Supporting information

Suitable Energy Avenue For Dimension-Matched Cascade Charge Transfer Mechanism in g-C₃N₄/TS-1 Heterostructure Co-Doped With Au-TiO₂ For Artificial Photosynthetic Green Fuel Production

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Figure S1. SEM image of TS nanosheets and (8CN/0.7M/TS).



Figure S2. EDX analysis of 1Au-4T/8CN/0.7hTS nanocomposites.



Figure S3. XRD patterns of TS and 8CN/TS a), 8CN/TS, 8CN/0.3hTS, 8CN/0.5hTS, 8CN/0.7hTS and 8CN/1hTS b). and c), XRD patterns of 8CN/0.7hTS, 2T/8CN/0.7hTS, 4T/8CN/0.7hTS, 6T/8CN/0.7hTS, 8T/8CN/0.7hTS and 1Au-4T/8CN/0.7hTS.



Figure S4. UV-vis diffuse reflectance spectra a) and b), and energy band gaps of different sulphated Mn modified samples c). The labels 0 #, 0.3 #, 0.5 #, 0.7 # and 1 # denote samples in which the wt.% sulfur (S) was 0%, 1%, 5%, 7%, and 10%, respectively.



Figure S5. N_2 adsorption/desorption isotherm curves and pore diameter of TS, 0.7hTS, 8CN/0.7hTS, and 1Au-4T/8CN/0.7hTS.





Figure S6. FTIR spectra of different samples.

Figure S7. XPS survey spectra a), XPS survey spectra of S2p b), Si2p c), and O1s d), of TS, 0.7hTS, 8CN/0.7hTS, 4T/8CN/0.7hTS and 1Au-4T/8CN/0.7hTS samples.



Figure S8. SS-SPS spectra of TS, 0.7hTS and 8CN/0.7hTS a), 8CN/0.3hTS, 8CN/0.5hTS, 8CN/0.7hTS, 8CN/1hTS b), and 8CN/0.7hTS, 2T/8CN/0.5hTS, 4T/8CN/0.5hTS, 6T/8CN/0.5hTS, 8T/8CN/0.7hTS and 1Au-4T/8CN/0.7hTS c).



Figure S9. Photoluminescence spectra of TS, 0.7hTS and 8CN/0.7hTS a), 0.7hTS, 8CN/0.3hTS, 8CN/0.5hTS, 8CN/0.7hTS and 8CN/1hTS b) and 2T/8CN/0.7hTS, 4T/8CN/0.7hTS, 6T/8CN/0.7hTS, 8T/8CN/0.7hTS) and 1Au-4T/8CN/0.7hTS c).



Figure S10. Fluorescence spectra related to the produced hydroxyl radical amounts of TS, 0.7hTS and 8CN/0.7hTS a), 0.7hTS, 8CN/0.3hTS, 8CN/0.5hTS, 8CN/0.7hTS and 8CN/1hTS b) and 2T/8CN/0.7hTS, 4T/8CN/0.7hTS, 6T/8CN/0.7hTS, 8T/8CN/0.7hTS and 1Au-4T/8CN/0.7hTS c).



Figure S11. Visible-light photocatalytic activities for CO₂ conversion of TS, 0.3hTS, 0.5hTS, 0.7hTS and 1hTS a) and 2CN/0.7hTS, 4CN/0.7hTS, 6CN/0.7hTS, 8CN/0.7hTS and 10CN/0.7hTS b), and 2T/8CN/0.7hTS, 4T/8CN/0.7MTS), 6T/8CN/0.7hTS, 8T(8CN/0.7hTS) and 1Au-4T/8CN/0.7hTS c).



Figure S12. The GC-Mass analysis of photocatalytic reduction products of CO_2 on 1Au-4T/8CN/0.7hTS after irradiation for 6h without methanol a) and with methanol b).



Figure S13. Three consecutive runs of CO_2 reduction by 1Au-4T/8CN/0.7hTS under visible-light irradiation.



Figure S14. XRD patterns of 0.7hTS, 8CN/6T/0.7hTS, 4T/8CN/0.7hTS and 1Au-4T/8CN/0.7hTS after long term stability test.



Figure S15. XPS survey spectra of Ti $2p_{3/2}$ a), XPS survey spectra of O1s b), C 1s c), and N 1s d), S 2p e) and f), Si 2p of TS, 0.7hTS, 8CN/0.7hTS, and 1Au-4T/8CN/0.7hTS samples after long term stability test.



Figure S16. Electrochemical reduction curves in the air-bubbled system of TS, 0.7hTS, 0.7hTS, 8CN/6T/0.7hTS, 4T/8CN/0.7hTS and 1Au-4T/8CN/0.7hTS. Electrochemical performance was measured in a 0.5 M Na₂SO₄ solution, and Hg/Hg₂Cl₂ (saturated KCl) electrode was used as the reference electrode.



Figure S17. GC-Mass Spectrometry analyses of intermediate products after visiblelight photocatalytic conversion of CO₂ over 1Au-4T/8CN/0.7hTS photocatalyst for 8h. The identified intermediate products, including CD (m/z=14), CD₂ (m/z=16), CD₃ (m/z=18), and CD₄ (m/z=20) are labeled in the figure.

Table S1. C, N, Ti, S, Au contents (mol%) in TS, 0.7hTS, 8CN/0.7hTS, and 1Au-4T/8CN/0.7hTS according to elemental analysis.

Samples	Ν	С	Ti	S	Au
TS	0	0	13.45	1.8	0
0.7hTS	0	0	13.12	1.82	0
8CN/0.7hTS	23.14	25.34	12.36	1.76	0
4T/8CN/0.7hTS	22.65	26.06	11.16	1.8	0
1Au-4T/8CN/0.7hTS	23.12	25.55	13.31	1.7	0.45
Elemental ICP analysis after long term CO ₂ reduction reaction:					
TS	0	0	13.25	1.6	0
0.7hTS	0	0	13.12	1.72	0
8CN/0.7hTS	22.17	24.31	11.48	1.79	0
4T/8CN/0.7hTS	22.87	25.00	12.09	1.75	0
1Au-4T/8CN/0.7hTS	22.79	25.35	13.21	1.66	0.45