Supporting information for

Infrared Photodissociation Spectroscopy of Mass-Selected $[TaO_3(CO_2)_n]^+$ (n = 2–5) Complexes in the Gas Phase

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of theory (bond lengths in Ångströms and bond angles in degrees).

Table S1. Binding energies of successive CO₂ ligands calculated for the singlet and triplet complex isomers at the PBE0-D3(BJ)/def2-TZVP level of theory (in kcal/mol).



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Figure S4. Optimized geometries of the $[TaO_3(CO_2)_5]^+$ complex isomers in the singlet and triplet states calculated at the PBE0-D3(BJ)/def2-TZVP level of theory (bond lengths in Ångströms and bond angles in degrees).



Figure S5. Relative energies (in kcal/mol) of the $[TaO_3(CO_2)_n]^+$ (n=2-5) complexes in the singlet and quintet states calculated at the PBE0-D3(BJ)/def2-TZVP level of theory (bond lengths in Ångströms and bond angles in degrees).

	Binding energy (kcal/mol)	
	singlet	triplet
$TaO_3(CO_2)^+$ - CO_2	29.5	17.8
$TaO_3(CO_2)_2^+$ - CO_2	20.2	15.1
$TaO_{3}(CO_{2})_{3}^{+}-CO_{2}$	12.8	5.8
$TaO_3(CO_2)_4^+$ – CO_2	5.6	5.7

Table S1. Binding energies of successive CO_2 ligands calculated for the singlet and triplet complex isomers at the PBE0-D3(BJ)/def2-TZVP level of theory (in kcal/mol).