

Chlorine-rich lithium argyrodite enables stable interfacial Li plating/stripping behavior in anode- free all-solid-state batteries

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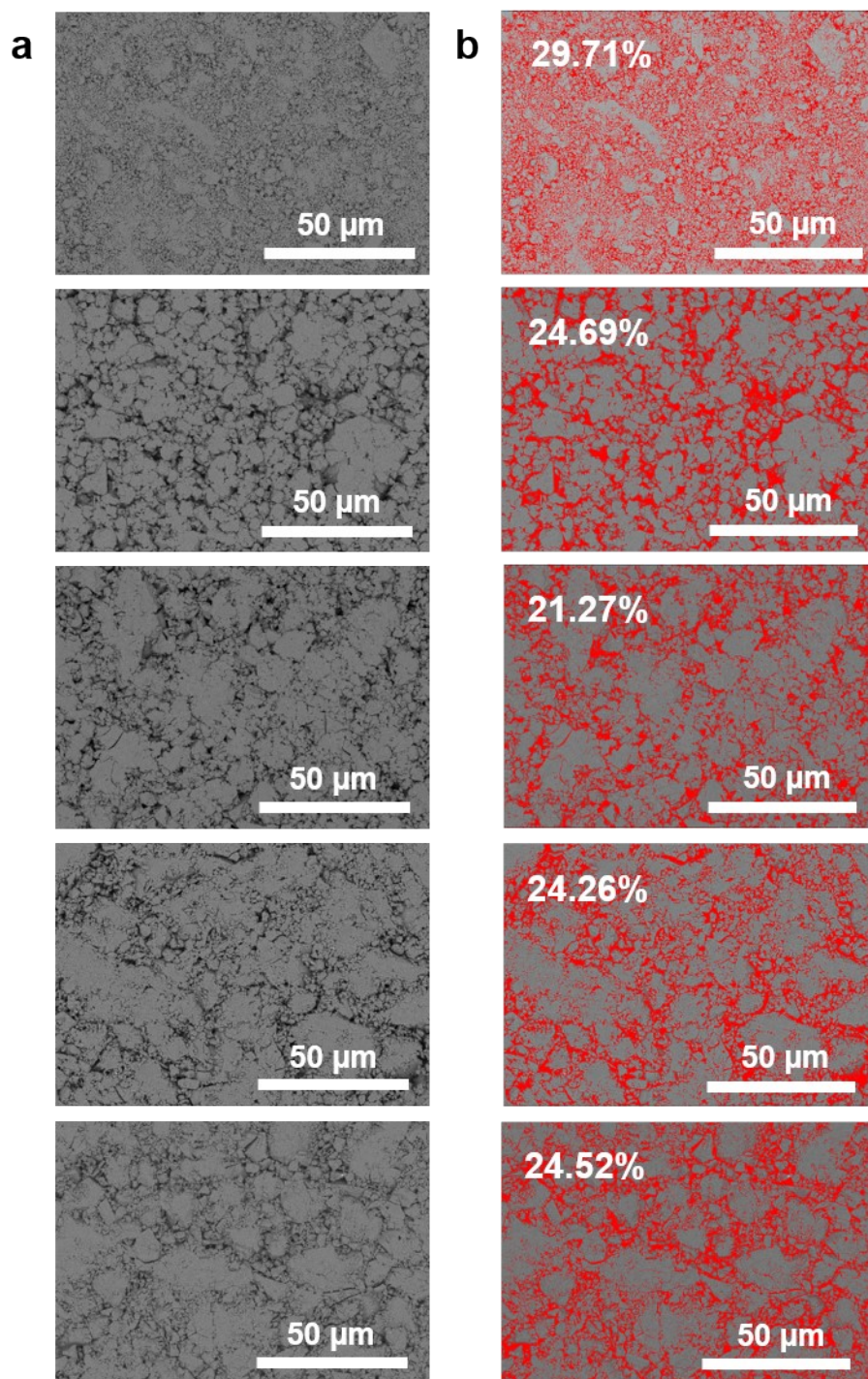


Fig. S1. Porosity analysis of SE pellets. (a) BSE images and (b) the corresponding porosity calculation of the surface for compressed SE pellets with Li_3YCl_6 , Li_3PS_4 , $(\text{Li}_3\text{PS}_4)_{0.7}(\text{LiI})_{0.3}$, $\text{Li}_6\text{PS}_5\text{Cl}$, and $\text{Li}_{5.5}\text{PS}_{4.5}\text{Cl}_{1.5}$ in order from top to bottom.

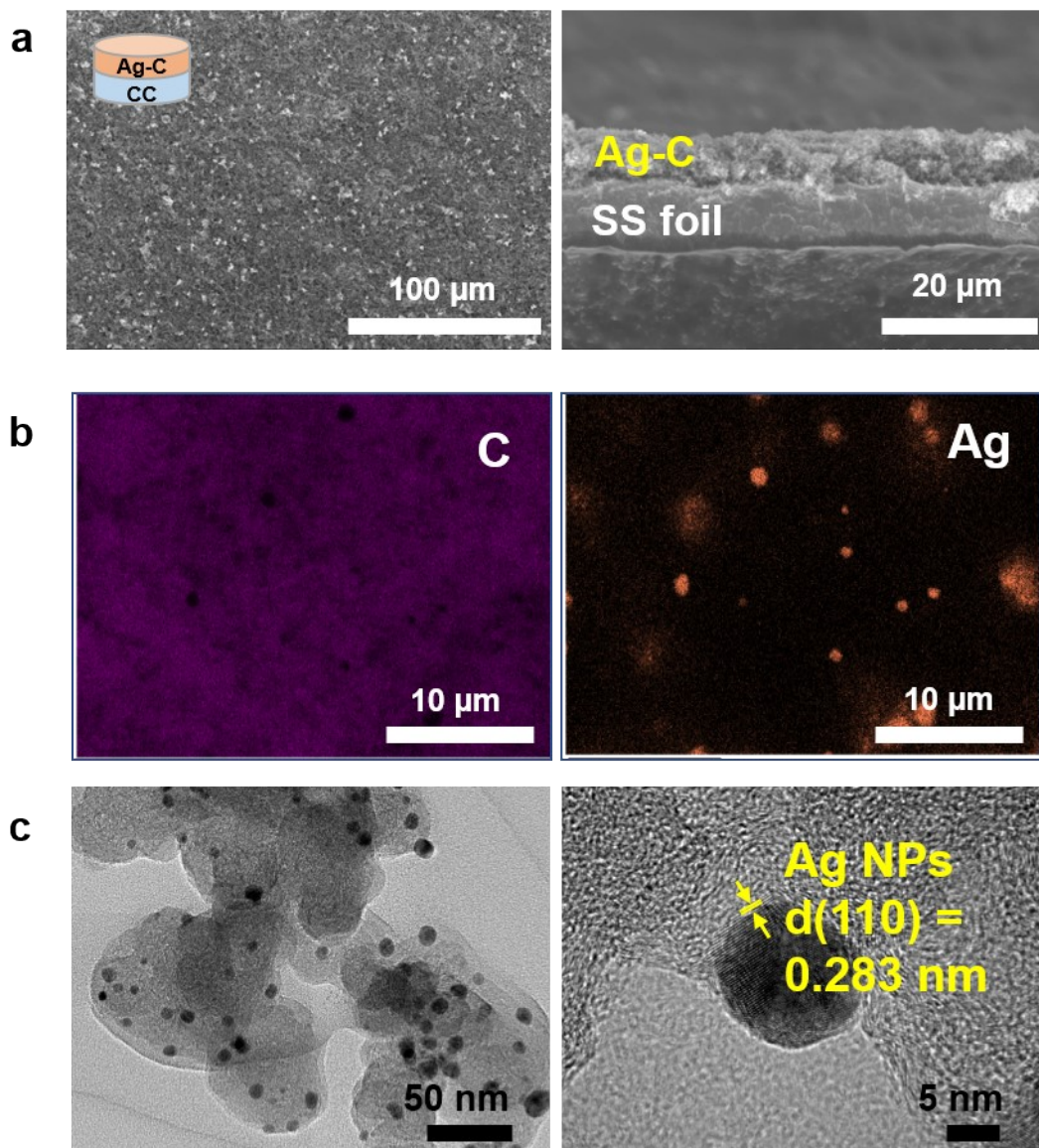


Fig. S2. Morphology and composition characterization of Ag-C layer. (a) Plane and cross-sectional view of SEM images, and (b) EDS mapping of Ag-C layer. (c) TEM image and HRTEM images of Ag-C composites.

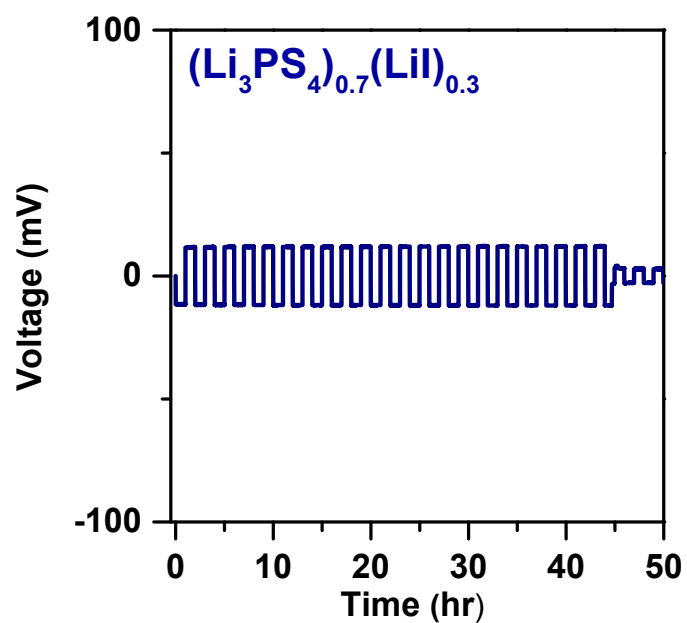


Fig. S3. Electrochemical performance of a symmetric cell $(\text{Li}||(\text{Li}_3\text{PS}_4)_{0.7}(\text{LiI})_{0.3}||\text{Li})$. Voltage-time profiles of the Li plating/stripping process in Li symmetric cell with $(\text{Li}_3\text{PS}_4)_{0.7}(\text{LiI})_{0.3}$ SE at the current density of 0.5 mAcm^{-2} .

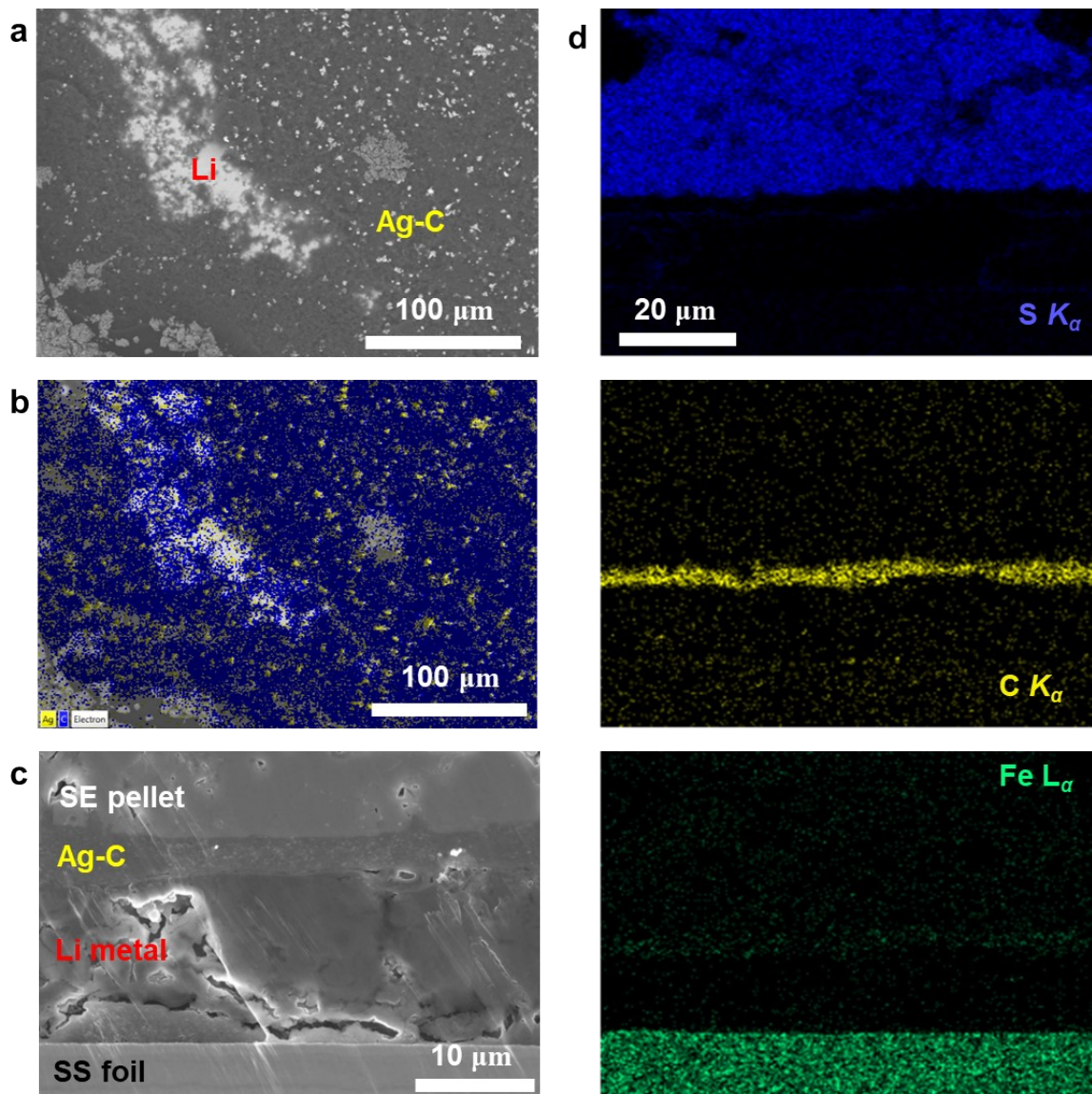


Fig. S4 Planar (a) SEM and (b) EDS images of Li plating with Ag-C layer. Cross-sectional (c) SEM and (d) EDS images of SE/Ag-C/Li/SS interfaces in the NCM||Ag-C full cell using $\text{Li}_{5.5}\text{PS}_{4.5}\text{Cl}_{1.5}$ solid electrolyte after charging at 0.1 C for 10 h.