

Dual-channel synthesis of H₂O₂ via photoelectrocatalytic water oxidation and oxygen reduction over TaON/Ta₃N₅/CuI/Cu foam electrode

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Table S1 Comparison of the activity of different catalysts for the synthesis of H₂O₂ by different methods

Catalyst	Synthesis method	Reaction pathway	H ₂ O ₂ production rate	reference
ZIF-8/C ₃ N ₄	PC	2e ⁻ ORR and WOR	2641.0 μmol g ⁻¹ h ⁻¹	1
Ni-CAT-CN ₆₀	PC	1e ⁻ ORR and 4e ⁻ WOR	1801.0 μmol g ⁻¹ h ⁻¹	2
C, N co-doped TiO ₂	EC	2e ⁻ WOR	0.3 μmol L ⁻¹ cm ⁻² h ⁻¹	3
α-Fe ₂ O ₃ -GDE	PEC	2e ⁻ ORR	58.8 μmol L ⁻¹ cm ⁻² h ⁻¹	4
(1T-2H)-MoSe ₂ /TiO ₂	PEC	1e ⁻ ORR	38.0 μmol cm ⁻² h ⁻¹	5
SnO _{2-x} -BiVO ₄	PEC	2e ⁻ and 4e ⁻ WOR	48.0 μmol cm ⁻² h ⁻¹	6
Mo-doped BiVO ₄	PEC	2e ⁻ ORR and 4e ⁻ WOR	9.6 μmol cm ⁻² h ⁻¹	7
NiFeO _x /BiVO ₄ -pTTh	PEC	2e ⁻ ORR	1.1 mmol L ⁻¹ cm ⁻² h ⁻¹	8
CuWO ₄	PEC	2e ⁻ ORR and WOR	0.7 mmol cm ⁻² h ⁻¹	9
WO ₃ /FPC	PEC	2e ⁻ ORR	0.2 mmol L ⁻¹ cm ⁻² h ⁻¹	10

Table S2 The fitting parameters of the R(C(RW)) equivalent circuit model

Photoanodes	Rs	CPE-T	CPE-P	R _p
	($\Omega \text{ cm}^2$)	($\Omega^{-1} \text{S}^n \text{ cm}^{-2}$) $\times 10^3$	($\Omega^{-1} \text{S}^n \text{ cm}^{-2}$) $\times 10^3$	($\Omega \text{ cm}^2$)
CuI/Cu	3.412	5.5718	468.41	39.792
TN/Cu	4.019	2.2699	539.27	37.817
CIT-1	3.6955	0.1342	41.517	16.892
CIT-2	4.940	2.383	494.01	70.398
CIT-3	3.397	13.899	409.91	22.542

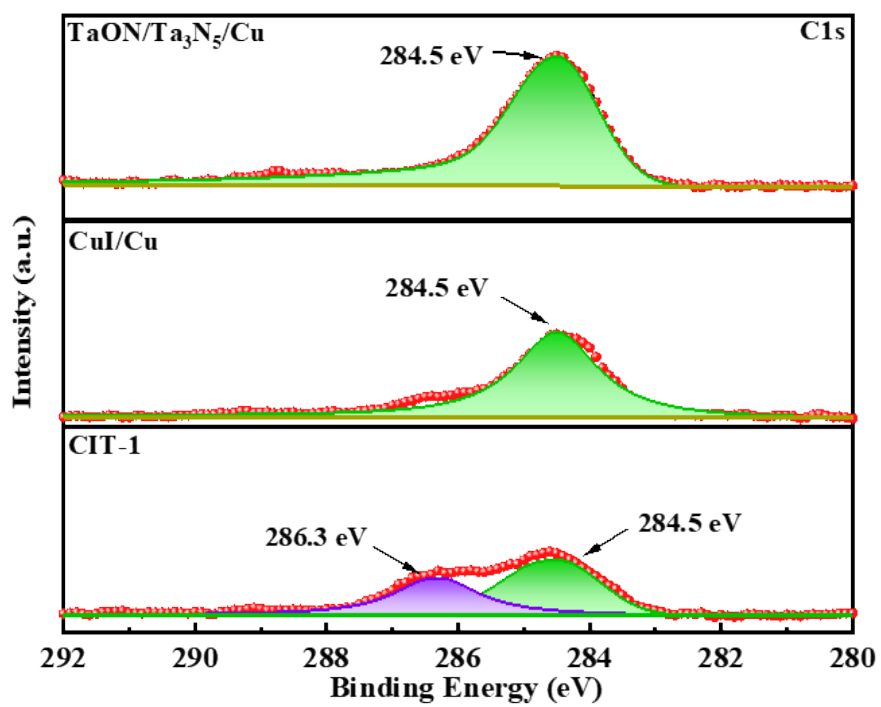


Fig. S1 C 1s high resolution XPS spectra of TaON/Ta₃N₅/Cu, CuI/Cu, and CIT-1.

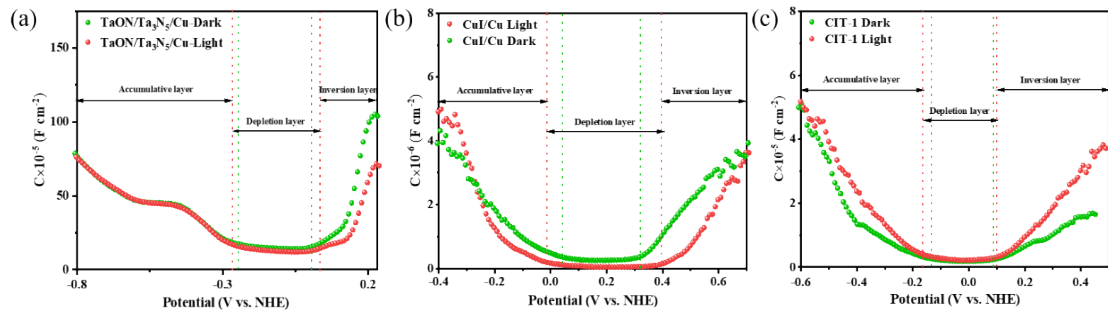


Fig. S2 The C-E curves of (a) TaON/Ta₃N₅/Cu, (b) CuI/Cu, and (c) CIT-1 under dark and light.

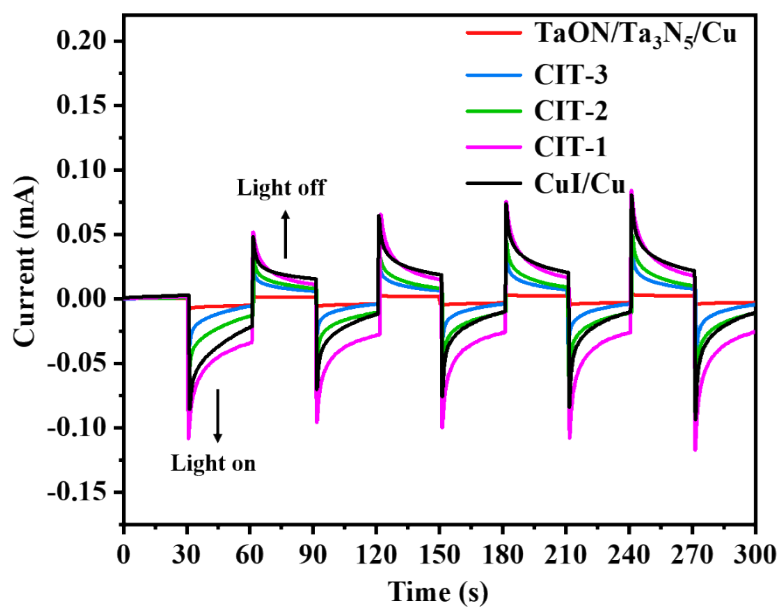


Fig. S3 The J-t curves of TaON/Ta₃N₅/Cu, CuI/Cu, CIT-1, CIT-2 and CIT-3.

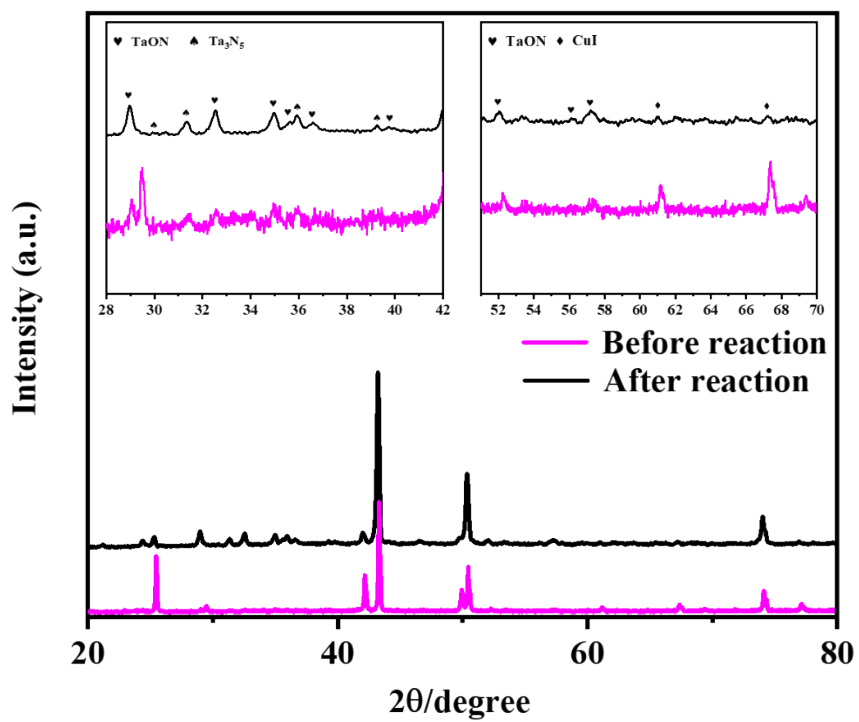


Fig. S4 The XRD patterns of CIT-1 before and after reaction.

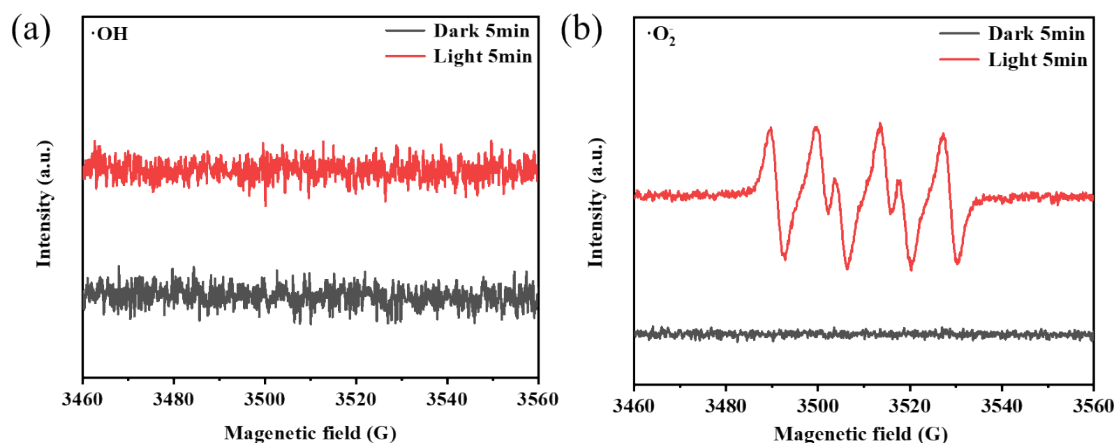


Fig. S5 EPR spectra of CIT-1 for (a) $\cdot\text{OH}$ and (b) $\cdot\text{O}_2^-$ under dark and light.

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