Supporting Information for

## Emergent ultrasmall multiferroics in paraelectric perovskite oxide by hole

## polarons

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**Fig. S1.** (a)Crystal structure of orthorhombic CaTiO<sub>3</sub>. The red dot box indicates the primitive cell. (b-1)(b-2)AFD displacement exist in this structure. The arrows indicate AFD displacement around each axis.



**Fig. S2.** (a)Schematic diagram of perovskite ABO<sub>3</sub>, which has uniaxial AFD displacement. Blue arrows indicate the displacement of O atoms due to AFD displacement. (b)Schematic diagram of ABO<sub>3</sub> with hole polaron(yellow area). The red arrows indicate the displacement of A1 and O atoms due to the localization of hole. The pink and blue arrows indicate the direction of polarization and the displacement of O atom due to AFD displacement, respectively.



**Fig. S3**. The evolution of the squared wave function of the hole polaron in CaTiO<sub>3</sub> along the migration pathway. The yellow area represents the iso-surface of charge densities of 0.02 Å<sup>-3</sup>. White arrows indicate the small displacement of atoms around the hole polaron.



Fig. S4. (a)O2 atoms in primitive cell (black dot box) of orthorhombic  $CaTiO_3$ . (b)The direction of polarization (white arrows) and magnetic moment (red arrows) in the case of the hole polaron at each O2 atom.

	HSE06	Exp
$E_{\rm gap}({\rm eV})$	3.86	$3.8 \sim 4.4$
a (Å)	5.36	5.38
<i>b</i> (Å)	5.43	5.44
<i>c</i> (Å)	7.61	7.64

**Table S1**. Band gap  $E_{gap}$ , lattice parameters *a*, *b*, and *c* of orthorhombic CaTiO<sub>3</sub> calculated using HSE06. The experimental values [1,2] are also shown for comparison.

**Table S2**. Activation energy *Ea* and the hole polaron mobility  $\mu$  for polaronic migration in CaTiO<sub>3</sub> from initial to the final structure.

Number of O2	O2-b,d	O2-c,e		
$E_{\rm a}({\rm meV})$	148.0	167.0		
$\mu(\times 10^{-3})(cm^2/Vs)$	2.52	1.19		
Number of O1	O1-a	O1-b	O1-c	O1-d
$E_{\rm a}({\rm meV})$	151.6	250.8	265.5	136.9
$\mu(\times 10^{-3})(cm^2/Vs)$	2.19	4.73×10 <sup>-2</sup>	2.69×10 <sup>-2</sup>	3.79

## Reference

[1] A. R. Chakhmouradian, R. H. Mitchell, J. Solid State Chem., 1998, 138, 272277.
[2] A. Krause, W. M. Weber, D. Pohl, B. Rellinghaus, A. Kersch, T. Mikolajick, J. Phys. D: Appl. Phys., 2015, 48, 415304.