

SUPPORTING INFORMATION

A DFT and *ab initio* molecular dynamics simulation study of the infrared spectrum of the protic ionic liquid 2-hydroxyethylammonium formate

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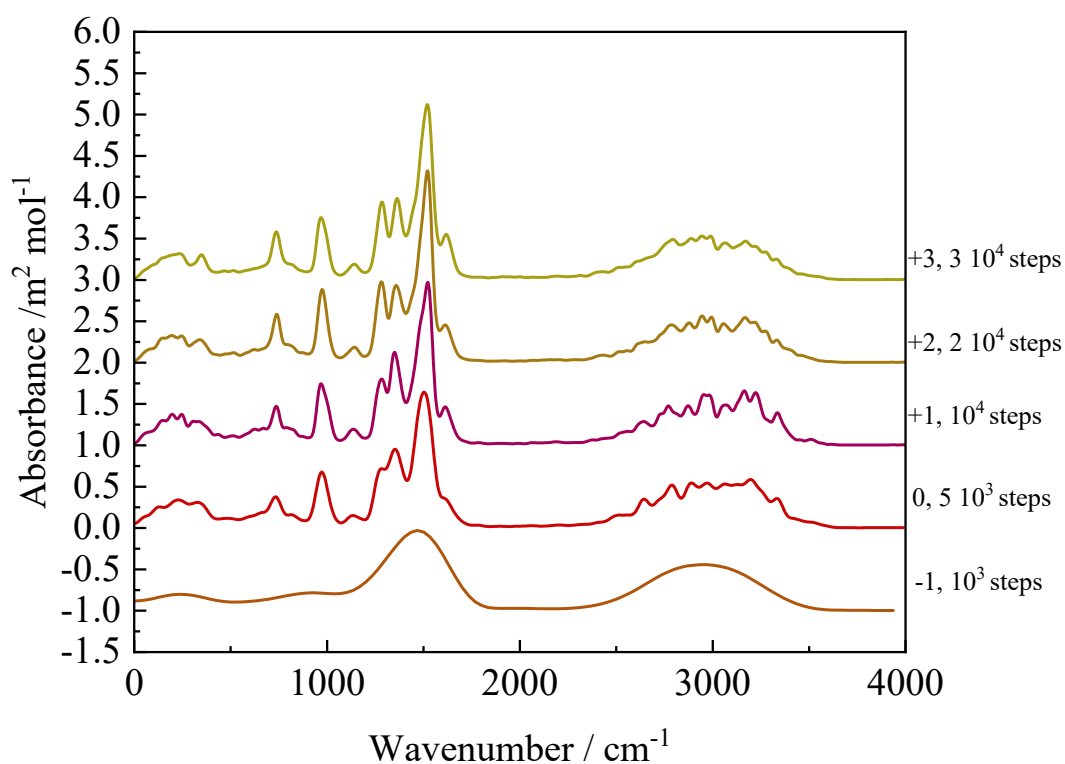


Figure S1. Calculated IR spectra from the AIMD trajectory considering different lengths, ranging from 10^3 steps (brown line, shifted by -1 units of absorbance) to the whole trajectory, 3×10^4 steps (gold line, shifted by three units of absorbance). The spectra are shifted vertically for better visualization.

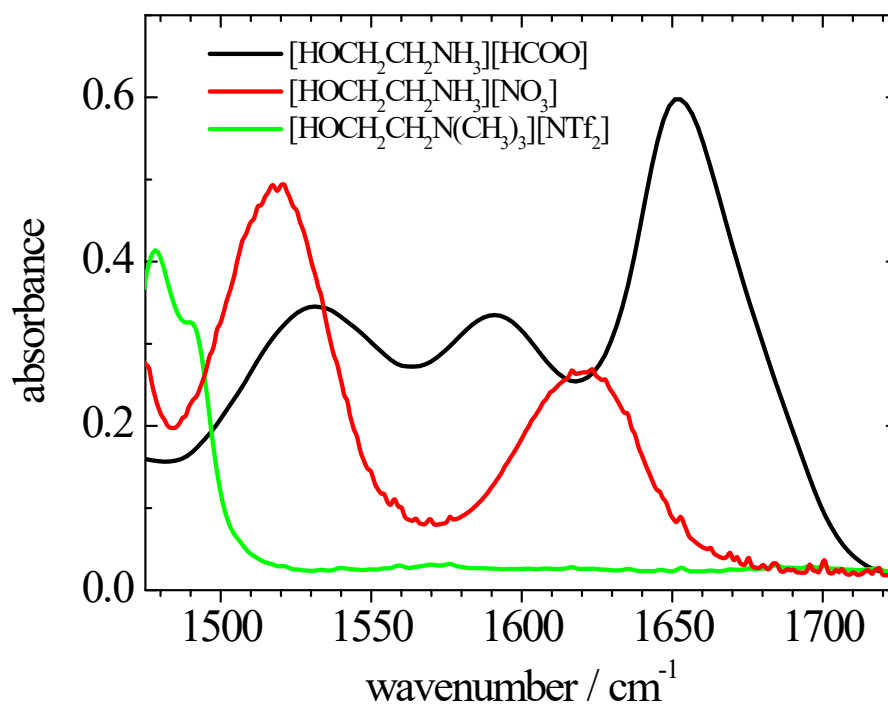


Figure S2. The 1500 – 1700 cm⁻¹ range of the infrared spectrum of the PIL under investigation, [HOCH₂CH₂NH₃][HCOO] (black line), compared with the spectra of 2-hydroxyethylammonium nitrate, [HOCH₂CH₂NH₃][NO₃] (red line), and choline bis(trifluoromethanesulfonyl)imide, [HOCH₂CH₂(NH₃)₃][NTf₂] (green line).

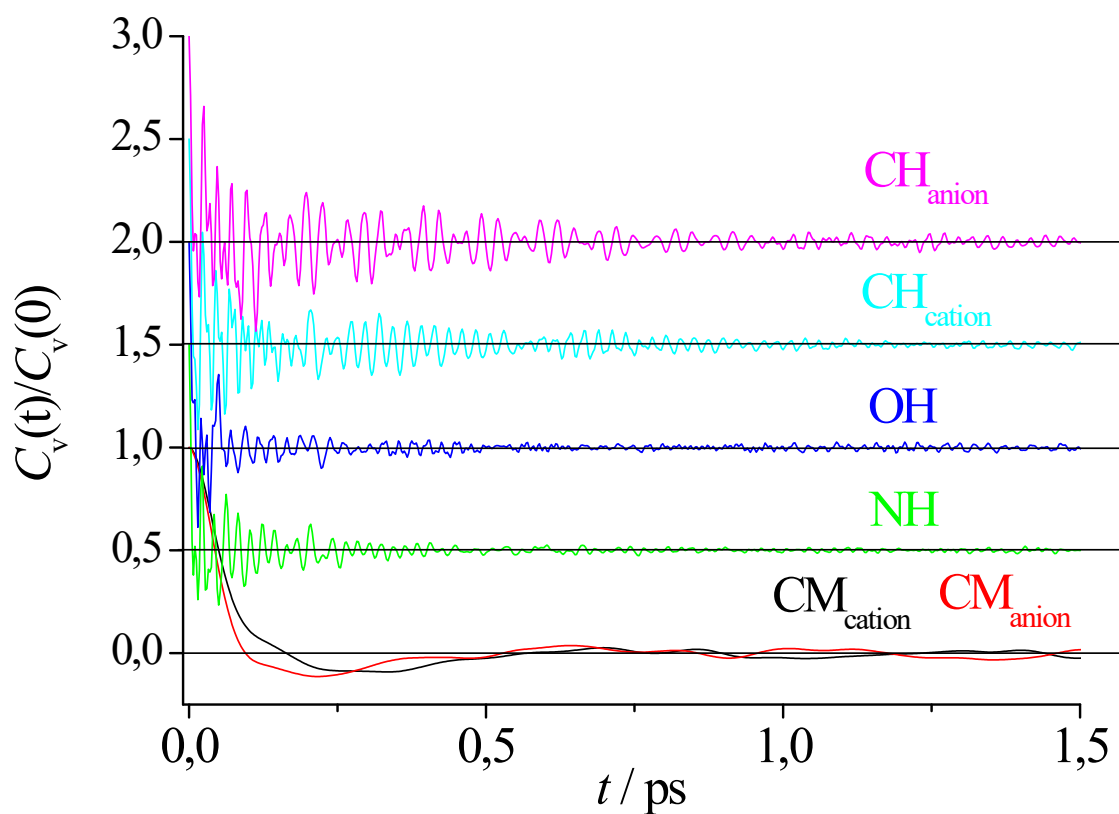


Figure S3. Single-particle time correlation functions of velocities, $C_v(t) = \langle \mathbf{v}_i(0) \cdot \mathbf{v}_i(t) \rangle$, calculated by the AIMD simulation of $[\text{HOCH}_2\text{CH}_2\text{NH}_3][\text{HCOO}]$. The functions were normalized by its initial value, $C_v(0)$, and the upper four functions were vertically shifted by 0.5 to help visualization. The black and red colored $C_v(t)$ at the bottom correspond, respectively, to the cation and anion center-of-mass velocity. The others $C_v(t)$ correspond to velocities of selected atoms as indicated in the figure: N and H atoms of the cation NH_3 group (green line), O and H atoms of the cation OH group (blue line), C and H atoms of the cation CH_2 groups (cyan line), and C and H atoms of the anion (magenta line).

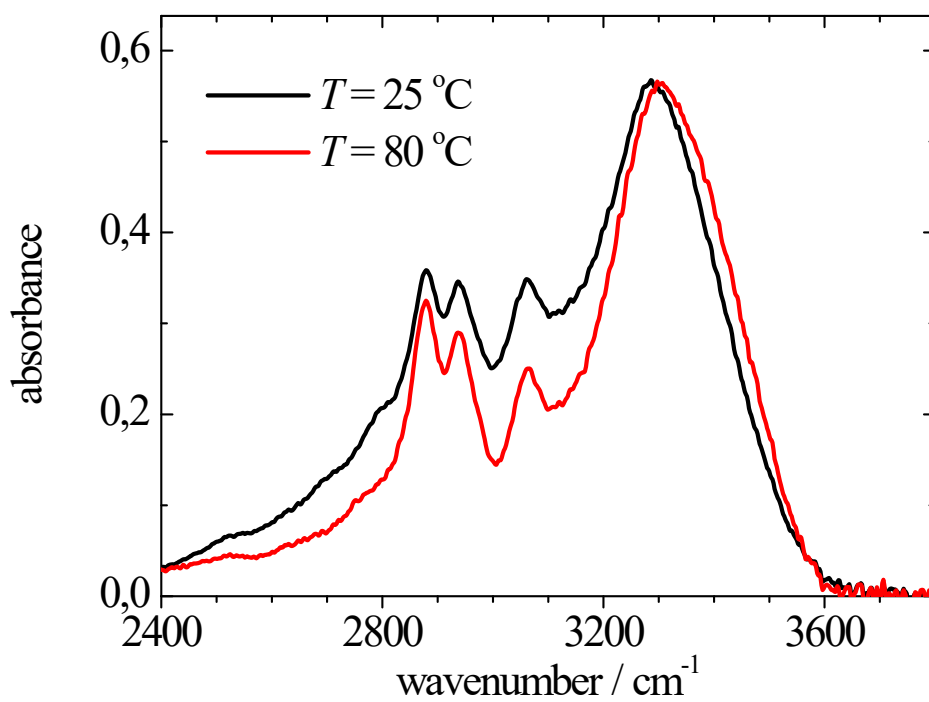


Figure S4. Infrared spectra of [HOCH₂CH₂NH₃][HCOO] at 25 °C (black line) and 80 °C (red line). The intensities have been normalized by the highest intensity peak.

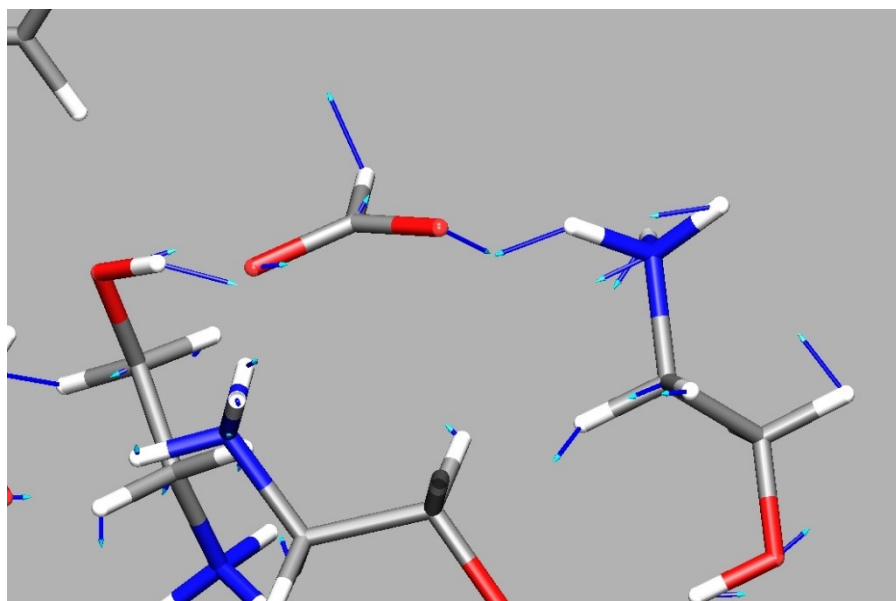


Figure S5. Atomic displacements of the normal mode calculated by DFT at 217 cm^{-1} . The image only emphasizes a portion of the four-ionic pair cluster in order to facilitate better viewing.