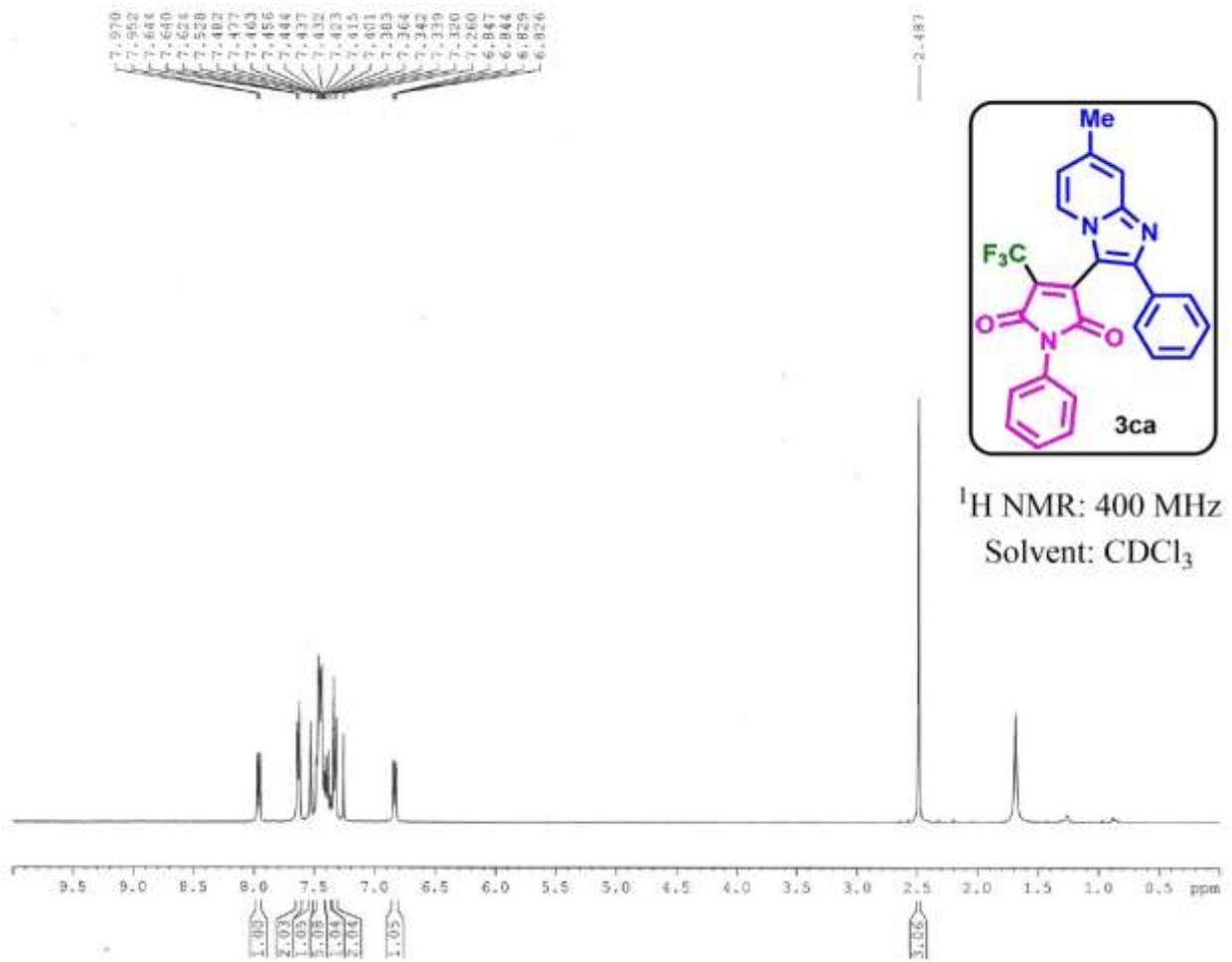
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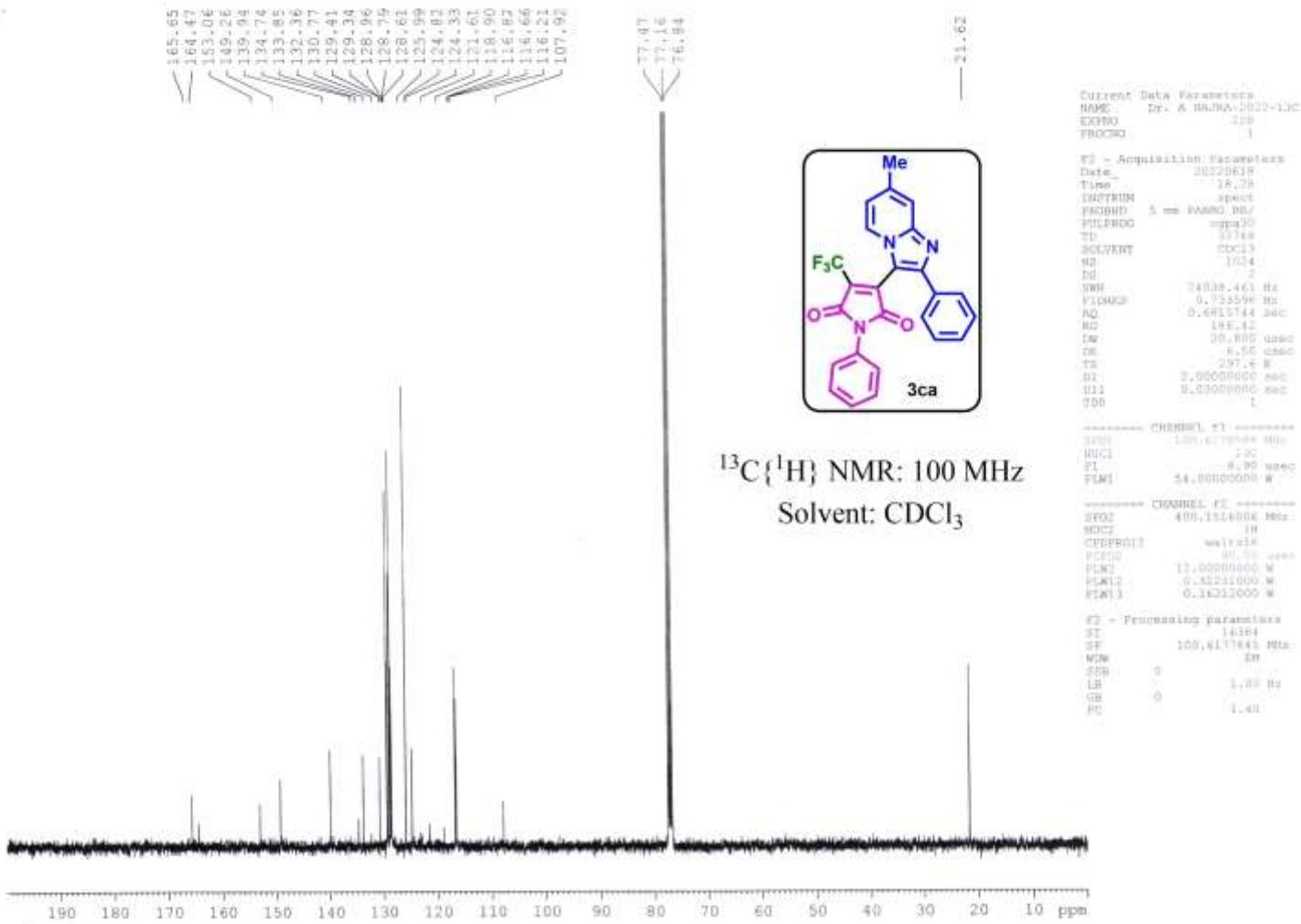


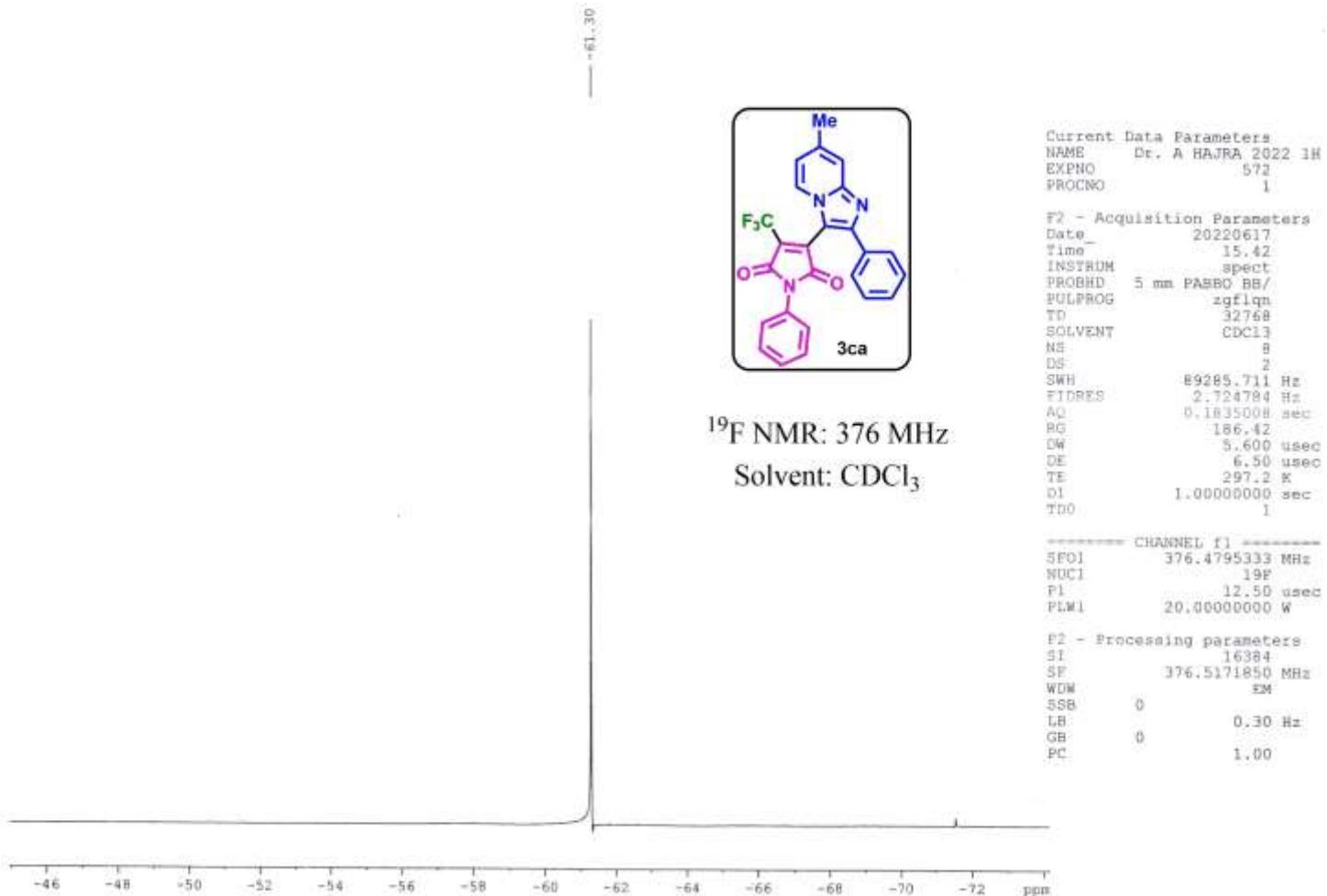


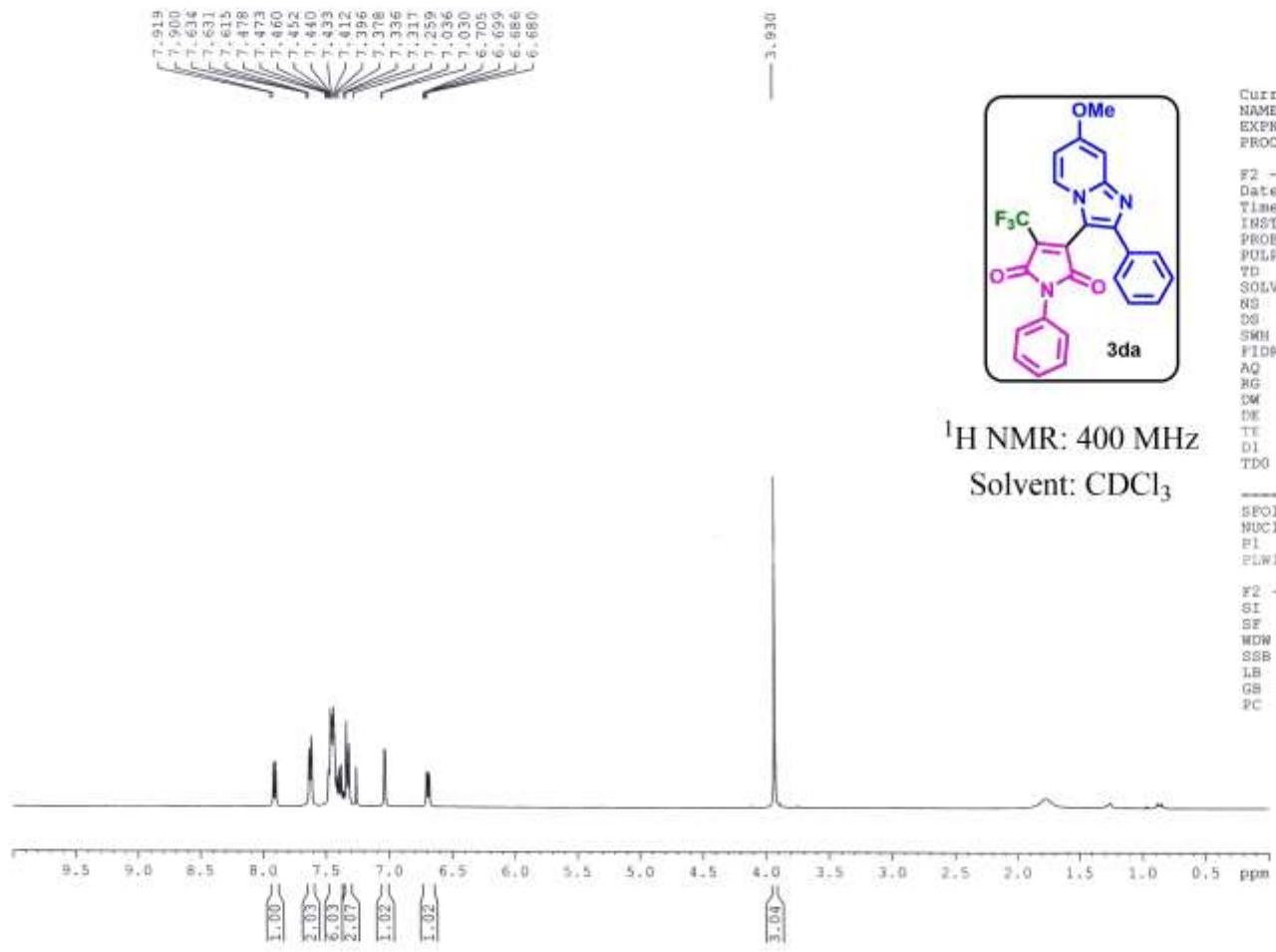


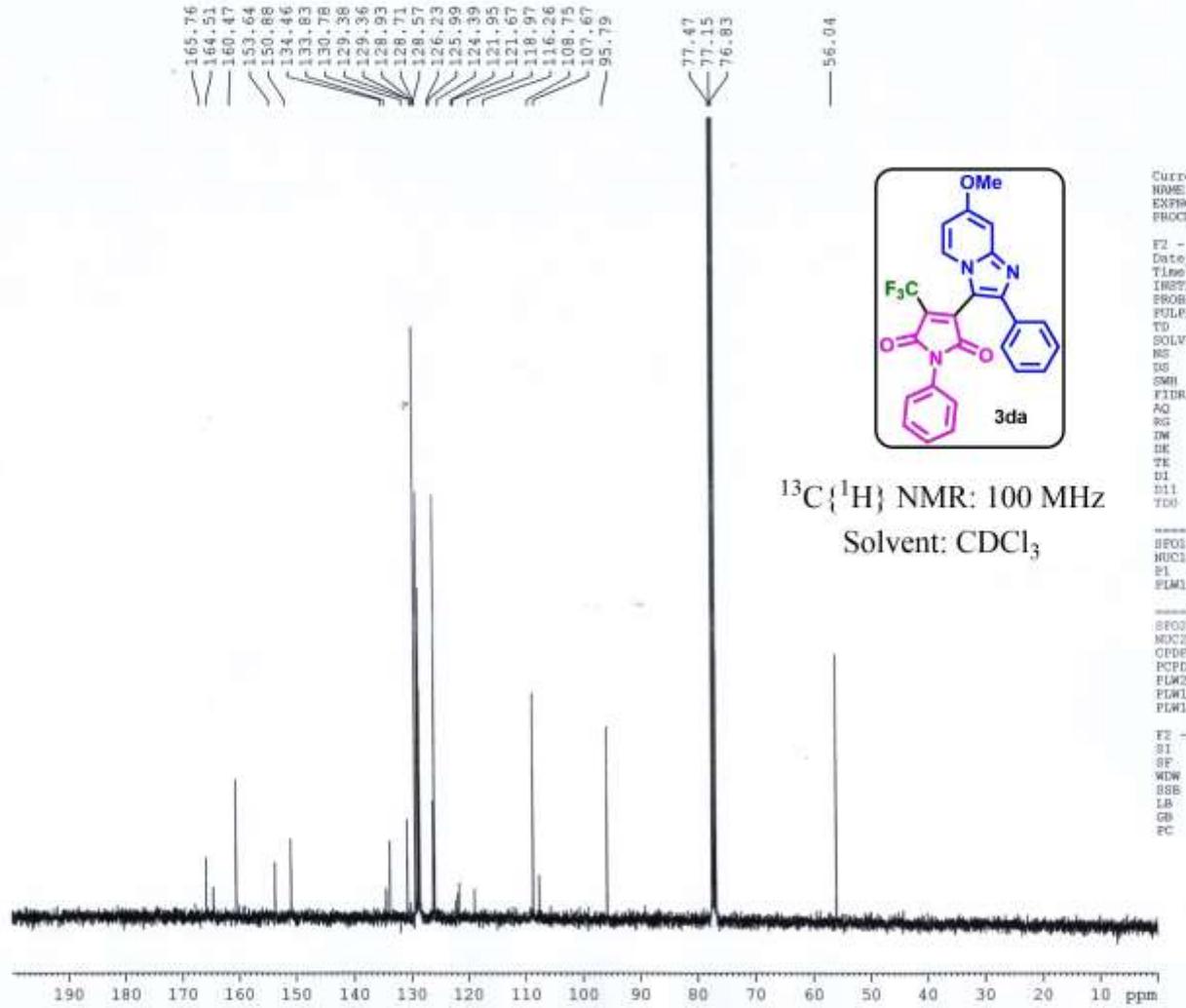


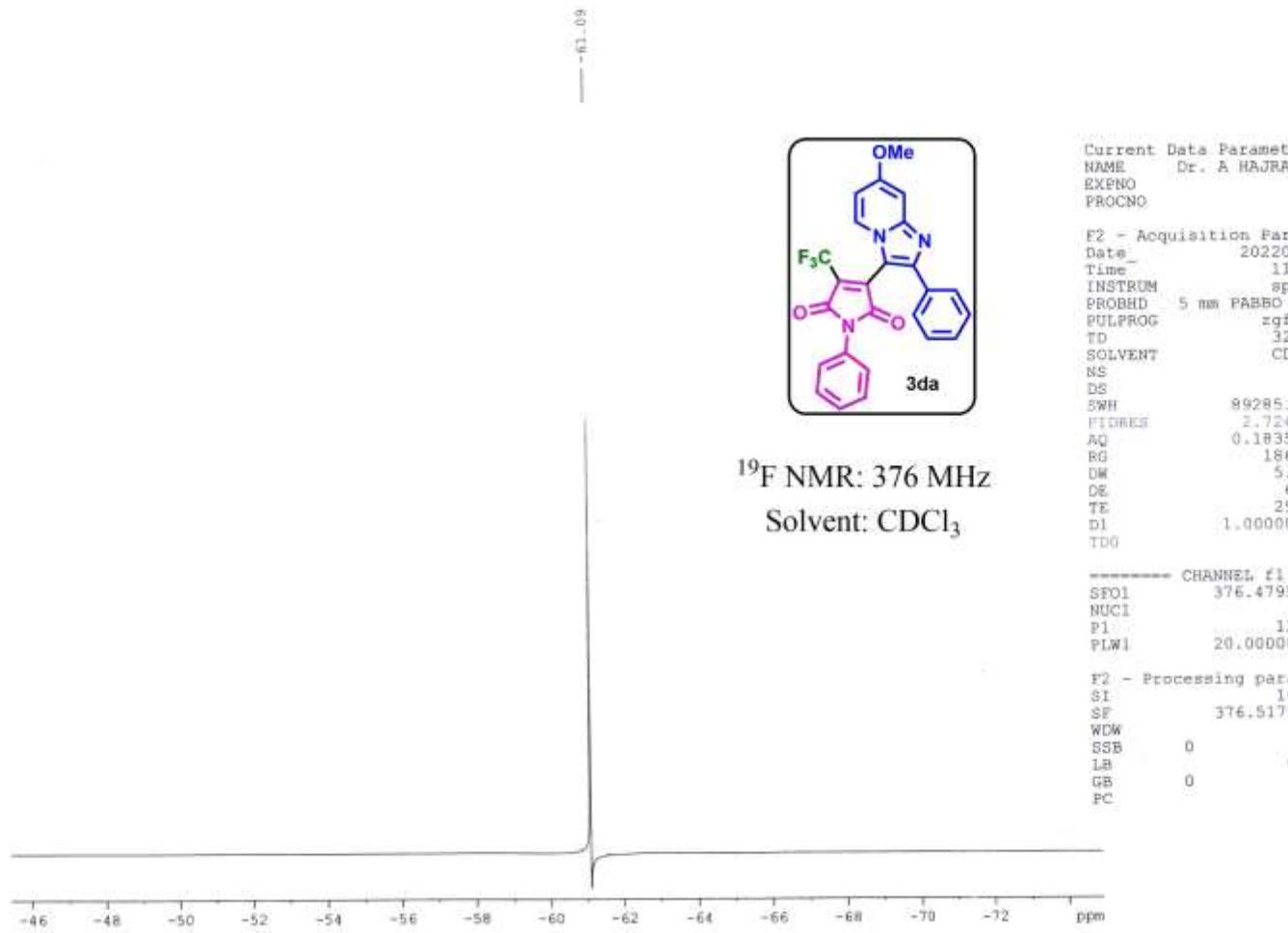


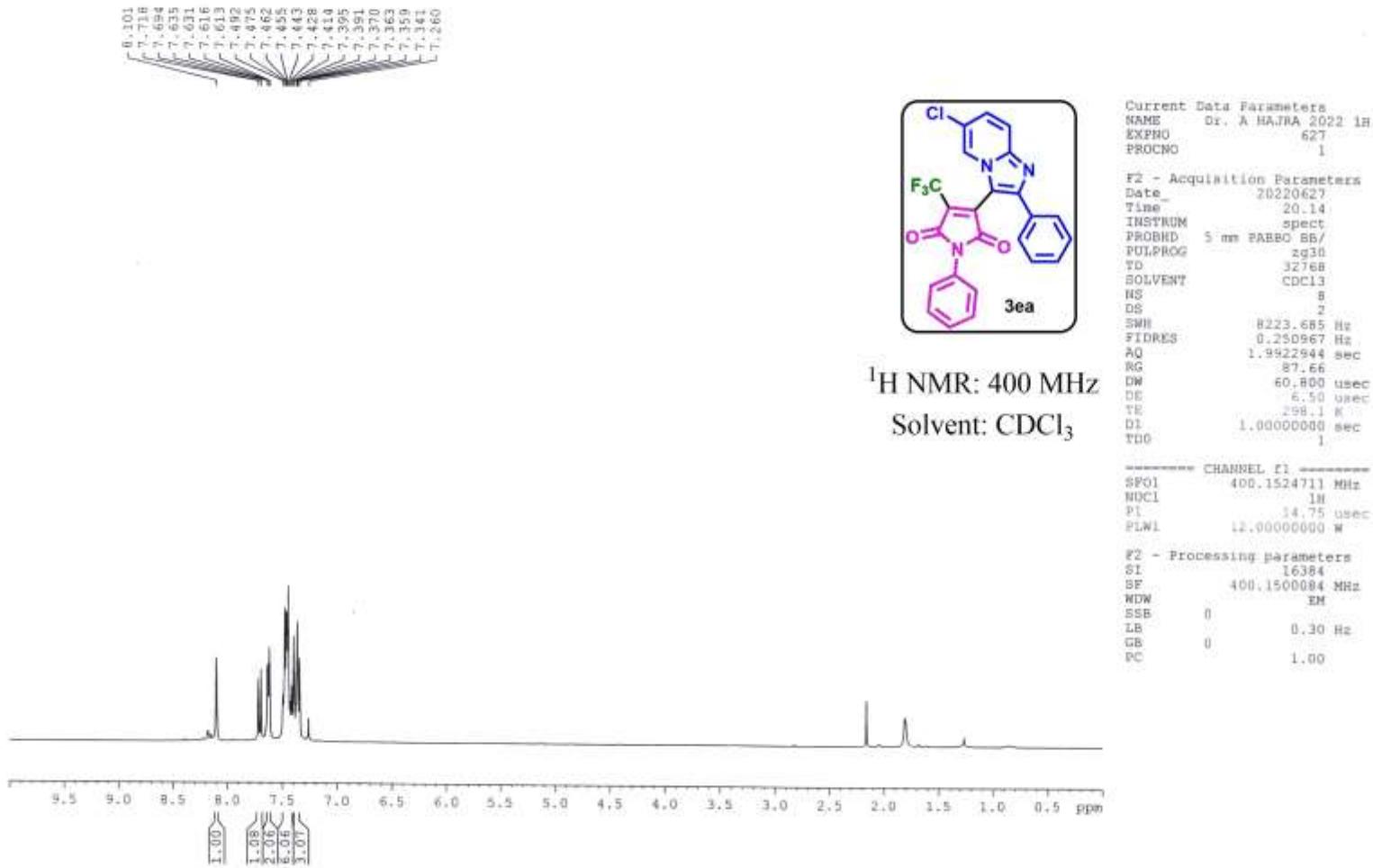


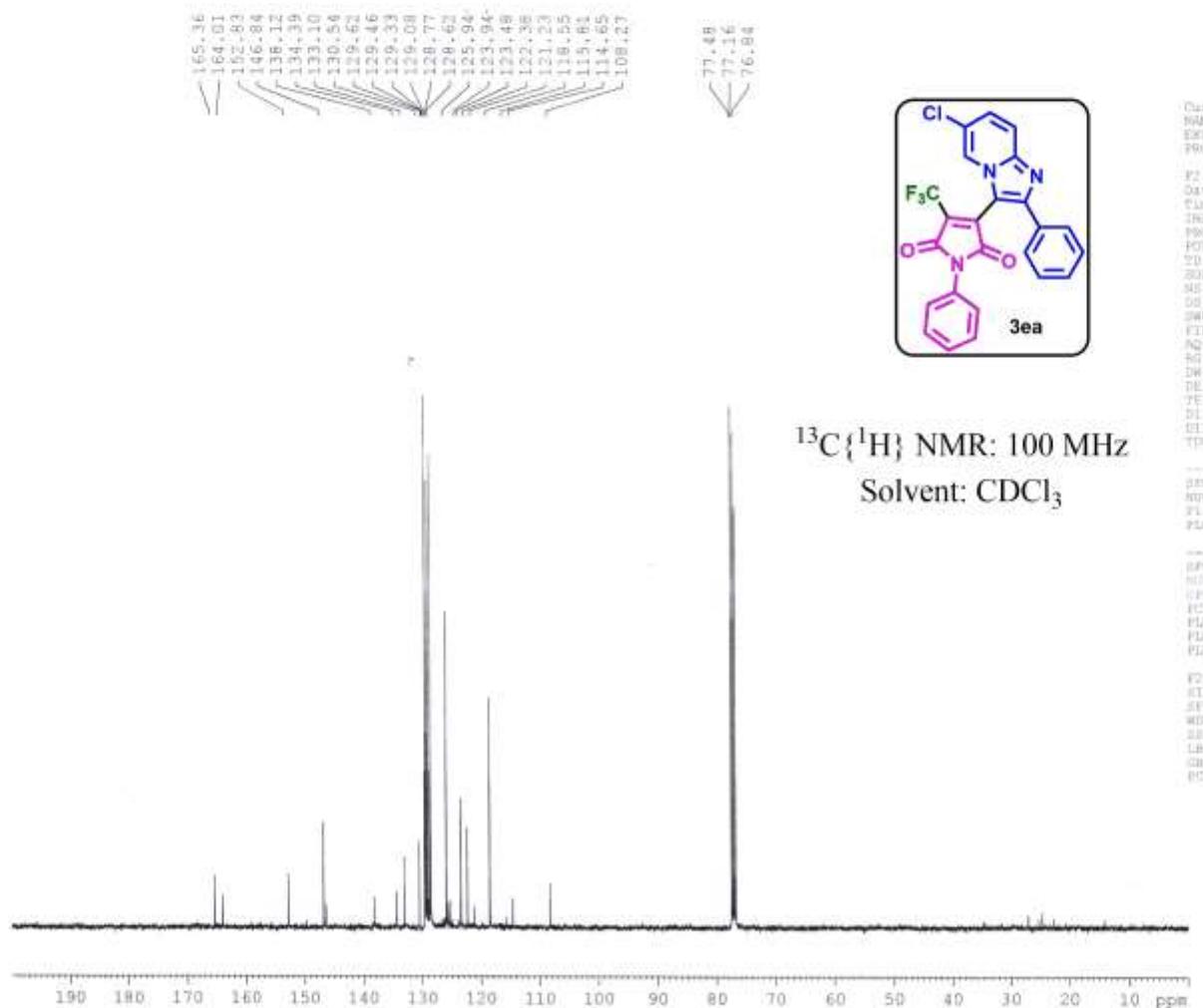


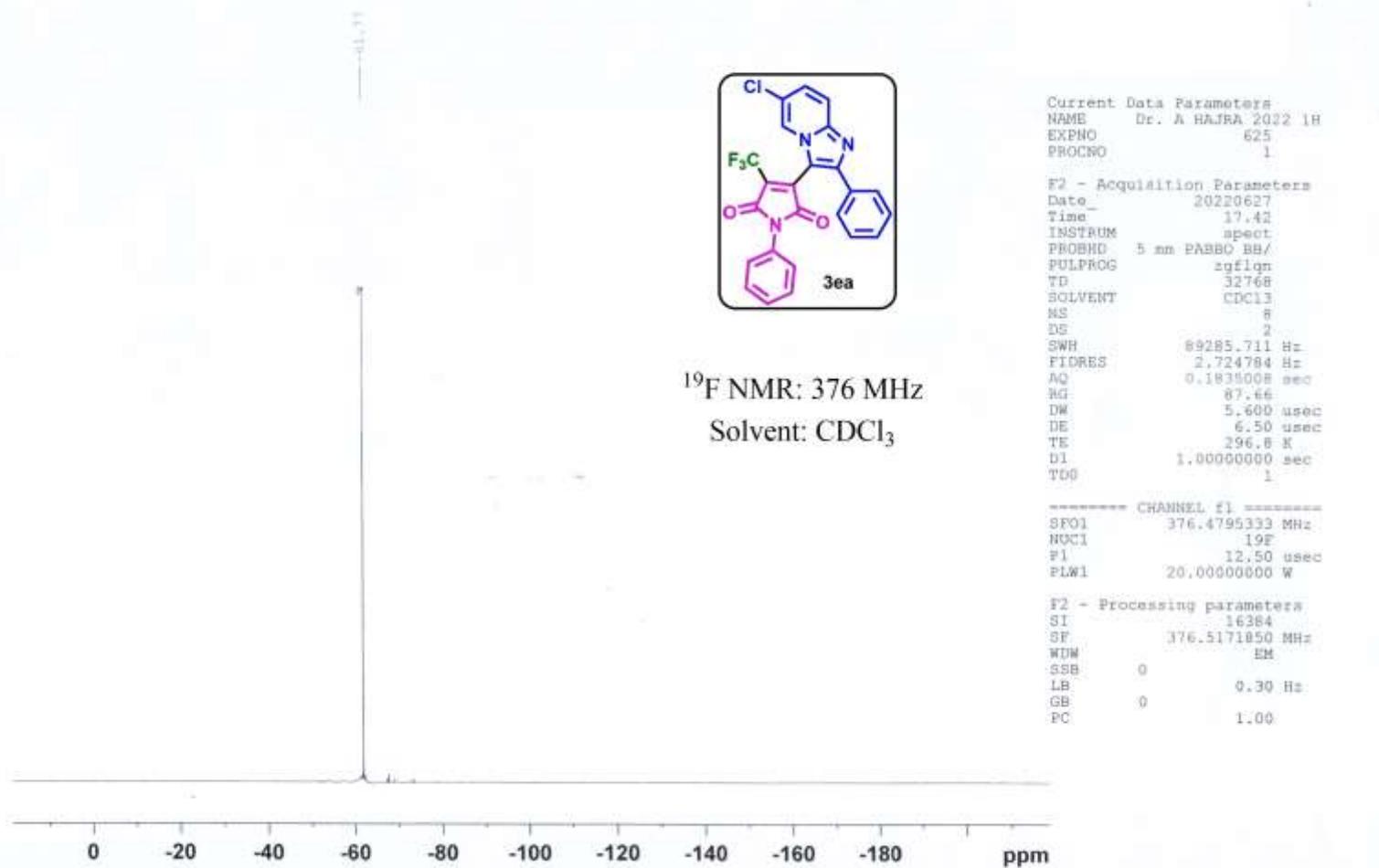


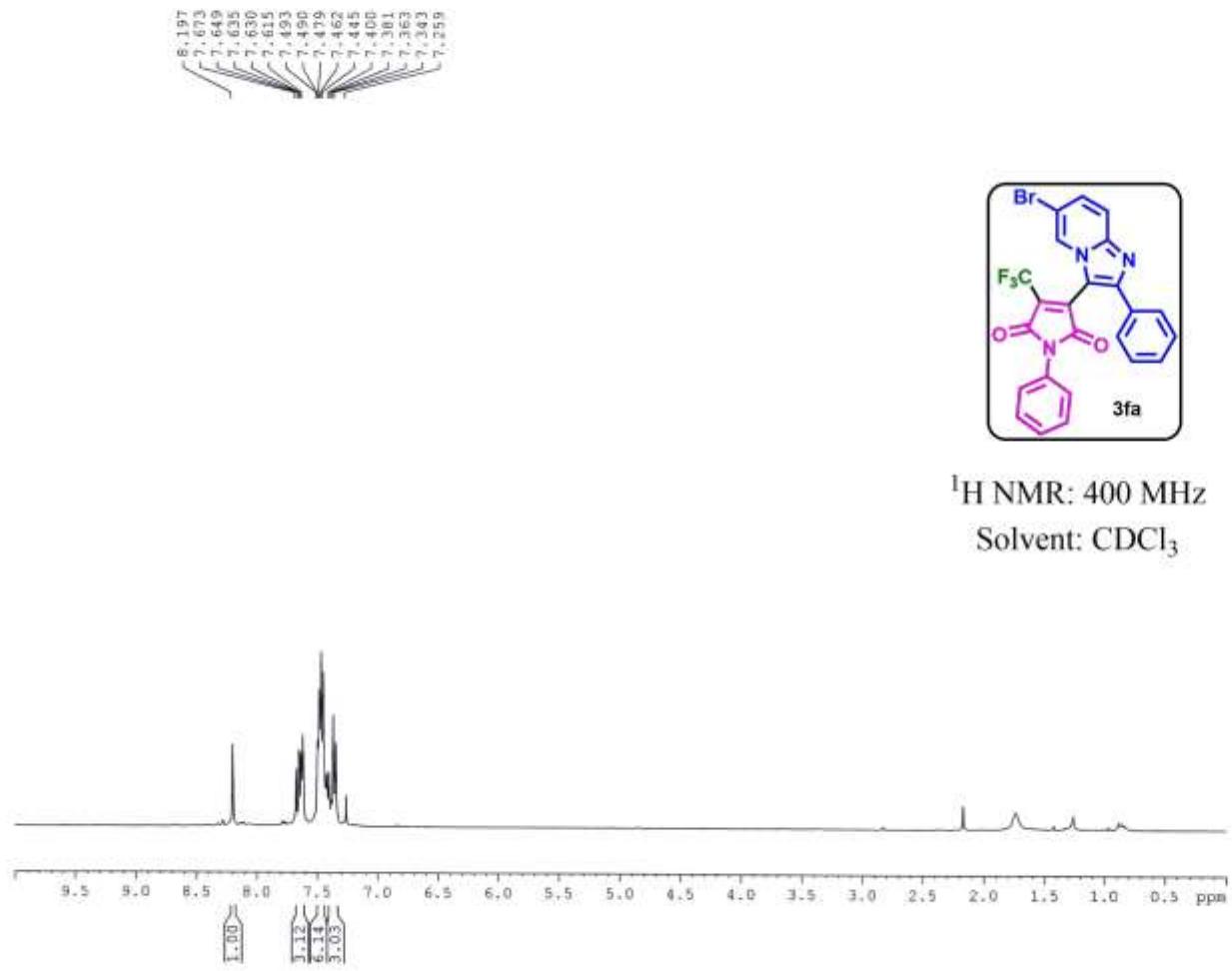












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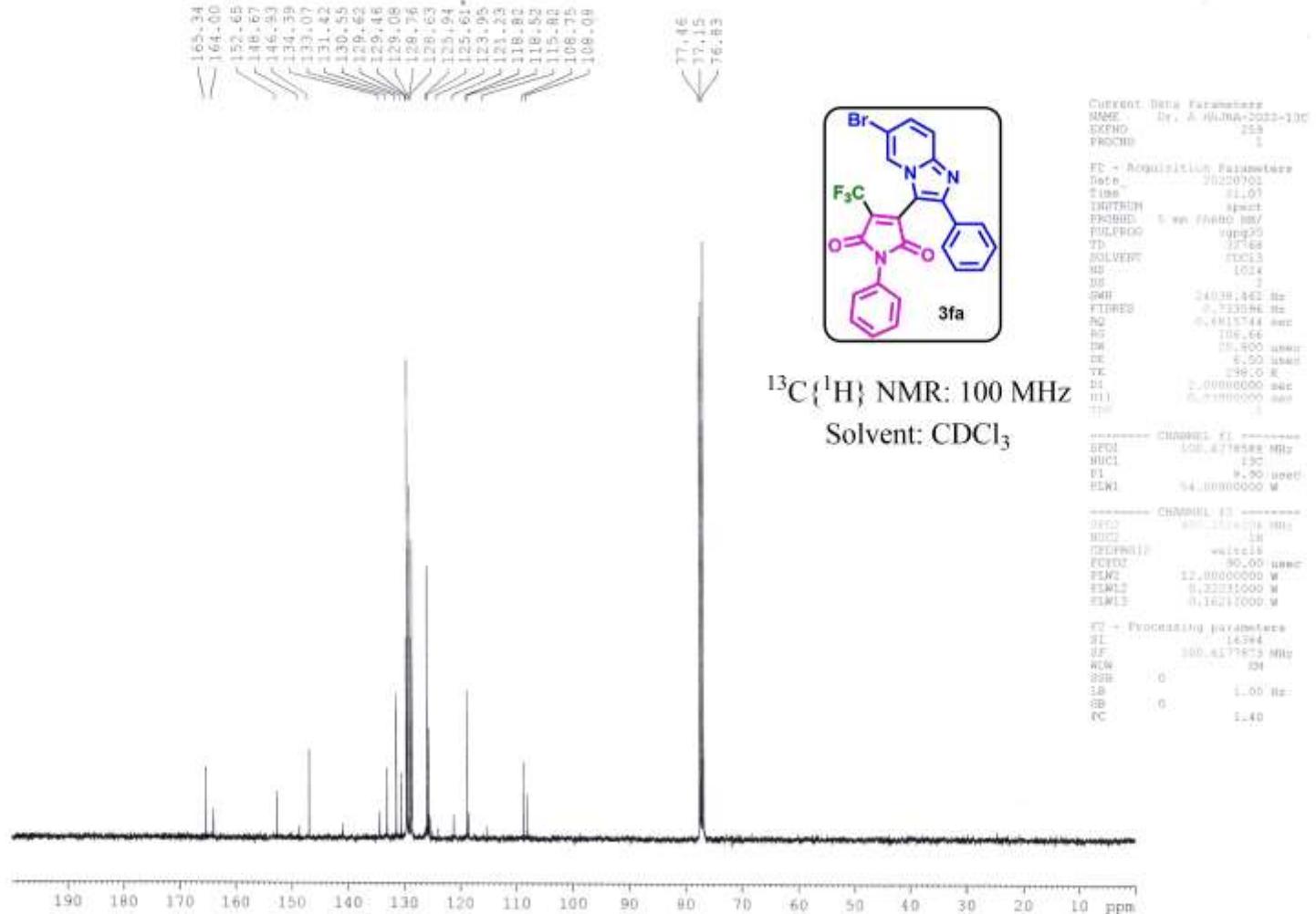
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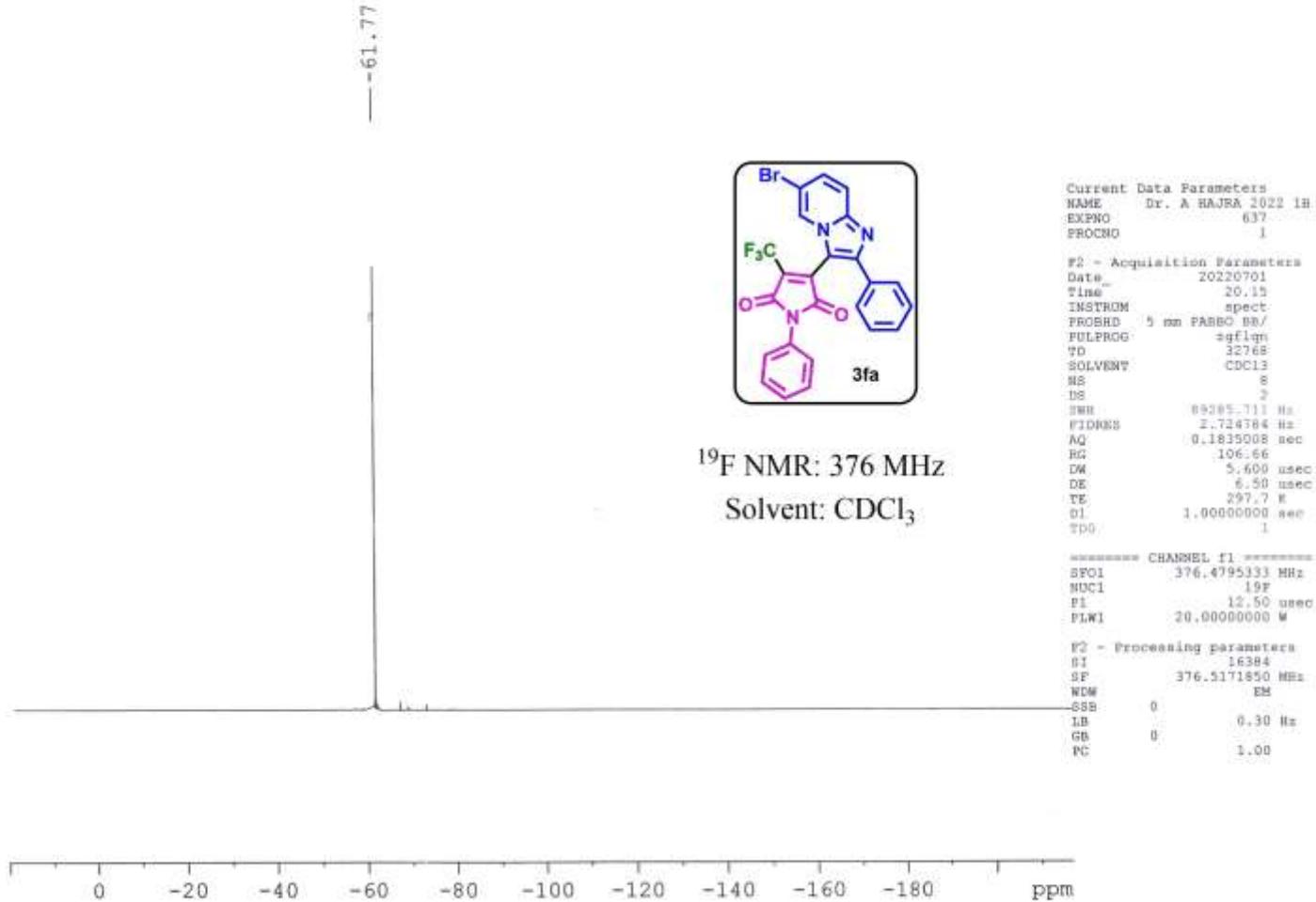
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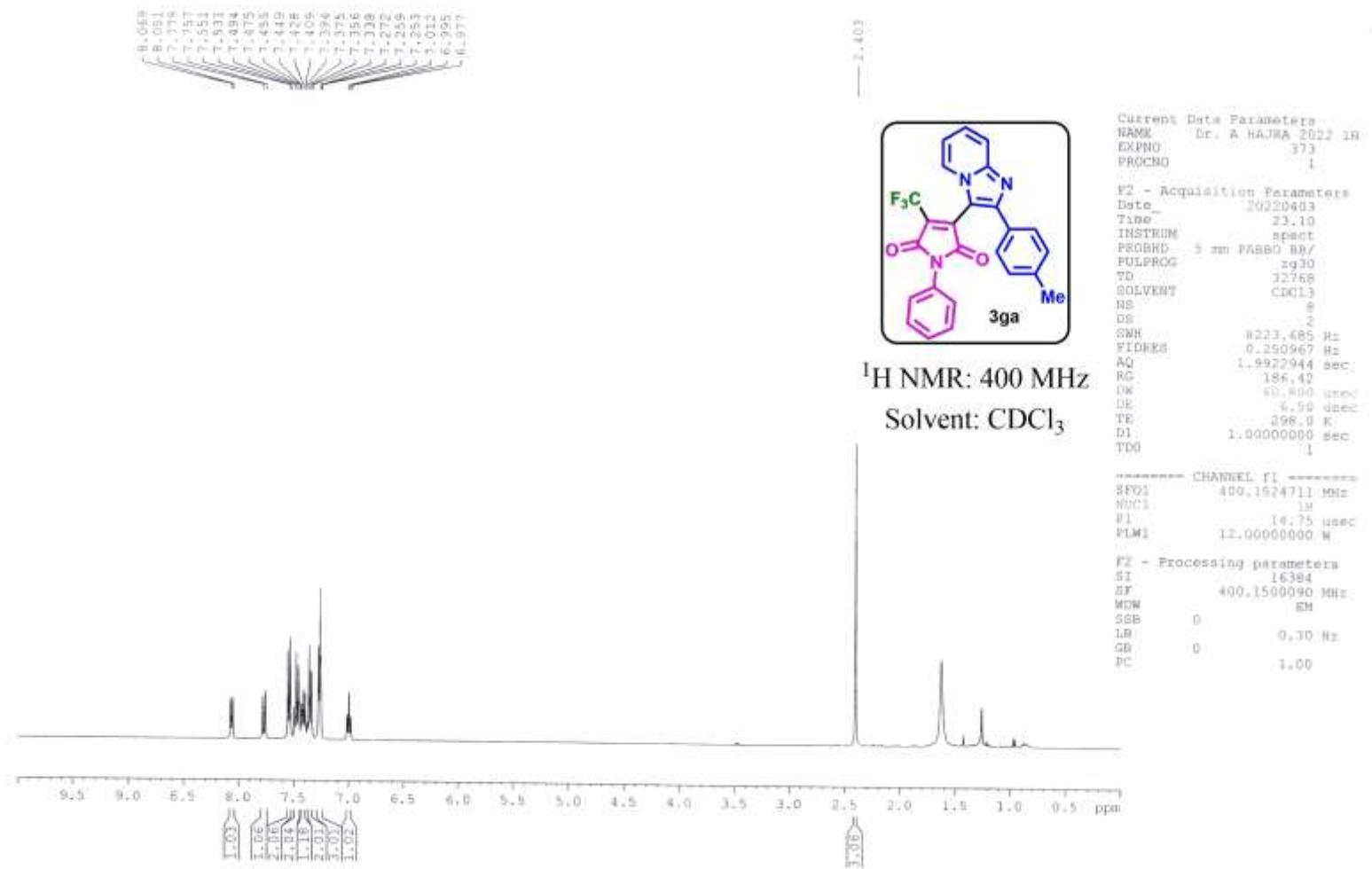
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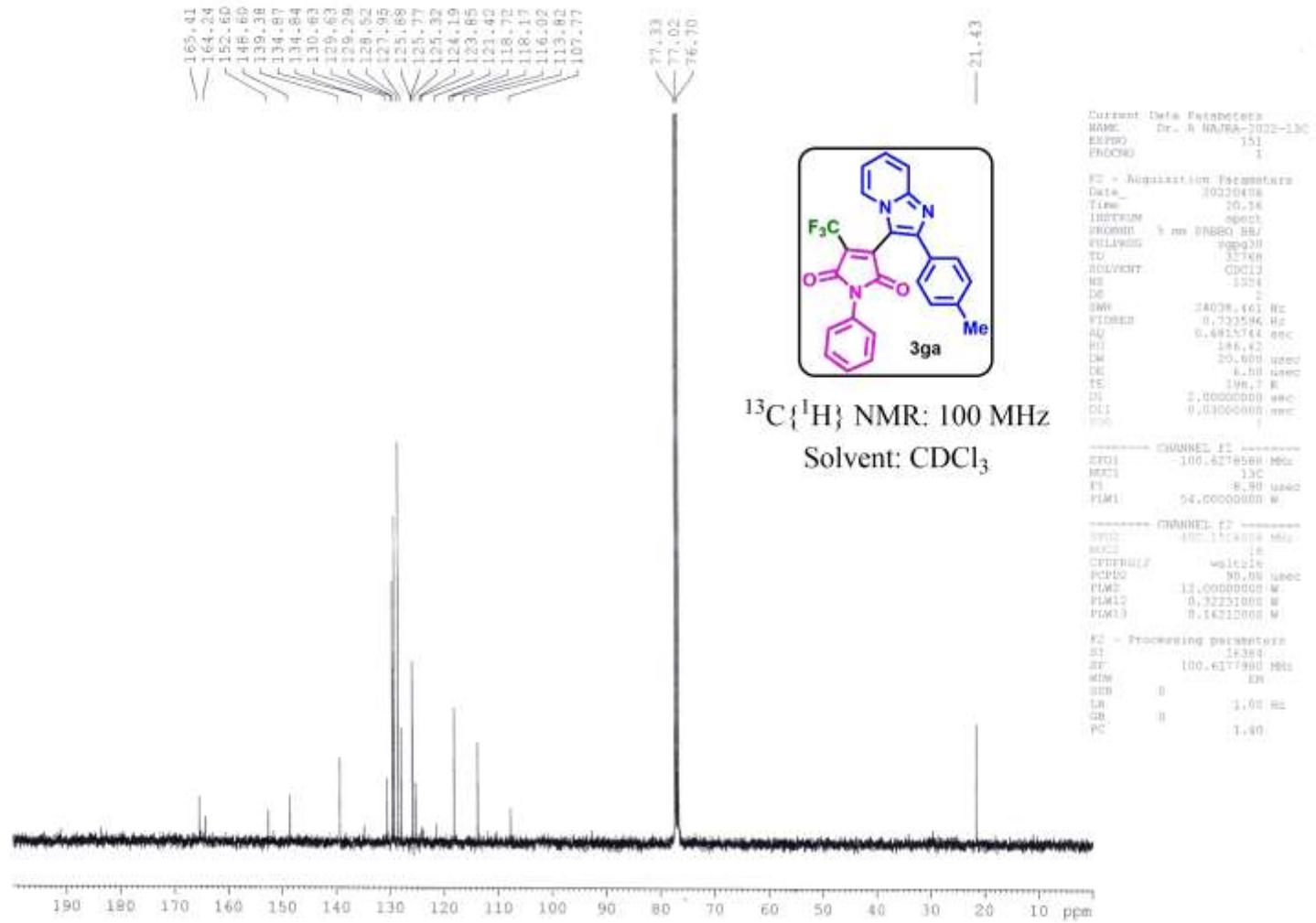
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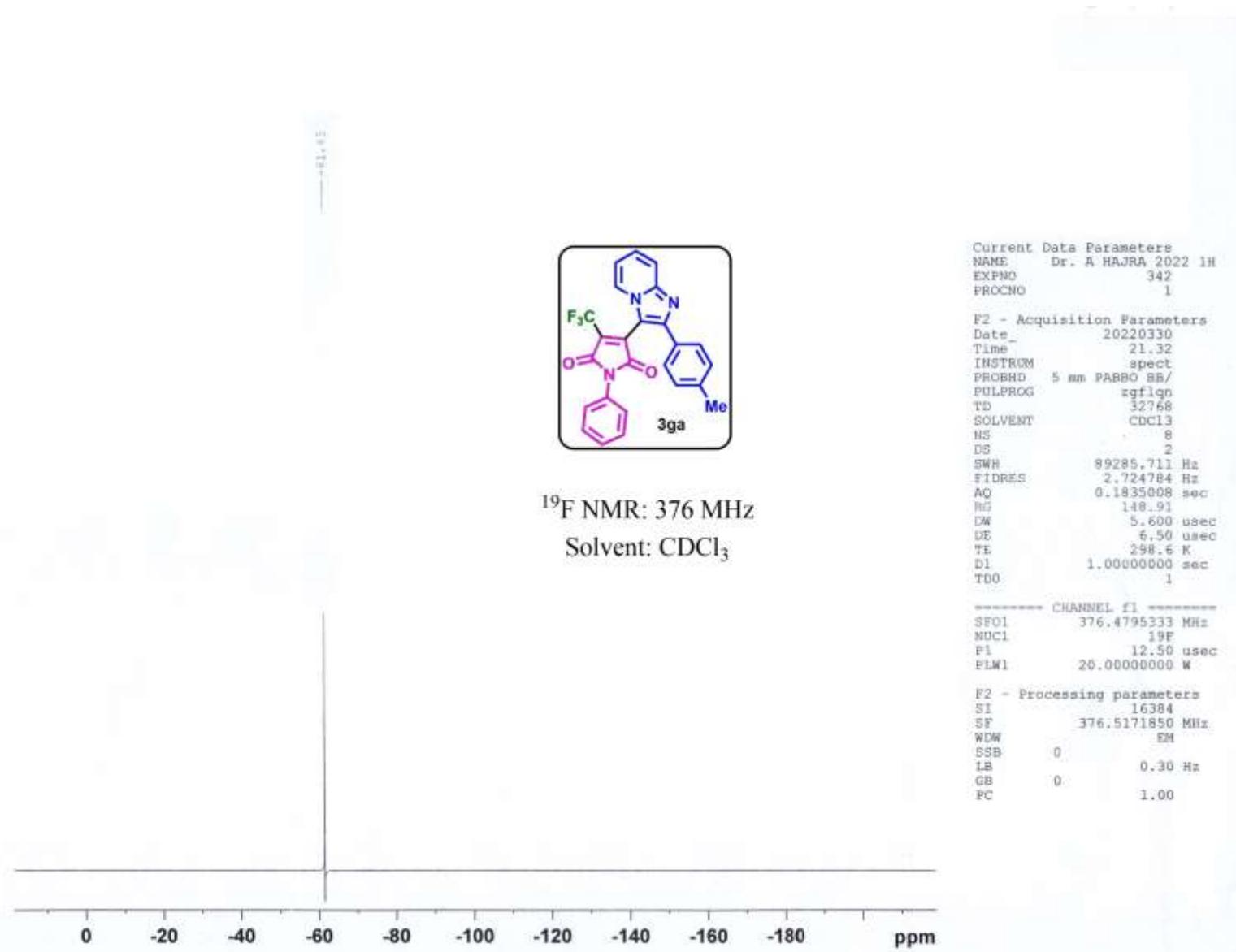
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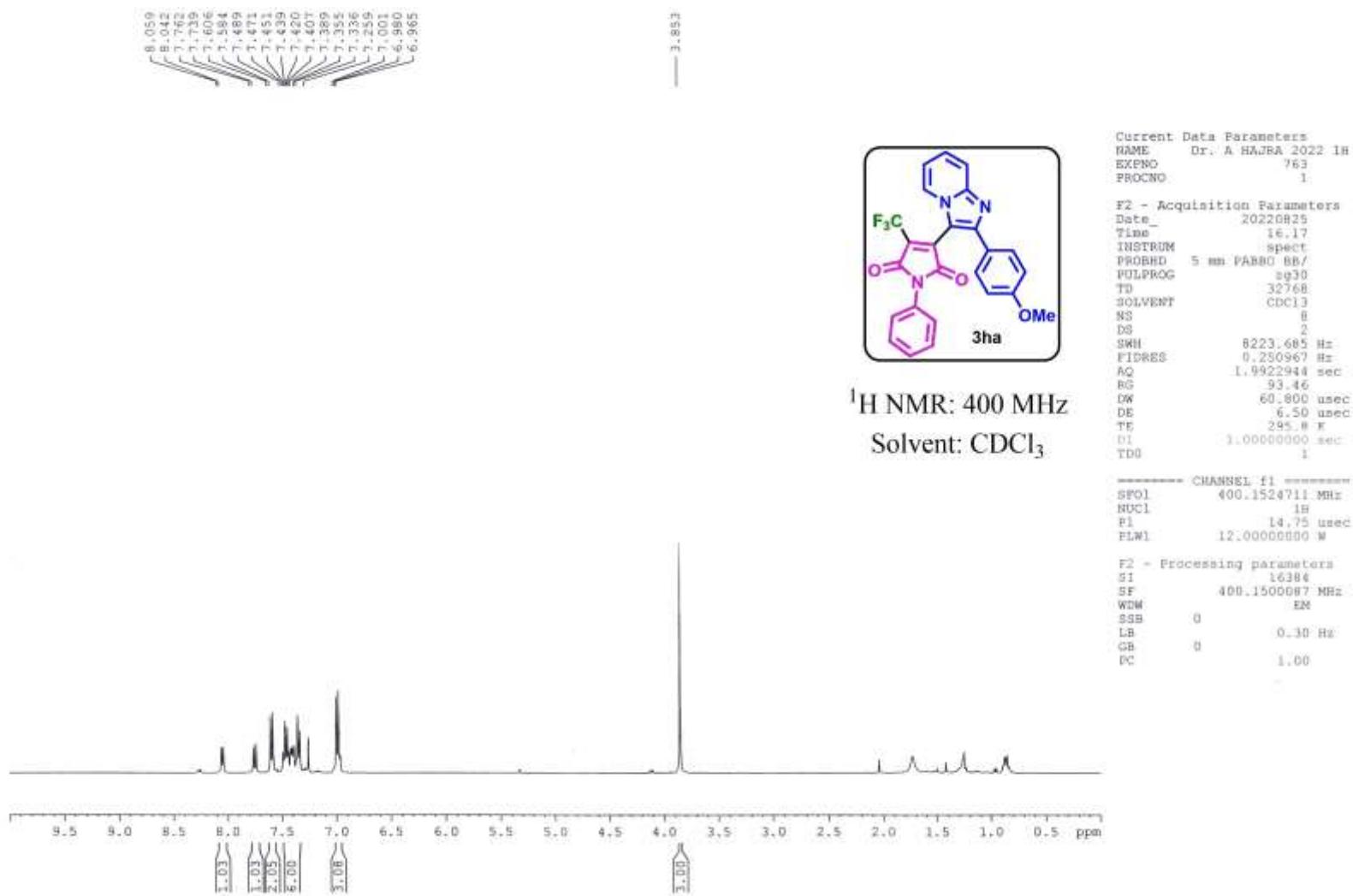


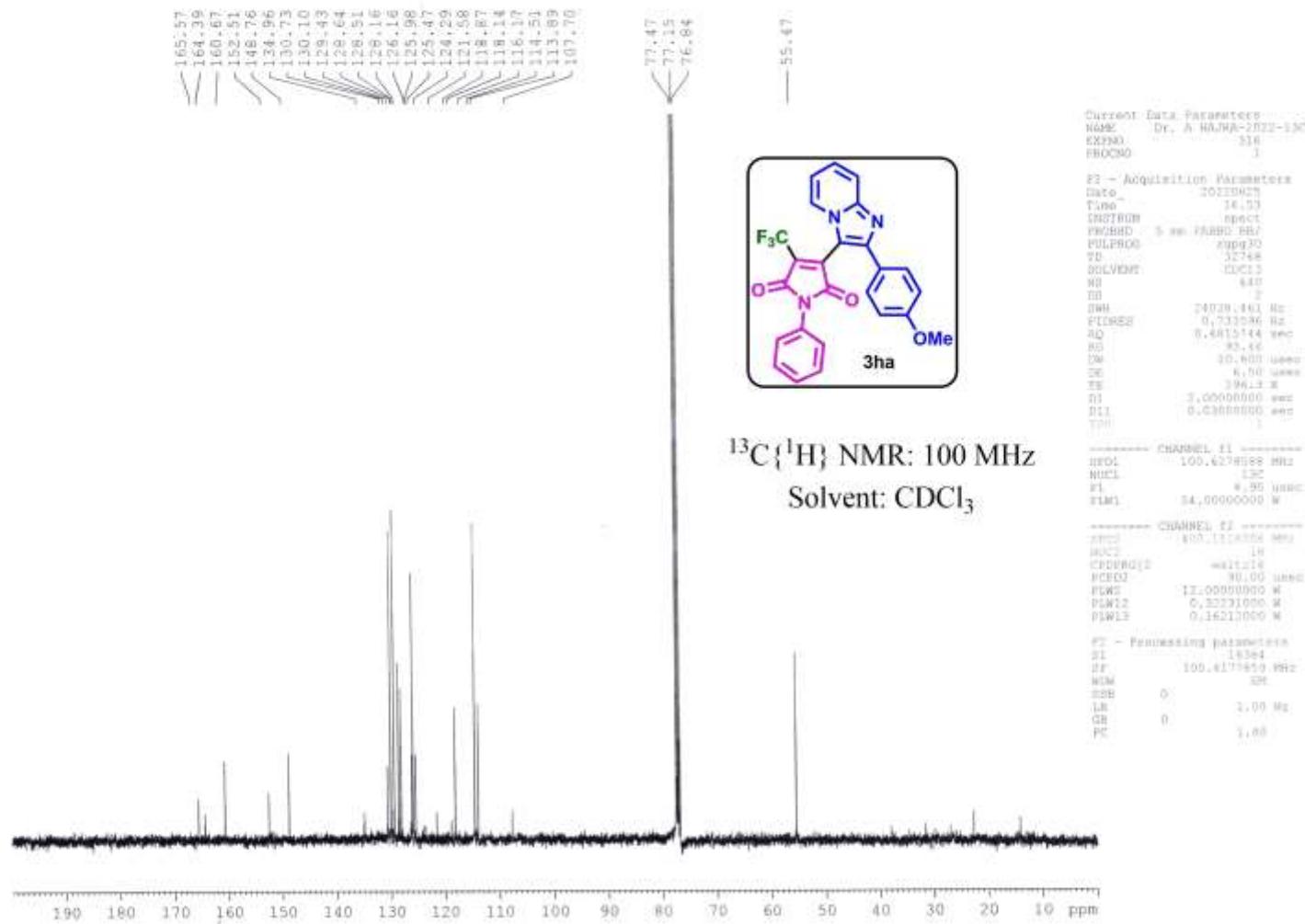


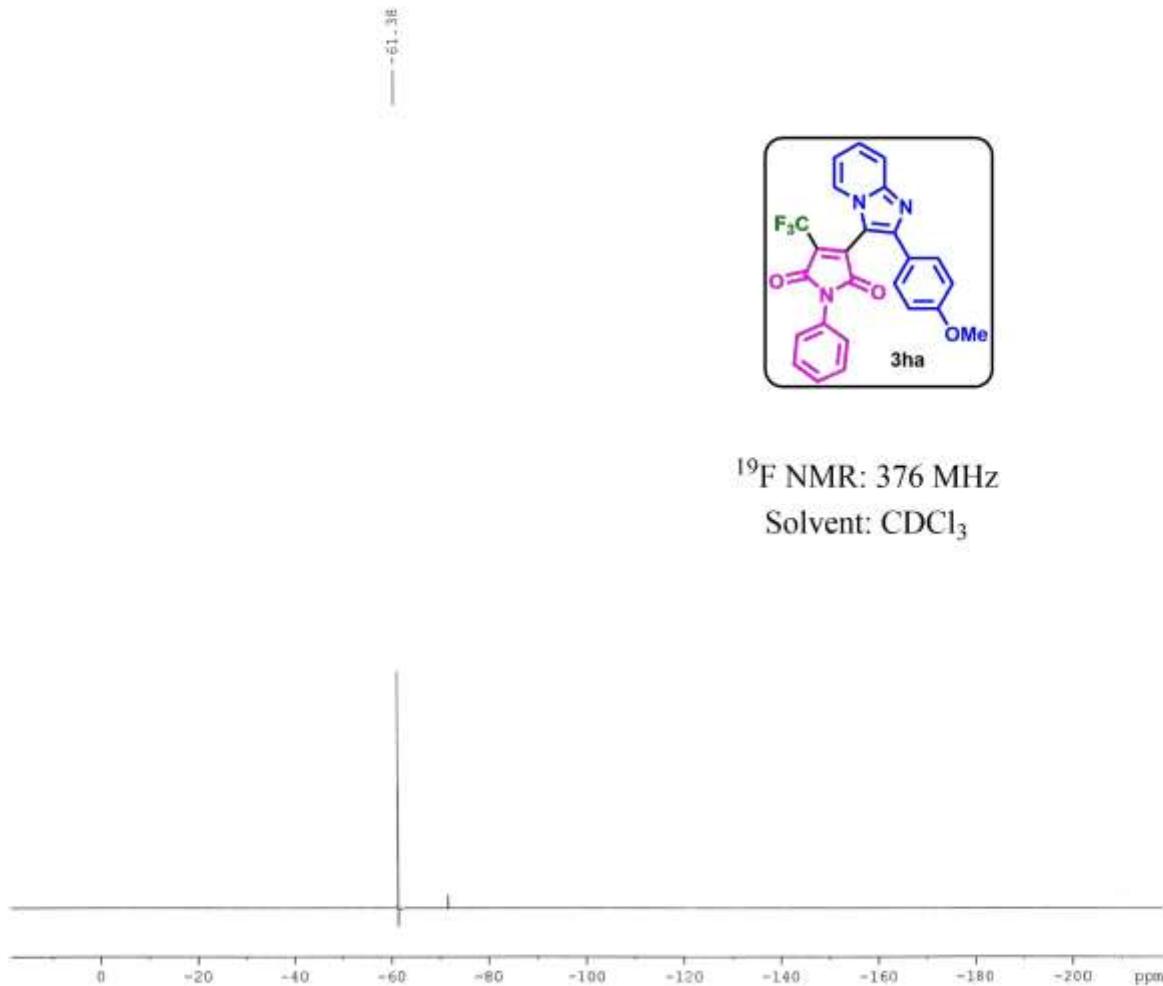


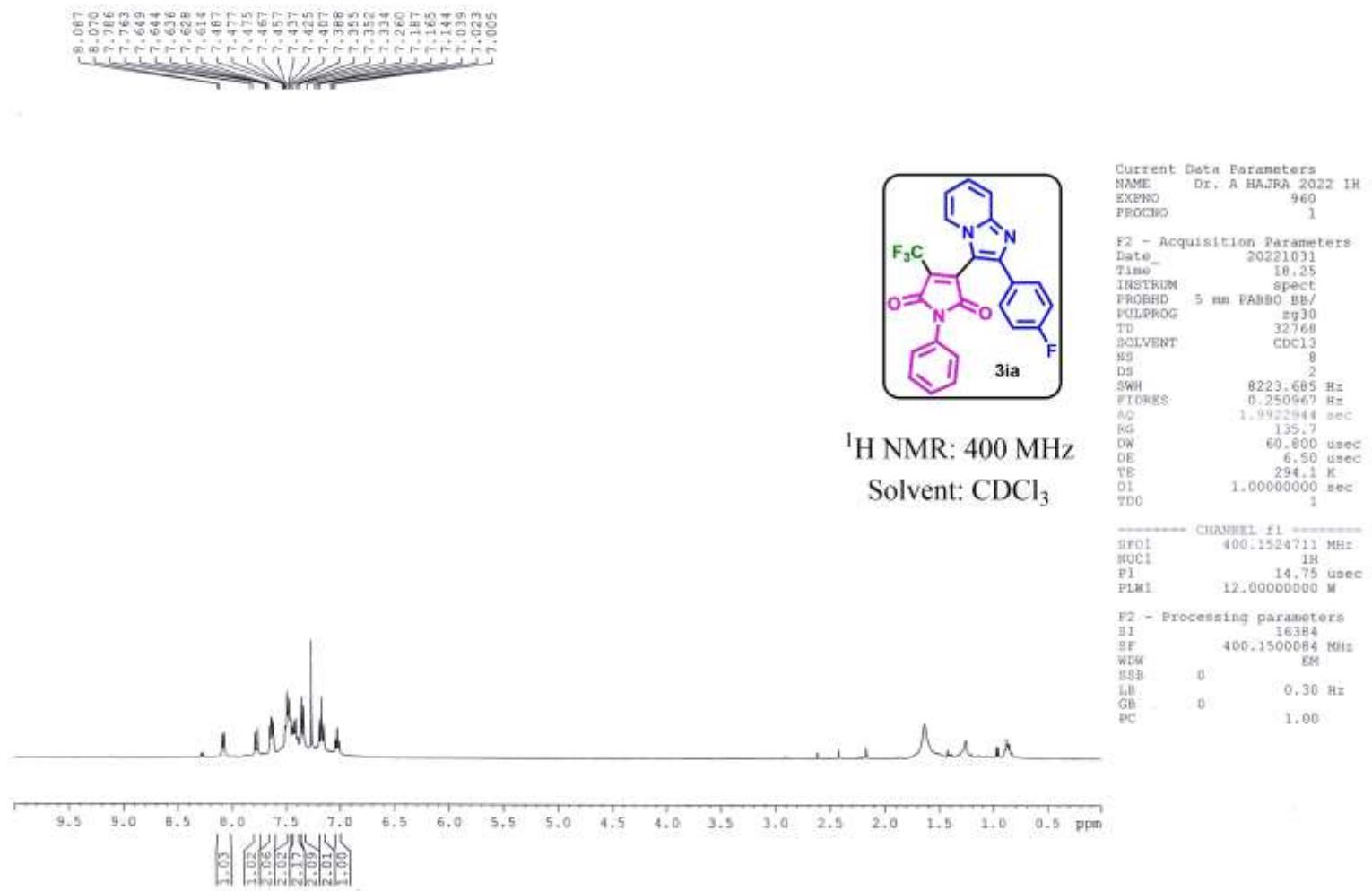


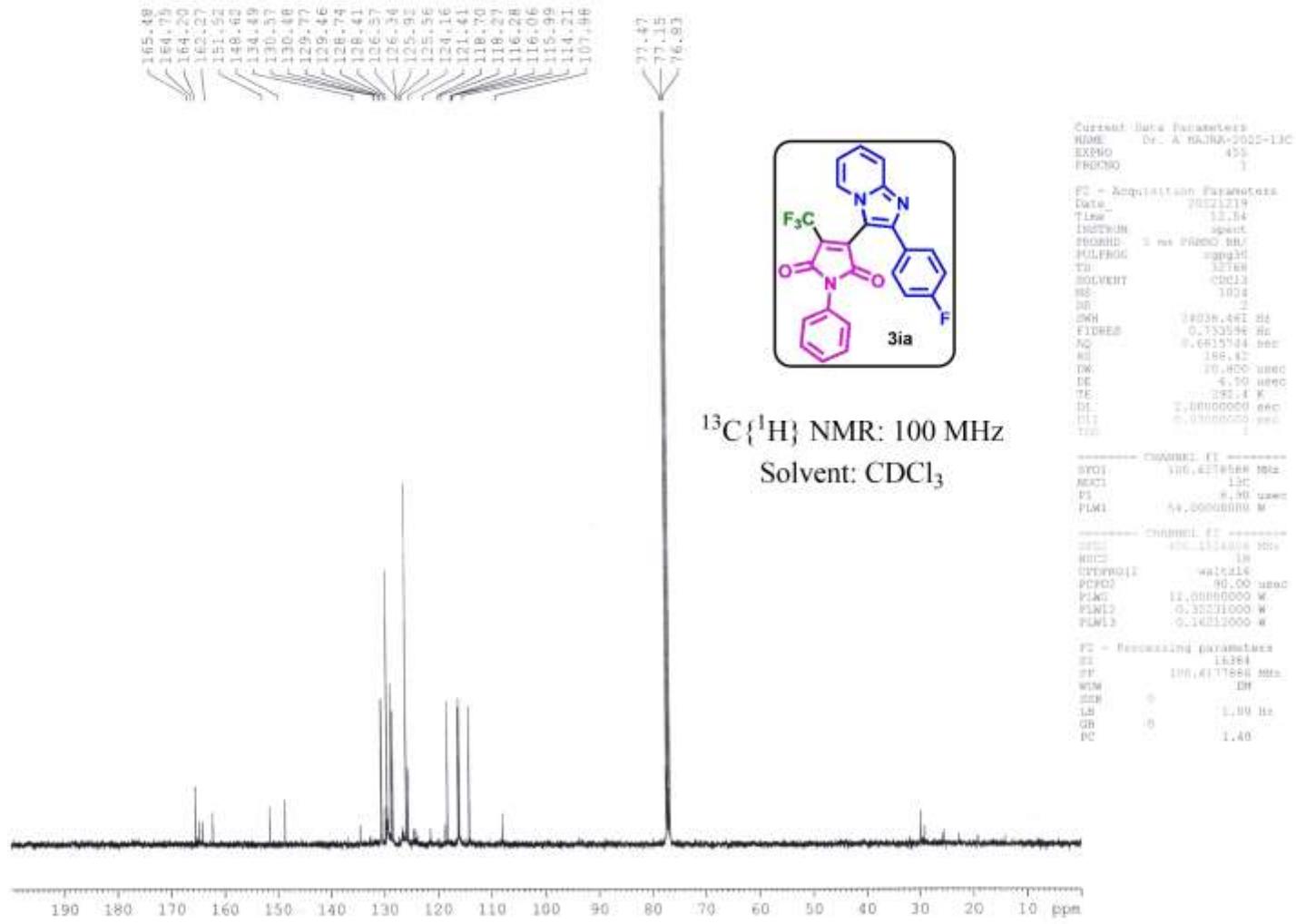


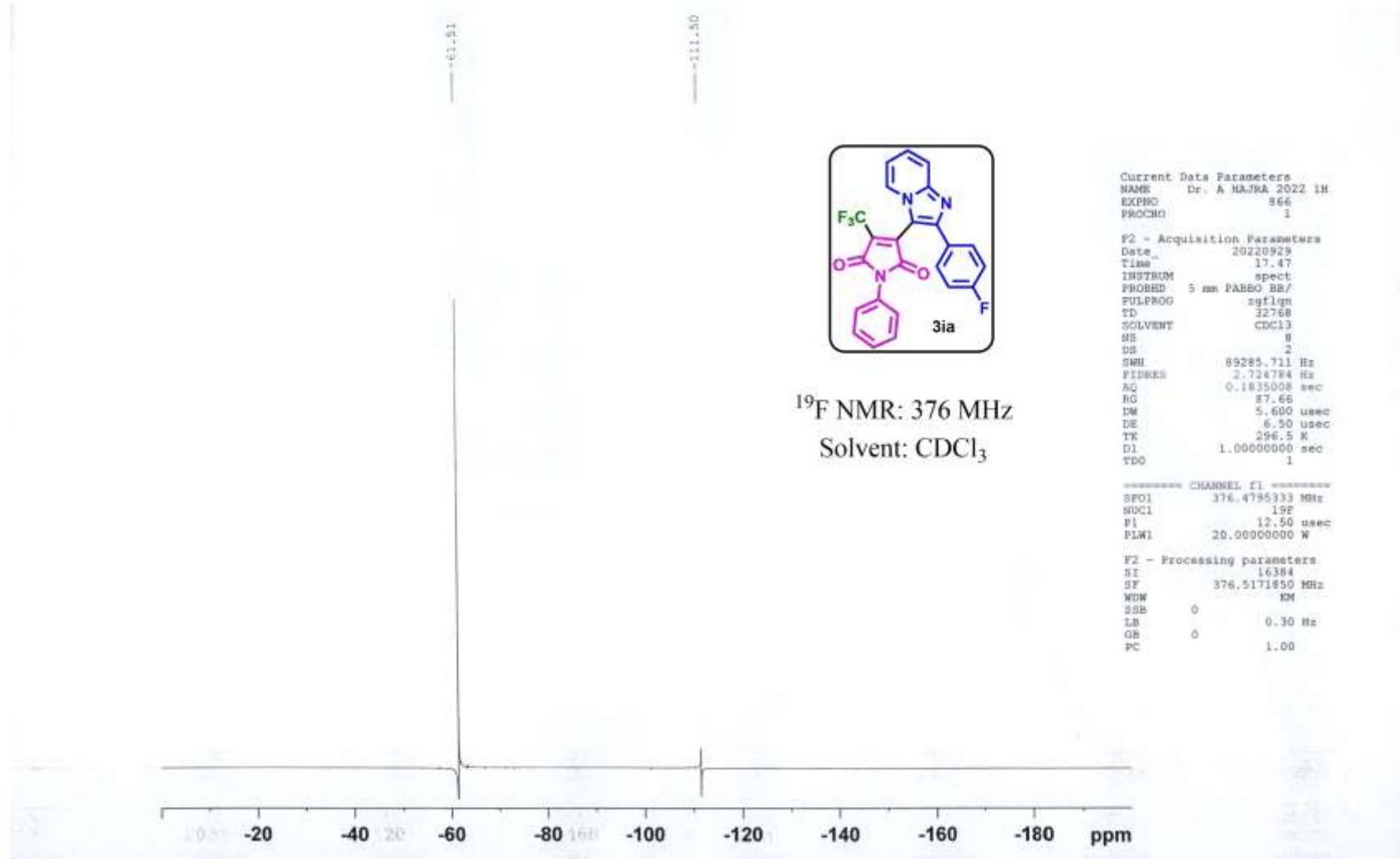


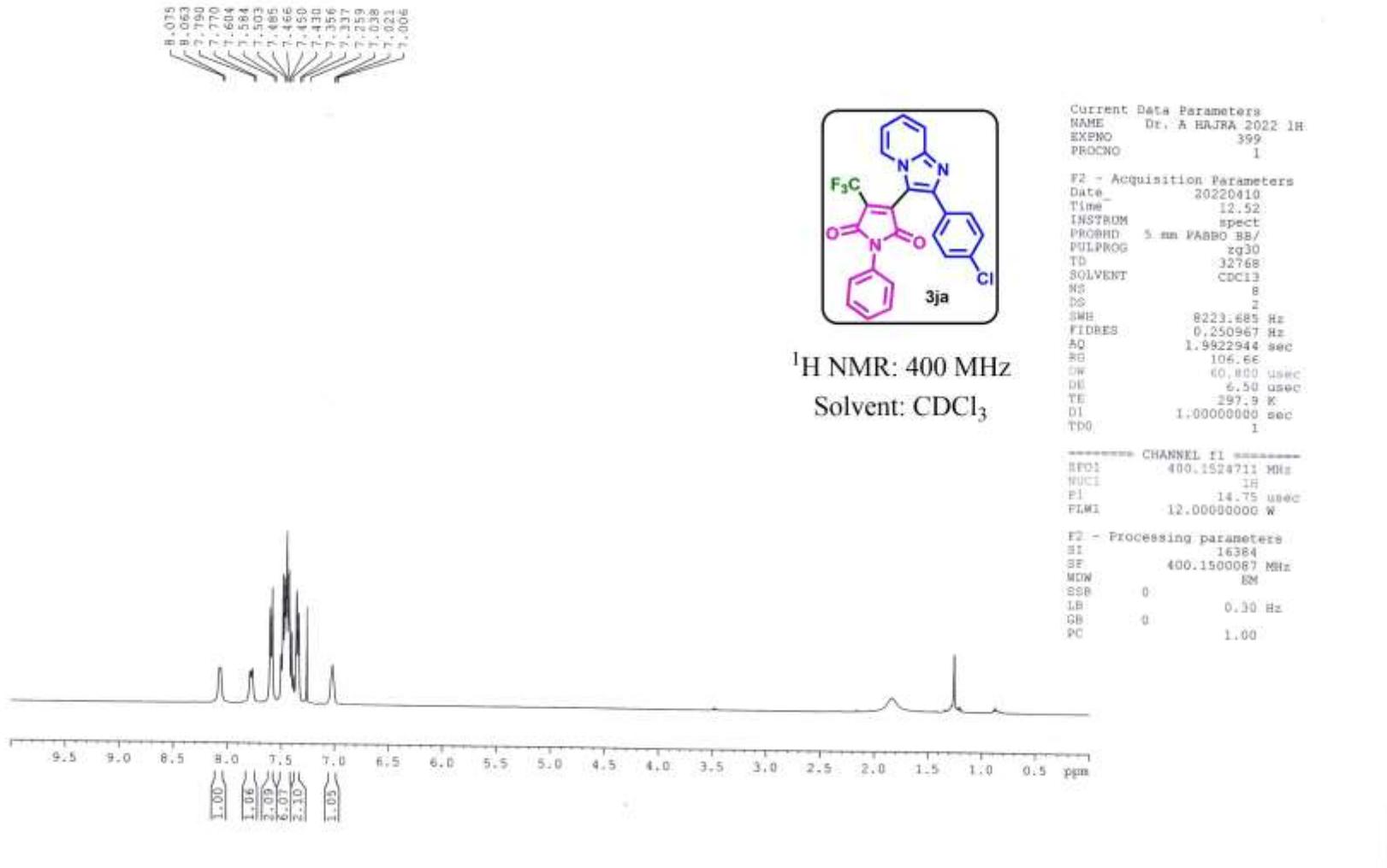


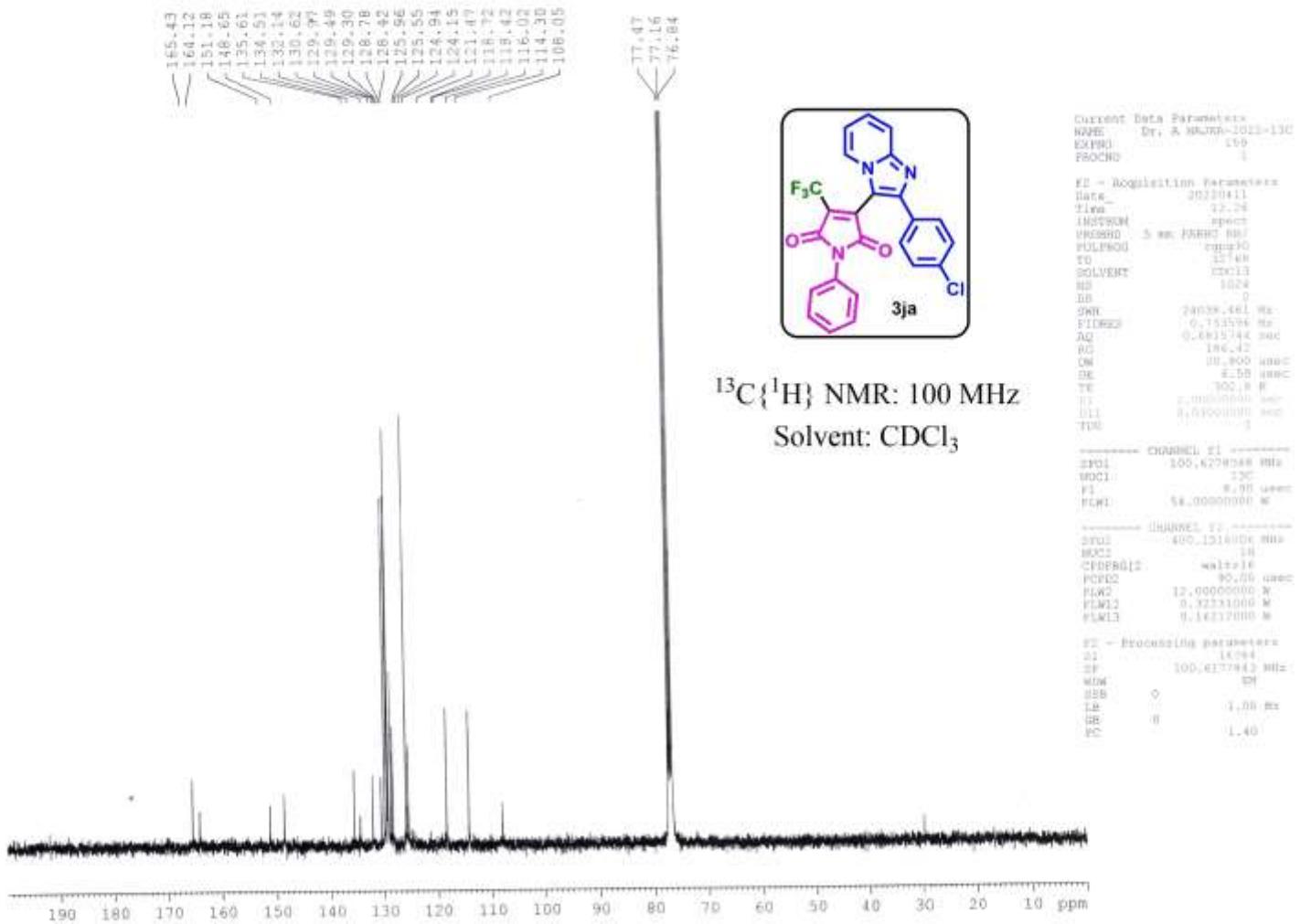


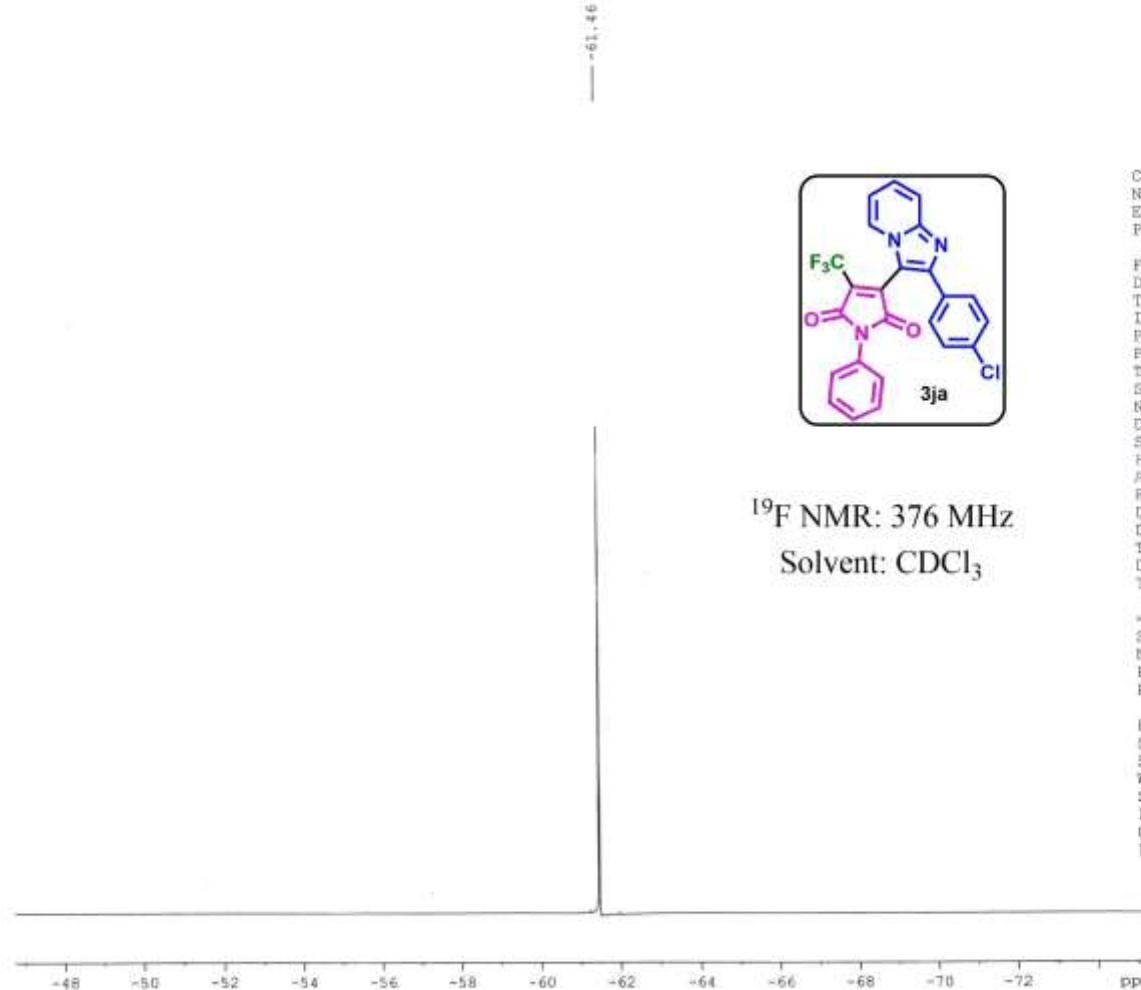


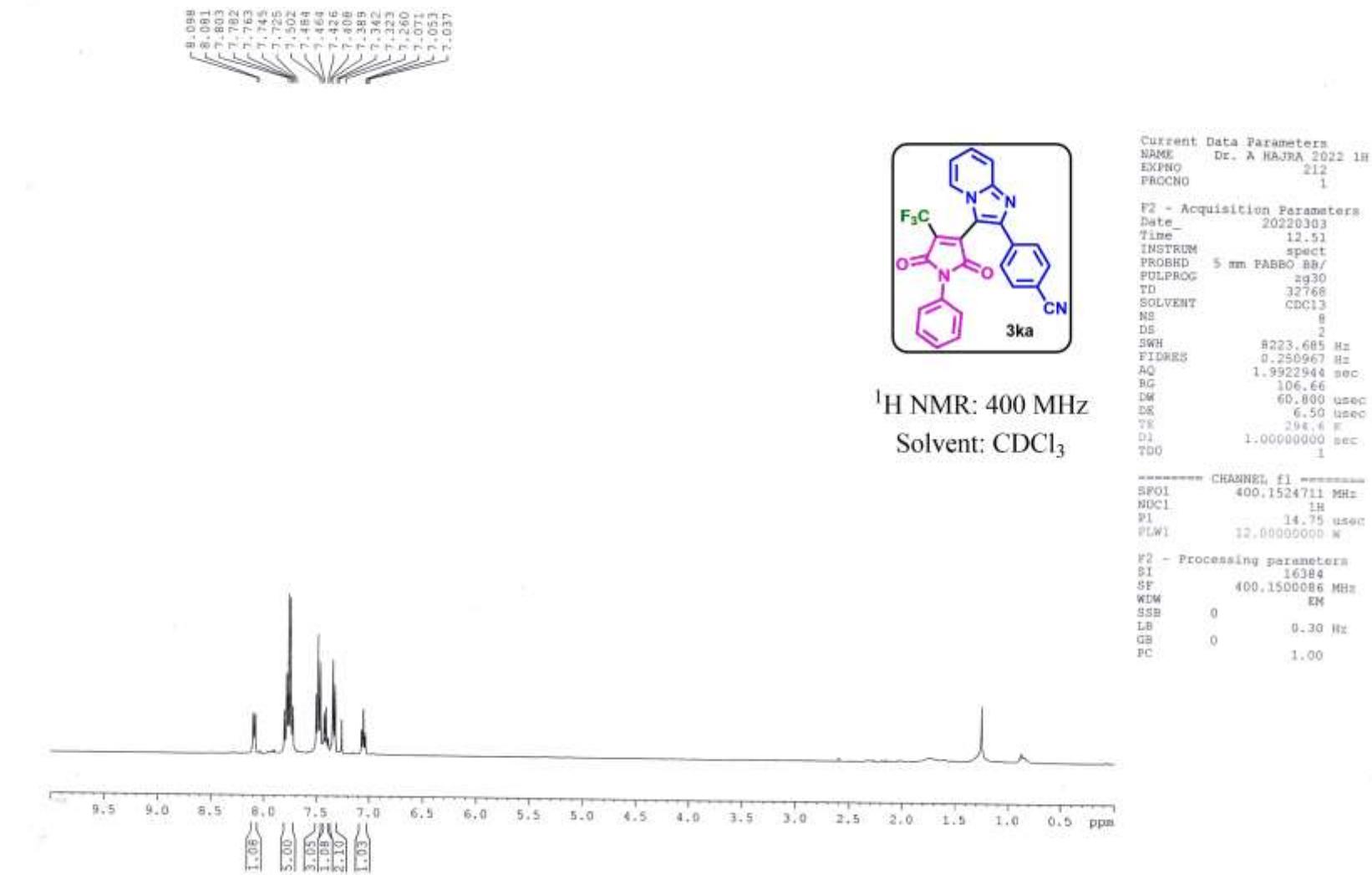


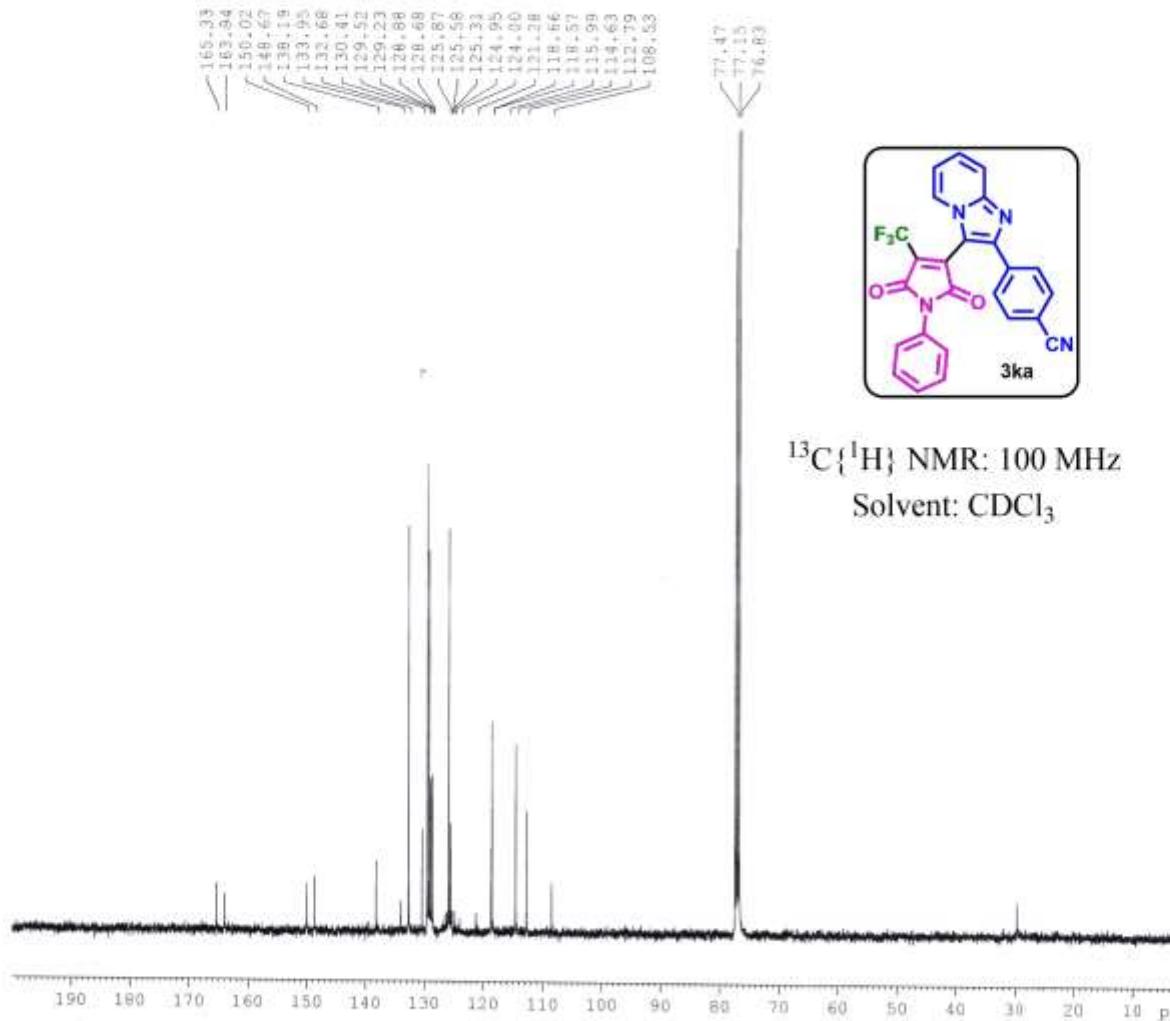




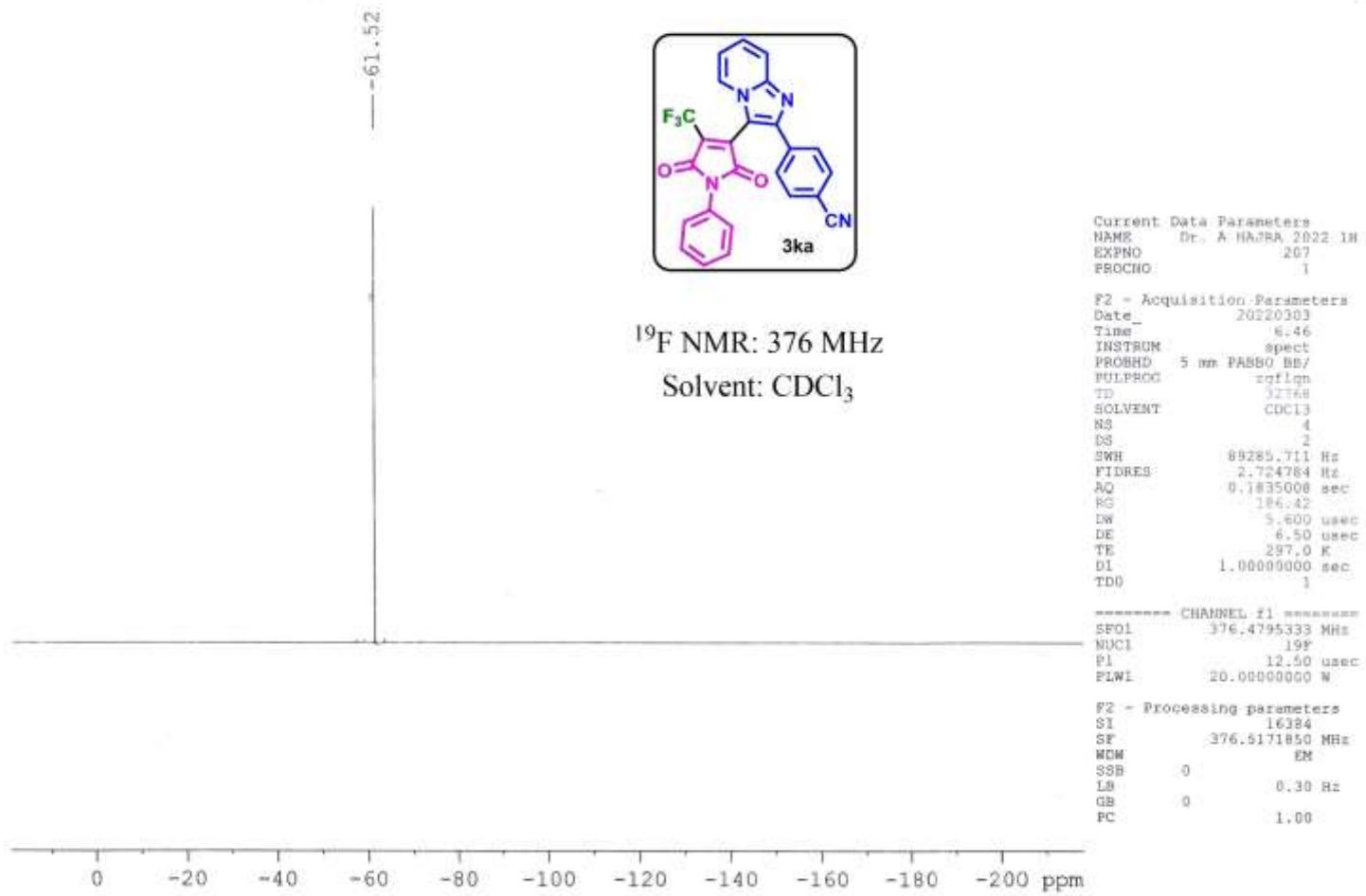


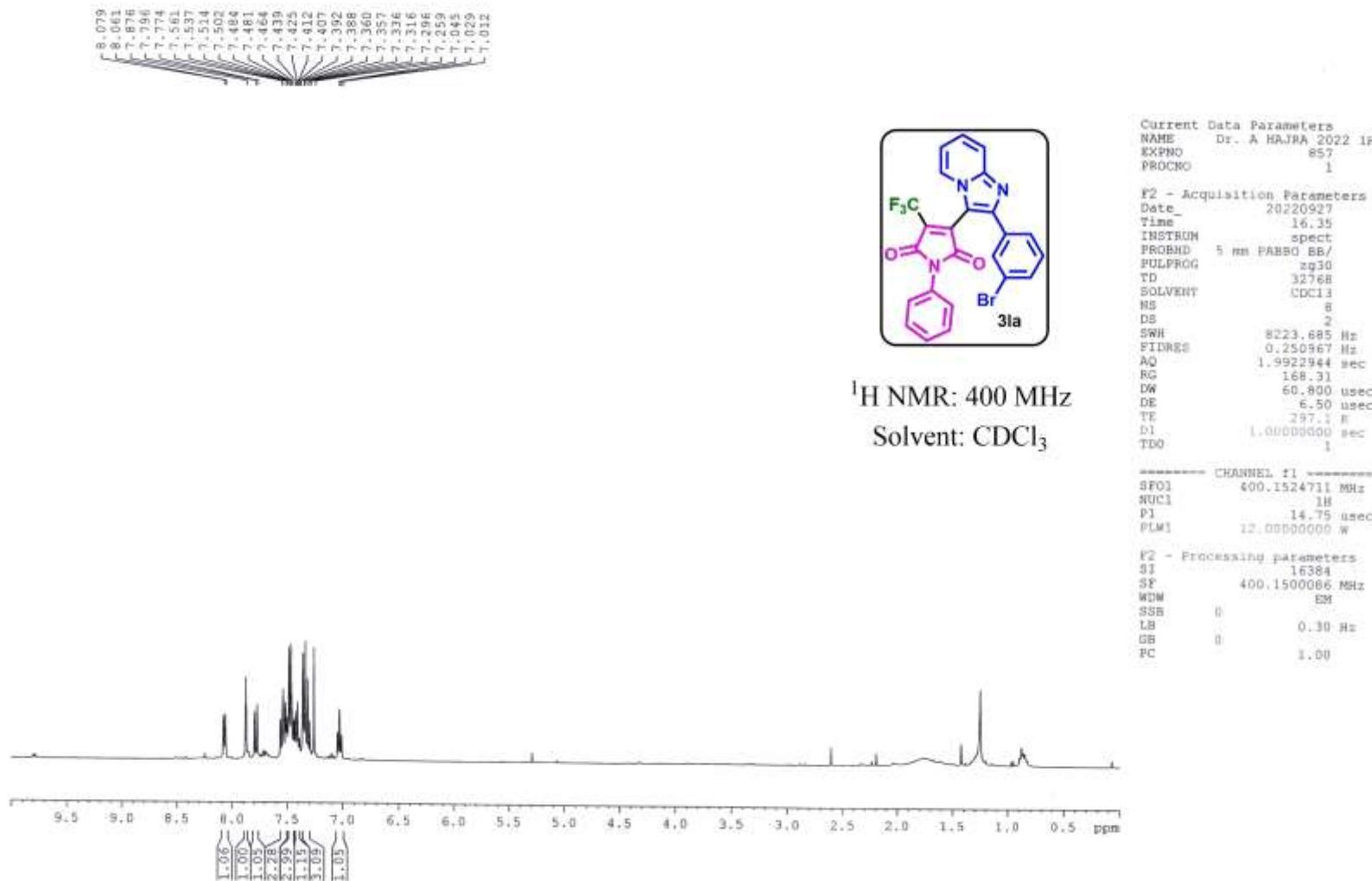


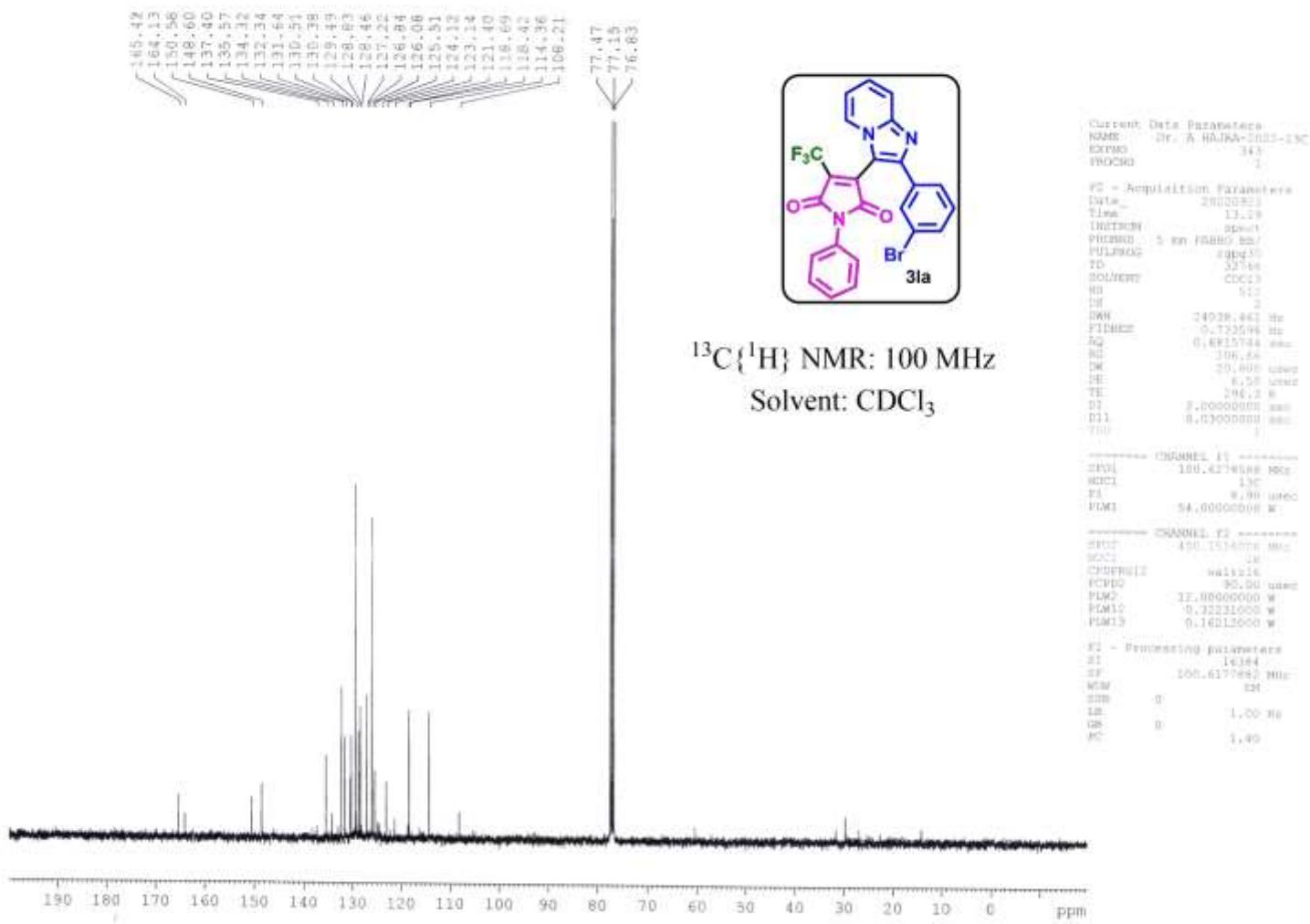


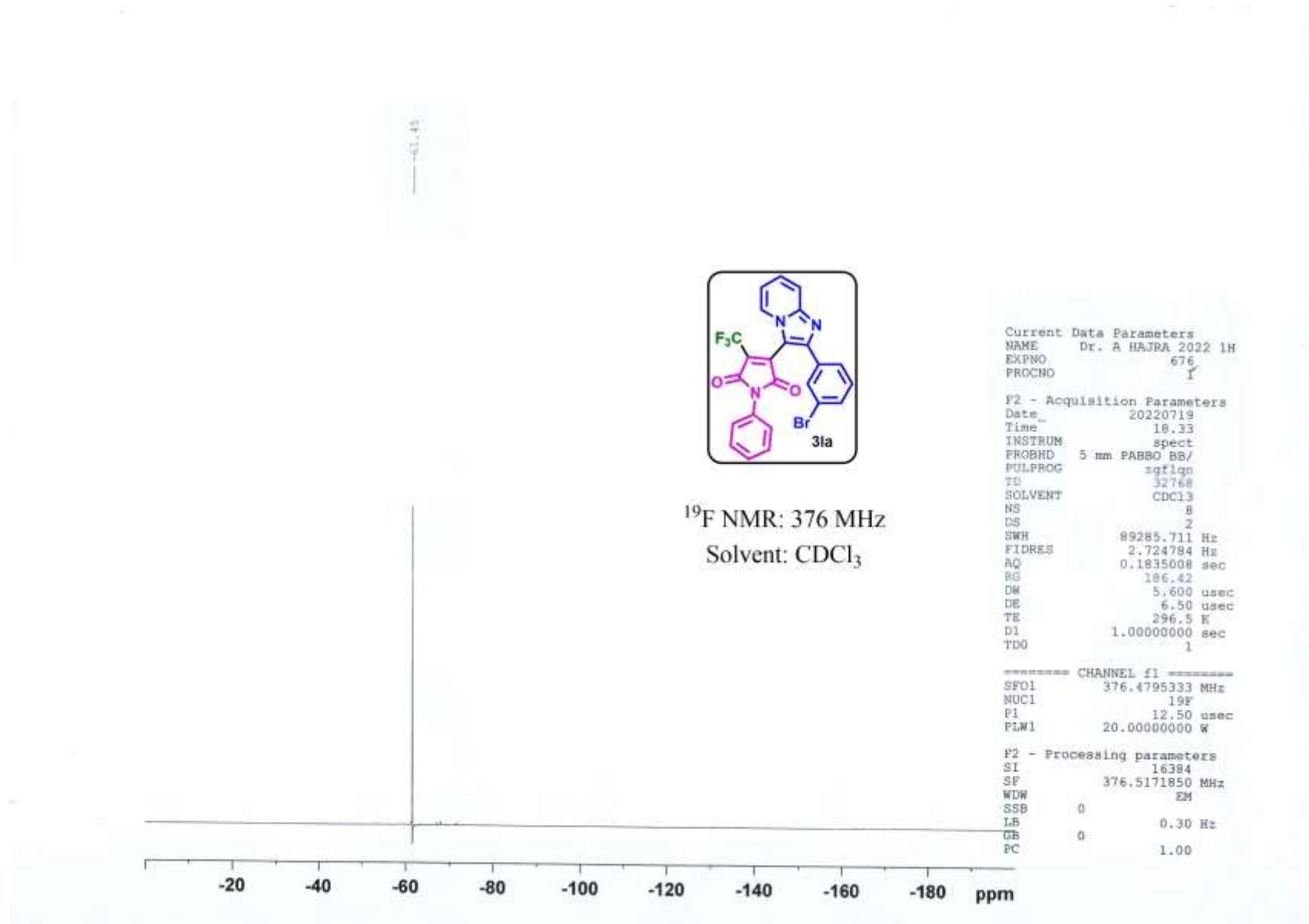


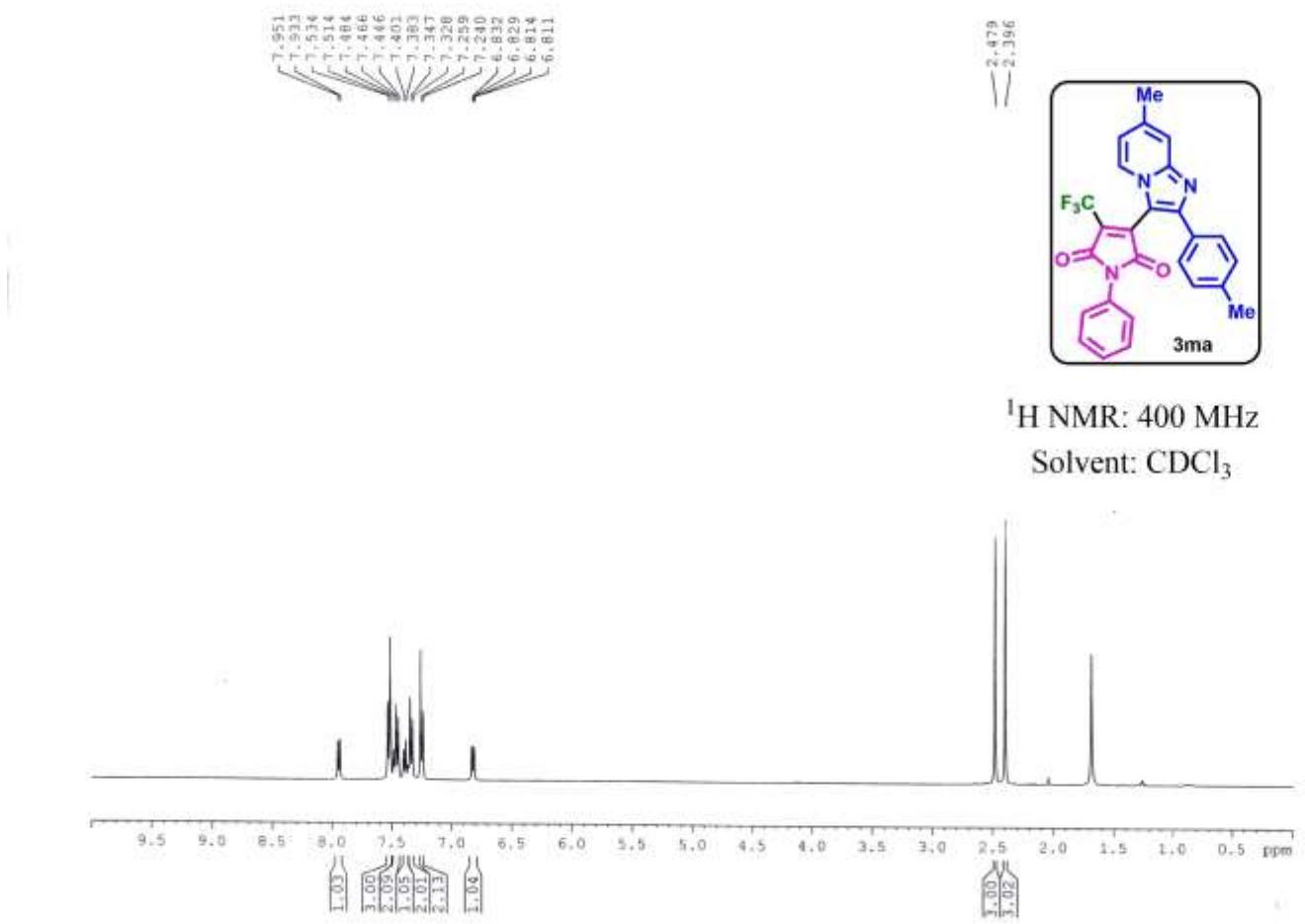
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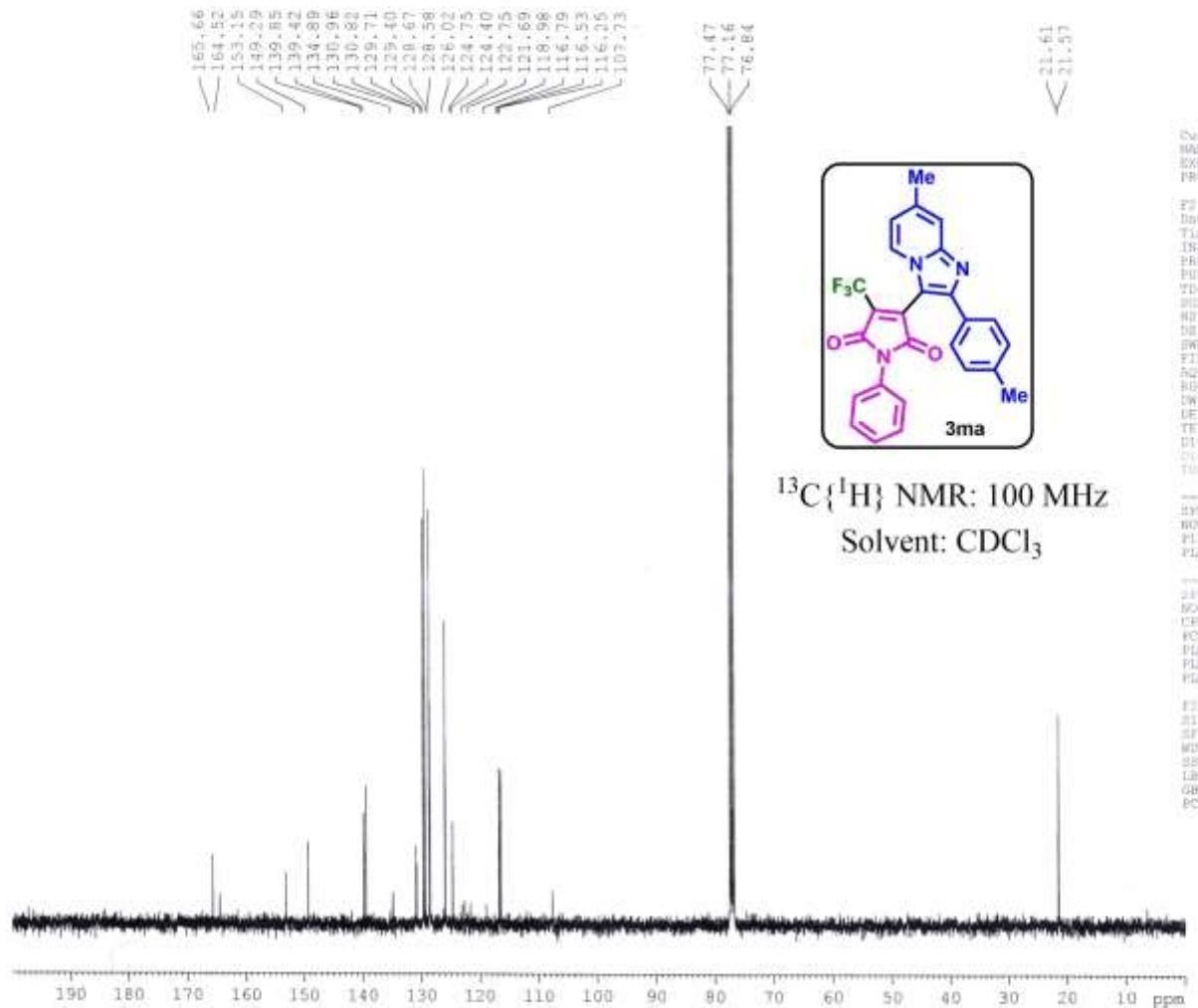












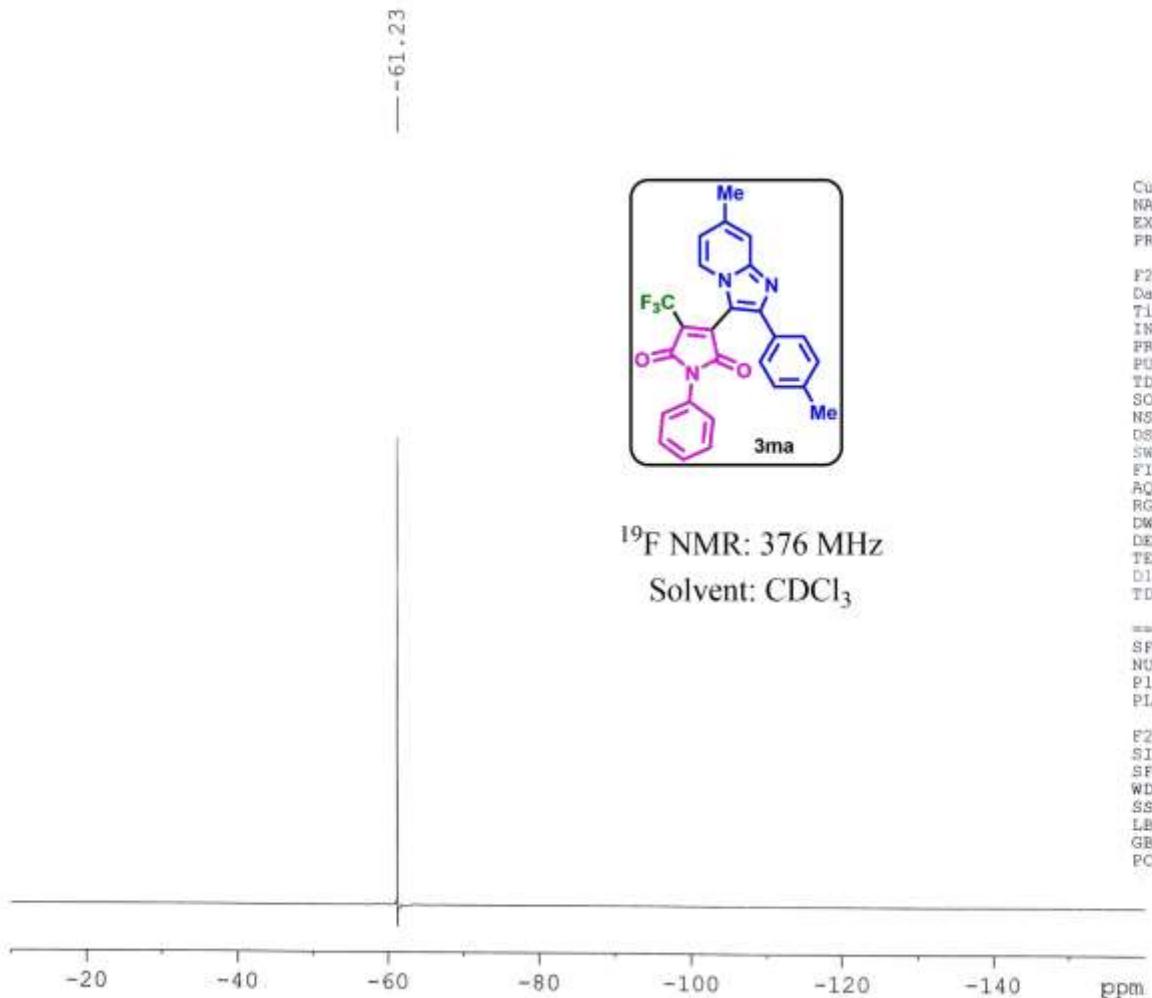
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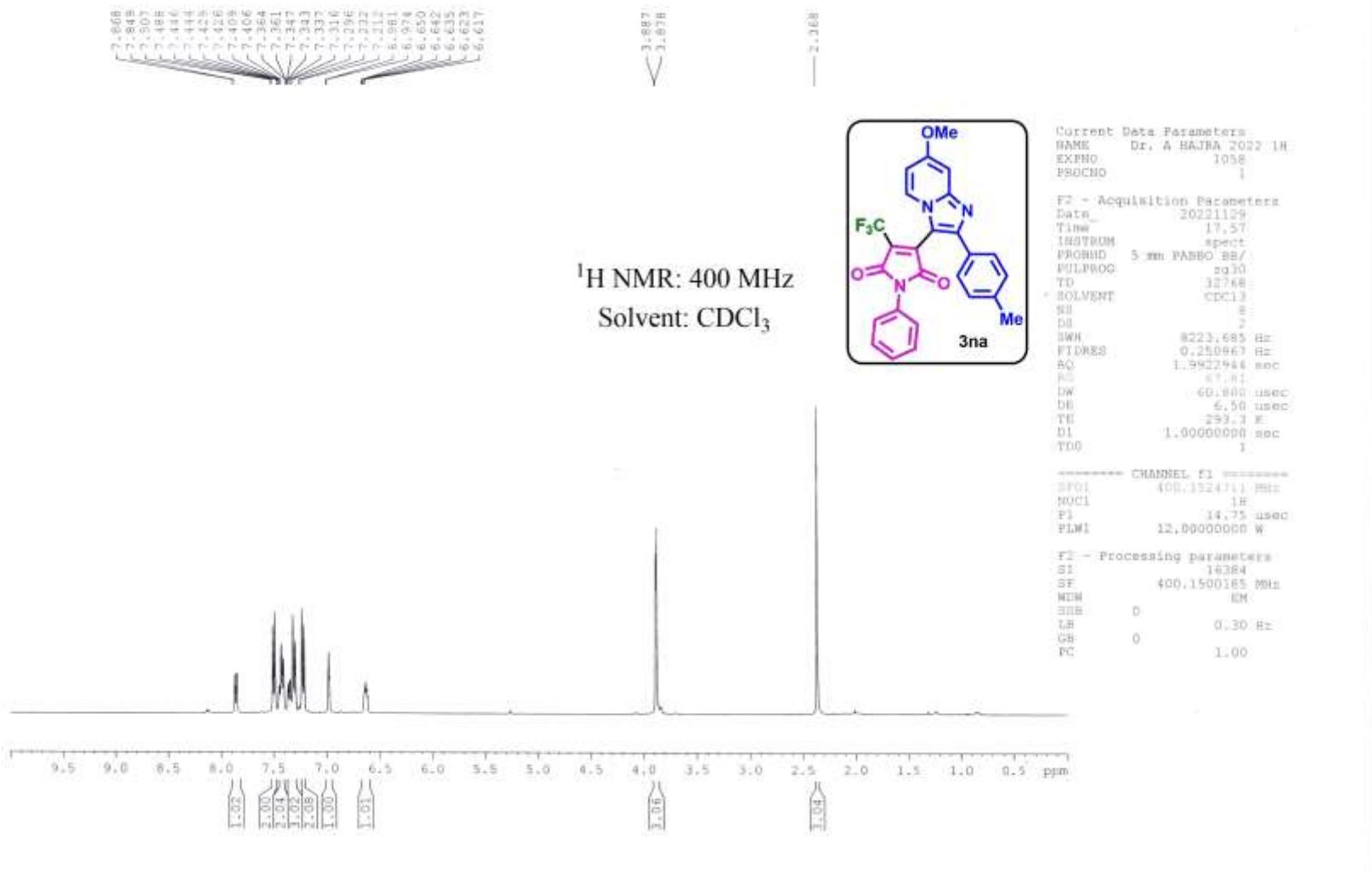
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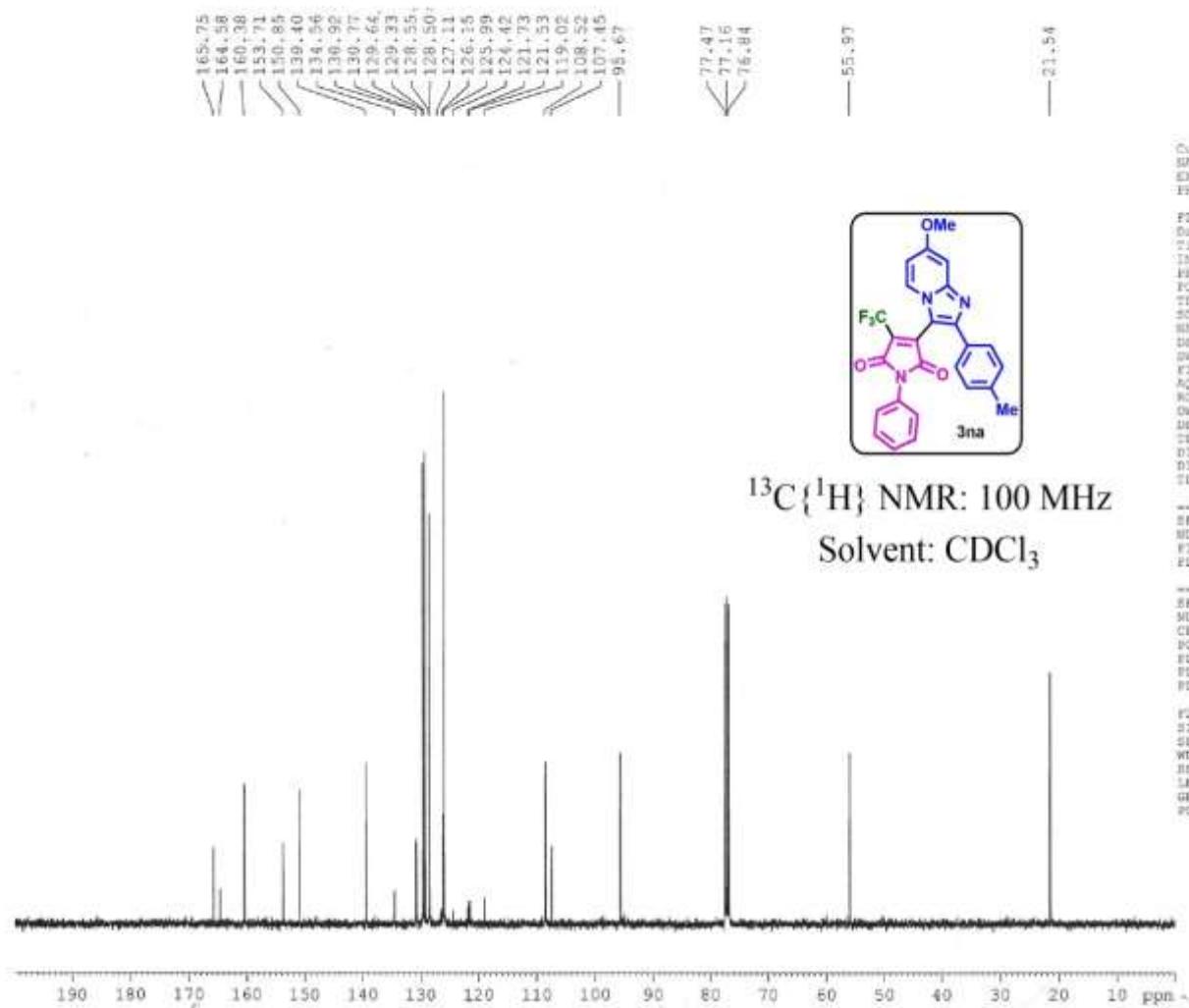
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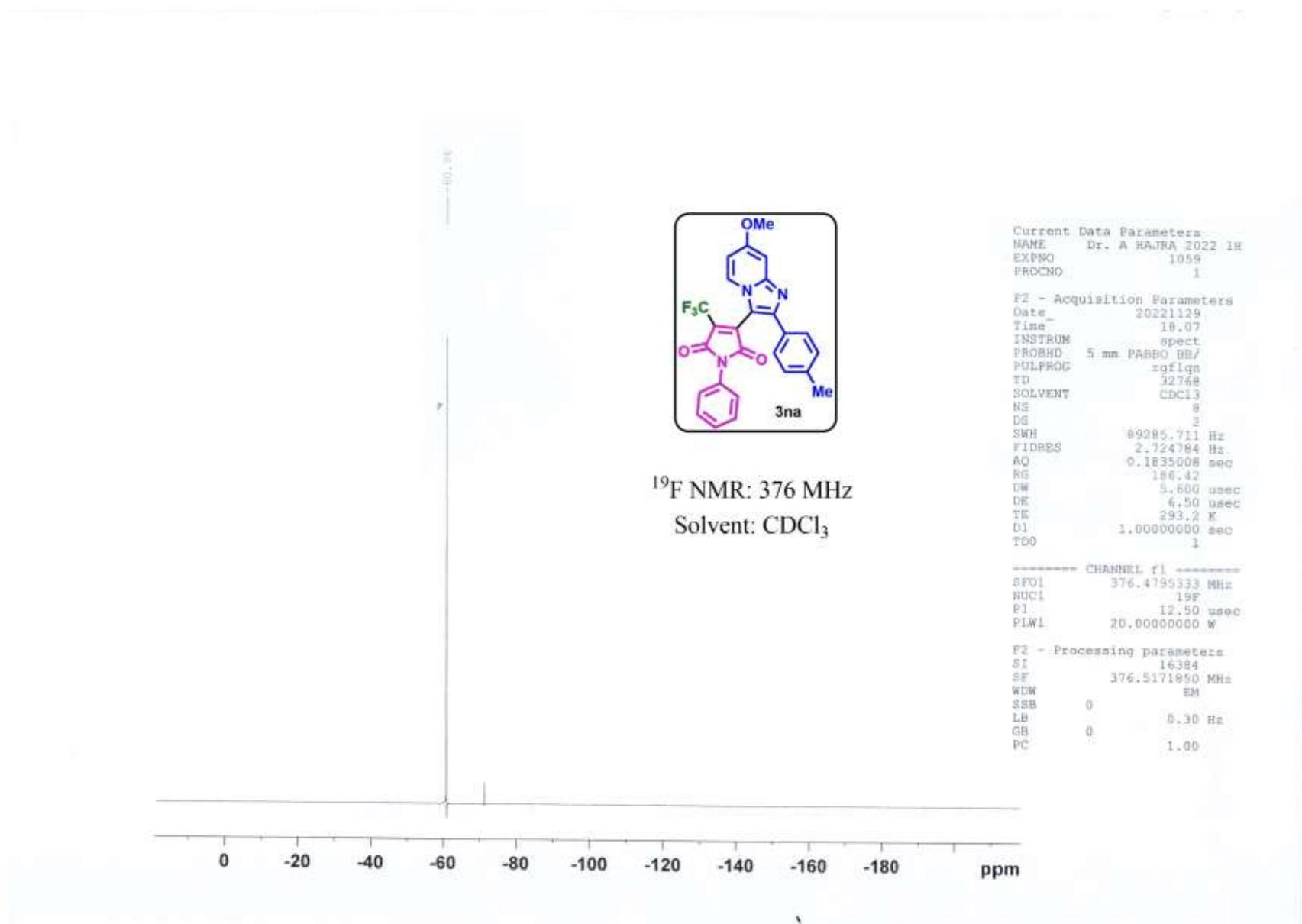
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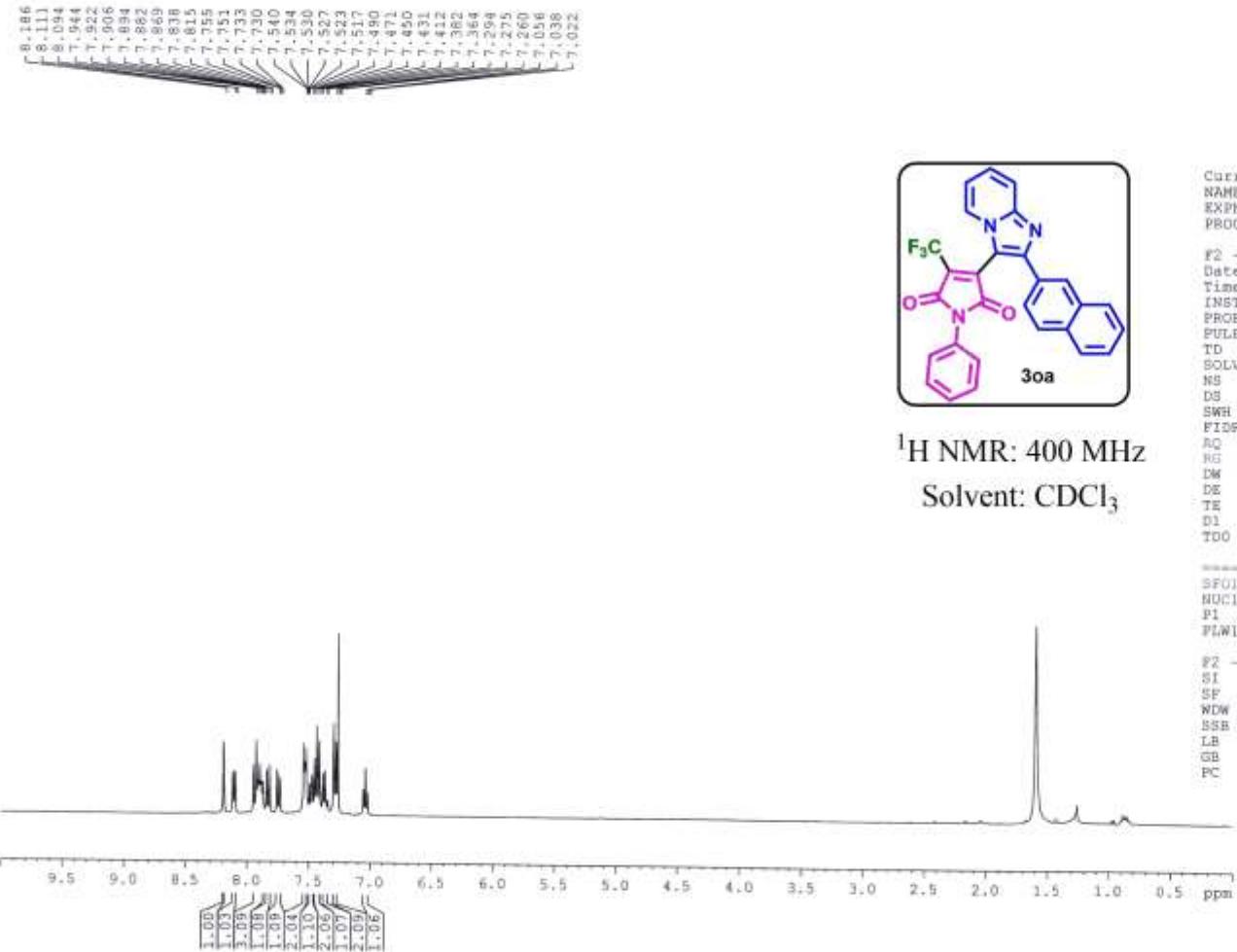
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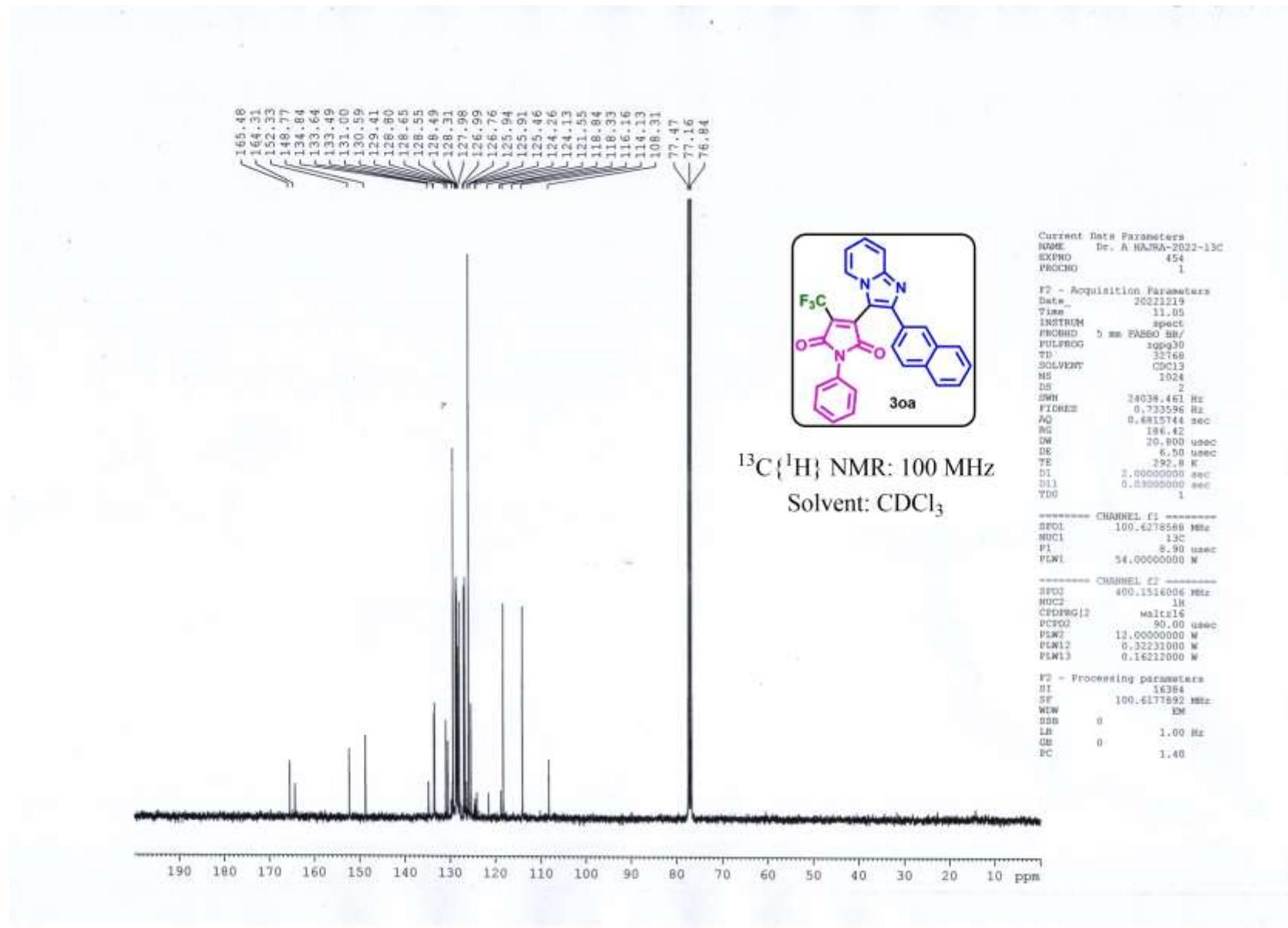
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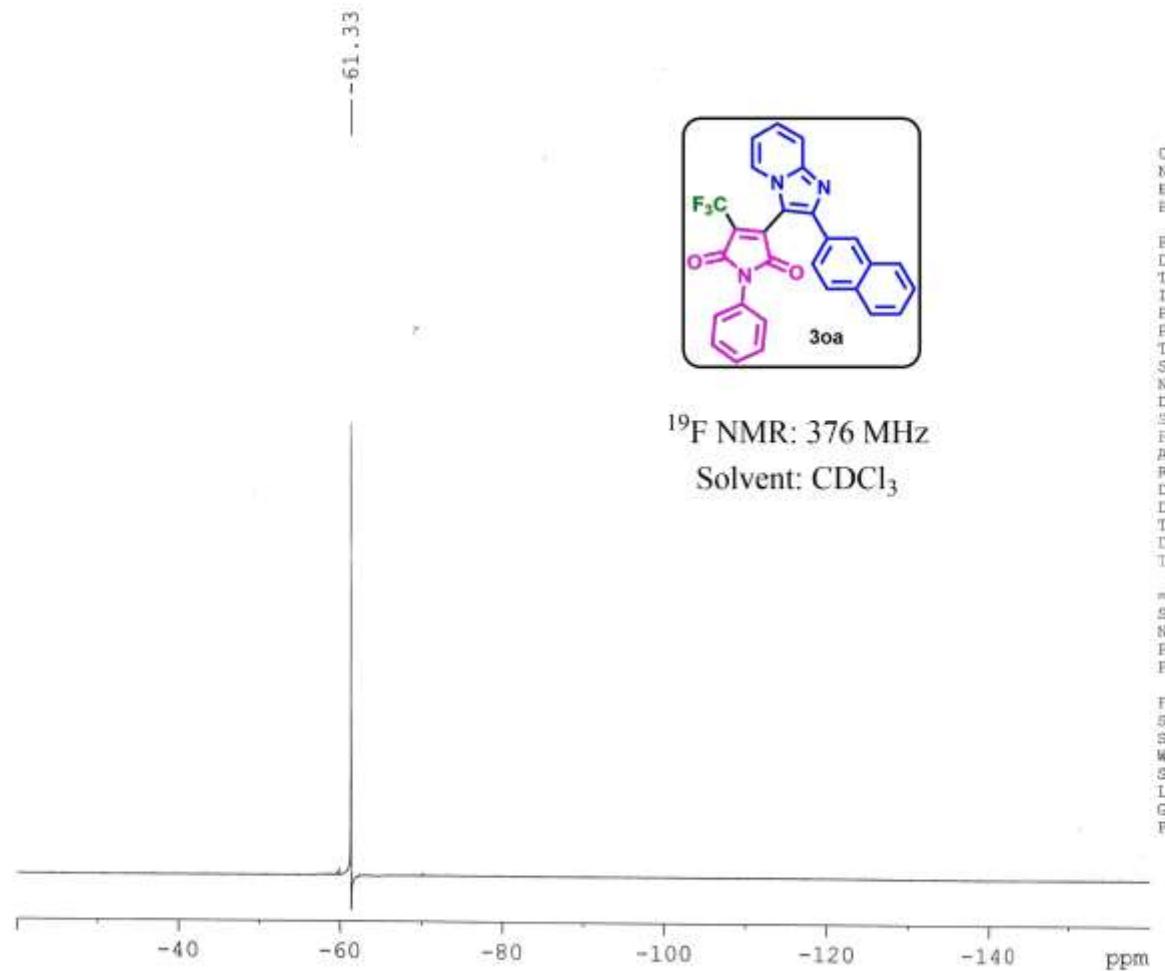
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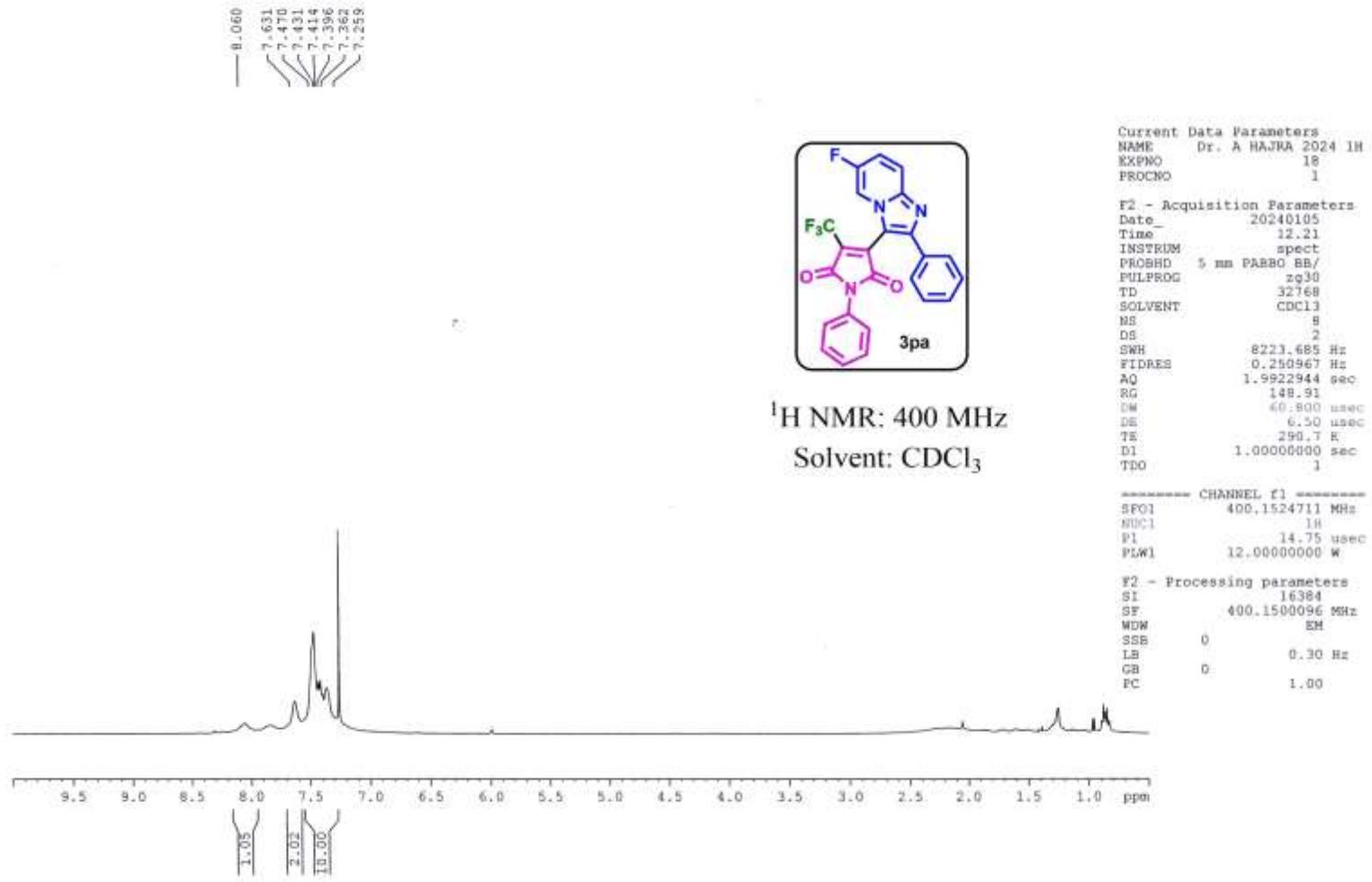
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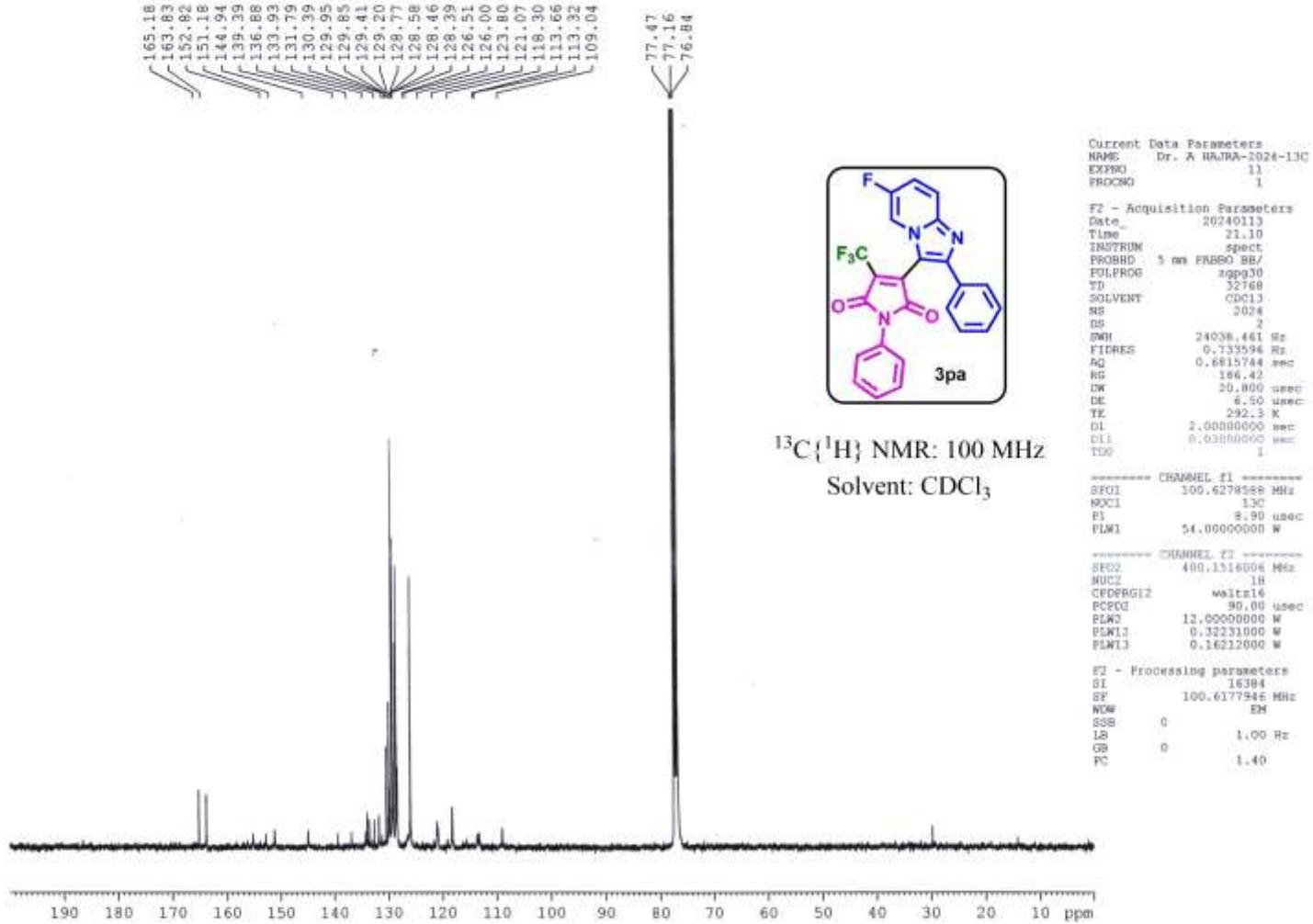


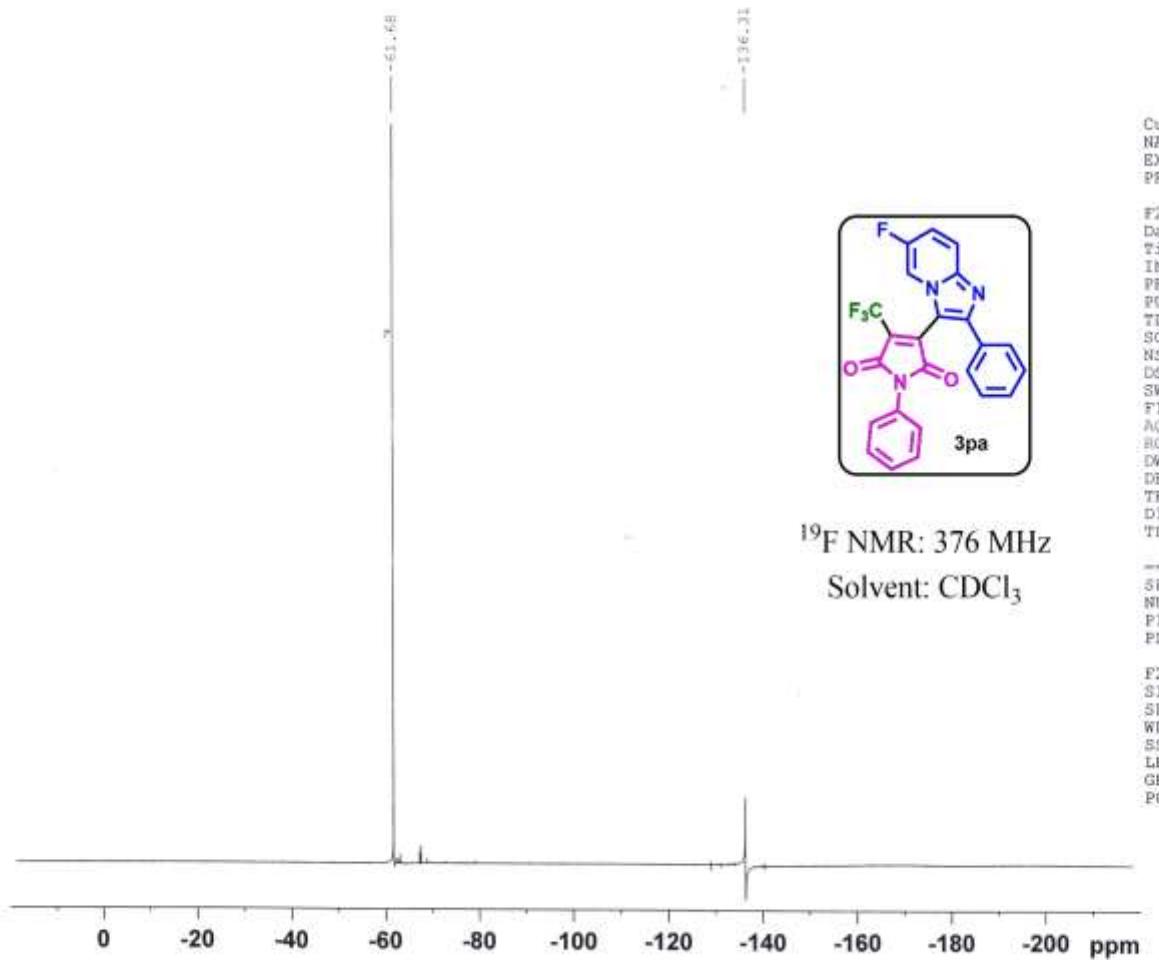


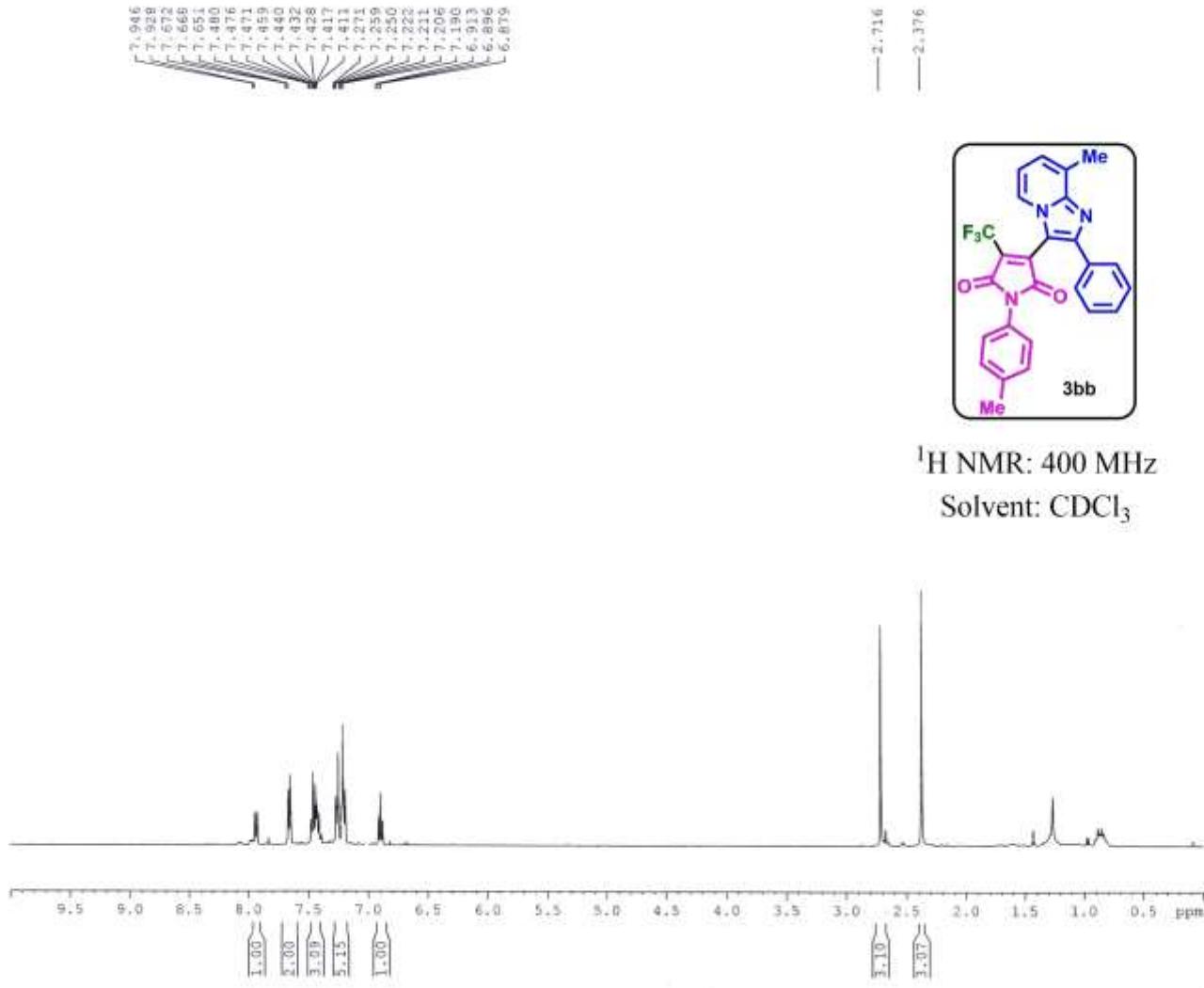










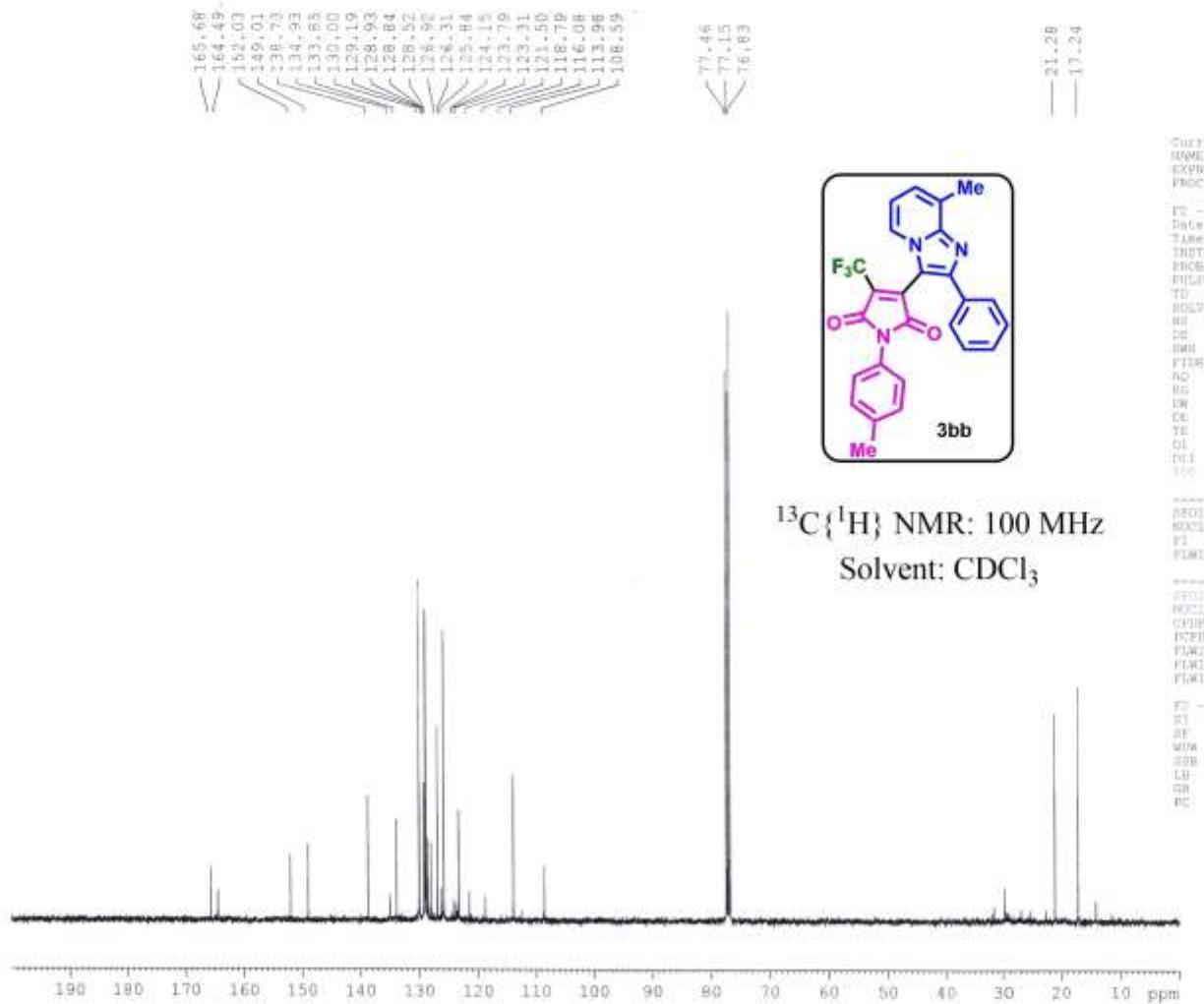


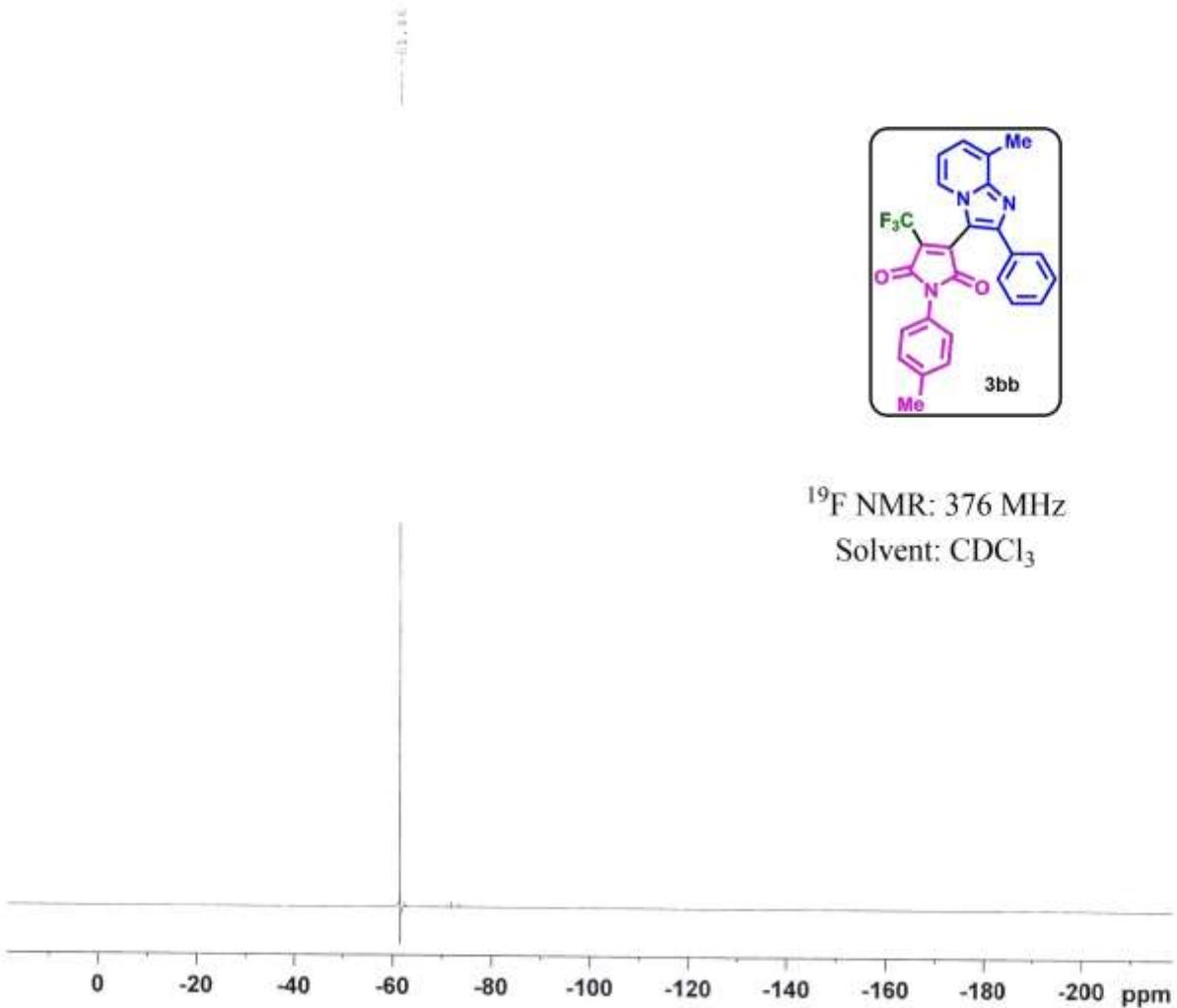
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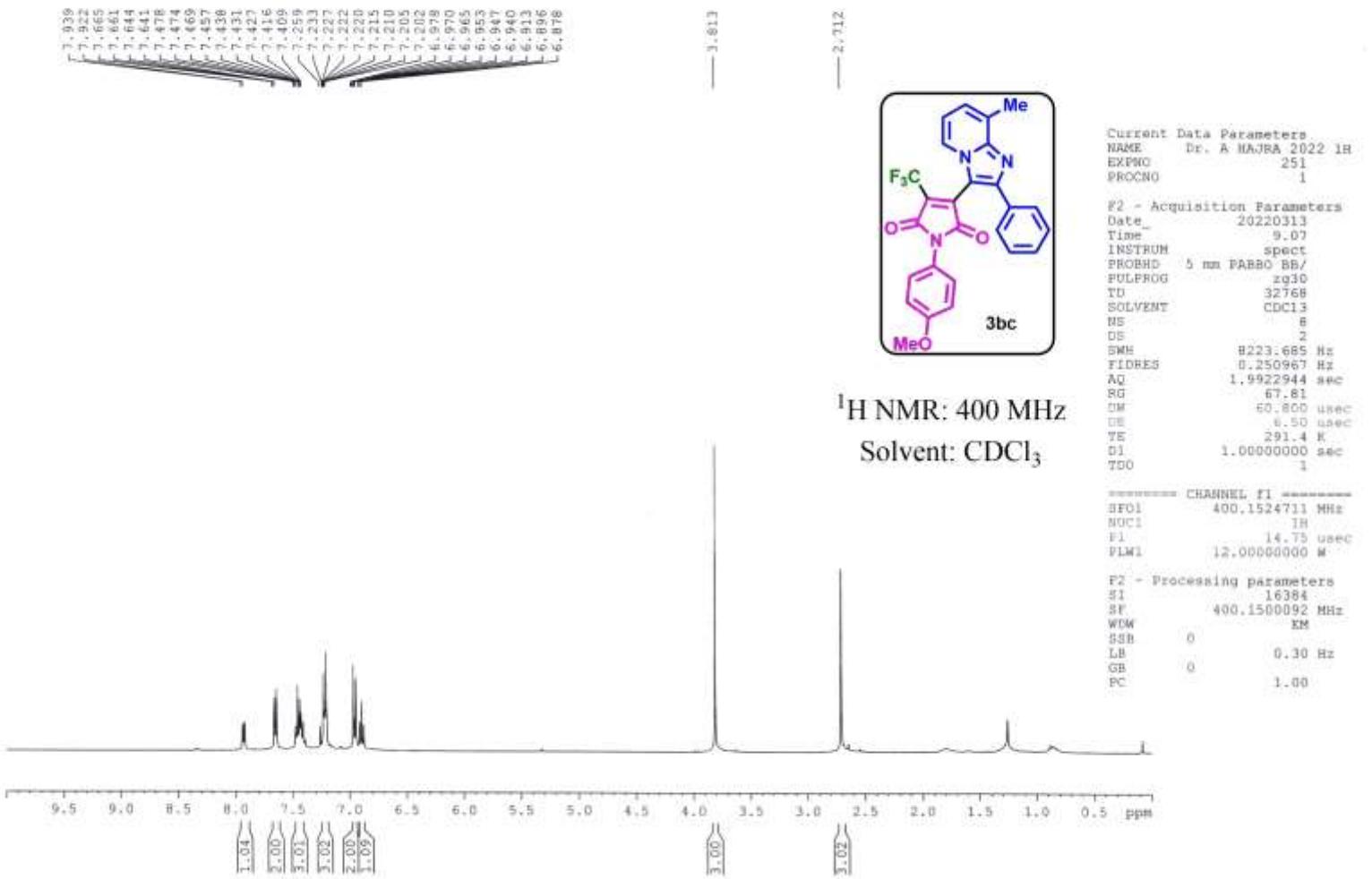
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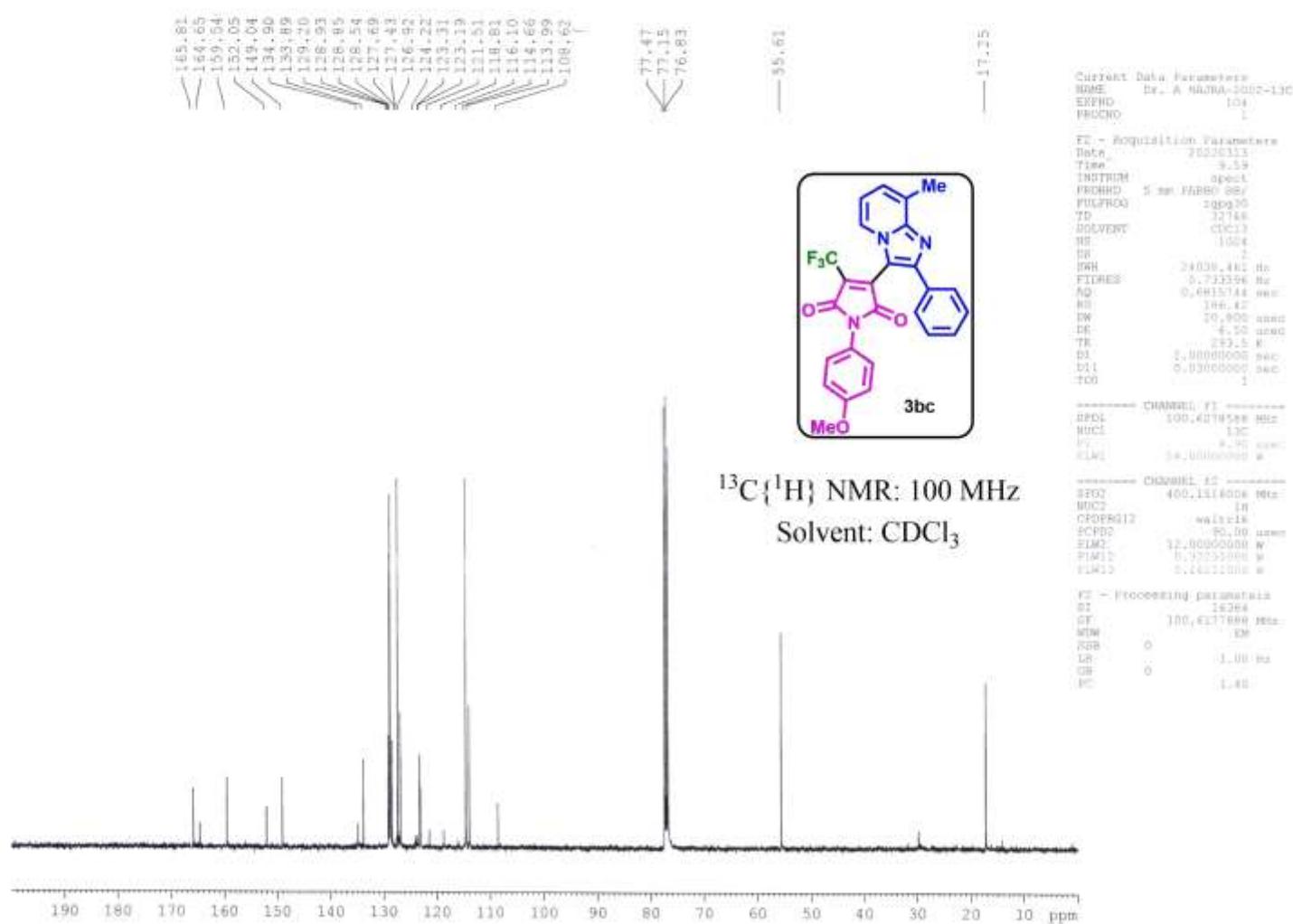
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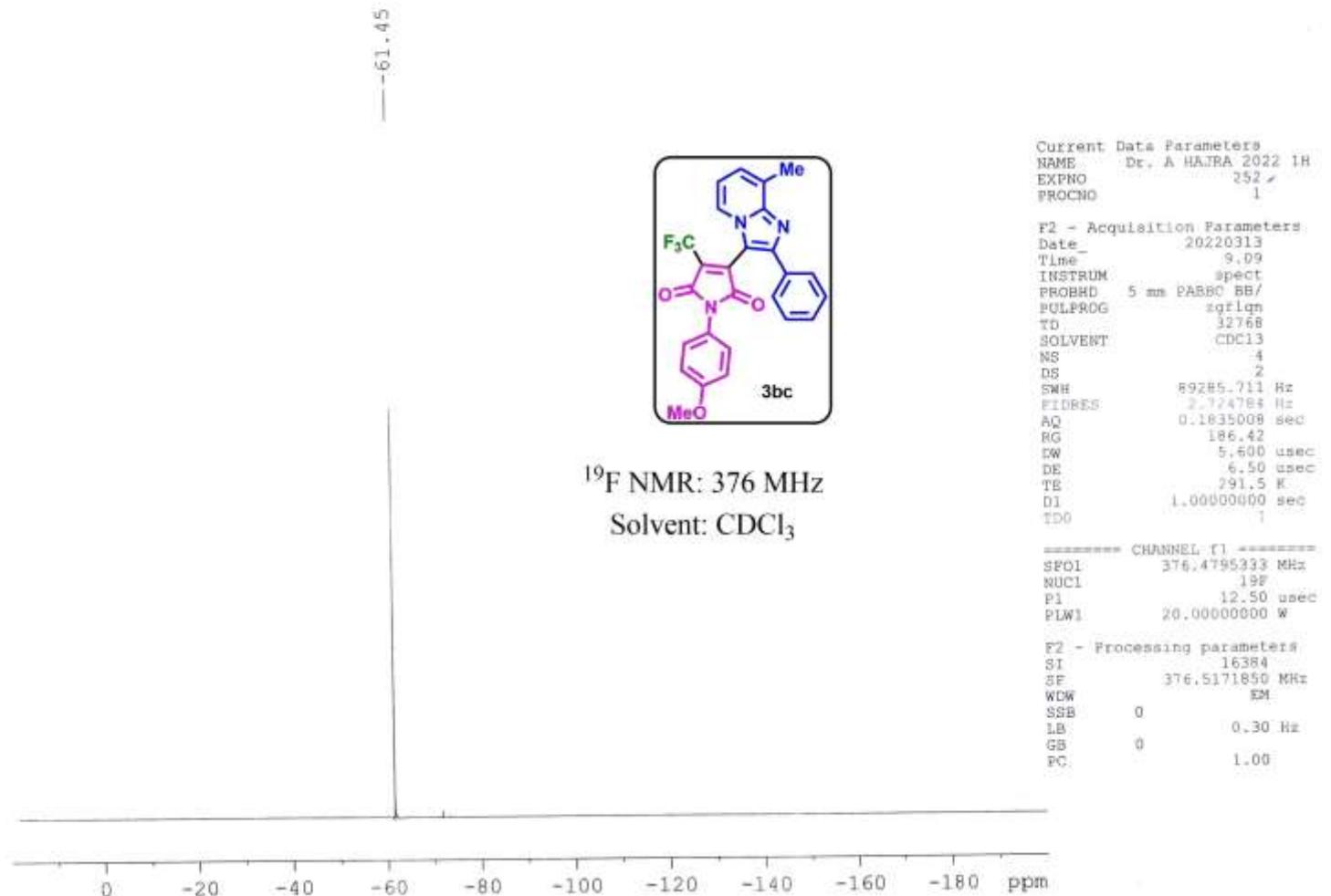
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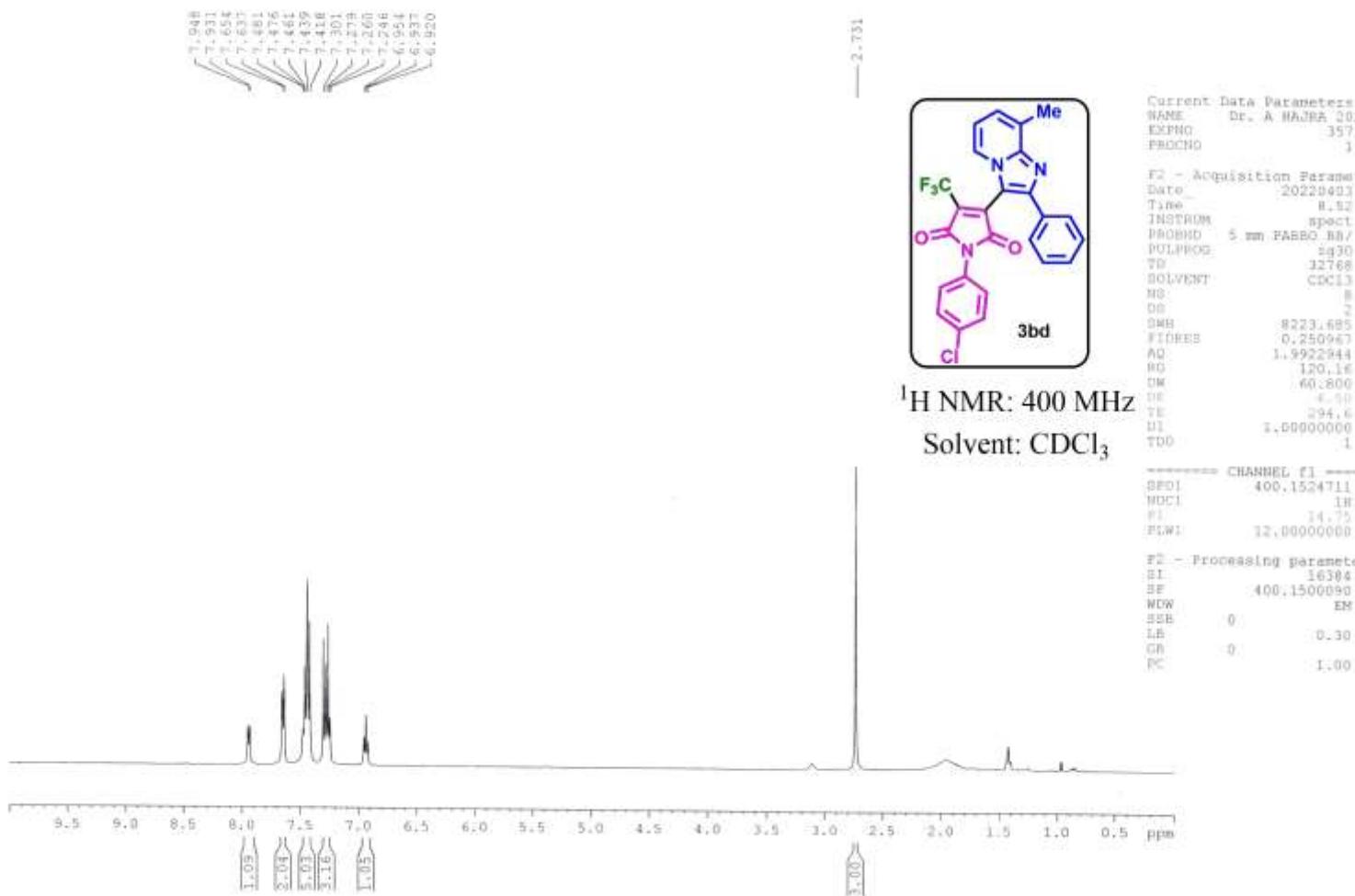


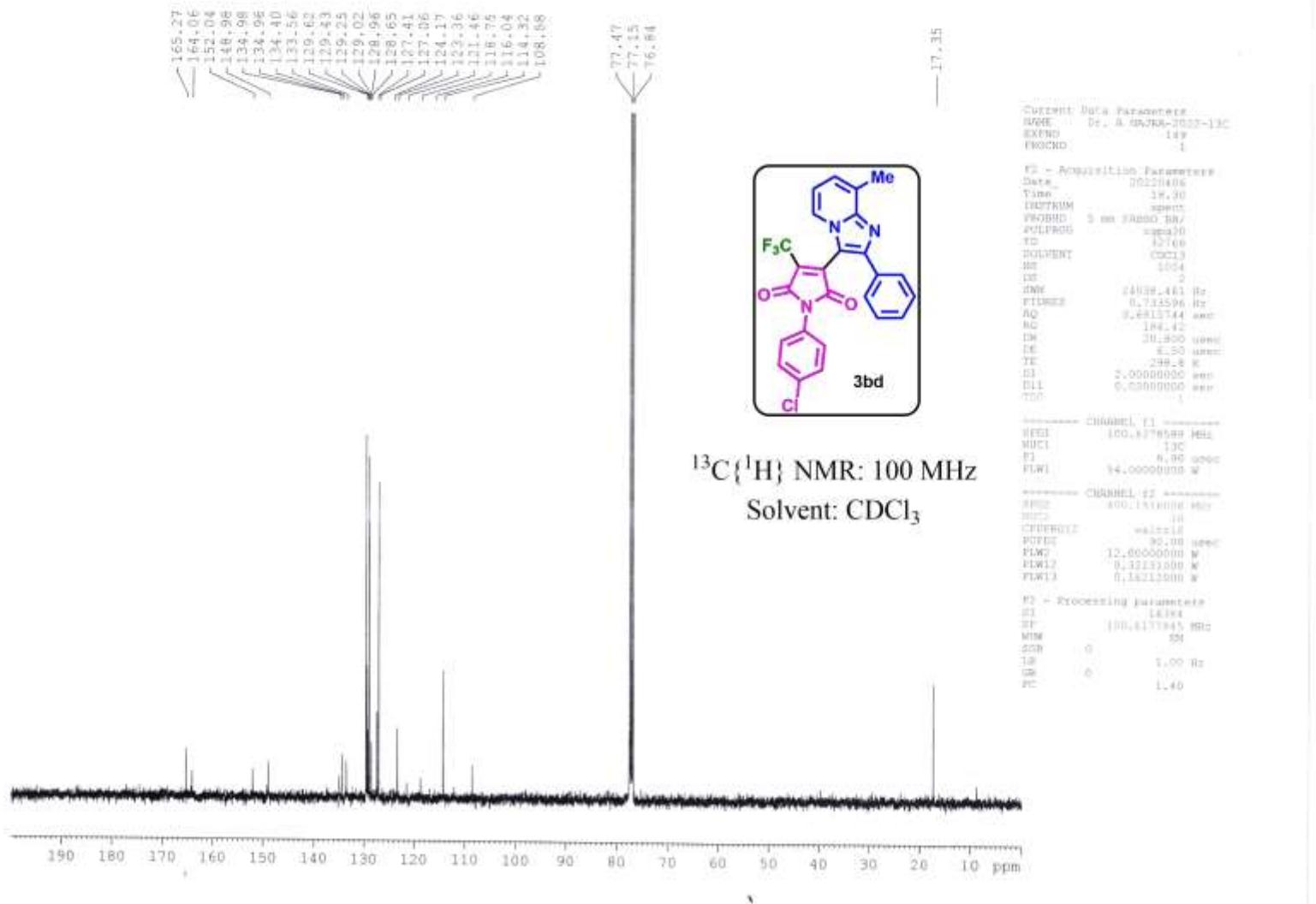




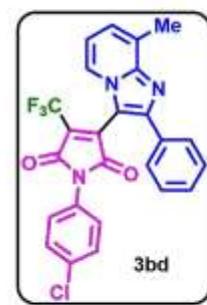








— +61.40



¹⁹F NMR: 376 MHz

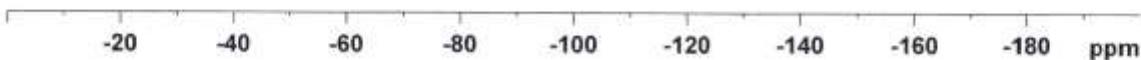
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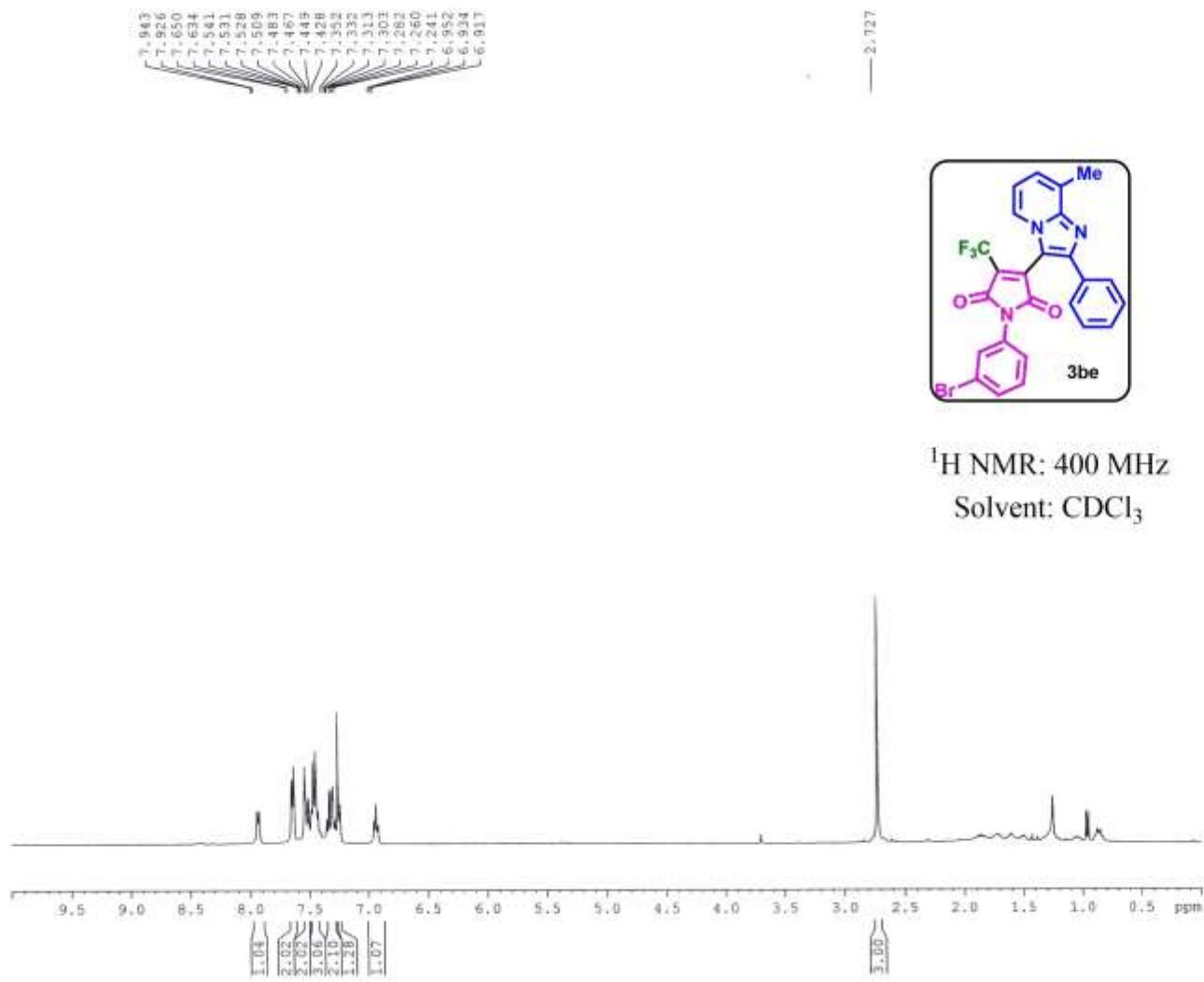
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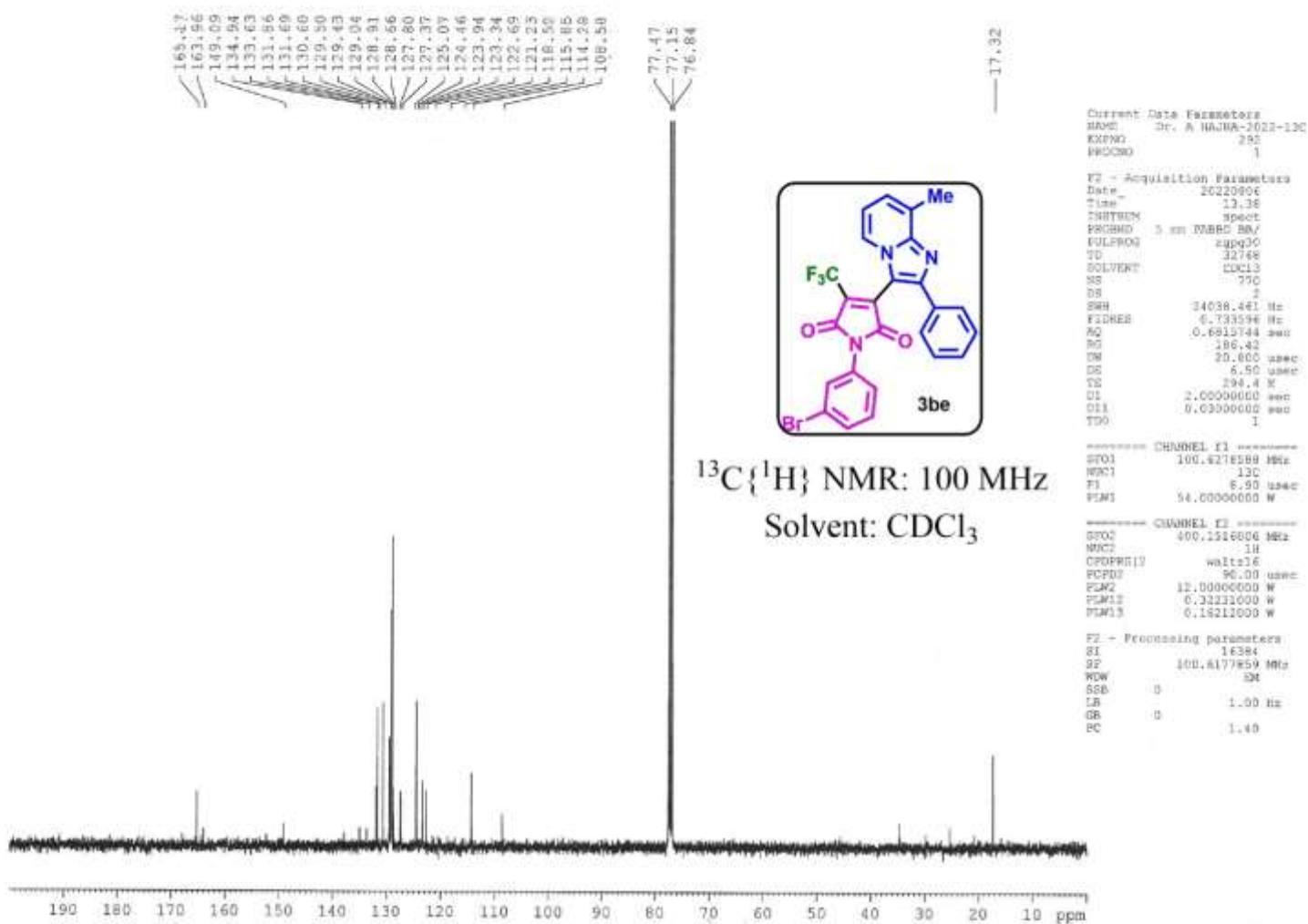
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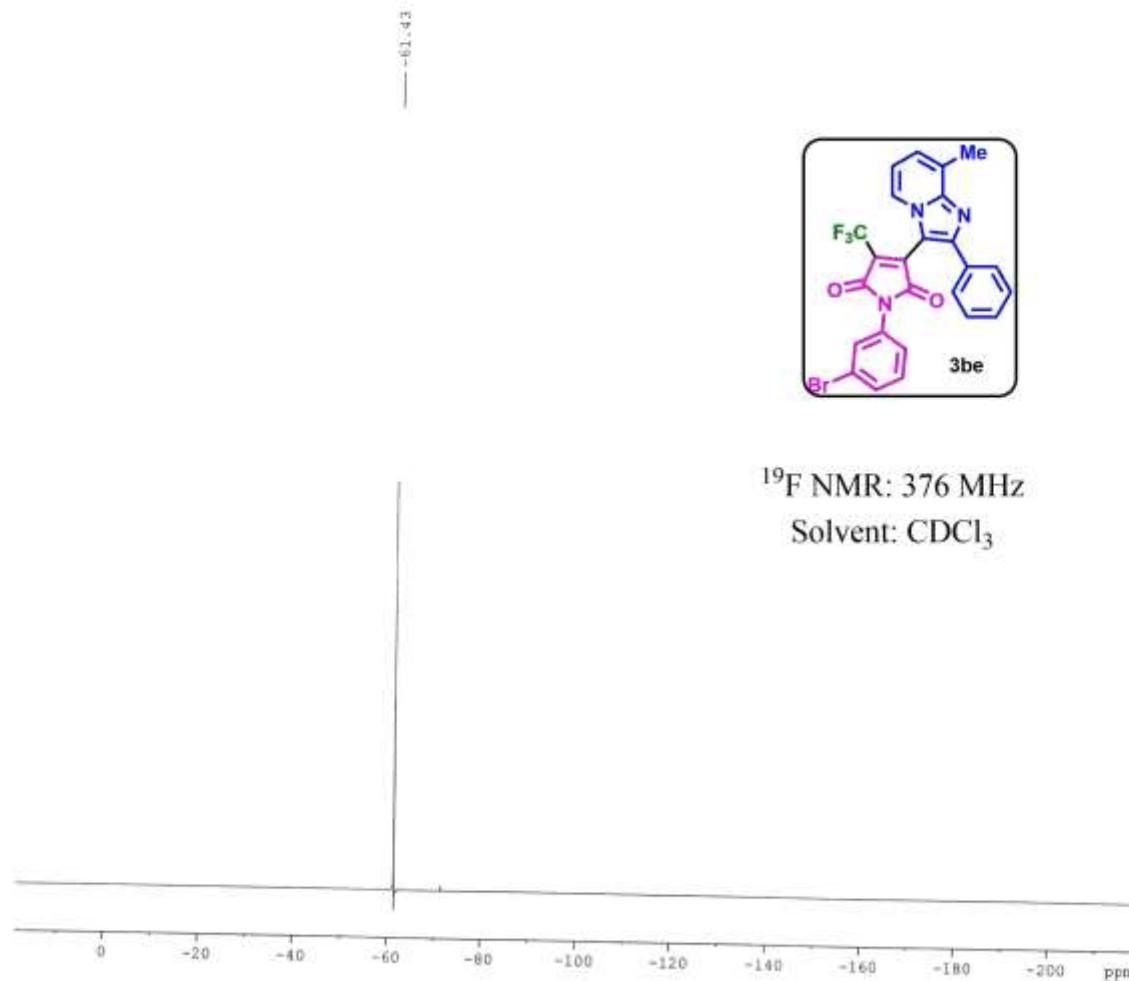
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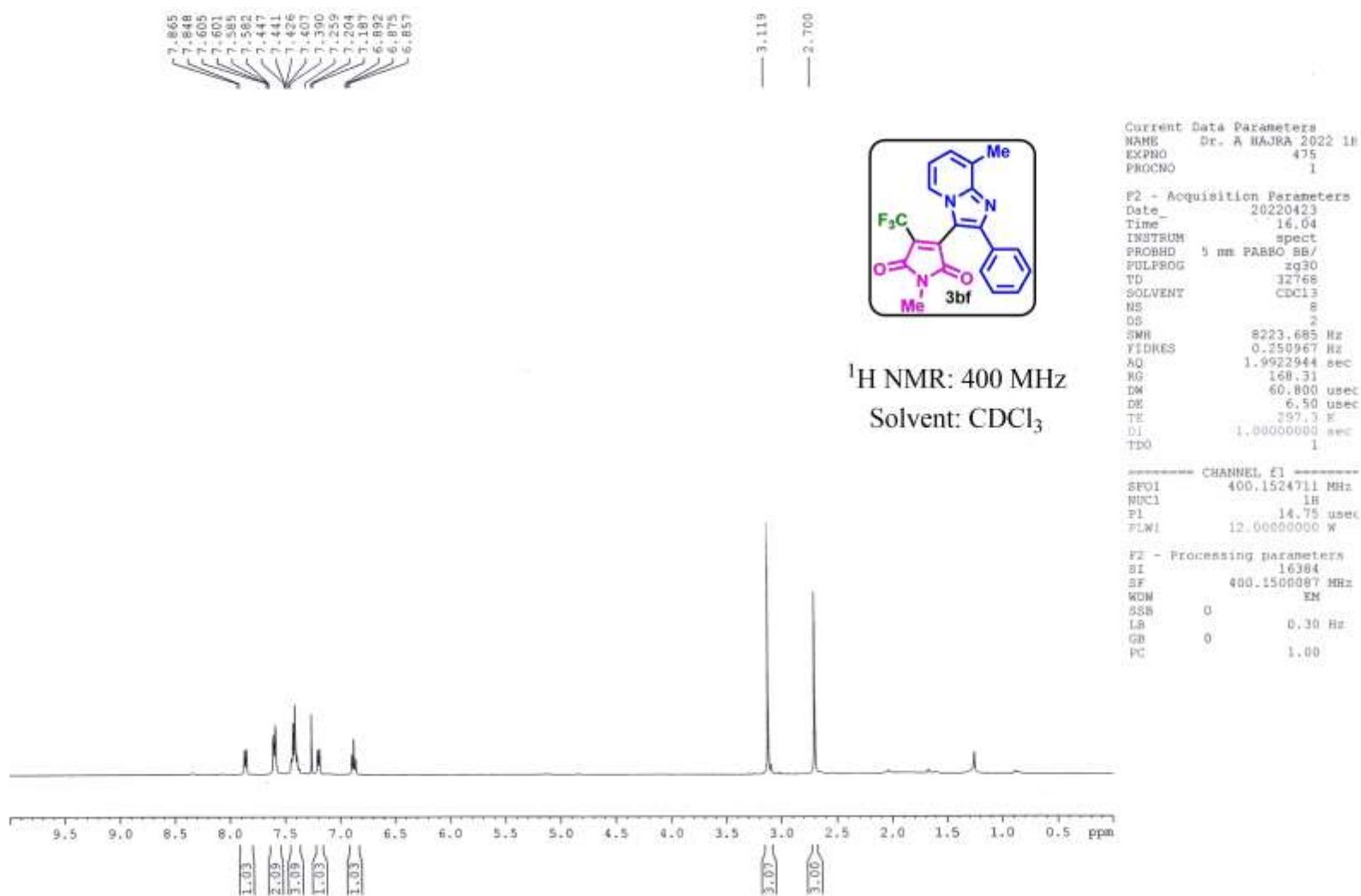


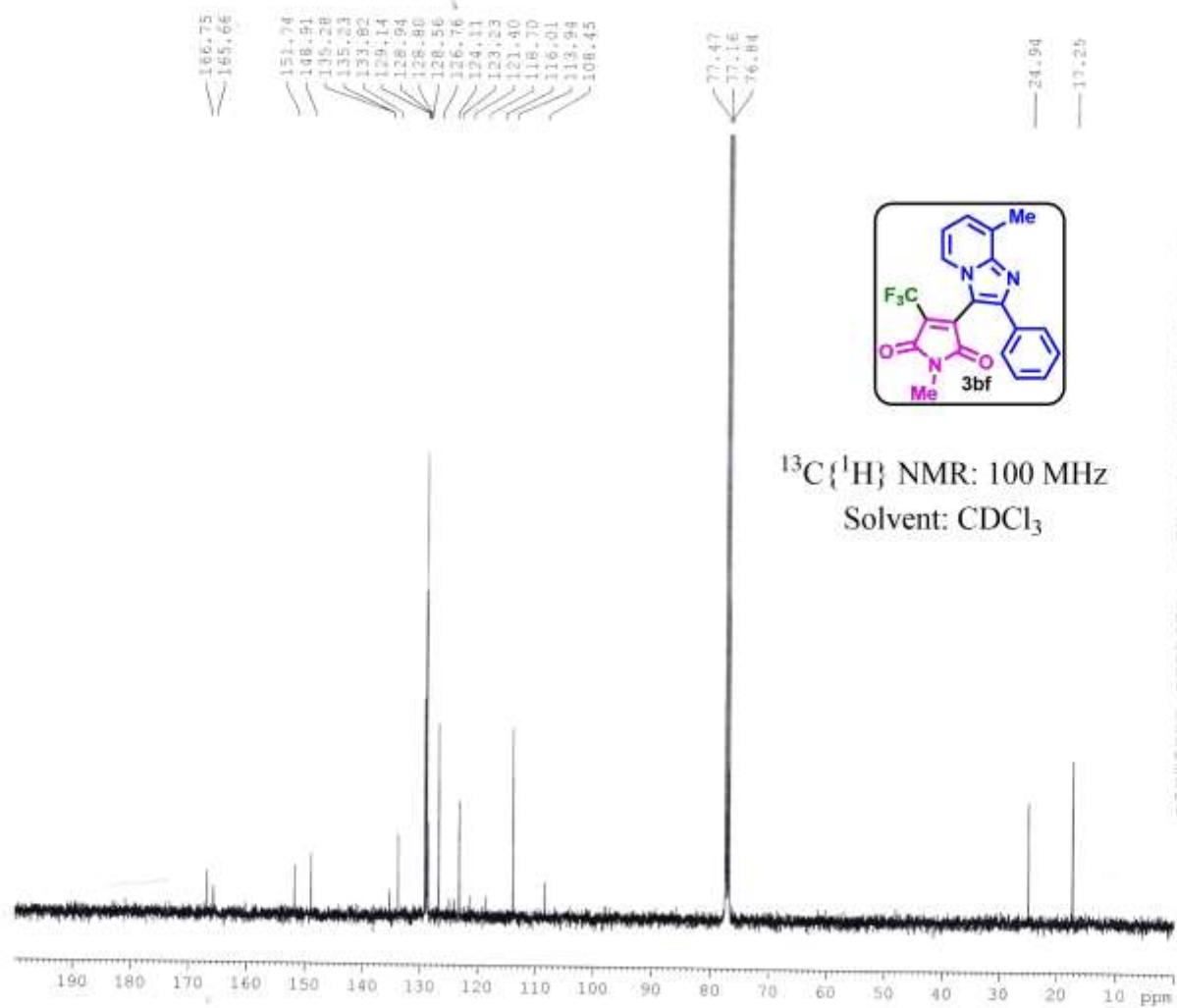
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 TE 293.6 K
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 TDO 1

===== CHANNEL f1 =====
 SFO1 376.4795333 MHz
 NUC1 ¹⁹F
 P1 12.50 usec
 PLW1 20.0000000 W

F2 - Processing parameters
 SI 16384
 SF 376.5171850 MHz
 WDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00





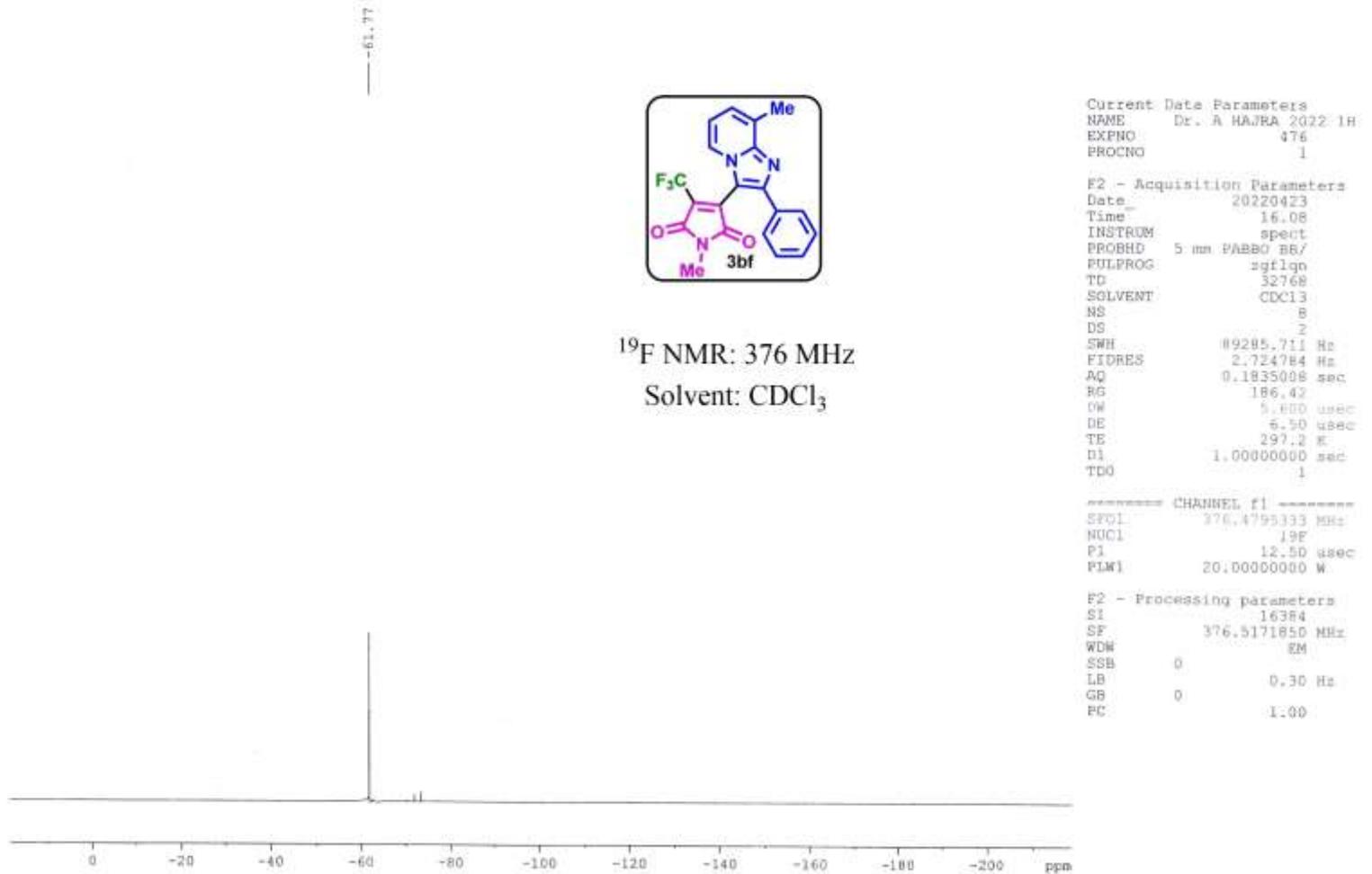
Current Data Parameters
NAME: Dr_A_RHJNv2022-13C
EXPNO: 384
PROCNO: 1
PCPMIN:

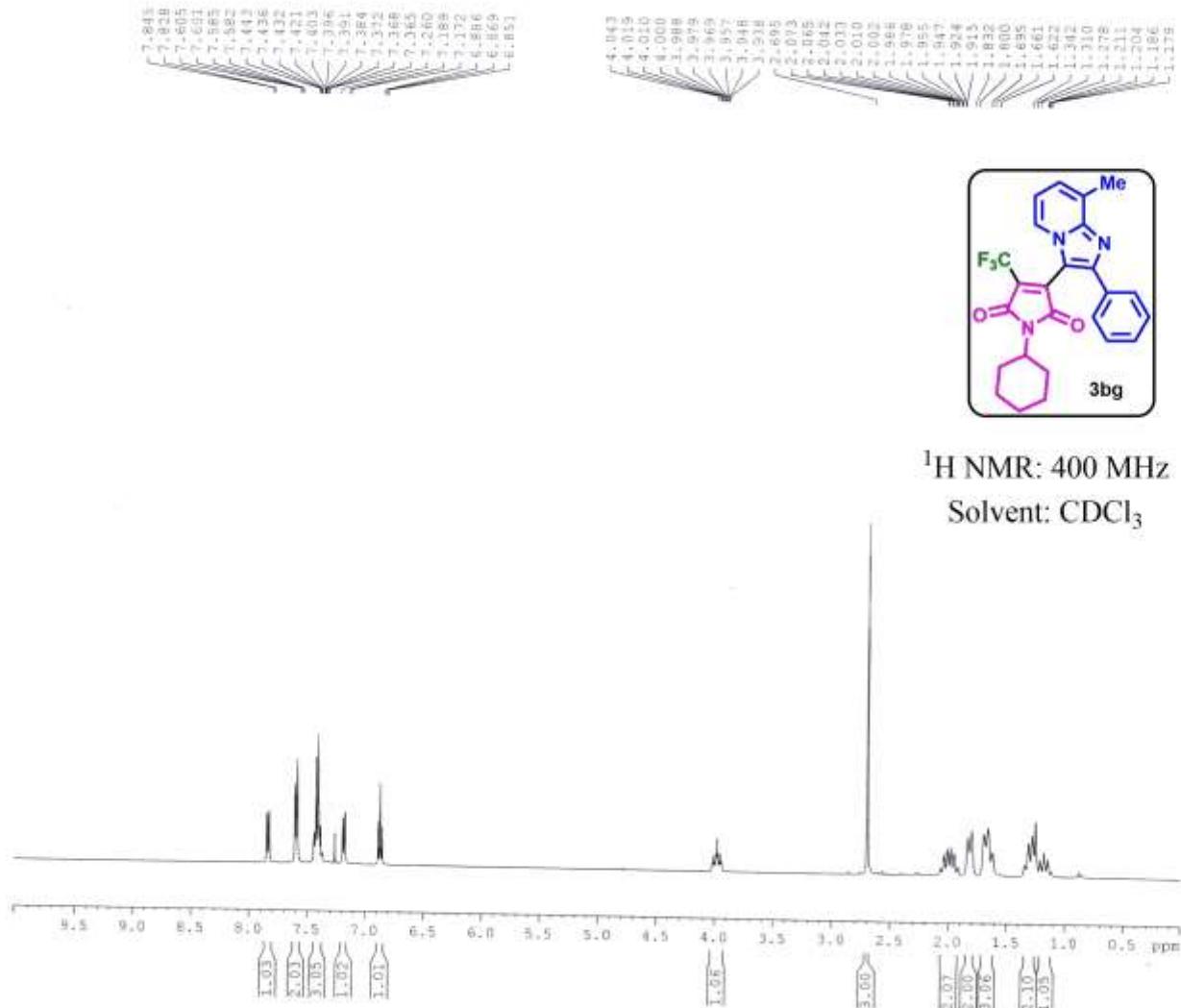
ED = Acquisition Parameters
Date: 20220422
Time: 16:50
D1/D1WM: 2.00000
P1/M1D: 3 ms PR1000000/
PULPROG: zg3g10
TD: 32768
SOLVENT: CDCl_3
NS: 1024
SW: 21050.143 Hz
T1SLW: 0.222096 ms
RG: 0.0015744 sec
D1: 1.00 ms
DW: 70.00 us/sec
DE: 6.50 deg
TE: 277.8 °
D11: 0.000000 sec
T11: 0.000000 sec
T90: 1

===== CHANNEL T1 =====
SF01: 100.6178688 MHz
NUC1: 1H
P1: 9.70 us/sec
PM1: 54.0000000 W

===== CHANNEL F2 =====
SF02: 100.1251604 MHz
NUC2: 1H
CTGPGR12: 90.00 us/sec
P1M2: 13.0000000 W
P1M12: 0.12231000 W
P1M13: 0.16212000 W

ED = Processing parameters
SI: 65536
SF: 100.6177643 MHz
SWB: 16384
LB: 1.00 Hz
GB: 0.00 Hz
PC: 3.40



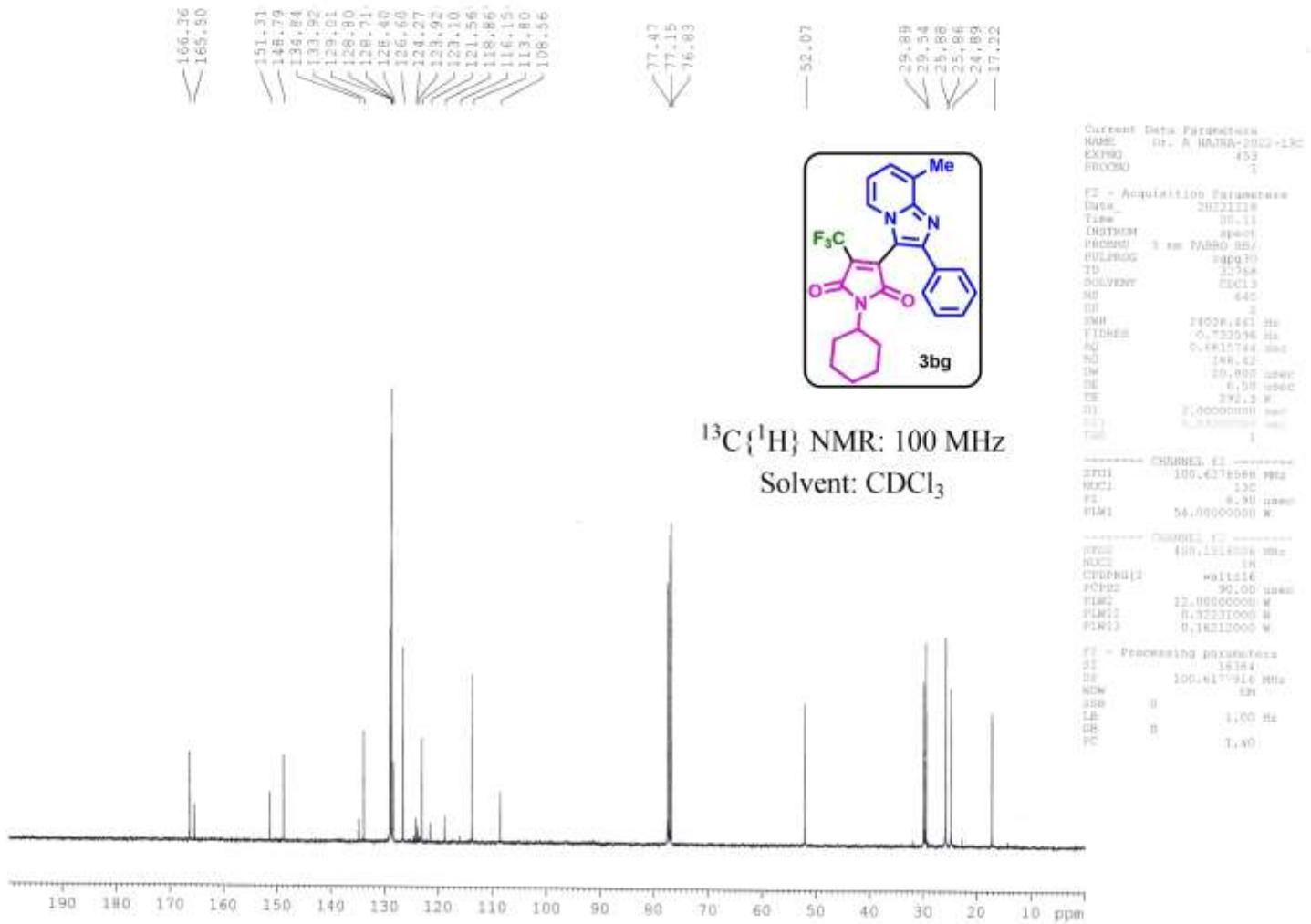


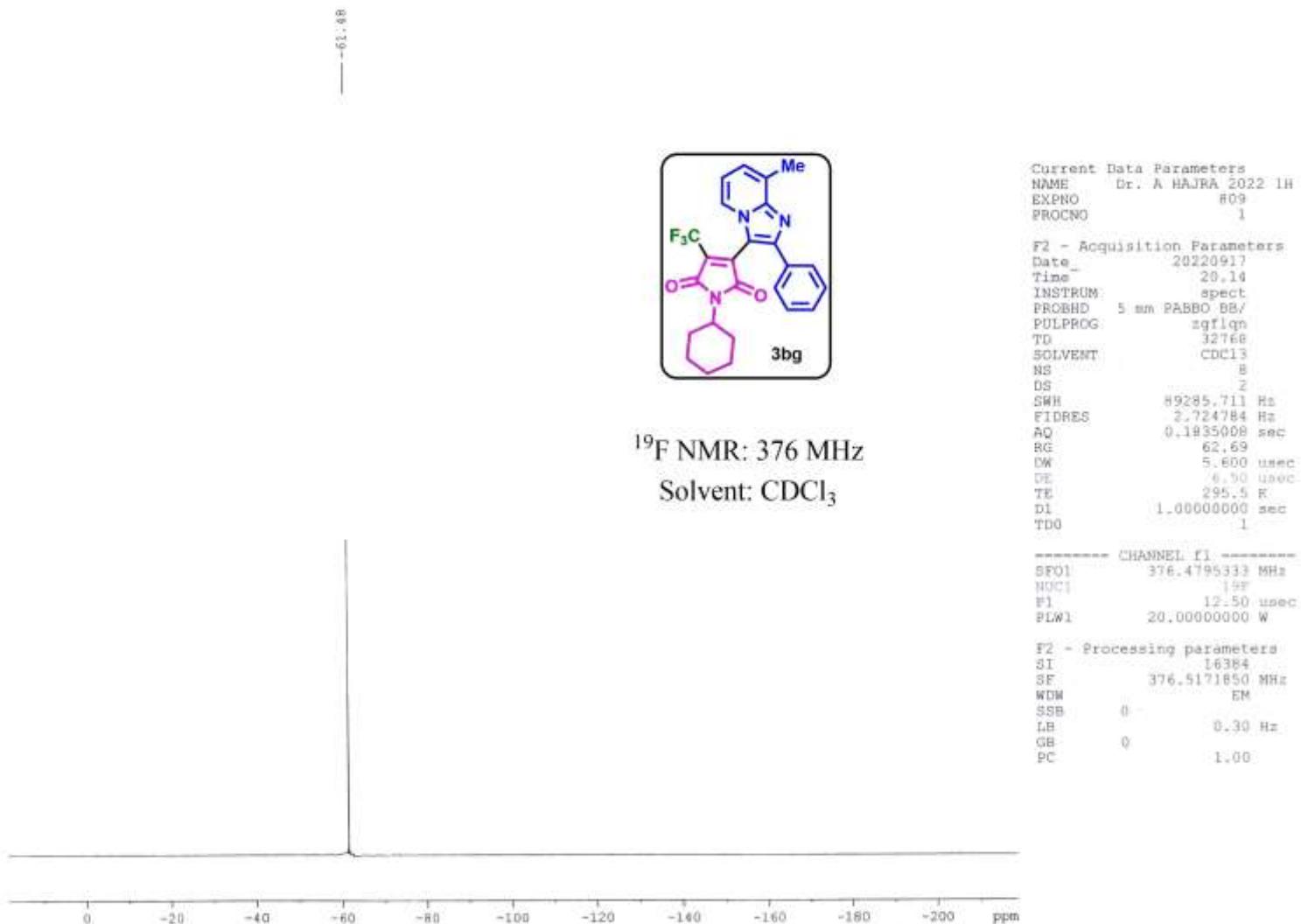
Current Data Parameters
NAME Dr. A HAJRA 2022_1H
EXPNO 1134
PROCNO 1

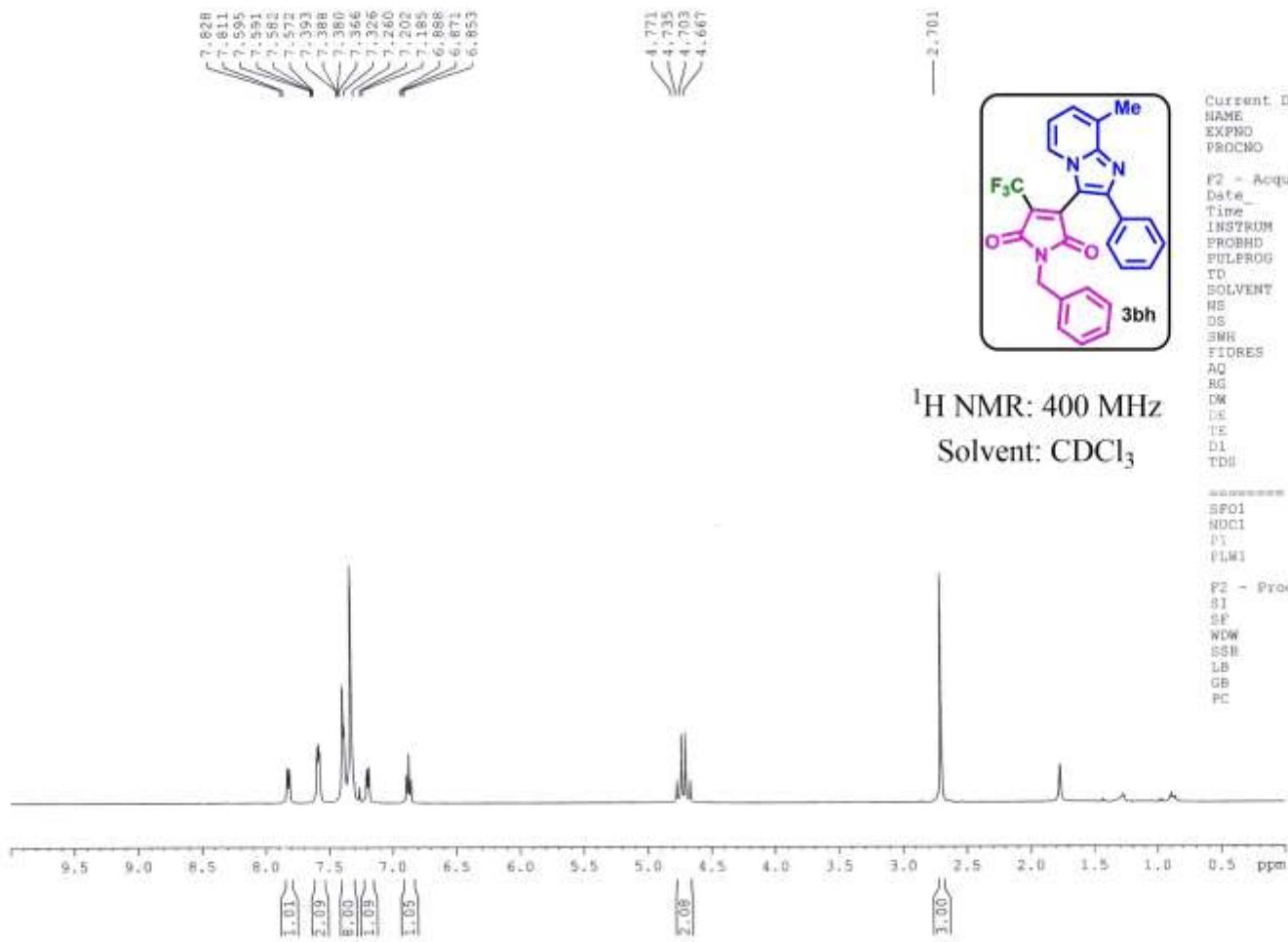
P1 - Acquisition Parameters
Date 20221218
Time 19:36
INSTRUM Spect
PROBHD 5 mm PABBO BB/
PULPROG zg30
TD 32768
SOLVENT CDCl₃
ME 8
DS 2
SWH 8223.665 Hz
FIDRES 0.250967 Hz
AQ 1.992294 sec
RG 47.25
DW 60,000 usec
DE 6.50 usec
TE 291.7 K
D1 1.0000000 sec
TDO :\$

CHANNEL f1
SF01 400.1524711 MHz
NUC1 IH
P1 14.75 usec
PLW1 12,00000000 M

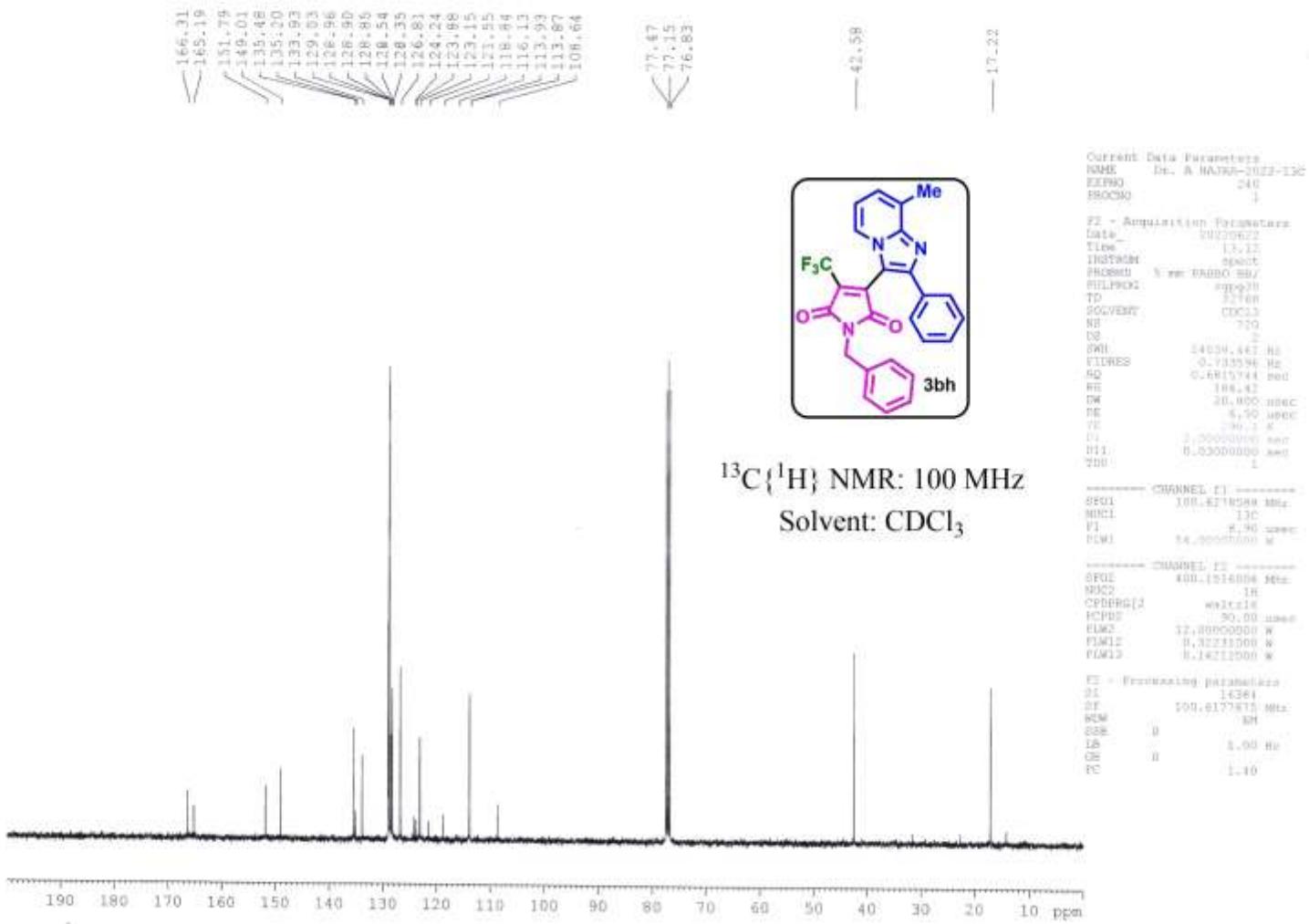
P2 - Processing parameters
SI 16384
SF 400.1500095 MHz
WDW EM
SSB 0
LB 0.30 Hz
GB 0
PC 1.00

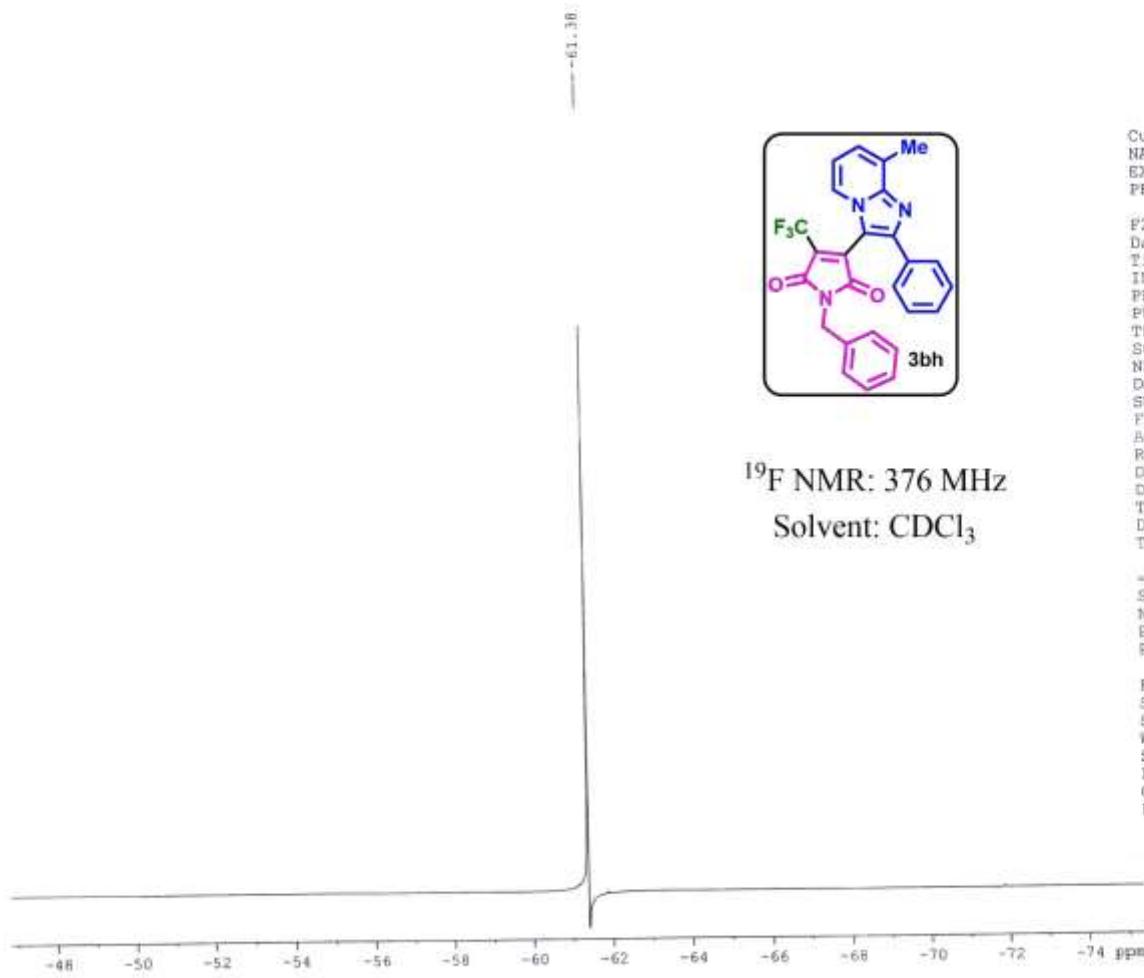






S110





Current Data Parameters
 NAME Dr. A HAJRA 2022 1H
 EXPNO 605
 PROCNO 1

F2 - Acquisition Parameters
 Date 20220622
 Time 12:37
 INSTRUM spect
 PROBHD 5 mm PARBO BB/
 PULPROG zgflq0
 TD 32768
 SOLVENT CDCl₃
 NS 8
 DS 2
 SWH 89285.711 Hz
 FIDRES 1.724784 Hz
 AQ 0.1835008 sec
 RG 186.42
 DW 5.600 usec
 DE 6.50 usec
 TE 297.5 K
 D1 1.00000000 sec
 TDD 1

===== CHANNEL f1 =====
 SP01 376.4795333 MHz
 NUC1 19F
 P1 12.50 usec
 PLW1 20.00000000 W

F2 - Processing parameters
 S1 16384
 SF 376.5171850 MHz
 WDW 8K
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00