

## **Supplementary Information for**

### **High-throughput computational materials screening of transition metal peroxides**

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Phase	Lattice Parameters (Å)	Atomic coordinates (fractional)
ScO <sub>2</sub> [ <i>P2<sub>1</sub>/c</i> ]	a = 3.724, b = 5.838, c = 5.902 $\alpha = 92.718, \beta = \gamma = 90.000$	Sc 0.028 0.224 0.839
		O1 0.460 0.859 0.317
		O2 0.845 0.548 0.440
TiO <sub>3</sub> [ <i>Pmmn</i> ]	a = 2.948, b = 3.948, c = 7.113 $\alpha = \beta = \gamma = 90.000$	Ti 0.000 0.500 0.350
		O1 0.500 0.500 0.545
		O2 0.000 0.184 0.818

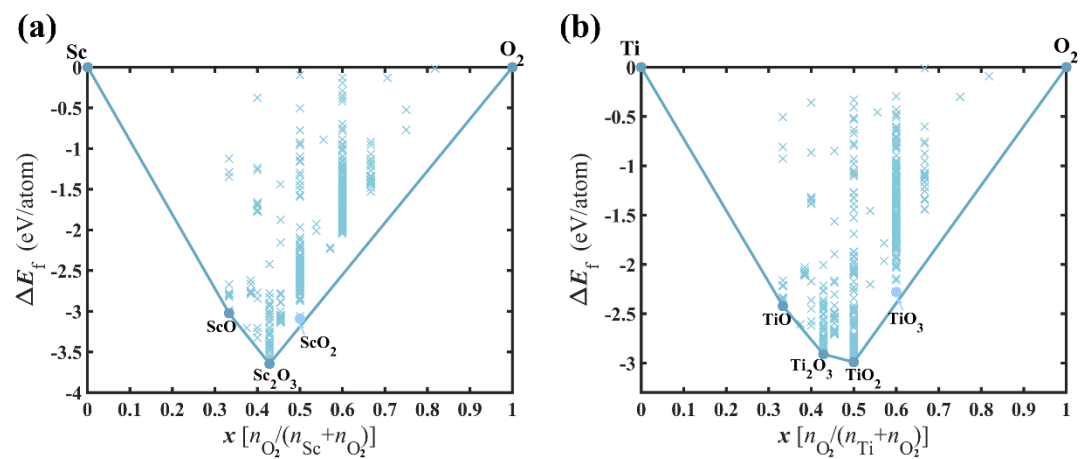
**Table S1.** Detailed structural information of the ScO<sub>2</sub> and TiO<sub>3</sub>.

Phase	Atom	Charge	$\sigma(e)$
ScO <sub>2</sub> [ <i>P2<sub>1</sub>/c</i> ]	Sc	1.22	-1.78
	O1	7.16	1.16
	O2	6.62	0.62
TiO <sub>3</sub> [ <i>Pmmn</i> ]	Ti	2.12	-1.88
	O1	6.98	0.98
	O2	6.45	0.45

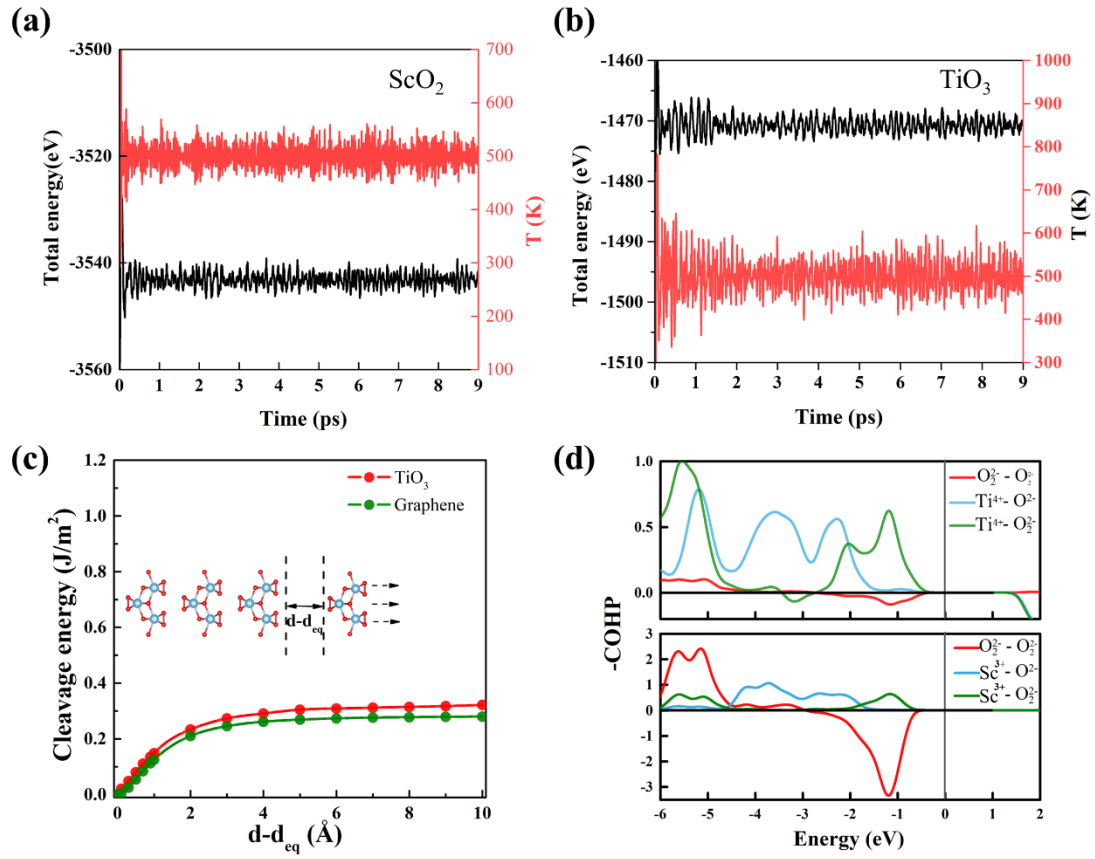
**Table S2.** Residual charges on metal and O atoms based on the Bader charge analysis in ScO<sub>2</sub> and TiO<sub>3</sub>. The positive and negative values of  $\sigma$  represent the number of charges gained and lost by the atom.

The calculated formula for the convex hulls of formation enthalpy ( $\Delta E_f$ ) is defined as follows:

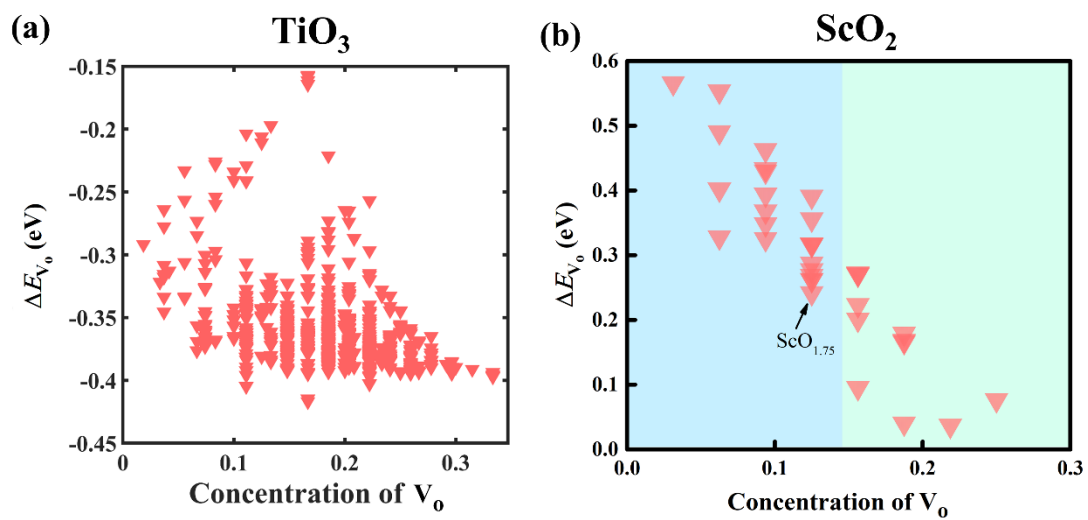
$$\Delta E_f(\text{MO}_x) = [E(\text{MO}_x) - E(\text{M}) - xE(\text{O})] / (1+x), \text{M}=\text{Sc, Ti}$$



**Fig. S1.** (a), (b) The formation enthalpy of various Sc-O and Ti-O compounds.



**Fig. S2** (a), (b) Fluctuations of total energy and temperature with respect to AIMD steps at 500 K for ScO<sub>2</sub> and TiO<sub>3</sub>. (c) Calculated cleavage energy of monolayer TiO<sub>3</sub> compared with the cleavage energy of monolayer graphene. (d) Negative crystal orbital Hamilton population (-COHP) of average  $O_2^{2-} - O_2^{2-}$ ,  $Ti^{4+} - O_2^{2-}$ ,  $Ti^{4+} - O_2^{2-}$ ,  $Sc^{3+} - O_2^{2-}$  and  $Sc^{3+} - O_2^{2-}$  bonds in the bulk ScO<sub>2</sub> and single layered TiO<sub>3</sub>.



**Fig. S3** (a) The average formation energy  $\Delta E_{V_0}$  varies with oxygen vacancy concentration for the  $2 \times 2 \times 1$ ,  $\sqrt{5} \times \sqrt{5} \times 1$ , and  $3 \times 3 \times 1$  supercell of  $\text{TiO}_3$ . (b) The average formation energy  $\Delta E_{V_0}$  varies with oxygen vacancy concentration for  $2 \times 2 \times 1$  supercell of  $\text{ScO}_2$ . The right part indicates that the defect structure exhibits a higher degree of distortion.