<Electronic Supplementary Information>

Single crystals of cyclodimeric zinc(II) complexes containing 9,10-bis((isoquinolin-5yloxy)methyl)anthracene: reversible adsorption of targeting molecules and recognition of CH₂I₂ in SCSC mode

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Fig. S1. IR spectra of L (a), $[ZnCl_2L]_2 \cdot 5C_4H_8O$ (b), $[ZnCl_2L]_2 \cdot 4o - X \cdot 2C_4H_8O \cdot 0.5CH_2Cl_2$ (c), $[ZnCl_2L]_2 \cdot 4m - X$ (d), $[ZnCl_2L]_2 \cdot 4p - X$ (e), and $[Zn_2Cl_4L(C_6H_5NH_2)_2] \cdot 2C_6H_5NH_2$ (f).



Fig. S2. TG curves of $[ZnCl_2L]_2 \cdot 5C_4H_8O$ (a), $[ZnCl_2L]_2 \cdot 4o - X \cdot 2C_4H_8O \cdot 0.5CH_2Cl_2$ (b), $[ZnCl_2L]_2 \cdot 4m - X$ (c), $[ZnCl_2L]_2 \cdot 4p - X$ (d), and $[Zn_2Cl_4L(C_6H_5NH_2)_2] \cdot 2C_6H_5NH_2$ (e).



Fig. S3. ¹H NMR spectra (dissociated in Me₂SO-*d*⁶) of L (a), $[ZnCl_2L]_2 \cdot 5C_4H_8O$ (b), $[ZnCl_2L]_2 \cdot 4o - X \cdot 2C_4H_8O \cdot 0.5CH_2Cl_2$ (c), $[ZnCl_2L]_2 \cdot 4m - X$ (d), $[ZnCl_2L]_2 \cdot 4p - X$ (e), and $[Zn_2Cl_4L(C_6H_5NH_2)_2] \cdot 2C_6H_5NH_2$ (f).



Fig. S4. Packing structures of $[ZnCl_2L]_2 \cdot 5C_4H_8O$ (a), $[ZnCl_2L]_2 \cdot 4o - X \cdot 2C_4H_8O \cdot 0.5CH_2Cl_2$ (b), and $[ZnCl_2L]_2 \cdot 4p - X$ (c). C_4H_8O (red), CH_2Cl_2 (green), o - X (blue), and p - X (yellow).



Fig. S5. PXRD patterns of $[ZnCl_2L]_2 \cdot 5C_4H_8O$ (a) and $[Zn_2Cl_4L(C_6H_5NH_2)_2] \cdot 2C_6H_5NH_2$ (b).



Fig. S6. Solid PL spectra at $\lambda_{ex} = 350$ nm of crystals of $[ZnCl_2L]_2 \cdot 5C_4H_8O$ (blue line), the crystals + one drop (20 µL) of the CH₂Cl₂ (pink line), the crystals + one drop of the CH₂Br₂ (orange line), and the crystals + one drop of the CH₂I₂ (sky blue line).