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## X-ray induced ultrafast charge transfer in thiophenebased conjugated polymers controlled by core-hole clock spectroscopy<sup>†</sup>. Electronic Supplementary Information.

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#### 1 APPENDIX A: SAMPLES CHARACTERISA-TION

Depending on the procedure used for a P3HT film preparation, two orientations of the polymer chains on the substrate are possible<sup>1</sup>. In the edge-on orientation, the self-organised lamella sheets are positioned perpendicular to the substrate, while in the face-on orientation, the lamellae are fixed parallel to the substrate.

Following the prescriptions provided in literature<sup>2</sup>, we have chosen the rotational spin-coating frequency of 2000 revolutions per minute and the chloroform solution concentration of 5 mg/ml that lead to formation of a film with face-on orientation. Furthermore, our x-ray diffraction analysis of the film, using a Cu K $\alpha$  source in a symmetric  $\theta/2\theta$  geometry, revealed a signature at 24° corresponding to the inter-chain  $\pi - \pi$  stacking, which indicates a face-on orientation of the film<sup>3,4</sup>. In addition, no sharp structures typical for an edge-on orientation were observed. The ~34 nm thickness of the P3HT film was determined using atomic force microscopy.



Fig. S1 2D maps of S KL<sub>2,3</sub>L<sub>2,3</sub> Auger spectra recorded in P3HT film at grazing incidence. Transitions to the  $2p^{-2}({}^{1}S)\pi^{*}$  and  $2p^{-2}({}^{1}S)\sigma^{*}$  final states are indicated with black crosses. Transitions to the  $2p^{-2}({}^{1}D)\pi^{*}$  and  $2p^{-2}({}^{1}D)\sigma^{*}$  states, indicated with white crosses, are labelled as  $\pi^{*}$  and  $\sigma^{*}$ . Blue lines indicate the corresponding resonant photon energies. Diagonal dotted lines show dispersive behaviour of resonant structures. The map is normalised to the background. The colour-coded intensities are on a logarithmic scale.

The measurements presented in Fig. 3 of the main text were performed at normal incidence with the polarisation of the incident x-ray beam parallel to the sample surface. Assuming a faceon orientation of the P3HT film, in this geometry one can expect a dominant excitation of the S  $1s^{-1}\sigma^*$  state. Indeed, this can be observed in Fig. 3 of the main text. For comparison, Fig. S1 shows a 2D map of S KL<sub>2,3</sub>L<sub>2,3</sub> Auger spectra for P3HT measured in grazing incidence geometry, where the x-ray beam polarisation is perpendicular to the sample surface, leading to a primary excitation of the S  $1s^{-1}\pi^*$  state. Note, that a dominant excitation of the S  $1s^{-1}\pi^*$  state in the case of grazing incidence geometry and a resulting overlap of the strong resonant  $2p^{-2}\pi^*$  feature with a weak low-energy peak, prevents us from a reliable fitting analysis in the whole photon energy range of the 2D map and therefore, does not allow revealing a possible CT process in this geometry.

A strong angular dependence observed for the P3HT sample, demonstrating dominant excitation of the S  $1s^{-1}\sigma^*$  state for normal incidence, and the S  $1s^{-1}\pi^*$  state for grazing incidence geometry, further corroborates the face-on orientation of the P3HT conjugate plane.

Note that our measurements did not reveal angular depen-

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dence for the PT powder sample, which is consistent with the isotropic morphology of the powdered polymer. Similarly, measurements (not shown here) of a PT film, prepared using a potentiostatic electrochemical deposition technique, lack angular dependence. This points at a poor organisation of the PT film structure, probably related to the imperfect deposition conditions. Therefore, these measurements were excluded from the data analysis.

#### 2 APPENDIX B: EHRENFEST MOLECULAR DY-NAMICS IN P3HT FILM



Fig. S2 Bader charge analysis in P3HT using EMD formalism for the first 3 fs of simulation considering only the excited thiophene ring and its nearest neighbours for (a)  $S 1s^{-1}\pi^*$  and (b)  $S 1s^{-1}\sigma^*$  states. The gray line denotes  $\Delta q = 0$ . Negative (positive) values indicate electron depletion (accumulation).

RT-TDDFT simulations can be compared to the Ehrenfest molecular dynamics (EMD) formalism<sup>5,6</sup>. In this case, the electronic wavefunction evolution is coupled to the nuclei motion in a meanfield approach, obtained by averaging over all quantum mechanical states. Bader charge analysis performed for P3HT using EMD for the first 3 fs of propagation (see Fig. S2), shows very similar results to RT-TDDFT simulations presented in Fig. 9 of the main text. Therefore, the choice of neglecting the nuclear motion in the RT-TDDFT approach does not seem to affect the CT process taking place within the first fs after these specific core-excitations. 3 APPENDIX C: RT-TDDFT SIMULATIONS IN PT FILM



Fig. S3 (a) Ball and stick representation of PT simulation cell employed in the RT-TDDFT simulations. colour code: brown for carbon, yellow for sulfur, green for chlorine (Z+1 approximation), light pink for hydrogen. (b) and (c) Bader charge analysis in S  $1s^{-1}\pi^*$  and S $1s^{-1}\sigma^*$  core-excited states for the first 2 fs of simulation considering the excited thiophene ring and its nearest neighbours. The gray line denotes  $\Delta q = 0$ . Negative (positive) values indicate electron depletion (accumulation).

#### Notes and references

- H. Sirringhaus, P. J. Brown, R. H. Friend, M. M. Nielsen, K. Bechgaard, B. M. W. Langeveld-Voss, A. J. H. Spiering, R. A. J. Janssen, E. W. Meijer, P. Herwig and D. M. de Leeuw, *Nature*, 1999, **401**, 685–688.
- 2 D. M. DeLongchamp, B. M. Vogel, Y. Jung, M. C. Gurau, C. A. Richter, O. A. Kirillov, J. Obrzut, D. A. Fischer, S. Sambasivan, L. J. Richter and E. K. Lin, *Chem. Mater.*, 2005, **17**, 5610–5612.
- 3 M. E. Nicho, W. Jaimes, M. E. Rivas-Aguilar, H. Hu and M. A. Quevedo-Lopez, *Rev. Mex. Fis.*, 2017, 516–524.
- 4 Y. D. Park, Bull. Korean Chem. Soc., 2014, 35, 2277–2280.

- 5 S. Andermatt, J. Cha, F. Schiffmann and J. VandeVondele, J. Chem. Theory Comput., 2016, **12**, 3214–3227.
- 6 B. F. E. Curchod, U. Rothlisberger and I. Tavernelli, *ChemPhysChem*, 2013, **14**, 1314–1340.