Supporting information

Highlights on the reversible nonpolar-to-polar P3₁21-P3₁ phase transition at low temperature in NaLa(SO₄)₂.H₂O: mechanism and piezoelectric properties

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Chemical formula	$H_{0.67}La_{0.33}Na_{0.33}O_3S_{0.67}$			
Formula weight	124.01 g/mol			
Temperature	100(2) K			
Wavelength		0.71073	Å	
Crystal system		Trigonal		
Space group	P31			
	a = 7.022	7(5) Å	$\alpha = 90^{\circ}$	
Unit cell dimensions	b = 7.022	7(5) Å	$\beta = 90^{\circ}$	
	c = 12.945	0(13) Å	$\gamma = 120^{\circ}$	
Volume		552.89(10)	Å ³	
Z		3		
Density (calculated)	1.117 g/cm ³			
Absorption coefficient	2.148 mm ⁻¹			
F(000)	174			
Theta range for data collection	3.15 to 52.27°			
Index ranges	-15<=h<=15, -15<=k<=15, -28<=l<=28			
Reflections collected	27080			
Independent reflections	8289 [R(int) = 0.0342]			
	020)	$[\mathbf{K}(\mathbf{m})] = 0$	0.0342]	
Refinement method	Full-mat	rix least-sq	uares on F^2	
Refinement method Refinement program	Full-mat	rix least-sq 2019/1 (She	uares on F ²	
Refinement method Refinement program Function minimized	Full-mate	$\frac{\text{rix least-sq}}{2019/1 \text{ (She}}$	uares on F^2 eldrick, 2019)	
Refinement methodRefinement programFunction minimizedData / restraints / parameters	Full-mati	rix least-sq 2019/1 (She $\Sigma w(F_o^2 - F_o^2)$ 3289 / 1 / 1	uares on F^2 eldrick, 2019) $(2^2)^2$	
Refinement method Refinement program Function minimized Data / restraints / parameters Goodness-of-fit on F ²	Full-matrix	$\frac{\text{rix least-sq}}{\text{c019/1 (She}}$ $\frac{\text{E w(F_o^2 - F_o^2)}}{\text{S289 / 1 / 1}}$ 1.041	1000000000000000000000000000000000000	
Refinement method Refinement program Function minimized Data / restraints / parameters Goodness-of-fit on F ²	Full-mati	$\frac{r(x(m)) - c}{r(x + m) - c}$ $\frac{r(x + m) - c}{r(x + m) - c}$ $\frac{r(x + m) - c}{r(x + m) - c}$ $\frac{r(x + m) - c}{r(x + m) - c}$ $\frac{r(x + m) - c}{r(x + m) - c}$ $\frac{r(x + m) - c}{r(x + m) - c}$ $\frac{r(x + m) - c}{r(x + m) - c}$ $\frac{r(x + m) - c}{r(x + m) - c}$ $\frac{r(x + m) - c}{r(x + m) - c}$ $\frac{r(x + m) - c}{r(x + m) - c}$ $\frac{r(x + m) - c}{r(x + m) - c}$	(1000000000000000000000000000000000000	
Refinement method Refinement program Function minimized Data / restraints / parameters Goodness-of-fit on F ² Final R indices	Full-mati SHELXL-2 2 8289 data; $I>2\sigma(I)$	$\frac{[\text{R(III)} = 0]}{\text{rix least-sq}}$ $\frac{2019/1 \text{ (She}}{2 \text{ w}(\text{F}_{0}^{2} - \text{F}_{0}^{2})}$ $\frac{3289 / 1 / 1}{1.041}$ $\frac{\text{R1}}{\text{wR2}}$	1000000000000000000000000000000000000	

Table S1: Crystal data collection and structure refinement for NaLa(SO₄)₂.H₂O at 100K

Weighting scheme	w=1/[$\sigma^{2}(F_{o}^{2})$] where P=($F_{o}^{2}+2F_{c}^{2}$)/3	
Absolute structure parameter	0.005(4)	
Extinction coefficient	0.0220(10)	
Largest diff. peak and hole	1.774 and -0.961 eÅ ⁻³	

Table S2: Atomic coordinates and equivalent isotropic atomic displacement parameters (Å²) for NaLa(SO₄)₂.H₂O at 100K

	x/a	y/b	z/c	U(eq)
La01	0.57677(2)	0.01318(2)	0.49361(2)	0.00218(10)
Na02	0.4649(2)	0.47075(16)	0.65869(8)	0.00657(12)
S003	0.01571(10)	0.57106(6)	0.57995(3)	0.00283(6)
S004	0.45530(10)	0.44530(10)	0.40693(3)	0.00282(7)
O001	0.8805(2)	0.6349(2)	0.64441(10)	0.00602(15)
O002	0.1800(2)	0.7643(2)	0.52274(10)	0.00682(17)
O003	0.8786(2)	0.4022(2)	0.50378(10)	0.00670(17)
O004	0.6349(2)	0.4846(3)	0.33468(10)	0.00673(18)
O005	0.5201(2)	0.6327(2)	0.47642(10)	0.00683(17)
O006	0.1200(3)	0.4889(2)	0.65298(10)	0.00572(15)
O007	0.2671(2)	0.4084(2)	0.34166(9)	0.00582(15)
O008	0.3956(2)	0.2495(2)	0.47172(10)	0.00674(16)
O009	0.9344(3)	0.0196(3)	0.52408(14)	0.0124(2)

Table S3: Bond lengths (Å) for NaLa(SO₄)₂.H₂O at 100K(With the Symmetry transformations : #1= x, y+1, z. #2= x+1, y, z. #3= x+1, y+1, z. #4= y+1, x-y, z+1/3.#5= -y+1, x-y+1, z+1/3 #6=-x+y, -x+1, z-1/3 . #7= -x+y+1, -x+2, z-1/3)

13 = y + 1, x y + 1	r, z+1/3 #0=	X + y, X + 1, Z = 1/	$J \cdot \Pi I = X + y + 1$
La01-O002	2.4677(14)	La01-O003#3	2.4867(15)
La01-O005	2.5072(14)	La01-O009	2.5200(16)
La01-O008#1	2.5620(14)	La01-O004#5	2.5738(14)
La01-O006#6	2.5908(13)	La01-O001#6	2.6013(13)
La01-O007#5	2.6104(13)	La01-S004#5	3.1962(7)
La01-S003#6	3.2026(7)	La01-Na02#7	4.0263(13)
Na02-O007#5	2.4154(17)	Na02-O004#4	2.4499(17)
Na02-O003#5	2.4838(17)	Na02-O006	2.4895(19)
Na02-O001#2	2.5524(17)	Na02-O005	2.5631(17)
Na02-O008	2.7845(18)	Na02-O002#5	2.9406(18)
Na02-S004	3.2628(11)	Na02-S003#5	3.3034(11)
S003-O002	1.4681(14)	S003-O003	1.4709(14)
S003-O006	1.4785(14)	S003-O001	1.4918(14)
S004-O005	1.4658(14)	S004-O007	1.4785(14)
S004-O004	1.4811(15)	S004-O008	1.4812(14)

Table S4: Bond angles (°) for NaLa(SO₄)₂.H₂O at 100K (With the Symmetry transformations: #1= x-1, y-1, z . #2= x-1, y, z . #3= x, y-1, z . #4= x, y+1, z. #5= x+1, y, z . #6= x+1, y+1, z . #7= -y+1, x-y, z+1/3 . #8= -y+1, x-y+1, z+1/3. #9= -y+2, x-y+1, z+1/3. #10=-x+y, -x+1, z-1/3. #11= -x+y+1, -x+1, z-1/3. #12= -x+y+1, -x+2, z-1/3)

, 2+1/0+1/10 A+J, A+1, 2 1/2	$j = \frac{1}{2}$	1, x + 1, z + 3, x + y + 1, z + 1,	,
O002-La01-O003#6	144.14(5)	O002-La01-O005	74.77(5)
O003#6-La01-O005	140.32(5)	O002-La01-O009	138.66(5)
O003#6-La01-O009	71.19(6)	O005-La01-O009	70.04(5)
O002-La01-O008#4	74.06(4)	O003#6-La01-O008#4	73.78(5)
O005-La01-O008#4	144.59(5)	O009-La01-O008#4	144.96(5)
O002-La01-O004#8	77.89(5)	O003#6-La01-O004#8	75.68(5)
O005-La01-O004#8	120.64(4)	O009-La01-O004#8	101.99(5)
O008#4-La01-O004#8	67.94(4)	O002-La01-O006#10	69.16(4)
O003#6-La01-O006#10	117.36(4)	O005-La01-O006#10	77.14(5)
O009-La01-O006#10	121.27(5)	O008#4-La01-O006#10	76.41(5)
O004#8-La01-O006#10	136.72(4)	O002-La01-O001#10	123.49(4)
O003#6-La01-O001#10	73.12(4)	O005-La01-O001#10	91.32(4)
O009-La01-O001#10	78.83(5)	O008#4-La01-O001#10	92.05(4)
O004#8-La01-O001#10	146.65(5)	O006#10-La01-O001#10	54.33(4)
O002-La01-O007#8	80.43(4)	O003#6-La01-O007#8	102.70(4)
O005-La01-O007#8	69.81(4)	O009-La01-O007#8	67.67(5)
O008#4-La01-O007#8	120.31(4)	O004#8-La01-O007#8	54.23(4)
O006#10-La01-O007#8	139.83(5)	O001#10-La01-O007#8	145.41(4)
O002-La01-S004#8	78.98(3)	O003#6-La01-S004#8	88.08(3)
O005-La01-S004#8	95.74(3)	O009-La01-S004#8	83.58(4)
O008#4-La01-S004#8	94.42(3)	O004#8-La01-S004#8	27.10(3)
O006#10-La01-S004#8	148.12(3)	O001#10-La01-S004#8	157.53(3)
O007#8-La01-S004#8	27.18(3)	O002-La01-S003#10	96.14(3)
O003#6-La01-S003#10	96.30(3)	O005-La01-S003#10	82.32(3)
O009-La01-S003#10	99.92(4)	O008#4-La01-S003#10	84.71(3)
O004#8-La01-S003#10	152.63(3)	O006#10-La01-S003#10	27.02(3)
O001#10-La01-S003#10	27.36(3)	O007#8-La01-S003#10	151.89(3)
S004#8-La01-S003#10	175.087(9)	O002-La01-Na02#12	142.51(4)
O003#6-La01-Na02#12	35.87(3)	O005-La01-Na02#12	124.74(4)
O009-La01-Na02#12	77.46(4)	O008#4-La01-Na02#12	74.84(3)
O004#8-La01-Na02#12	108.84(4)	O006#10-La01-Na02#12	83.60(4)
O001#10-La01-Na02#12	38.19(3)	O007#8-La01-Na02#12	134.24(4)
S004#8-La01-Na02#12	123.941(18)	S003#10-La01-Na02#12	60.51(2)
O007#8-Na02-O004#7	152.98(7)	O007#8-Na02-O003#8	88.70(6)

O004#7-Na02-O003#8	75.12(6)	O007#8-Na02-O006	76.35(6)
O004#7-Na02-O006	129.78(7)	O003#8-Na02-O006	113.21(6)
O007#8-Na02-O001#5	78.68(5)	O004#7-Na02-O001#5	76.15(6)
O003#8-Na02-O001#5	74.02(6)	O006-Na02-O001#5	153.72(7)
O007#8-Na02-O005	72.02(5)	O004#7-Na02-O005	113.18(6)
O003#8-Na02-O005	151.25(7)	O006-Na02-O005	83.40(6)
O001#5-Na02-O005	81.25(5)	O007#8-Na02-O008	124.64(6)
O004#7-Na02-O008	66.13(5)	O003#8-Na02-O008	140.95(7)
O006-Na02-O008	95.72(6)	O001#5-Na02-O008	91.75(6)
O005-Na02-O008	52.62(5)	O007#8-Na02-O002#8	86.61(5)
O004#7-Na02-O002#8	99.08(5)	O003#8-Na02-O002#8	51.00(5)
O006-Na02-O002#8	63.24(5)	O001#5-Na02-O002#8	123.33(6)
O005-Na02-O002#8	144.05(7)	O008-Na02-O002#8	138.64(5)
O007#8-Na02-S004	97.75(5)	O004#7-Na02-S004	90.42(5)
O003#8-Na02-S004	157.98(6)	O006-Na02-S004	88.80(4)
O001#5-Na02-S004	86.57(4)	O005-Na02-S004	25.74(3)
O008-Na02-S004	26.89(3)	O002#8-Na02-S004	149.96(5)
O007#8-Na02-S003#8	87.15(4)	O004#7-Na02-S003#8	86.68(4)
O003#8-Na02-S003#8	24.62(3)	O006-Na02-S003#8	89.03(5)
O001#5-Na02-S003#8	97.81(5)	O005-Na02-S003#8	158.97(5)
O008-Na02-S003#8	148.11(5)	O002#8-Na02-S003#8	26.39(3)
S004-Na02-S003#8	174.02(4)	O007#8-Na02-La01#9	88.74(5)
O004#7-Na02-La01#9	65.42(4)	O003#8-Na02-La01#9	35.92(4)
O006-Na02-La01#9	146.92(5)	O001#5-Na02-La01#9	39.06(3)
O005-Na02-La01#9	120.18(5)	O008-Na02-La01#9	116.85(5)
O002#8-Na02-La01#9	86.87(4)	S004-Na02-La01#9	122.78(4)
S003#8-Na02-La01#9	60.50(2)	O007#8-Na02-La01	37.22(3)
O004#7-Na02-La01	146.43(5)	O003#8-Na02-La01	125.16(5)
O006-Na02-La01	71.04(4)	O001#5-Na02-La01	84.12(4)
O005-Na02-La01	35.83(3)	O008-Na02-La01	87.95(4)
O002#8-Na02-La01	114.47(4)	S004-Na02-La01	61.17(2)
S003#8-Na02-La01	123.17(3)	La01#9-Na02-La01	113.38(3)
O002-S003-O003	107.58(9)	O002-S003-O006	111.67(9)
O003-S003-O006	111.50(9)	O002-S003-O001	109.19(8)
O003-S003-O001	111.02(9)	O006-S003-O001	105.89(8)
O002-S003-La01#8	122.20(6)	O003-S003-La01#8	130.20(6)
O006-S003-La01#8	52.77(6)	O001-S003-La01#8	53.26(5)
O002-S003-Na02#10	62.89(6)	O003-S003-Na02#10	44.71(6)
O006-S003-Na02#10	127.13(6)	O001-S003-Na02#10	126.08(6)

La01#8-S003-Na02#10	174.90(3)	O005-S004-O007	110.93(8)
O005-S004-O004	111.54(10)	O007-S004-O004	105.97(9)
O005-S004-O008	107.62(9)	O007-S004-O008	110.34(9)
O004-S004-O008	110.47(9)	O005-S004-La01#10	130.04(6)
O007-S004-La01#10	53.76(5)	O004-S004-La01#10	52.34(6)
O008-S004-La01#10	122.35(6)	O005-S004-Na02	49.40(6)
O007-S004-Na02	124.94(6)	O004-S004-Na02	128.92(6)
O008-S004-Na02	58.25(6)	La01#10-S004-Na02	178.59(3)
S003-O001-Na02#2	131.35(8)	S003-O001-La01#8	99.38(7)
Na02#2-O001-La01#8	102.75(5)	S003-O002-La01	144.99(9)
S003-O002-Na02#10	90.73(7)	La01-O002-Na02#10	105.57(5)
S003-O003-Na02#10	110.67(8)	S003-O003-La01#1	140.89(8)
Na02#10-O003-La01#1	108.20(5)	S004-O004-Na02#11	143.40(9)
S004-O004-La01#10	100.56(7)	Na02#11-O004-La01#10	115.39(6)
S004-O005-La01	145.47(8)	S004-O005-Na02	104.86(8)
La01-O005-Na02	107.42(5)	S003-O006-Na02	136.35(9)
S003-O006-La01#8	100.21(7)	Na02-O006-La01#8	116.33(6)
S004-O007-Na02#10	139.69(9)	S004-O007-La01#10	99.06(7)
Na02#10-O007-La01#10	108.75(5)	S004-O008-La01#3	131.48(8)
S004-O008-Na02	94.86(7)	La01#3-O008-Na02	105.10(5)

Table S5: Anisotropic atomic displacement parameters (Å²) for NaLa(SO₄)₂.H₂O at 100K

	U11	U22	U33	U23	U13	U12
La01	0.00257(3)	0.00165(3)	0.00199(2)	0.00000(2)	0.00016(2)	0.00080(2)
Na02	0.0080(4)	0.0068(2)	0.00599(18)	-0.0003(3)	-0.0009(2)	0.0045(3)
S003	0.00262(14)	0.00273(15)	0.00280(14)	0.00015(8)	0.00016(13)	0.00107(11)
S004	0.00369(13)	0.00293(15)	0.00236(12)	0.00001(11)	-0.00010(12)	0.00205(10)
O001	0.0060(4)	0.0084(4)	0.0055(3)	0.0001(3)	0.0012(3)	0.0051(3)
O002	0.0054(4)	0.0052(4)	0.0079(4)	0.0025(3)	0.0022(3)	0.0011(3)
O003	0.0066(4)	0.0059(4)	0.0051(3)	-0.0017(3)	-0.0025(3)	0.0013(3)
O004	0.0050(4)	0.0112(5)	0.0046(3)	0.0002(3)	0.0011(3)	0.0045(4)
O005	0.0108(5)	0.0057(4)	0.0057(3)	-0.0031(3)	-0.0027(3)	0.0053(4)
O006	0.0066(4)	0.0077(4)	0.0046(3)	0.0006(3)	-0.0002(3)	0.0049(4)
O007	0.0043(4)	0.0085(4)	0.0053(3)	-0.0005(3)	-0.0017(3)	0.0037(3)
O008	0.0080(4)	0.0048(4)	0.0078(3)	0.0030(3)	0.0003(3)	0.0034(3)
O009	0.0069(4)	0.0096(6)	0.0214(6)	0.0057(5)	0.0018(4)	0.0047(4)

Figure S1: Experimental mid-infrared spectra between 1800 and 3000 cm⁻¹ as a function of temperature

