Supplementary Information

Efficient Exploration of Transition-Metal Decorated MXene for Carbon Monoxide Sensing Using Integrated Active Learning and Density Functional Theory

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Surface	E_{ads} (eV)	Surface	E_{ads} (eV)	Surface	E_{ads} (eV)
Ti ₂ CO ₂	-0.14	Zr ₂ CO ₂	-0.13	Hf_2CO_2	-0.14
Ti ₃ C ₂ O ₂	-0.14	$Zr_3C_2O_2$	-0.15	$Hf_3C_2O_2$	-0.15
Ti ₄ C ₃ O ₂	-0.15	$Zr_4C_3O_2$	-0.15	$Hf_4C_3O_2$	-0.15
Ti ₂ CS ₂	-0.13	Zr_2CS_2	-0.13	Hf_2CS_2	-0.13
Ti ₃ C ₂ S ₂	-0.14	$Zr_3C_2S_2$	-0.13	$Hf_3C_2S_2$	-0.13
Ti ₄ C ₃ S ₂	-0.14	$Zr_4C_3S_2$	-0.13	$Hf_4C_3S_2$	-0.16

 Table S1 CO adsorption on pristine MXene

Table S2 Pre-computed initial dataset for initializing the active learning screening

Initial Dataset							
Surface	Method	E _{ads} (eV)	Surface	Method	E _{ads} (eV)		
Fe@Zr ₂ CO ₂	Random	-1.32	Cu@Ti ₃ C ₂ O ₂	Random	-1.75		
Fe@Hf ₃ C2O ₂	Random	-1.83	Os@Ti ₃ C ₂ O ₂	Random	-2.12		
$Co@Hf_4C_3S_2$	Random	-1.45	Nb@Ti ₃ C ₂ S ₂	Random	-1.65		
Ta@Ti ₃ C2O ₂	Random	-1.55	Ta@Ti ₃ C ₂ S ₂	Random	-1.97		
Fe@Hf ₂ CO ₂	Random	-1.73	Ru@Ti ₄ C ₃ O ₂	Random	-1.55		
Au@Hf ₂ CO ₂	Random	-1.80	Nb@Ti ₄ C ₃ O ₂	Random	-1.54		
Ta@Hf ₂ CS ₂	Random	-1.89	Ta@Ti ₄ C ₃ S ₂	Random	-1.98		
Y@Hf ₂ CS ₂	Random	-0.67	Ir@Ti ₄ C ₃ S ₂	Random	-1.46		
W@Hf ₃ C ₂ O ₂	Random	-1.97	Ag@Zr ₂ CO ₂	Random	-1.12		
Rh@Hf ₃ C ₂ O ₂	Random	-1.96	Os@Zr ₂ CO ₂	Random	-2.53		
Re@Hf ₃ C ₂ S ₂	Random	-2.12	Nb@Zr ₂ CS ₂	Random	-1.54		
Mo@Hf ₃ C ₂ S ₂	Random	-1.81	Au@Zr ₂ CS ₂	Random	-1.63		
Pd@Hf ₄ C3O ₂	Random	-1.76	Hf@Zr ₃ C2O ₂	Random	-1.17		
Co@Hf ₄ C3O ₂	Random	-2.33	Ni@Zr ₃ C2O ₂	Random	-1.88		
V@Hf ₄ C3S ₂	Random	-1.54	Pt@Zr ₃ C2S ₂	Random	-1.24		
Fe@Hf ₄ C3S ₂	Random	-1.19	Ti@Zr ₃ C2S ₂	Random	-1.21		
Ta@Ti ₂ CO ₂	Random	-1.51	Sc@Zr ₄ C3O ₂	Random	-0.80		
Ti@Ti ₂ CO ₂	Random	-1.08	$Zr@Zr_4C3O_2$	Random	-1.08		
Mo@Ti ₂ CS ₂	Random	-1.86	Ni@Zr ₄ C3S ₂	Random	-1.23		
Mn@Ti ₂ CS ₂	Random	-1.09	Cr@Zr ₄ C3S ₂	Random	-1.17		

Table S3 Predicted and actual adsorption energy with their value difference of query datapoints at

each iteration.

Iteration 1					Iteration 2					
Surface	Method	E _{ads,D}	_{FT} (eV)	E _{ads,ML} (eV)	Error (eV)	Surface	Method	E _{ads,DFT} (eV)	E _{ads,ML} (eV)	Error (eV)
Ag@Hf2CO2	Explore	-1	.07	-1.66	0.59	Ag@Ti ₃ C ₂ S ₂	Explore	-1.07	-1.63	0.56
Au@Ti ₂ CO ₂	Explore	-2	2.80	-1.32	-1.48	Co@Ti ₂ CO ₂	Explore	-1.68	-1.36	-0.32
Cr@Ti ₂ CO ₂	Explore	-1	.52	-1.36	-0.16	Cu@Zr ₂ CS ₂	Explore	-1.06	-1.49	0.43
Ir@Zr ₂ CO ₂	Explore	-2	2.56	-1.58	-0.98	Ir@Hf2CO2	Explore	-2.76	-1.66	-1.10
Mn@Ti ₂ CO ₂	Explore	-1	.19	-1.39	0.20	Ni@Hf2CS2	Explore	-1.35	-1.52	0.17
Pt@Hf ₂ CO ₂	Explore	-3	.45	-1.65	-1.80	Pd@Hf ₂ CS ₂	Explore	-1.20	-1.50	0.30
Rh@Zr ₄ C ₃ O ₂	Explore	-1	.96	-1.38	-0.58	Pt@Ti ₂ CS ₂	Explore	-2.01	-1.65	-0.36
Ta@Hf ₃ C ₂ O ₂	Explore	-1	.74	-1.66	-0.08	Ta@Hf ₄ C ₃ O ₂	Explore	-1.77	-1.64	-0.13
V@Hf ₂ CO ₂	Explore	-1	.39	-1.66	0.27	Ti@Hf ₂ CO ₂	Explore	-1.28	-1.69	0.41
Zr@Zr ₂ CO ₂	Explore	-1	.17	-1.56	0.39	W@Zr ₂ CS ₂	Explore	-2.13	-1.51	-0.62
Explore RM	SE = 0.8586 e	V		Exploit RMSE	= N/A	Explore RM	SE = 0.5127	eV	Exploit RMSE	= N/A
		Itera	tion 3					Iteration 4		
Surface	Method	E _{ads,D}	_{FT} (eV)	E _{ads,ML} (eV)	Error (eV)	Surface	Method	E _{ads,DFT} (eV)	E _{ads,ML} (eV)	Error (eV)
Au@Hf2CS2	Explore	-1	.84	-1.31	-0.53	Pt@Hf ₃ C ₂ O ₂	Explore	-3.37	-2.80	-0.57
Au@Ti ₂ CS ₂	Explore	-1	.82	-1.62	-0.20	Sc@Hf ₂ CO ₂	Explore	-1.06	-0.45	-0.61
Ir@Ti ₂ CO ₂	Explore	-2	2.81	-1.61	-1.20	Sc@Hf ₃ C ₂ O ₂	Explore	-0.93	-1.06	0.13
Pd@Hf ₂ CO ₂	Explore	-1	.76	-1.78	0.02	Sc@Ti ₂ CO ₂	Explore	-0.75	-0.32	-0.43
Pd@Zr ₂ CS ₂	Explore	-1	.13	-1.41	0.28	Sc@Zr ₂ CO ₂	Explore	-0.93	-0.29	-0.64
Pt@Hf ₂ CS ₂	Explore	-1	.49	-1.59	0.10	Sc@Ti ₃ C ₂ O ₂	Exploit	-0.71	-0.78	0.07
Pt@Hf ₃ C ₂ S ₂	Explore	-1	.34	-1.60	0.26	Sc@Zr ₃ C ₂ S ₂	Exploit	-0.75	-0.78	0.03
Pt@Ti ₃ C ₂ S ₂	Explore	-1	.97	-2.07	0.10	Y@Zr ₂ CS ₂	Exploit	-0.65	-0.73	0.08
Pt@Zr ₂ CS ₂	Explore	-1	.38	-1.79	0.41	Y@Zr ₃ C ₂ O ₂	Exploit	-0.72	-0.67	-0.05
Sc@Ti ₂ CS ₂	Explore	-().77	-1.57	0.80	Y@Zr ₄ C ₃ O ₂	Exploit	-0.70	-0.76	0.06
Explore RM	SE = 0.5230 e	V		Exploit RMSE	= N/A	Explore RM	e RMSE = 0.5115 eV Exploit RMSE = 0.0605			0.0605 eV
		Itera	tion 5					Iteration 6		
Surface	Method	E _{ads,D}	_{FT} (eV)	E _{ads,ML} (eV)	Error (eV)	Surface	Method	E _{ads,DFT} (eV)	E _{ads,ML} (eV)	Error (eV)
Fe@Zr ₄ C ₃ S ₂	Exploit	-1	.15	-0.78	-0.37	Sc@Hf ₄ C ₃ S ₂	Exploit	-0.76	-0.89	0.13
Sc@Hf ₂ CS ₂	Exploit	-(0.78	-0.78	0.00	Sc@Ti ₄ C ₃ O ₂	Exploit	-0.71	-0.78	0.07
Sc@Zr ₂ CS ₂	Exploit	-(0.76	-0.67	-0.09	Ti@Zr ₂ CO ₂	Exploit	-1.22	-0.93	-0.29
Sc@Zr ₃ C ₂ O ₂	Exploit	-(0.80	-0.67	-0.13	Y@Hf ₃ C ₂ O ₂	Exploit	-0.85	-0.89	0.04
Sc@Zr ₄ C ₃ S ₂	Exploit	-().74	-0.67	-0.07	Y@Hf ₃ C ₂ S ₂	Exploit	-0.65	-0.76	0.11
Y@Hf ₄ C ₃ S ₂	Exploit	-().64	-0.80	0.16	Y@Hf ₄ C ₃ O ₂	Exploit	-0.80	-0.93	0.13
Y@Ti ₂ CO ₂	Exploit	-(0.67	-0.62	-0.05	Y@Ti ₂ CS ₂	Exploit	-0.67	-0.64	-0.03
Y@Ti ₃ C ₂ O ₂	Exploit	-0.61		-0.64	0.03	Y@Ti ₃ C ₂ S ₂	Exploit	-0.66	-0.82	0.16
Y@Ti ₄ C ₃ O ₂	Exploit	-(0.60	-0.80	0.20	Y@Zr ₃ C ₂ S ₂	Exploit	-0.63	-0.58	-0.05
Y@Zr ₂ CO ₂	Exploit	-(.86	-0.70	-0.16	Y@Zr ₄ C ₃ S ₂	Exploit	-0.63	-0.63	0.00
Explore R	MSE = N/A		Ех	xploit RMSE = 0).1617 eV	Explore l	RMSE = N/A	E	xploit RMSE =	0.1286 eV

		Iterat	ion 7		
Surface	Method	E _{ads,DI}	rt (eV)	E _{ads,ML} (eV)	Error (eV)
Ag@Ti ₂ CS ₂	Exploit	-1.	.06	-1.04	-0.02
Ag@Zr ₃ C ₂ S ₂	Exploit	-1.	.04	-1.10	0.06
Ag@Zr ₄ C ₃ S ₂	Exploit	-1.	.05	-1.05	0.00
Fe@Ti ₃ C ₂ S ₂	Exploit	-1.	.43	-1.10	-0.33
Sc@Hf ₃ C ₂ S ₂	Exploit	-0.	.76	-0.84	0.08
Sc@Hf ₄ C ₃ O ₂	Exploit	-0.	.89	-0.97	0.08
Sc@Ti ₃ C ₂ S ₂	Exploit	-0.	.77	-0.80	0.03
Sc@Ti ₄ C ₃ S ₂	Exploit	-0.	.78	-1.00	0.22
Y@Hf2CO2	Exploit	-1.	.04	-1.03	-0.01
Y@Ti ₄ C ₃ S ₂	Exploit	-0.	.67	-0.81	0.14
Explore RMSE = N/A Exploit RMSE = 0.1395 eV					

Table S4 Summary of candidates having adsorption energy between -0.6 and -0.8 eV. The distance between the anchored-TM atom and the C atom of CO (d(TM-C)), CO adsorption energy (^{E}ads), recovery time (τ) calculated at 300 K, formation energy (^{E}f), and charge transfer (Δq) are listed.

No	Surface	Iteration	Method	E_{ads} (eV)	τ (s)	E_f (eV)	Δq (e-)
1	Y@Hf ₄ C ₃ O ₂	6	Exploit	-0.80	27.24	-1.89	-0.29
2	$Y@Zr_3C_2O_2$	4	Exploit	-0.72	1.24	-1.73	-0.25
3	Y@Ti ₂ CO ₂	5	Exploit	-0.67	0.18	-2.53	-0.22
4	$Y@Zr_4C_3O_2$	4	Exploit	-0.70	0.57	-2.82	-0.22
5	$Sc@Zr_3C_2O_2$	5	Exploit	-0.80	27.24	-1.49	-0.18
6	$Sc@Zr_4C_3O_2$	0	Random	-0.80	27.24	-2.55	-0.18
7	Y@Ti ₃ C ₂ O ₂	5	Exploit	-0.61	0.02	-4.63	-0.15
8	Sc@Ti ₂ CO ₂	4	Explore	-0.75	3.94	-2.29	-0.15
9	Y@Ti ₄ C ₃ O ₂	5	Exploit	-0.60	0.01	-3.40	-0.14
10	Y@Ti ₃ C ₂ S ₂	6	Exploit	-0.66	0.12	-3.02	-0.12
11	$Sc@Ti_3C_2O_2$	4	Exploit	-0.71	0.84	-4.38	-0.12
12	Y@Ti ₂ CS ₂	6	Exploit	-0.67	0.18	-2.83	-0.12
13	Sc@Ti ₂ CS ₂	3	Explore	-0.77	8.54	-2.61	-0.10
14	$Sc@Ti_3C_2S_2$	7	Exploit	-0.77	8.54	-2.81	-0.09
15	Y@Ti ₄ C ₃ S ₂	7	Exploit	-0.67	0.18	-2.98	-0.09
16	Y@Hf ₂ CS ₂	0	Random	-0.67	0.18	-3.12	-0.09
17	$Sc@Ti_4C_3O_2$	6	Exploit	-0.71	0.84	-3.11	-0.07
18	$Sc@Ti_4C_3S_2$	7	Exploit	-0.78	12.57	-2.75	-0.07
19	$Y@Hf_3C_2S_2$	6	Exploit	-0.65	0.08	-3.26	-0.07
20	$Y@Hf_4C_3S_2$	5	Exploit	-0.64	0.06	-3.61	-0.07
21	Sc@Hf ₂ CS ₂	5	Exploit	-0.78	12.57	-2.83	-0.07
22	$Sc@Hf_3C_2S_2$	7	Exploit	-0.76	5.80	-2.97	-0.05
23	$Y@Zr_4C_3S_2$	6	Exploit	-0.63	0.04	-3.78	-0.05
24	$Y@Zr_2CS_2$	4	Exploit	-0.65	0.08	-3.75	-0.05
25	$Sc@Hf_4C_3S_2$	6	Exploit	-0.76	5.80	-3.33	-0.05
26	$Sc@Zr_2CS_2$	5	Exploit	-0.76	5.80	-3.43	-0.04
27	$Y@Zr_3C_2S_2$	6	Exploit	-0.63	0.04	-3.73	-0.03
28	$Sc@Zr_4C_3S_2$	5	Exploit	-0.74	2.68	-3.48	-0.03
29	$Sc@Zr_3C_2S_2$	4	Exploit	-0.75	3.94	-3.41	-0.02

Surface	CO adsorption energy (eV)				
	400 eV	450 eV	500 eV	550 eV	
$Sc@Zr_3C_2O_2$	-0.84	-0.80	-0.80	-0.81	
$Y@Zr_3C_2O_2$	-0.74	-0.72	-0.73	-0.72	
Ag@Ti ₂ CS ₂	-1.08	-1.06	-1.05	-1.06	
Pd@Hf ₄ C ₃ O ₂	-1.79	-1.76	-1.75	-1.76	

Table S5. The example of the CO adsorption energy at different kinetic energy cutoff. The structure is sampled from different type of decorated transition metal and MXene.

Table S6. The hyperparameter, total number of fitting parameter, and the average RMSE of test set from five different seeds. The hyperparameter optimization is done on initial dataset. The training set, validation set, and test set ratio are 0.8:0.1:0.1 in hyperparameter optimization.

Batch	Learning	# Convolution Number of fitting		Test set RMSE
Size	Rate	Layer	parameter	(eV)
4	0.00003	2	58433	0.3668
4	0.00003	4	102465	0.3575
4	0.00003	6	146497	0.3862
4	0.0003	2	58433	0.3294
4	0.0003	4	102465	0.3225
4	0.0003	6	146497	0.3273
4	0.003	2	58433	0.3377
4	0.003	4	102465	0.3529
4	0.003	6	146497	0.3852
8	0.00003	2	58433	0.3627
8	0.00003	4	102465	0.3696
8	0.00003	6	146497	0.3556
8	0.0003	2	58433	0.3617
8	0.0003	4	102465	0.3212
8	0.0003	6	146497	0.3637
8	0.003	2	58433	0.3832
8	0.003	4	102465	0.3329
8	0.003	6	146497	0.3494
16	0.00003	2	58433	0.3433
16	0.00003	4	102465	0.3744
16	0.00003	6	146497	0.3073
16	0.0003	2	58433	0.3641
16	0.0003	4	102465	0.3634
16	0.0003	6	146497	0.3062
16	0.003	2	58433	0.3655
16	0.003	4	102465	0.3567
16	0.003	6	146497	0.4034



Fig. S1 Actual adsorption energy of labeled datapoints and predicted adsorption energy of unlabeled datapoints at iteration 1. The yellow shaded area represents the target adsorption energy range of -0.6 to -0.8 eV.



Fig. S2 mMDS of embedding after training of surrogate model at iteration 1. Circles denote unlabeled data, squares denote labeled data, and crosses denote queried data by K-Center



Fig. S3 Actual adsorption energy of labeled datapoints and predicted adsorption energy of unlabeled datapoints at iteration 2. The yellow shaded area represents the target adsorption energy range of -0.6 to -0.8 eV.



Fig. S4 mMDS of embedding after training of surrogate model at iteration 2. Circles denote unlabeled data, squares denote labeled data, and crosses denote queried data by K-Center



Fig. S5 Actual adsorption energy of labeled datapoints and predicted adsorption energy of unlabeled datapoints at iteration 3. The yellow shaded area represents the target adsorption energy range of -0.6 to -0.8 eV.



Fig. S6 mMDS of embedding after training of surrogate model at iteration 3. Circles denote unlabeled data, squares denote labeled data, and crosses denote queried data by K-Center



Fig. S7 Actual adsorption energy of labeled datapoints and predicted adsorption energy of unlabeled datapoints at iteration 4. The yellow shaded area represents the target adsorption energy range of -0.6 to -0.8 eV. The dash lines denote the range of queried data by Greedy.



Fig. S8 mMDS of embedding after training of surrogate model at iteration 4. Circles denote unlabeled data, squares denote labeled data, crosses denote queried data by K-Center, and stars denote queried data by Greedy.



Fig. S7 Actual adsorption energy of labeled datapoints and predicted adsorption energy of unlabeled datapoints at iteration 5. The yellow shaded area represents the target adsorption energy range of -0.6 to -0.8 eV. The dash lines denote the range of queried data by Greedy.



Fig. S8 mMDS of embedding after training of surrogate model at iteration 5. Circles denote unlabeled data, squares denote labeled data, and stars denote queried data by Greedy.



Fig. S9 Actual adsorption energy of labeled datapoints and predicted adsorption energy of unlabeled datapoints at iteration 6. The yellow shaded area represents the target adsorption energy range of -0.6 to -0.8 eV. The dash lines denote the range of queried data by Greedy.



Fig. S10 mMDS of embedding after training of surrogate model at iteration 6. Circles denote unlabeled data, squares denote labeled data, and stars denote queried data by Greedy.



Fig. S11 Actual adsorption energy of labeled datapoints and predicted adsorption energy of unlabeled datapoints at iteration 7. The yellow shaded area represents the target adsorption energy range of -0.6 to -0.8 eV. The dash lines denote the range of queried data by Greedy.



Fig. S12 mMDS of embedding after training of surrogate model at iteration 7. Circles denote unlabeled data, squares denote labeled data, and stars denote queried data by Greedy.



Fig. S13 Actual adsorption energy of labeled datapoints and predicted adsorption energy of unlabeled datapoints at iteration 8. The yellow shaded area represents the target adsorption energy range of -0.6 to -0.8 eV.



Fig. S14 mMDS of embedding after training of surrogate model at iteration 8. Circles denote unlabeled data, and squares denote labeled data.



Fig. S15 T TSNE of the embedding at iteration 1-4, **a-d** respectively. Colors are based on (Left) different surfaces. (Middle) different anchored metal. These two columns use the same color as the legends of Fig. S1 and S2 respectively. (Right) Colors represent the actual E_{ads} for labeled data and the predicted E_{ads} for unlabeled data, with lower E_{ads} color being red and higher E_{ads} being blue. Circles denote unlabeled data, and squares denote labeled data.



Fig. S16 TSNE of the embedding at iteration 5-8, a-d respectively. Colors are based on (Left) different surfaces. (Middle) different anchored metal. These two columns use the same color as the legends of Fig. S1 and S2 respectively. (Right) Colors represent the actual E_{ads} for labeled data and the predicted E_{ads} for unlabeled data, with lower E_{ads} color being red and higher E_{ads} being blue. Circles denote unlabeled data, and squares denote labeled data.



Fig. S17 Energy fluctuation of CO adsorbed on Y@ $Zr_3C_2O_2$ from AIMD simulation at the temperature of 300 K for 8 ps with a time step of 2 fs. The total simulation consists of 2 ps for the equilibration period and 6 ps for the production period.



Fig. S18 Snapshot of initial and final structure from AIMD simulation of CO adsorbed on $Y@Zr_3C_2O_2$. (a-b) side views at 0 and 8 ps (c-d) top views at 0 and 8 ps. The distance between Y and C atoms are represented in angstrom.



Fig. S19 Energy fluctuation of CO adsorbed on $Sc@Zr_3C_2O_2$ from AIMD simulation at the temperature of 300 K for 8 ps with a time step of 2 fs. The total simulation consists of 2 ps for the equilibration period and 6 ps for the production period.



Fig. S20 Snapshot of initial and final structure from AIMD simulation of CO adsorbed on $Sc@Zr_3C_2O_2$. (a-b) side views at 0 and 8 ps (c-d) top views at 0 and 8 ps. The distance between Sc and C atoms are represented in angstrom.