

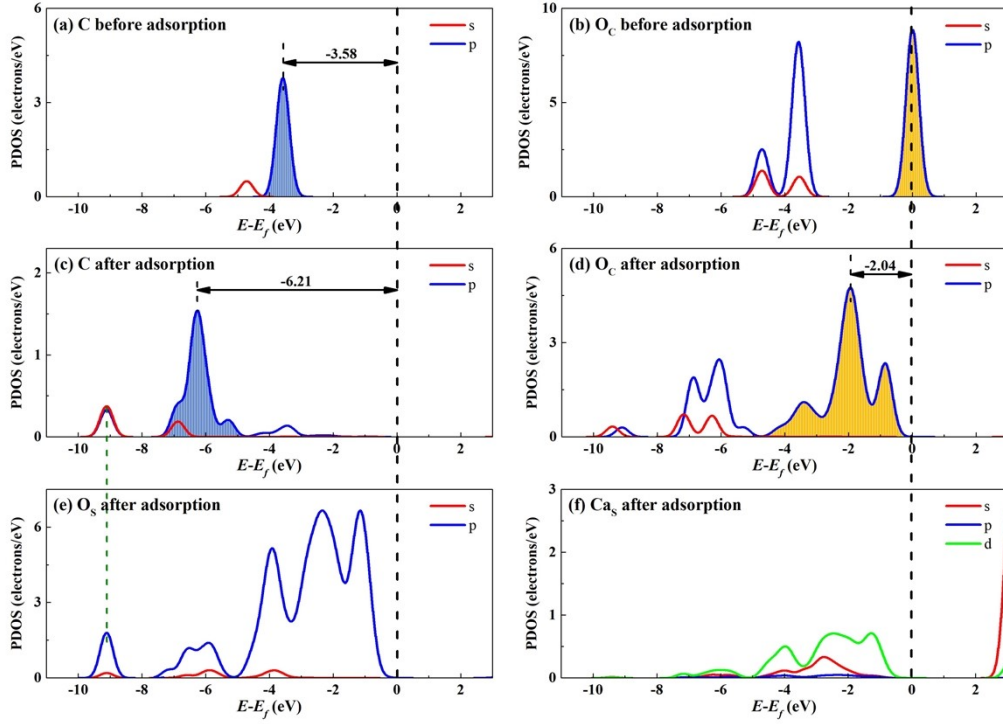
## **CO<sub>2</sub> capture by Li<sub>2</sub>CaSiO<sub>4</sub> and enhancement with alkali carbonates**

Zhen Wang<sup>a</sup>, Chenteng Sun<sup>a\*</sup>, Qian Xu<sup>a\*</sup>, Xingli Zou<sup>a</sup>, Hongwei Cheng<sup>a</sup>, Xionggang Lu<sup>a</sup>

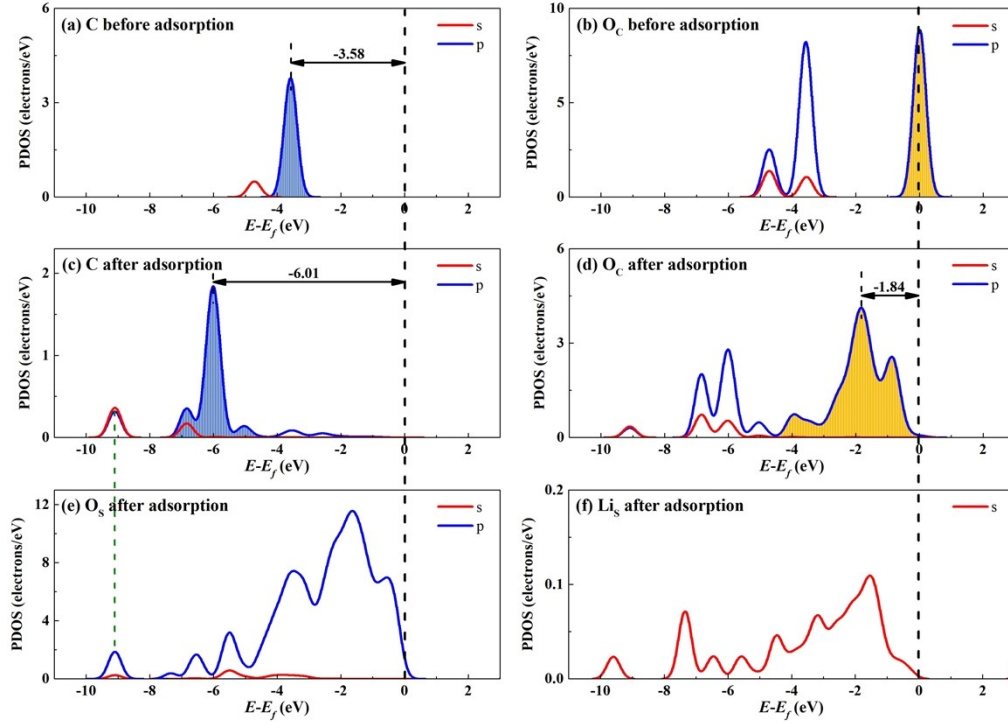
<sup>a</sup> *The State Key Laboratory of Advanced Special Steel,*

*Shanghai Key Laboratory of Advanced Ferrometallurgy,*

*School of Materials Science and Engineering, Shanghai University, Shanghai 200072, P. R. China.*



**Fig. S1.** Density of states analyses of CO<sub>2</sub> before and after adsorption on Li<sub>2</sub>CaSiO<sub>4</sub> (101) surface with top atoms of O and Ca.



**Fig. S2.** Density of states analyses of CO<sub>2</sub> before and after adsorption on Li<sub>2</sub>CaSiO<sub>4</sub> (101) surface with top atoms of O and Li.

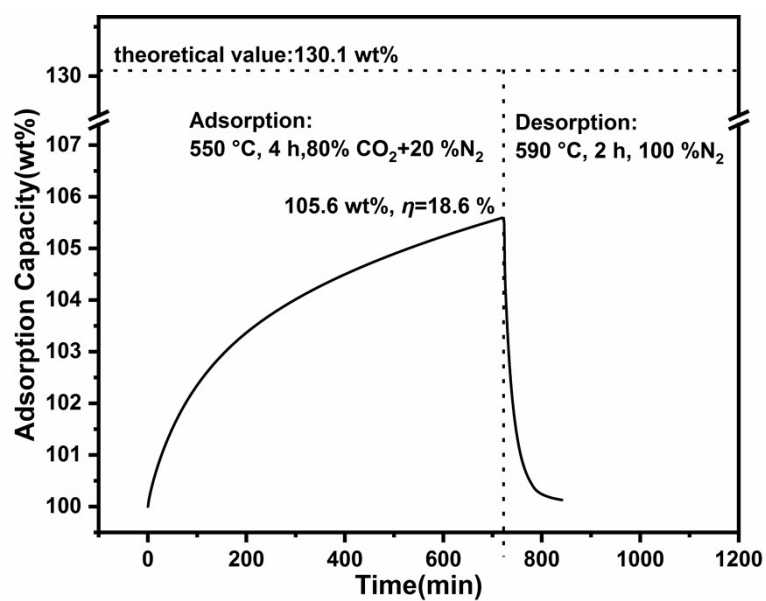


Fig. S3. Isothermal analysis of adsorption at 550 °C in the atmosphere of 80% CO<sub>2</sub> and desorption at 590 °C in 100% N<sub>2</sub>.

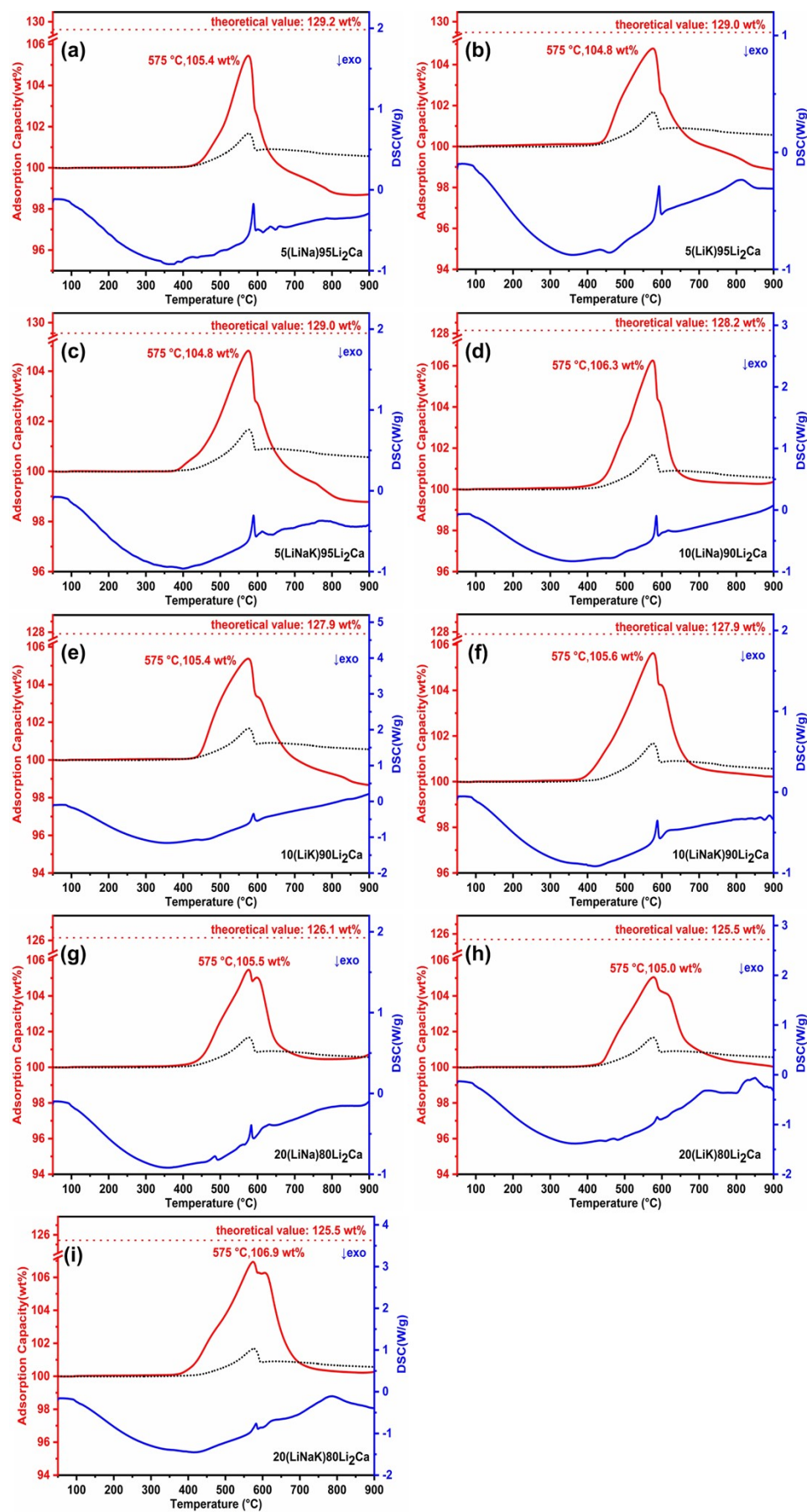


Fig. S4. Dynamic thermal analyses of  $\text{Li}_2\text{CaSiO}_4$  mixed with different alkali carbonates from 50

to 900 °C with a heating rate of 10 °C·min<sup>-1</sup> under the atmosphere containing 80% CO<sub>2</sub>. Dashed line: TG curve of pristine Li<sub>2</sub>CaSiO<sub>4</sub>.

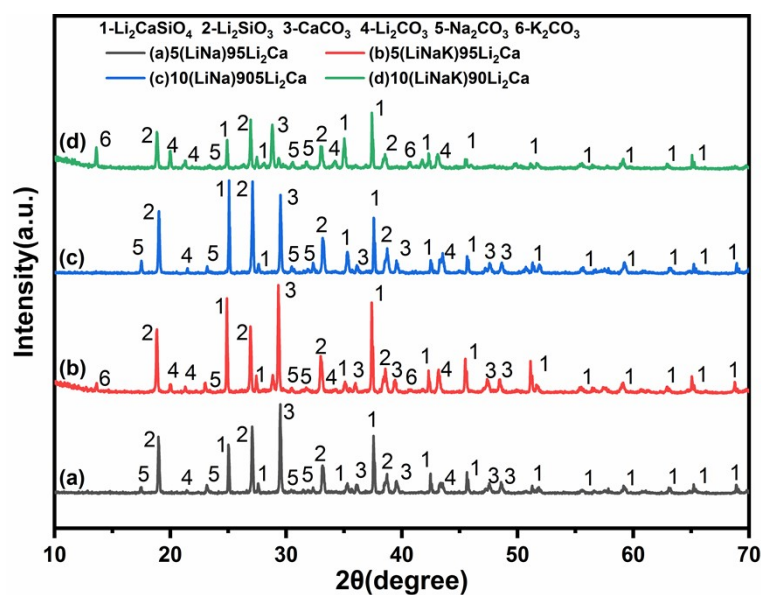


Fig. S5. XRD patterns of the 5(LiNa)95Li<sub>2</sub>Ca(a), 5(LiNaK)95Li<sub>2</sub>Ca(b), 10(LiNa)90Li<sub>2</sub>Ca(c), 10(LiNaK)90Li<sub>2</sub>Ca(d), Li<sub>2</sub>CaSiO<sub>4</sub> after CO<sub>2</sub> adsorption at 550 °C in the 80 % CO<sub>2</sub> for 4 h.

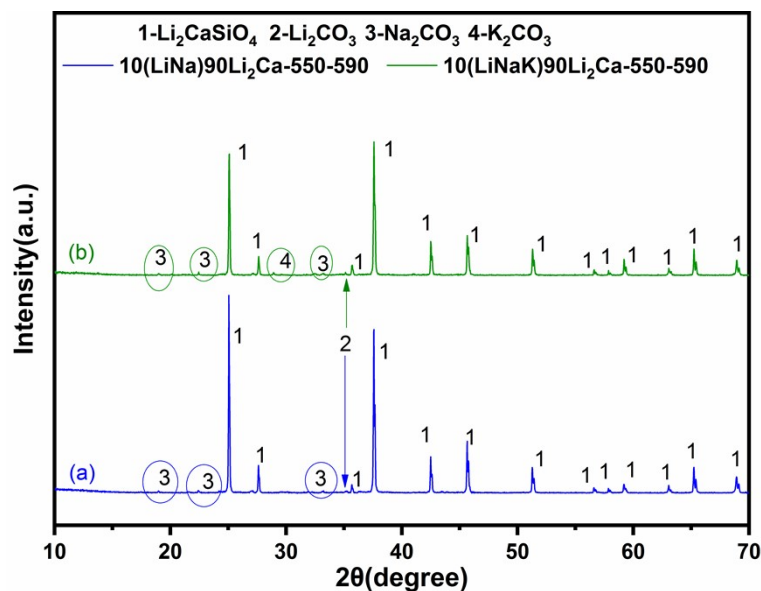


Fig. S6. XRD patterns of the 10(LiNa)90Li<sub>2</sub>Ca(a), 10(LiNaK)90Li<sub>2</sub>Ca(b) after CO<sub>2</sub> desorption at 590 °C in the 100 % N<sub>2</sub> for 2 h.

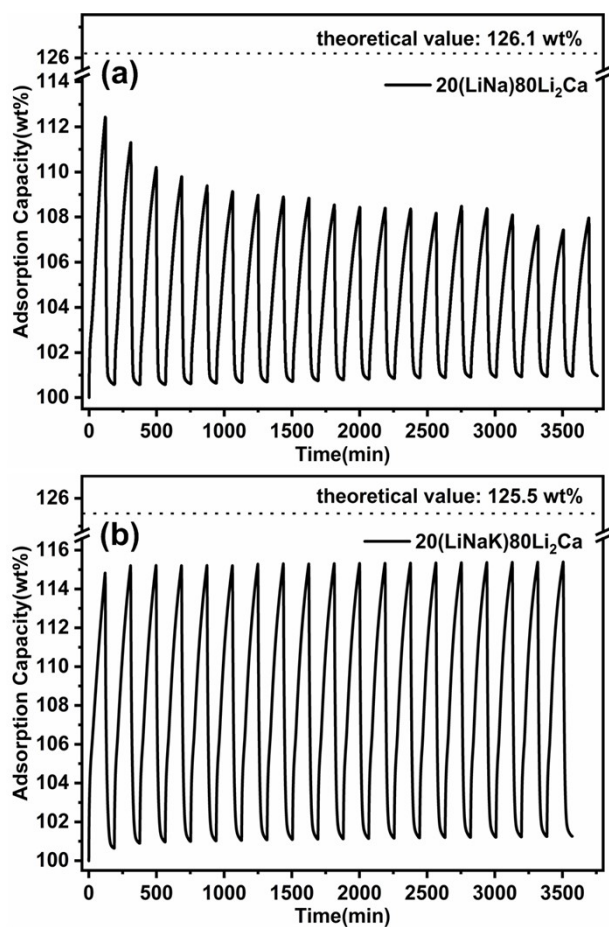


Fig. S7. Cyclic stability of 20(LiNa)80Li<sub>2</sub>Ca(a), and 20(LiNaK)80Li<sub>2</sub>Ca(b) (adsorption: 550°C, 2h, 80% CO<sub>2</sub>, 50 mL·min<sup>-1</sup>; desorption: 590°C, 1h, 100% N<sub>2</sub>, 50 mL·min<sup>-1</sup>)

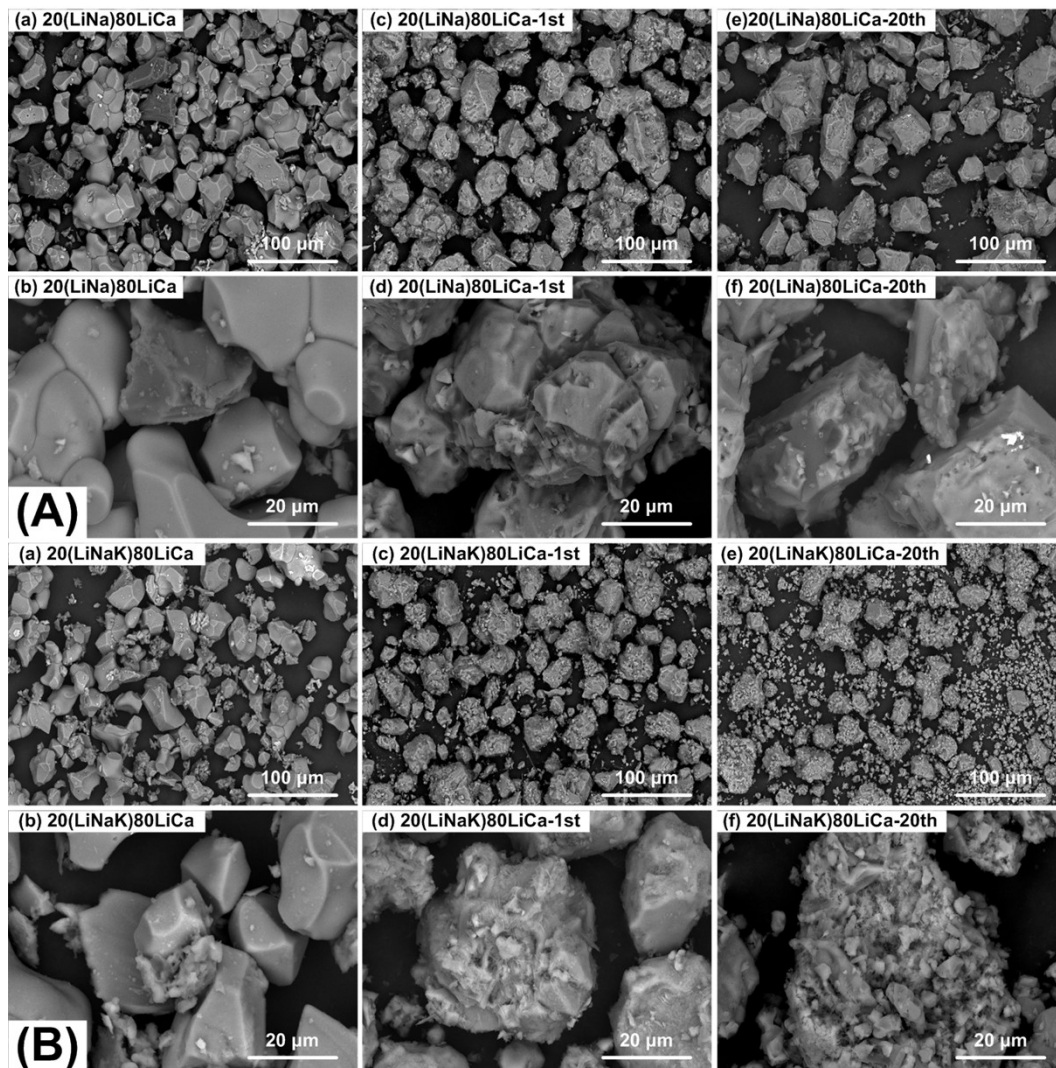


Fig. S8. SEM images of 20(LiNa)80Li<sub>2</sub>Ca (A), and 20(LiNaK)80Li<sub>2</sub>Ca (B) after different cycles

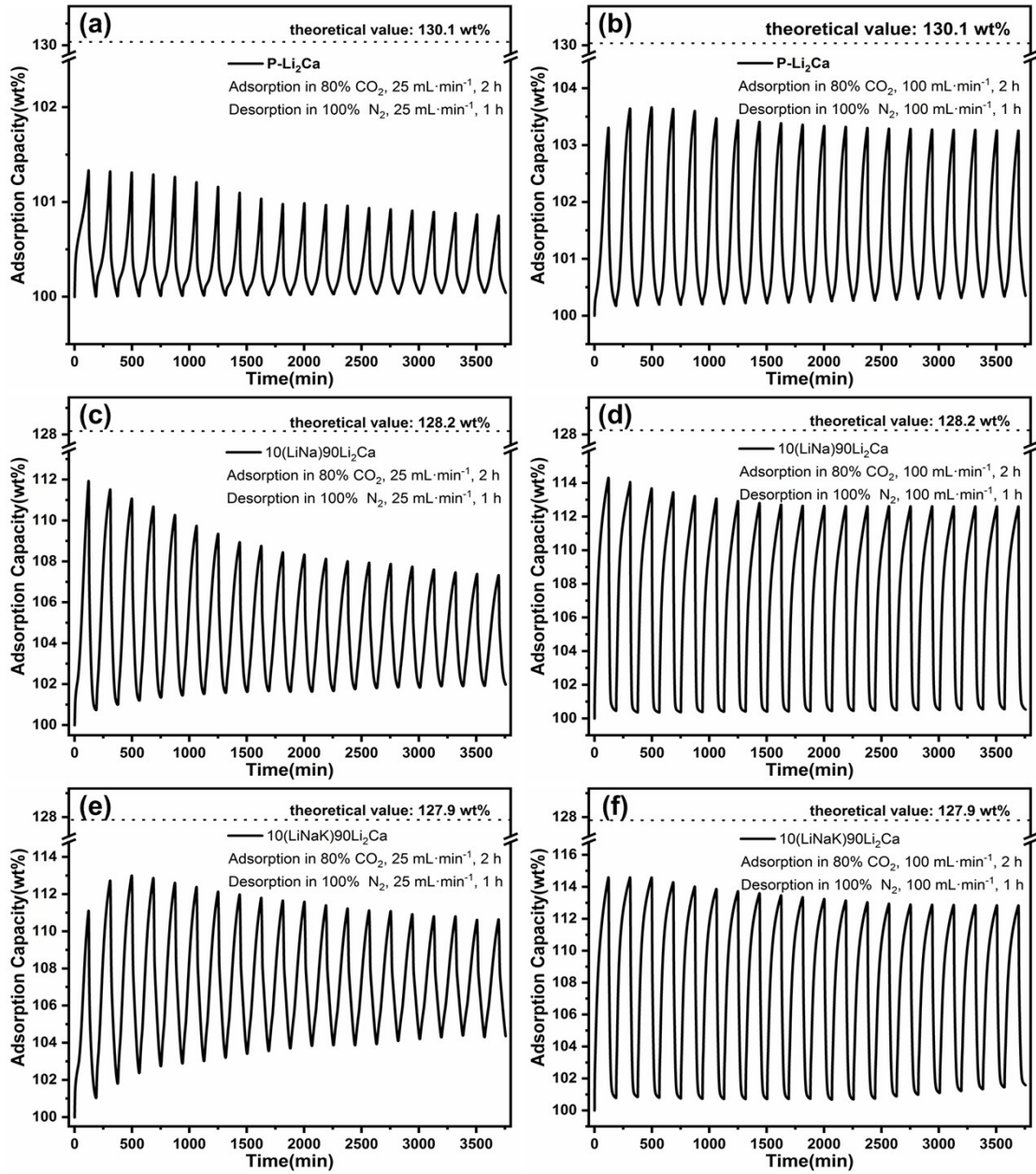


Fig. S9. Cyclic stability of P-Li<sub>2</sub>Ca(a,b), 10(LiNa)90Li<sub>2</sub>Ca(c,d), and 10(LiNaK)90Li<sub>2</sub>Ca(e,f) at different flow rates during 20 capture/release cycles.



**Table S1. Weight gain of P-Li<sub>2</sub>Ca, 10(LiNa)90Li<sub>2</sub>Ca, and 10(LiNaK)90Li<sub>2</sub>Ca at different flow rates during 20 capture/release cycles.**

Weight gain Flow rate	Samples	P-Li <sub>2</sub> Ca	10(LiNa)90Li <sub>2</sub> Ca	10(LiNaK)90Li <sub>2</sub> Ca
	25mL/min	Minimum	0.8 wt%	5.4 wt%
Median		1.0 wt%	7.0 wt%	8.1 wt%
Maximum		1.3 wt%	11.9 wt%	11.1 wt%
50mL/min	Minimum	1.2 wt%	10.6 wt%	9.6 wt%
	Median	1.4 wt%	10.8 wt%	10.4 wt%
	Maximum	1.6 wt%	12.2 wt%	12.3 wt%
100mL/min	Minimum	2.9 wt%	11.1 wt%	11.4 wt%
	Median	3.1 wt%	11.3 wt%	12.7 wt%
	Maximum	3.3 wt%	14.3 wt%	14.6 wt%