

## Supplementary Material

### Stability, electronic, and optical properties of two-dimensional iridium trihalides with promising applications in photocatalytic water splitting

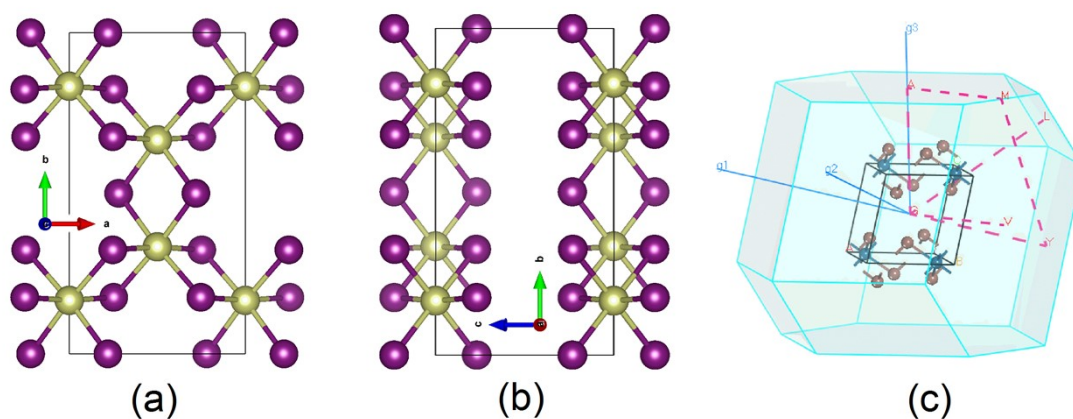
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#### Contents:

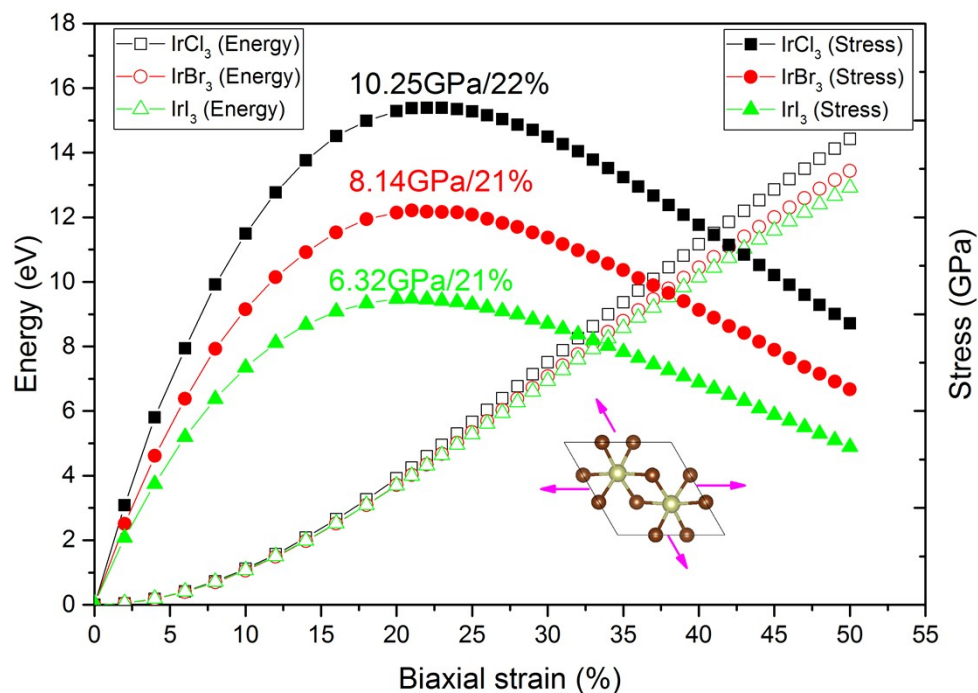
#### Figure S1:



**Figure S1** (a) Top and (b) side views of the crystal structure of the bulk phase of  $\alpha$ - $\text{IrX}_3$  ( $X = \text{Cl}, \text{Br}, \text{I}$ ) in the representation of the conventional cell. (c) High symmetry  $k$ -points of the Brillouin zone for  $\alpha$ - $\text{IrX}_3$  and the path along which the band structure is evaluated.

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**Figure S2:**

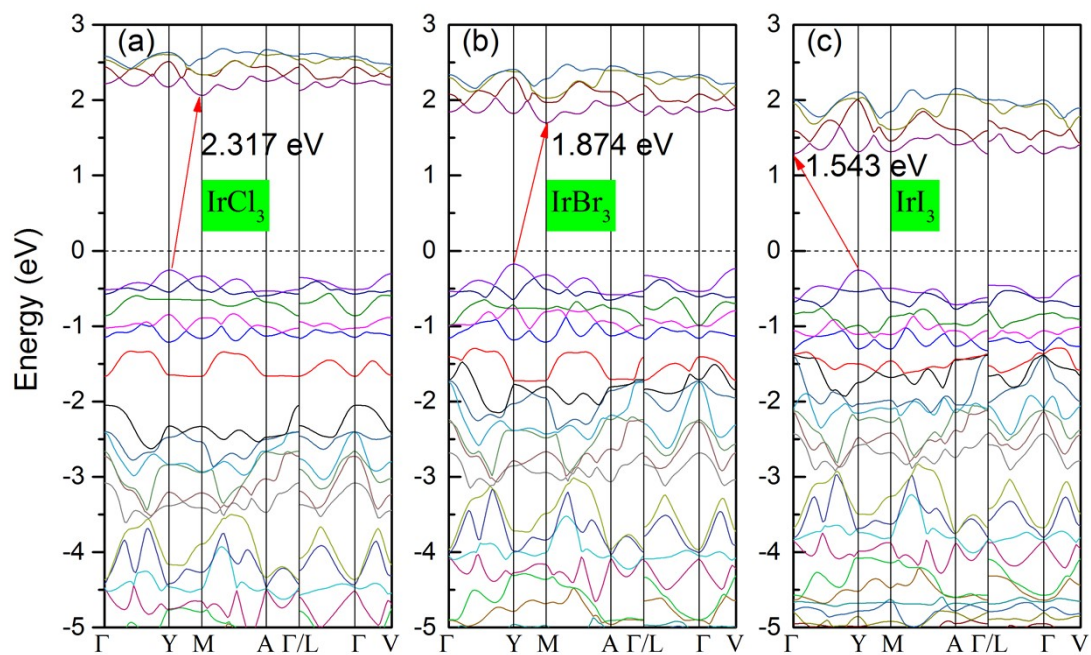


**Figure S2.** The change of strain energy and the created stress in the  $\text{IrX}_3$  monolayers under applied biaxial strain. Because a vacuum space is left along the  $z$  direction in the unit cell, the calculated in-plane stress components must be rescaled<sup>[1,2]</sup> based on the effective layer thickness  $d_0$  of  $\text{IrX}_3$  monolayers. We chose the lattice constant  $c$  of bulk  $\text{IrX}_3$  as the  $d_0$ , as list in Table S1, and rescale the in-plane stress components by  $h/d_0$ , where  $h$  is the length of the cell along  $z$  axis.

Table S1. Calculated lattice constant (in Å) at PBE level, and band-gap  $E_g$  (in eV) of  $\alpha\text{-IrX}_3$  bulk at different levels. The asterisk (\*) means the experimental results.

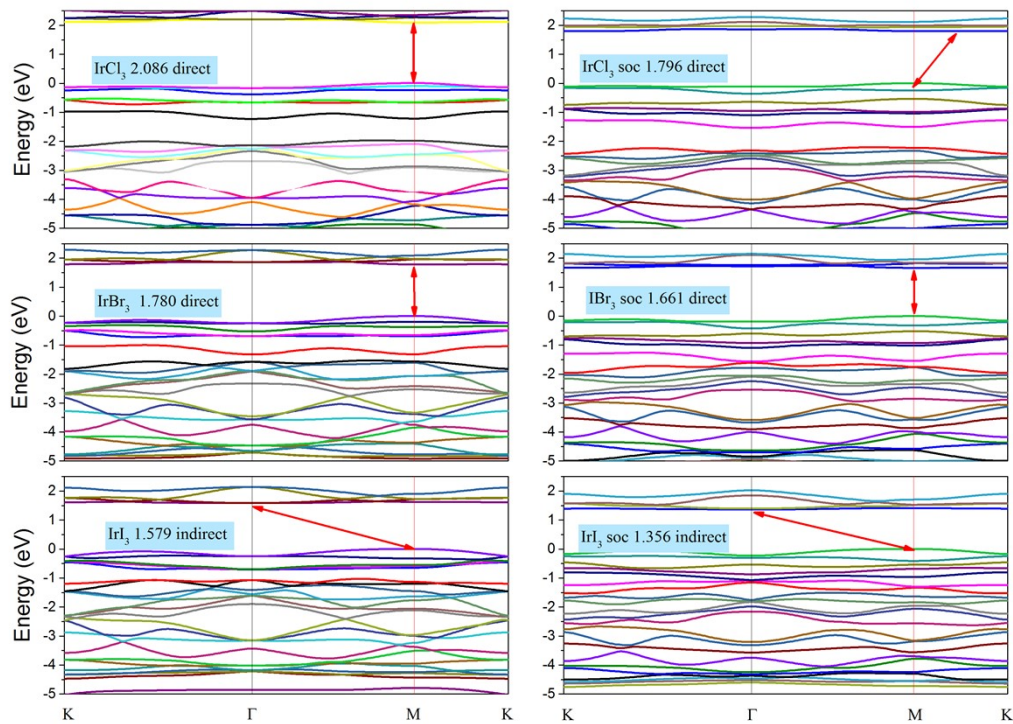
Structure	$a$	$b$	$c$	$E_g$ with different methods			
				PBE	PBE-SOC	HSE06	experiment
$\text{IrCl}_3$	6.092	10.558	5.629	1.846	1.750	3.264	
$\text{IrCl}_3^{[3]}$	$5.99 \pm 0.01$	$10.37 \pm 0.02$	$5.99 \pm 0.01$				
$\text{IrBr}_3$	6.408	11.116	5.984	1.432	1.380	2.704	
$\text{IrBr}_3^{* [4]}$	6.30	10.98	6.34				
$\text{IrI}_3$	6.867	11.901	6.490	1.175	1.123	2.217	
$\text{IrI}_3^{[5]}$	6.802	11.791	6.864	1.0			1.49
$\text{IrI}_3^{*[11]}$	6.74	11.75	6.80				

**Figure S3:**



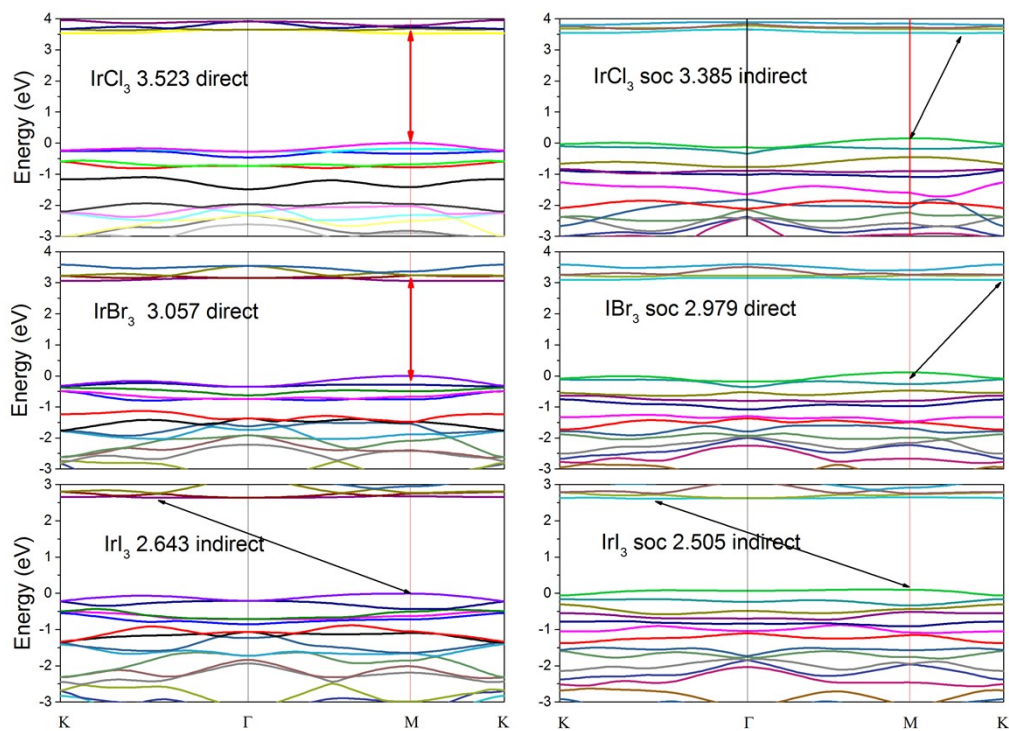
**Figure S3.** Electronic structure of bulk  $\text{IrX}_3$  obtained from HSE functional with 10% of the exact exchange.

**Figure S4:**



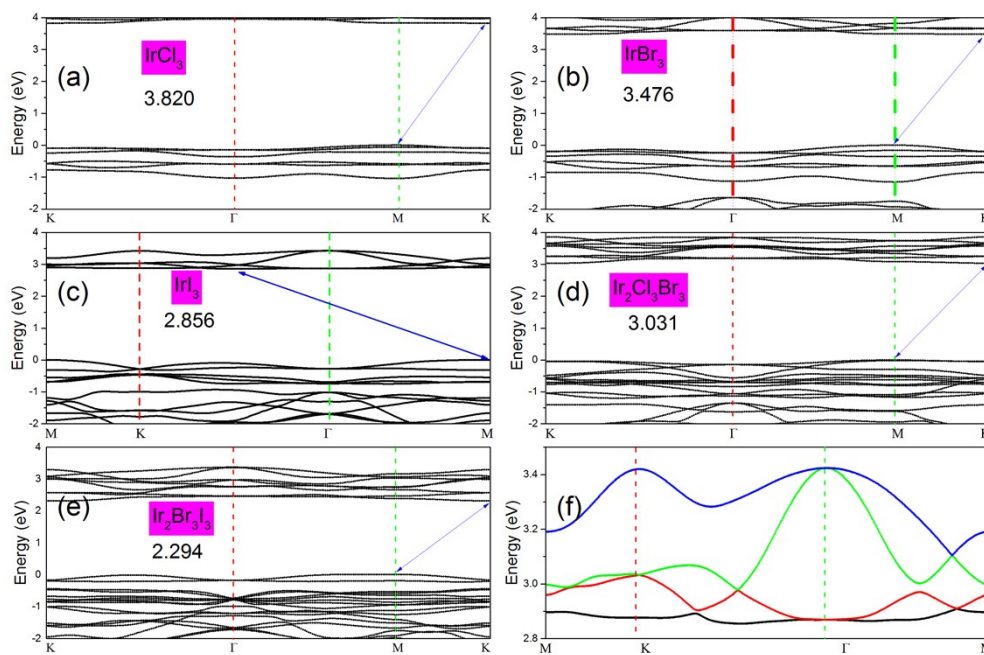
**Figure S4.** Electronic structure of monolayer IrX<sub>3</sub> obtained from (left panel) PBE and (right panel) PBE + SOC methods.

**Figure S5:**



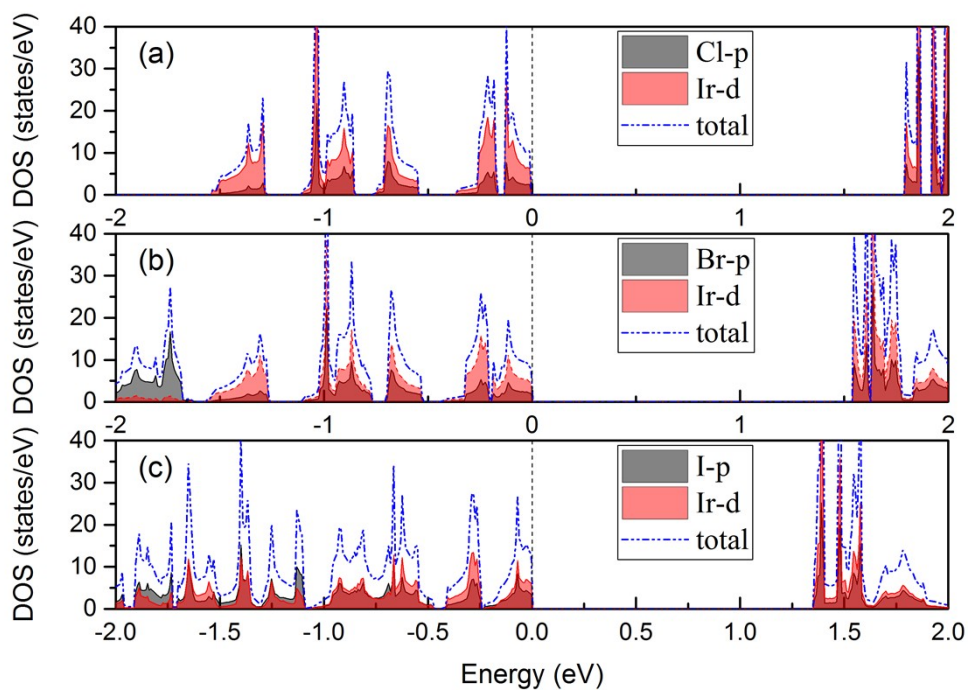
**Figure S5.** Electronic structure of monolayer  $\text{IrX}_3$  obtained from (left panel) HSE06 and (right panel) HSE06 + SOC methods.

**Figure S6.**



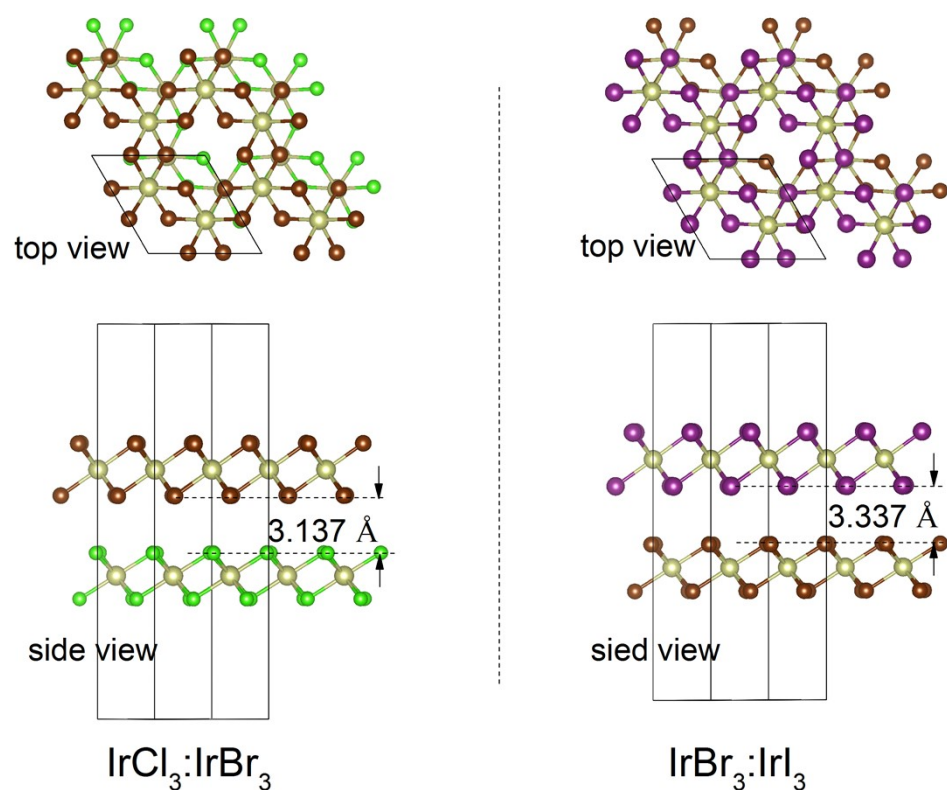
**Figure S6.** Electronic structure of monolayers (a) IrCl<sub>3</sub>, (b) IrBr<sub>3</sub>, (c) IrI<sub>3</sub>, and their heterojunctions (d) IrCl<sub>3</sub>:IrBr<sub>3</sub> and (e) IrBr<sub>3</sub>:IrI<sub>3</sub> obtained from G<sub>0</sub>W<sub>0</sub> method. (f) The band structures of IrI<sub>3</sub> near the valence band maximum.

**Figure S7.**



**Figure S7.** Total and projected DOS of monolayers (a) IrCl<sub>3</sub>, (b) IrBr<sub>3</sub>, (c) IrI<sub>3</sub> obtained at PBE+SOC level.

**Figure S8.**



**Figure S8.** Top and lateral views of IrCl<sub>3</sub>:IrBr<sub>3</sub> and IrBr<sub>3</sub>:Irl<sub>3</sub> van der Waals heterostructures.

## References

- [1] F. Liu, P. Ming and J. Li, Ab initio calculation of ideal strength and phonon instability of graphene under tension, *Phys. Rev. B*, 2007, 76(6), 064120.
- [2] T. Li, Ideal strength and phonon instability in single-layer MoS<sub>2</sub>, *Phys. Rev. B*, 2012, 85(23), 235407.
- [3] K. Brodersen, F. Moers and H. Schnering, Zur Struktur des Iridium (III)-und des Ruthenium (III)-chlorids, *Sci. Nat.*, 1965, 52(9), 205-206.
- [4] K. Brodersen, Structure of  $\beta$ -RuCl<sub>3</sub>, RuI<sub>3</sub>, IrBr<sub>3</sub>, and IrI<sub>3</sub>, *Angew. Chem. Int. Ed. in English*, 1968, 7(2), 148-148.
- [5] D. Ni, K. P. Devlin, G. Cheng, X. Gui, W. Xie, N. Yao and R. J. Cava, The honeycomb and hyperhoneycomb polymorphs of IrI<sub>3</sub>, *J. Solid State Chem.*, 2022, 312, 123240.



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- <sup>1</sup> F. Liu, P. Ming and J. Li, Ab initio calculation of ideal strength and phonon instability of graphene under tension, *Phys. Rev. B*, 2007, 76(6), 064120.
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- <sup>4</sup> K. Brodersen, Structure of  $\beta$ -RuCl<sub>3</sub>, RuI<sub>3</sub>, IrBr<sub>3</sub>, and IrI<sub>3</sub>, *Angew. Chem. Int. Ed. in English*, 1968, 7(2), 148-148.
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