

Supporting Information

Theoretical insights into the catalytic mechanism of propylene hydroformylation over Co-N-C materials

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Section1.

Formula for product selectivity

$$N_i = A_i - B_i \quad (S1)$$

$$S_{propane} = \frac{N_{propane}}{N_{isobutyraldehyde} + N_{n-butyraldehyde} + N_{propane}} \quad (S2)$$

$$S_{aldehyde} = \frac{N_{n-butyraldehyde} + N_{isobutyraldehyd}}{N_{n-butyraldehyde} + N_{isobutyraldehyd} + N_{propane}} \quad (S3)$$

$$S_{n-butyraldehyde} = \frac{N_{n-butyraldehyde}}{N_{n-butyraldehyde} + N_{isobutyraldehyd} + N_{propane}} \quad (S4)$$

$$S_{isobutyraldehyde} = \frac{N_{isobutyraldehyd}}{N_{n-butyraldehyde} + N_{isobutyraldehyd} + N_{propane}} \quad (S5)$$

where N represents the actual occurrence frequency of species generated per unit time, the i represents propane, n-butyraldehyde, and isobutyraldehyde, A represents the total occurrence frequency for all reactions that generate the species, and B represents the occurrence frequency for all reactions that consume the species.

Section2.

Table S1 The activation energy of each elementary process of propylene hydroformylation.

Serial Number	elementary process	Activation barrier /eV		
		I	II	III
1	$H_2 \rightleftharpoons H^* + H^*$	0	0	0
2	$H^* + * \rightleftharpoons * + H^*$	\	0.1	0.2

3	$\text{CO} \rightleftharpoons \text{CO}^*$	0	0	0
4	$\text{CH}_3\text{CH}=\text{CH}_2 \rightleftharpoons \text{CH}_3\text{CH}=\text{CH}_2^*$	0	0	0
5	$\text{CH}_3\text{CH}=\text{CH}_2^* + \text{H}^* \rightleftharpoons \text{CH}_3\text{CH}_2\text{CH}_2^*$	0.33	0.58	0.49
6	$\text{CH}_3\text{CH}_2\text{CH}_2^* \rightleftharpoons \text{CH}_3\text{CH}=\text{CH}_2^* + \text{H}^*$	0.8	1.66	1.28
7	$\text{CH}_3\text{CH}=\text{CH}_2^* + \text{H}^* \rightleftharpoons \text{CH}_3\text{CHCH}_3^*$	0.16	0.57	0.13
8	$\text{CH}_3\text{CHCH}_3^* \rightleftharpoons \text{CH}_3\text{CH}=\text{CH}_2^* + \text{H}^*$	0.67	0.84	1.01
9	$\text{CH}_3\text{CH}_2\text{CH}_2^* + \text{H}^* \rightleftharpoons \text{CH}_3\text{CH}_2\text{CH}_3^*$	0.77	0.83	0.88
10	$\text{CH}_3\text{CH}_2\text{CH}_3^* \rightleftharpoons \text{CH}_3\text{CH}_2\text{CH}_2^* + \text{H}^*$	0.26	0.95	0.43
11	$\text{CH}_3\text{CHCH}_3^* + \text{H}^* \rightleftharpoons \text{CH}_3\text{CH}_2\text{CH}_3^*$	0.66	0.86	0.72
12	$\text{CH}_3\text{CH}_2\text{CH}_3^* \rightleftharpoons \text{CH}_3\text{CHCH}_3^* + \text{H}^*$	0.25	0.98	0.49
13	$\text{CH}_3\text{CH}_2\text{CH}_2^* + \text{CO}^* \rightleftharpoons \text{CH}_3\text{CH}_2\text{CH}_2\text{CO}^*$	0.79	0.52	0.70
14	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CO}^* \rightleftharpoons \text{CH}_3\text{CH}_2\text{CH}_2^* + \text{CO}^*$	0.85	1.03	1.11
15	$\text{CH}_3\text{CHCH}_3^* + \text{CO}^* \rightleftharpoons \text{CH}_3\text{CHCH}_3\text{CO}^*$	0.41	0.62	0.45
16	$\text{CH}_3\text{CHCH}_3\text{CO}^* \rightleftharpoons \text{CH}_3\text{CHCH}_3^* + \text{CO}^*$	0.11	0.82	0.88
17	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CO}^* + \text{H}^* \rightleftharpoons$ $\text{CH}_3\text{CH}_2\text{CH}_2\text{CHO}^*$	0.42	0.72	0.95
18	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CHO}^* \rightleftharpoons \text{CH}_3\text{CH}_2\text{CH}_2\text{CO}^* +$ H^*	0.60	0.77	1.30
19	$\text{CH}_3\text{CHCH}_3\text{CO}^* + \text{H}^* \rightleftharpoons \text{CH}_3\text{CHCH}_3\text{CHO}^*$	0.59	0.70	1.01
20	$\text{CH}_3\text{CHCH}_3\text{CHO}^* \rightleftharpoons \text{CH}_3\text{CHCH}_3\text{CO}^* + \text{H}^*$	0.67	0.14	1.70
21	$\text{H}^* + \text{H}^* \rightleftharpoons \text{H}_2$	0.75	0.95	0.55
22	$\text{CO}^* \rightleftharpoons \text{CO}$	2.11	2.72	1.07
23	$\text{CH}_3\text{CH}=\text{CH}_2^* \rightleftharpoons \text{CH}_3\text{CH}=\text{CH}_2$	1.54	1.83	0.32
24	$\text{CH}_3\text{CH}_2\text{CH}_2^* \rightleftharpoons \text{CH}_3\text{CH}_2\text{CH}_2$	3.24	2.36	2.28
25	$\text{CH}_3\text{CHCH}_3^* \rightleftharpoons \text{CH}_3\text{CHCH}_3$	3.04	2.16	2.10
26	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CO}^* \rightleftharpoons \text{CH}_3\text{CH}_2\text{CH}_2\text{CO}$	3.35	2.76	2.34
27	$\text{CH}_3\text{CHCH}_3\text{CO}^* \rightleftharpoons \text{CH}_3\text{CHCH}_3\text{CO}$	3.07	3.89	2.09
28	$\text{CH}_3\text{CH}_2\text{CH}_3^* \rightleftharpoons \text{CH}_3\text{CH}_2\text{CH}_3$	1.62	1.85	0.77
29	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CHO}^* \rightleftharpoons \text{CH}_3\text{CH}_2\text{CH}_2\text{CHO}$	1.86	1.16	0.93

*I, II, and III in the table represent the three catalyst models of Co-N₃-C, 0N-bridged and 2N-bridged, respectively.

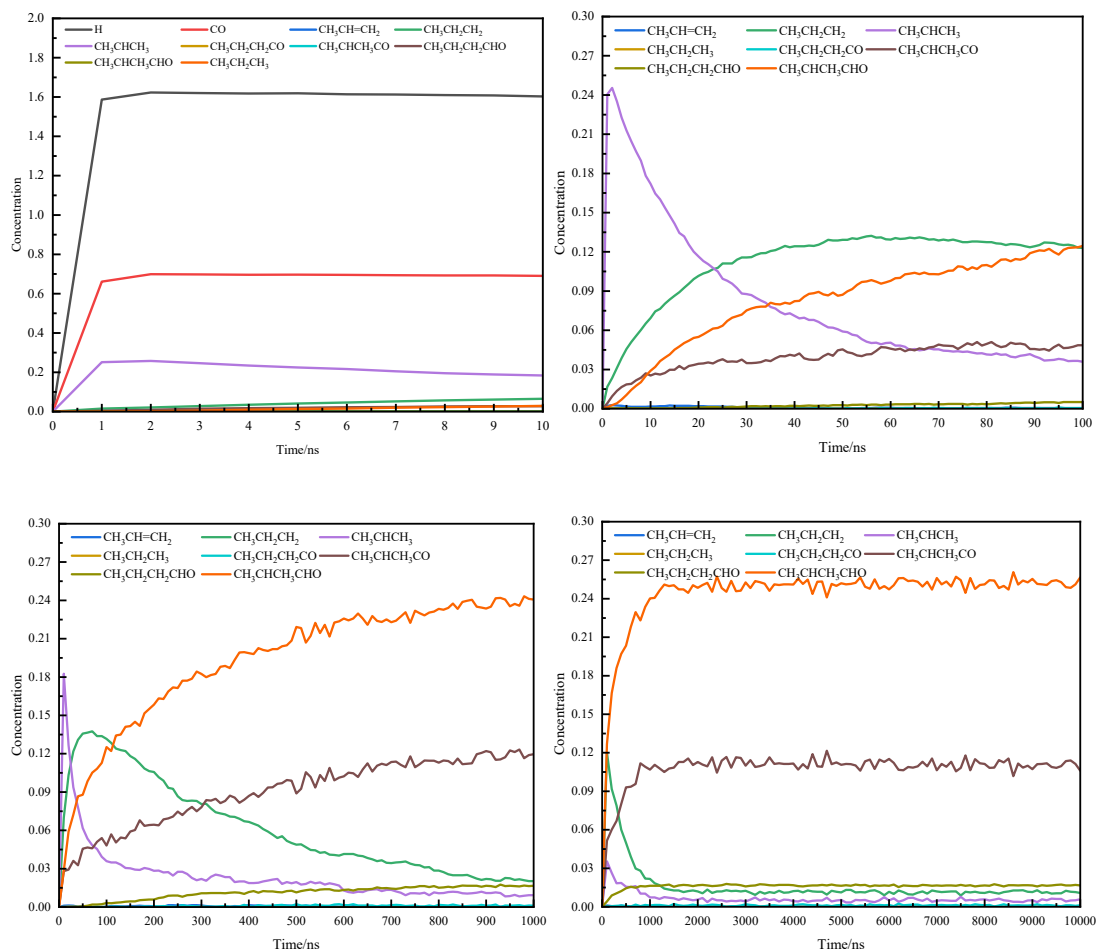


Figure S1 The change of coverage over time within 10, 100, 1000 and 10000 ns on Co-N₃-C

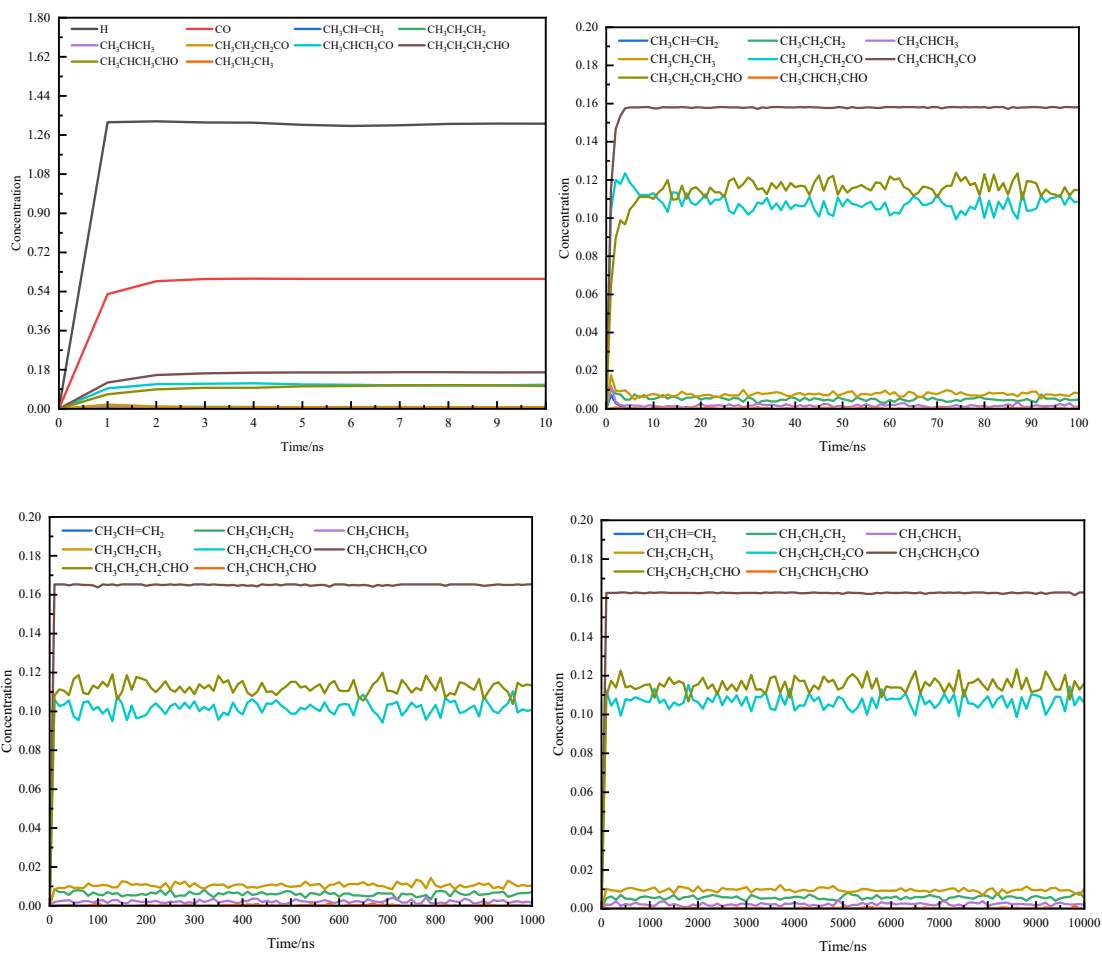


Figure S2 The change of coverage over time within 10, 100, 1000 and 10000 ns on ON-bridged

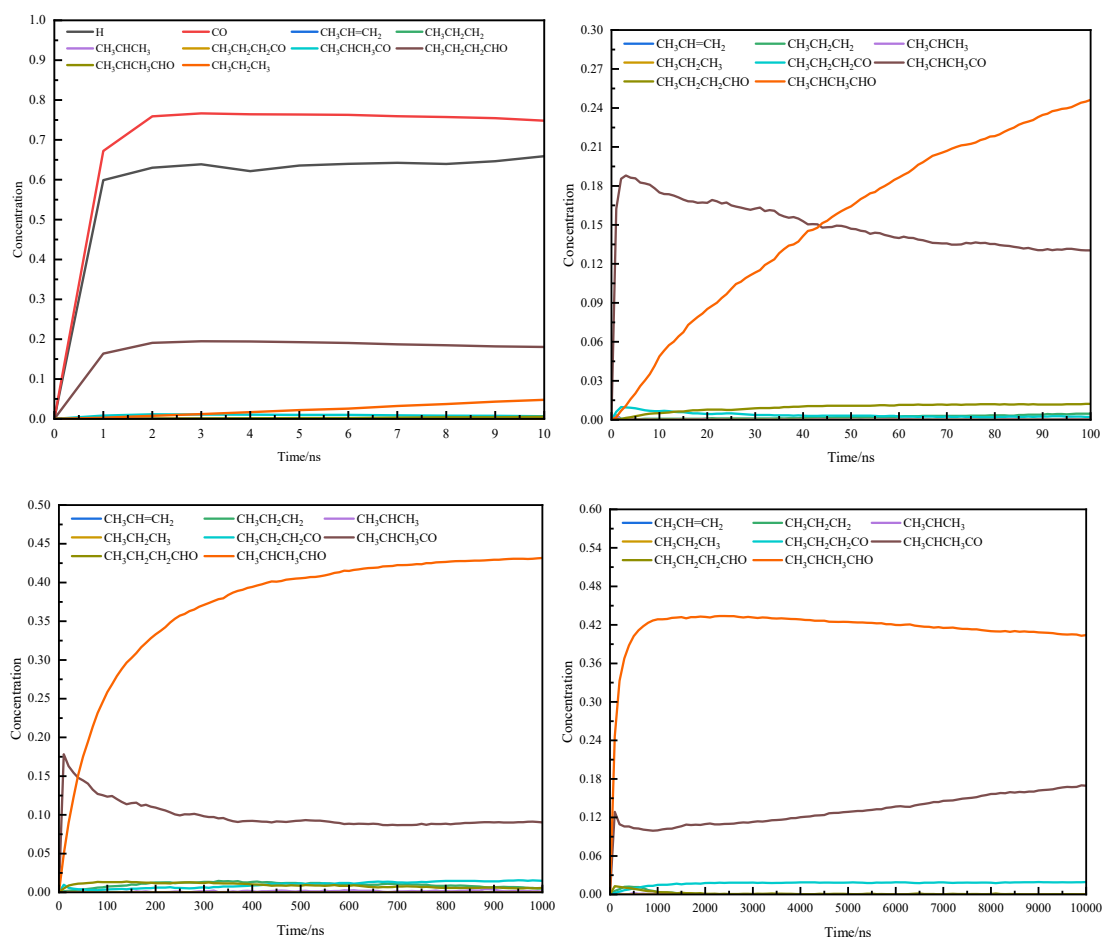


Figure S3 The change of coverage over time within 10,100, 1000 and 10000 ns on 2N-bridged

Table S2 The occurrence frequency of each elementary process in propylene hydroformylation on three models during 10000 ns simulation time.

Serial Number	elementary process	Activation barrier /eV		
		I	II	III
1	$H_2 \rightleftharpoons H^* + H^*$	303289	1529254	3547111
2	$H^* + * \rightleftharpoons * + H^*$	\	325373	1079617
3	$CO \rightleftharpoons CO^*$	4028	4033	404489
4	$CH_3CH=CH_2 \rightleftharpoons CH_3CH=CH_2^*$	1626	1672	164709
5	$CH_3CH=CH_2^* + H^* \rightleftharpoons CH_3CH_2CH_2^*$	910	8071	50
6	$CH_3CH_2CH_2^* \rightleftharpoons CH_3CH=CH_2^* + H^*$	3433	9	235
7	$CH_3CH=CH_2^* + H^* \rightleftharpoons CH_3CHCH_3^*$	23517	8921	4209
8	$CH_3CHCH_3^* \rightleftharpoons CH_3CH=CH_2^* + H^*$	19368	15311	1595
9	$CH_3CH_2CH_2^* + H^* \rightleftharpoons CH_3CH_2CH_3^*$	3696	29578	447
10	$CH_3CH_2CH_3^* \rightleftharpoons CH_3CH_2CH_2^* + H^*$	6336	22506	712
11	$CH_3CHCH_3^* + H^* \rightleftharpoons CH_3CH_2CH_3^*$	10269	9076	265
12	$CH_3CH_2CH_3^* \rightleftharpoons CH_3CHCH_3^* + H^*$	7629	16123	0
13	$CH_3CH_2CH_2^* + CO^* \rightleftharpoons CH_3CH_2CH_2CO^*$	73	1027	203
14	$CH_3CH_2CH_2CO^* \rightleftharpoons CH_3CH_2CH_2^* + CO^*$	0	64	124
15	$CH_3CHCH_3^* + CO^* \rightleftharpoons CH_3CHCH_3CO^*$	623542	1356	22176
16	$CH_3CHCH_3CO^* \rightleftharpoons CH_3CHCH_3^* + CO^*$	622057	710	19827
17	$CH_3CH_2CH_2CO^* + H^* \rightleftharpoons CH_3CH_2CH_2CHO^*$	158389	2260731	270
18	$CH_3CH_2CH_2CHO^* \rightleftharpoons CH_3CH_2CH_2CO^* + H^*$	158321	2260214	270
19	$CH_3CHCH_3CO^* + H^* \rightleftharpoons CH_3CHCH_3CHO^*$	639722	385517	2211

20	$\text{CH}_3\text{CHCH}_3\text{CHO}^* \rightleftharpoons \text{CH}_3\text{CHCH}_3\text{CO}^* + \text{H}^*$	638671	385516	556
21	$\text{H}^* + \text{H}^* \rightleftharpoons \text{H}_2$	299200	1525471	3543024
22	$\text{CO}^* \rightleftharpoons \text{CO}$	0	0	400417
23	$\text{CH}_3\text{CH}=\text{CH}_2^* \rightleftharpoons \text{CH}_3\text{CH}=\text{CH}_2$	0	0	162280
24	$\text{CH}_3\text{CH}_2\text{CH}_2^* \rightleftharpoons \text{CH}_3\text{CH}_2\text{CH}_2$	0	0	0
25	$\text{CH}_3\text{CHCH}_3^* \rightleftharpoons \text{CH}_3\text{CHCH}_3$	0	0	0
26	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CO}^* \rightleftharpoons \text{CH}_3\text{CH}_2\text{CH}_2\text{CO}$	0	0	0
27	$\text{CH}_3\text{CHCH}_3\text{CO}^* \rightleftharpoons \text{CH}_3\text{CHCH}_3\text{CO}$	0	0	0
28	$\text{CH}_3\text{CH}_2\text{CH}_3^* \rightleftharpoons \text{CH}_3\text{CH}_2\text{CH}_3$	0	0	0
29	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CHO}^* \rightleftharpoons \text{CH}_3\text{CH}_2\text{CH}_2\text{CHO}$	0	0	0
30	$\text{CH}_3\text{CHCH}_3\text{CHO}^* \rightleftharpoons \text{CH}_3\text{CHCH}_3\text{CHO}^*$	0	0	0

*I, II, and III in the table represent the three catalyst models of Co-N₃-C, 0N-bridged and 2N-bridged, respectively.

Table S3 The product selectivity on three models

	Co-N ₃	0N-bridged	2N-bridged
S _{propane} /%	0	4.6	0
S _{aldehyde} /%	100	95.4	100
S _{n-butyraldehyde} /%	6.1	95.2	0
S _{isobutyraldehyde} /%	93.9	0.2	100