

# **Fabrication of High-performance CeO<sub>2</sub>-MnO<sub>x</sub>/TiO<sub>2</sub>/Ti Monolithic Catalysts for Low-temperature and Stable CO Oxidation**

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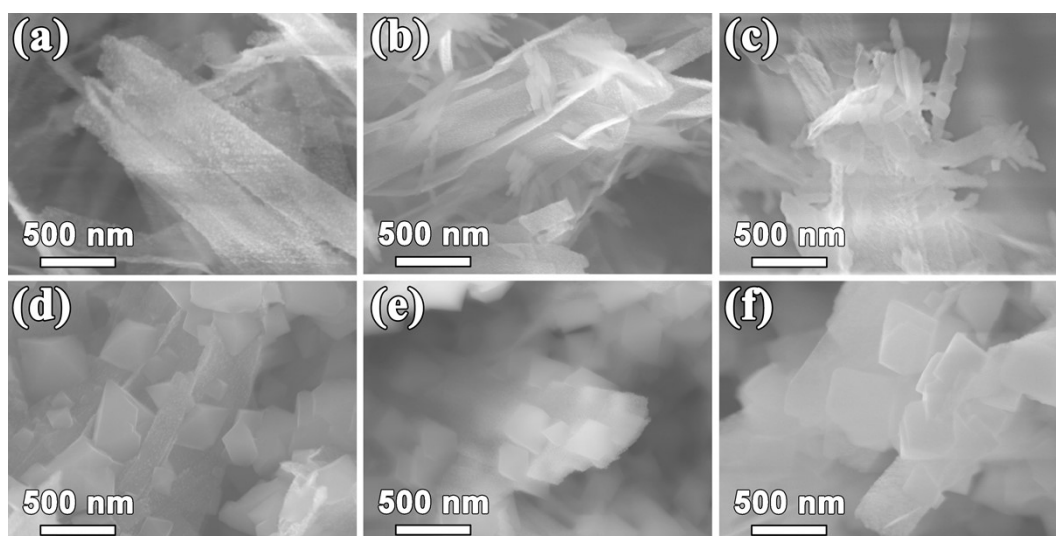
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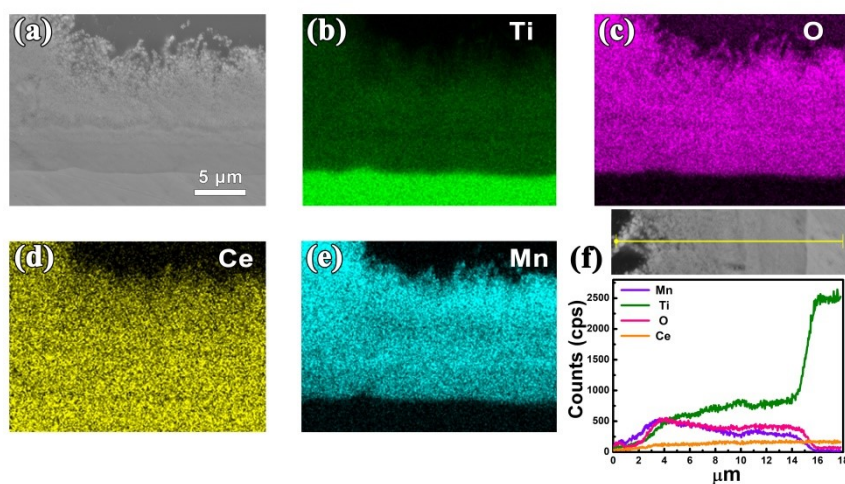
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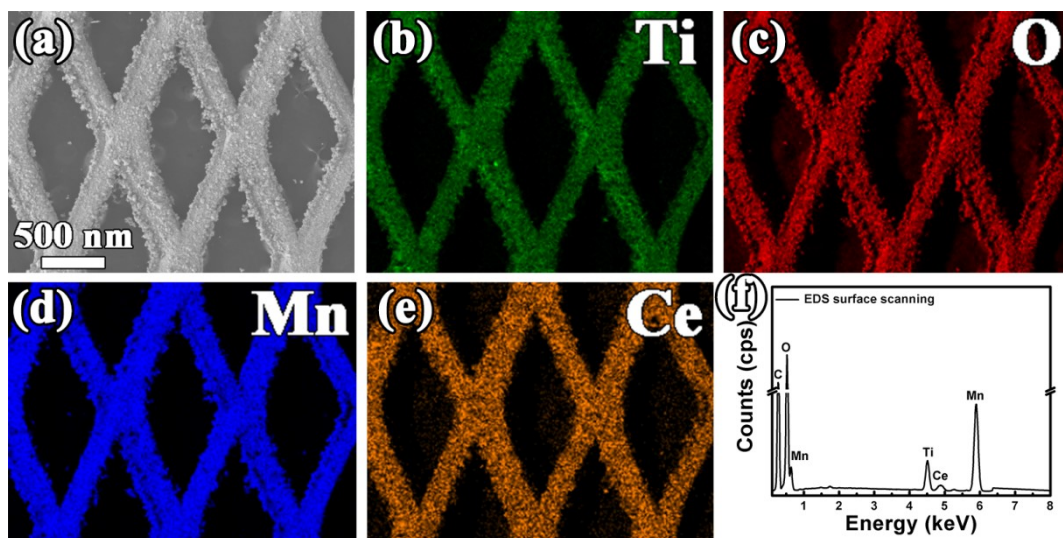
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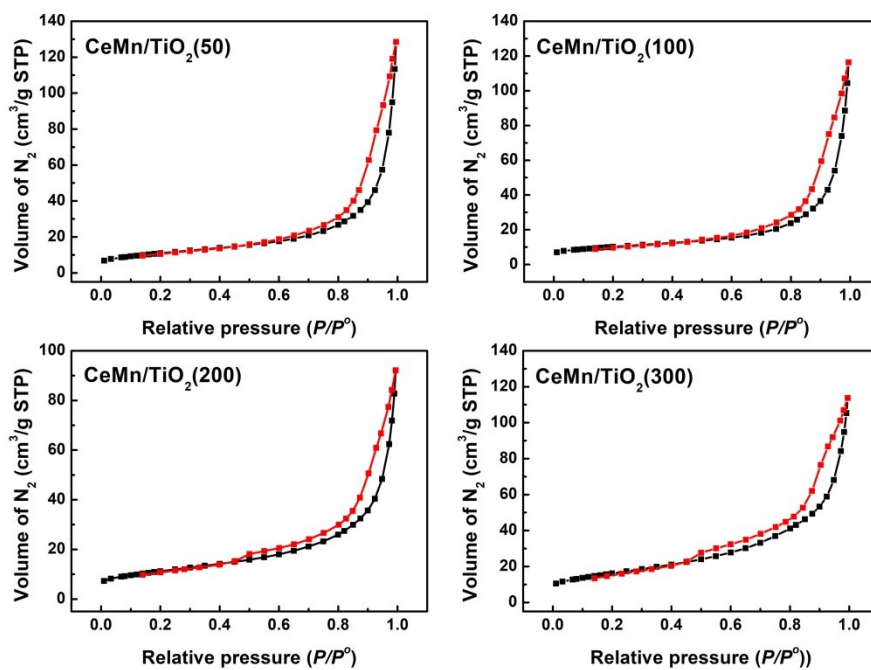
**Figure S1.** SEM images of the  $\text{CeO}_2\text{-MnO}_x/\text{TiO}_2/\text{Ti}$  catalysts prepared at different deposition time: (a) 3 h; (b) 6 h; (c) 9 h; (d) 12 h; (e) 18 h; (f) 24 h.



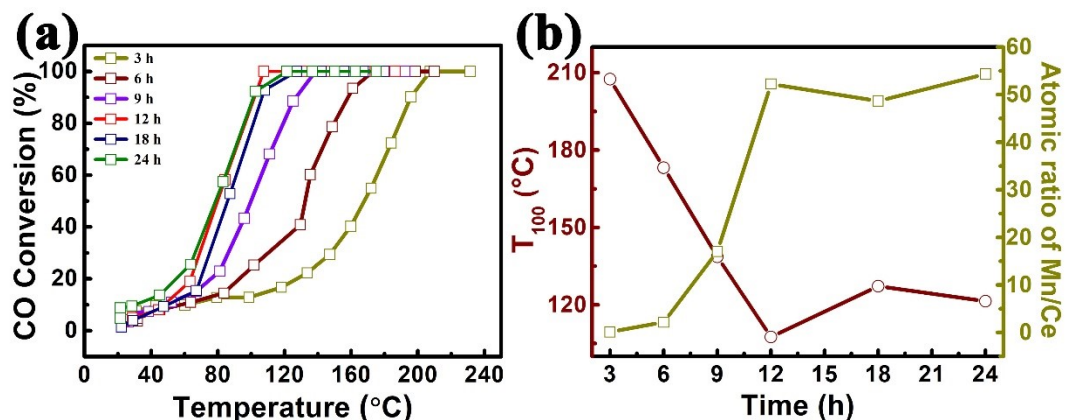
**Figure S2.** (a) SEM images of cross-section of  $\text{CeMn}/\text{TiO}_2(200)$  catalyst and corresponding EDS mapping results: (b-e) Ti, O, Ce and Mn, respectively; (f) elemental line scan profiles along the yellow line direction.



**Figure S3.** (a) SEM images of CeMn/TiO<sub>2</sub>(200) catalysts on Ti mesh and corresponding EDS mapping results: (b) Ti; (c) O; (d) Mn; (e) Ce; (f) EDS spectrum.



**Figure S4.** N<sub>2</sub> adsorption-desorption isotherms curves of CeO<sub>2</sub>-MnO<sub>x</sub>/TiO<sub>2</sub>/Ti monolithic catalysts with different urea concentrations.



**Figure S5.** (a) Catalytic CO oxidation light-off curves of as-prepared CeO<sub>2</sub>-MnO<sub>x</sub>/TiO<sub>2</sub>/Ti catalysts obtained by different deposition time; (b) Relationship of T<sub>100</sub> and the atomic ratio of Mn/Ce as a dependence with deposition time.

**Table S1.** Comparison of the activity for CO oxidation on different CeO<sub>2</sub> catalysts in literatures.

Catalyst	Reaction temperature	Reaction condition	Reference
CeO <sub>2</sub> /TiO <sub>2</sub> /Ti	T <sub>50</sub> = 350 °C	$p(\text{CO}) = 0.998\%$ , $p(\text{O}_2) = 20.04\%$ , in He.	[1]
CeO <sub>2</sub> microspheres	T <sub>90</sub> = 338 °C	1 vol.% CO, 4 vol.% O <sub>2</sub> , balanced He.	[2]
CeO <sub>2</sub>	T <sub>98</sub> = 370 °C	1 vol.% CO, 20 vol.% O <sub>2</sub> , 79 vol.% Ar.	[3]
CeO <sub>2</sub> NR <sub>s</sub>	T <sub>100</sub> > 390 °C	1 vol.% CO, 20 vol.% O <sub>2</sub> , 79 vol.% Ar.	[4]
Ce/TiO <sub>2</sub>	T <sub>100</sub> = 334 °C	$p(\text{CO}) = 0.998\%$ , $p(\text{O}_2) = 20.04\%$ , in He.	This work

**Table S2.** Mass loss of a series of CeO<sub>2</sub>-MnO<sub>x</sub>/TiO<sub>2</sub>/Ti catalysts after ultrasonic treatment for 10 min (wt%).

Catalyst	Initial mass (mg)	Final mass (mg)	Loss mass (mg)	Mass loss ratio
CeMn/TiO <sub>2</sub> (50)	411.8	405	6.8	1.65 wt%
CeMn/TiO <sub>2</sub> (100)	434	427	7	1.61 wt%
CeMn/TiO <sub>2</sub> (200)	383.4	377.1	6.3	1.64 wt%
CeMn/TiO <sub>2</sub> (300)	453.3	416	37.3	8.23 wt%

#### Notes and references

1. X. Liu, K. Wang, Y. Zhou, et al., *In-situ* fabrication of Ce-rich CeO<sub>2</sub> nanocatalyst for efficient CO oxidation, *J. Alloys Compd.*, 2019, **792**, 644-651.
2. D. Jampaiah, V. K. Velisoju, D. Devaiah, et al., Flower-like Mn<sub>3</sub>O<sub>4</sub>/CeO<sub>2</sub> microspheres as an efficient catalyst for diesel soot and CO oxidation: Synergistic effects for enhanced catalytic performance, *Appl. Surf. Sci.*, 2019, **473**, 209-221.
3. X. Zhang, F. Hou, Y. Yang, et al., A facile synthesis for cauliflower like CeO<sub>2</sub> catalysts from Ce-BTC precursor and their catalytic performance for CO oxidation, *Appl. Surf. Sci.*, 2017, **423**, 771-779.
4. Y. Deng, P. Tian, S. Liu, et al., Enhanced catalytic performance of atomically dispersed Pd on Pr-doped CeO<sub>2</sub> nanorod in CO oxidation, *J. Hazard. Mater.*, 2022, **426**, 127793.