"Supporting Information"

First-principles study on discharge electrochemical and catalytic performance of sulfur cathode host $Fe_{0.875}M_{0.125}S_2$ (M = Ti, V)

Cheng-Dong Wei^a, Hong-Tao Xue^{a,b}, Yu-Xia Hu^a, Qing-Shan Zhao^a, Fu-Ling Tang^{a,b*}

^a School of Materials Science and Engineering, Lanzhou University of Technology, Lanzhou 730050, China

^b State Key Laboratory of Advanced Processing and Recycling of Non-ferrous Metals, Lanzhou University of Technology, Lanzhou 730050, China

* Corresponding author.

E-mail address: tfl@lut.edu.cn (F. L. Tang)

Li ₂ FeS ₂	a (Å)	b	С	α (°)	ß	γ
Ibam	5.87	10.03	5.08	90.00	90.00	90.00
Immm	3.39	4.66	11.04	90.00	90.00	90.00
<i>P</i> 3 <i>m</i> 1	3.89	3.89	6.21	90.00	90.00	120.00
$P\overline{3}m1$	3.81	3.81	6.08	90.00	90.00	120.00
$P2_{1}/c$	6.71	6.38	8.12	90.00	92.83	90.00
P42mcn	5.77	5.77	5.69	90.00	90.00	90.00

Table S1. The lattice constant of Li_2FeS_2 with different space groups

Table S2. Chemical potentials for elements in $Li_2Fe_{0.875}M_{0.125}S_2$

$\mu_{ m M}$	Ti	V	Fe	Li	S
DFT	-7.834	-8.988	-8.309	-1.875	-4.113
Ref	-7.770	-8.941	-8.308	-1.897	-4.114

Table S3. Formation energy of Fe in Li_2FeS_2 substituted by M atom (M = Ti, V)

Li ₂ FeS ₂	Ibam	Immm	P3m1	$P\overline{3}m1$	$P2_{1}/c$	P42mcn
$Li_2Fe_{0.875}Ti_{0.125}S_2$	-2.57	-1.69	-0.76	-2.77	-2.34	-0.10
$Li_2Fe_{0.875}V_{0.125}S_2$	-3.49	-0.64	-0.81	-0.80	-1.23	-0.57

Figure S1. The structures of Li₂FeS₂ with different *Ibam* (a), *Immm* (b), P3m1 (c), $P\bar{3}m1$ (d), $P2_1/c$ (e) and P42mcm (f) space groups. (yellow, pink and green are for S, Fe and Li atoms.)



Figure S2. The formation energy of mixing enthalpy of $\text{Li}_{1-x}(\text{Fe}_{0.875}\text{Ti}_{0.125})_{1/2}\text{S}$ with different *Ibam* (a), *Immm* (c), P3m1 (e), $P\bar{3}m1$ (g), $P2_1/c$ (i) and P42mcm (m) is a function of Li concentration; the voltage of $\text{Li}_{1-x}(\text{Fe}_{0.875}\text{Ti}_{0.125})_{1/2}\text{S}$ with different *Ibam* (b), *Immm* (d), P3m1 (f), $P\bar{3}m1$ (h), $P2_1/c$ (j) and P42mcm (n) increases with the increase of theoretical capacity.





Figure S3. The formation energy of mixing enthalpy of $\text{Li}_{1-x}(\text{Fe}_{0.875}\text{V}_{0.125})_{1/2}\text{S}$ with different *Ibam* (a), *Immm* (c), P3m1 (e), $P\bar{3}m1$ (g), $P2_1/c$ (i) and P42mcm (m) is a function of Li concentration; the voltage of $\text{Li}_{1-x}(\text{Fe}_{0.875}\text{V}_{0.125})_{1/2}\text{S}$ with different *Ibam* (b), *Immm* (d), P3m1 (f), $P\bar{3}m1$ (h), $P2_1/c$ (j) and P42mcm (n) increases with the increase of theoretical capacity.





Space group	a (Å)	b	С	α (°)	β	γ
	7.5359	7.5359	12.0754	90	90	120
<i>P</i> 3 <i>m</i> 1			Ato	om position		
	Fe1	Fe1 0.50000000		0.500000000	0.000000000	
	Fe2	0.5000	00000	0.000000000	0.0000	00000
	Fe3	0.0000	00000	0.500000000	0.0000	00000
	Fe4	0.0000	00000	0.000000000	0.0000	00000
	Fe5	0.5000	00000	0.500000000	0.5000	00000
	Fe6	0.5000	00000	0.000000000	0.5000	00000
	Fe7	0.0000	00000	0.500000000	0.5000	00000
	Li1	0.8340	79981	0.668160021	0.1783	60000
	Li2	0.1659	20019	0.331839979	0.8216	40015
	Li3	0.3318	39979	0.165919960	0.1783	60000
	Li4	0.6681	60021	0.834080040	0.8216	40015
	Li5	0.8340	80040	0.165920019	0.1783	60000
	Li6	0.1659	19960	0.834079981	0.8216	40015
	Li7	0.6606	10020	0.830299973	0.3153	40012
	Li8	0.3393	89980	0.169700027	0.6846	59958
	Li9	0.1697	00027	0.830310047	0.31534	40012
	Li10	0.8302	299973	0.169689953	0.6846	59958
	Li11	0.1696	589953	0.339389980	0.3153	40012
	Li12	0.8303	310047	0.660610020	0.6846	59958
	Li13	0.3333	333373	0.666666687	0.1793	80000
$L_{12}Fe_{0.875}T_{10.125}S_2$	Li14	0.6666	666627	0.333333313	0.820620000	
	Li15	0.6666	666687	0.333333373	0.321550012	
	Li16	0.3333	333313	0.666666627	0.678449988	
	S1	0.832710028		0.665409982	0.887520015	
	S2	0.167289972		0.334590018	0.112479985	
	S 3	0.3345	90018	0.167300045	0.887520015	
	S 4	0.6654	09982	0.832699955	0.112479985	
	S 5	0.8326	99955	0.167289972	0.887520015	
	S6	0.1673	00045	0.832710028	0.1124	79985
	S7	0.8370	19980	0.674040020	0.386209995	
	S8	0.1629	80020	0.325959980	0.613790035	
	S 9	0.3259	59980	0.162979960	0.386209995	
	S10	0.6740	40020	0.837020040	0.6137	90035
	S11	0.8370	20040	0 162980020	0.3862	09995
	S12	0.1629	79960	0.837019980	0.6137	90035
	S12 S13	0.3333	33373	0.666666687	0 8849	09987
	S14	0.6666	66627	0 333333313	0.1150	90013
	S15	0.3333	33373	0.666666687	0 3924	39991
	S16	0.5555	66627	0.333333313	0.6075	60039
	Ti1	0.0000	000027	0.000000000	0.5000	00000
	7 49471	7 48774	5 12 1707	8 90 00340	90	120 0307/
Cm	/.┭ノ┭/1	1.10//2	Δt	$\frac{1}{2}$ $\frac{1}$	70	120.03074
		Fe1	0.006220	0.006531	0.000752	
		Fel	0.000220	0.000331	0.000733	
Li (Ee Ti) C		Fe2	0.777044	0.477200	0.990220	
$L_{1.875}(1 - 0.875 + 10.125)_{1/2}$		Fe4	0.300311	0.000331	0.000733	
		Fe4	0.475201	0.490321	0.001113	
		res	0.730234	0.900219	0.303238	

Table S4. Lattice parameters and atomic position for $Li_{1-x}(Fe_{0.875}Ti_{0.125})_{1/2}S$

		Fe6	0.999025	0.498051	0.497200	
		Fe7	0.521985	0.960219	0.503258	
		Li1	0.823272	0.676340	0.816541	
		Li2	0.817713	0.135426	0.808488	
		Li3	0.666772	0.327516	0.680880	
		Li4	0.353068	0.676340	0.816541	
		Li5	0.173978	0.847956	0.673712	
		Li6	0.336503	0.173008	0.815816	
		Li7	0.160743	0.327516	0.680880	
		Li8	0.845004	0.666962	0 312759	
		Li9	0.666385	0.832767	0 170354	
		Lil	0.838691	0 177383	0.321953	
		Lill	0.665229	0 334926	0.185383	
		Li11	0.321960	0.554720	0.312759	
		Li12	0.321700	0.835/136	0.182589	
			0.331281	0.162562	0.102507	
		L114 1 ;15	0.351201	0.334026	0.185383	
		S1	0.109098	0.554920	0.105505	
		51 62	0.03/373	0.004394	0.112334	
		52 52	0.003921	0.031041	0.903290	
		33 84	0.63/191	0.174562	0.110007	
		54	0.0/1/9/	0.552512	0.004900	
		33	0.326800	0.004394	0.112354	
		56	0.16/448	0.834896	0.883324	
		S/	0.333502	0.16/004	0.11464/	
		<u>S8</u>	0.160/14	0.332512	0.884906	
		S9	0.825891	0.667832	0.617553	
		S10	0.662228	0.824455	0.373286	
		S11	0.831237	0.162474	0.607702	
		S12	0.666455	0.342169	0.390339	
		S13	0.341943	0.667832	0.617553	
		S14	0.166273	0.832544	0.396835	
		S15	0.342485	0.184970	0.607964	
		S16	0.175714	0.342169	0.390339	
		Ti1	0.518415	0.536830	0.505578	
Cm	6.64332	6.72032	2 12.56177	90	78.21938	120.38419
			Ator	n position		
		Fe1	0.933420	0.972605	0.009684	
		Fe2	0.933420	0.460815	0.009684	
		Fe3	0.478210	0.989105	0.002973	
		Fe4	0.484065	0.492032	0.001250	
		Fe5	0.999532	0.987758	0.497122	
		Fe6	0.999532	0.511773	0.497122	
		Fe7	0.550615	0.025307	0.494937	
		Li1	0.103324	0.801661	0.734841	
$Li_{0.75}(Fe_{0.875}Ti_{0.125})_{1/2}S$		Li2	0.102895	0.301447	0.729379	
0.75 0.075 0.1257 172		Li3	0.017014	0.258507	0.244112	
		Li4	0.466682	0.725909	0.256293	
		Li5	0.012186	0.756092	0.244453	
		Li6	0.466681	0.240771	0.256293	
		<u>_1</u>	0.735724	0.617861	0.115573	
		S2	0.681681	0.843651	0.898259	
		<u>S</u> 3	0.724080	0.112041	0.119190	
		S4	0.681680	0.338029	0.898259	
		ти	0.001000	5.550047	0.070407	

		S5	0.244583	0.620014	0.102603	
		S 6	0.162257	0.831128	0.917398	
		S 7	0.244582	0.124567	0.102603	
		S 8	0.172243	0.336121	0.912581	
		S 9	0.807151	0.653575	0.619333	
		S10	0.762204	0.890456	0.375879	
		S11	0.815875	0.157938	0.598778	
		S12	0.762203	0.371747	0.375879	
		S13	0.298212	0.660719	0.594793	
		S14	0.250610	0.875304	0.396509	
		S15	0.298211	0.137492	0.594793	
		S16	0.226145	0.363072	0.399310	
		Ti1	0.572867	0.536434	0.494838	
0	6.45880	6.3124	1 12.90679	90.33601	90	120.77020
Cm			Ato	om position		
		Fe1	0.975268	0.952891	0.995568	
		Fe2	0.023410	0.546820	0.996876	
		Fe3	0.477623	0.952891	0.995568	
		Fe4	0.522435	0.544871	0.998691	
		Fe5	0.987341	0.015468	0.509253	
		Fe6	0.986926	0.473853	0.507206	
		Fe7	0.528127	0.015468	0.509253	
		Li1	0.313123	0.126245	0.269154	
		S 1	0.833707	0.664660	0.109255	
		S2	0.666913	0.833825	0.884570	
		S 3	0.833496	0.166992	0.081029	
		S 4	0.667709	0.335125	0.910557	
Li _{0 125} (Fe _{0 875} Ti _{0 125}) _{1/2} S		S5	0.330954	0.664660	0.109255	
		S6	0.168634	0.837266	0.883998	
		S 7	0.332286	0.164572	0.081639	
		S 8	0.167415	0.335125	0.910557	
		S 9	0.850868	0.674518	0.612766	
		S10	0.677411	0.854820	0.393412	
		S11	0.834969	0.169940	0.621511	
		S12	0.674662	0.327274	0.392700	
		S13	0.323652	0.674518	0.612766	
		S14	0.170058	0.840115	0.418967	
		S15	0.327251	0.154503	0.612994	
		S16	0.152612	0.327274	0.392700	
		Ti1	0.503453	0.506906	0.505098	
		111	0.000-00	0.200700	0.000000	

Space group	а	a b		α	β	γ	
	7.5258	7.5258	12.0294	90	90	120	
<i>P</i> 3 <i>m</i> 1		Atom position					
	Fe1	0.50000000	0.500	000000	0.0000	000000	
	Fe2	0.50000000	0.000	000000	0.0000	000000	
	Fe3	0.000000000	0.500	000000	0.0000	000000	
	Fe4	0.50000000	0.500	000000	0.5000	000000	
	Fe5	0.50000000	0.000	000000	0.5000	000000	
	Fe6	0.000000000	0.500	000000	0.5000	000000	
	Fe7	0.000000000	0.000	000000	0.5000	000000	
	Li1	0.831239998	0.662	479997	0.1840	000000	
	Li2	0.168760002	0.337	520003	0.8159	999985	
	Li3	0.337520003	0.168	760002	0.1840	000000	
	Li4	0.662479997	0.831	239998	0.8159	999985	
	Li5	0.831239998	0.168	760002	0.1840	000000	
	Li6	0 168760002	0.831	239998	0.8159	999985	
	Li7	0.668760002	0.834	380031	0 320	719987	
	Li8	0 331239998	0.165	619999	0.6792	280043	
	Li9	0.165619999	0.834	380031	0.3207	719987	
	Lil	0.834380031	0.051	619999	0.679	280043	
	Lill	0.165619999	0.105	239998	0.320	719987	
	Li12	0.834380031	0.551	760002	0.520	280043	
	Li12	0.333333377	0.000	666687	0.075	7600043	
$Li_2Fe_{0.875}V_{0.125}S_2$	L113	0.555555575	0.000	333313	0.175	739969	
	L114 1 ;15	0.666666687	0.333	3333/3	0.824	237707	
	LIIJ	0.333333313	0.555	666677	0.520	750070	
	\$1	0.3333333313	0.000	8400027	0.079	270001	
	51	0.030419974	0.070	150007	0.0000	20000	
	52 52	0.101380020	0.525	570067	0.1111	277777	
	55 54	0.323139993	0.101.	420022	0.0000	20000	
	54 85	0.070840007	0.030	+20033 580026	0.1111	27777	
	55 56	0.030420033	0.101.	410074	0.0000	20000	
	50	0.1013/990/	0.656	0.653/0060		29999	
	57	0.852009975	0.005	549900 c=0010	0.50/0	20002	
	38	0.16/33002/	0.5540	210002	0.0123	200028	
	59	0.334650010	0.167.	519983	0.38/0	20002	
	510	0.005349900	0.832	0.852079987		580028	
	511	0.8320/998/		330027	0.38/0	520002 290029	
	512	0.10/319983	0.832	0099/3	0.012.	380028 240001	
	513	0.333333373		00008/	0.891	240001	
	S14	0.666666627	0.333	333313	0.108	/59999	
	S15	0.333333373	0.666	66668/	0.3834	159985	
	S16	0.666666627	0.333	333313	0.616	540015	
	Til	0.000000000	0.000		0.0000		
<i>C</i> 2	6.70890) 6.59793	11.77052	90.5514	0 90 12	20.55796	
			Atom pos	sition	0.000005		
		Fe1 0.013	3527 1.00	00000	0.000000		
		Fe2 0.477	/290 0.40	53676	0.006250		
Lio 75 (Feo 875 Vo 125) 1/2S		Fe3 0.013	615 0.5.	36324	0.993750		
0.73(0.875 · 0.125/1/29		Fe4 0.995	5452 0.00	00000	0.500000		
		Fe5 0.468	8747 0.4 [°]	78895	0.501288		
		Fe6 0.989	9852 0.52	21105	0.498712		

Table S5. Lattice parameters and atomic position for $Li_{1-x}(Fe_{0.875}V_{0.125})_{1/2}S$

		Fe7	0.463237	0.000000	0.500000
		Li1	0.503136	0.004249	0.747193
		Li2	0.961172	0.005961	0.748951
		Li3	0.498424	0.507848	0.755684
		Li4	0.498886	0.995752	0.252807
		Li5	0.990576	0.492152	0.244316
		Li6	0.955212	0.994039	0.251049
		S1	0 308904	0 165561	0 127959
		S2	0 143344	0.834440	0.872041
		S3	0.818444	0.662501	0.100870
		S4	0.667977	0 339879	0.892548
		S5	0.328098	0.660122	0.107452
		S6	0.155943	0.337499	0.899130
		S7	0.133743	0.177810	0.09/130
		57	0.629/12	0.822100	0.094701
		50	0.031902	0.822190	0.903299
		S9 S10	0.310360	0.102033	0.010901
		S10 S11	0.14/952	0.657540	0.363099
		511	0.803731	0.003923	0.002138
		512	0.055528	0.337056	0.396357
		513	0.316272	0.662945	0.603643
		S14	0.141826	0.3360/5	0.397842
		\$15	0.815900	0.173118	0.594296
		S16	0.642782	0.826882	0.405704
		V1	0.435289	0.000000	0.000000
C^{2}	6.60466	6.42	2753 12.2	2412 90.36	223 90 120.91589
			At	om position	
		Fe1	0.018225	1.000000	0.000000
		Fe2	0.468200	0.451352	0.007567
		Fe3	0.016848	0.548648	0.992433
		Fe4	0.035706	1.000000	0.500000
		Fe5	0.485387	0.456577	0.505667
		Fe6	0.028810	0.543423	0.494333
		Fe7	0.489539	0.000000	0.500000
		Li1	0.990685	0.483494	0.743413
		Li2	0.519300	0.526618	0.758429
		Li3	0.992682	0.473382	0.241571
		Li4	0.507191	0.516506	0.256587
		S 1	0.307986	0.148640	0.128285
		S2	0.159347	0.851361	0.871715
$Li_{0.5}(Fe_{0.875}V_{0.125})_{1/2}S$		S 3	0.815669	0.659189	0.100574
0.5 (0.675 0.1257172		S 4	0.670252	0.342891	0.893170
		S 5	0 327362	0.657110	0 106830
		S6	0.156480	0 340811	0.899426
		S7	0.836514	0 182837	0.088357
		58	0.653676	0.817163	0.911643
		50	0.338018	0 1/0330	0.610331
		SJ S10	0.188670	0.147559	0.017551
		S10 S11	0.1000/9	0.650002	0.50009
		S11 S12	0.651029	0.00124/	0.001051
		S12 S12	0.003207	0.330270	0.373004
		010 014	0.343009	0.001/23	0.004190
		S14	0.109/82	0.338/33	0.578747
		212	0.851194	0.1/8/45	0.588058
		010	0.070440	0.001055	0 411040

		V1	0.4438	810	0.00	0000	1.000	0000	
$\mathcal{C}\mathcal{D}$	6.49310	6.28	8227 1	13.037	767	89.031	1 90) 12	1.11646
C2				Ator	m pos	ition			
		Fe1	0.0319	964	1.00	0000	0.000	0000	
		Fe2	0.4643	327	0.43	8343	0.008	3413	
		Fe3	0.0259	984	0.56	1657	0.99	1587	
		Fe4	0.0314	452	1.00	0000	0.500	0000	
		Fe5	0.4743	333	0.44	5079	0.502	2826	
		Fe6	0.0292	254	0.55	4921	0.497	7174	
		Fe7	0.4720)53	0.00	0000	0.500	0000	
		Li1	0.6974	132	0.85	9633	0.734	688	
		Li2	0.8377	799	0.14	0367	0.265	5312	
		S 1	0.3111	77	0.14	0185	0.124	265	
		S 2	0.1709	92	0.85	9816	0.875	5735	
		S 3	0.8137	27	0.64	9949	0.094	979	
$\mathbf{L}_{\mathbf{i}} = (\mathbf{F}_{\mathbf{e}} = \mathbf{V}_{\mathbf{e}}) = \mathbf{S}$		S 4	0.6771	48	0.35	2051	0.896	5957	
$L_{10.25}(1 \times 0.875 \times 0.125)1/2S$		S 5	0.3250)97	0.64	7950	0.103	3044	
		S 6	0.1637	78	0.35	0051	0.905	5021	
		S 7	0.8434	02	0.18	1553	0.083	3070	
		S 8	0.6618	849	0.81	8447	0.916	5930	
		S9	0.3236	680	0.14	3879	0.615	5007	
		S10	0.1798	302	0.85	6122	0.384	1993	
		S11	0.8148	372	0.65	2833	0.597	7355	
		S12	0.6882	232	0.34	6212	0.40	1902	
		S13	0.3420)19	0.65	3789	0.598	8098	
		S14	0.1620)39	0.34	7167	0.402	2645	
		S15	0.8372	294	0.17	1505	0.584	4562	
		S16	0.665	789	0.82	8495	0.415	5438	
		V1	0.4511	188	0.00	0000	1.000	0000	



Figure S4. Partial density of states for $Li_2Fe_{0.875}Ti_{0.125}S_2$ (a), $Li_{1.875}Fe_{0.875}Ti_{0.125}S_2$ (b), $Li_{0.75}Fe_{0.875}Ti_{0.125}S_2$ (c), $Li_{0.125}Fe_{0.875}Ti_{0.125}S_2$ (d) and $Fe_{0.875}Ti_{0.125}S_2$ (e); charge distribution of Fe and Ti atoms (f).



Figure S5. Partial density of states for $Li_2Fe_{0.875}V_{0.125}S_2$ (a), $Li_{0.75}Fe_{0.875}V_{0.125}S_2$ (b), $Li_{0.5}Fe_{0.875}V_{0.125}S_2$ (c), $Li_{0.25}Fe_{0.875}V_{0.125}S_2$ (d), and $Fe_{0.875}V_{0.125}S_2$ (e); charge distribution of Fe and V atoms (f).







Figure S7. The structures for $Li_2FeS_2(100)$ (a), $Li_2FeS_2(110)$ (b), $Li_2FeS_2(001)$ (c), $Li_2FeS_2(002)$ (d), $Li_2Fe_{0.875}Ti_{0.125}S_2(002)$ (e) and $Li_2Fe_{0.875}V_{0.125}S_2(002)$ (f)



Figure S8. The structures for Li_2S_2 - $Li_2FeS_2(002)$ (a), Li_3S_2 - $Li_2FeS_2(002)$ (b), LiS- $Li_2FeS_2(002)$ (c) and Li_2S - $Li_2FeS_2(002)$ (d)



Figure S9. The structures for Li_2S_2 - $Li_2Fe_{0.875}Ti_{0.125}S_2(002)$ (a), Li_3S_2 - $Li_2Fe_{0.875}Ti_{0.125}S_2(002)$ (b), $LiS-Li_2Fe_{0.875}Ti_{0.125}S_2(002)$ (c) and Li_2S - $Li_2Fe_{0.875}Ti_{0.125}S_2(002)$ (d)



Figure S10. The structures for Li_2S_2 - $Li_2Fe_{0.875}V_{0.125}S_2(002)$ (a), Li_3S_2 - $Li_2Fe_{0.875}V_{0.125}S_2(002)$ (b), $LiS-Li_2Fe_{0.875}V_{0.125}S_2(002)$ (c) and Li_2S - $Li_2Fe_{0.875}V_{0.125}S_2(002)$ (d)



Figure S11. Li_2S decomposition paths on $Li_2FeS_2(002)$ (a), $Li_2Fe_{0.875}Ti_{0.125}S_2(002)$ (b) and $Li_2Fe_{0.875}V_{0.125}S_2(002)$.