

Supporting Information

Effect of oxygen vacancies and crystal phases in defective Pt/ZrO_{2-x} on its photocatalytic activity toward hydrogen production

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Table S1. Lattice parameters of the models for DFT calculation.

system	a	b	c	α	β	γ
Monoclinic (111) slab	7.3207	7.4378	15.2518	90	90	114.68
Tetragonal (101) slab	6.4049	3.6400	19.9195	90	90	90

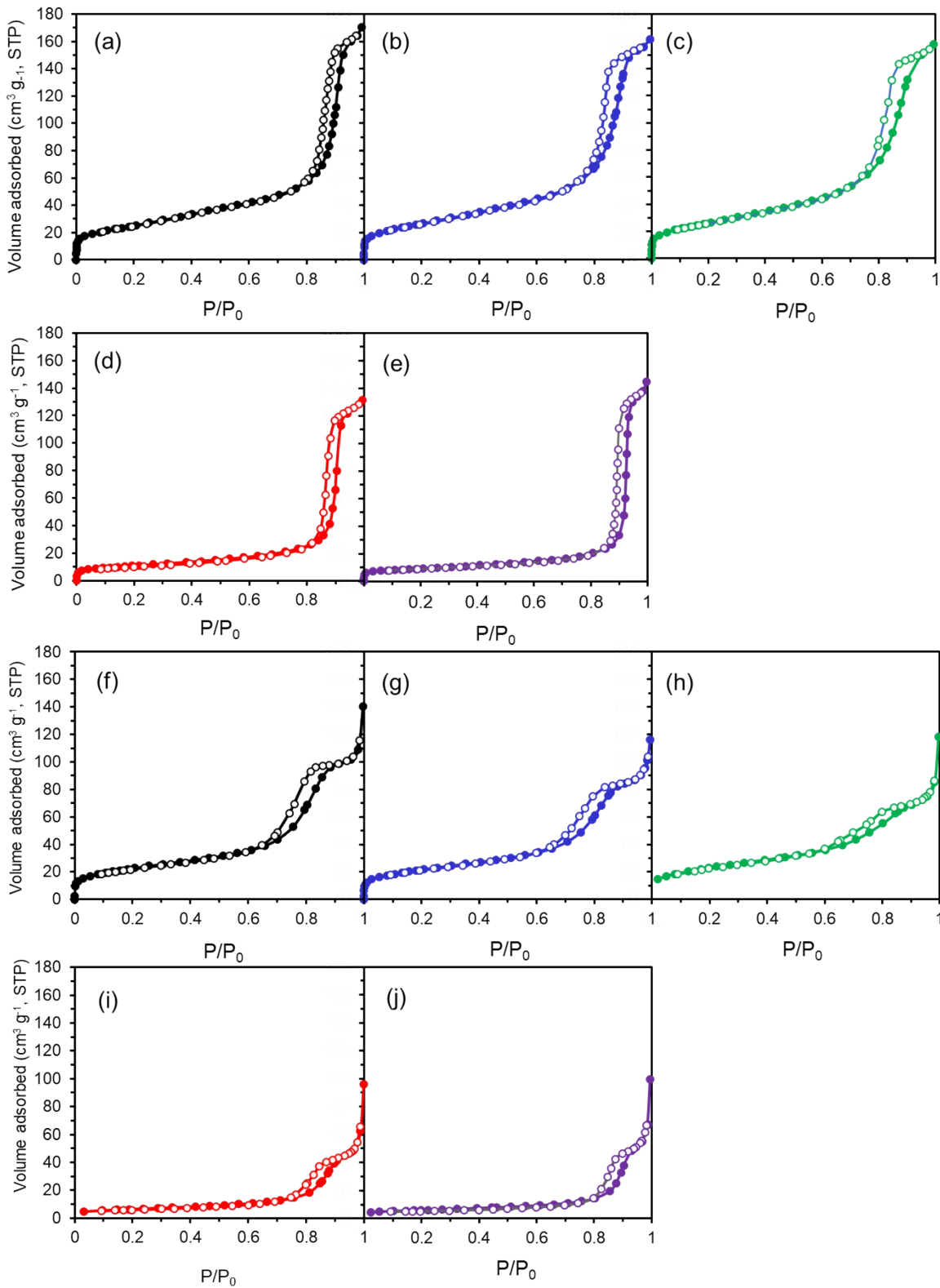


Figure S1. N_2 adsorption/desorption isotherms of (a) as-synthesized m - ZrO_2 , (b) Pt/m - $ZrO_{2-x}200$, (c) Pt/m - $ZrO_{2-x}400$, (d) Pt/m - $ZrO_{2-x}600$, (e) Pt/m - $ZrO_{2-x}600_6h$, (f) as-synthe

sized t - ZrO_2 , (g) Pt/t - $ZrO_{2-x}200$, (h) Pt/t - $ZrO_{2-x}400$, (i) Pt/t - $ZrO_{2-x}600$, and (j) Pt/t - $ZrO_{2-x}600_6h$.

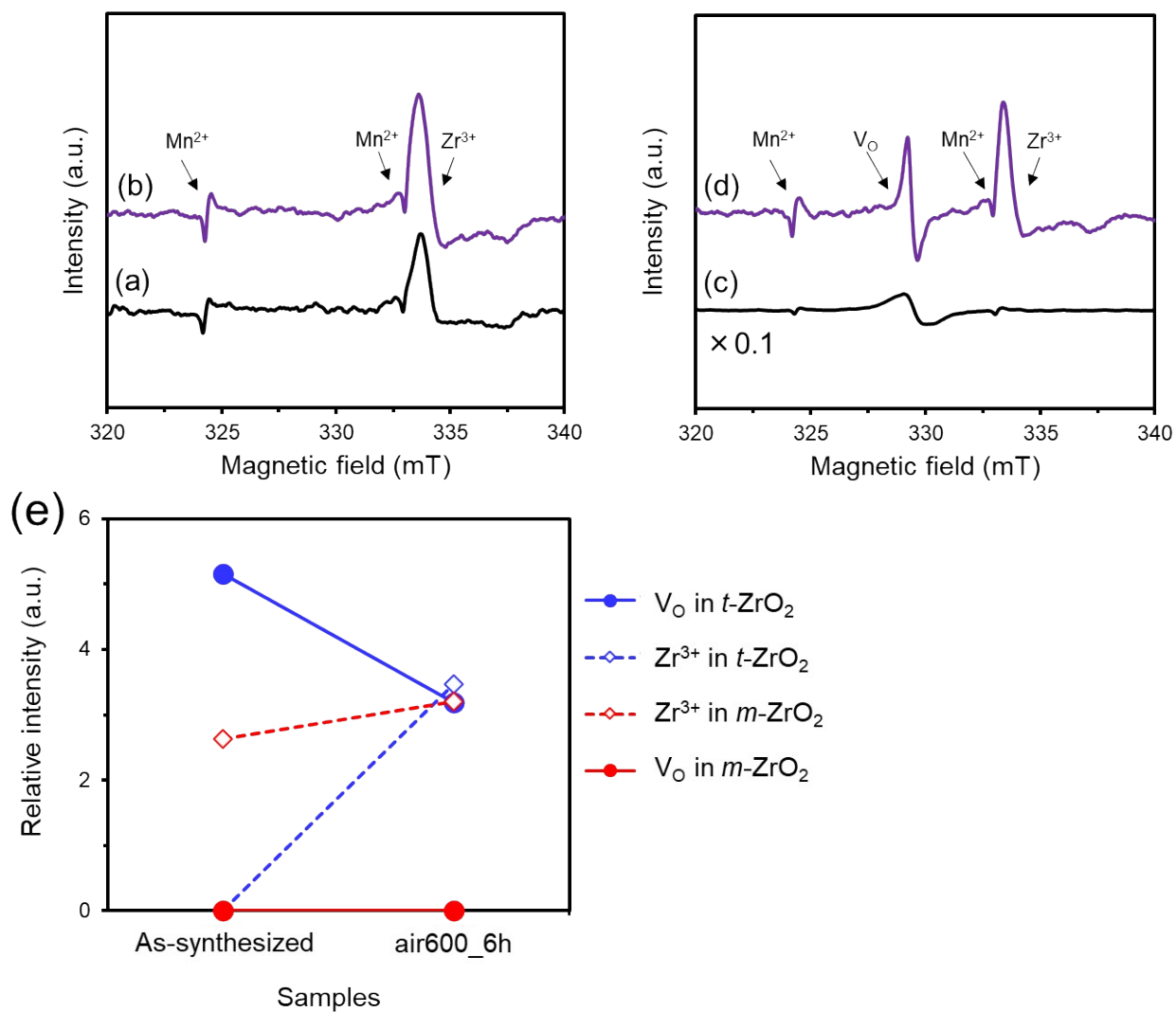


Figure S2. ESR spectra of (a) as-synthesized *m*-ZrO₂, (b) *m*-ZrO₂ after calcination, (c) as-synthesized *t*-ZrO₂, and (d) *t*-ZrO₂ after calcination at 600 °C for 6 h. (e) Relative intensity of the ESR signals based on Mn²⁺.

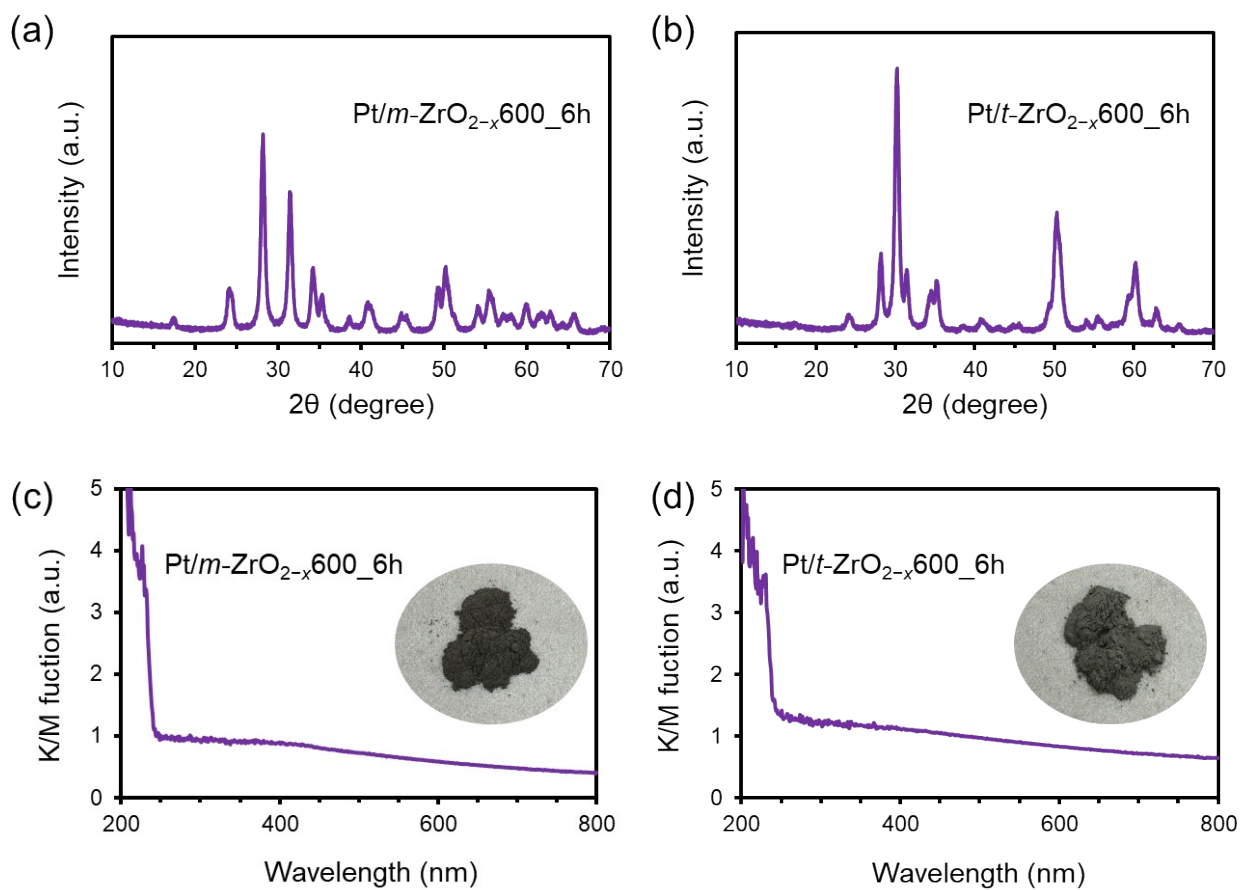


Figure S3. XRD patterns of (a) Pt/m-ZrO_{2-x}600_6h and (b) Pt/t-ZrO_{2-x}600_6h. Diffuse-reflectance spectra and photographs of (c) Pt/m-ZrO_{2-x}600_6h and (d) Pt/t-ZrO_{2-x}600_6h.

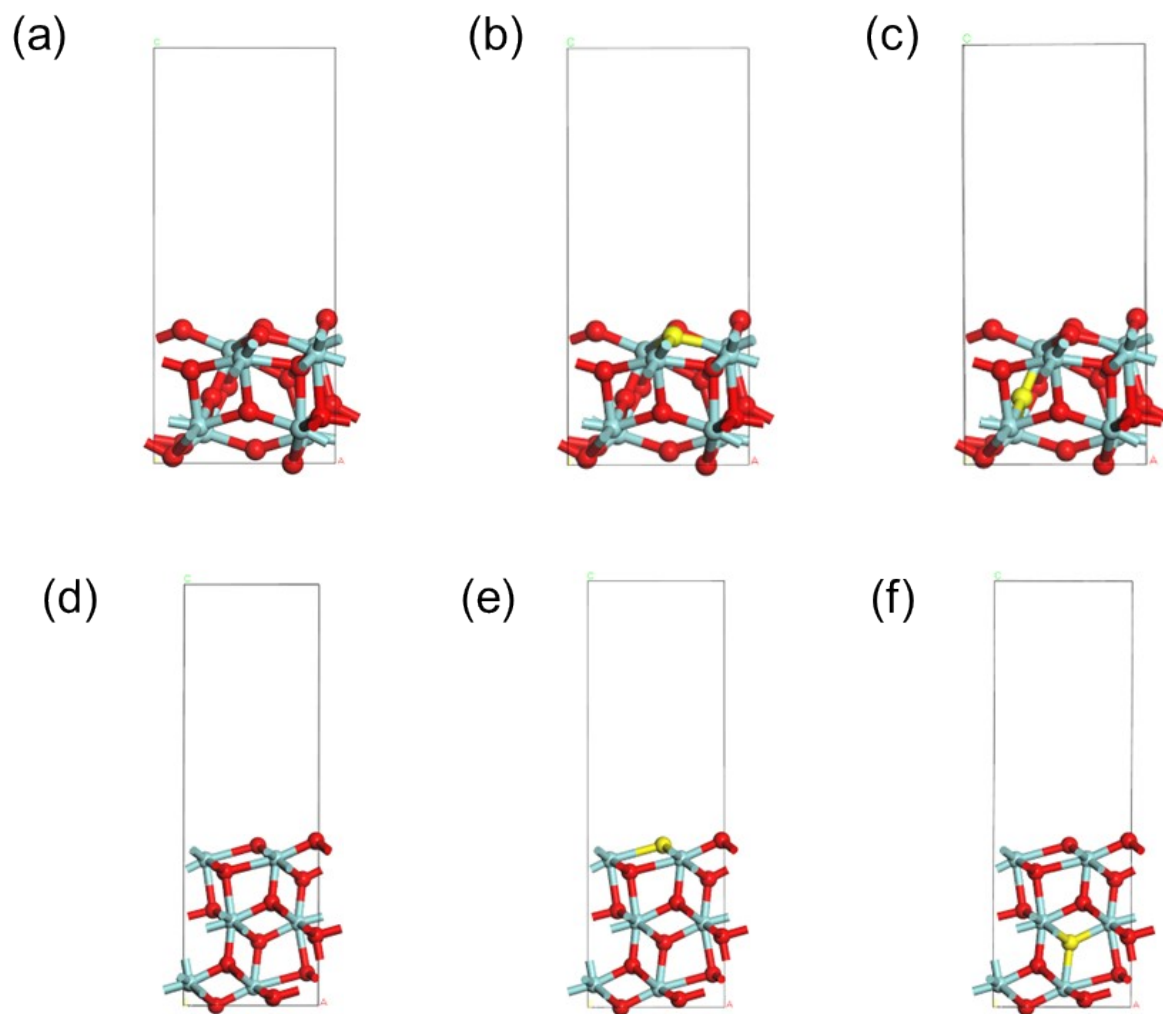


Figure S4. Models of monoclinic ZrO₂ (111) with (a) no defect, (b) a surface V_O, (c) a bulk V_O, and tetragonal ZrO₂ (101) with (d) no defect, (e) a surface V_O, (f) a bulk V_O. Red and blue balls display O and Zr atom, respectively. Yellow balls are O atoms removed to form V_O prior to the calculation.