## **Supplementary Information**

## Magnetron sputtering platinum on nitrogen-doped polypyrrole carbon nanotubes as an efficient and stable cathode for lithium-carbon dioxide batteries

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Figure S1 EDX mapping of C and N for the selected area of NPPy-CNTs.

Table S1 Chemical composition from EDS analysis carried out on NPPy-CNTs.

Element	Weight %	Atomic %
С	93.8	94.6
Ν	6.2	5.4



Figure S2 The SEM images of Pt-NPPy-CNTs (a) and commercial CNTs (b).



Figure S3 (a)  $N_2$  adsorption and desorption curve of NPPy-CNTs; (b) pore size distribution diagram of NPPy-CNTs.



Figure S4 TEM and EDX images of Pt-NPPy-CNTs.



Figure S5 The charge-discharge cycle curves of Li-CO<sub>2</sub> batteries with different catalysts at a current density of 100mA g<sup>-1</sup> and a cut-off capacity of 1000mAh g<sup>-1</sup>: (a) Commercial CNTs; (b) NPPy-CNTs.



Figure S6 Charge and discharge curves of Li-CO<sub>2</sub> batteries with different catalysts at different current densities and cut-off capacities of 1000mAh g<sup>-1</sup>: (a) Commercial CNTs; (b) NPPy-CNTs.



Figure S7 Commercial CNTs cathodes in different charge and discharge stages: (a) Ex-situ XRD pattern; (b) ex-situ EIS map.



Figure S8 Comparison of discharge specific capacity and cycle performance of different noble metal doped carbon materials as cathode catalyst of Li-CO<sub>2</sub> batteries.<sup>1-7</sup>



Figure S9 (a) First complete charge-discharge diagram of Pt-NPPy-CNTs with different magnetron sputtering time at 100 mAg<sup>-1</sup> current density; charge-discharge cycles of Pt-NPPy-CNTs with different magnetron sputtering time at 100 mA g<sup>-1</sup> current density and 1000 mAh g<sup>-1</sup> cut-off capacity: (b) 60s; (c) 120s; (d) 180s.

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