

Supporting Information

A synergetic promotion of surface stability for high-voltage LiCoO₂ by multi-element surface doping: a first-principles study

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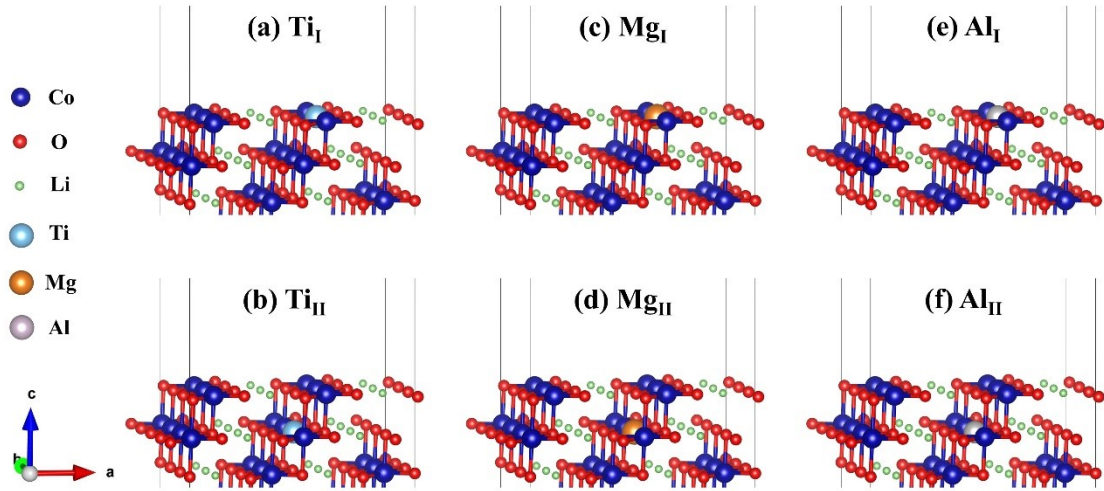


Fig. S1 Schematic illustrations of (a, b) two types of doping configurations for Ti-doped LiCoO₂ (104) surface, (c, d) two types of doping configurations for Mg-doped LiCoO₂ (104) surface and (e, f) two types of doping configurations for Al-doped LiCoO₂ (104) surface.

Table S1 Calculated formation energies (E_{form}) and energy differences (ΔE_{form} , relative to the surface layer doping configurations) of various doping configurations for Ti-doped, Mg-doped and Al-doped LiCoO₂ (104) surfaces.

System	Doping configurations	Formation energy E_{form} (eV)	Energy difference ΔE_{form} (eV)
Ti-doped LiCoO ₂ (104)	Ti _I	-2.50	0
	Ti _{II}	-2.02	0.48
Mg-doped LiCoO ₂ (104)	Mg _I	-3.65	0
	Mg _{II}	-4.43	-0.78
Al-doped LiCoO ₂ (104)	Al _I	-5.75	0
	Al _{II}	-7.99	-2.24

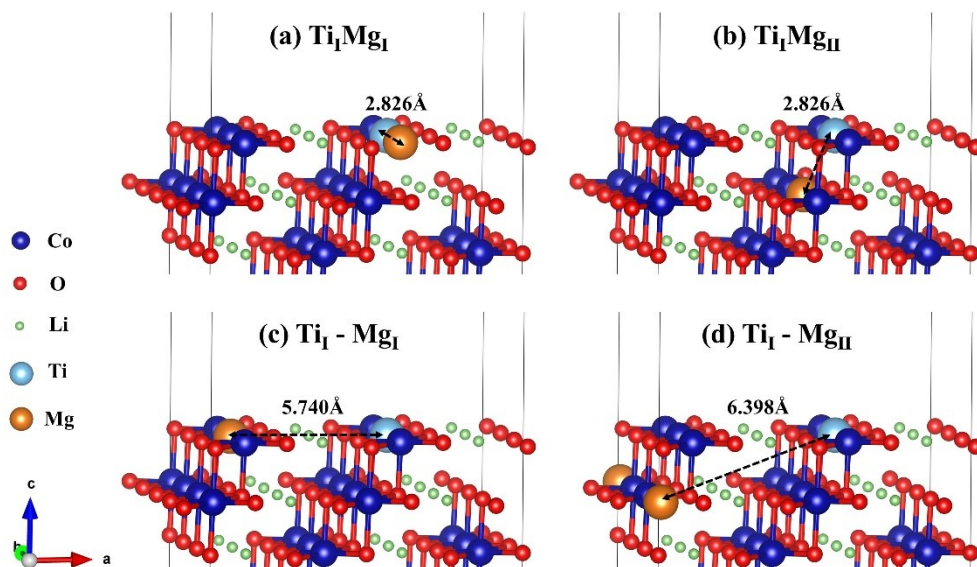


Fig. S2 Schematic illustrations of four types of doping configurations for Ti-Mg co-doped LiCoO_2 (104) surface. The distances between nearest doping Ti and Mg at (104) surface are 2.826 Å, 2.826 Å, 5.740 Å and 6.398 Å in these four doping configurations respectively.

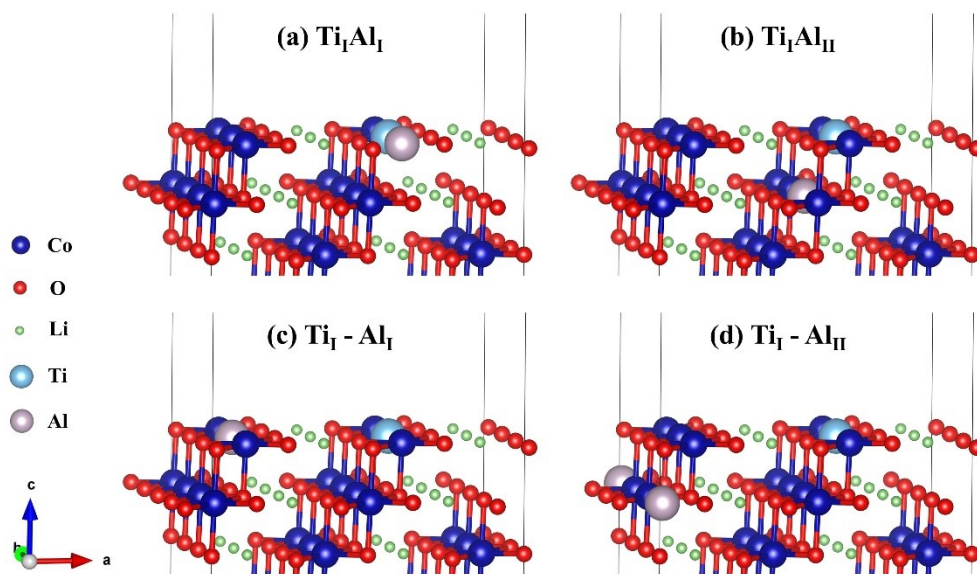


Fig. S3 Schematic illustrations of four types of doping configurations for Ti-Al co-doped LiCoO_2 (104) surface.

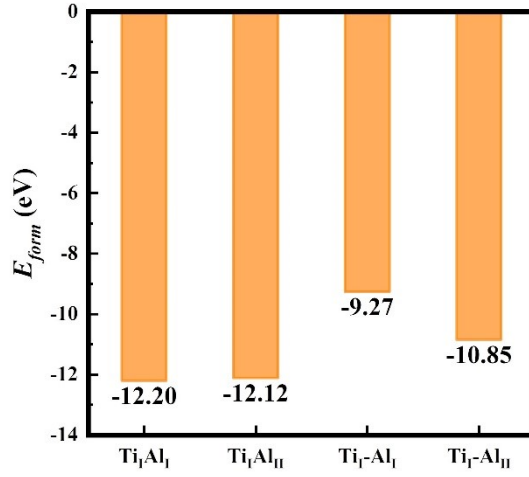


Fig.S4 Calculated formation energies (E_{form}) of various doping configurations for Ti-Al co-doped LiCoO₂ (104) surface.

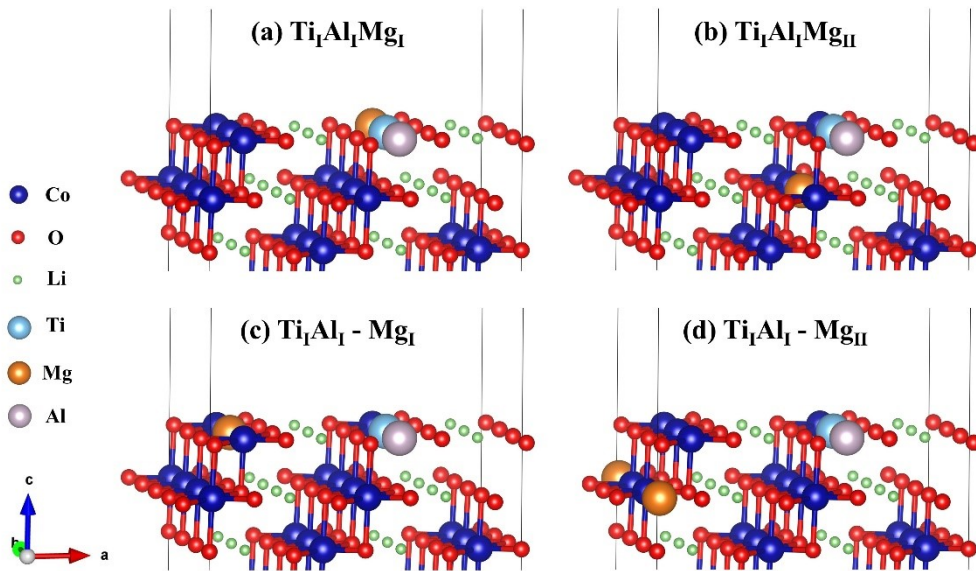


Fig. S5 Schematic illustrations of four types of doping configurations for Ti-Al-Mg co-doped LiCoO₂ (104) surface.

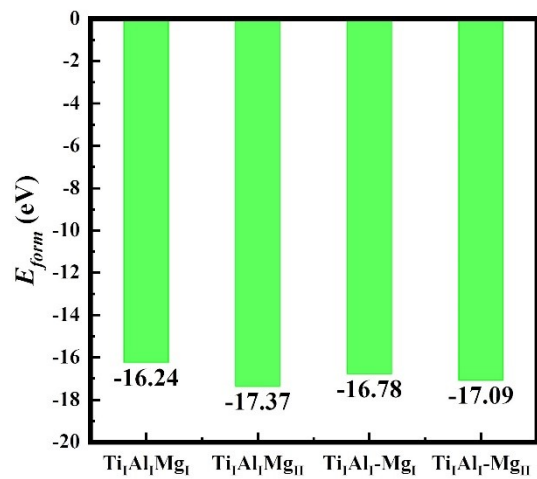


Fig. S6 Calculated formation energies (E_{form}) of various doping configurations for Ti-Al-Mg co-doped $LiCoO_2$ (104) surface.