

**Direct Z-scheme WO₃/Covalent organic frameworks (COFs)
heterostructure for enhanced photocatalytic hydrogen peroxide
production in water**

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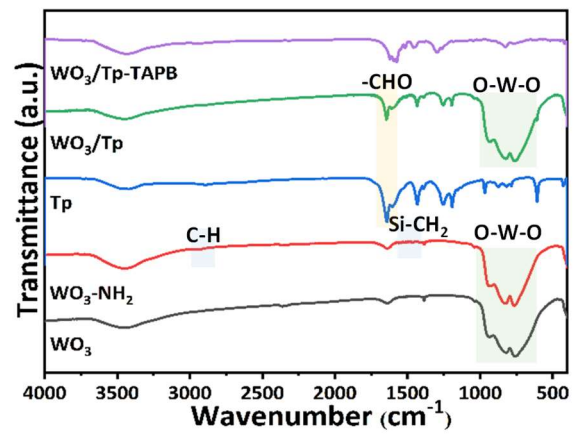


Fig. S1. FT-IR survey spectra of WO_3 , $\text{WO}_3\text{-NH}_2$, Tp, WO_3/Tp and $\text{WO}_3/\text{Tp-TAPB}$

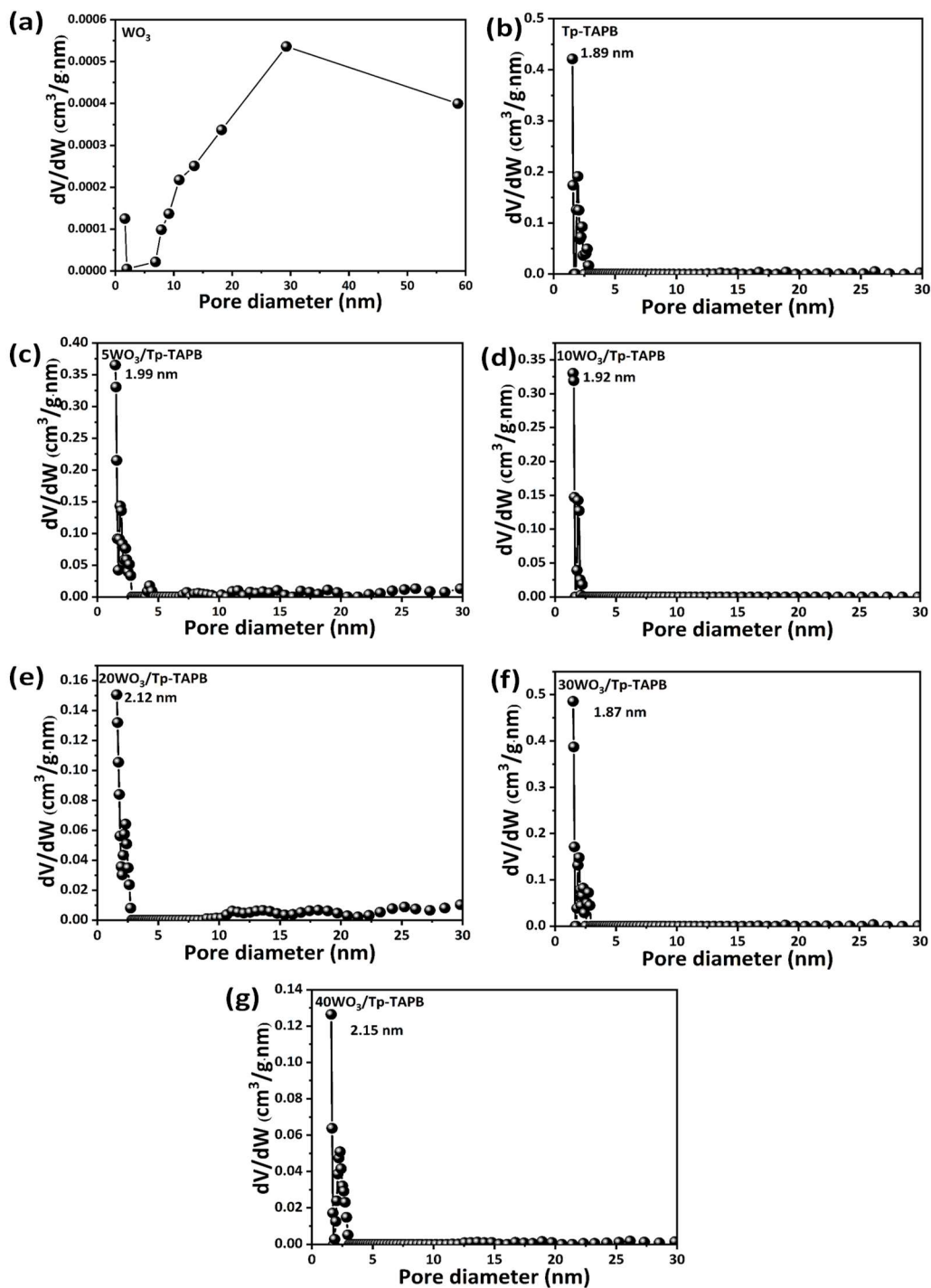


Fig. S2. (a) Pore size distributions estimated by BJH of WO_3 ; (b) Pore size distributions estimated by NLDFT of Tp-TAPB; (c) Pore size distributions estimated by NLDFT of $5\text{WO}_3/\text{Tp-TAPB}$; (d) Pore size distributions estimated by NLDFT of $10\text{WO}_3/\text{Tp-TAPB}$; (e) Pore size distributions estimated by NLDFT of $20\text{WO}_3/\text{Tp-TAPB}$; (f) Pore size distributions estimated by NLDFT of $30\text{WO}_3/\text{Tp-TAPB}$; (g) Pore size distributions estimated by NLDFT of $40\text{WO}_3/\text{Tp-TAPB}$

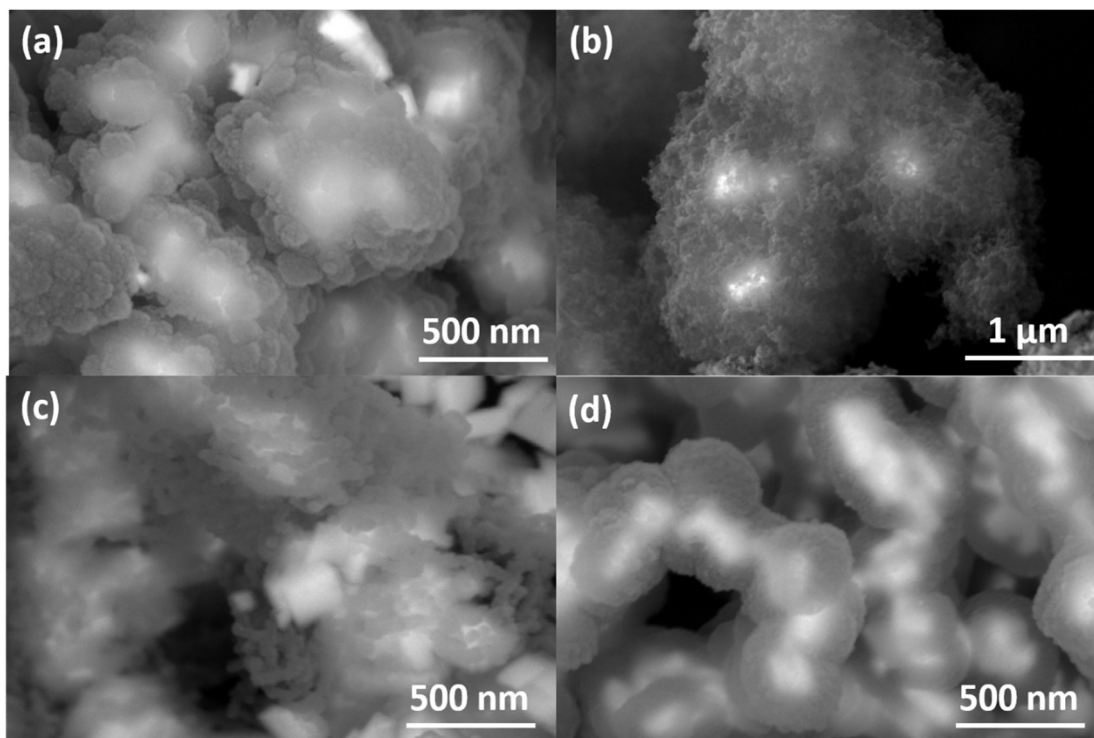


Fig. S3. SEM and image of (a) 5WO₃/Tp-TAPB; (b) 10WO₃/Tp-TAPB; (c) 30WO₃/Tp-TAPB; (d) 40WO₃/Tp-TAPB

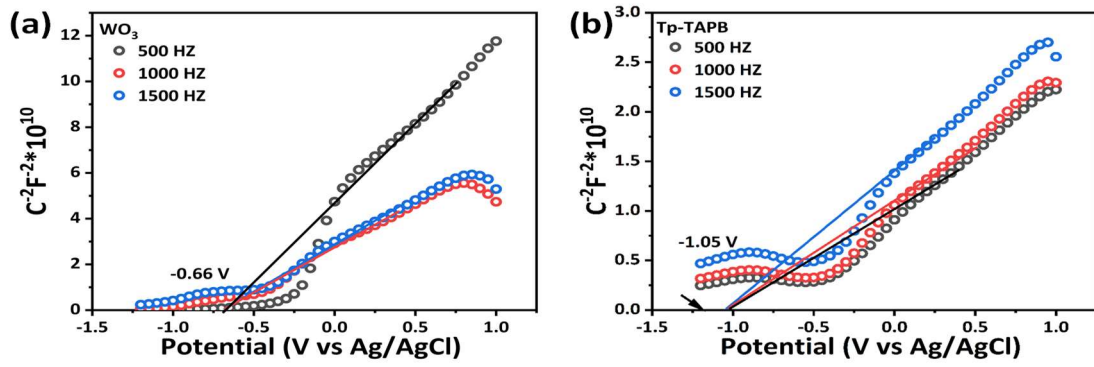


Fig. S4. (a) Mott-Schottky plots of WO_3 , (b) Mott-Schottky plots of Tp-TAPB

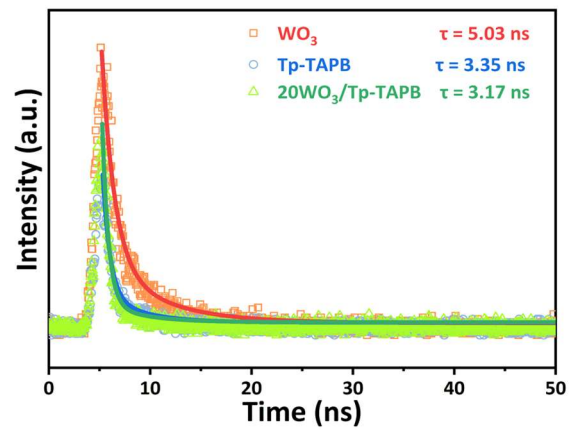


Fig. S5. TRPL spectra of WO_3 , Tp-TAPB and $20WO_3/Tp-TAPB$

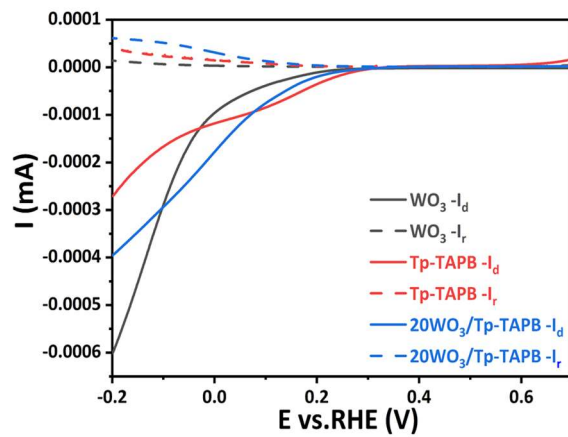


Fig. S6. LSV curves of WO_3 , Tp-TAPB and $20\text{WO}_3/\text{Tp-TAPB}$ obtained using an RRDE with a rotating speed of 1600 rpm

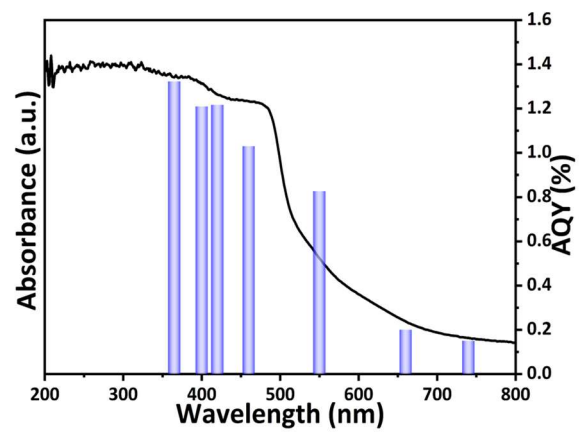


Fig. S7. AQY values of 20WO₃/Tp-TAPB under different monochromatic light irradiation and UV-visible DRS spectra

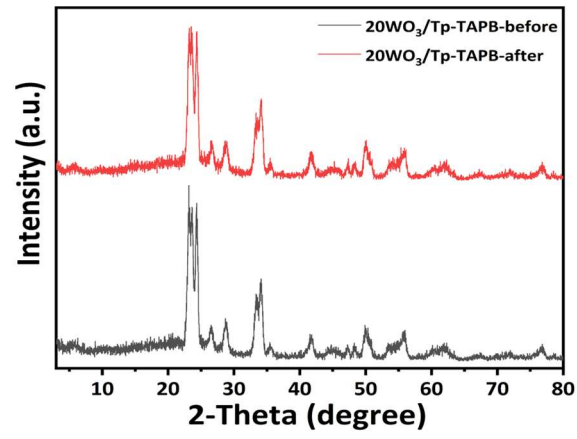


Fig. S8. XRD pattern of 20WO₃/Tp-TAPB before and after photocatalytic reaction

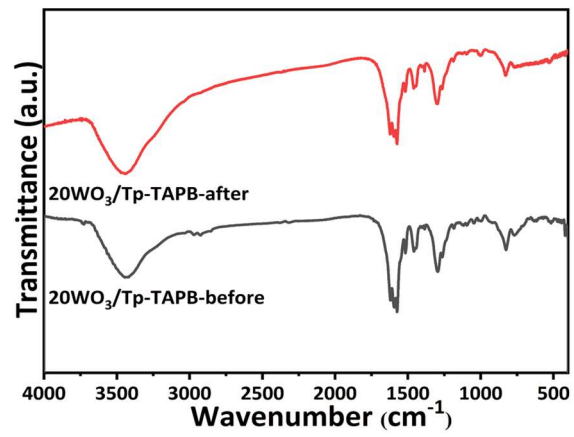


Fig. S9. FT-IR survey spectra of 20WO₃/Tp-TAPB before and after photocatalytic reaction

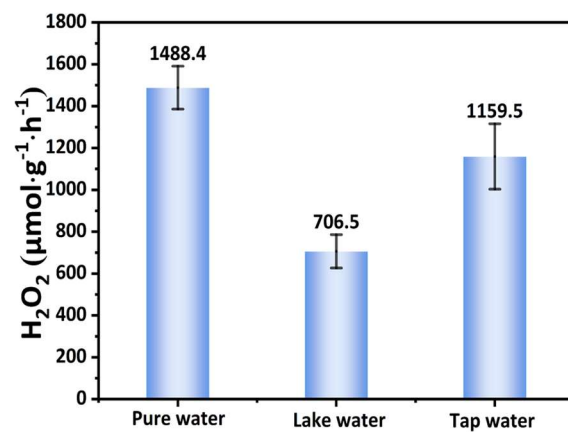


Fig. S10. The photocatalytic H₂O₂ production over 20WO₃/Tp-TAPB in tap water and lake water (Condition: 5 mg catalyst, 10 mL H₂O + O₂, LED light ($\lambda > 420$ nm))

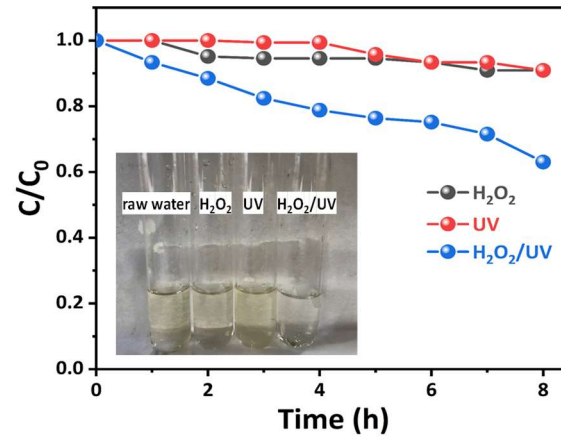


Fig. S11. The removal efficiency of COD by H₂O₂, UV and H₂O₂/UV (Raw water COD concentration = 110mg/L, 50 mL Coking water + 5 mL H₂O₂)

Table S1. BET surface areas of the samples

Samples	BET surface areas (m ² /g)
WO ₃	7.02
Tp-TAPB	608.26
5WO ₃ / Tp-TAPB	547.15
10WO ₃ / Tp-TAPB	489.89
20WO ₃ / Tp-TAPB	306.60
30WO ₃ / Tp-TAPB	301.89
40WO ₃ / Tp-TAPB	246.37

Table S2. The WO₃ composition of WO₃/Tp-TAPB

Samples	WO ₃ composition (%)
5WO ₃ /Tp-TAPB	3.08
10WO ₃ /Tp-TAPB	11.62
20WO ₃ /Tp-TAPB	21.92
30WO ₃ /Tp-TAPB	26.29
40WO ₃ /Tp-TAPB	34.60

Table S3. Comparison of photocatalytic H₂O₂ production with other photocatalysts

Samples	H ₂ O ₂ generation rate	Solvent	References
CHF-DPDA	256 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	H ₂ O	1
TPB-DMTP-COF	2882 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	H ₂ O	2
COF-TfpBpy	1970 $\mu\text{mol}\cdot\text{L}^{-1}\cdot\text{h}^{-1}$	H ₂ O	3
TiCOF-spn	489.94 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	H ₂ O	4
HEP-TAPT-COF	87.50 $\mu\text{mol}\cdot\text{h}^{-1}$	H ₂ O	5
TAPB-PDA-H ₂	1349.3 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	H ₂ O	6
TAPB-PDA-OH	1841.3 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	H ₂ O	6
TAPB-PDA-OCH ₃	869.1 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	H ₂ O	6
TAPB-PDA-CH ₃	857.8 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	H ₂ O	6
TAPD-(Me) ₂ COF	234.52 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	H ₂ O + EtOH	7
CTF-NS-5BT	1630 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	H ₂ O + BA	8
TpMa/CN-5	880.46 $\mu\text{mol}\cdot\text{L}^{-1}$	H ₂ O + IPA	9
TiO ₂ /COF	740 $\mu\text{mol}\cdot\text{L}^{-1}\cdot\text{h}^{-1}$	furfuryl alcohol	10
ZnIn ₂ S ₄ /TpPa-1	516 $\mu\text{mol}\cdot\text{L}^{-1}$ (2h)	H ₂ O + EtOH	11
CDs@CTFs	535.41 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	H ₂ O	12
ZnO/COF	2443 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	H ₂ O + EtOH	13
MnIn ₂ S ₄ /WO ₃	1188 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	H ₂ O	14
ZnO/WO ₃	6788 $\mu\text{mol}\cdot\text{L}^{-1}\cdot\text{h}^{-1}$	H ₂ O + EtOH	15
KCN10/WO ₃	1.33 mmol/L	H ₂ O + EtOH	16
Au/WO ₃	544 μM	H ₂ O + MeOH	17
Tp-TAPB	528.3 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	H ₂ O	This work
20WO ₃ /Tp-TAPB	1488.4 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	H ₂ O	This work

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