

Supporting Information

Constructing crystalline-amorphous hydrated niobium pentoxide homojunction for superior photocatalytic CO₂ reduction into CH₄ with high selectivity

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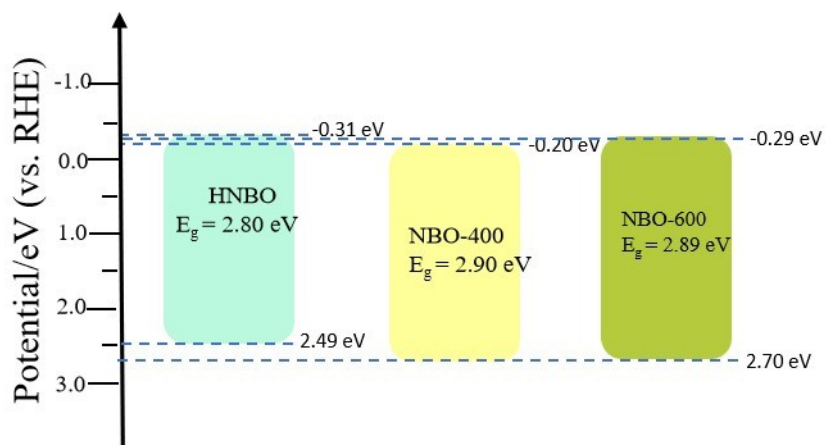


Fig. S1 Band structure of HNBO, NBO-400, and NBO-600.

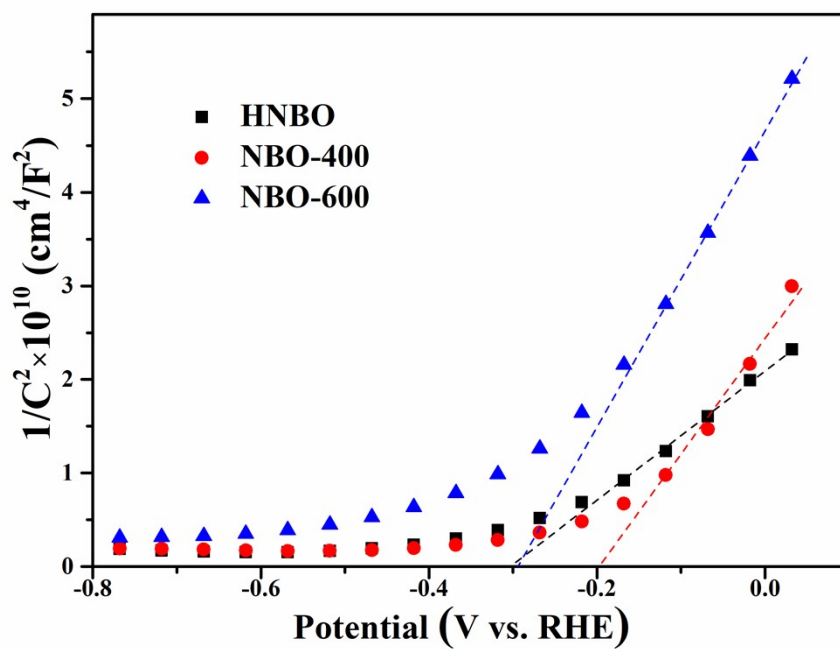


Fig. S2 Mott-Schottky plots of HNBO, NBO-400, and NBO-600

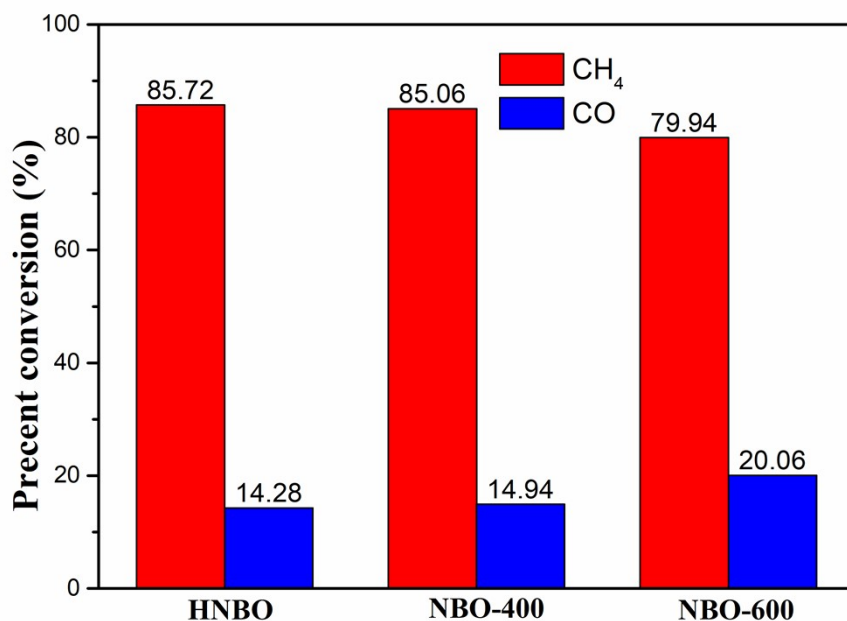


Fig. S2 CH₄ and CO selectivity comparison over HNBO, NBO-400, and NBO-600.

Table S1 Comparison of CO₂ photoreduction activity and selectivity for CH₄ over HNBO with other representative works reported in the literature.

Photocatalytic Materials	Light Source	Reaction condition	Production rate of CH ₄	Selectivity of CH ₄	Ref.
HBNO	300 W Xe lamp	Gas-solid, H ₂ O	31.39 μmol/g/h	85.7%	This work
Bi ₄ O ₅ Br ₂ /AgBr	300 W Xe lamp	Gas-solid, H ₂ O	1.8 μmol/g/h	21.9%	[1]
O-doped g-C ₃ N ₄ /N-doped Nb ₂ O ₅	300 W Xe lamp	Gas-solid, H ₂ O	68.11 μmol/g/h	21.1%	[2]
NiO _x /Nb ₂ O ₅	UV light	Gas-solid, H ₂ O	138.9 μmol/g/h	40.9%	[3]
g-C ₃ N ₄ /Pt/(TiO ₂ @carbon)	Simulated sunlight	Gas-solid, H ₂ O	6.56 μmol/g/h	81.7%	[4]
MoO ₂ /g-C ₃ N ₄	300 W Xe lamp	Gas-solid, H ₂ O	0.145 μmol/g/h	20.3%	[5]
CuO/Nb ₂ O ₅	UV light	Liquid-solid, H ₂ O	2 μmol/g/h	68.6%	[6]
Black Nb ₂ O _{5-x}	300 W Xe lamp	Liquid-solid, H ₂ O	19.5 μmol/g/h	64.8%	[7]

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